

Supplementary Materials for

Universal and scalable synthesis of ultrahigh-density single-atom libraries on metal oxides

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Supplementary Methods

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Supplementary References

1. Syntheses

1.1 Synthesis of 2D-Fe₂O₃

A piece of iron foil (1 cm², 0.25 mm thick, 99.99%, Alfa) was ultrasonically cleaned in acetone and ethanol for 10 min for degreasing, followed by dried under a nitrogen stream. To prepare α -Fe₂O₃ nanosheets grown on the iron substrate, the foil was thermally annealed at 430 °C for 2 h in air using a HF-Kejing furnace (Model KSL-1100X), with a heating rate of 10 °C min⁻¹. This thermal annealing process resulted in the formation of nano-leaf-like structures across the entire sample surface.

1.2 Synthesis of 1D-Cu₂O

The copper foil (1 cm², 0.1 mm thick, 99.99%, Alfa) was ultrasonically cleaned in acetone and ethanol for 10 min for degreasing, followed by dried under a nitrogen stream. The cleaned foil was then electrochemically anodized in a 3 M KOH solution at a current density of 10 mA cm⁻² until the potential reached 2 V to form Cu(OH)₂ nanowires. Subsequently, the as-grown Cu(OH)₂ nanowires were converted into one-dimensional Cu₂O by annealing in an Ar atmosphere at 550 °C for 5 h with a heating rate of 10 °C min⁻¹.

1.3 Synthesis of 2D-ZnIn₂S₄

The precursor solution was prepared by dissolving ZnCl₂ (0.2045 g, 99.95%, Alfa), InCl₃·4H₂O (0.8795 g, 99.9%, Aladdin), and thiourea (0.4565 g, 99%, Aladdin) in 150 mL of ultrapure water. Subsequently, 20 mL of the solution and a clean FTO substrate (conductive side down) were loaded into a Teflon-lined autoclave, which was heated at 160 °C for 6 h. The resulting sample was collected, washed with water and ethanol, and vacuum-dried at 50 °C for 24 h.

1.4 Synthesis of 2D-C₃N₄

Bulk 2D-C₃N₄ was synthesized by directly annealing 10 g of urea (99.9%, Aladdin) in a crucible at 550 °C for 3 h under ambient air.

1.5 Synthesis of 3D-Fe₂O₃

Following the preparation of a precursor solution from FeCl₃·6H₂O (0.15 M, 99%, Aladdin), NaNO₃ (1 M, 99%, Aladdin), and concentrated HNO₃ (150 μ L) in 100 mL of ultrapure water, a clean FTO substrate (conductive side down) was placed in a Teflon-lined autoclave with 20 mL of the solution. The autoclave was heated at 100 °C for 4 h. The obtained product was washed with ultrapure water/ethanol and dried.

Subsequently, the FeOOH sample was then subjected to annealing in argon at 600 °C for 1 h, followed by ramping to 750 °C at 2 °C min⁻¹ and holding for 20 min before natural cooling.

1.6 Synthesis of 3D-TiO₂

Following the preparation of a precursor solution from tetrabutyl titanate (1 mL, 99%, Aladdin) and NaNO₃ (1 M, 99%, Aladdin) in a mixture of ultrapure water (30 mL) and concentrated HCl (30 mL), a clean FTO substrate (conductive side down) was subjected to a hydrothermal treatment with 20 mL of the solution at 150 °C for 12 h. The resulting product was washed with deionized water, and subsequently annealed at 450 °C for 1 h in air.

2. Characterizations

2.1 Structure and morphology

The crystal structure and composition were characterized by X-ray diffraction (XRD, Rigaku RINT-2000, Cu K α radiation at 40 kV and 40 mA) and X-ray photoelectron spectroscopy (XPS, ESCALAB 250xi, Thermo Fisher Scientific). The morphology was examined using field-emission scanning electron microscopy (FE-SEM, Supra 55, Zeiss, Germany) and spherical aberration-corrected high-resolution scanning transmission electron microscopy (JEM GRAND ARM 300F, Japan). Electron paramagnetic resonance (EPR) measurements were recorded at a low temperature of -150 °C using a JES-FA200 spectrometer.

2.2 XAFS Measurements

X-ray absorption fine structure (XAFS) measurements were performed with Si (111) crystal monochromators at the BL14W Beam line at Shanghai Synchrotron Radiation Facility (SSRF) (Shanghai, China). Before the analysis at the beamline, samples were placed into aluminum sample holders and sealed using Kapton tape film. The obtained XAFS data was processed in Athena (version 0.9.26) for background, pre-edge line and post-edge line calibrations. Then Fourier transformed fitting was carried out in Artemis (version 0.9.26). The k^3 weighting, k -range of 3~11.5 Å⁻¹ and R range of 1-4 Å were used for the fitting of UHL-SACs/Fe₂O₃. For Wavelet Transform analysis, the $\chi(k)$ exported from Athena was imported into the Hama Fortran code. The parameters were listed as follow: R range, 0-4 Å. k range, 0-16 Å⁻¹. k weight, 2. and Morlet function with $\kappa=10$, $\sigma=1$ was used as the mother wavelet to provide the overall distribution.

2.3 Electrochemical performance measurements

The electrochemical OER and HER performances of the as-prepared W_{PM}/C_3N_4 and W_M/C_3N_4 samples were evaluated in a standard three-electrode system using a 0.5 M H_2SO_4 electrolyte. The measurements were conducted with a CHI 760E electrochemical workstation, and all recorded potentials were converted to the reversible hydrogen electrode (RHE) scale.

2.4 Density functional theory calculations

All density functional theory (DFT) calculations were performed using the Vienna Ab-initio Simulation Package (VASP).¹ The electron-ion interactions were described by the projector-augmented wave (PAW) potentials, which accurately describe the core-valence electron interactions.² The exchange-correlation energy was calculated within the generalized gradient approximation (GGA) using the Perdew-Burke-Ernzerhof (PBE) functional, which is widely adopted for solid-state calculations due to its reliable performance in describing the electronic structure and structural properties of many materials.³ To account for the strong correlation effects of d electrons, the DFT+U correction was employed. The effective U value for the Fe elements was set to 4 eV, which was chosen based on previous literatures or calibration calculations.⁴ Additionally, the DFT+D3 correction was incorporated to consider the long-range van der Waals (vdW) interactions, which is crucial for systems involving weak intermolecular or interfacial interactions.⁵ For the calculation of density of states (DOS), hybrid density functional theory (HSE06) calculations were performed to improve the accuracy of electronic structure predictions.⁶ The cutoff energy for the plane-wave basis set was set to 500 eV, which was tested to ensure the total energy convergence. For the k-point sampling in the first Brillouin zone, a Monkhorst-Pack grid of $3 \times 3 \times 1$ was used for optimization, and the convergence of total energy with respect to k-point density was verified.⁷

The adsorption energy of the system is defined by:

$$E_{\text{form}} = E_{\text{total}} - E_{\text{sub}} - E_{\text{bulk}}/n \quad (1)$$

where E_{total} is the total energy of SAC on Fe_2O_3 (001), E_{sub} is the energy of Fe_2O_3 (001), E_{bulk} is the energy of metals in bulk, and n is the number of metal atoms in the bulk.

The cohesive energy of the system is defined by:

$$E_{\text{coh}} = (E_{\text{bulk}} - n \cdot E_{\text{atom}})/n \quad (2)$$

where E_{bulk} is the energy of bulk metal and E_{atom} is the energy of single metal in this system. The lower the cohesive energy, the easier it is for atoms to agglomerate.

The binding energy of the system is defined by:

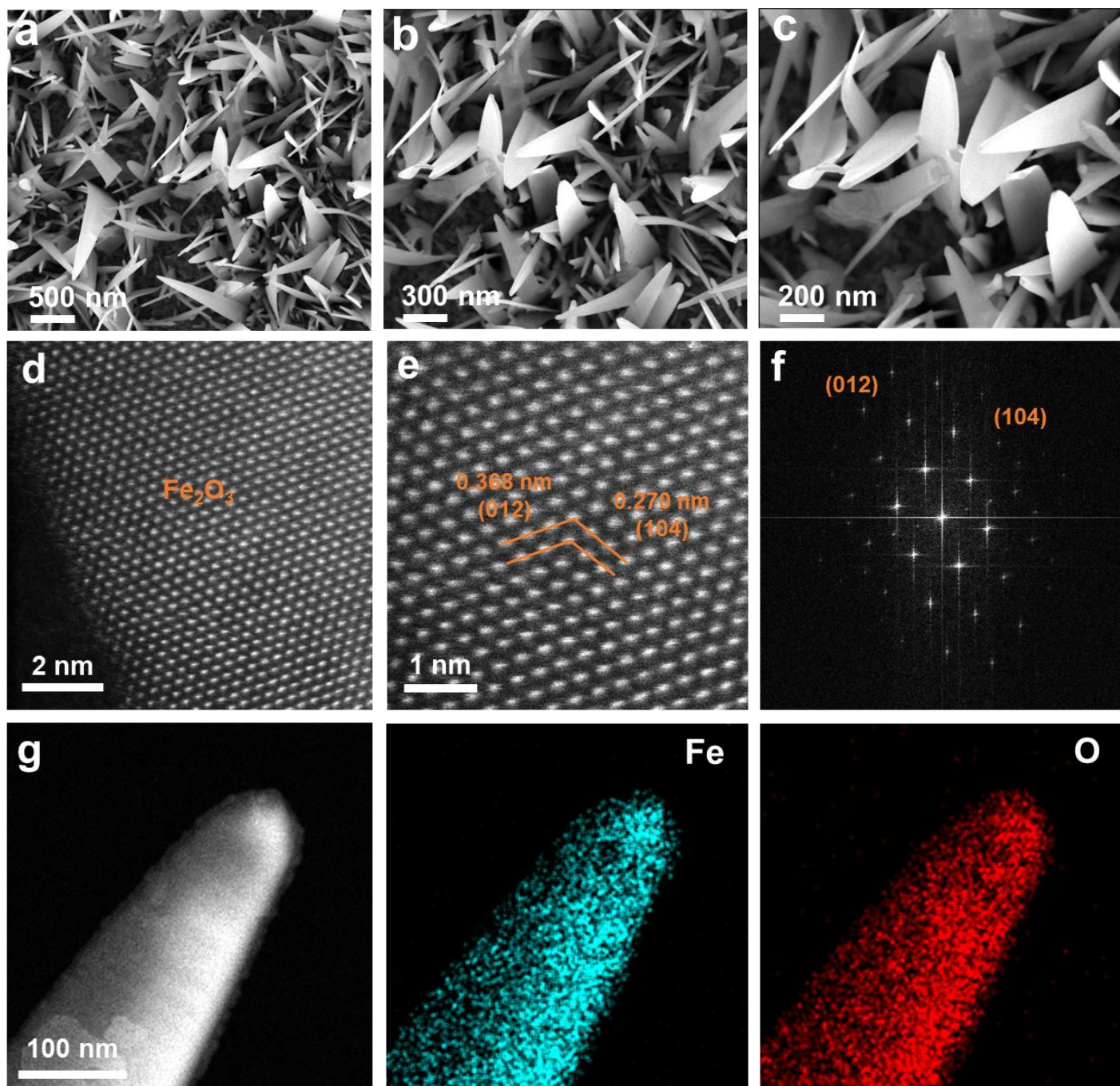
$$E_{\text{bind}} = E_{\text{total}} - E_{\text{substrate}} - E_{\text{SAC}} \quad (3)$$

where $E_{\text{substrate}}$ is the energy of the substrate, and E_{total} is the total energy of the system.

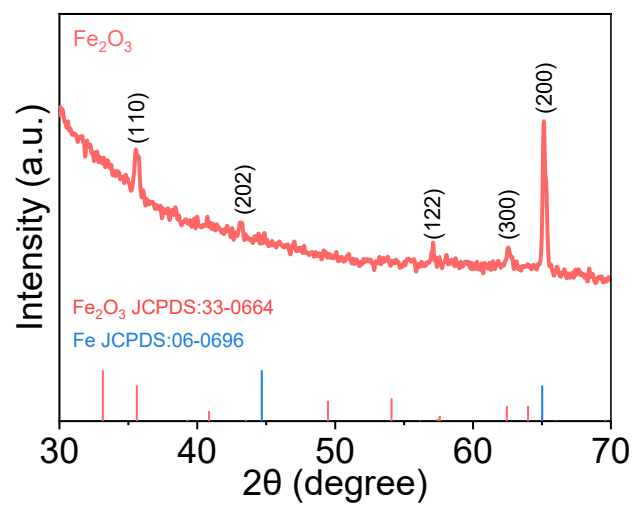
The Gibbs free energy of the system is defined by:

$$\Delta G = \Delta H - T\Delta S \quad (4)$$

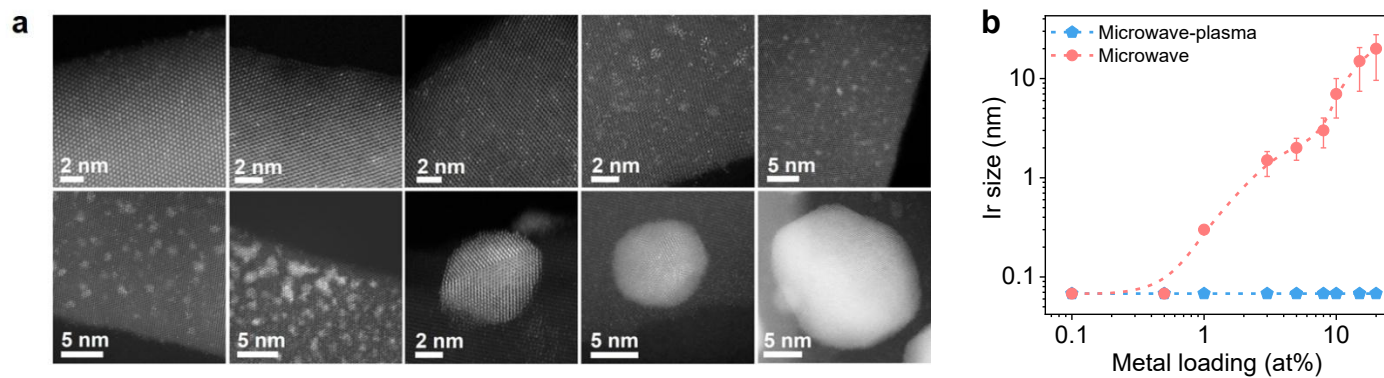
where ΔH is enthalpy, T is the temperature, and ΔS is the entropy of the system. Negative ΔG means that the reaction occurs spontaneously.



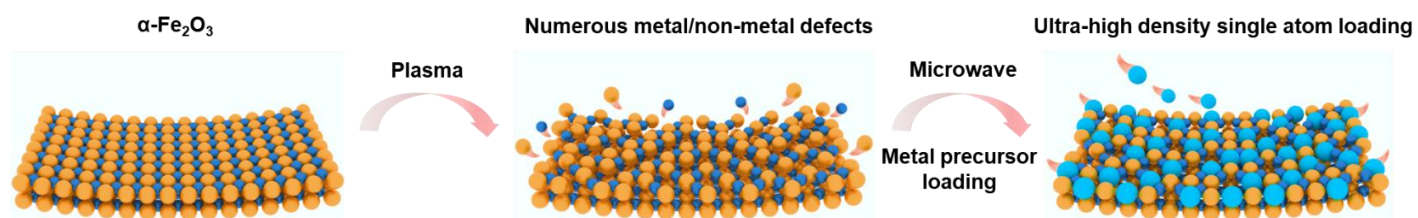
Supplementary Fig. 1 SEM, HAADF-STEM and EDS elemental mappings images of Fe_2O_3 . a-c, SEM images of Fe_2O_3 . d, e, high resolution HAADF-STEM images of Fe_2O_3 . f, FFT pattern. g, elemental mappings of Fe_2O_3 .



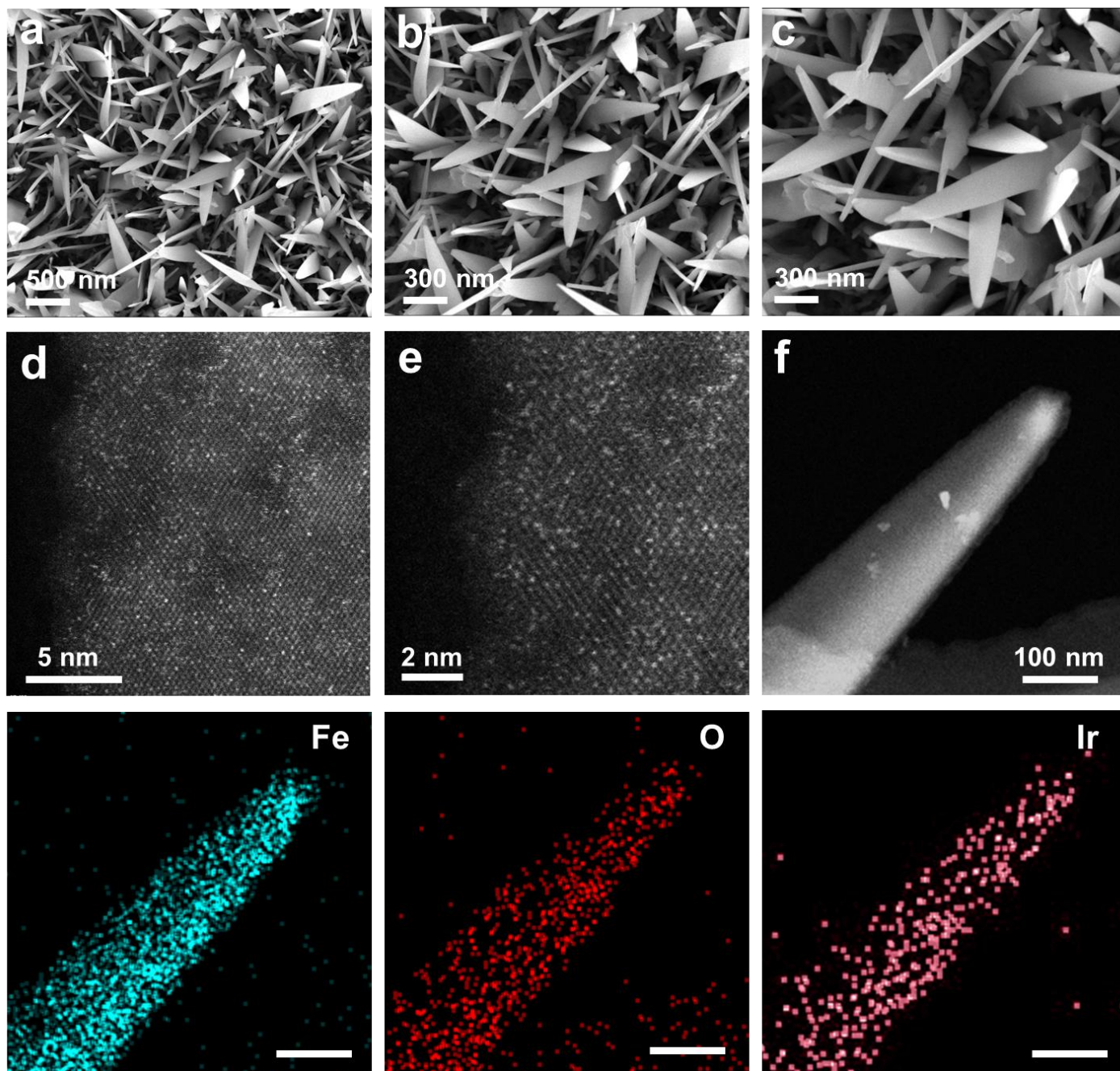
Supplementary Fig. 2 X-ray diffraction (XRD) pattern. XRD pattern of Fe_2O_3 .



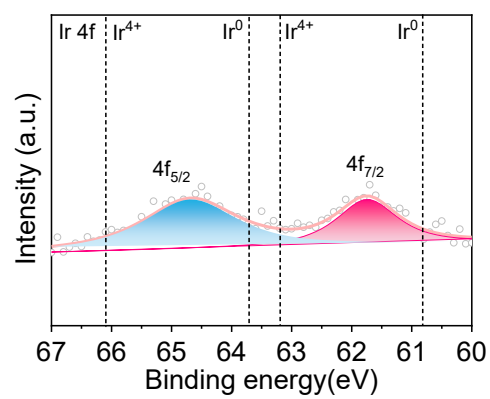
Supplementary Fig. 3 a,b, The HAADF-STEM images (**a**) and the size of Ir (**b**) on Fe_2O_3 with various Ir loadings (0.1-20 wt%).



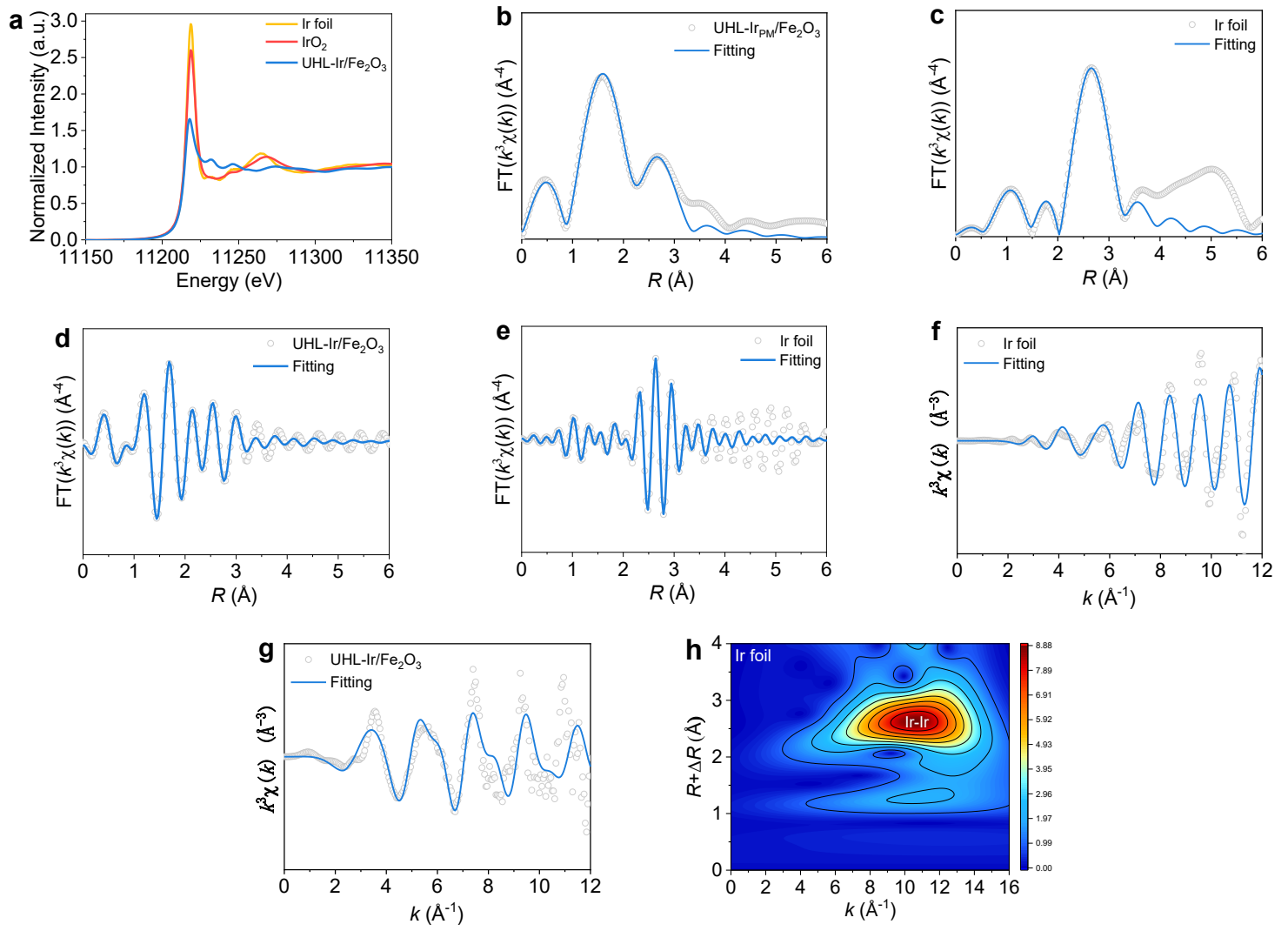
Supplementary Fig. 4 Illustration of the synthesis approach. Schematic illustration for the synthesis of UHL-SACs/Fe₂O₃.



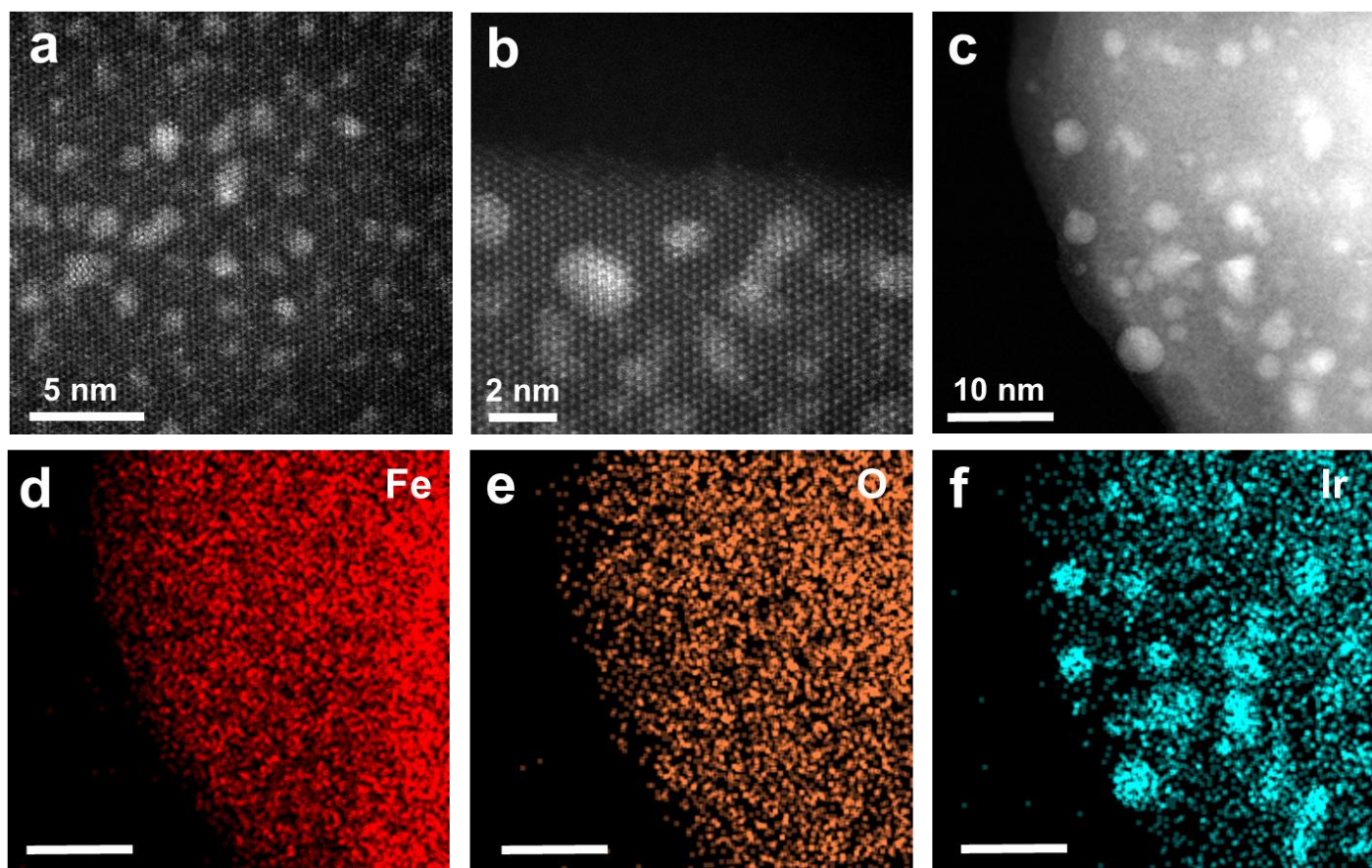
Supplementary Fig. 5 SEM, HAADF-STEM and EDS elemental mappings images of UHL-Ir/Fe₂O₃. a-c, SEM images of UHL-Ir/Fe₂O₃. d, e, high resolution HAADF-STEM images of UHL-Ir/Fe₂O₃. f, elemental mappings of UHL-Ir/Fe₂O₃.



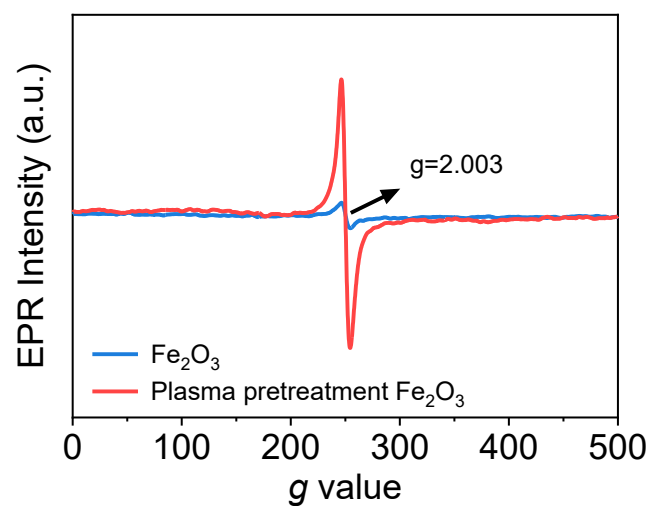
Supplementary Fig. 6 X-ray photoelectron spectroscopy (XPS) characterization. Ir 4f XPS spectra for UHL-Ir/Fe₂O₃.



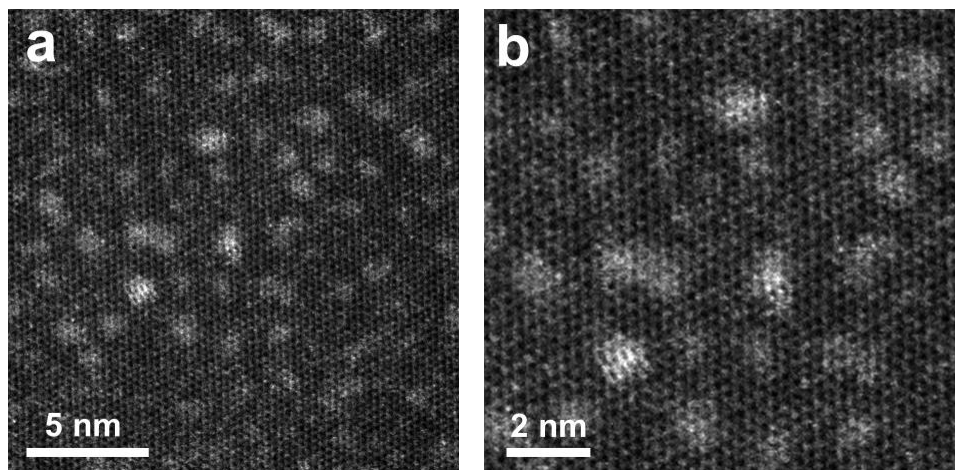
Supplementary Fig. 7 Atom structural characterization of UHL-Ir/Fe₂O₃. **a**, Ir L_3 -edge XANES spectra of UHL-Ir/Fe₂O₃, Ir foil and IrO₂. **b-e**, Ir K -edge EXAFS data and fit for UHL-Ir/Fe₂O₃ in R -space and **f, g**, k -space. **h**, WT-EXAFS plot of UHL-Ir/Fe₂O₃.



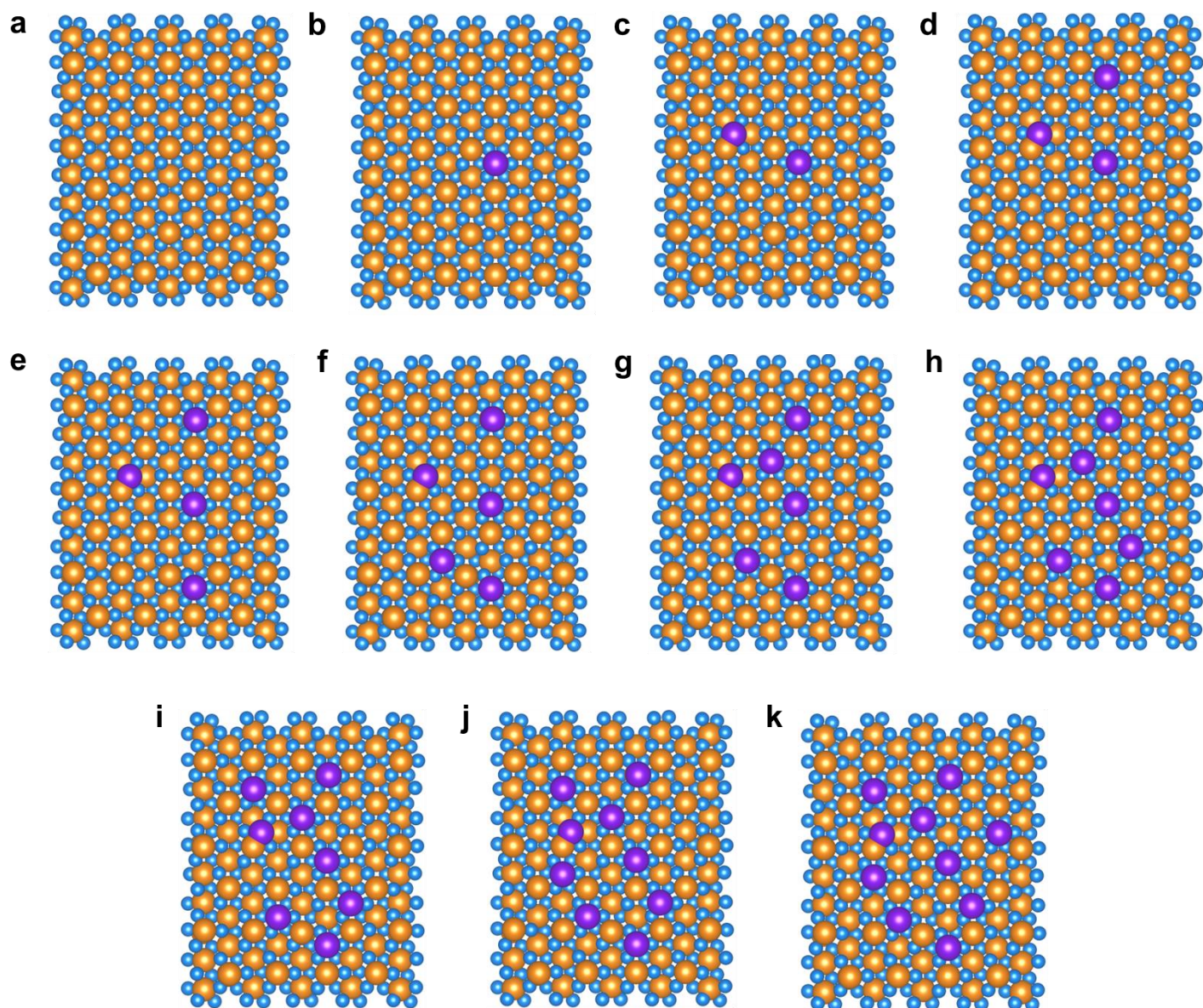
Supplementary Fig. 8 HAADF-STEM and EDS elemental mappings images of UHL-Ir/Fe₂O₃ without plasma pretreatment. a,b, HAADF-STEM images of UHL-Ir/Fe₂O₃ without plasma pretreatment. **c-f,** elemental mappings of UHL-Ir/Fe₂O₃ without plasma pretreatment.



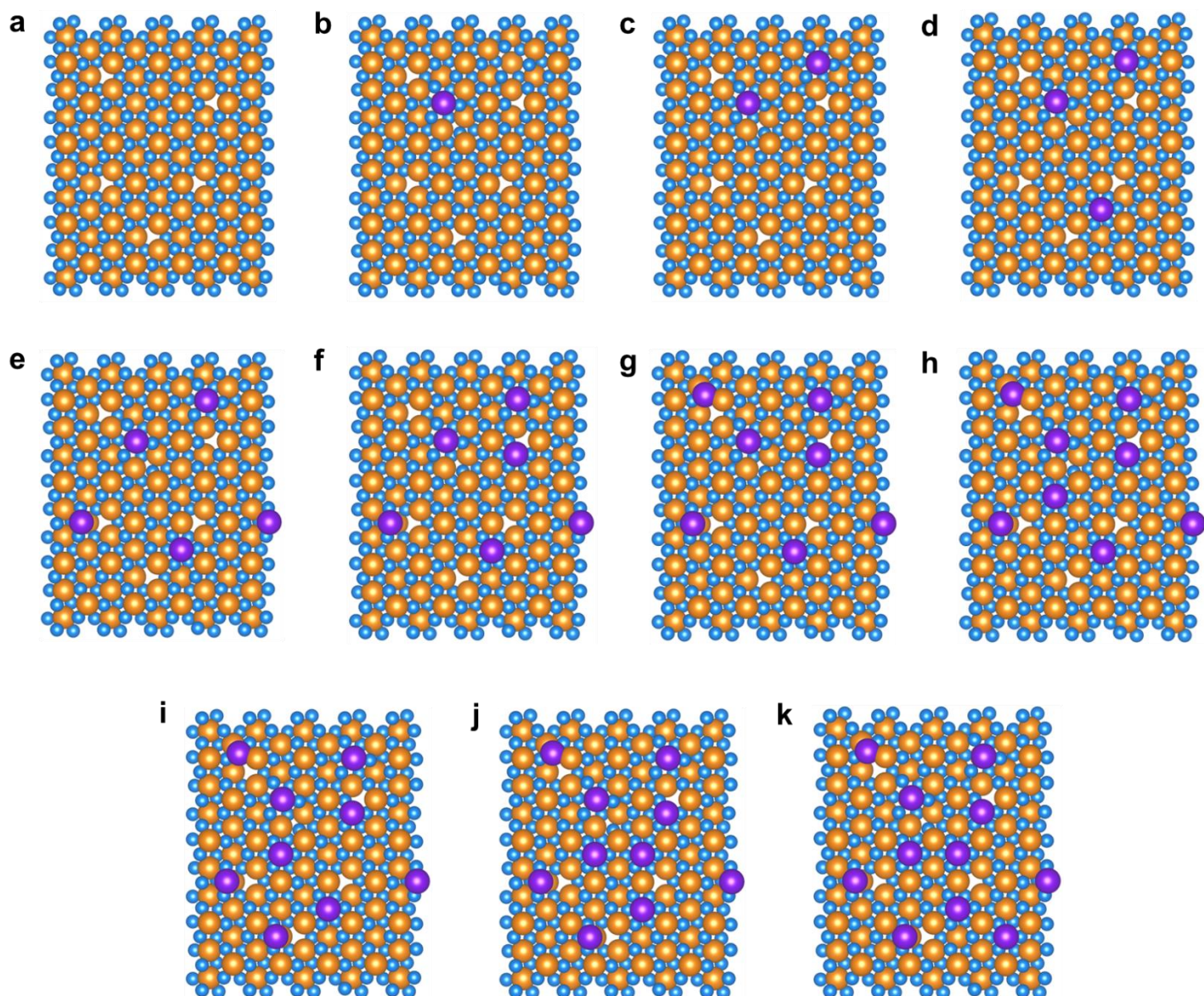
Supplementary Fig. 9 Electron paramagnetic resonance (EPR) spectroscopy. EPR spectra of Fe₂O₃ before and after plasma pretreatment.



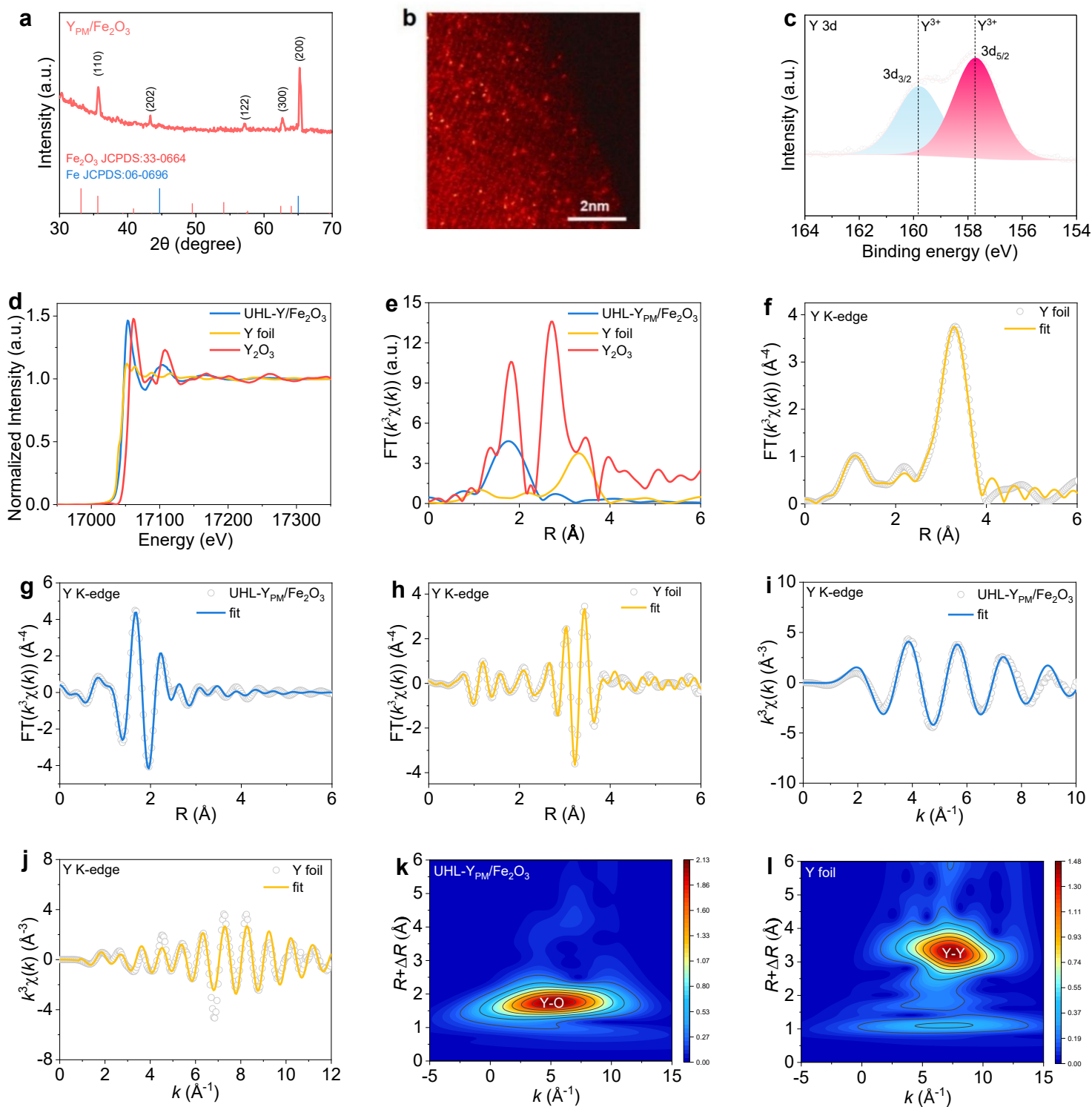
Supplementary Fig. 10 HAADF-STEM images. HAADF-STEM image of Ir precursor loaded onto Fe₂O₃ after plasma pretreatment, following conventional sluggish annealing.



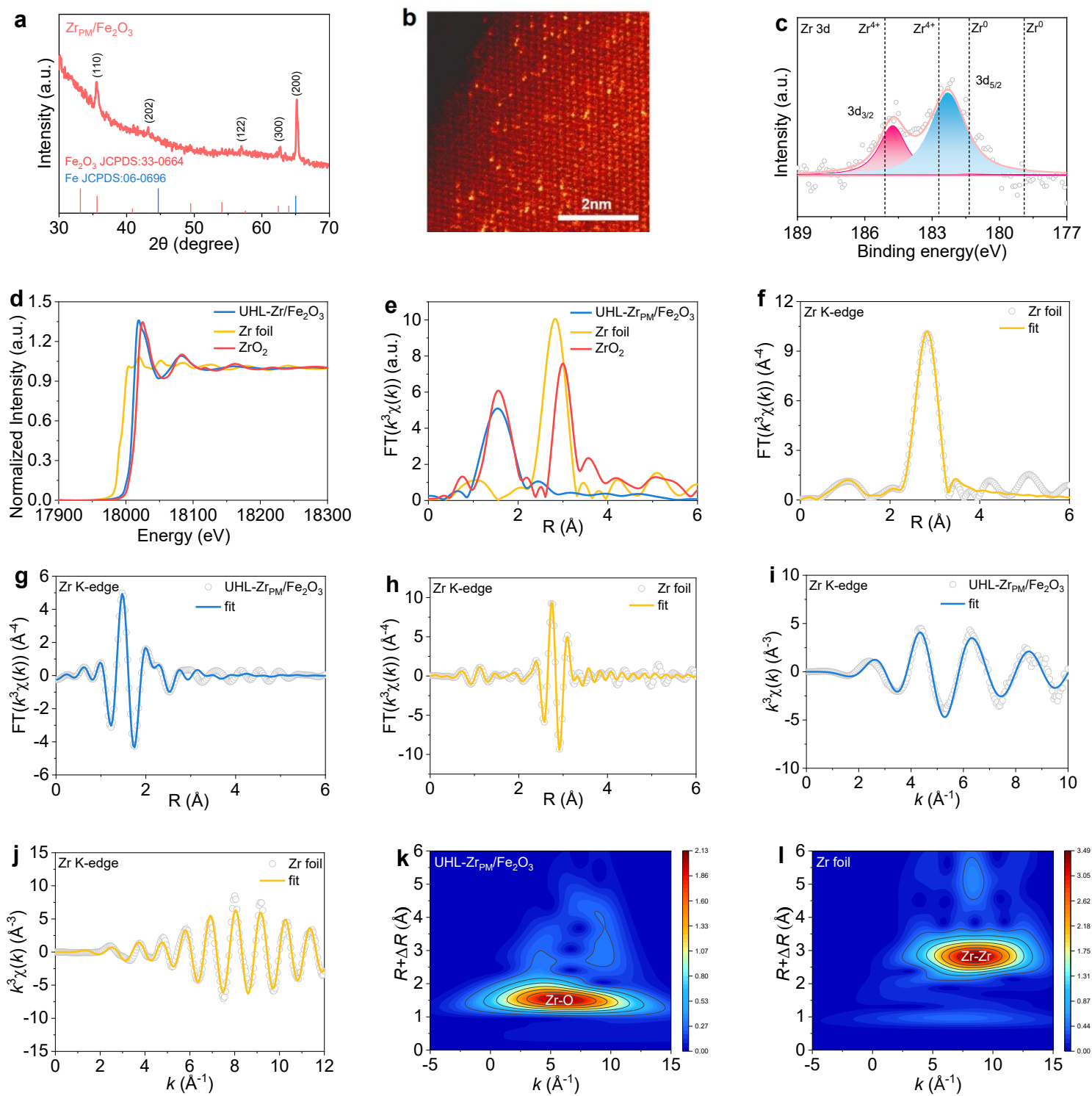
Supplementary Fig. 11 Structural modes. a-k, Structural modes of pristine Fe_2O_3 surfaces loaded with 1-10 dispersed Ir atoms.



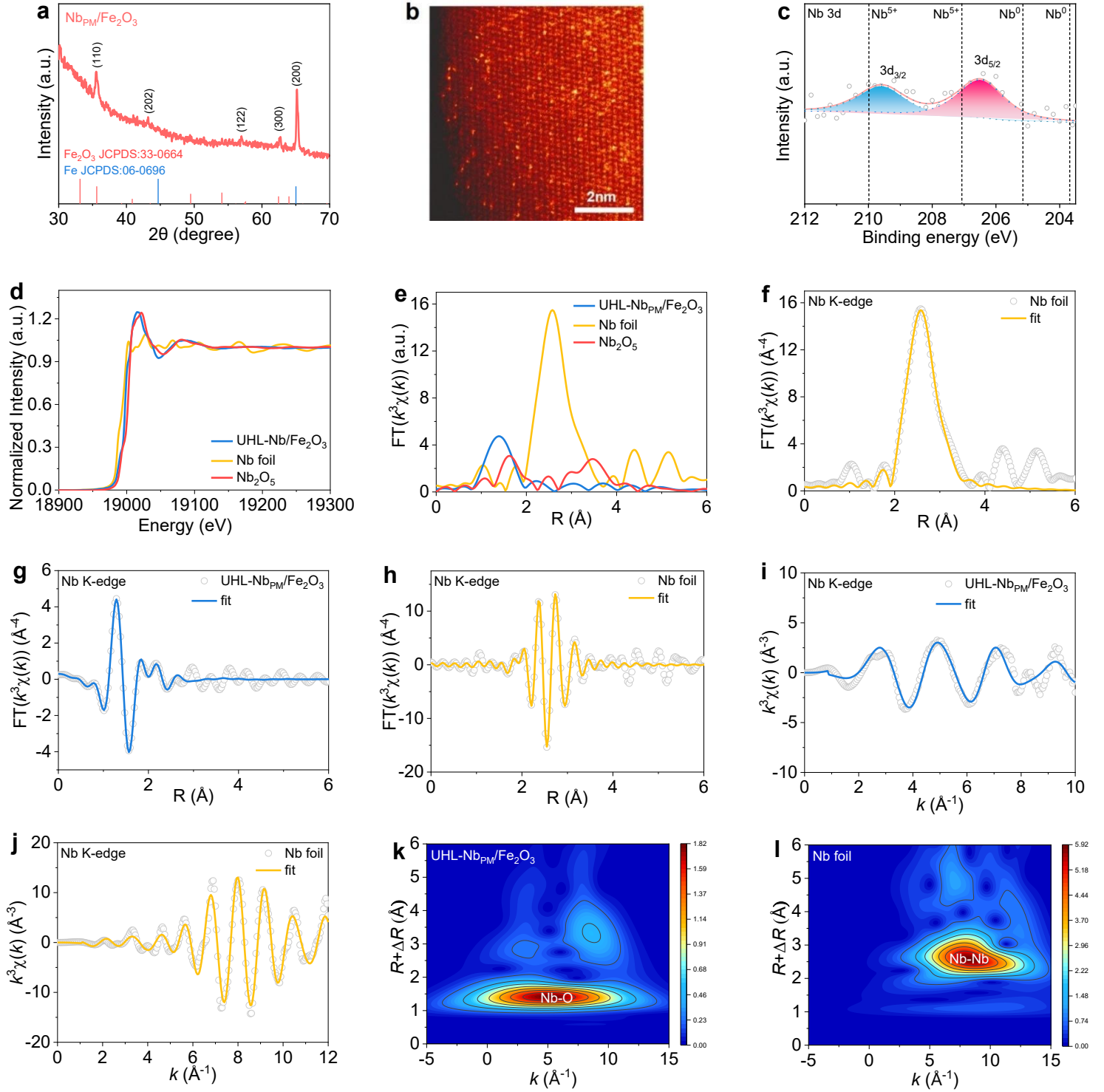
Supplementary Fig. 12 Structural modes. a-k, Structural modes of plasma-treated Fe_2O_3 surfaces loaded with 1-10 dispersed Ir atoms.



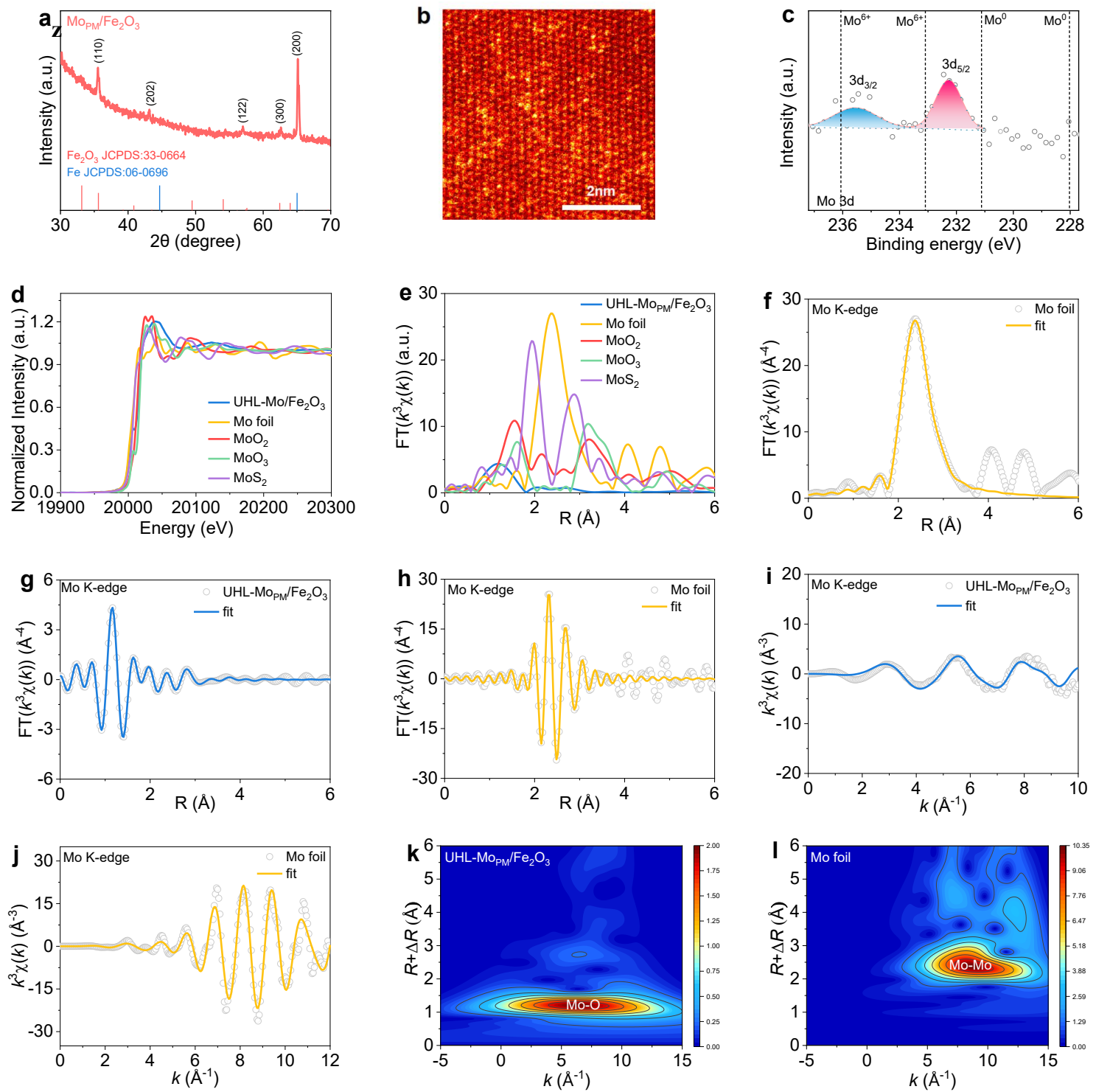
Supplementary Fig. 13 Structural characterization of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Y 3d XPS spectra of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Y K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Y K-edge EXAFS signals. **f-h**, Y K-edge EXAFS data and fit for UHL- $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Y foil.



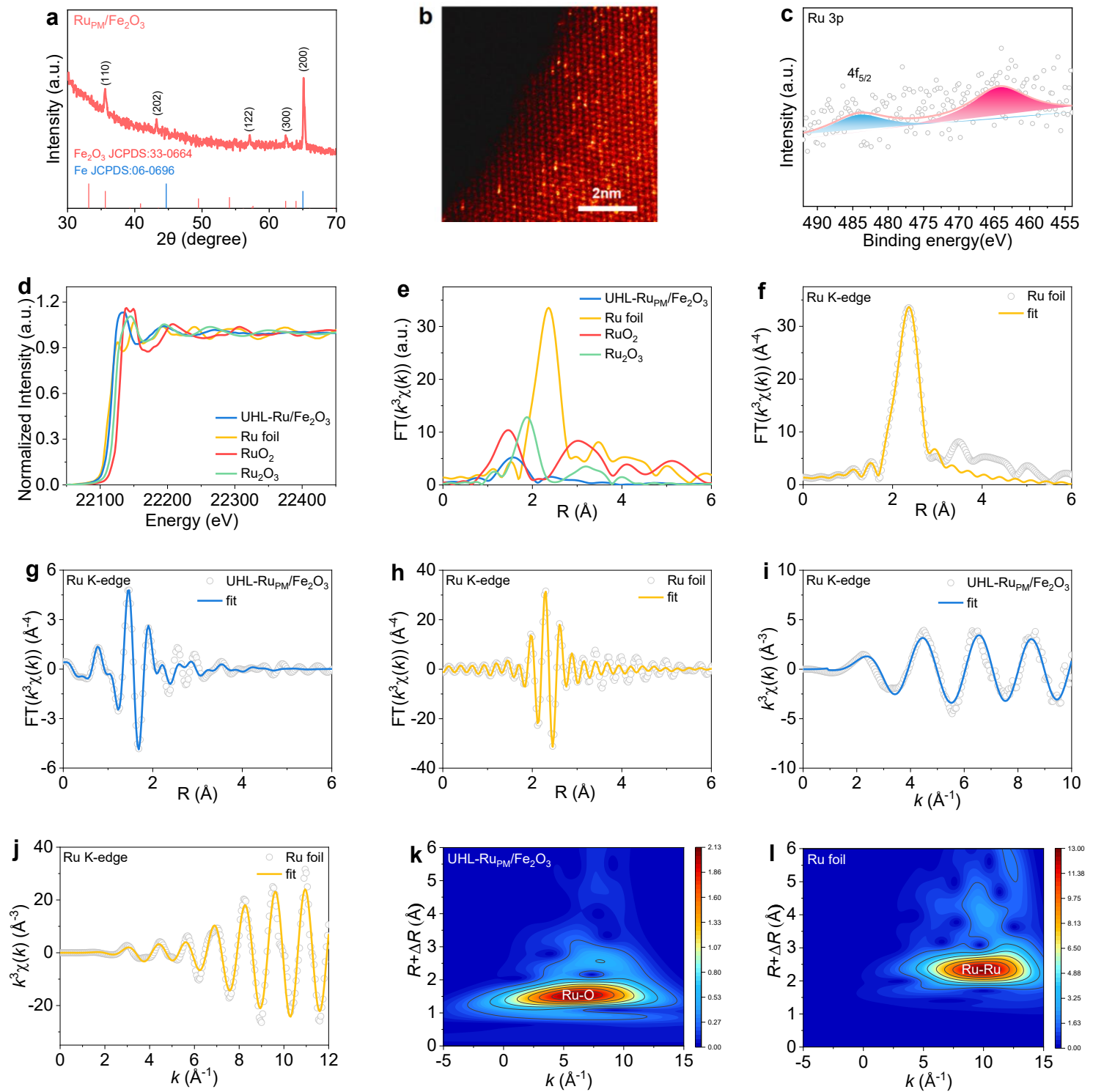
Supplementary Fig. 14 Structural characterization of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Zr 3d XPS spectra of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Zr K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Zr K-edge EXAFS signals. **f-h**, Zr K-edge EXAFS data and fit for UHL- $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Zr foil.



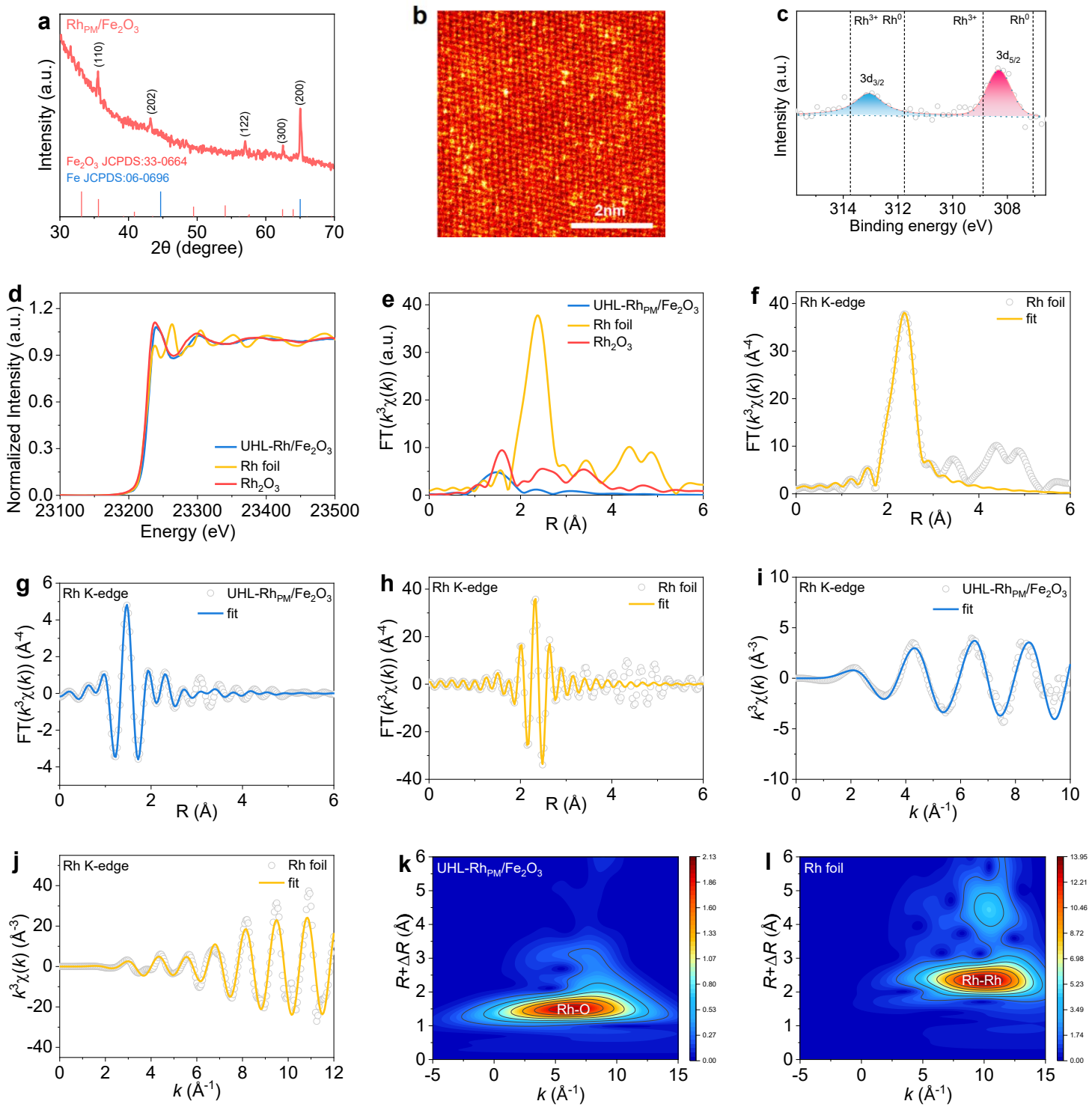
Supplementary Fig. 15 Structural characterization of $\text{Nb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Nb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Nb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Nb 3d XPS spectra of $\text{Nb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Nb K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Nb K-edge EXAFS signals. **f-h**, Nb K-edge EXAFS data and fit for UHL-Nb/Fe₂O₃ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL-Nb/Fe₂O₃ and Nb foil.



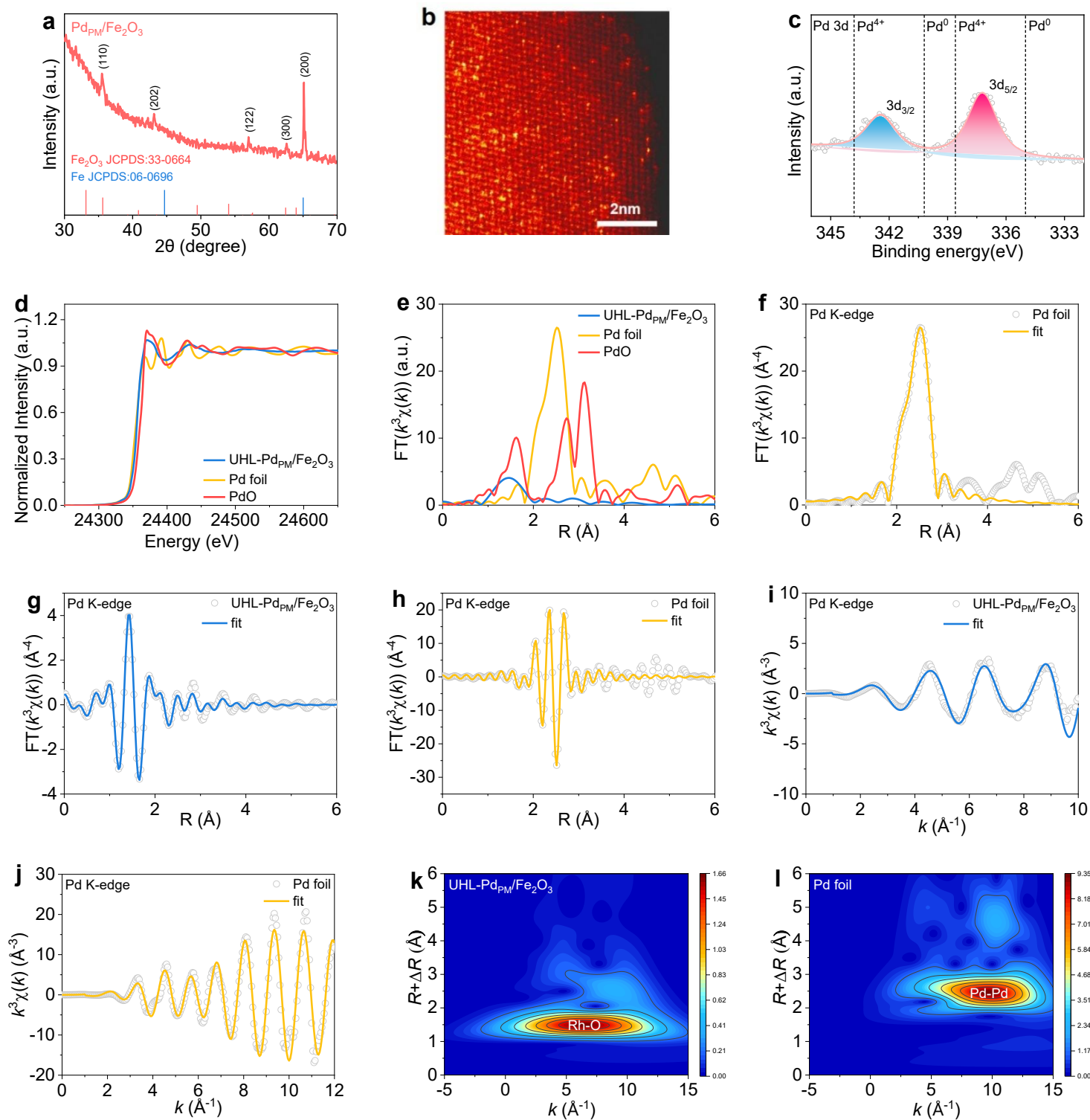
Supplementary Fig. 16 Structural characterization of $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Mo 3d XPS spectra of $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Mo K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Mo K-edge EXAFS signals. **f-h**, Mo K-edge EXAFS data and fit for UHL- $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Mo}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Mo foil.



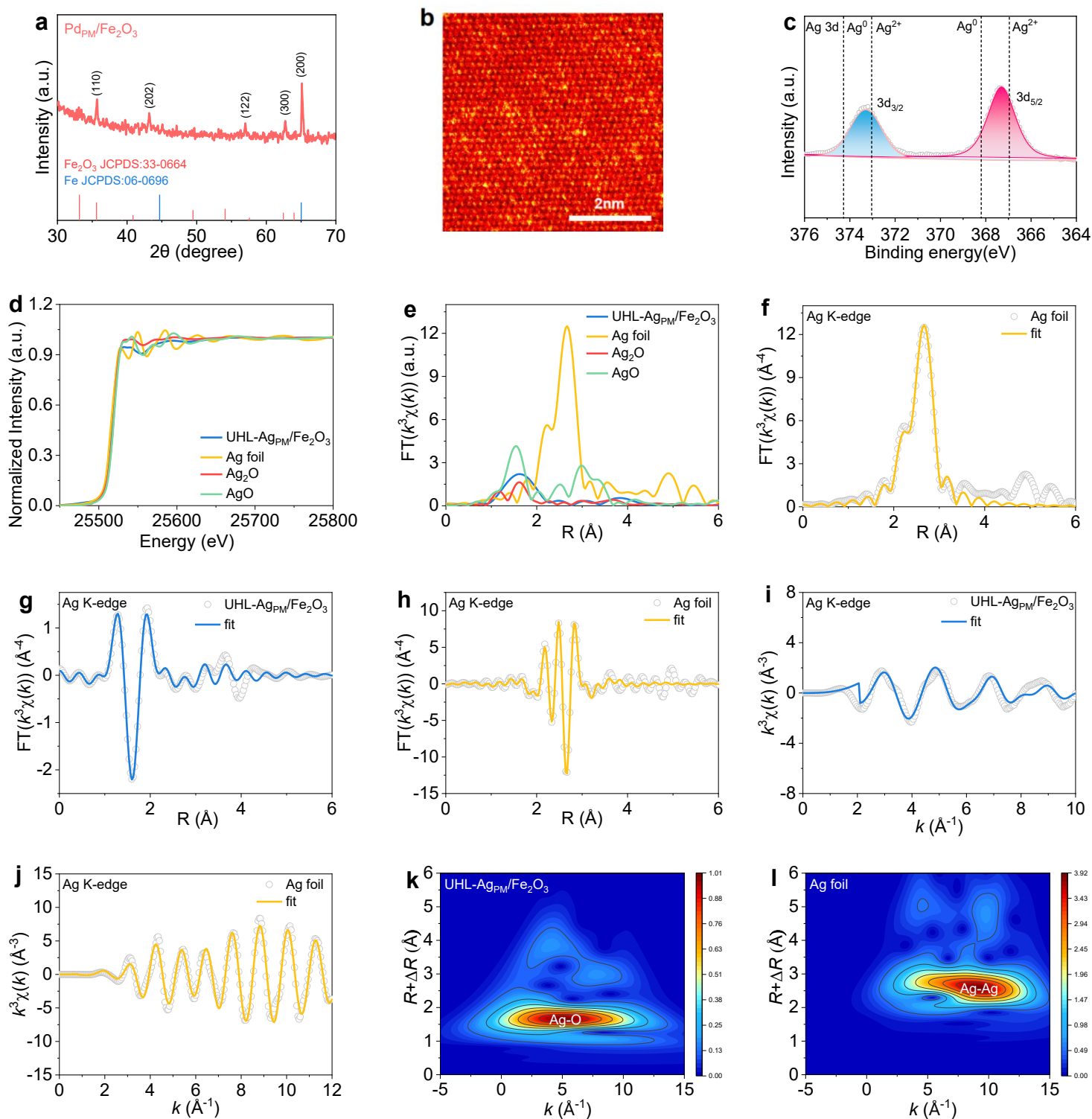
Supplementary Fig. 17 Structural characterization of $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Ru 3p XPS spectra of $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Ru K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Ru K-edge EXAFS signals. **f-h**, Ru K-edge EXAFS data and fit for UHL- $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Ru}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Ru foil.



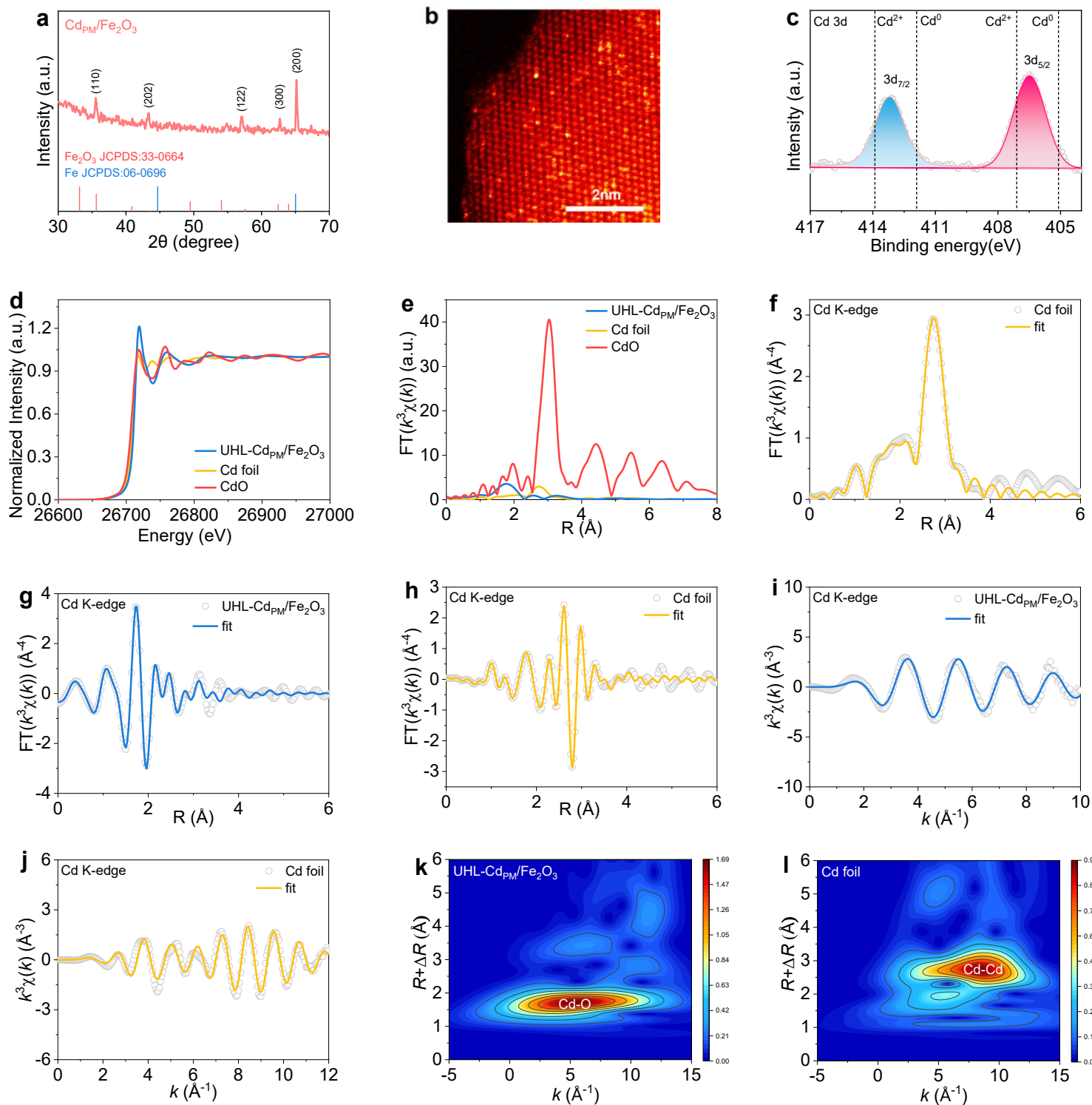
Supplementary Fig. 18 Structural characterization of Rh_{PM}/Fe₂O₃. **a**, XRD patterns of Rh_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Rh_{PM}/Fe₂O₃. **c**, Rh 3d XPS spectra of Rh_{PM}/Fe₂O₃. **d**, Normalized Rh K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Rh K-edge EXAFS signals. **f-h**, Rh K-edge EXAFS data and fit for UHL-Rh_{PM}/Fe₂O₃ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL-Rh_{PM}/Fe₂O₃ and Rh foil.



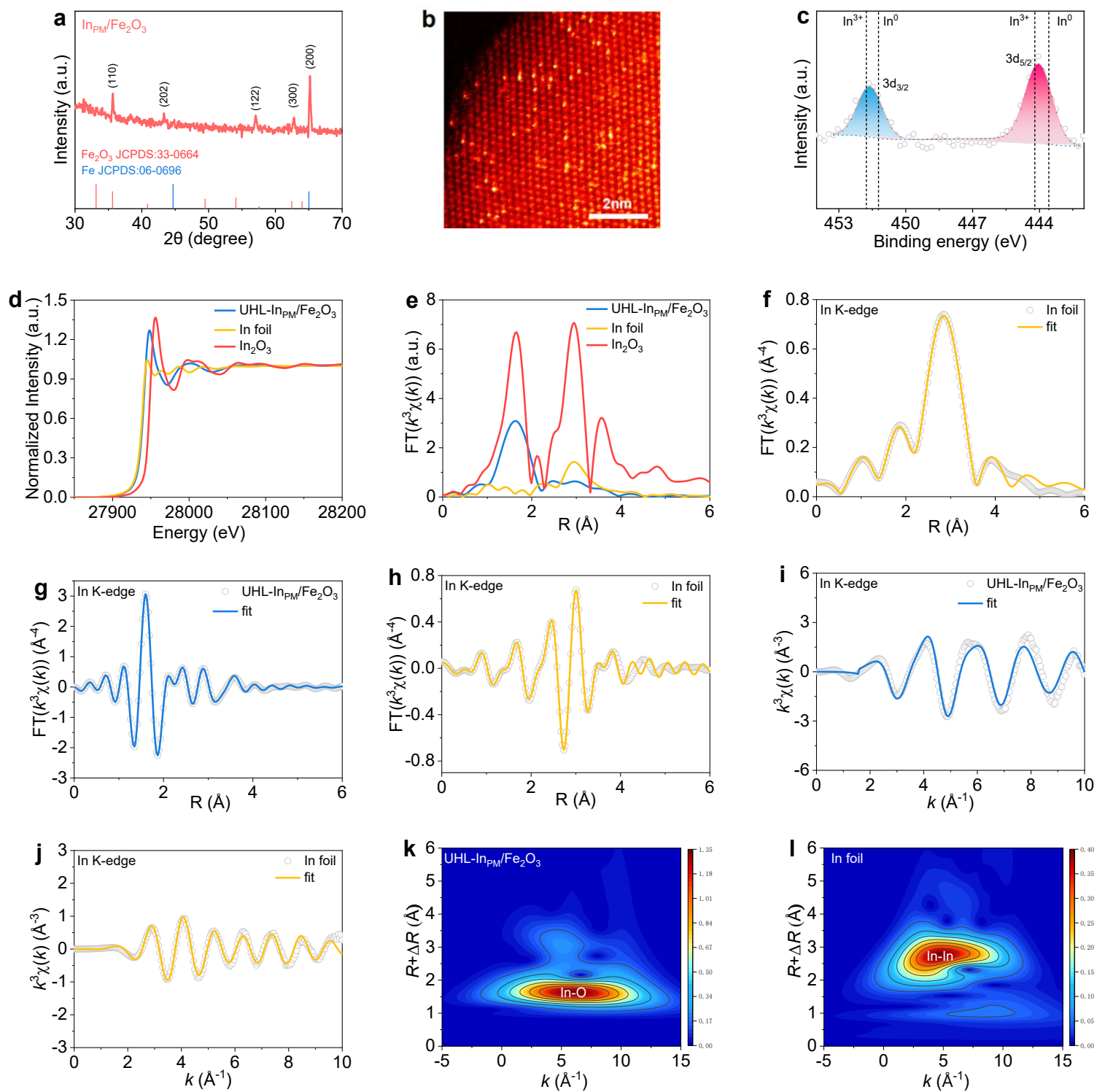
Supplementary Fig. 19 Structural characterization of Pd_{PM}/Fe₂O₃. **a**, XRD patterns of Pd_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Pd_{PM}/Fe₂O₃. **c**, Pd 3d XPS spectra of Pd_{PM}/Fe₂O₃. **d**, Normalized Pd K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Pd K-edge EXAFS signals. **f-h**, Pd K-edge EXAFS data and fit for UHL-Pd_{PM}/Fe₂O₃ in R-space and **i, j**, k-space. **k,l**, WT-EXAFS plots of UHL-Pd_{PM}/Fe₂O₃ and Pd foil.



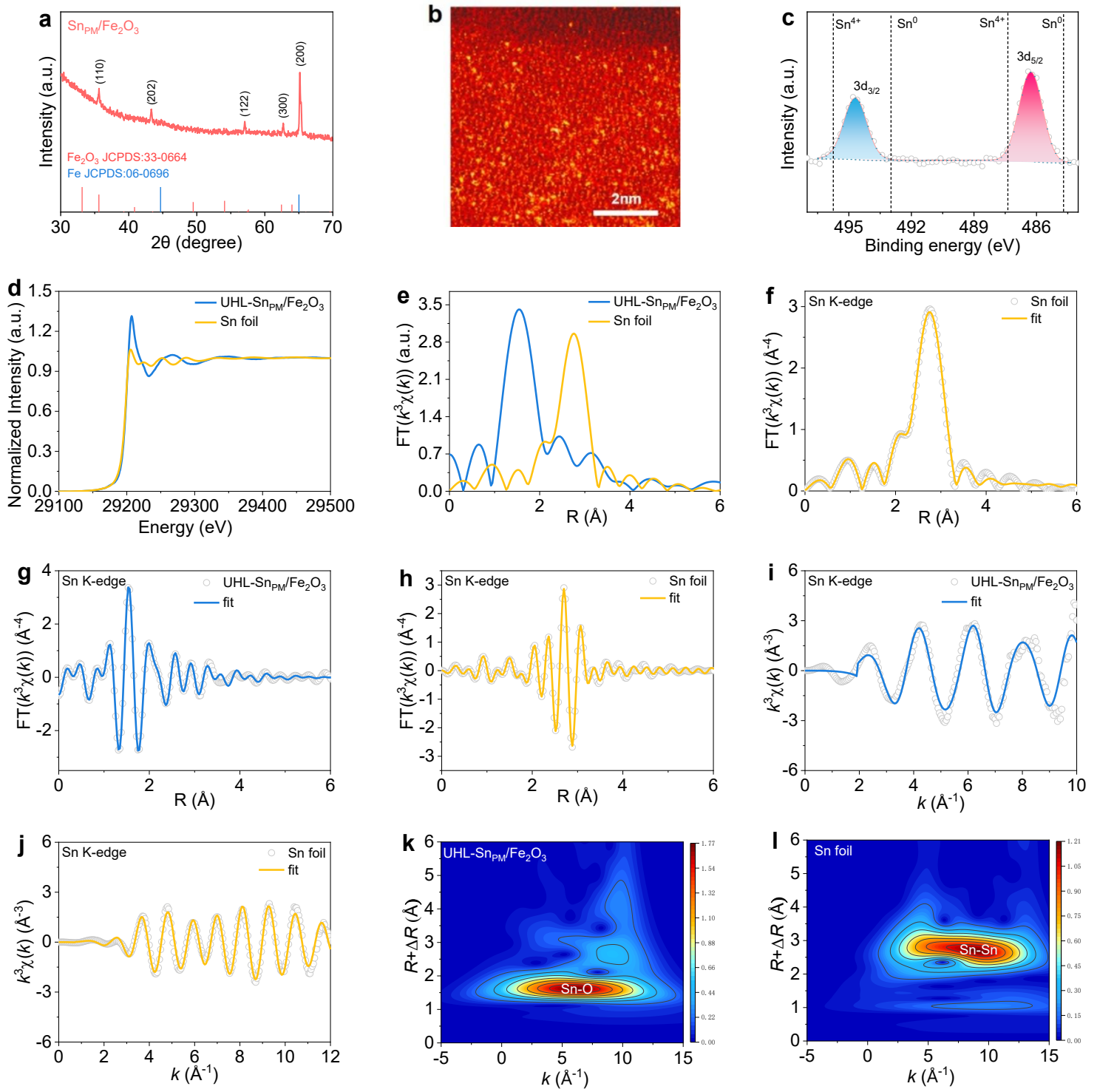
Supplementary Fig. 20 Structural characterization of $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Ag 3d XPS spectra of $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Ag K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Ag K-edge EXAFS signals. **f-h**, Ag K-edge EXAFS data and fit for UHL- $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Ag}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Ag foil.



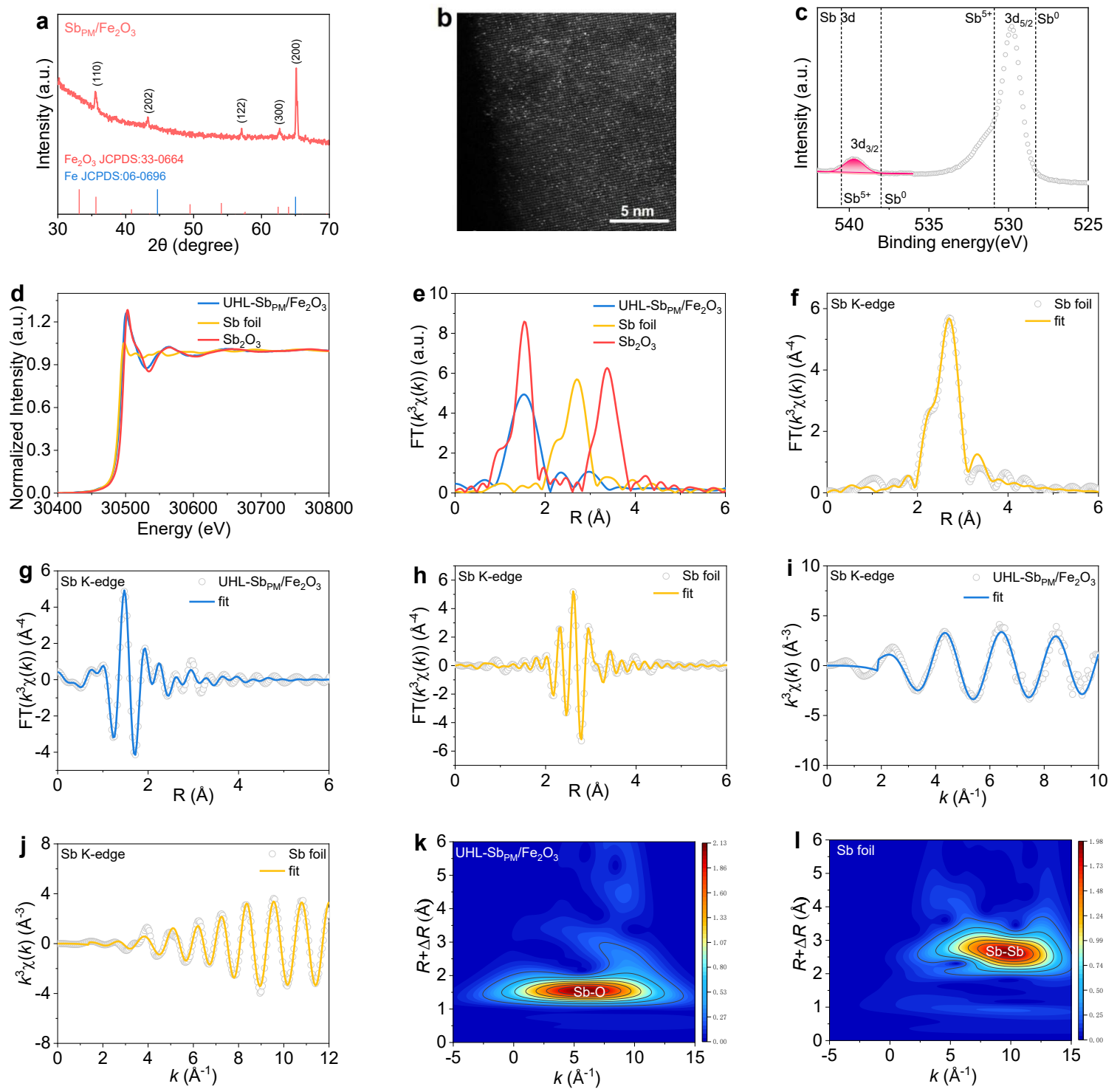
Supplementary Fig. 21 Structural characterization of Cd_{PM}/Fe₂O₃. **a**, XRD patterns of Cd_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Cd_{PM}/Fe₂O₃. **c**, Cd 3d XPS spectra of Cd_{PM}/Fe₂O₃. **d**, Normalized Cd K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Cd K-edge EXAFS signals. **f-h**, Cd K-edge EXAFS data and fit for UHL-Cd_{PM}/Fe₂O₃ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL-Cd_{PM}/Fe₂O₃ and Cd foil.



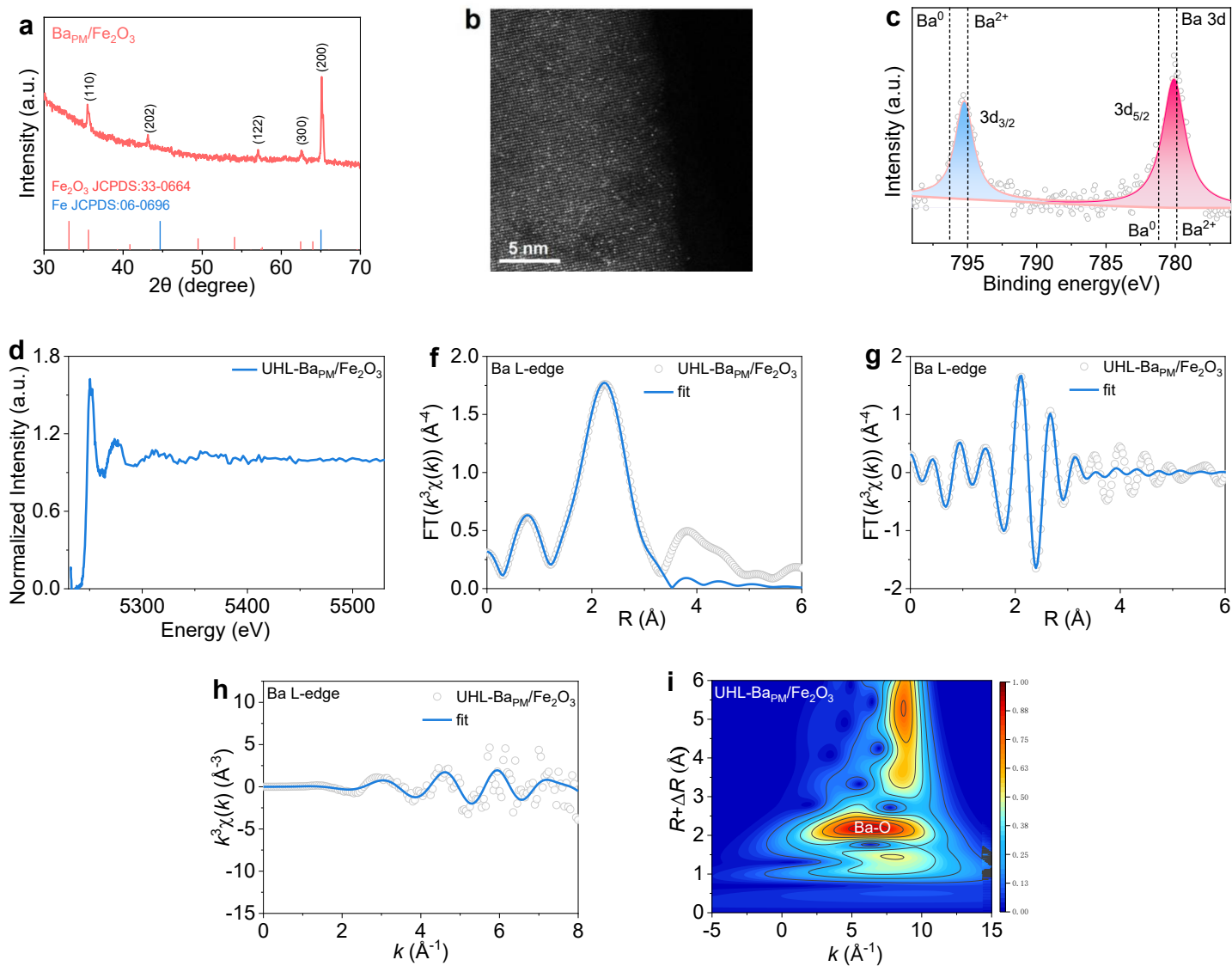
Supplementary Fig. 22 Structural characterization of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, In 3d XPS spectra of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized In K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental In K-edge EXAFS signals. **f-h**, In K-edge EXAFS data and fit for UHL- $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and In foil.



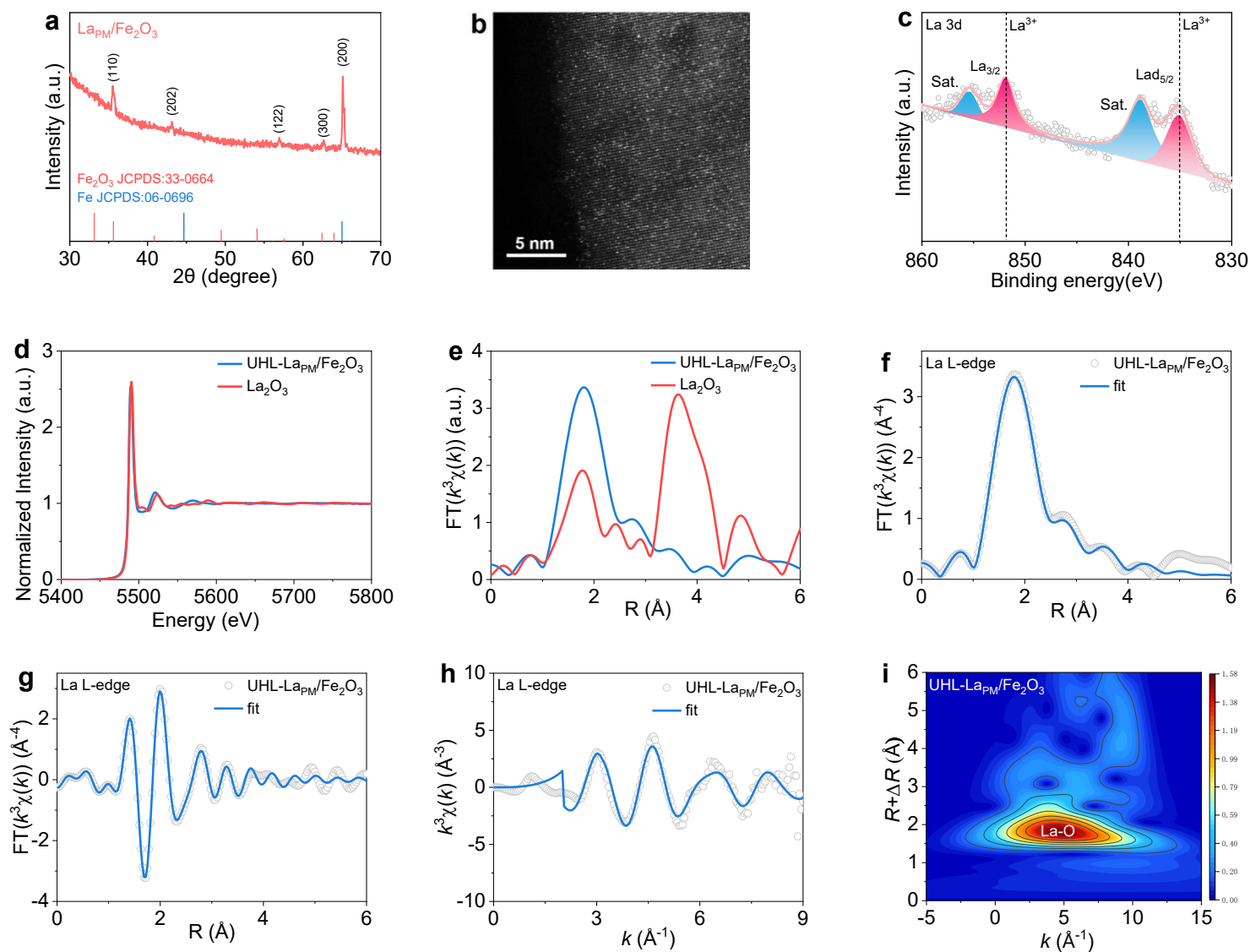
Supplementary Fig. 23 Structural characterization of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Sn 3d XPS spectra of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Sn K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Sn K-edge EXAFS signals. **f-h**, Sn K-edge EXAFS data and fit for UHL- $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Sn foil.



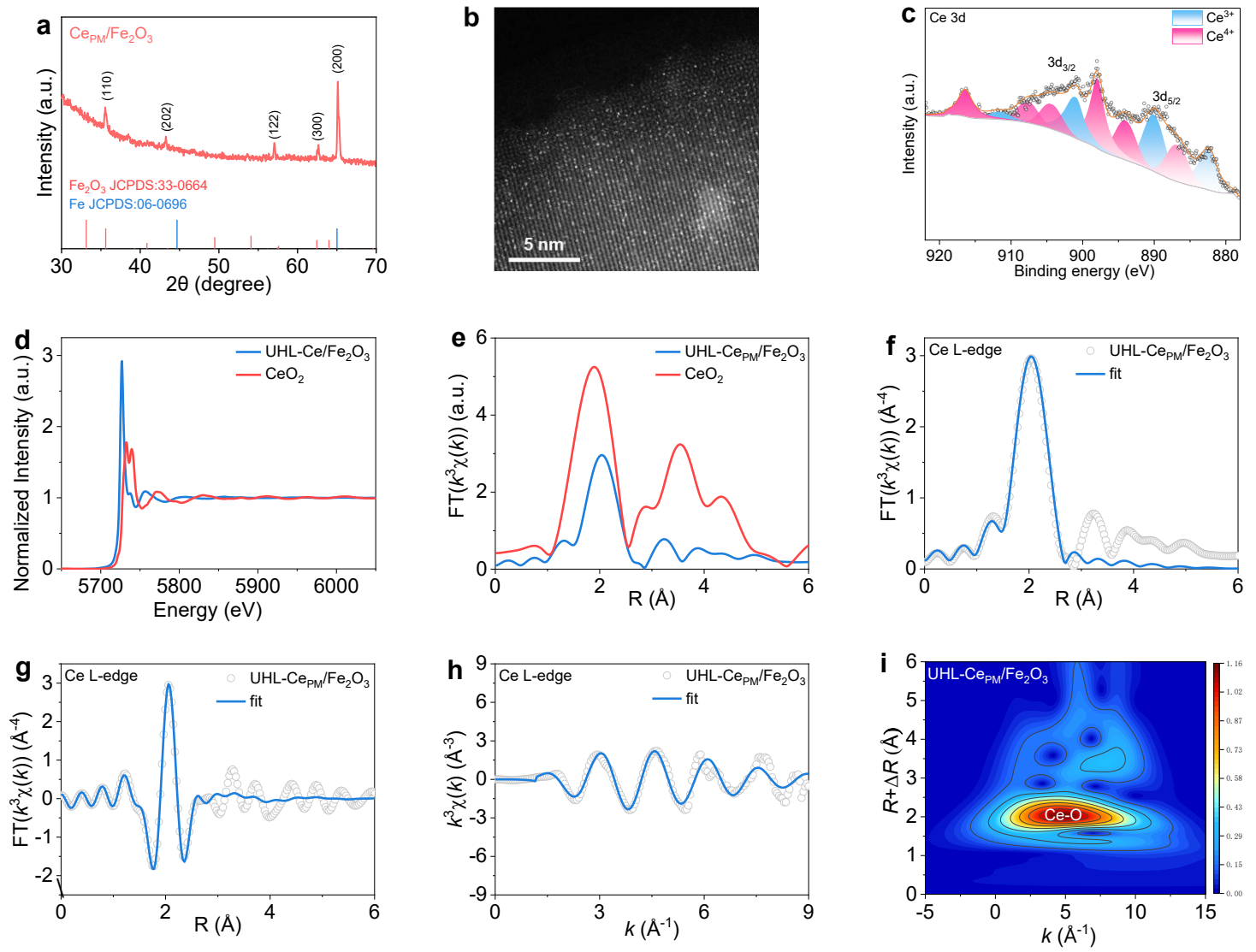
Supplementary Fig. 24 Structural characterization of Sb_{PM}/Fe₂O₃. **a**, XRD patterns of Sb_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Sb_{PM}/Fe₂O₃. **c**, Sb 3d XPS spectra of Sb_{PM}/Fe₂O₃. **d**, Normalized Sb K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Sb K-edge EXAFS signals. **f-h**, Sb K-edge EXAFS data and fit for UHL-Sb_{PM}/Fe₂O₃ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL-Sb_{PM}/Fe₂O₃ and Sb foil.



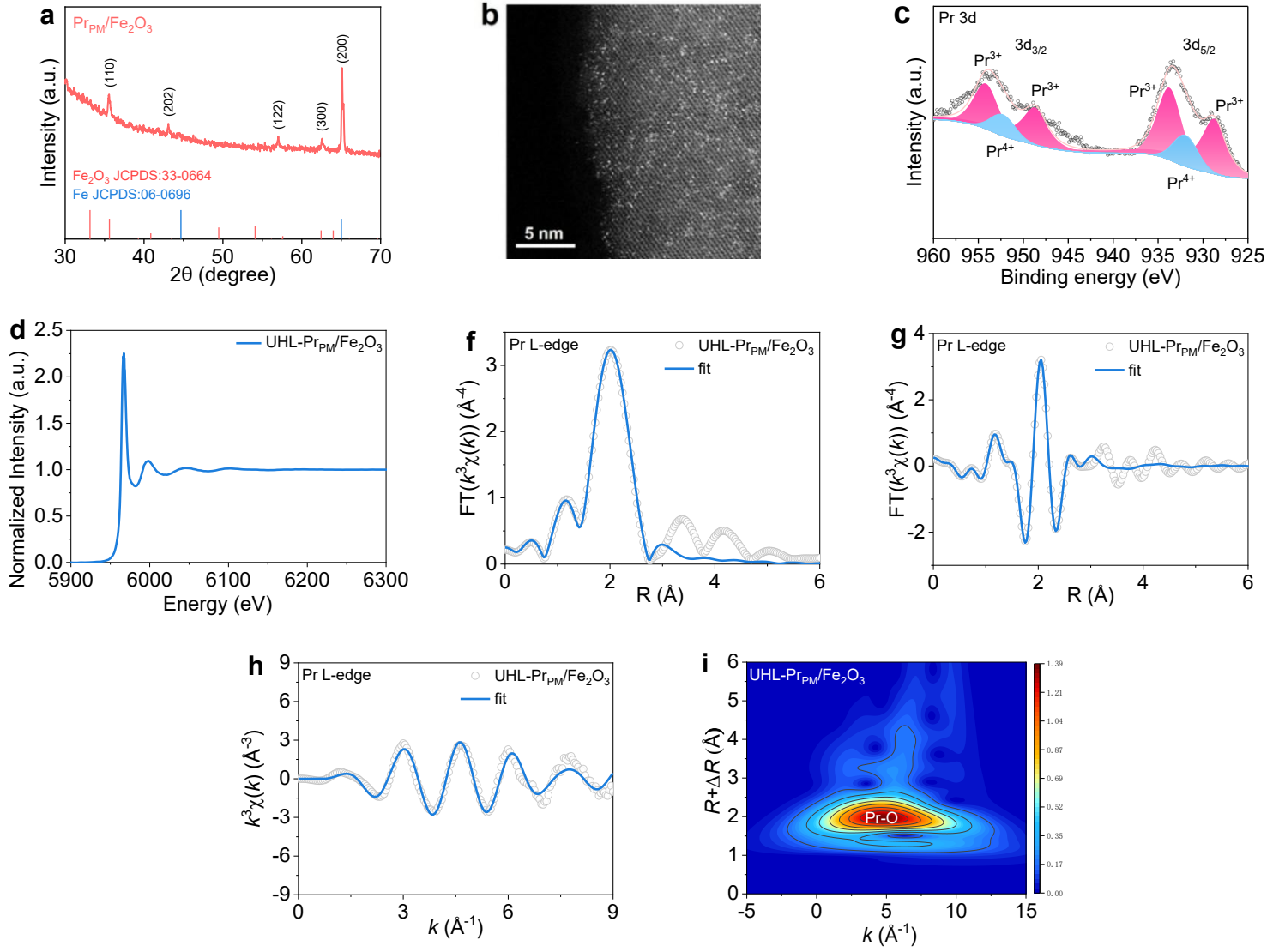
Supplementary Fig. 25 Structural characterization of $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Ba 3d XPS spectra of $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Ba L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Ba L-edge EXAFS signals. **f-h**, Ba L-edge EXAFS data and fit for UHL- $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k,l**, WT-EXAFS plots of UHL- $\text{Ba}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Ba foil.



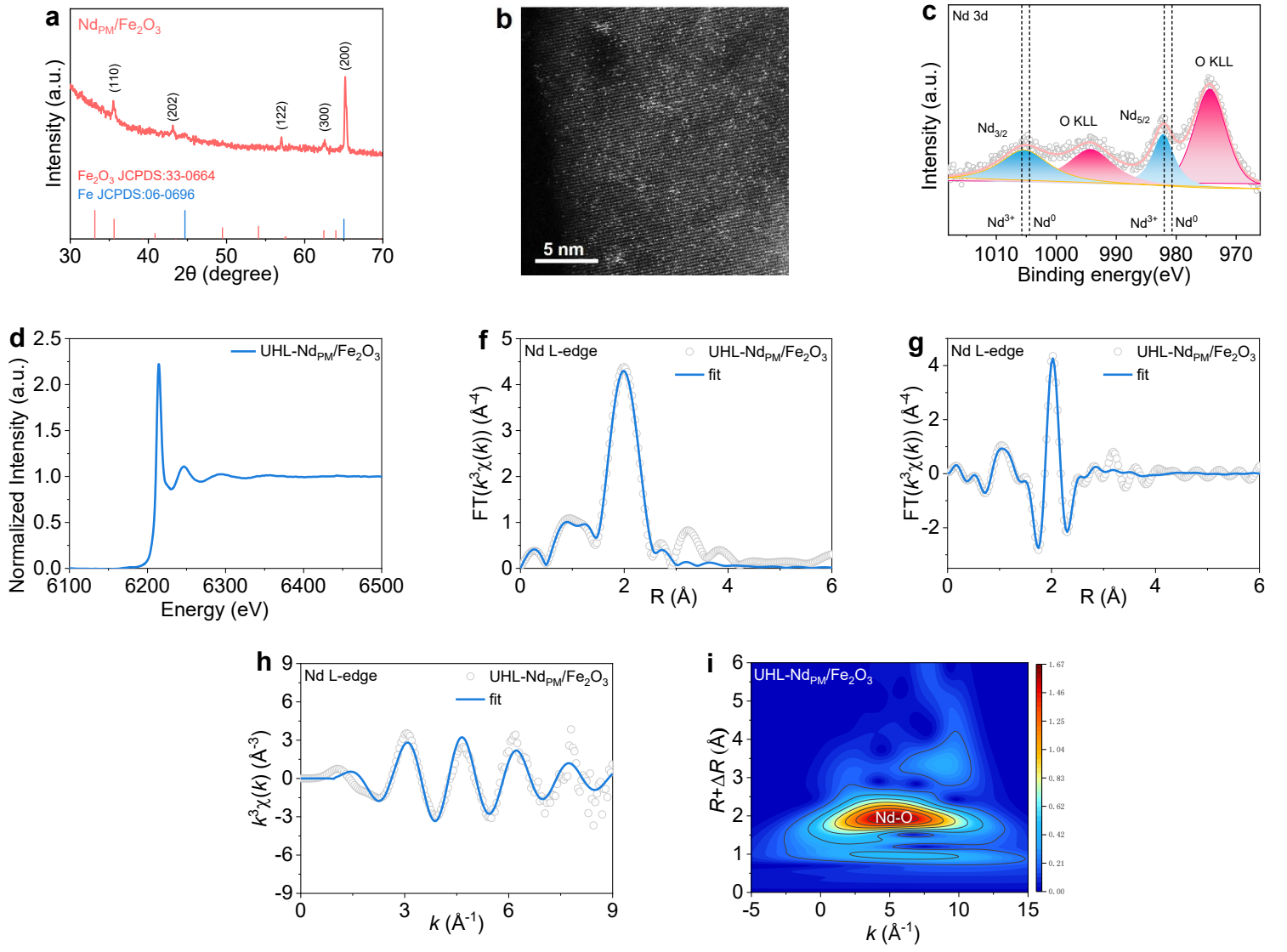
Supplementary Fig. 26 Structural characterization of $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, La 3d XPS spectra of $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized La L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental La L-edge EXAFS signals. **f-h**, La L-edge EXAFS data and fit for UHL- $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{La}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and La foil.



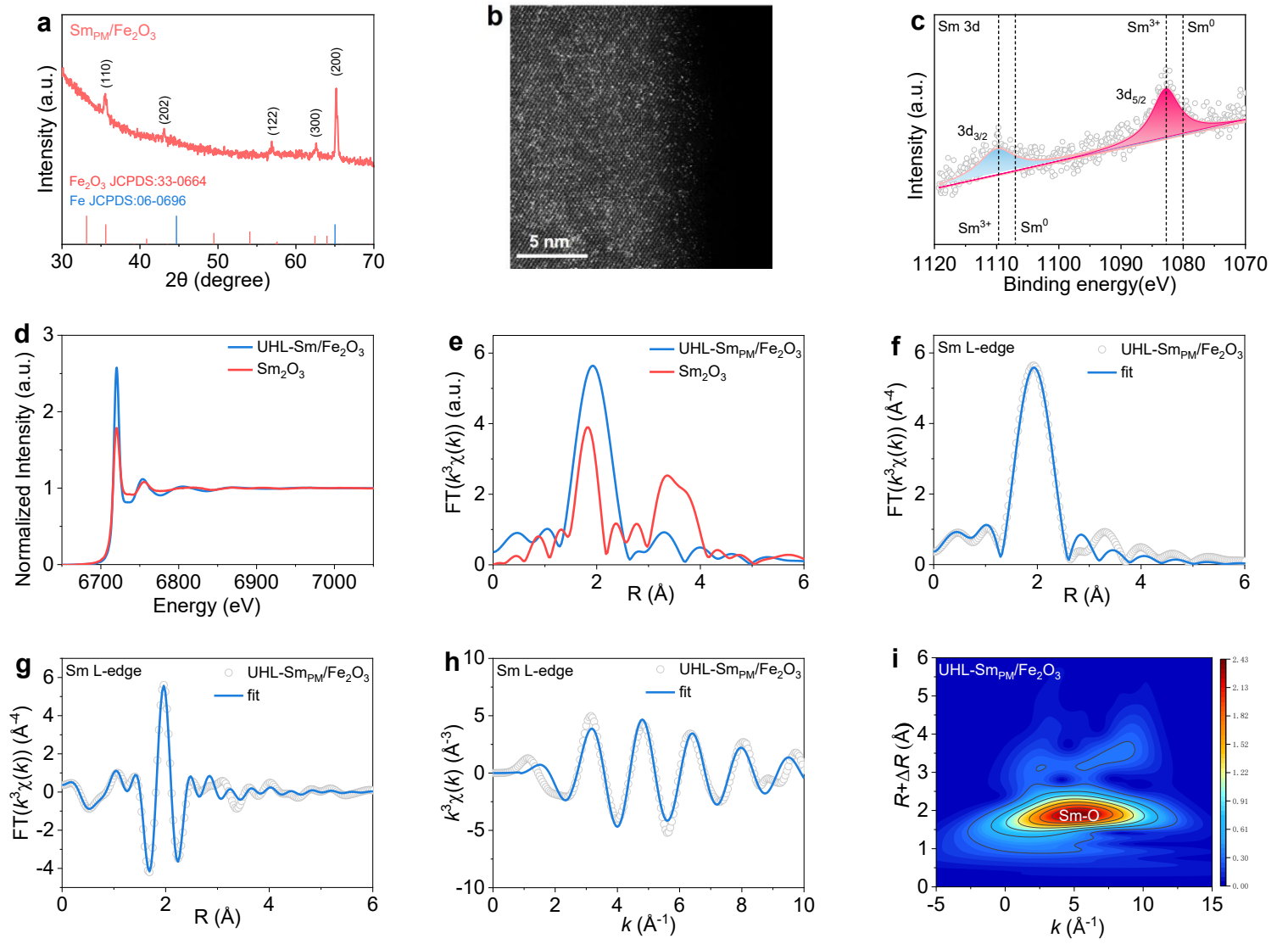
Supplementary Fig. 27 Structural characterization of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Ce 3d XPS spectra of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Ce *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed *R*-space of the experimental Ce *L*-edge EXAFS signals. **f-h**, Ce *L*-edge EXAFS data and fit for UHL- $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in *R*-space and **i, j**, *k*-space. **k,l**, WT-EXAFS plots of UHL- $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Ce foil.



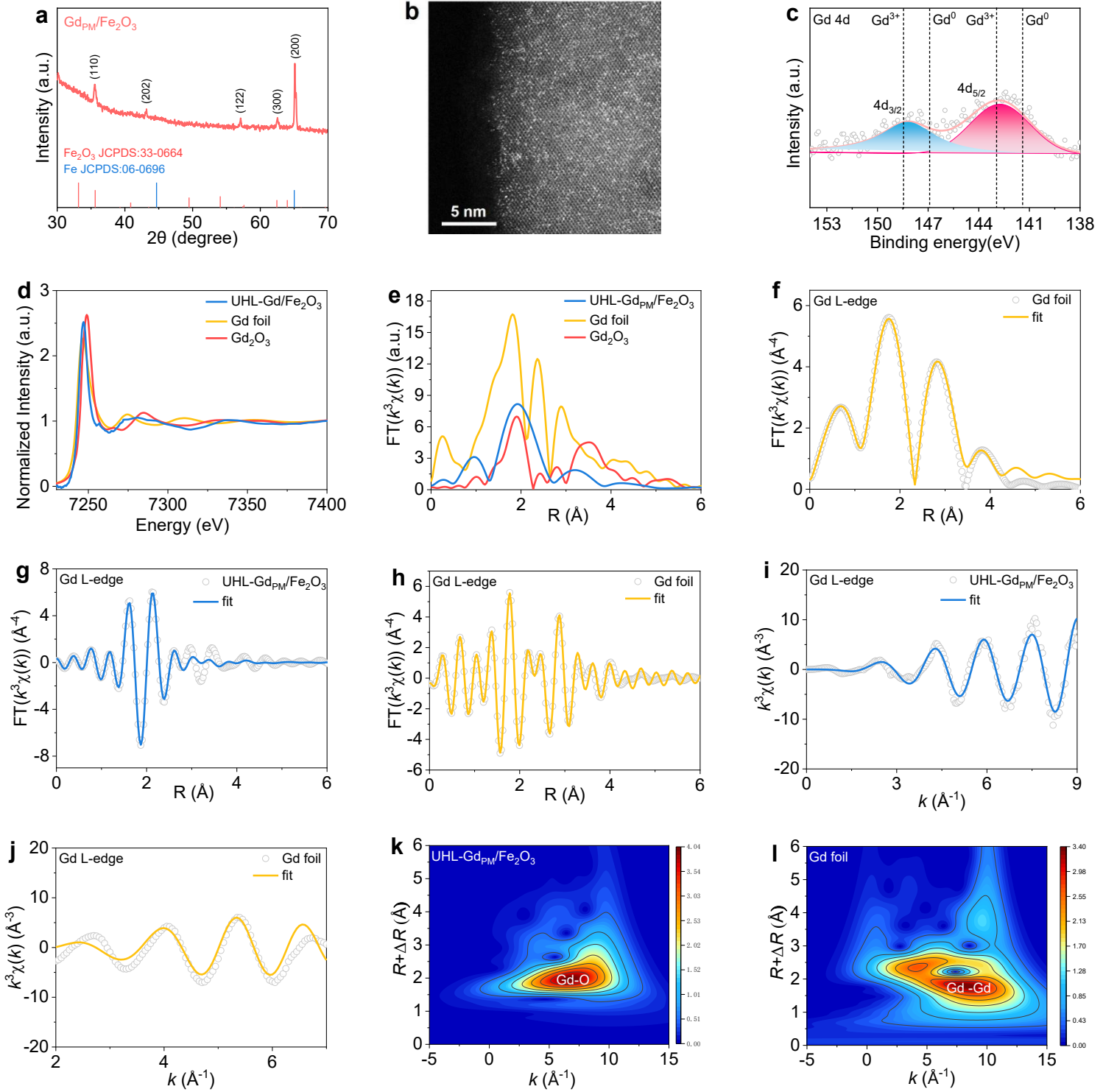
Supplementary Fig. 28 Structural characterization of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Pr 4f XPS spectra of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Pr K-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Pr L-edge EXAFS signals. **f-h**, Pr L-edge EXAFS data and fit for UHL- $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Pr foil.



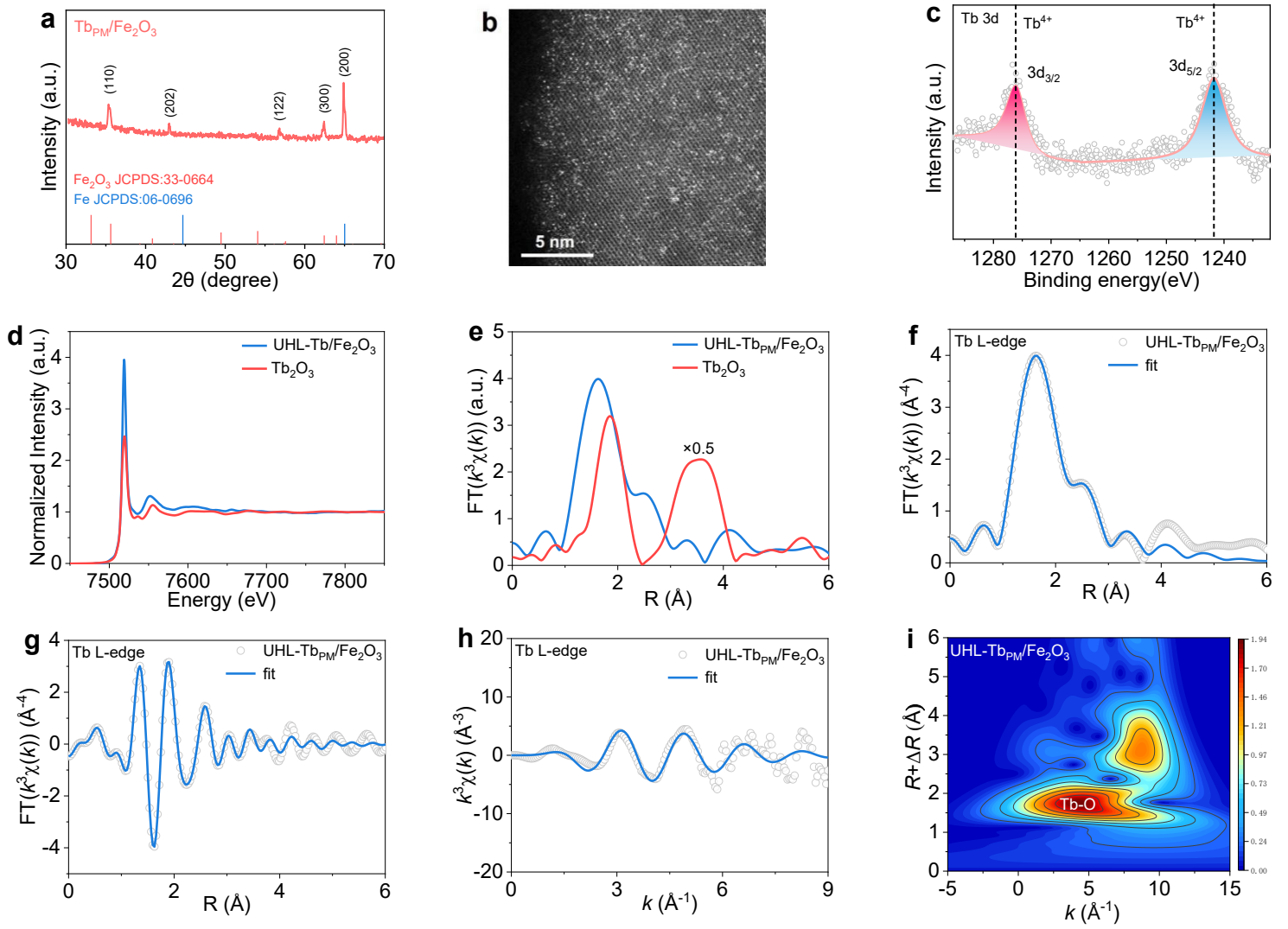
Supplementary Fig. 29 Structural characterization of $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Nd 3d XPS spectra of $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Nd L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Nd L-edge EXAFS signals. **f-h**, Nd L-edge EXAFS data and fit for UHL- $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k,l**, WT-EXAFS plots of UHL- $\text{Nd}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Nd foil.



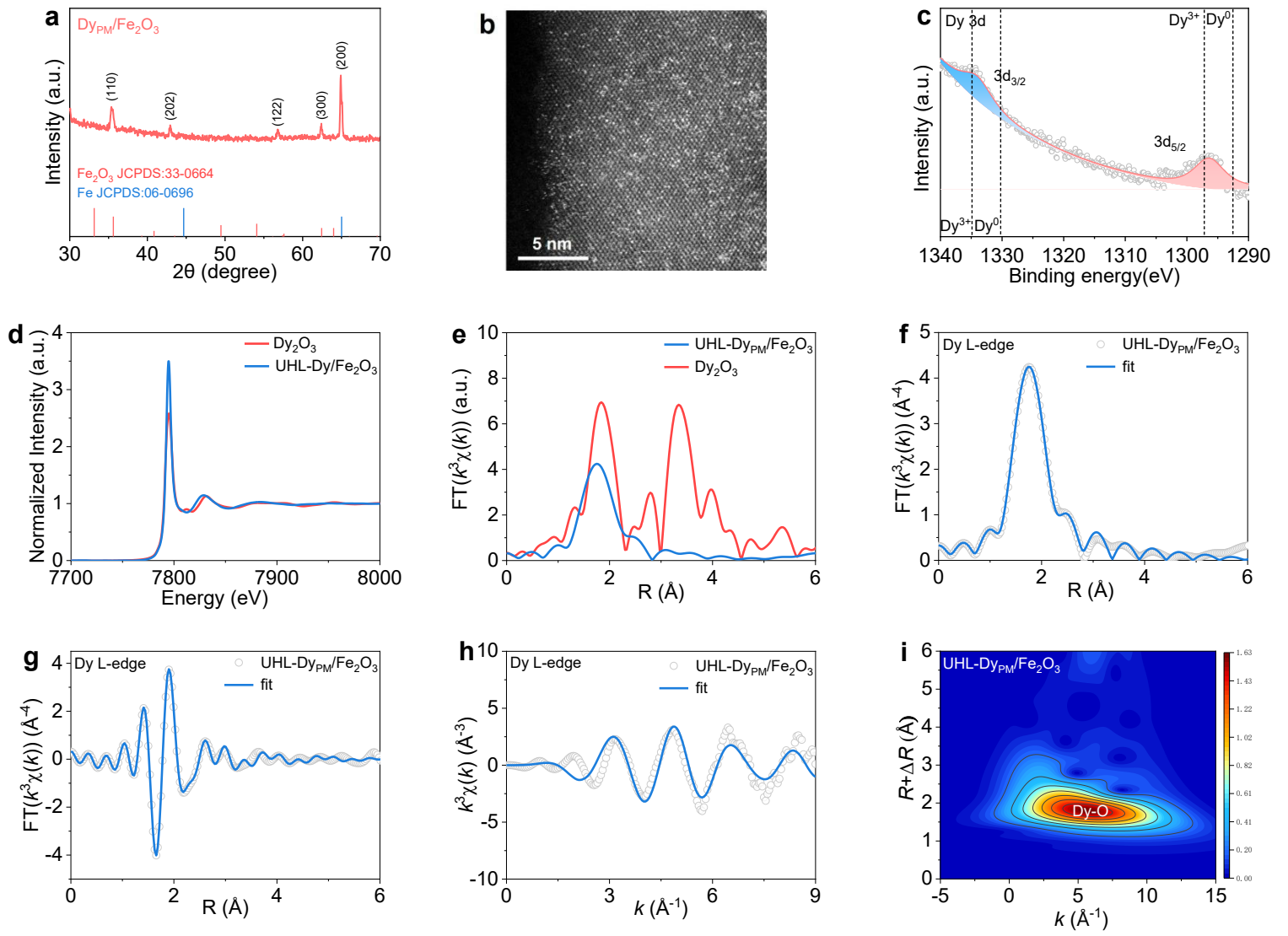
Supplementary Fig. 30 Structural characterization of $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Sm 3d XPS spectra of $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Sm *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Sm *L*-edge EXAFS signals. **f-h**, Sm *L*-edge EXAFS data and fit for UHL- $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R -space and **i, j**, k -space. **k, l**, WT-EXAFS plots of UHL- $\text{Sm}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Sm foil.



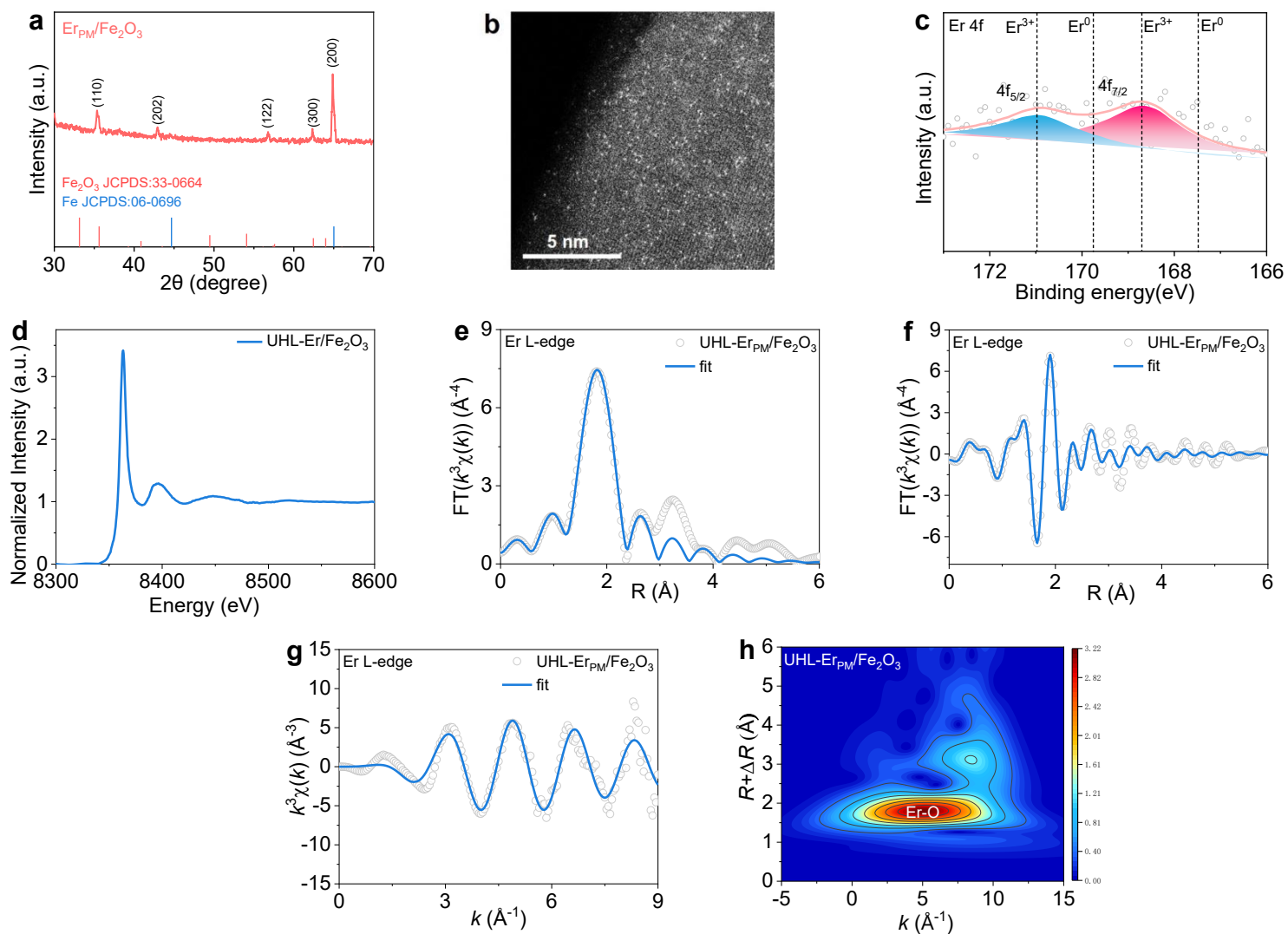
Supplementary Fig. 31 Structural characterization of Gd_{PM}/Fe₂O₃. **a**, XRD patterns of Gd_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Gd_{PM}/Fe₂O₃. **c**, Gd 3d XPS spectra of Gd_{PM}/Fe₂O₃. **d**, Normalized Gd *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed *R*-space of the experimental Gd *L*-edge EXAFS signals. **f-h**, Gd *L*-edge EXAFS data and fit for UHL-Gd_{PM}/Fe₂O₃ in *R*-space and **i, j**, *k*-space. **k, l**, WT-EXAFS plots of UHL-Gd_{PM}/Fe₂O₃ and Gd foil.



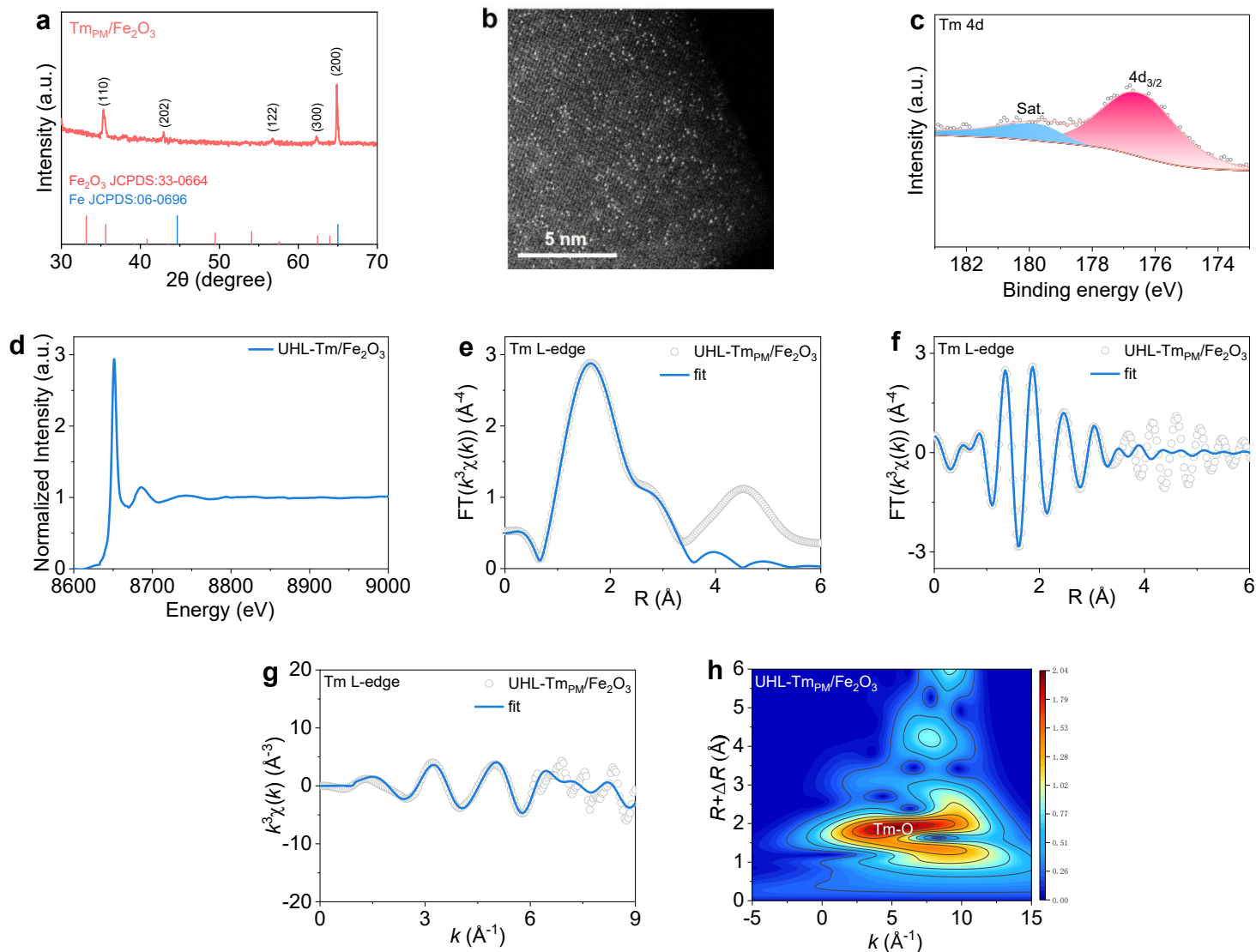
Supplementary Fig. 32 Structural characterization of $\text{Tb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Tb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Tb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Tb 3d XPS spectra of $\text{Tb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Tb L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Tb L-edge EXAFS signals. **f-g**, Tb L-edge EXAFS data and fit for UHL-Tb/Fe₂O₃ in R-space and **h**, k-space. **i**, WT-EXAFS plots of UHL-Tb/Fe₂O₃ and Tb foil.



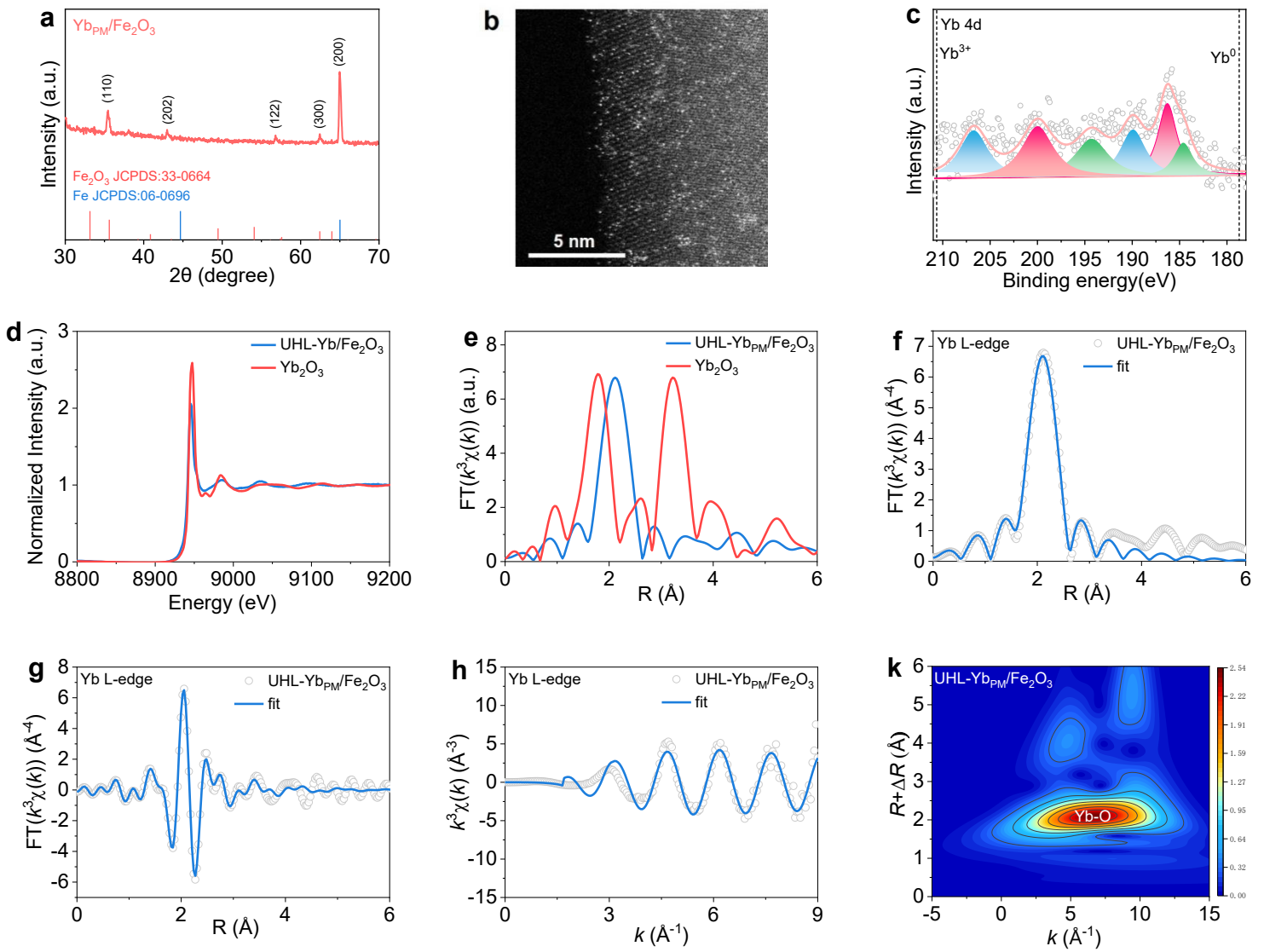
Supplementary Fig. 33 Structural characterization of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Dy 3d XPS spectra of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Dy L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Dy L-edge EXAFS signals. **f-g**, Dy L-edge EXAFS data and fit for UHL-Dy $_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **h**, k -space. **i**, WT-EXAFS plots of UHL-Dy $_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Dy foil.



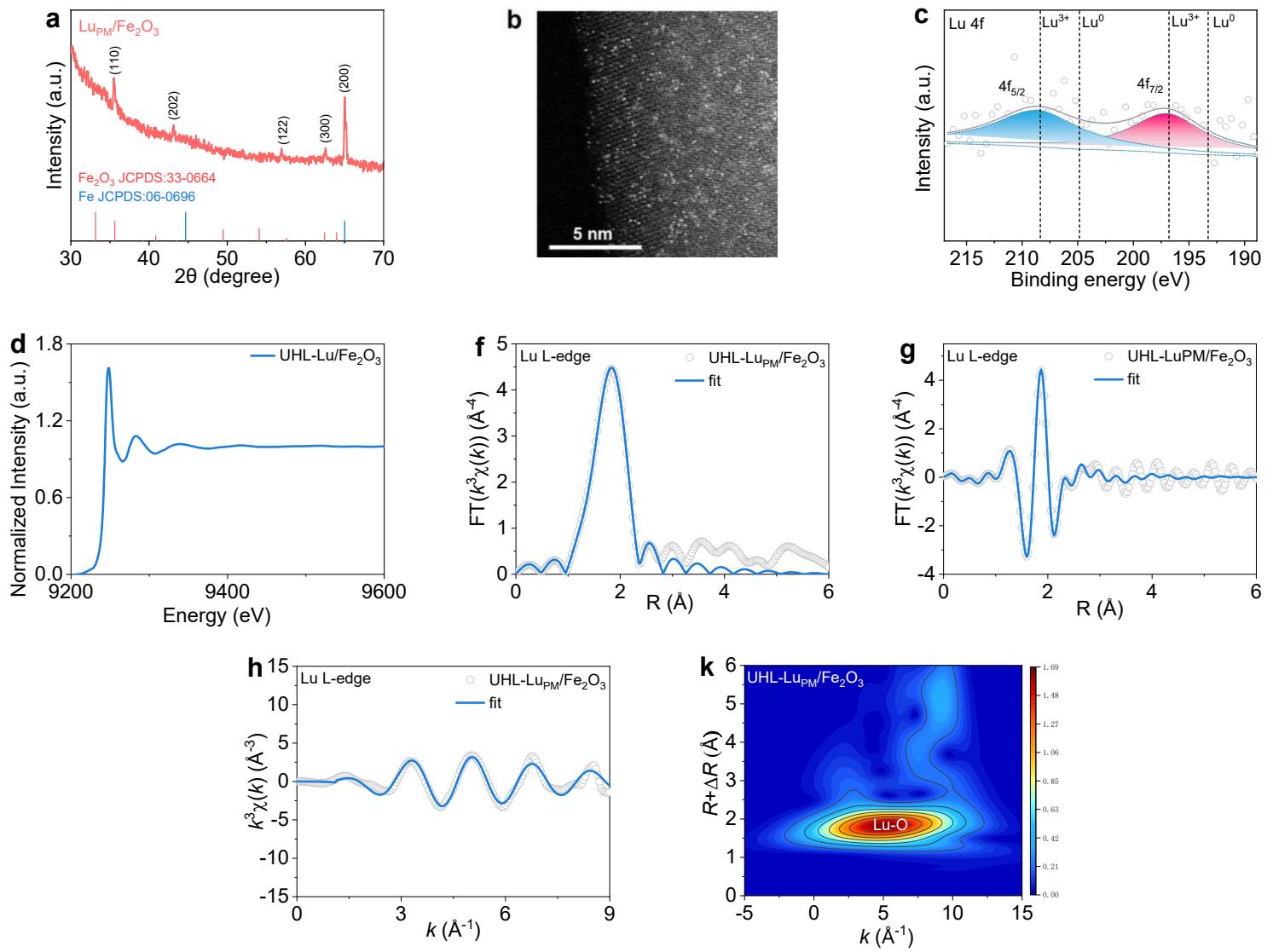
Supplementary Fig. 34 Structural characterization of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Er 4f XPS spectra of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Er L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Er L-edge EXAFS signals. **f**, Er L-edge EXAFS data and fit for UHL- $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **g**, k-space. **h**, WT-EXAFS plot of UHL- $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$.



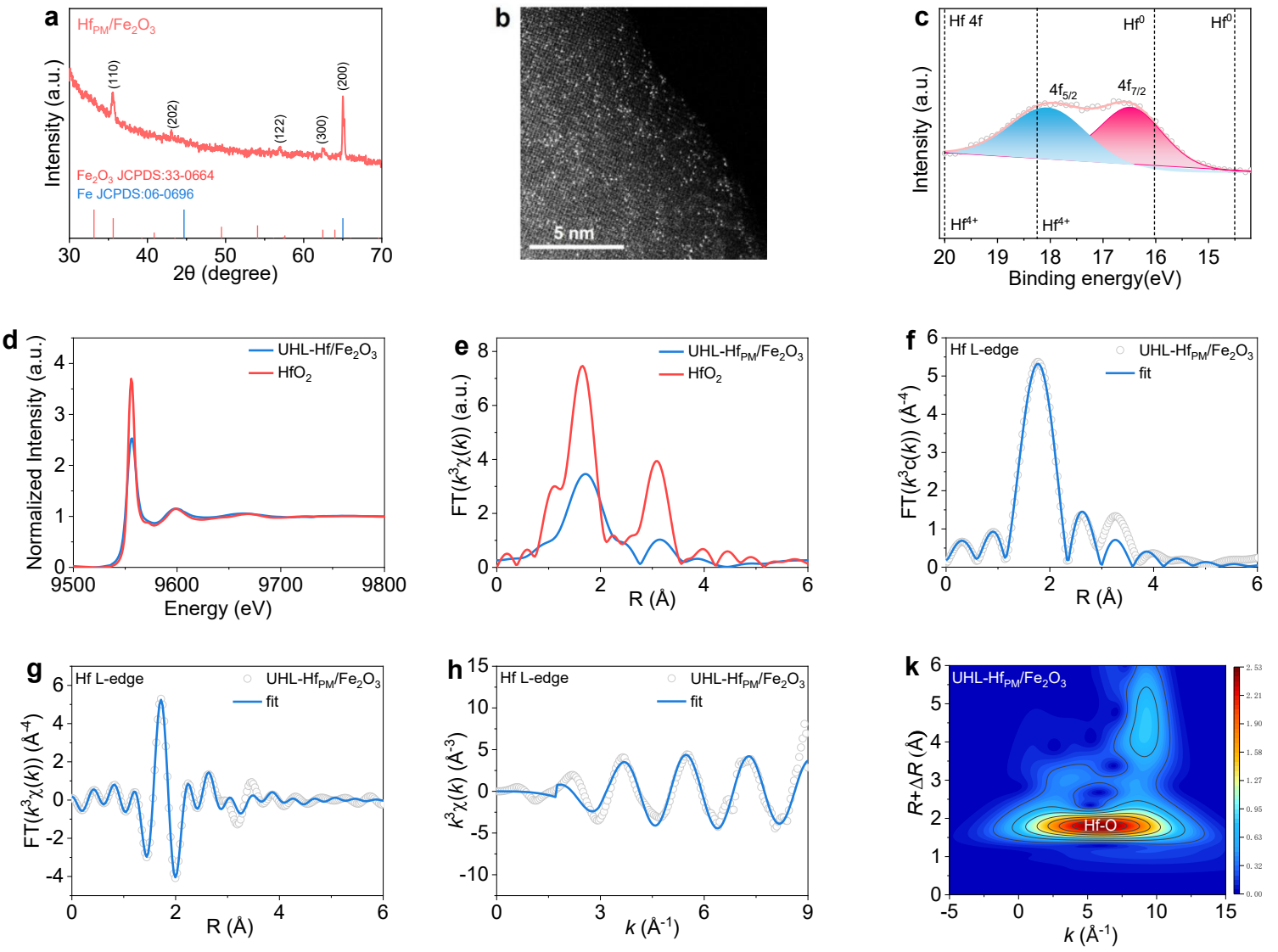
Supplementary Fig. 35 Structural characterization of $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Tm 4d XPS spectra of $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Tm *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Tm *L*-edge EXAFS signals. **f**, Tm *L*-edge EXAFS data and fit for UHL- $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in *R*-space and **g**, *k*-space. **h**, WT-EXAFS plot of UHL- $\text{Tm}_{\text{PM}}/\text{Fe}_2\text{O}_3$.



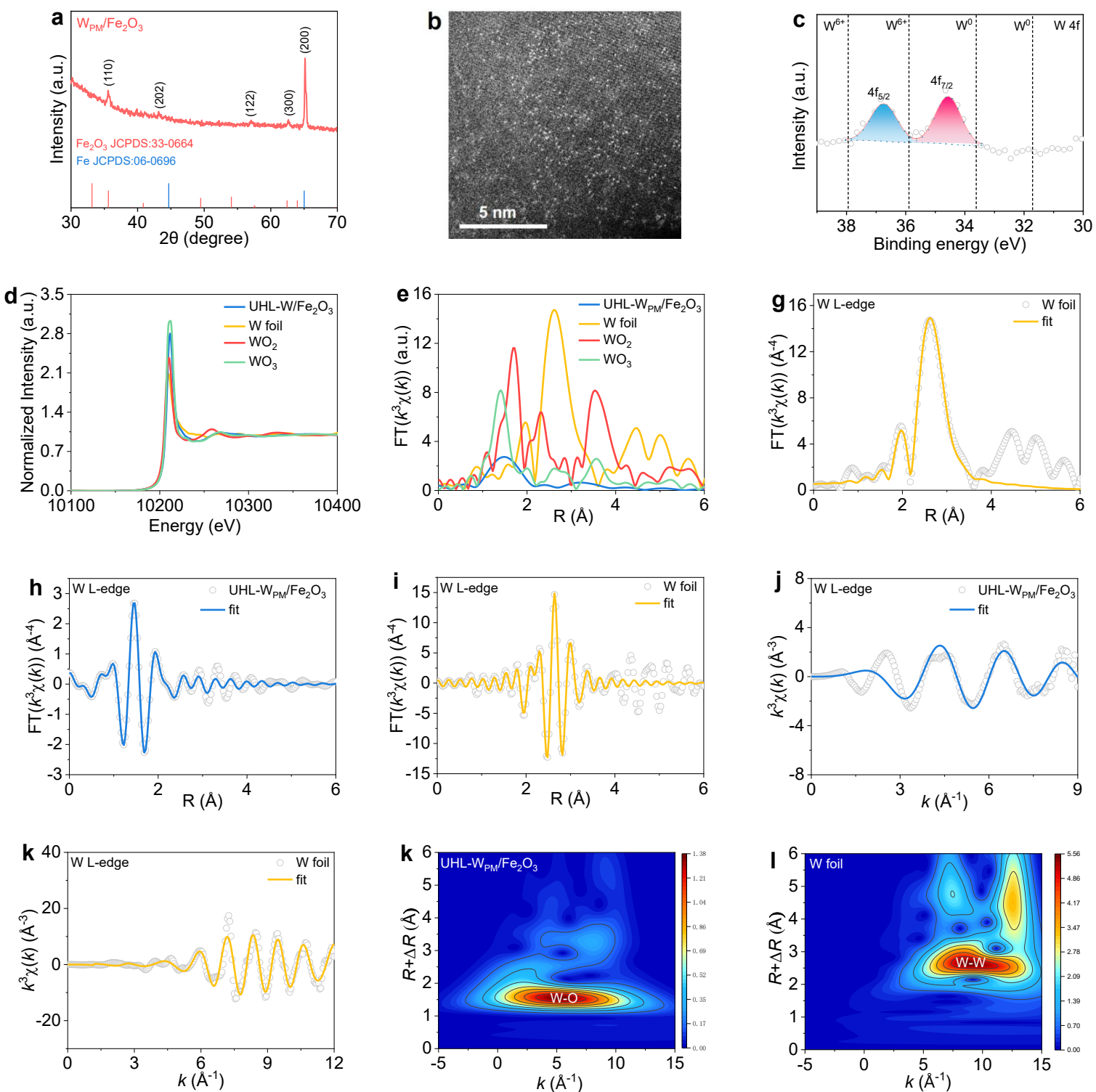
Supplementary Fig. 36 Structural characterization of Yb_{PM}/Fe₂O₃. **a**, XRD patterns of Yb_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Yb_{PM}/Fe₂O₃. **c**, Yb 4d XPS spectra of Yb_{PM}/Fe₂O₃. **d**, Normalized Yb *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed *R*-space of the experimental Yb *L*-edge EXAFS signals. **f,g**, Yb *L*-edge EXAFS data and fit for UHL-Yb_{PM}/Fe₂O₃ in *R*-space and **h**, *k*-space. **k,l**, WT-EXAFS plots of UHL-Yb_{PM}/Fe₂O₃ and Yb foil.



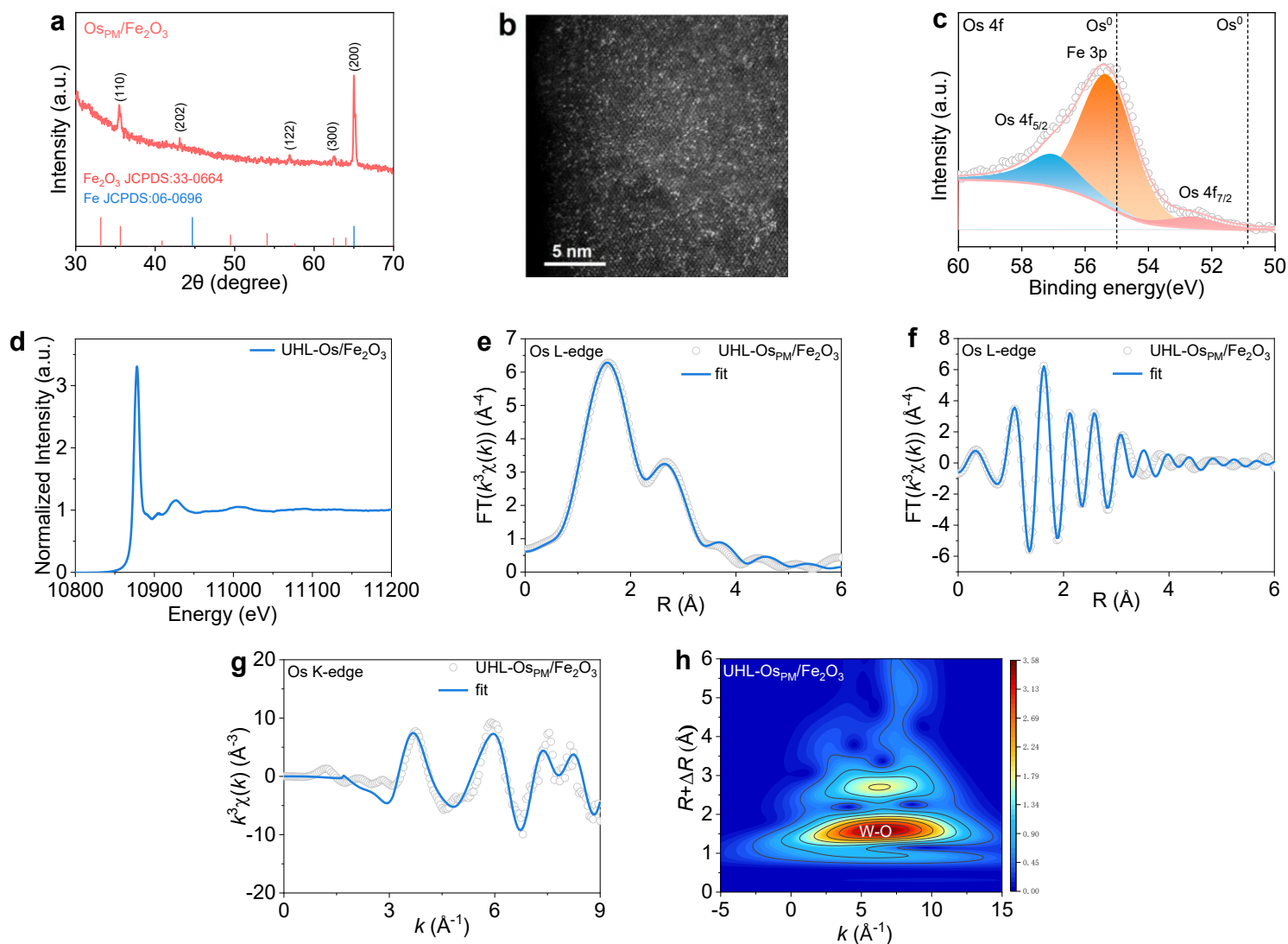
Supplementary Fig. 37 Structural characterization of $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Lu 4f XPS spectra of $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Lu L -edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R -space of the experimental Lu L -edge EXAFS signals. **f,g**, Lu L -edge EXAFS data and fit for UHL- $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R -space and **h**, k -space. **k**, WT-EXAFS plot of UHL- $\text{Lu}_{\text{PM}}/\text{Fe}_2\text{O}_3$.



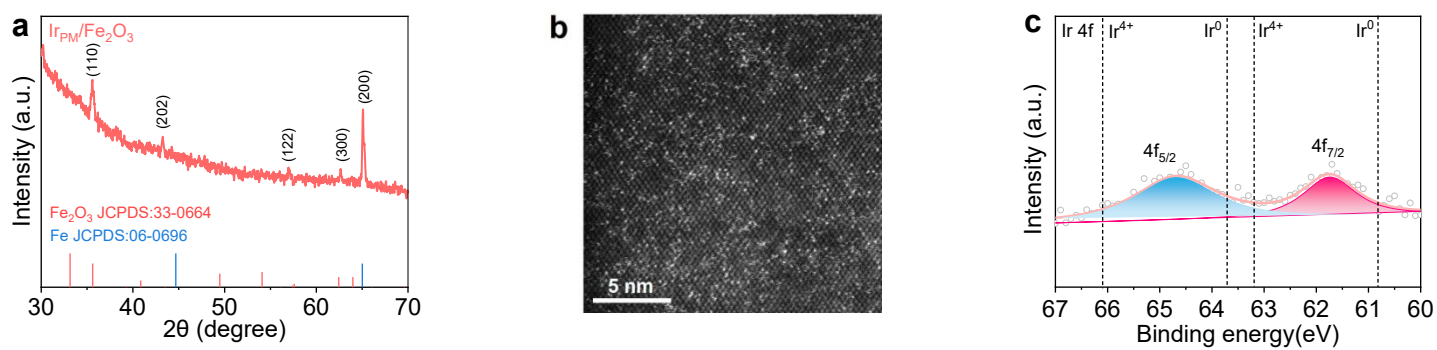
Supplementary Fig. 38 Structural characterization of $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Hf 4f XPS spectra of $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Hf L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Hf L-edge EXAFS signals. **f,g**, Hf L-edge EXAFS data and fit for UHL- $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R -space and **h**, k -space. **k**, WT-EXAFS plots of UHL- $\text{Hf}_{\text{PM}}/\text{Fe}_2\text{O}_3$.



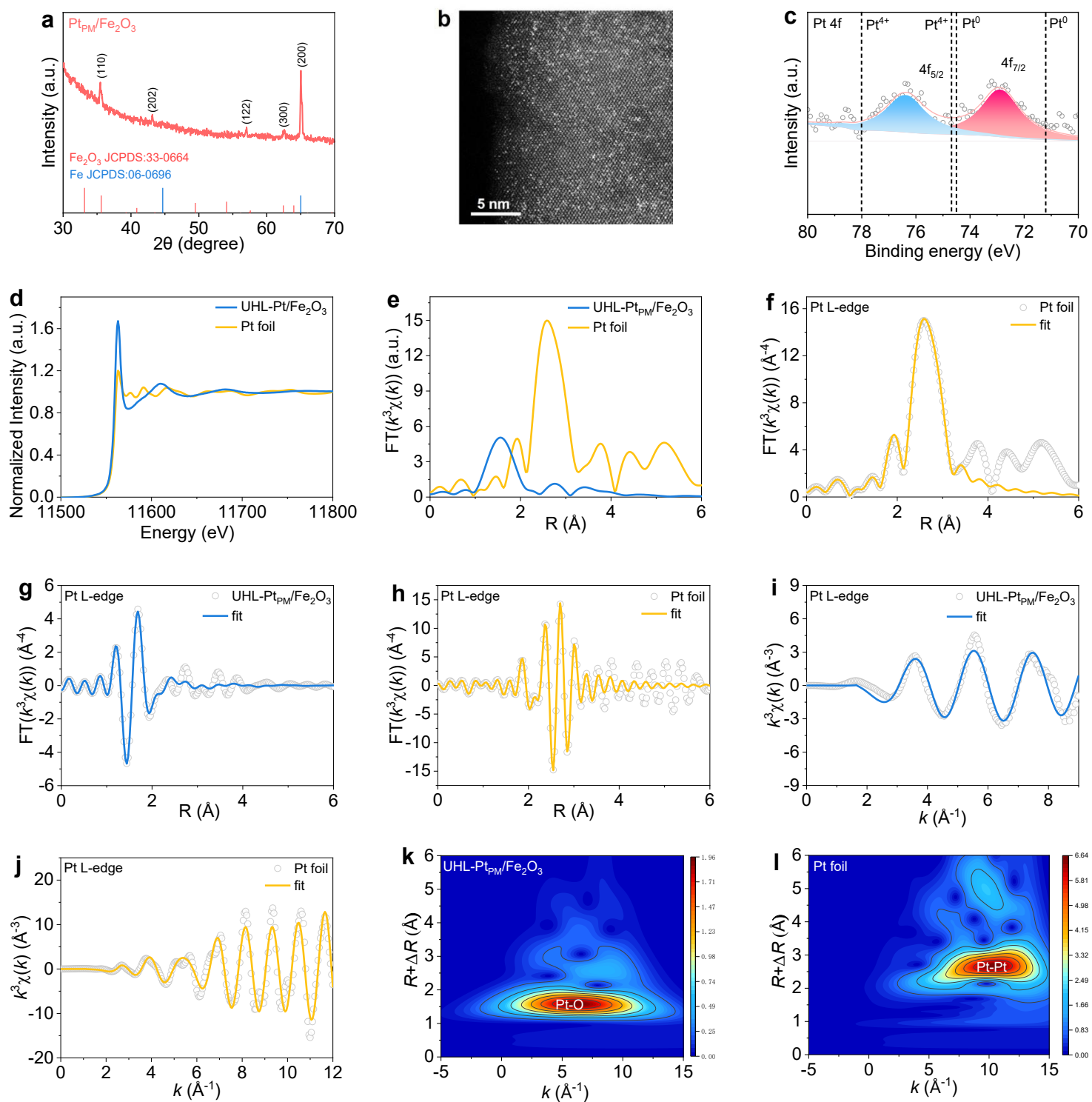
Supplementary Fig. 39 Structural characterization of W_{PM}/Fe_2O_3 . **a**, XRD patterns of W_{PM}/Fe_2O_3 . **b**, HAADF-STEM image of W_{PM}/Fe_2O_3 . **c**, W 4f XPS spectra of W_{PM}/Fe_2O_3 . **d**, Normalized W L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental W L-edge EXAFS signals. **f-h**, W L-edge EXAFS data and fit for UHL- W_{PM}/Fe_2O_3 in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- W_{PM}/Fe_2O_3 and W foil.



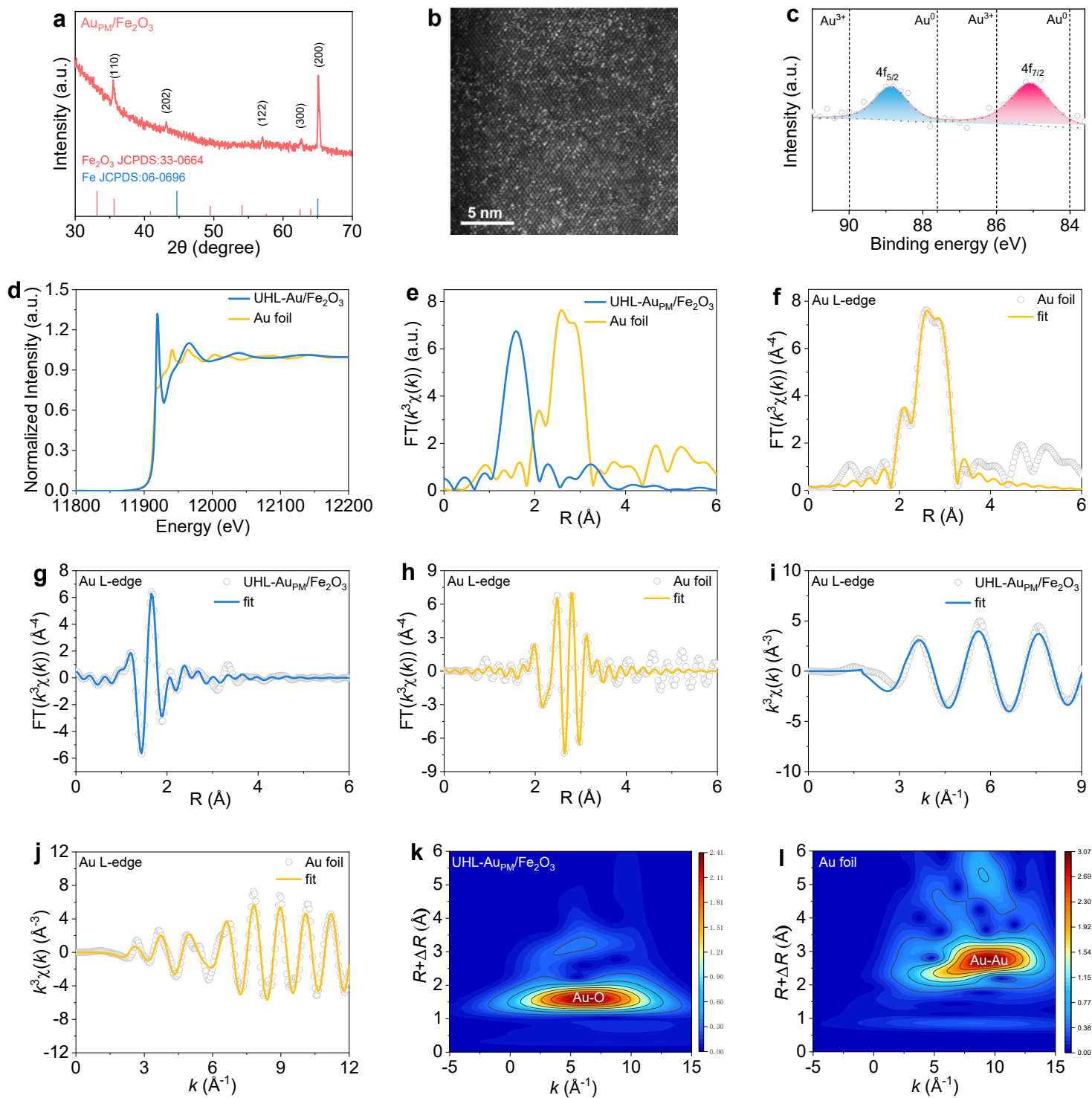
Supplementary Fig. 40 Structural characterization of $\text{O}_{\text{SPM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{O}_{\text{SPM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{O}_{\text{SPM}}/\text{Fe}_2\text{O}_3$. **c**, Os 4f XPS spectra of $\text{O}_{\text{SPM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Os L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Os L-edge EXAFS signals. **f**, Os L-edge EXAFS data and fit for UHL-Os_{PM}/Fe₂O₃ in R-space and **g**, k-space. **h**, WT-EXAFS plots of UHL-Os_{PM}/Fe₂O₃.



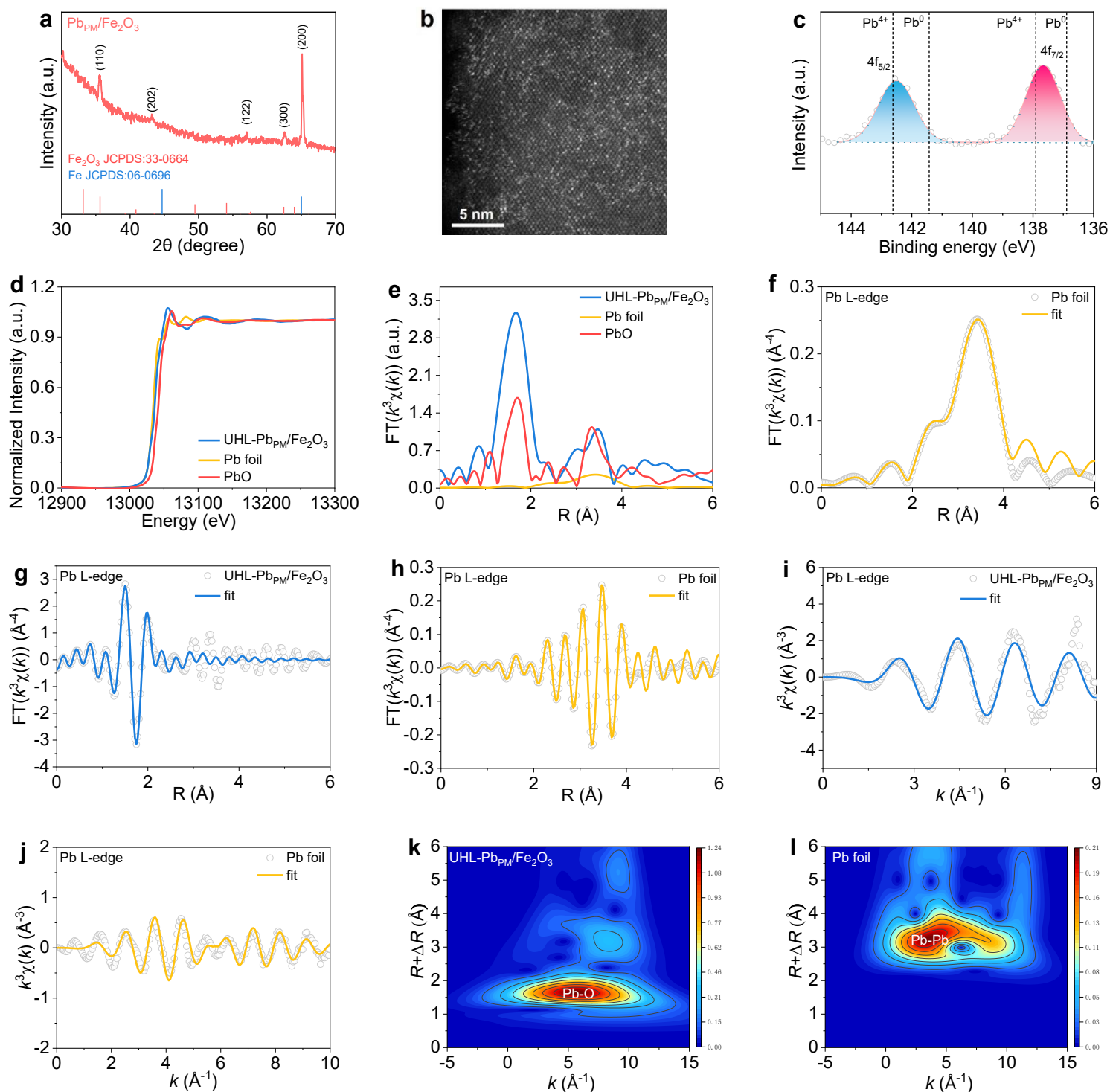
Supplementary Fig. 41 Structural characterization of $\text{Ir}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Ir}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Ir}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Ir 4f XPS spectra of $\text{Ir}_{\text{PM}}/\text{Fe}_2\text{O}_3$.



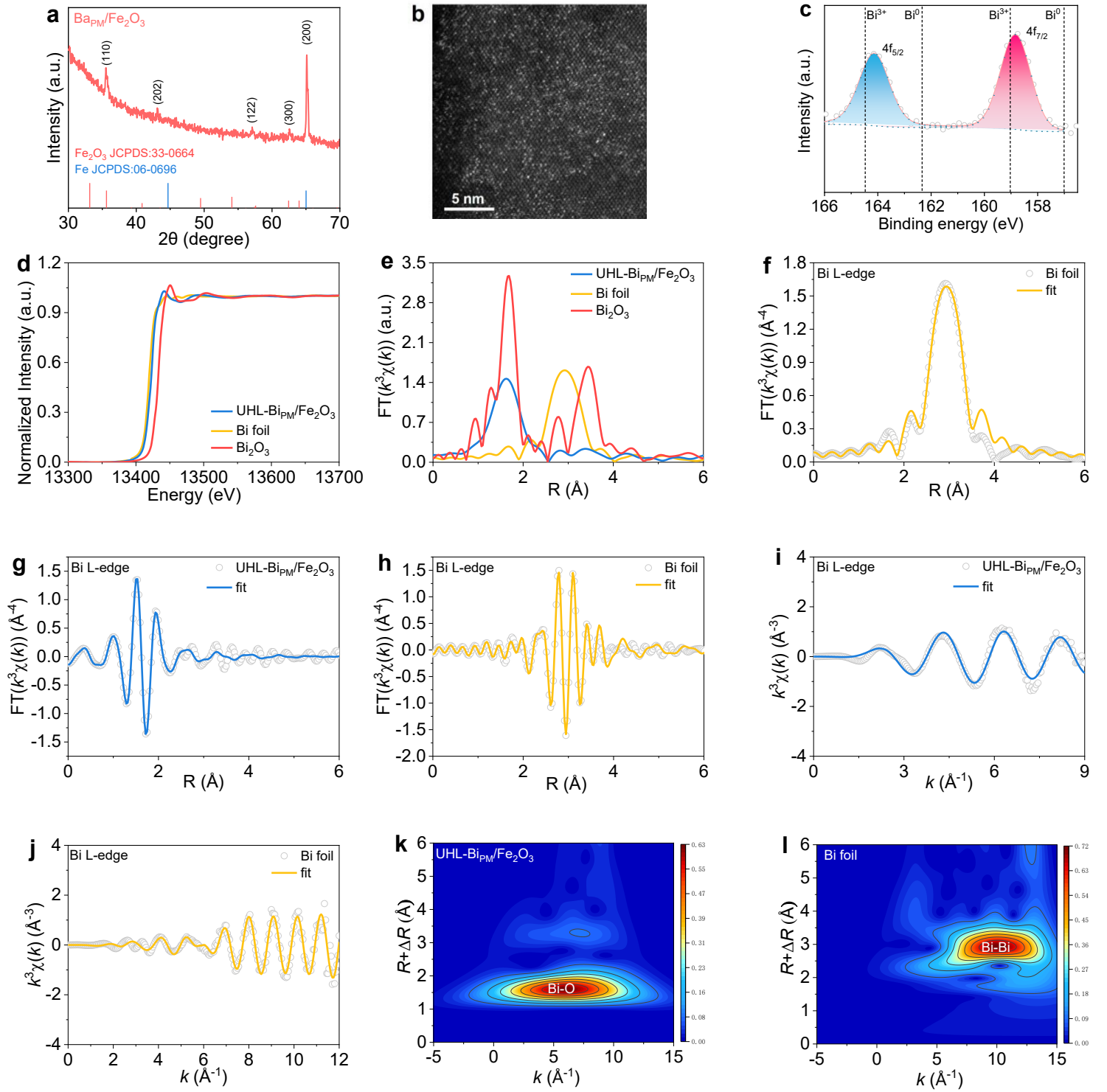
Supplementary Fig. 42 Structural characterization of Pt_{PM}/Fe₂O₃. **a**, XRD patterns of Pt_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Pt_{PM}/Fe₂O₃. **c**, Pt 4f XPS spectra of Pt_{PM}/Fe₂O₃. **d**, Normalized Pt L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Pt L-edge EXAFS signals. **f-h**, Pt L-edge EXAFS data and fit for UHL-Pt_{PM}/Fe₂O₃ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL-Pt_{PM}/Fe₂O₃ and Pt foil.



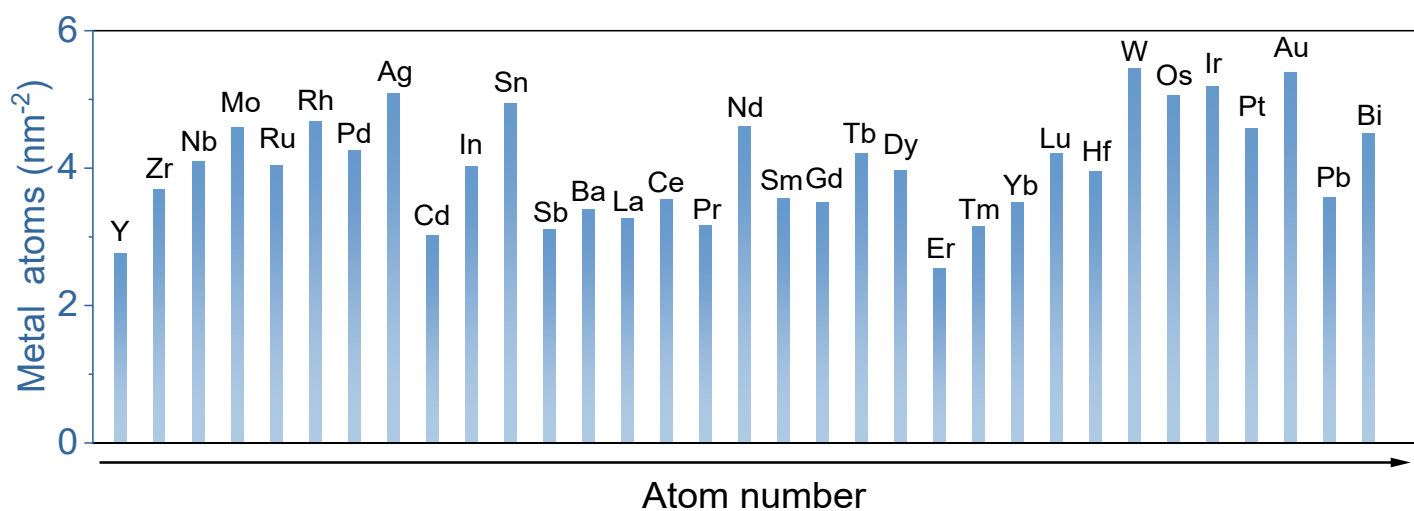
Supplementary Fig. 43 Structural characterization of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Au 4f XPS spectra of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Au L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Au L-edge EXAFS signals. **f-h**, Au L-edge EXAFS data and fit for UHL- $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Au foil.



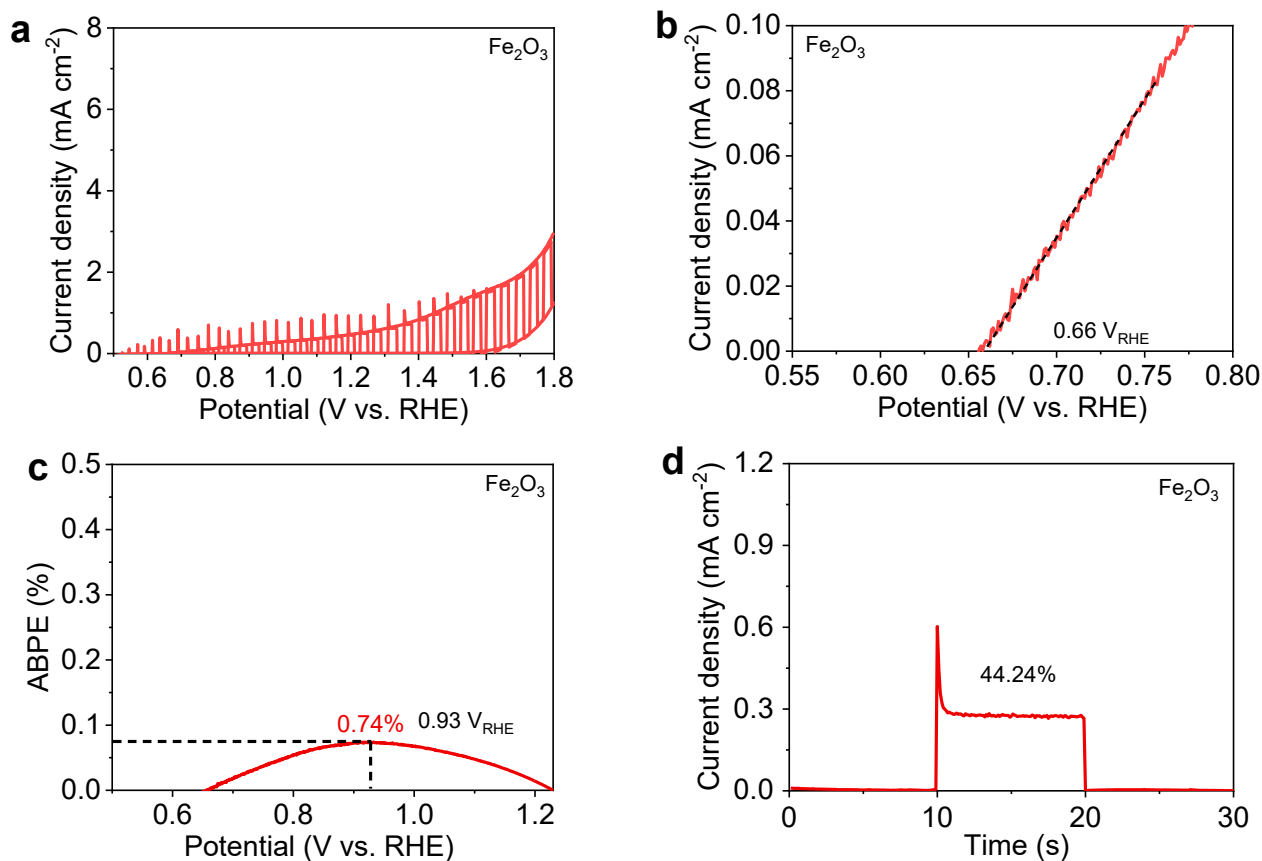
Supplementary Fig. 44 Structural characterization of Pb_{PM}/Fe₂O₃. **a**, XRD patterns of Pb_{PM}/Fe₂O₃. **b**, HAADF-STEM image of Pb_{PM}/Fe₂O₃. **c**, Pb 4f XPS spectra of Pb_{PM}/Fe₂O₃. **d**, Normalized Pb *L*-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed *R*-space of the experimental Pb *L*-edge EXAFS signals. **f-h**, Pb *L*-edge EXAFS data and fit for UHL-Pb_{PM}/Fe₂O₃ in *R*-space and **i, j**, *k*-space. **k, l**, WT-EXAFS plots of UHL-Pb_{PM}/Fe₂O₃ and Pb foil.



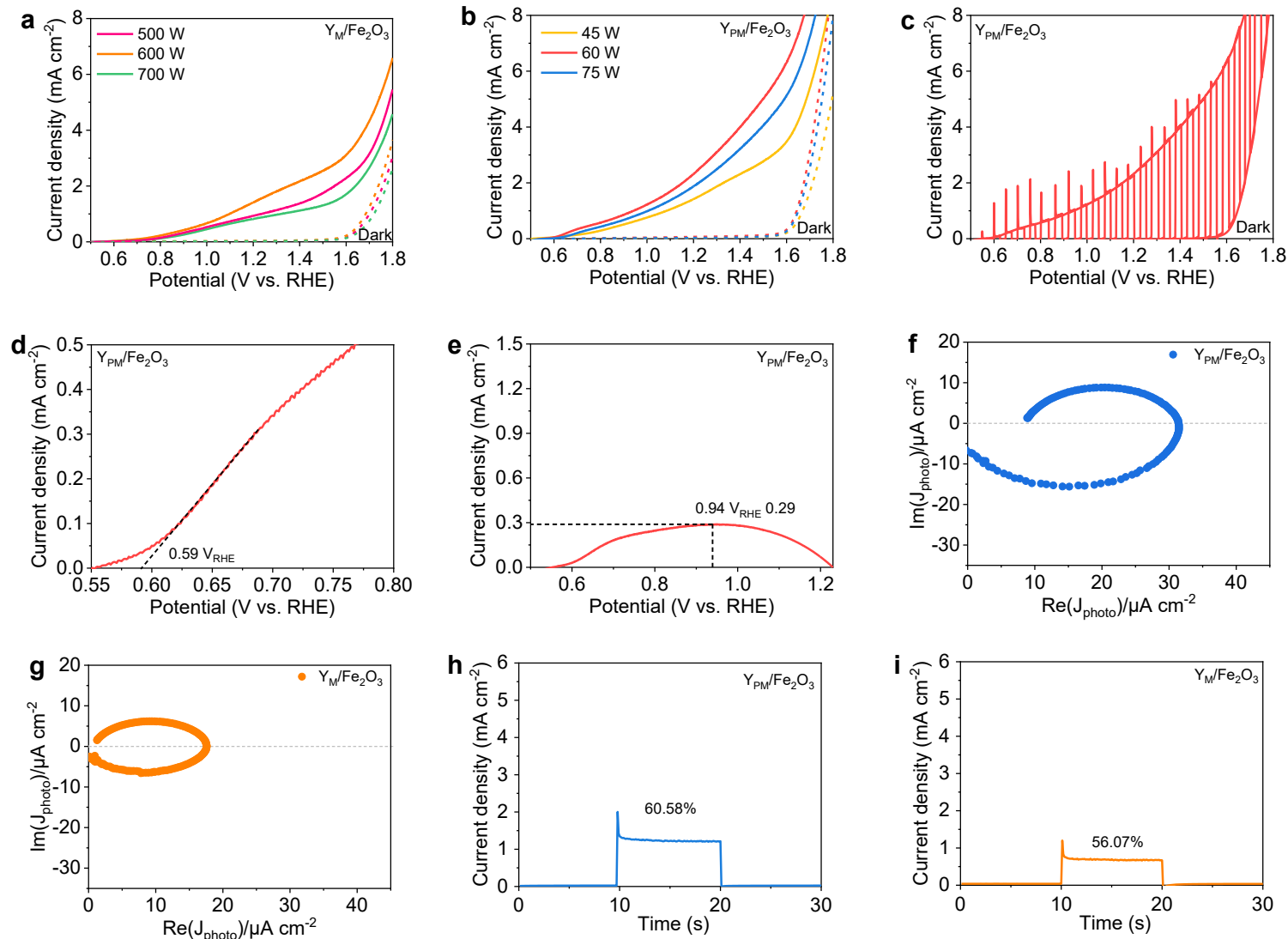
Supplementary Fig. 45 Structural characterization of $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, XRD patterns of $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **b**, HAADF-STEM image of $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Bi 4f XPS spectra of $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Normalized Bi L-edge X-ray absorption near-edge structure (XANES) spectra. **e**, Fourier-transformed R-space of the experimental Bi L-edge EXAFS signals. **f-h**, Bi L-edge EXAFS data and fit for UHL- $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$ in R-space and **i, j**, k-space. **k, l**, WT-EXAFS plots of UHL- $\text{Bi}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Bi foil.



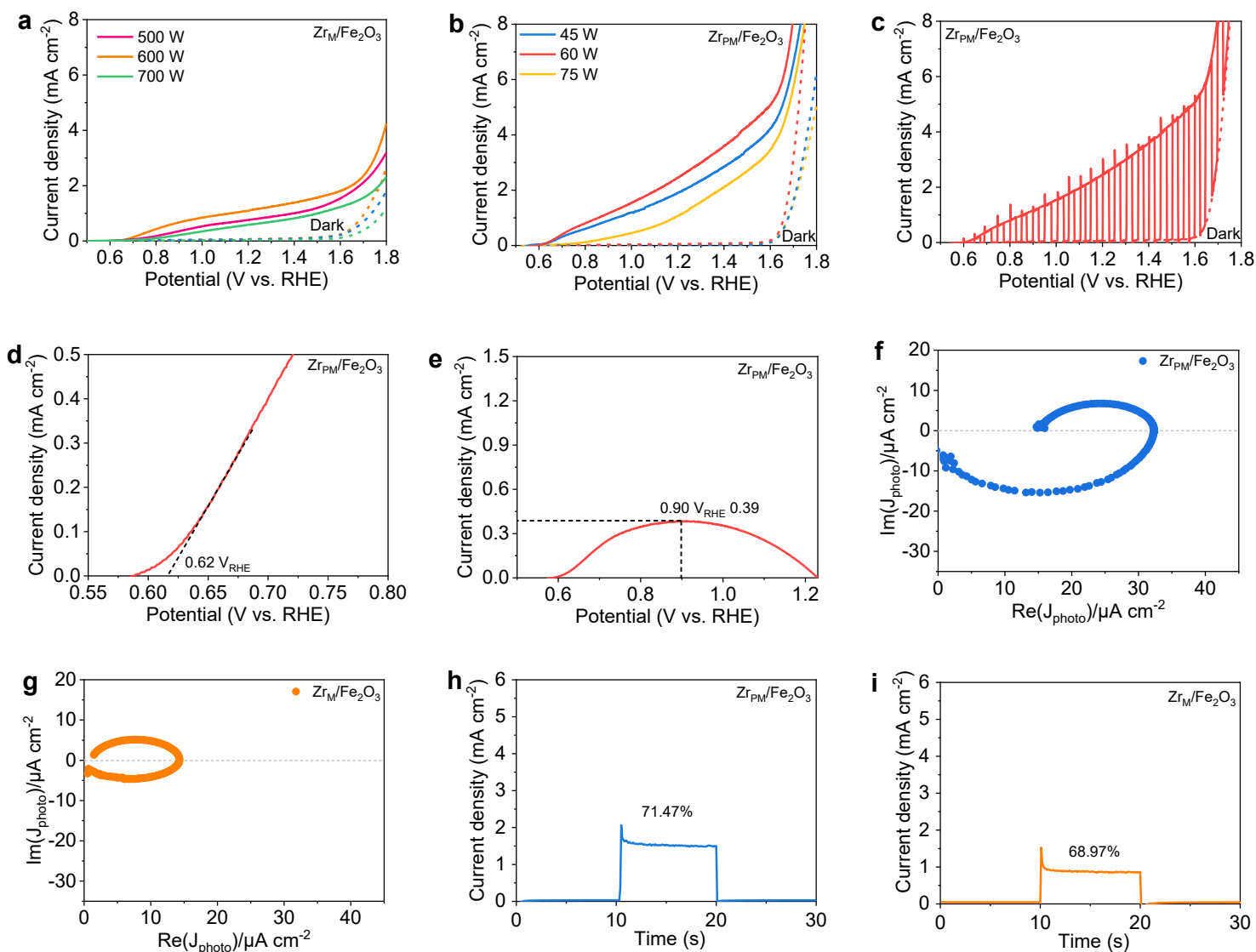
Supplementary Fig. 46 Surface concentration of metal atoms. Number of metal atoms per square nanometre on Fe₂O₃ carrier surface for UHL-SACs



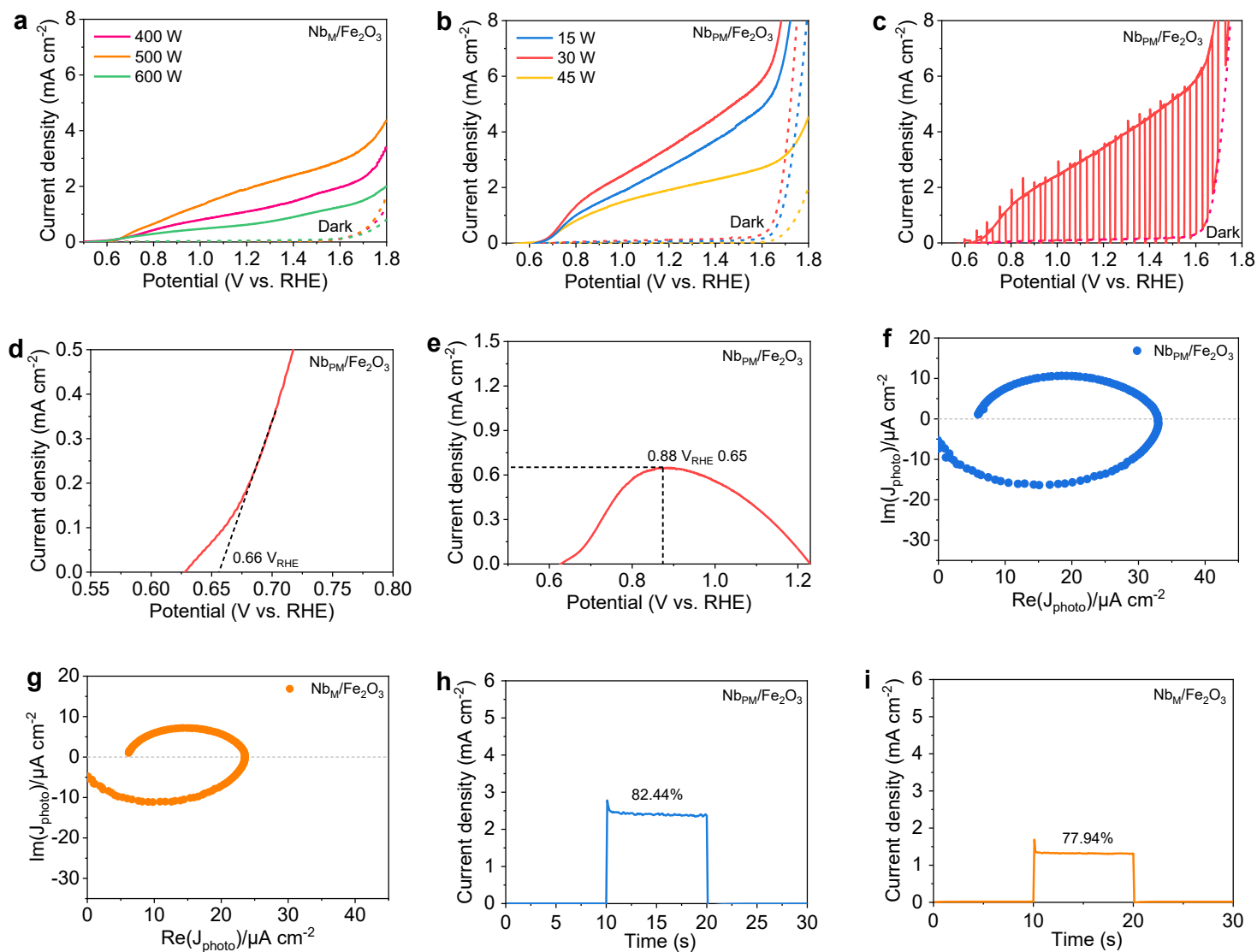
Supplementary Fig. 47 Photoelectrochemical performance measurement of Fe_2O_3 . **a**, J - V and chopped curves of Fe_2O_3 . **b**, Onset potential of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Applied bias photon-to-current efficiency of $\text{Y}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Transient photocurrent of Fe_2O_3 . All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



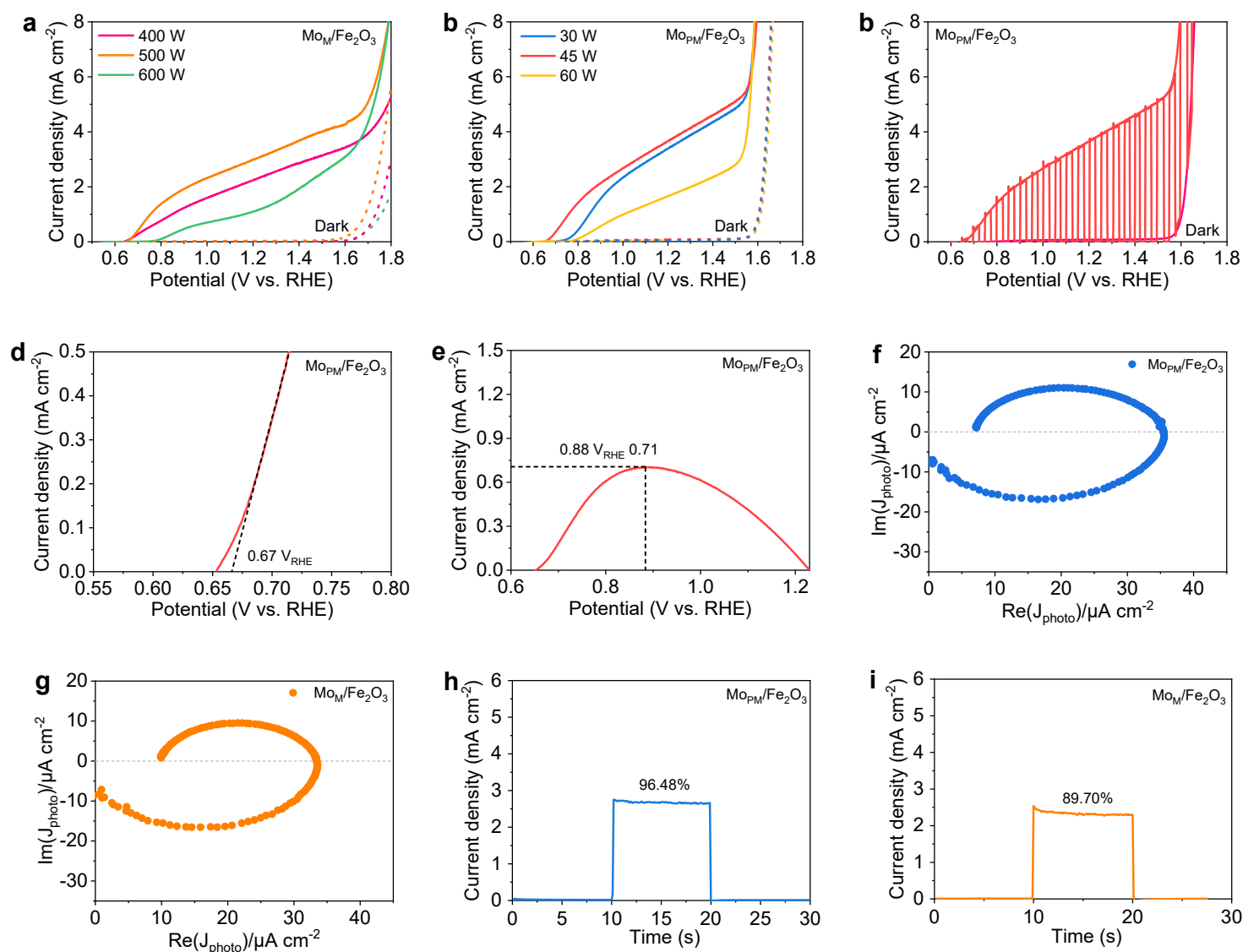
Supplementary Fig. 48 Photoelectrochemical performance measurement of $Y_{PM}Fe_2O_3$. **a**, J - V curves of Y -loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment, **b**, J - V curves of Y single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W), **c**, Chopped J - V curve of Y_{PM}/Fe_2O_3 . **d**, Onset potential of Y_{PM}/Fe_2O_3 . **e**, Applied bias photon-to-current efficiency of Y_{PM}/Fe_2O_3 . **f, g**, IMPS responses of Y_{PM}/Fe_2O_3 and Y_M/Fe_2O_3 applied at $1\ V_{RHE}$. **h, i**, Transient photocurrent density of Y_{PM}/Fe_2O_3 and Y_M/Fe_2O_3 . All measurements were performed in 1 M KOH under AM 1.5G illumination ($100\ mW\ cm^{-2}$).



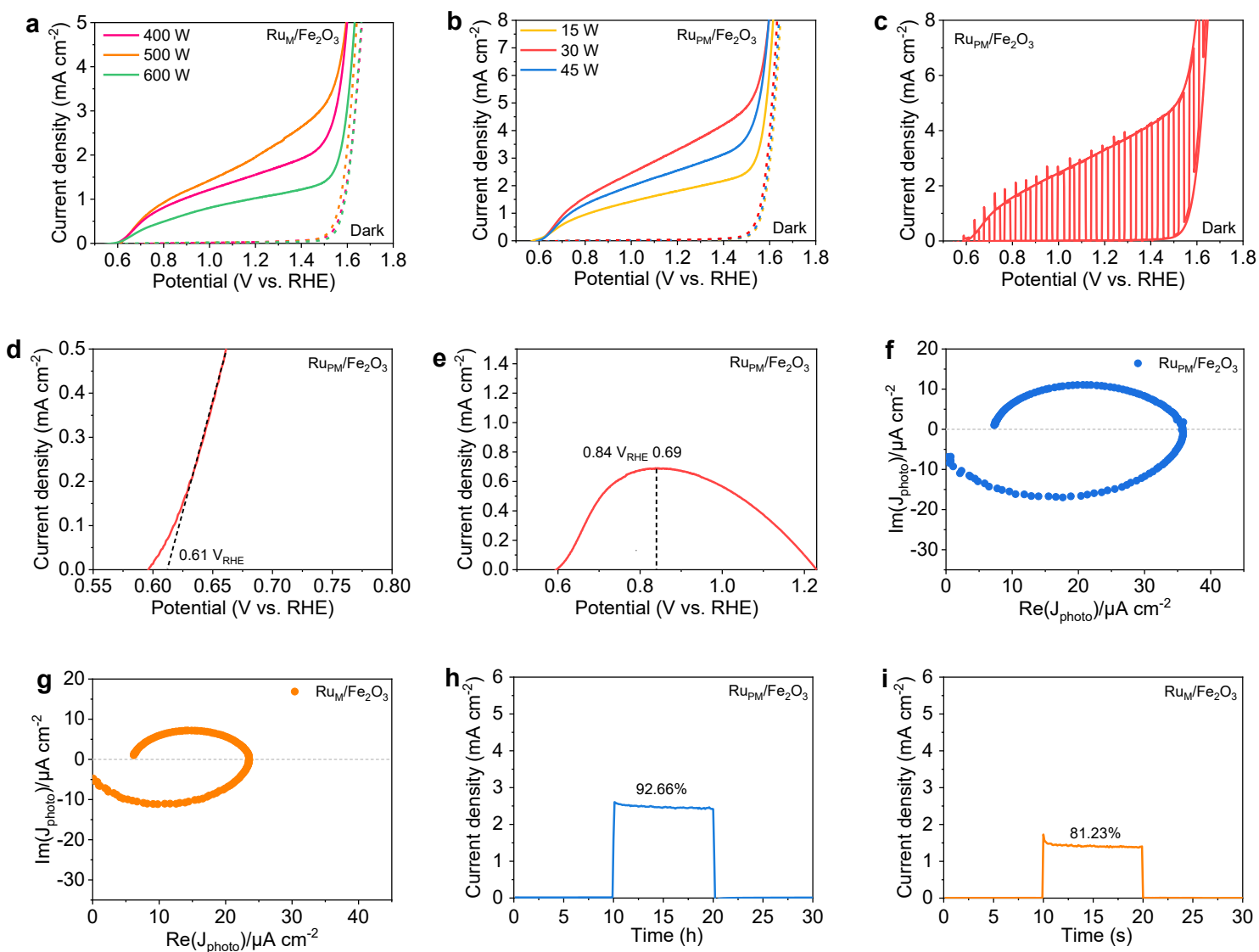
Supplementary Fig. 49 Photoelectrochemical performance measurement of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **A**, J - V curves of Zr-loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **B**, J - V curves of Zr single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W). **C**, Chopped J - V curve of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **D**, Onset potential of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **E**, Applied bias photon-to-current efficiency of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **F**, **G**, IMPS responses of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Zr}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **H**, **I**, Transient photocurrent density of $\text{Zr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Zr}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



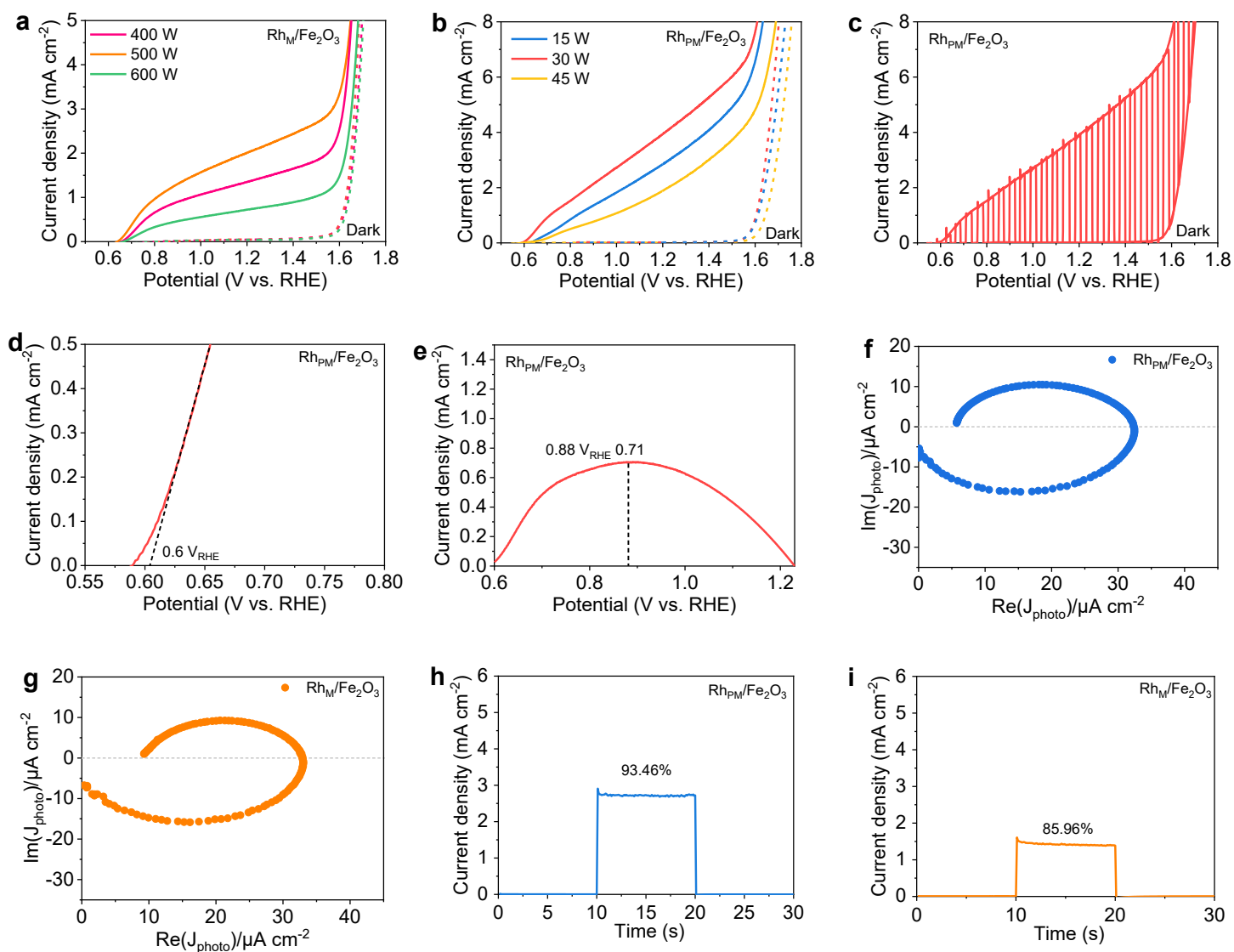
Supplementary Fig. 50 Photoelectrochemical performance measurement of Nb_{PM}/Fe₂O₃. **a**, J - V curves of Nb-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Nb single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of Nb_{PM}/Fe₂O₃. **d**, Onset potential of Nb_{PM}/Fe₂O₃. **e**, Applied bias photon-to-current efficiency of Nb_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Nb_{PM}/Fe₂O₃ and Nb_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Nb_{PM}/Fe₂O₃ and Nb_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



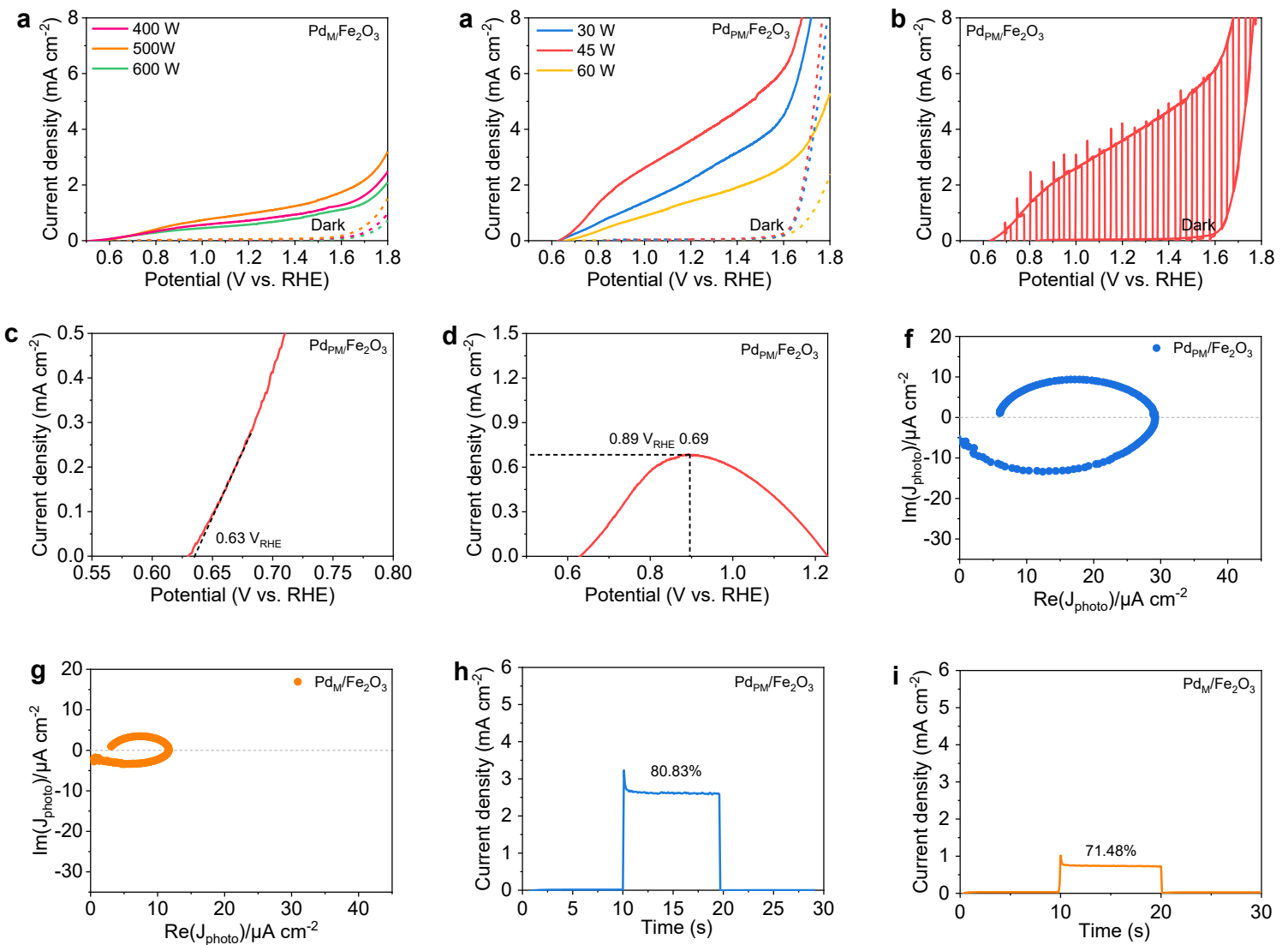
Supplementary Fig. 51 Photoelectrochemical performance measurement of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Mo -loaded Fe_2O_3 annealed directly by microwave at various power (400 W, 600 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Mo single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$. **e** Applied bias photon-to-current efficiency of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Mo}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Mo}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Mo}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



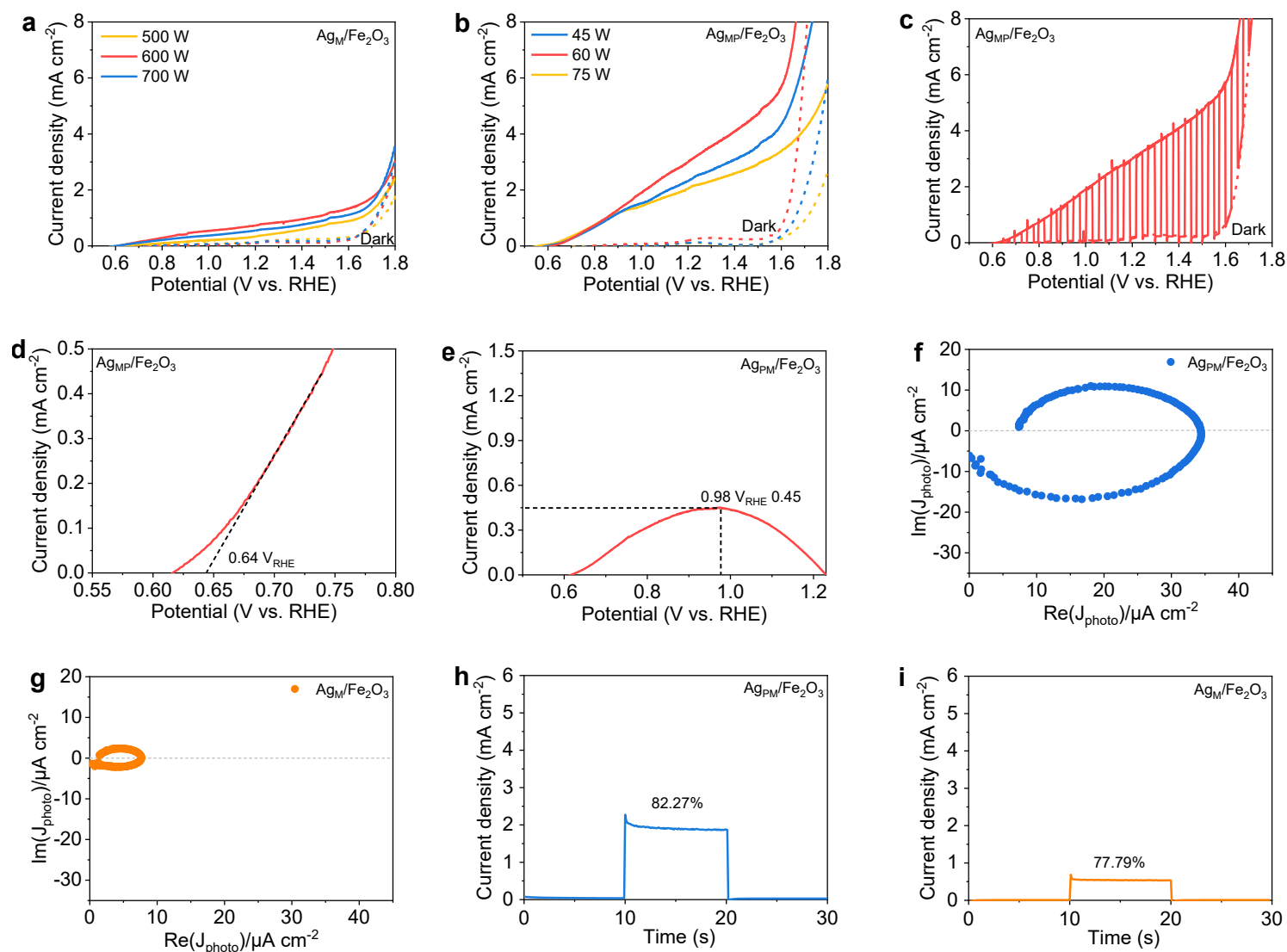
Supplementary Fig. 52 Photoelectrochemical performance measurement of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Ru -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Ru single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$. **e** Applied bias photon-to-current efficiency of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Ru}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_RHE . **h**, **i**, Transient photocurrent density of $\text{Ru}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Ru}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



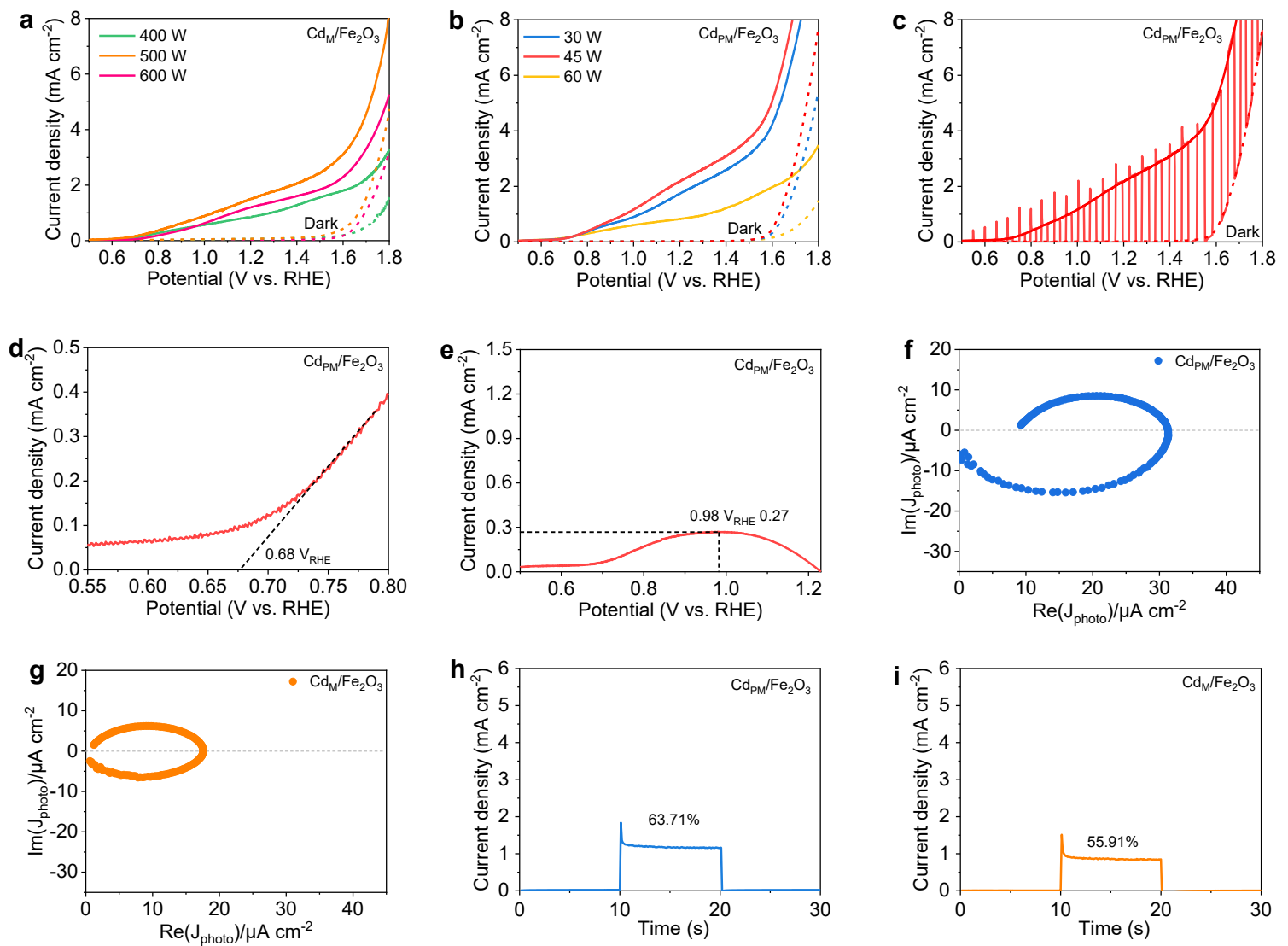
Supplementary Fig. 53 Photoelectrochemical performance measurement of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Rh-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Rh single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e** Applied bias photon-to-current efficiency of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Rh}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Rh}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Mo}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



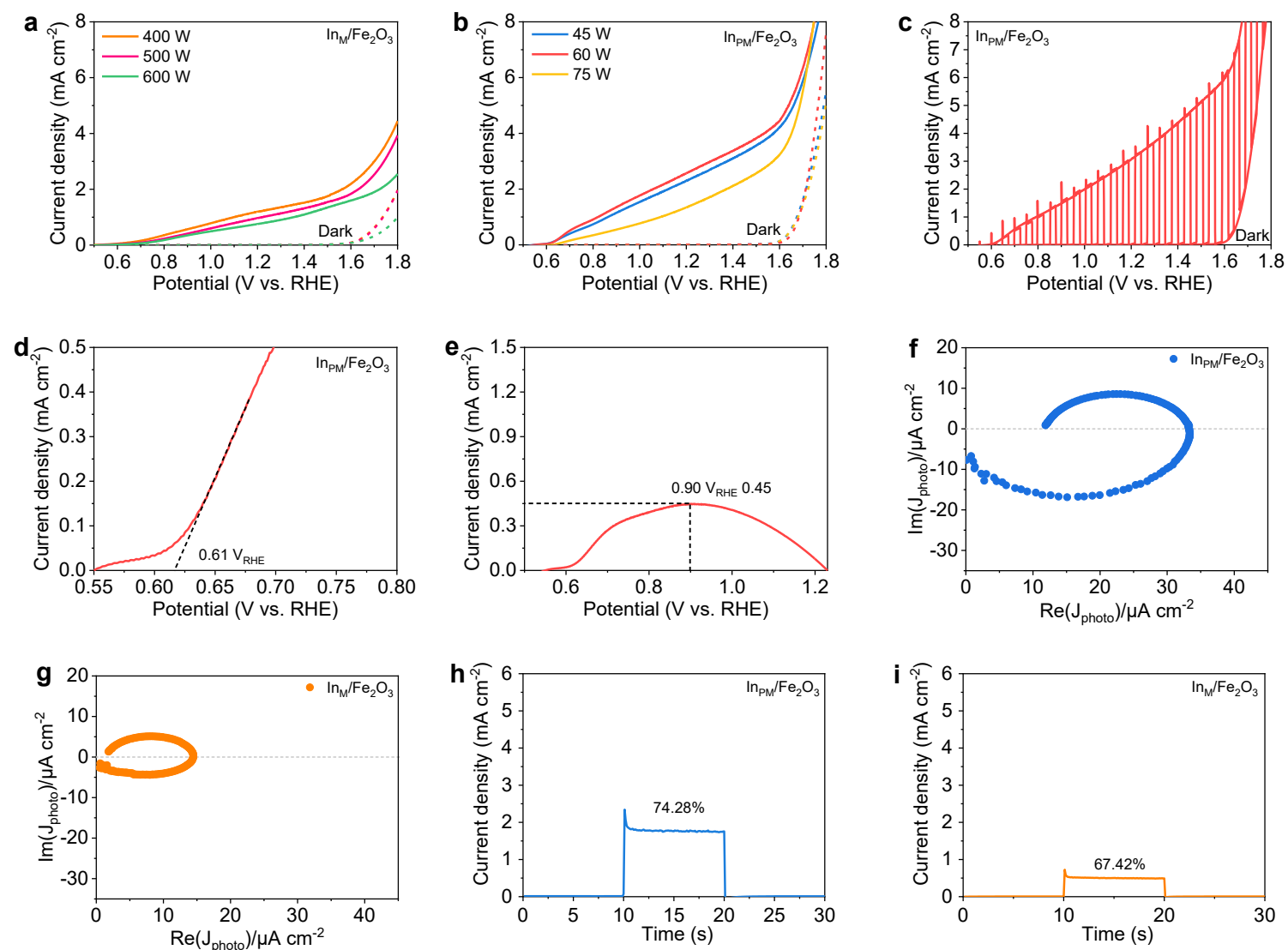
Supplementary Fig. 54 Photoelectrochemical performance measurement of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Pd -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Pd single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Pd}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Pd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Mo}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



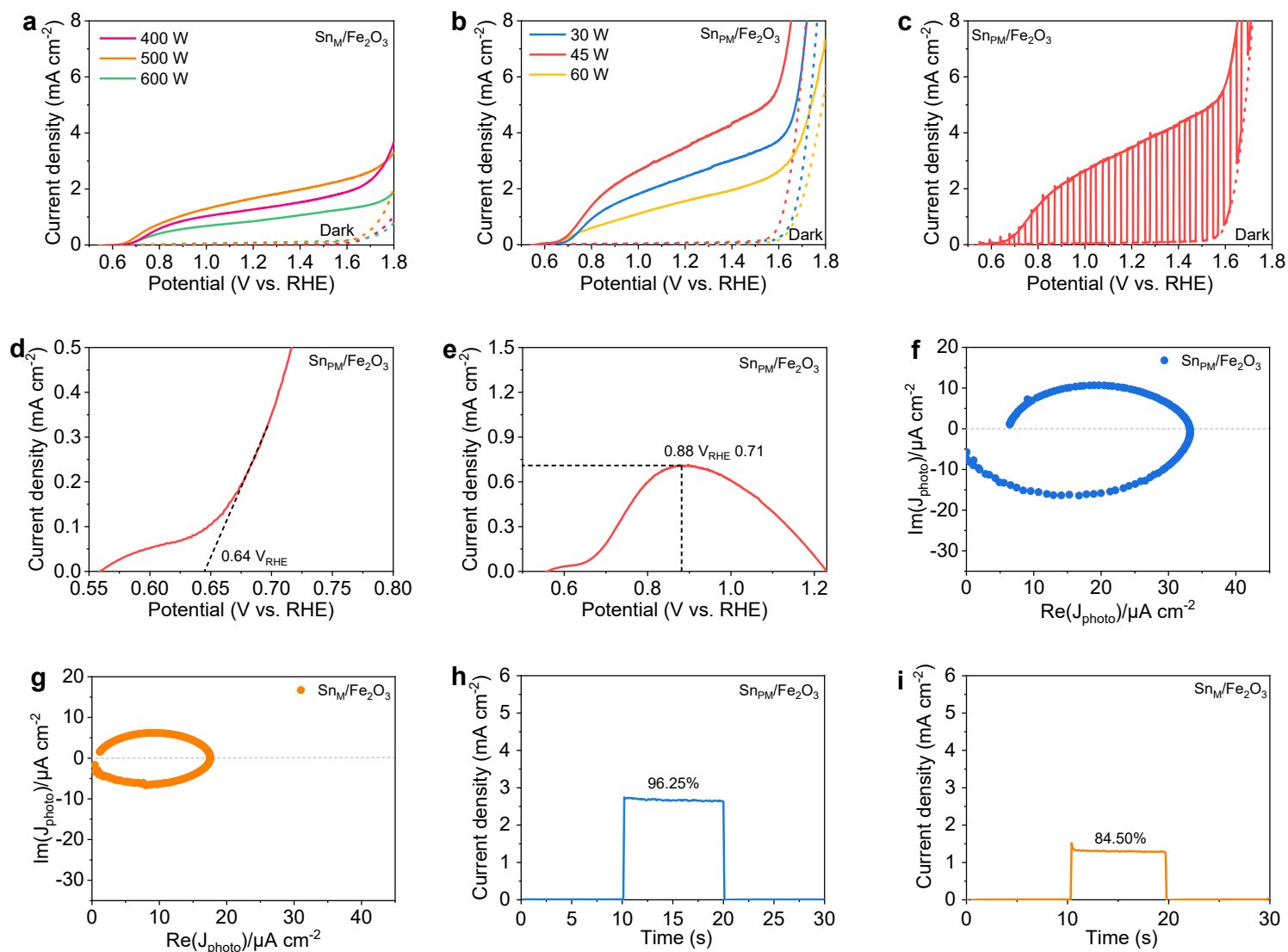
Supplementary Fig. 55 Photoelectrochemical performance measurement of AgPM/Fe₂O₃. **a**, *J-V* curves of Ag-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, *J-V* curves of Ag single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped *J-V* curve of AgPM/Fe₂O₃. **d**, Onset potential of AgPM/Fe₂O₃. **e**, Applied bias photon-to-current efficiency of AgPM/Fe₂O₃. **f**, **g**, IMPS responses of AgPM/Fe₂O₃ and AgM/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of AgPM/Fe₂O₃ and AgM/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



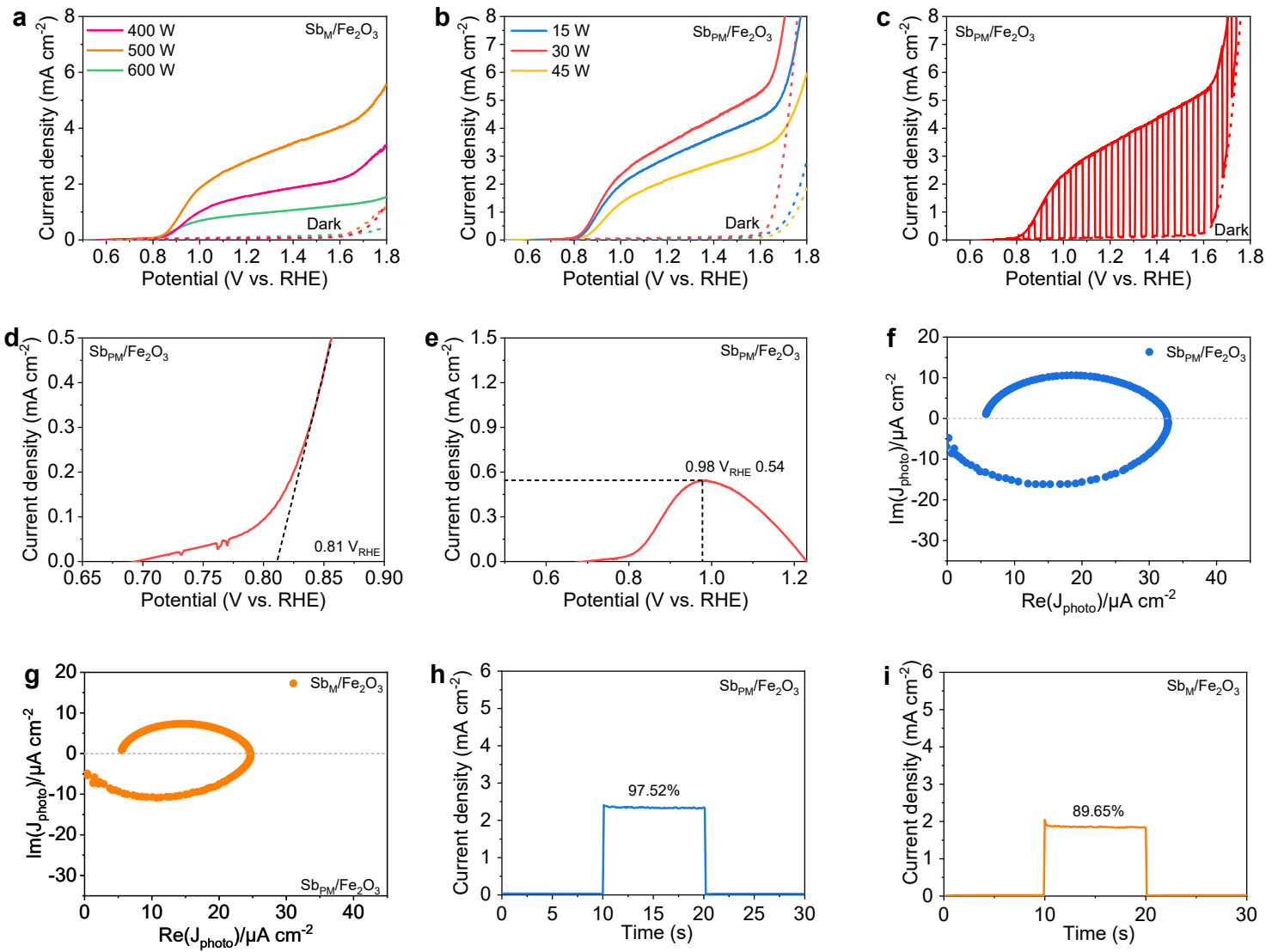
Supplementary Fig. 56 Photoelectrochemical performance measurement of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Cd -loaded Fe_2O_3 annealed directly by microwave at various power (400 W, 600 W, and 800 W) without plasma pretreatment. **b**, J - V curves of Cd single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Cd}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Cd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Cd}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



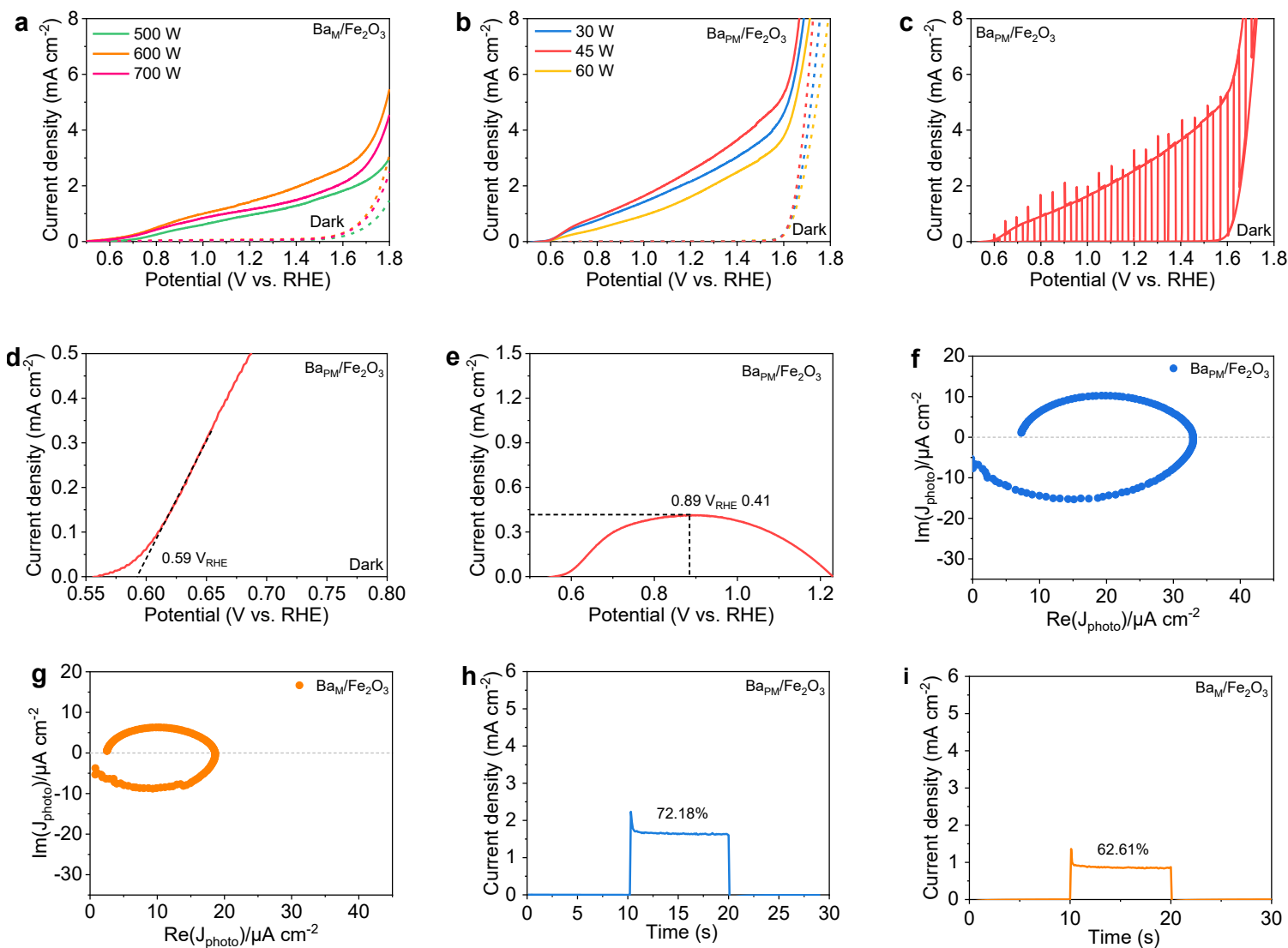
Supplementary Fig. 57 Photoelectrochemical performance measurement of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of In-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of In single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W). **c**, Chopped J - V curve of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{In}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{In}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{In}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



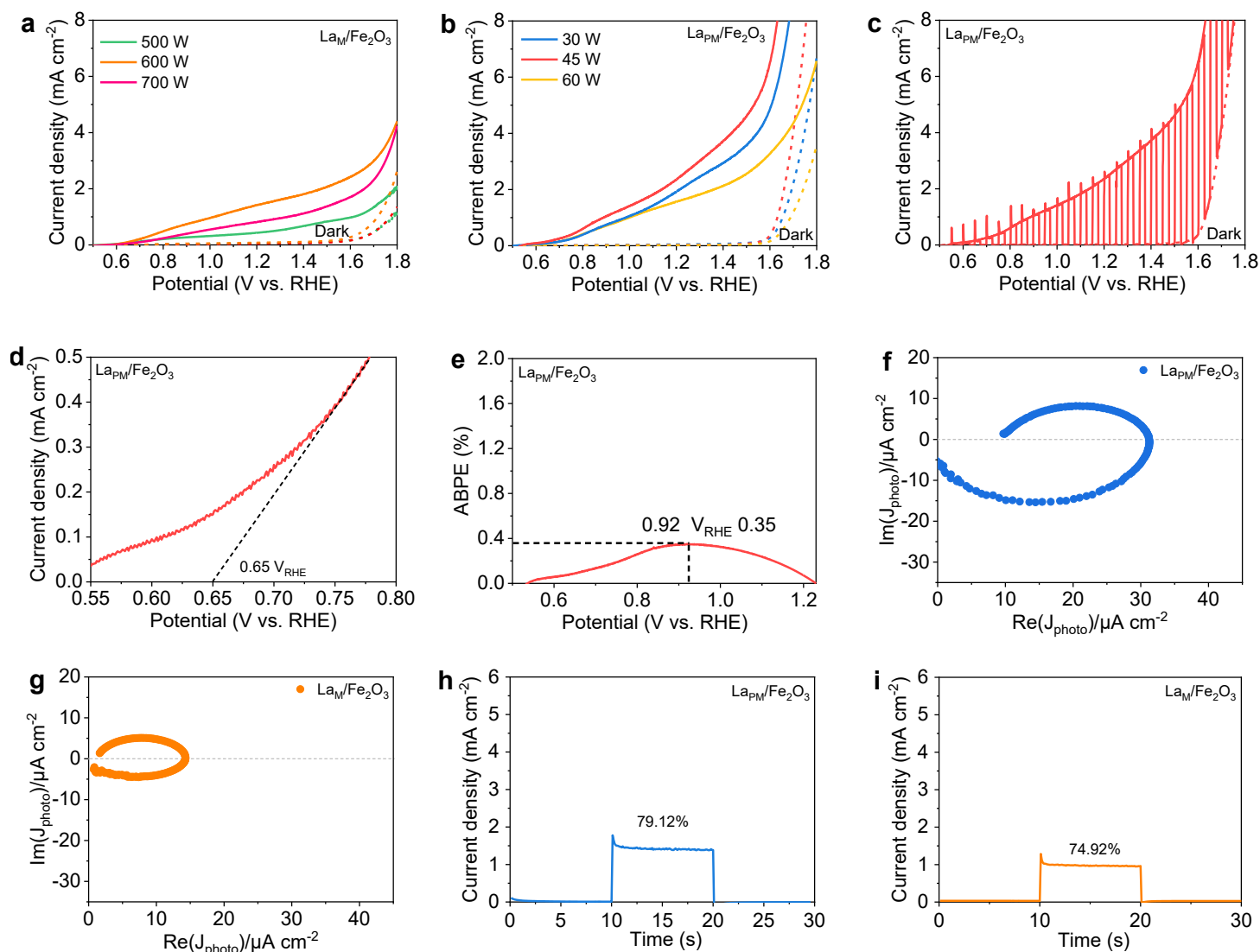
Supplementary Fig. 58 Photoelectrochemical performance measurement of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Sn-loaded Fe_2O_3 annealed directly by microwave at various power (400 W, 600 W, and 800 W) without plasma pretreatment. **b**, J - V curves of Sn single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f, g**, IMPS responses of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Sn}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h, i**, Transient photocurrent density of $\text{Sn}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Sn}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



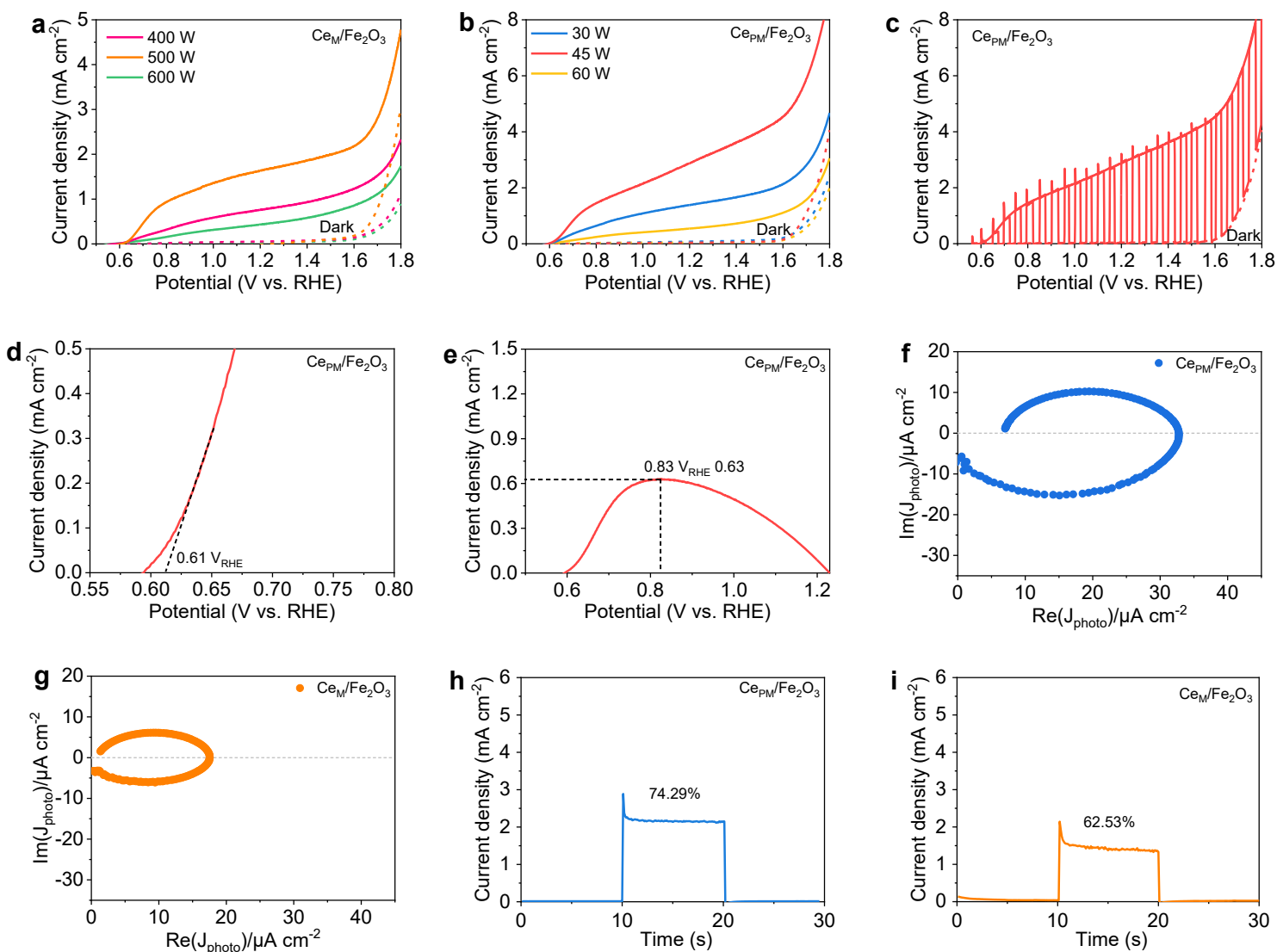
Supplementary Fig. 59 Photoelectrochemical performance measurement of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Sb -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Sb single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Sb}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Sb}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Sb}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



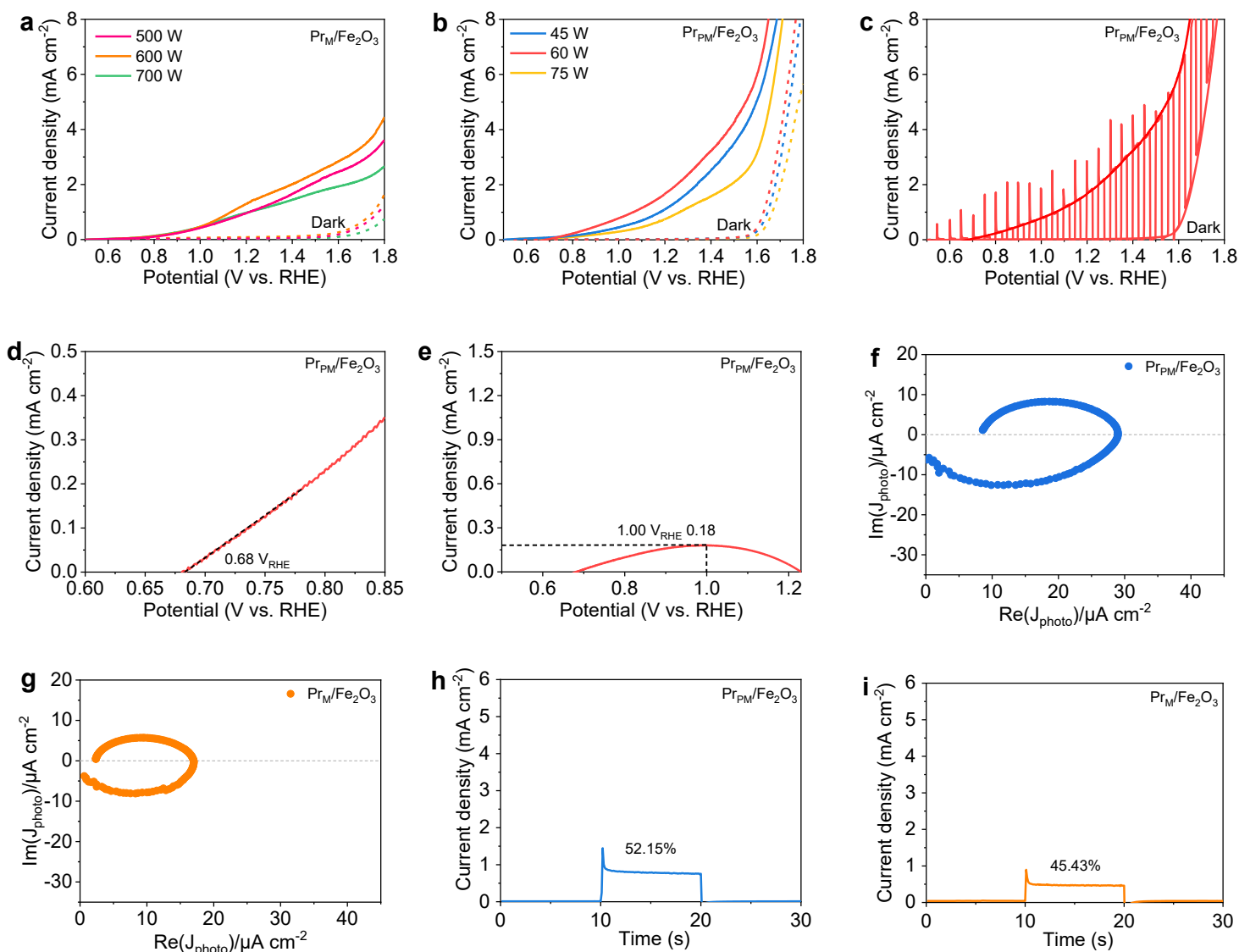
Supplementary Fig. 60 Photoelectrochemical performance measurement of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Ba-loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, J - V curves of Ba single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$. **f, g**, IMPS responses of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Ba}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h, i**, Transient photocurrent density of $\text{Ba}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Ba}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



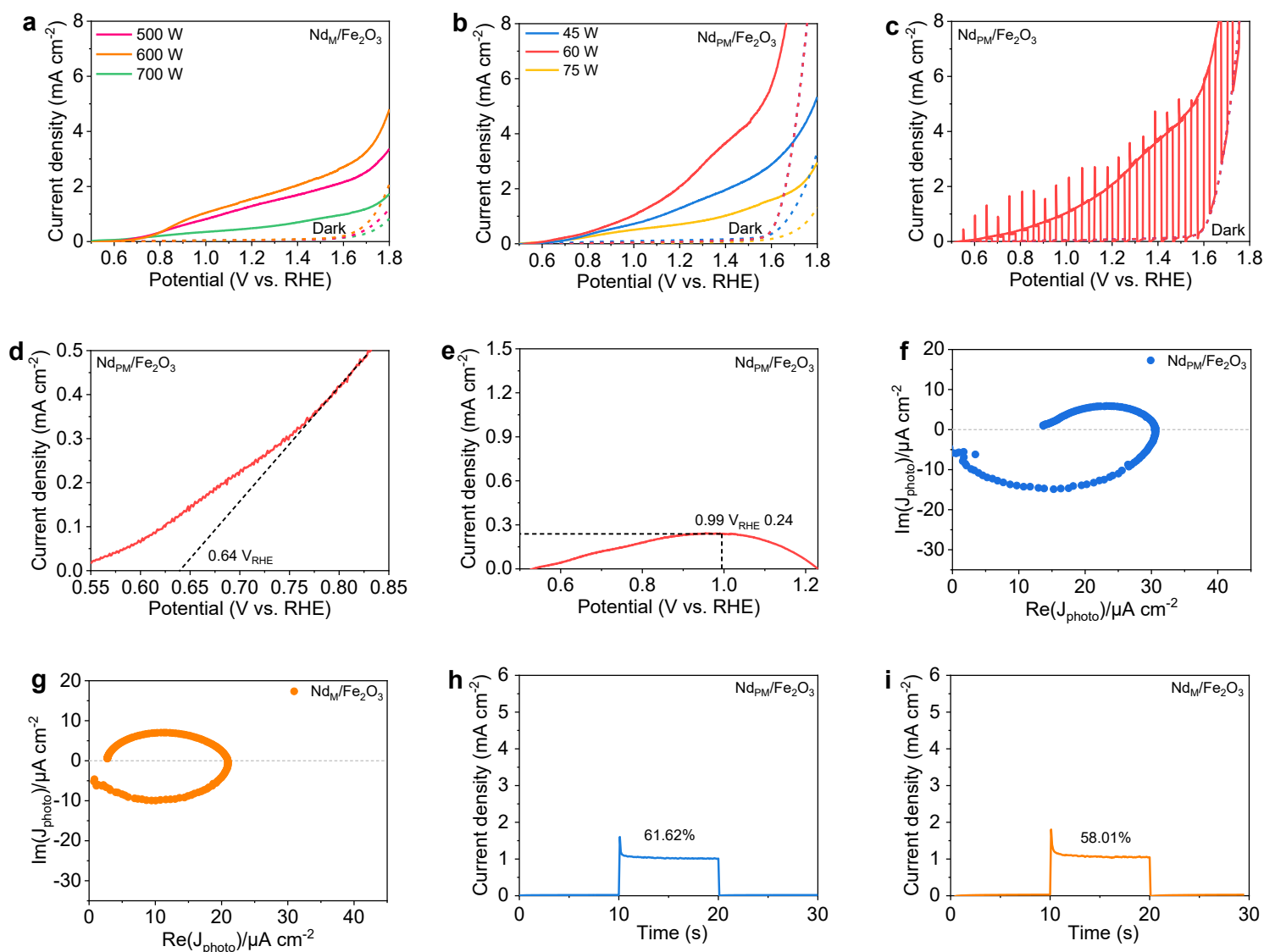
Supplementary Fig.61 Photoelectrochemical performance measurement of $\text{LaPM/Fe}_2\text{O}_3$. **a**, J - V curves of La -loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, J - V curves of La single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{LaPM/Fe}_2\text{O}_3$. **d**, Onset potential of $\text{LaPM/Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{LaPM/Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{LaPM/Fe}_2\text{O}_3$ and $\text{LaM/Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{LaPM/Fe}_2\text{O}_3$ and $\text{LaM/Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



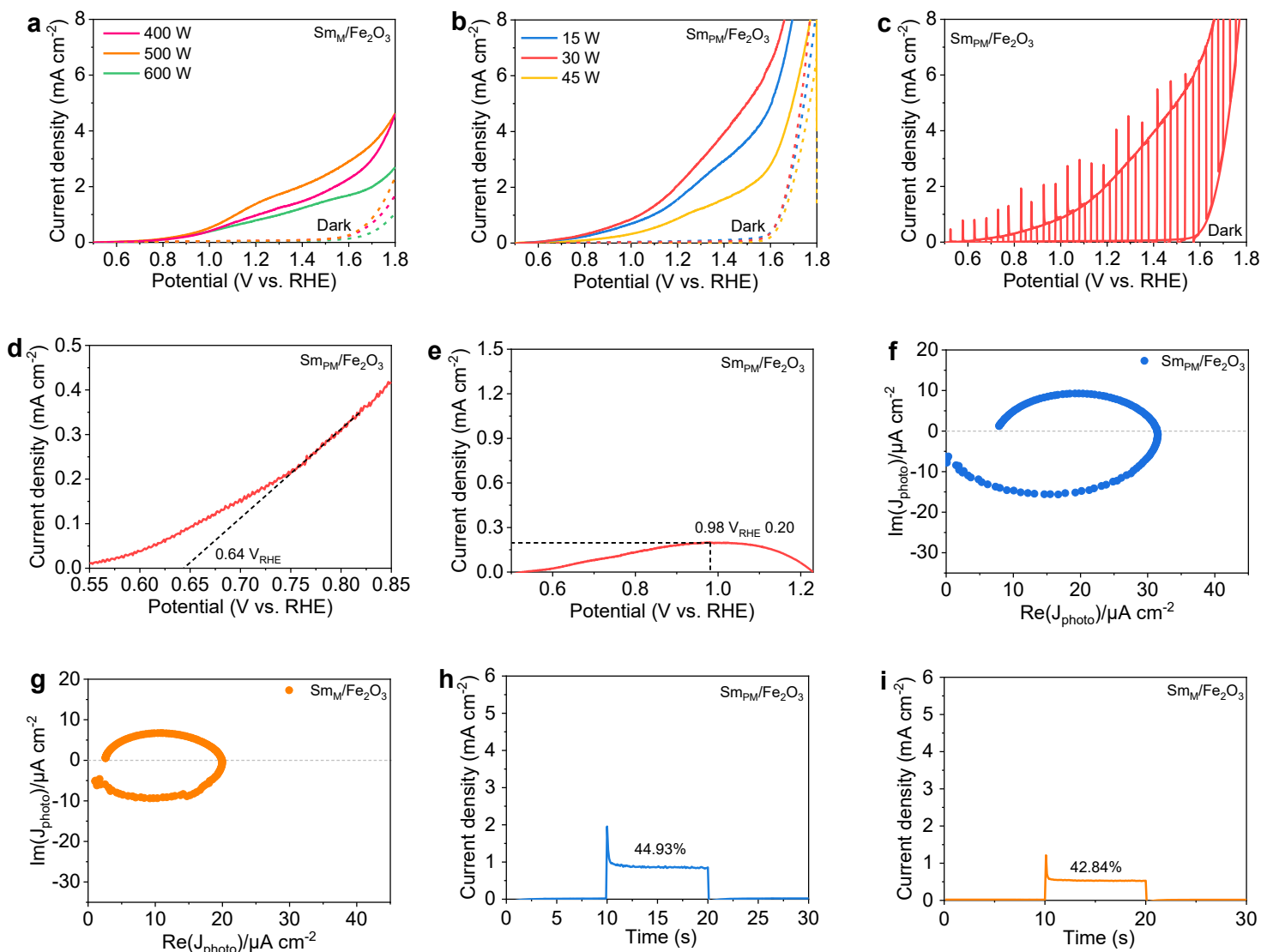
Supplementary Fig. 62 Photoelectrochemical performance measurement of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Ce-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Ce single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Ce}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Ce}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Ce}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



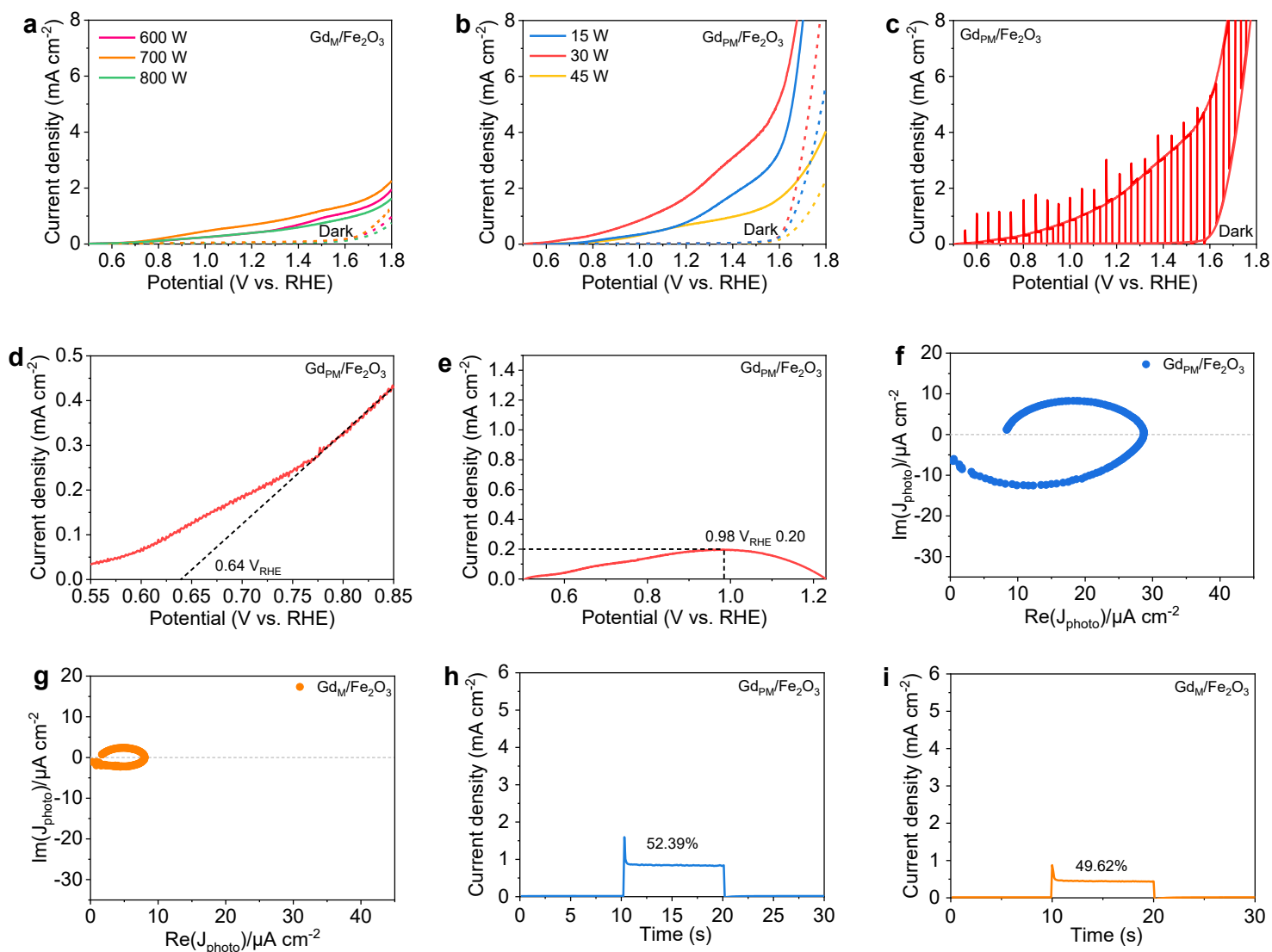
Supplementary Fig. 63 Photoelectrochemical performance measurement of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Pr-loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, J - V curves of Pr single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W). **c**, Chopped J - V curve of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pr}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Pr}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pr}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



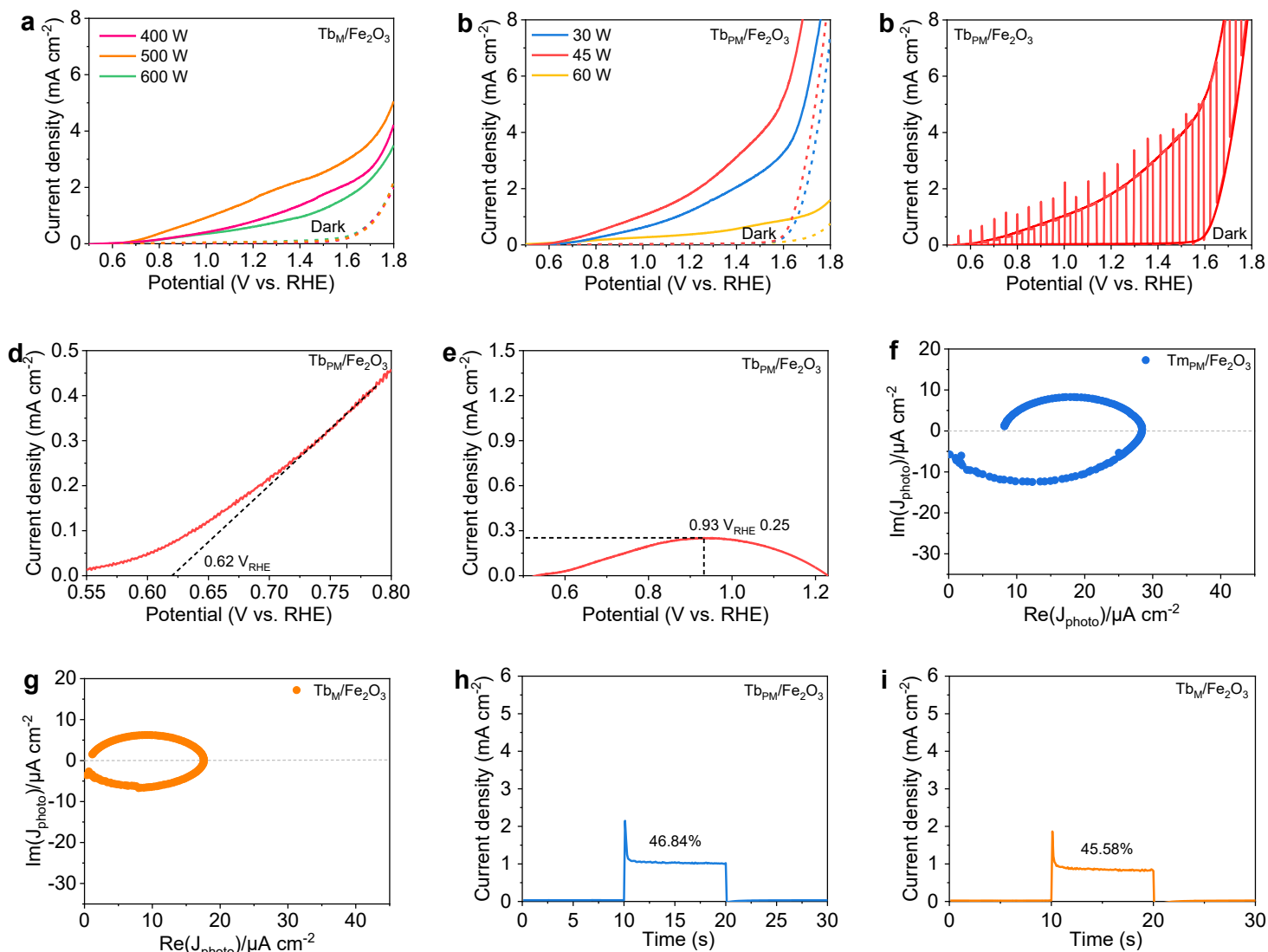
Supplementary Fig. 64 Photoelectrochemical performance measurement of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Nd -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Nd single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W). **c**, Chopped J - V curve of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Nd}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Nd}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Nd}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



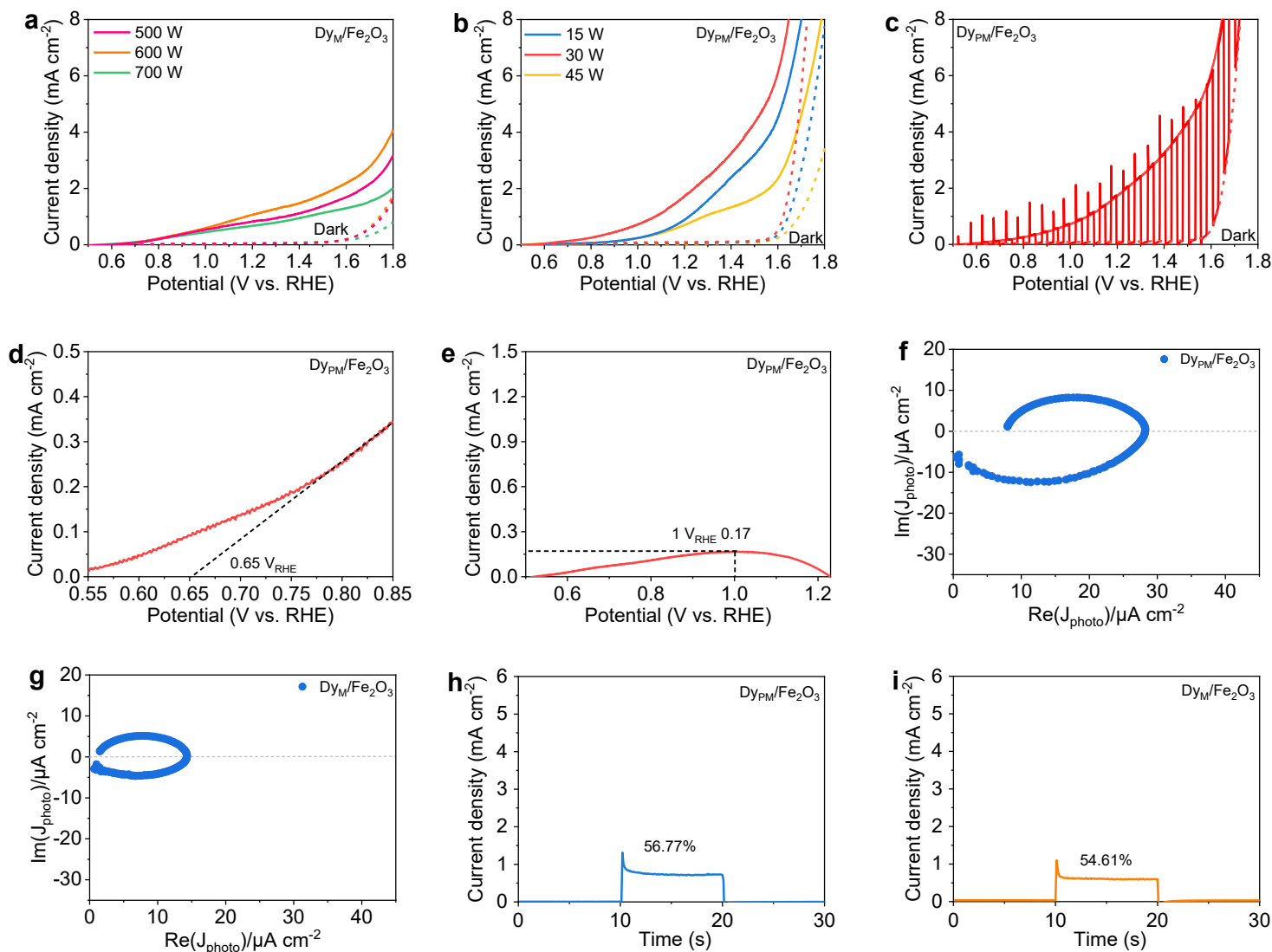
Supplementary Fig. 65 Photoelectrochemical performance measurement of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Sm -loaded Fe_2O_3 annealed directly by microwave at various power (400 W, 600 W, and 800 W) without plasma pretreatment. **b**, J - V curves of Sm single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Sm}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Sm}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Sm}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



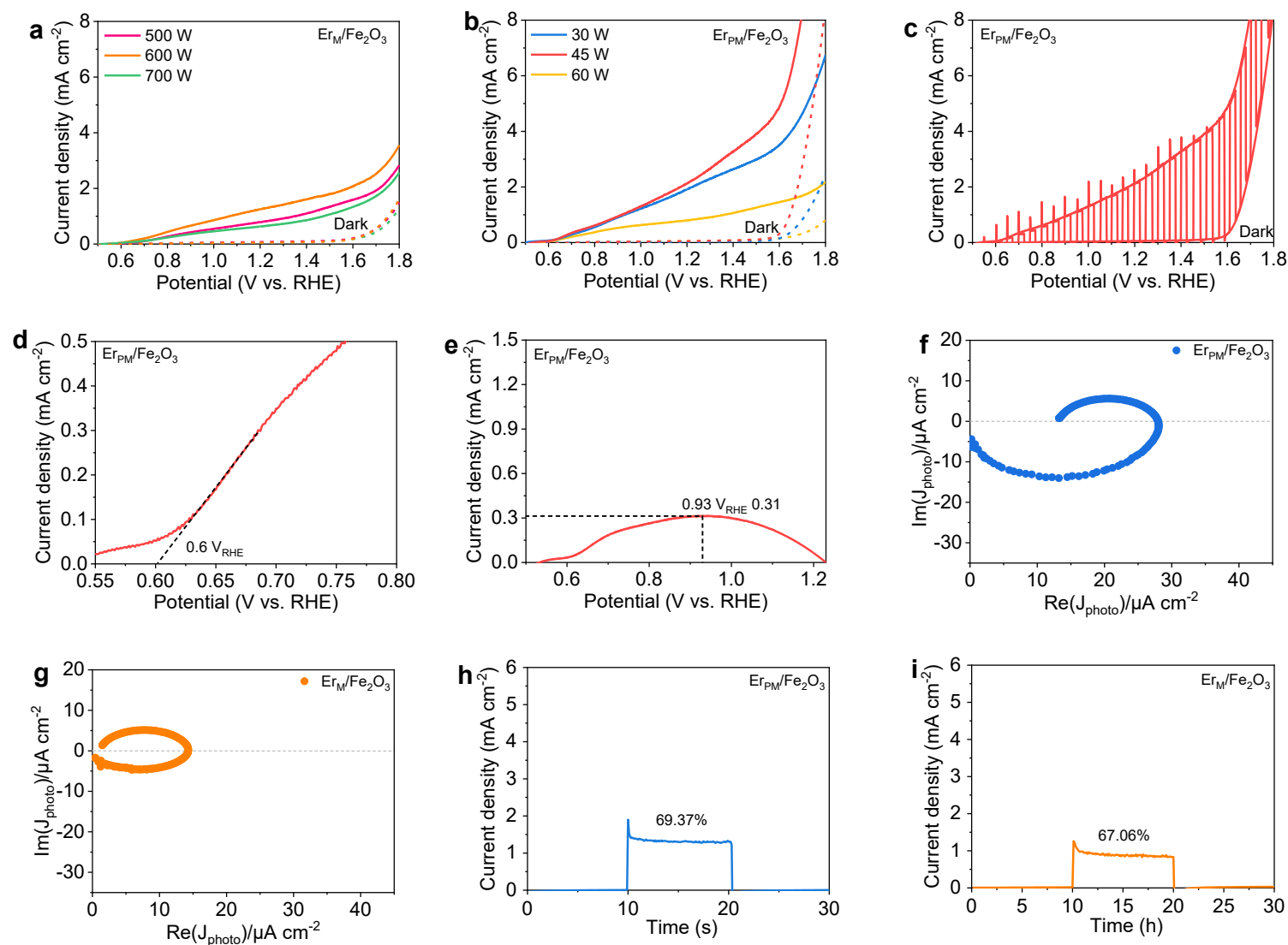
Supplementary Fig.66 Photoelectrochemical performance measurement of Gd_{PM}/Fe₂O₃. **a**, J - V curves of Gd-loaded Fe₂O₃ annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, J - V curves of Gd single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of Gd_{PM}/Fe₂O₃. **d**, Onset potential of Gd_{PM}/Fe₂O₃. **e** Applied bias photon-to-current efficiency of Gd_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Gd_{PM}/Fe₂O₃ and Gd_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Gd_{PM}/Fe₂O₃ and Gd_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



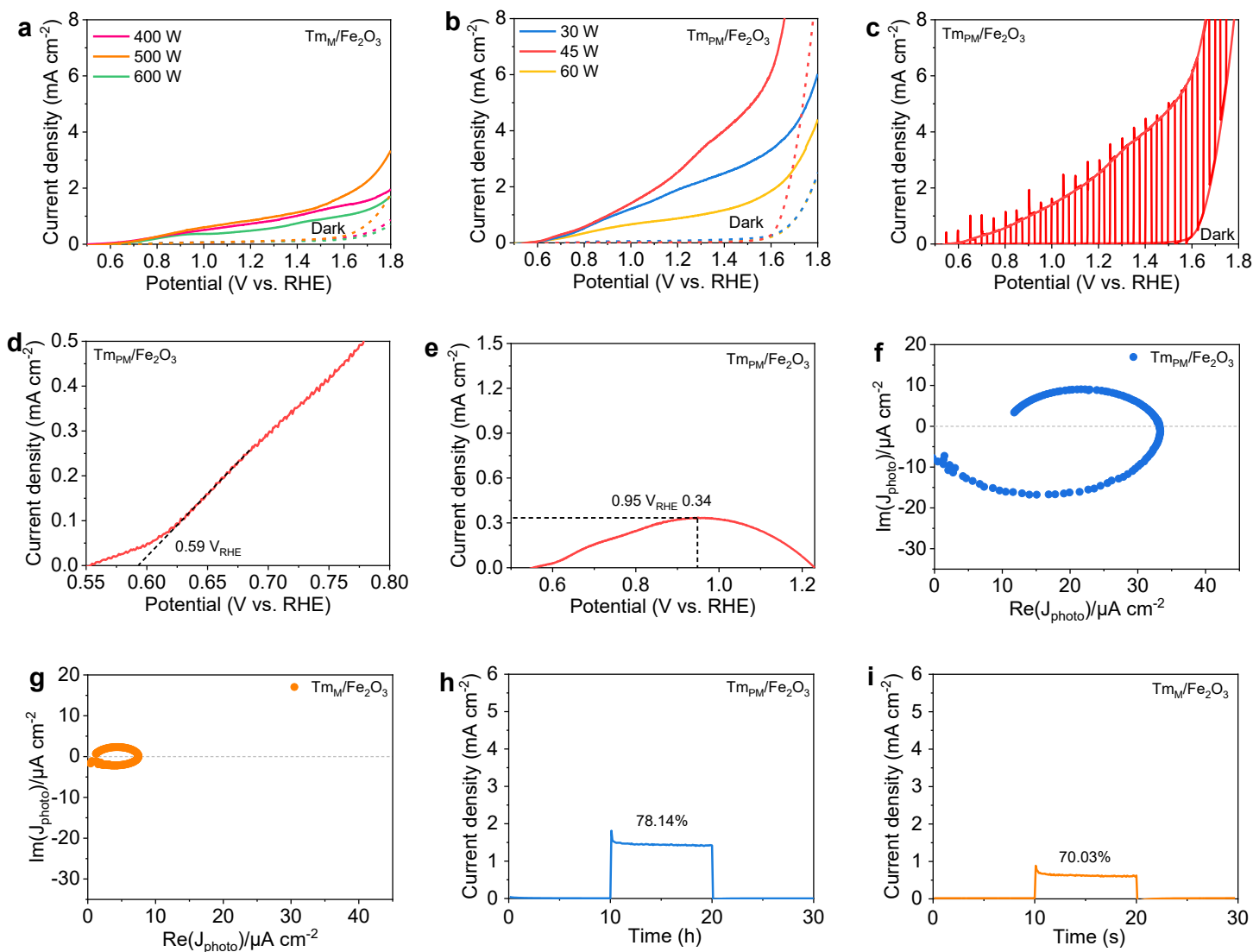
Supplementary Fig.67 Photoelectrochemical performance measurement of Tb_{PM}/Fe₂O₃. **a**, J - V curves of Td-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Td single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of Td_{PM}/Fe₂O₃. **d**, Onset potential of Td_{PM}/Fe₂O₃. **e**, Applied bias photon-to current efficiency of Td_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Td_{PM}/Fe₂O₃ and Td_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Td_{PM}/Fe₂O₃ and Td_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



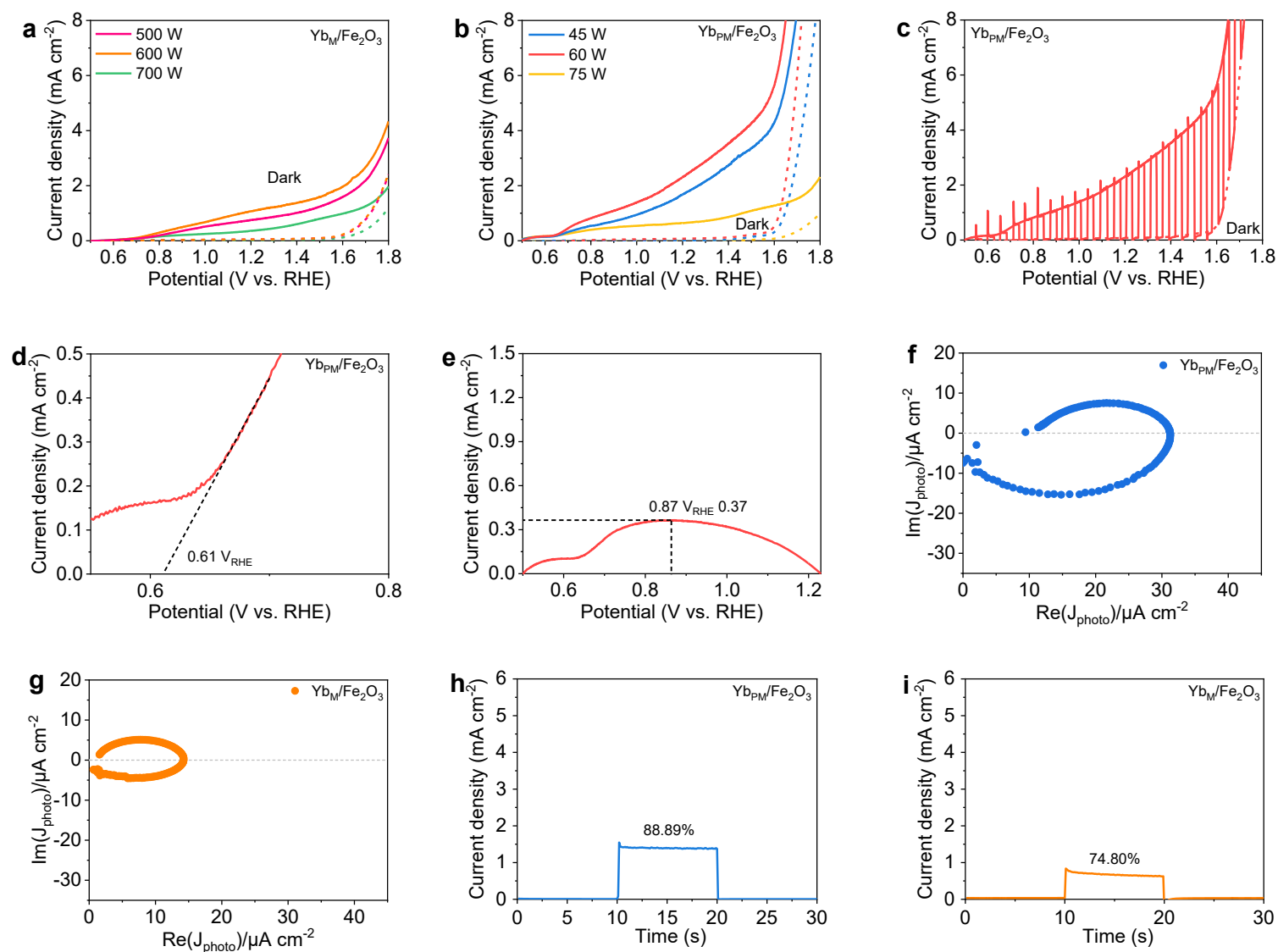
Supplementary Fig. 68 Photoelectrochemical performance measurement of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Dy-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Dy single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Dy}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Dy}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Dy}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



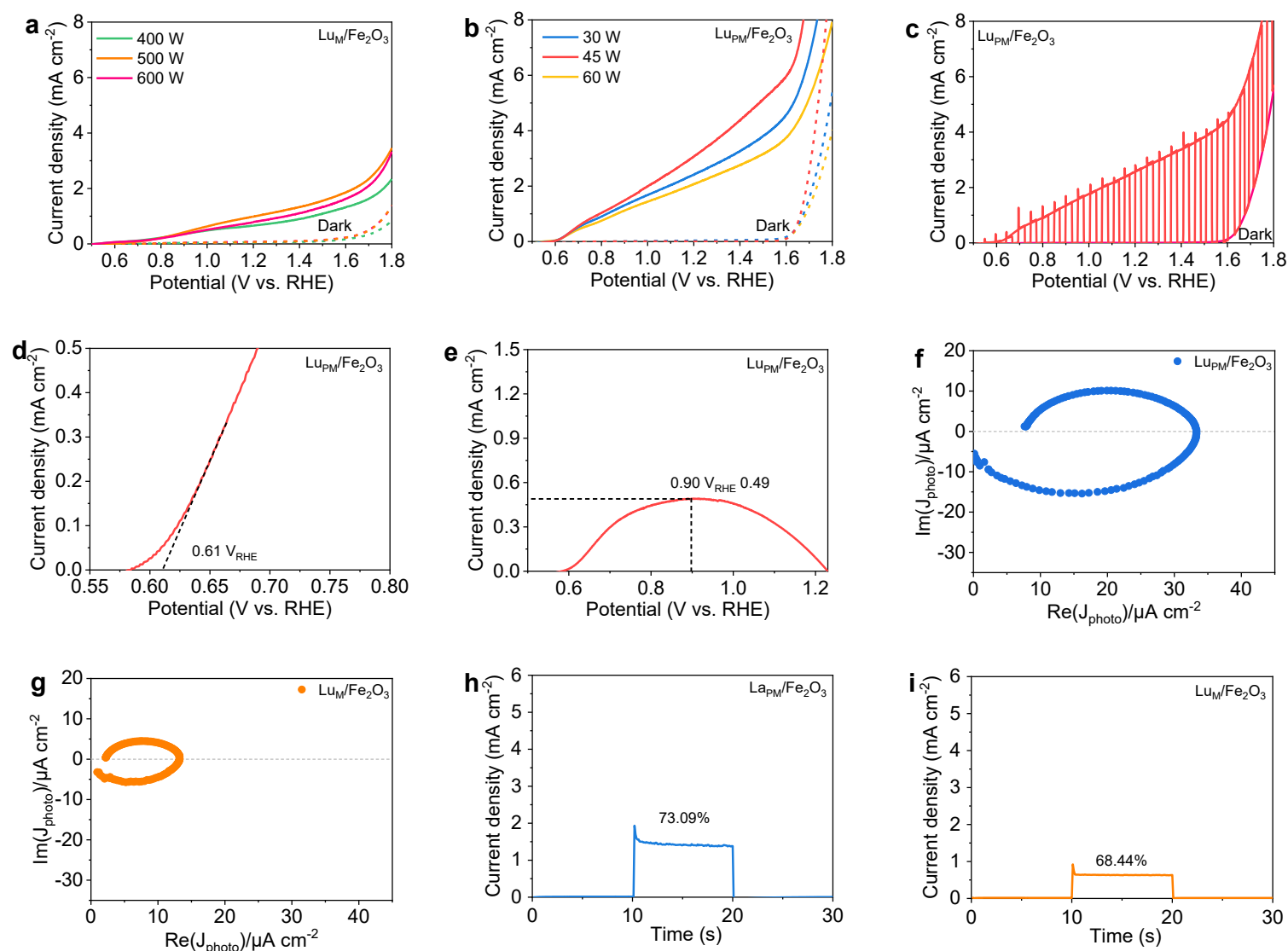
Supplementary Fig. 69 Photoelectrochemical performance measurement of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Er-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Er single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Er}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Er}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Er}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



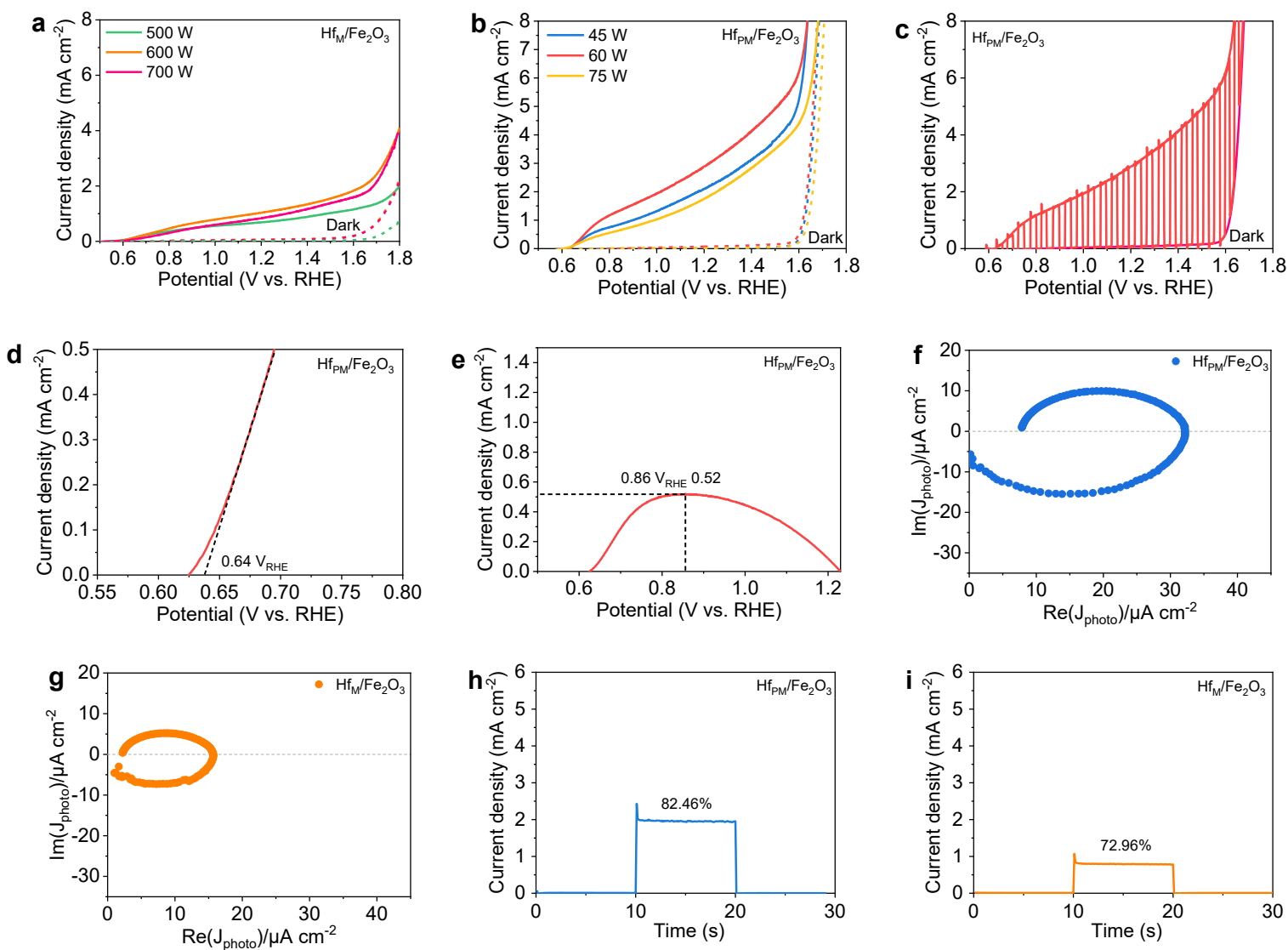
Supplementary Fig. 70 Photoelectrochemical performance measurement of Tm_{PM}/Fe₂O₃. **a**, *J-V* curves of Tm-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, *J-V* curves of Tm single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped *J-V* curve of Tm_{PM}/Fe₂O₃. **d**, Onset potential of Tm_{PM}/Fe₂O₃. **e**, Applied bias photon-to-current efficiency of Tm_{PM}/Fe₂O₃. **f, g**, IMPS responses of Tm_{PM}/Fe₂O₃ and Tm_M/Fe₂O₃ applied at 1 V_{RHE}. **h, i**, Transient photocurrent density of Tm_{PM}/Fe₂O₃ and Tm_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



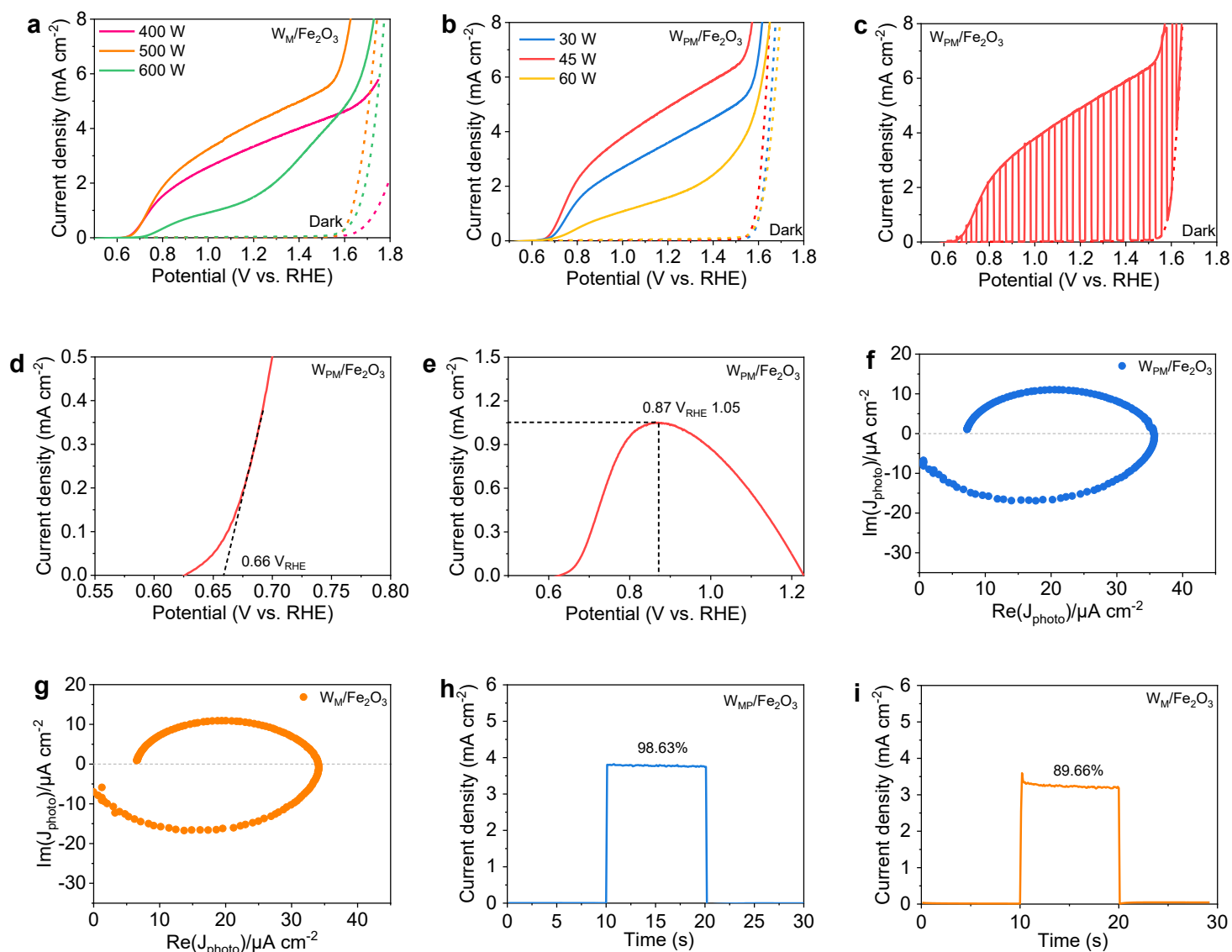
Supplementary Fig. 71 Photoelectrochemical performance measurement of $\text{Yb}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Y-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Y single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Y}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Y}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Y}_\text{PM}/\text{Fe}_2\text{O}_3$. **f, g**, IMPS responses of $\text{Y}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Y}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h, i**, Transient photocurrent density of $\text{Y}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Y}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



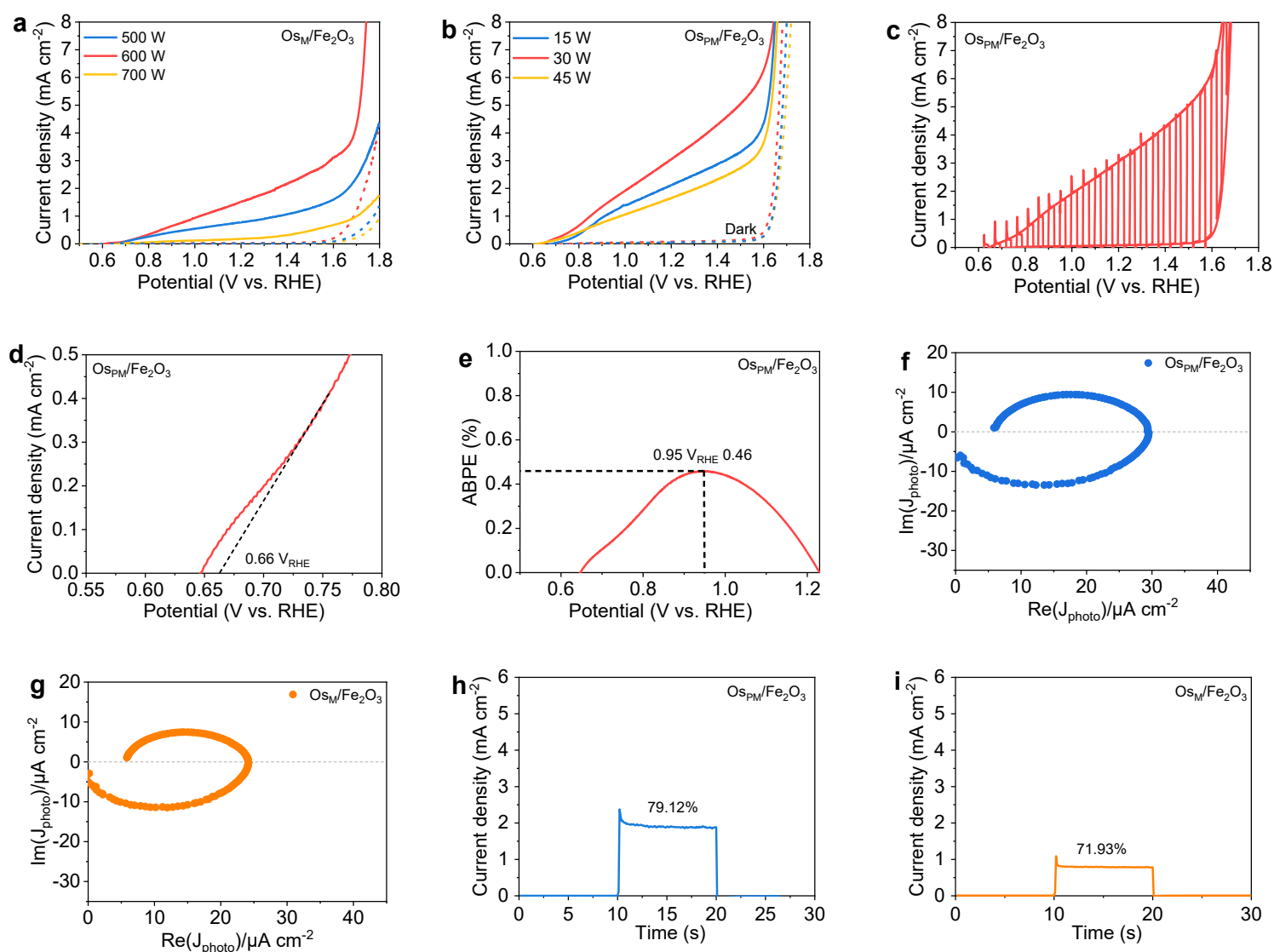
Supplementary Fig. 72 Photoelectrochemical performance measurement of Lu_{PM}/Fe₂O₃. **a**, J - V curves of Lu-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Lu single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of Lu_{PM}/Fe₂O₃. **d**, Onset potential of Lu_{PM}/Fe₂O₃. **e** Applied bias photon-to-current efficiency of Lu_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Lu_{PM}/Fe₂O₃ and Lu_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Lu_{PM}/Fe₂O₃ and Lu_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



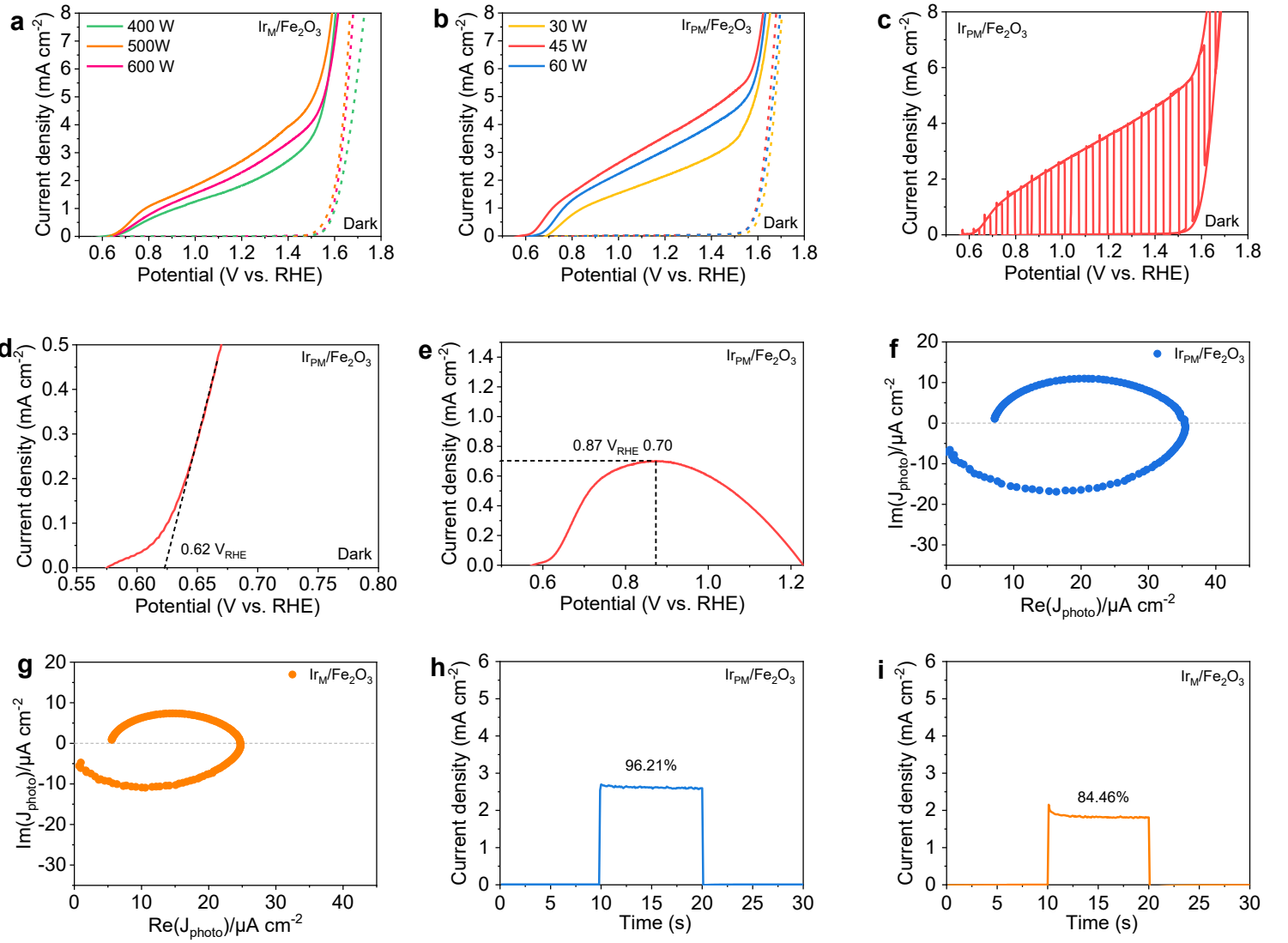
Supplementary Fig. 73 Photoelectrochemical performance measurement of Hf_{PM}/Fe₂O₃. **a**, J - V curves of Hf-loaded Fe₂O₃ annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Hf single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of Hf_{PM}/Fe₂O₃. **d**, Onset potential of Hf_{PM}/Fe₂O₃. **e**, Applied bias photon-to-current efficiency of Hf_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Hf_{PM}/Fe₂O₃ and Hf_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Hf_{PM}/Fe₂O₃ and Hf_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



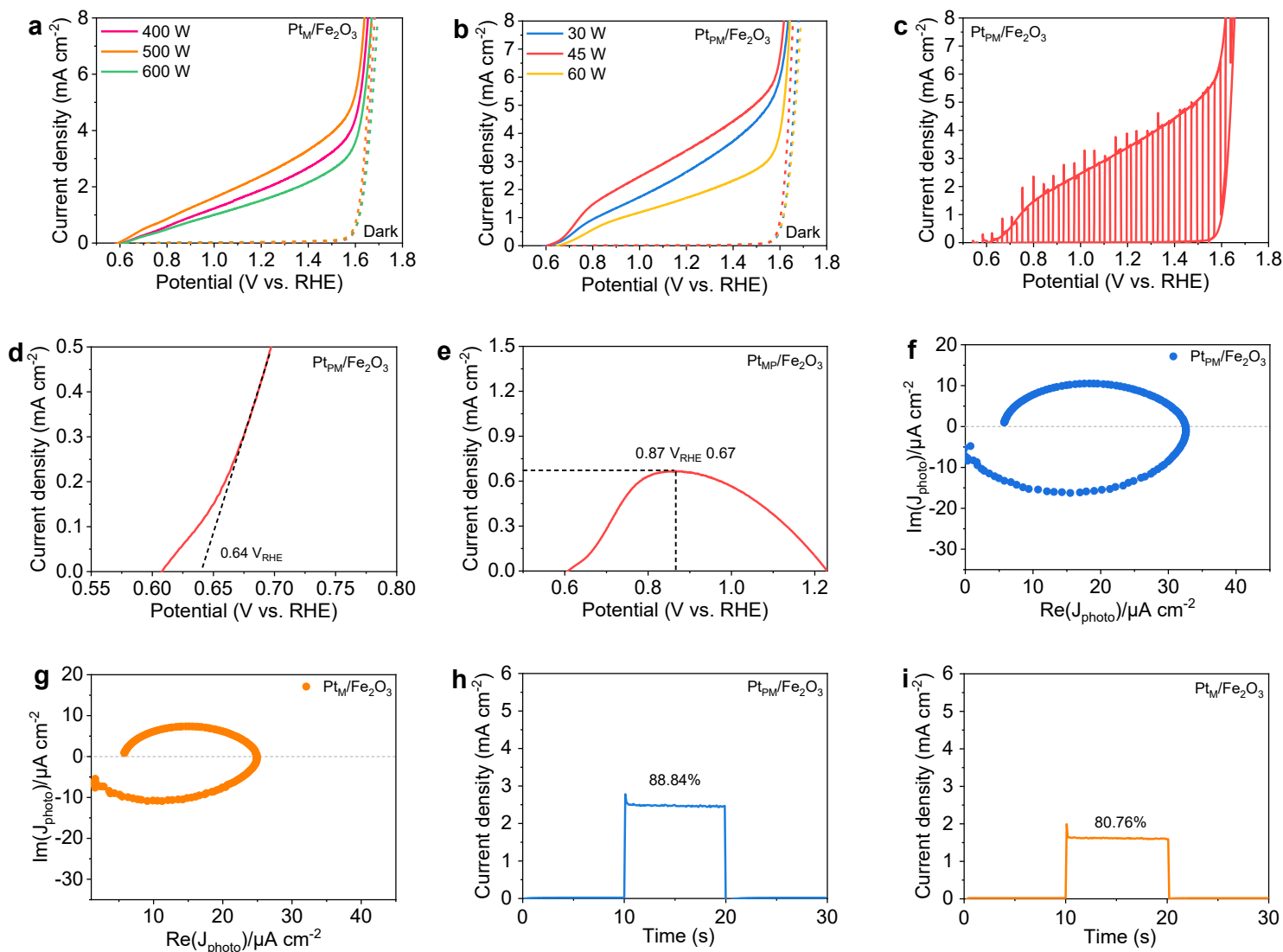
Supplementary Fig. 74 Photoelectrochemical performance measurement of W_{PM}/Fe_2O_3 . **a**, J - V curves of W -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of W single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of W_{PM}/Fe_2O_3 . **d**, Onset potential of W_{PM}/Fe_2O_3 . **e**, Applied bias photon-to-current efficiency of W_{PM}/Fe_2O_3 . **f**, **g**, IMPS responses of W_{PM}/Fe_2O_3 and W_M/Fe_2O_3 applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of W_{PM}/Fe_2O_3 and W_M/Fe_2O_3 . All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



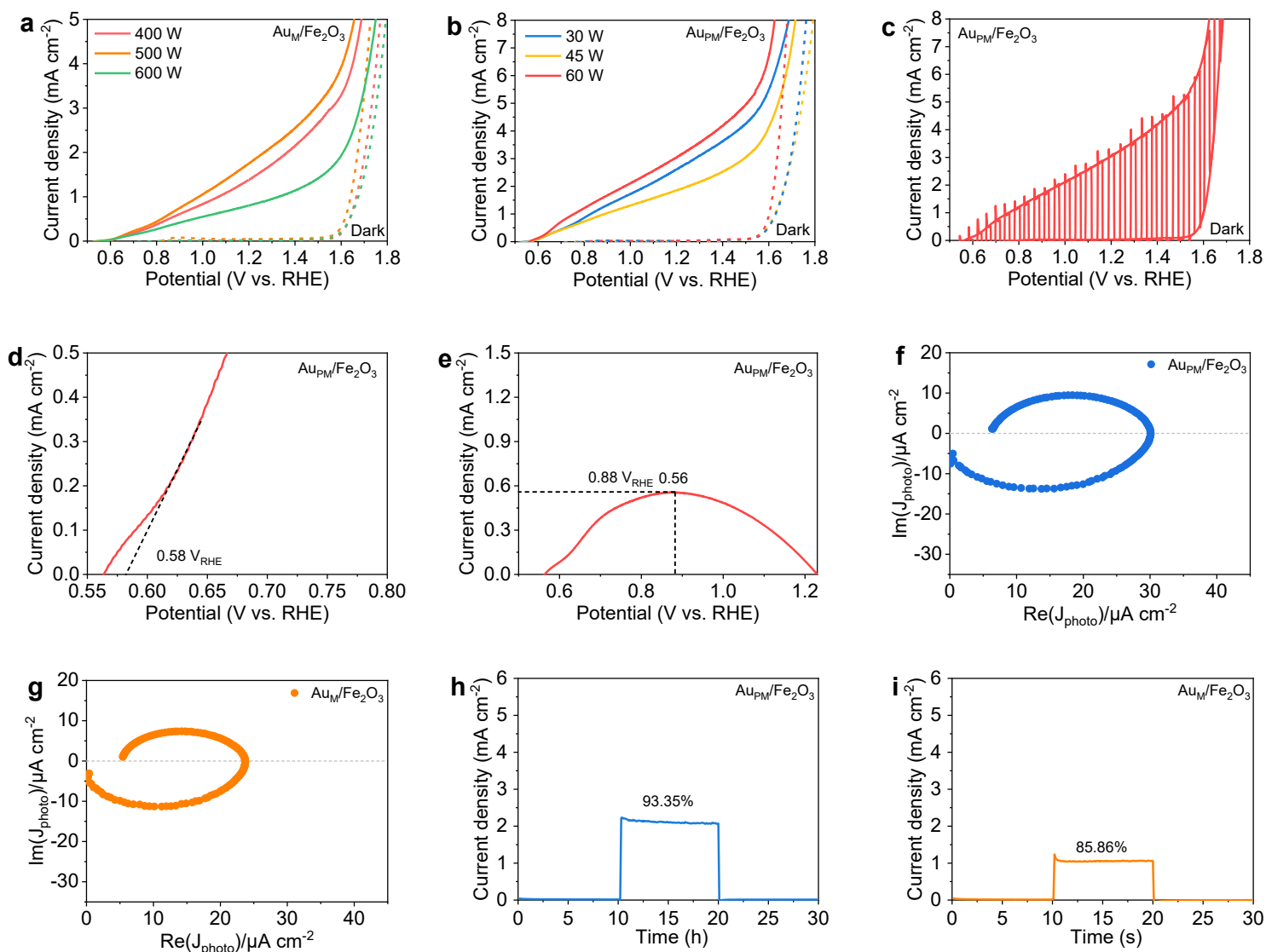
Supplementary Fig. 75 Photoelectrochemical performance measurement of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Os-loaded Fe_2O_3 annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, J - V curves of Os single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **c**, Chopped J - V curve of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Os}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Os}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Os}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



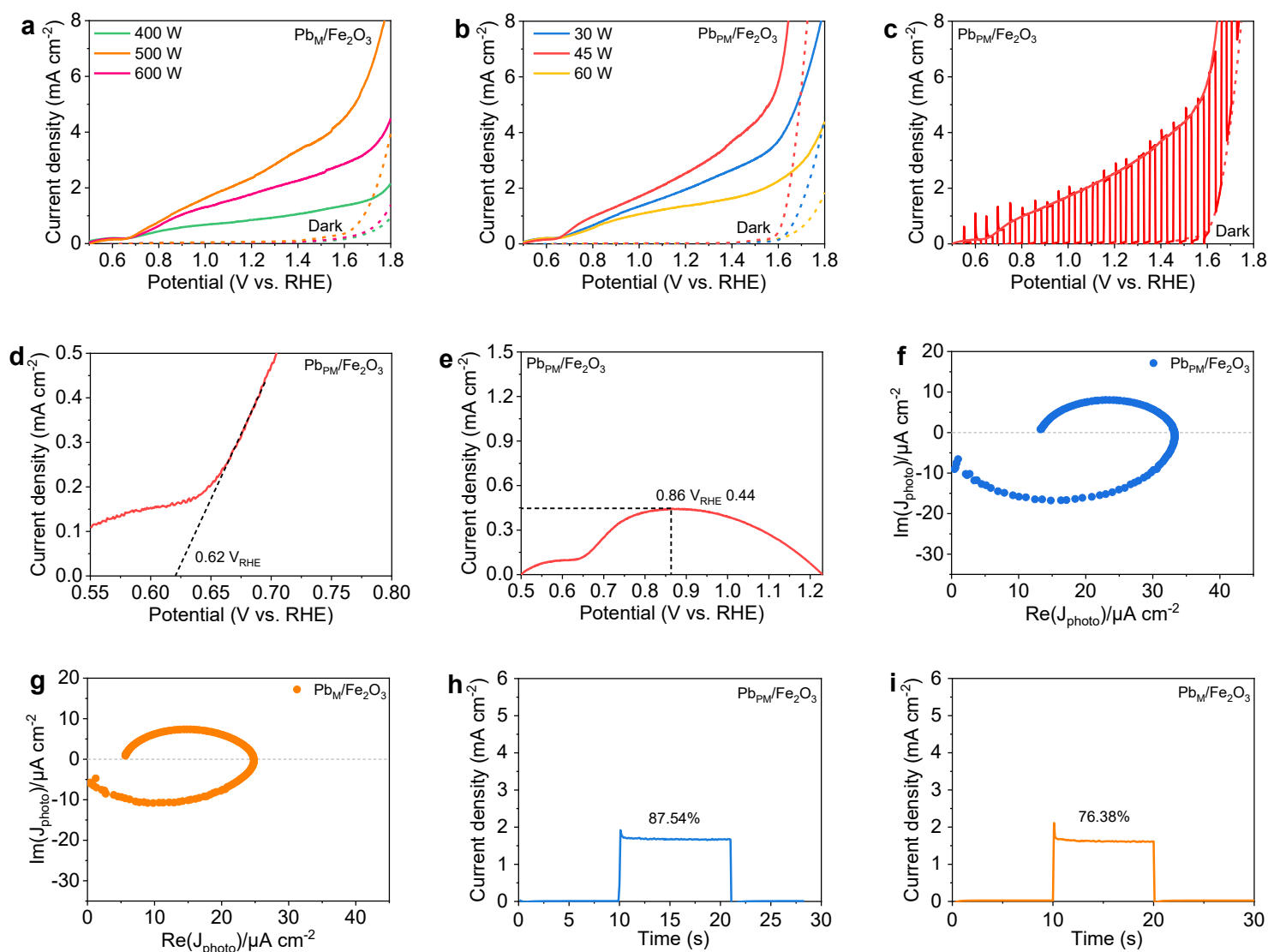
Supplementary Fig. 76 Photoelectrochemical performance measurement of Ir_{PM}/Fe₂O₃. **a**, J - V curves of Ir-loaded Fe₂O₃ annealed directly by microwave at various power (400 W, 600 W, and 800 W) without plasma pretreatment. **b**, J - V curves of Ir single atoms anchored on Fe₂O₃ via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of Ir_{PM}/Fe₂O₃. **d**, Onset potential of Ir_{PM}/Fe₂O₃. **e**, Applied bias photon-to-current efficiency of Ir_{PM}/Fe₂O₃. **f**, **g**, IMPS responses of Ir_{PM}/Fe₂O₃ and Ir_M/Fe₂O₃ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of Ir_{PM}/Fe₂O₃ and Ir_M/Fe₂O₃. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



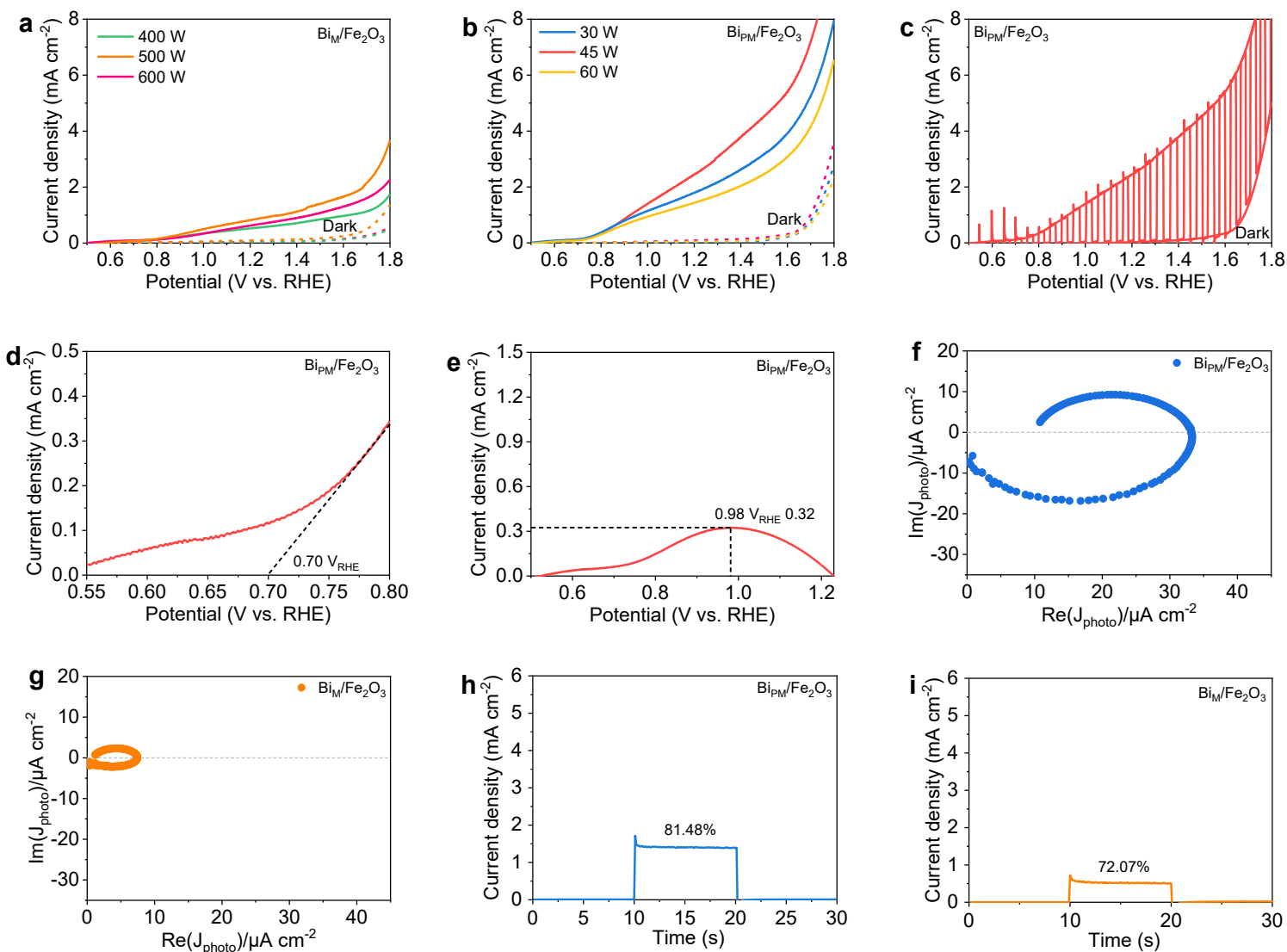
Supplementary Fig. 77 Photoelectrochemical performance measurement of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Pt-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Pt single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pt}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Pt}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pt}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



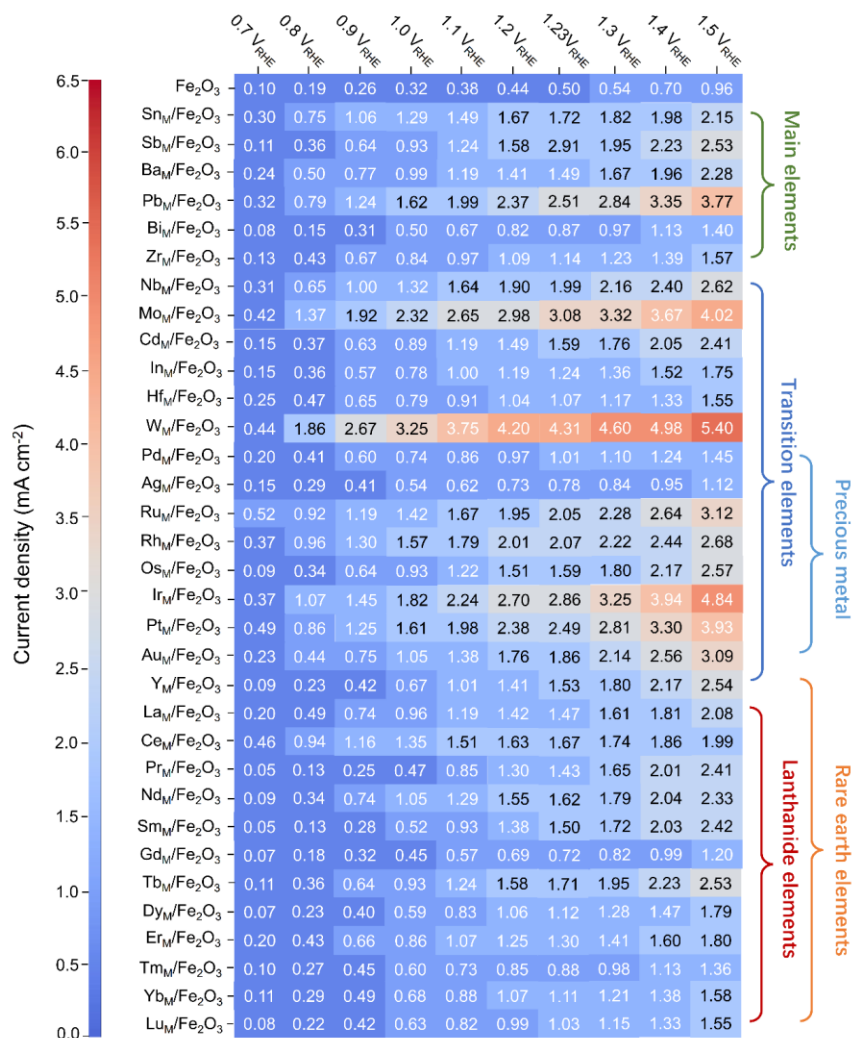
Supplementary Fig. 78 Photoelectrochemical performance measurement of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Au-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Au single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Au}_{\text{M}}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE} . **h**, **i**, Transient photocurrent density of $\text{Au}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Au}_{\text{M}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



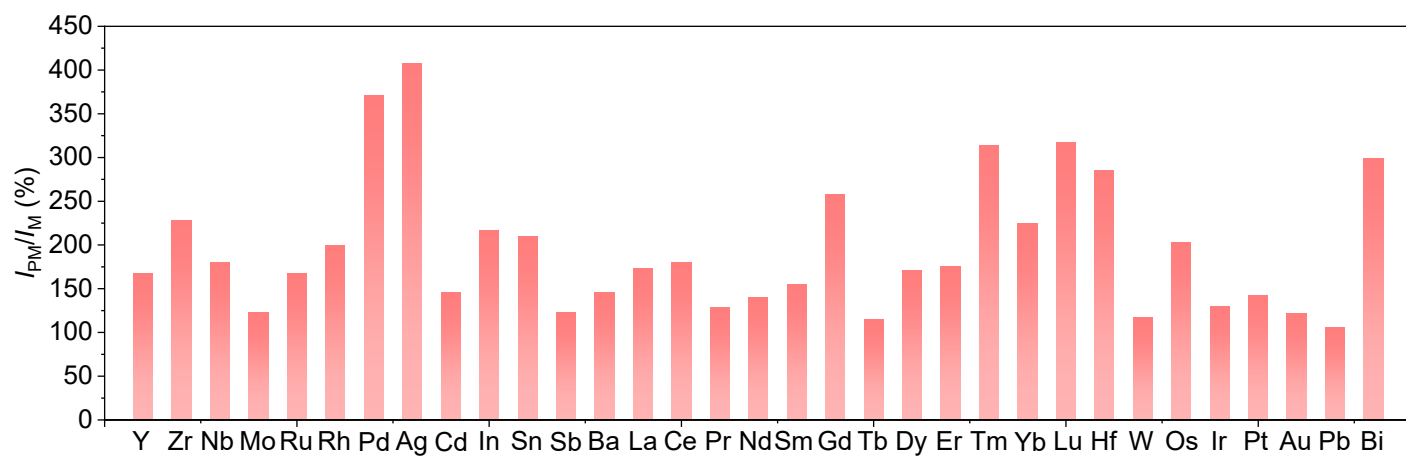
Supplementary Fig. 79 Photoelectrochemical performance measurement of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Pb -loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Au single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pb}_M/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Pb}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and $\text{Pb}_M/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



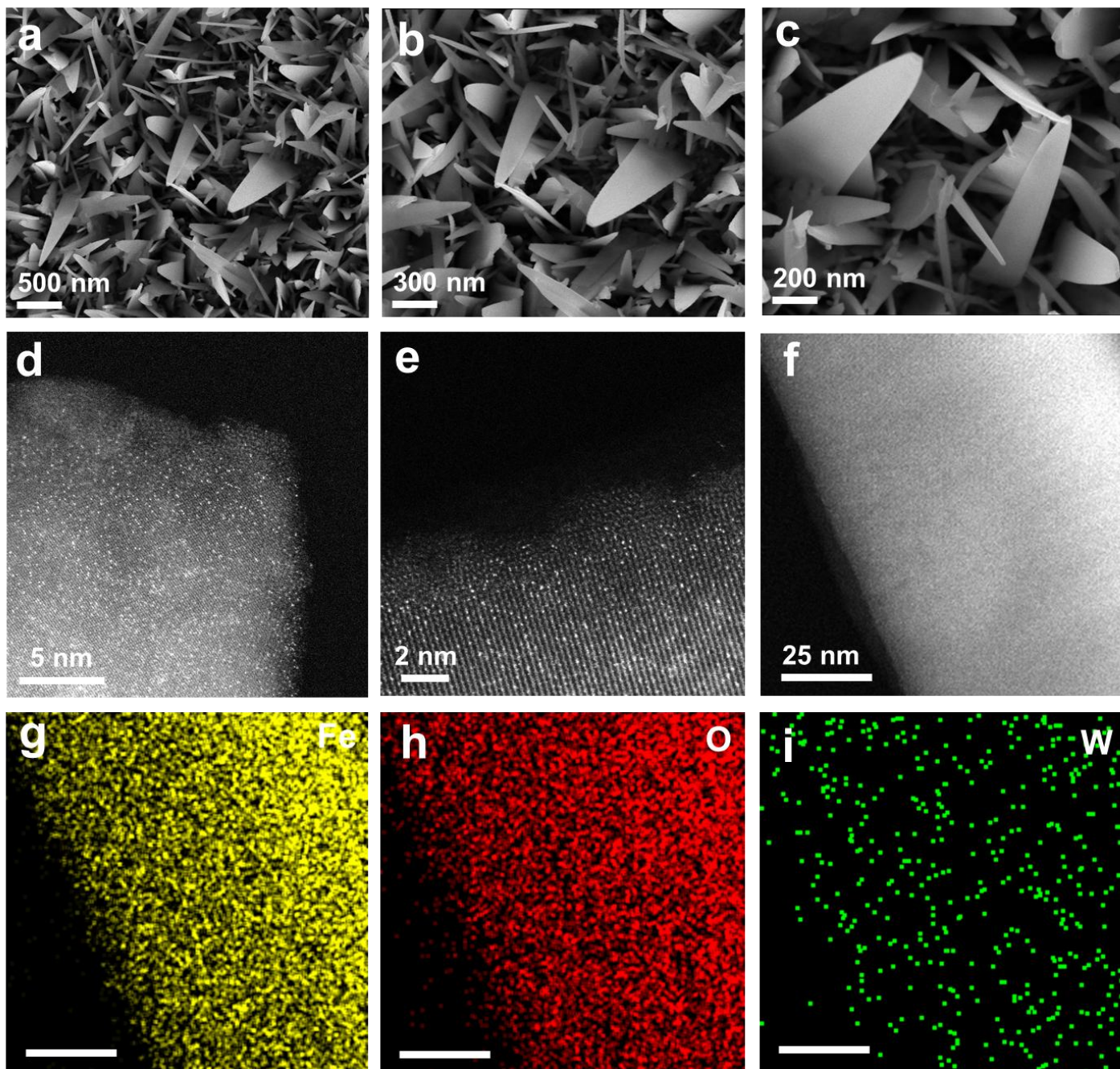
Supplementary Fig. 80 Photoelectrochemical performance measurement of $\text{Nb}_\text{PM}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Bi-loaded Fe_2O_3 annealed directly by microwave at various power (600 W, 800 W, and 1000 W) without plasma pretreatment. **b**, J - V curves of Bi single atoms anchored on Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **c**, Chopped J - V curve of $\text{Bi}_\text{PM}/\text{Fe}_2\text{O}_3$. **d**, Onset potential of $\text{Bi}_\text{PM}/\text{Fe}_2\text{O}_3$. **e**, Applied bias photon-to-current efficiency of $\text{Bi}_\text{PM}/\text{Fe}_2\text{O}_3$. **f**, **g**, IMPS responses of $\text{Bi}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Bi}_\text{M}/\text{Fe}_2\text{O}_3$ applied at 1 V_{RHE}. **h**, **i**, Transient photocurrent density of $\text{Bi}_\text{PM}/\text{Fe}_2\text{O}_3$ and $\text{Bi}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



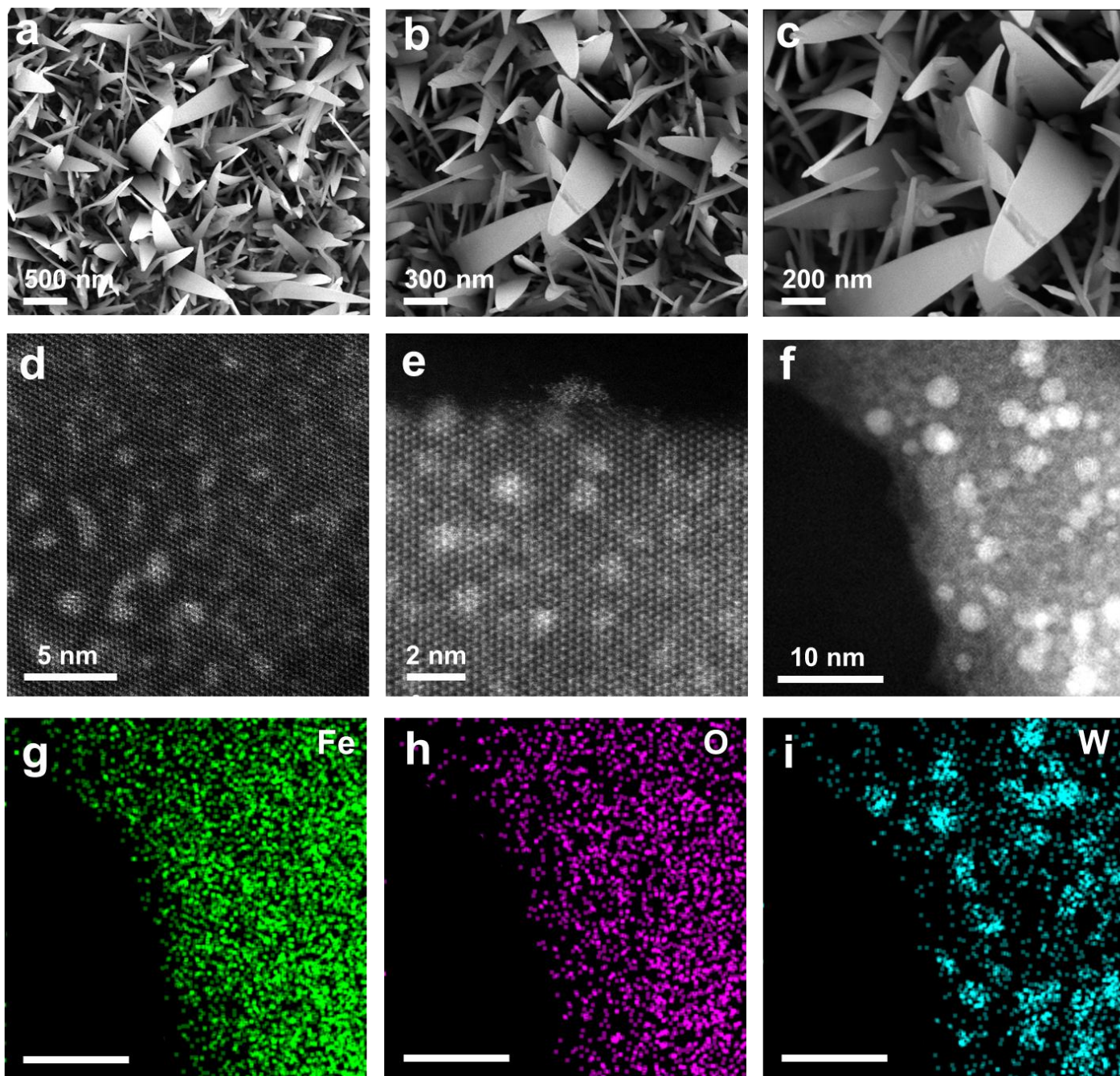
Supplementary Fig.81 The photoelectrochemical performance of 33 photoelectrodes without plasma pretreatment. OER activities of Fe₂O₃ loaded with 33 metallic elements without plasma pretreatment.



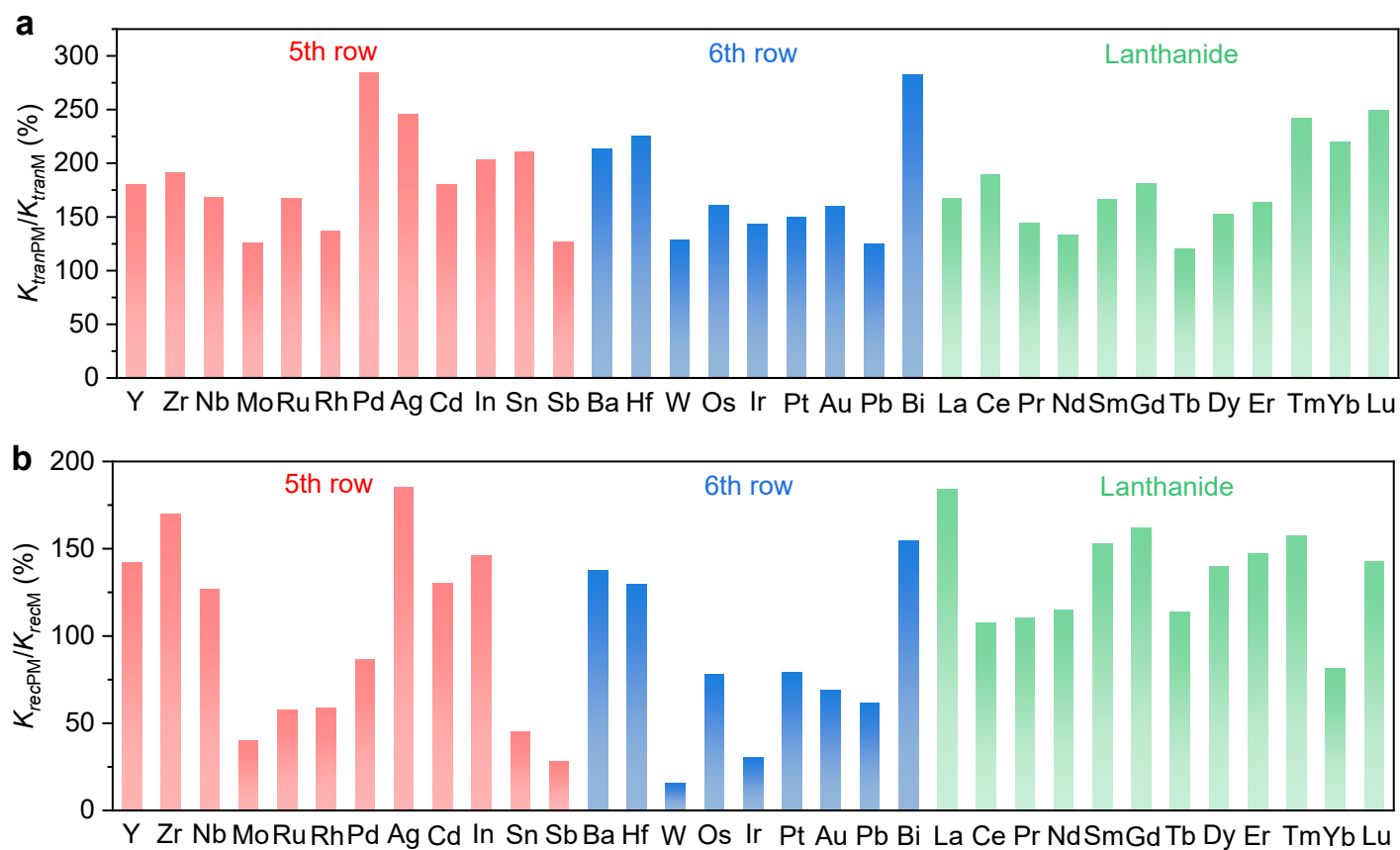
Supplementary Fig.82 Photocurrent density ratio. Photoelectrochemical current density comparison between plasma-microwave and microwave-only methods.



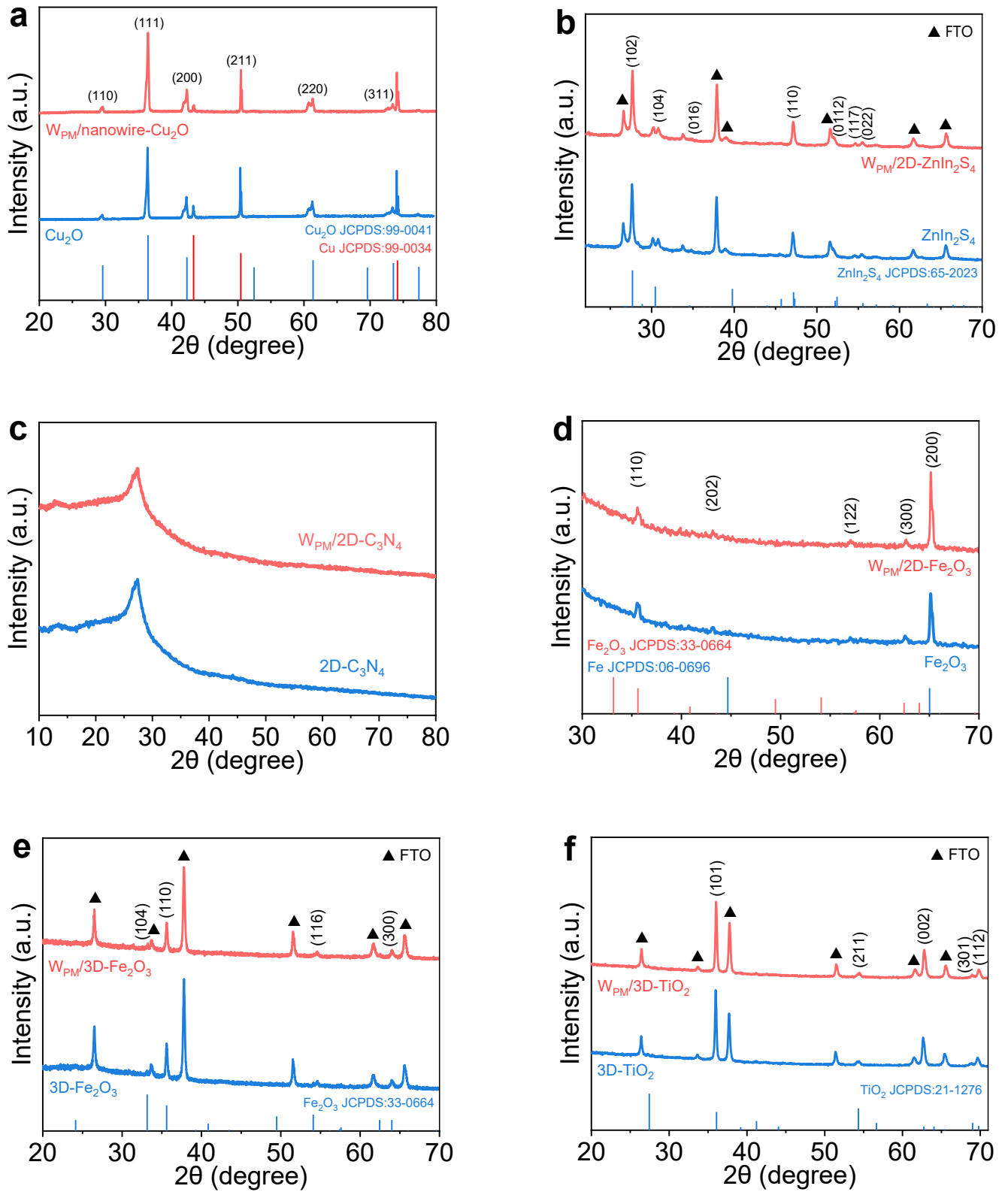
Supplementary Fig. 83 SEM, HAADF-STEM and EDS elemental mappings of W_{PM}/Fe_2O_3 . **a-c**, SEM images of W_{PM}/Fe_2O_3 . **d, e**, High resolution HAADF-STEM images of W_{PM}/Fe_2O_3 . **f-i**, elemental mappings of W_{PM}/Fe_2O_3 .



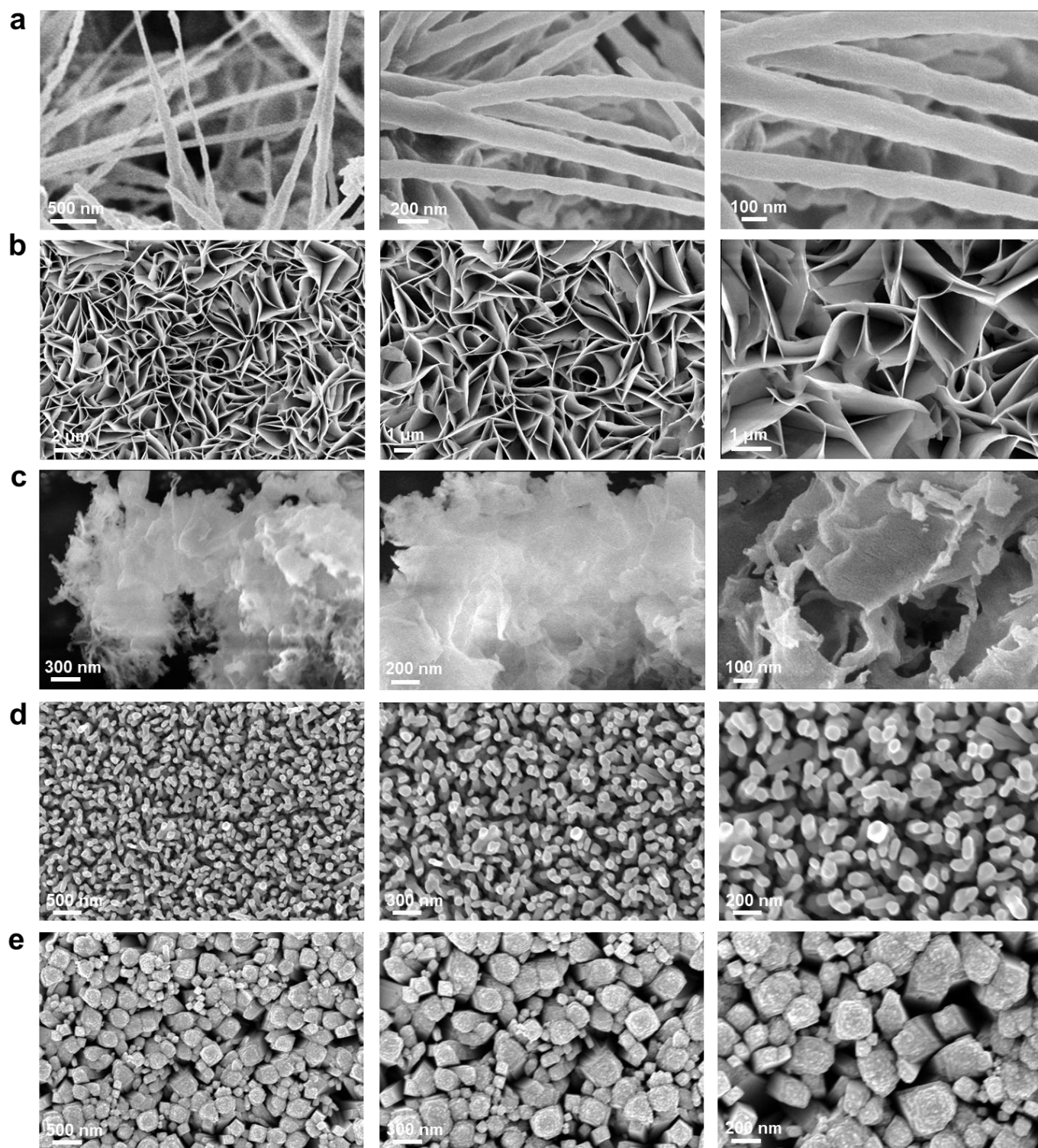
Supplementary Fig. 84 SEM, HAADF-STEM and EDS elemental mappings of W_M/Fe_2O_3 . **a-c**, SEM images of W_M/Fe_2O_3 . **d, e**, High resolution HAADF-STEM images of W_M/Fe_2O_3 . **f**, elemental mappings of W_M/Fe_2O_3 .



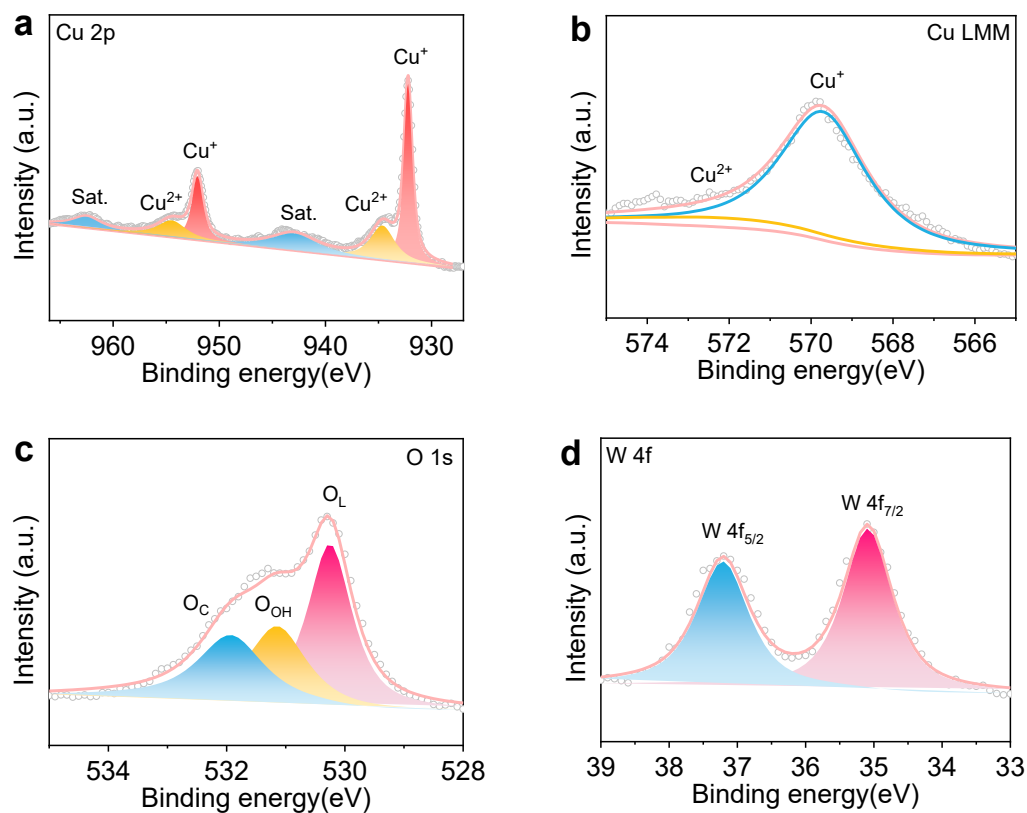
Supplementary Fig. 85 Charge transfer and recombination rate ratios. The charge transfer and recombination rate ratios of plasma-treated and untreated Fe_2O_3 loaded with 33 various metals.



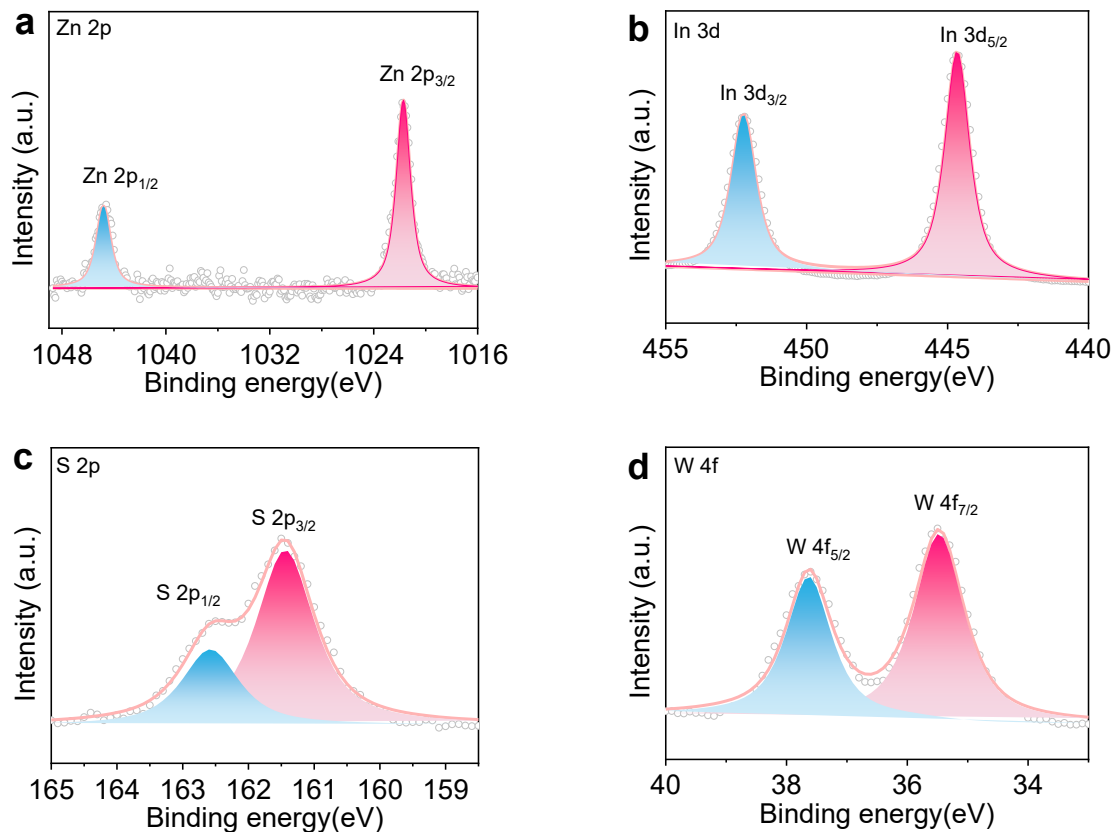
Supplementary Fig. 86 X-ray diffraction (XRD) patterns. a-f, XRD patterns of 1D- Cu_2O , $W_{PM}/1D-Cu_2O$ (a), 2D- $ZnIn_2S_4$, $W_{PM}/2D-2D-ZnIn_2S_4$ (b), 2D- C_3N_4 , $W_{PM}/2D-C_3N_4$ (c), 2D- Fe_2O_3 , $W_{PM}/2D-Fe_2O_3$ (d), 3D- Fe_2O_3 , $W_{PM}/3D-Fe_2O_3$ (e), 3D- TiO_2 and $W_{PM}/3D-TiO_2$ (f).



Supplementary Fig. 87 SEM observation. SEM images of 1D-Cu₂O (a), 2D-ZnIn₂S₄ (b), 2D-C₃N₄ (c), 3D-Fe₂O₃ (d), and 3D-TiO₂ (e).

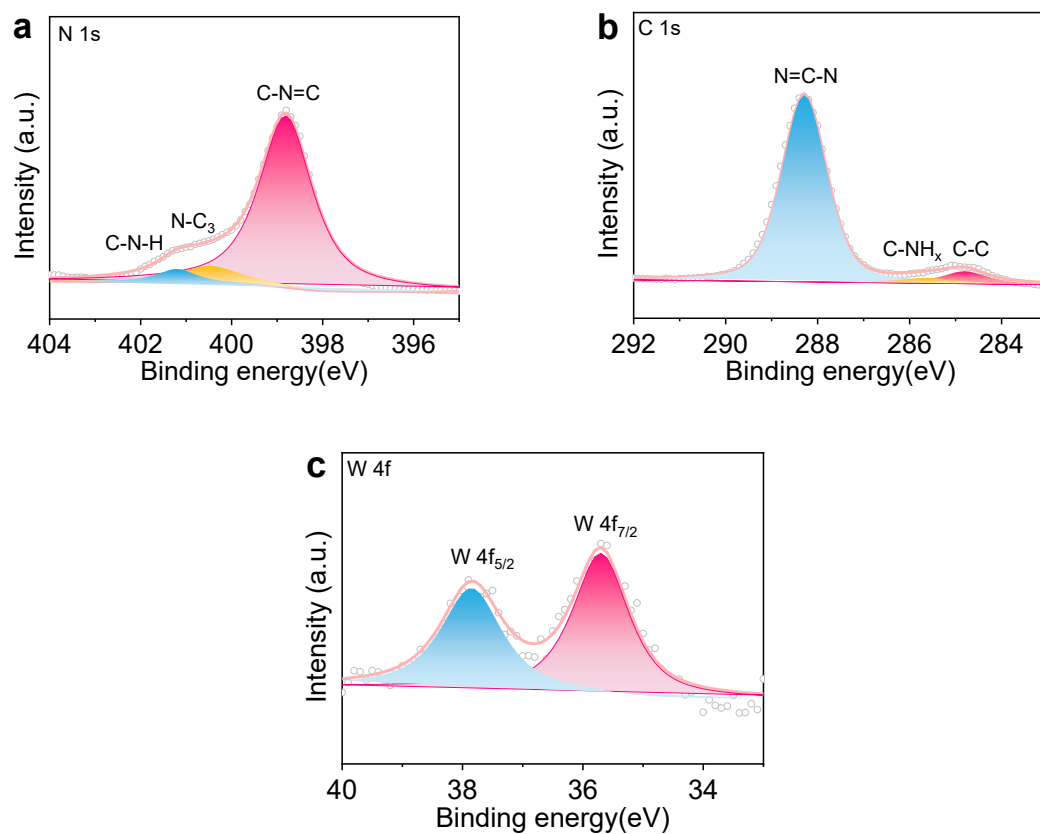


Supplementary Fig. 88 X-ray photoelectron spectroscopy (XPS) characterization of W_{PM}/nanowire-Cu₂O. **a**, Cu 2p, **b**, Cu LMM, **c**, O 1s, and **d**, W 4f XPS spectra for W_{PM}/1D-Cu₂O.

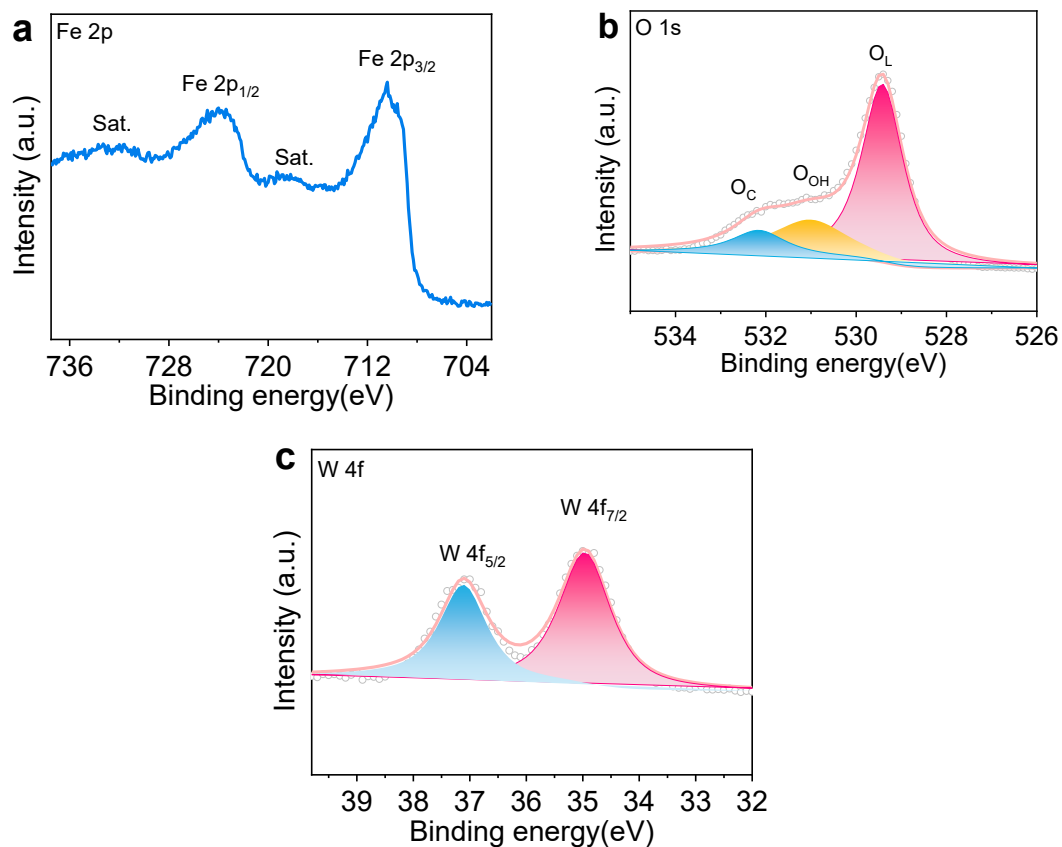


Supplementary Fig. 89 X-ray photoelectron spectroscopy (XPS) characterization of W_{PM}/2D-ZnIn₂S₄.

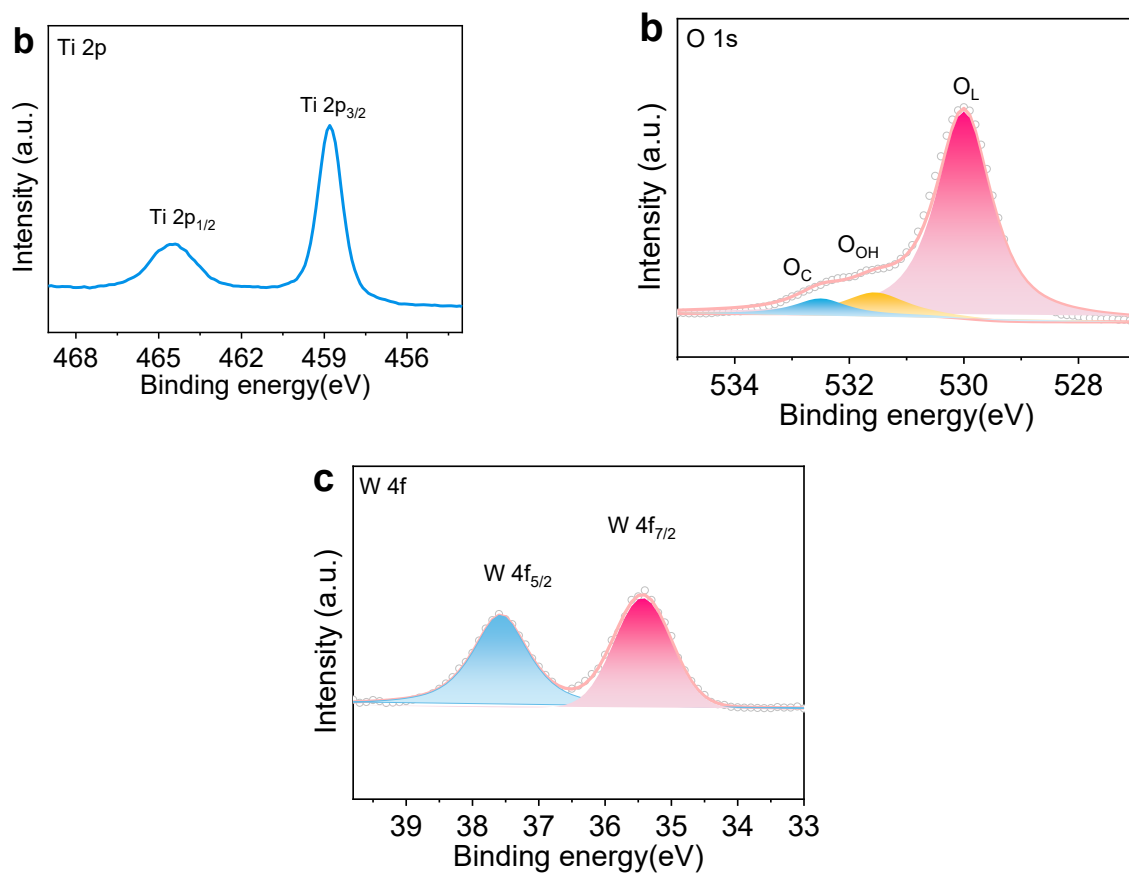
a, Zn 2p, b, In 3d, c, S 2p, and d, W 4f XPS spectra for W_{PM}/2D-ZnIn₂S₄.



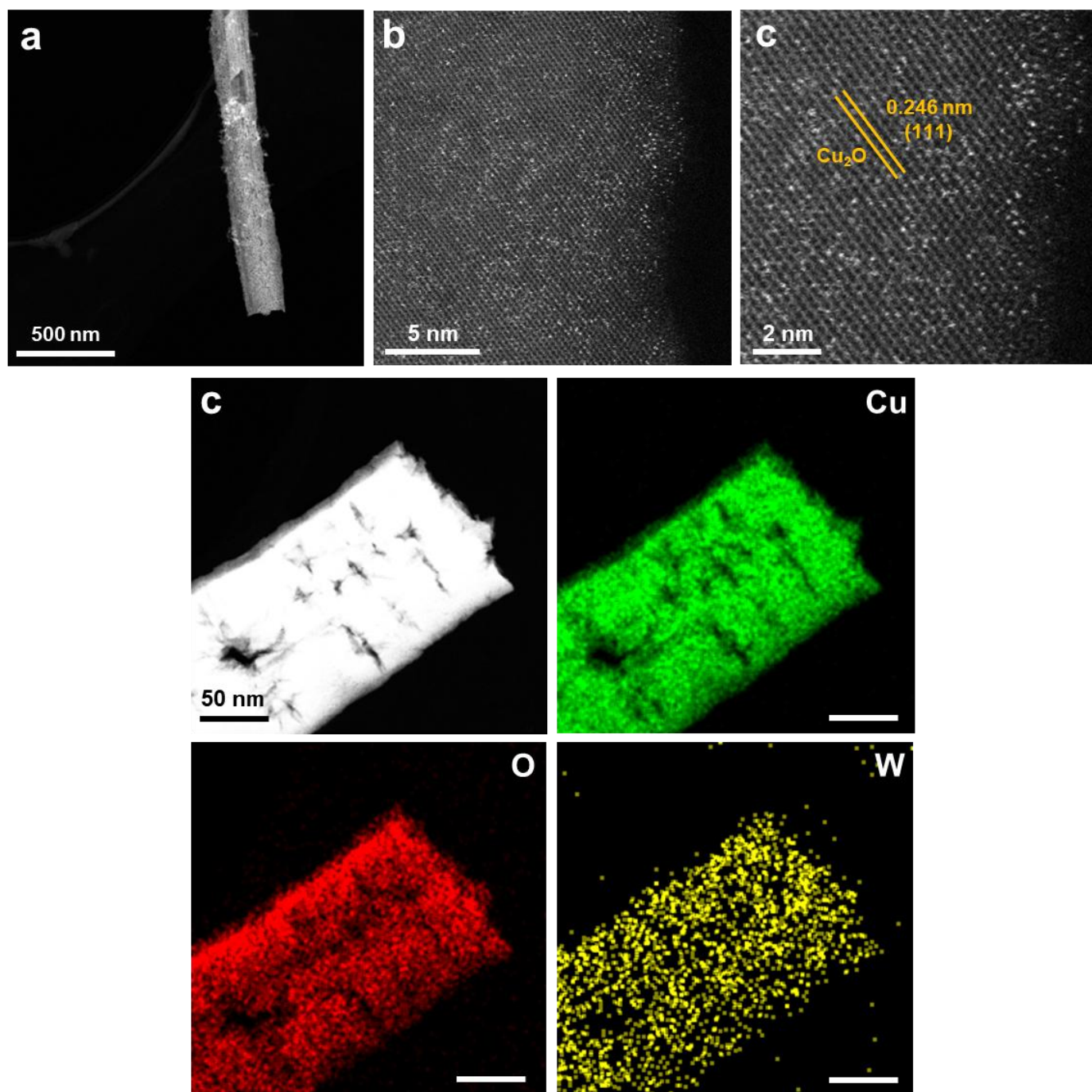
Supplementary Fig. 90 X-ray photoelectron spectroscopy (XPS) characterization of $W_{PM}/2D-C_3N_4$. a, N 1s b, C 1s, and c, W 4f XPS spectra for $W_{PM}/2D-C_3N_4$.



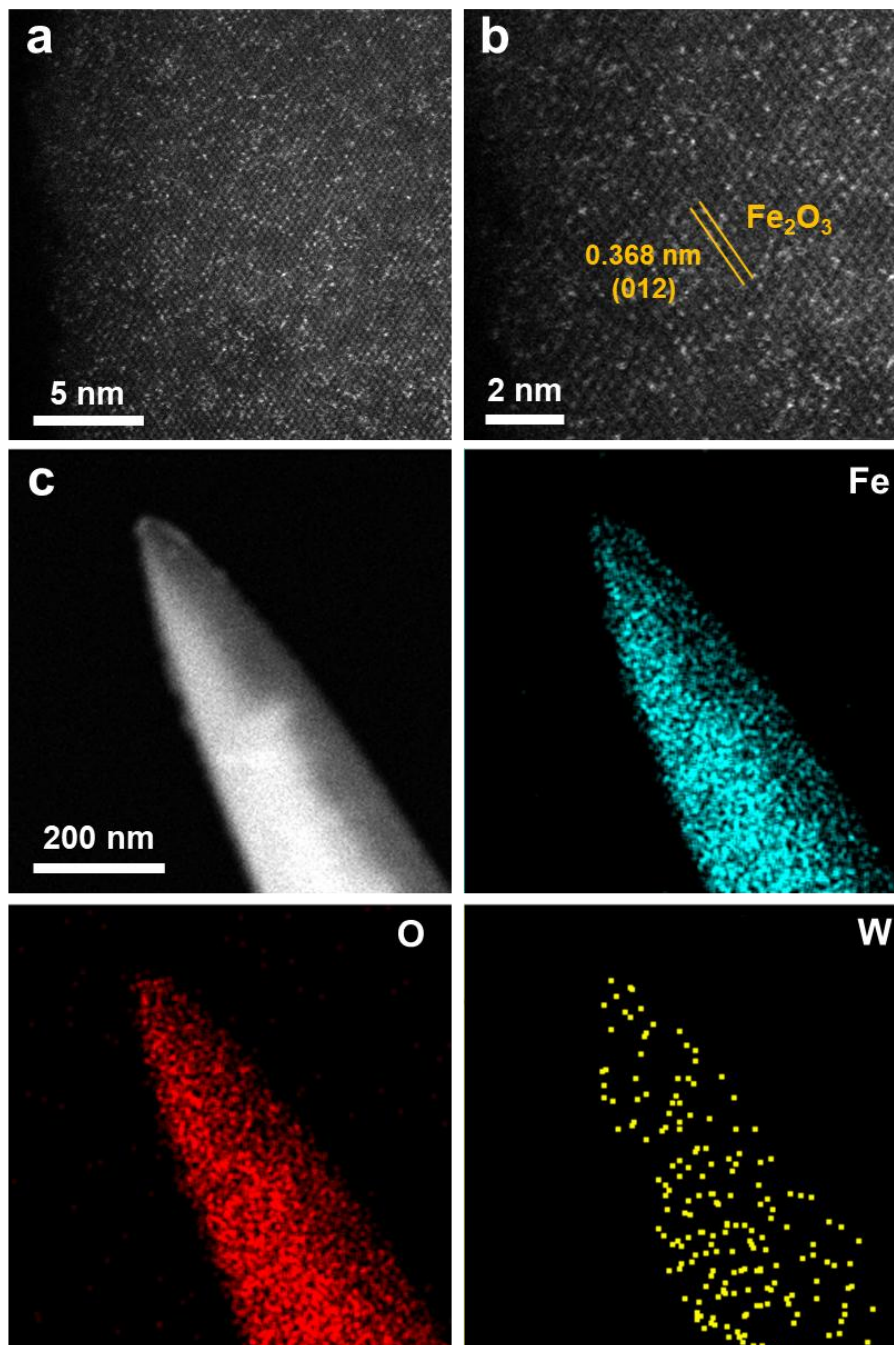
Supplementary Fig. 91 X-ray photoelectron spectroscopy (XPS) characterization of $W_{PM}/3D-Fe_2O_3$. a, Fe 2p, b, O 1s, and c, W 4f XPS spectra for $W_{PM}/3D-Fe_2O_3$.



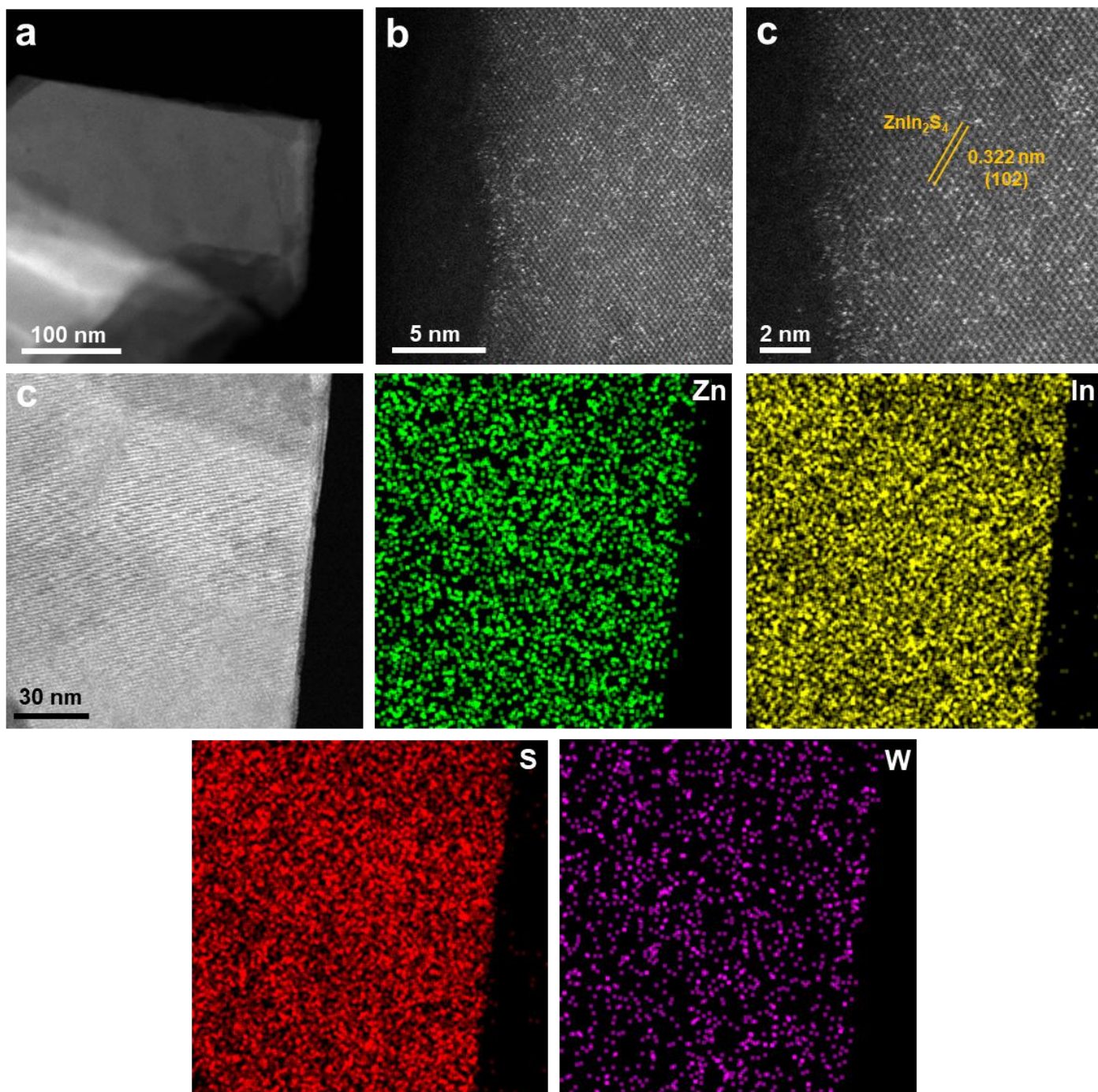
Supplementary Fig. 92 X-ray photoelectron spectroscopy (XPS) characterization of W_{PM}/3D-TiO₂. a, Ti 2p, b, O 1s, and c, W 4f XPS spectra for W_{PM}/3D-TiO₂.



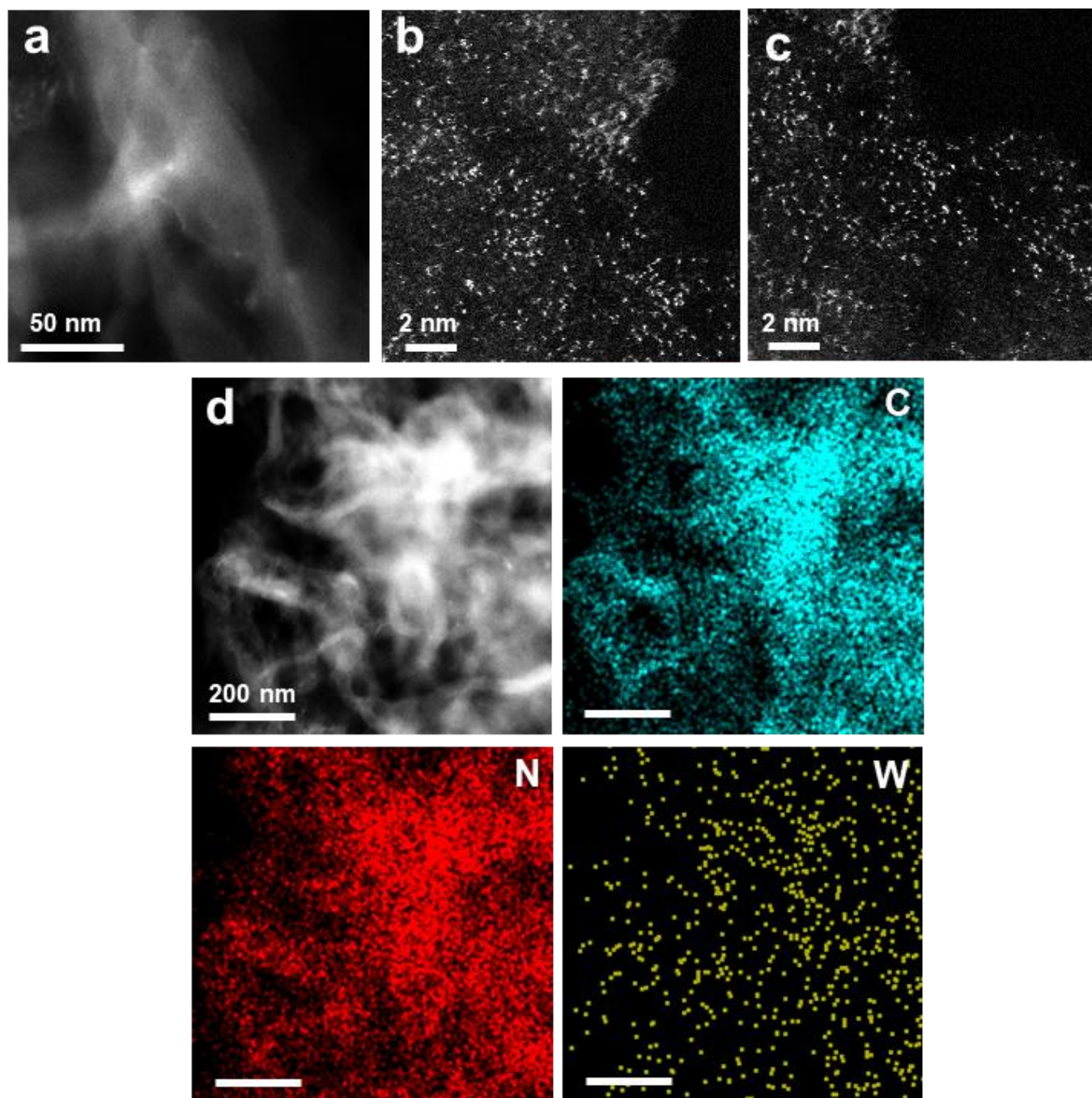
Supplementary Fig. 93 HAADF-STEM and EDS elemental mappings of W/1D-Cu₂O. a-c, HAADF-STEM images and d, EDS elemental mappings of W/1D-Cu₂O.



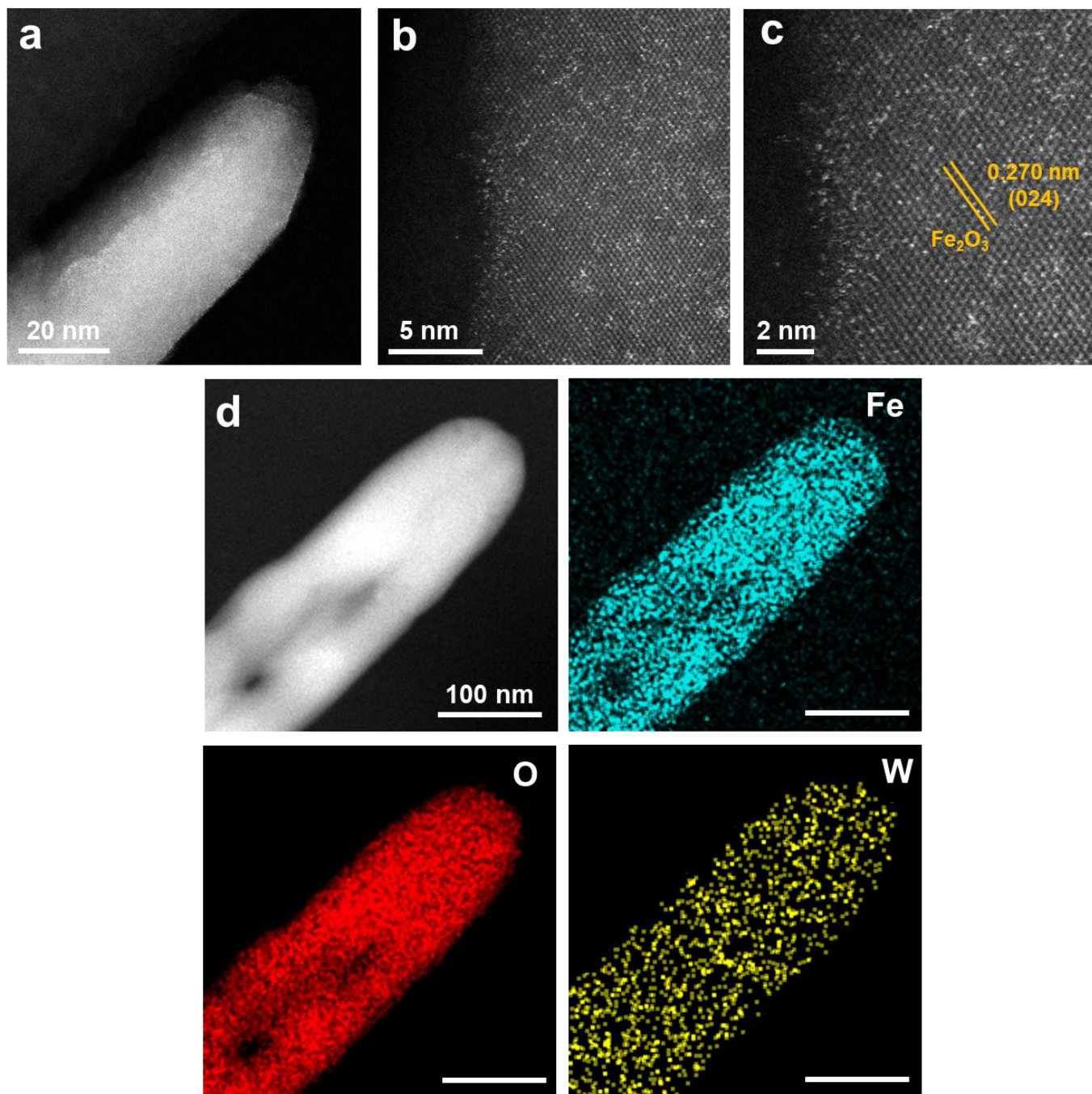
Supplementary Fig. 94 HAADF-STEM and EDS elemental mappings of W/2D-Fe₂O₃. a,b, HAADF-STEM images and c, EDS elemental mappings of W/2D-Fe₂O₃.



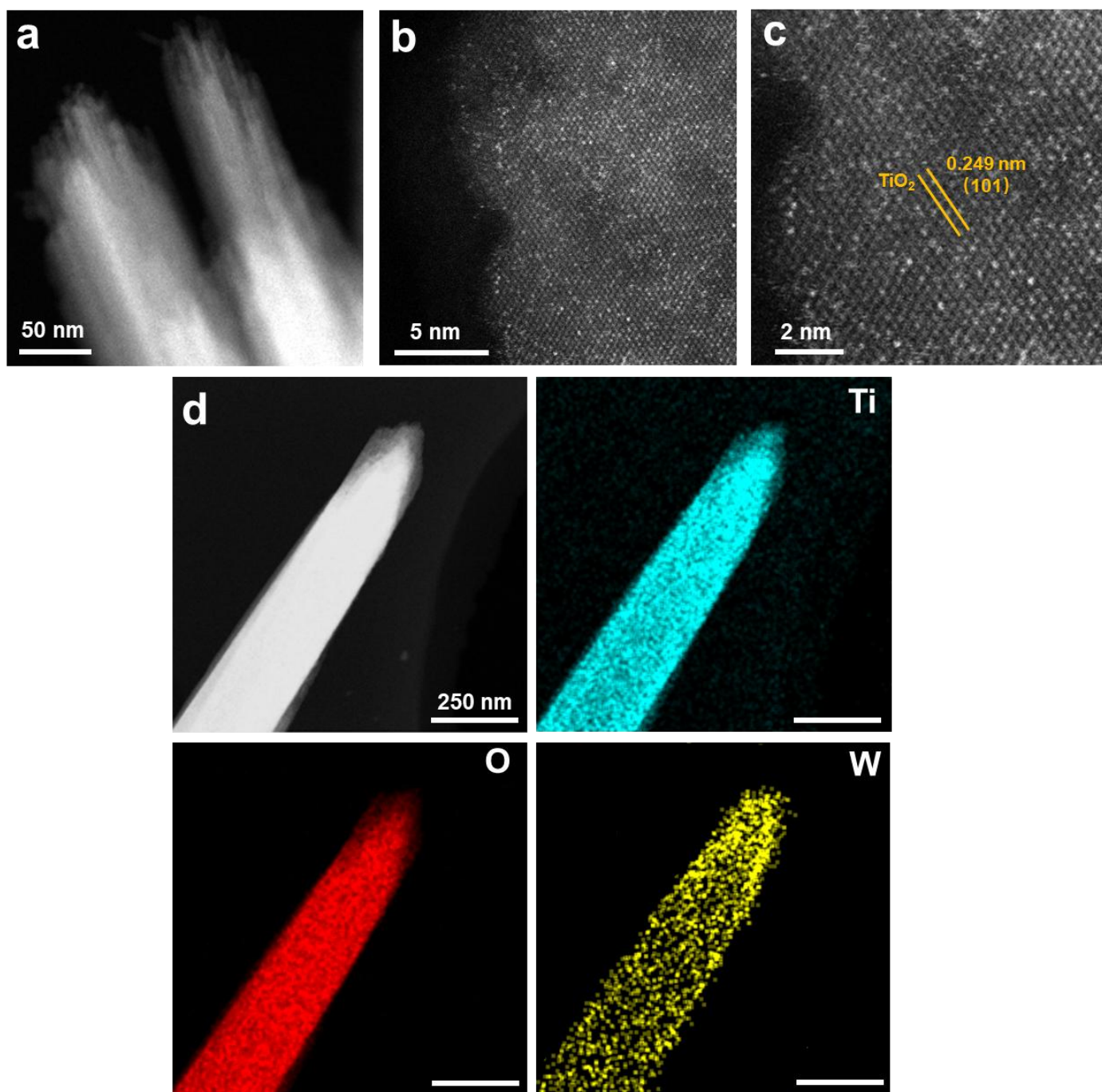
Supplementary Fig. 95 HAADF-STEM and EDS elemental mappings of W/2D-ZnIn₂S₄. a,b, HAADF-STEM images and c, EDS elemental mappings of W/2D-ZnIn₂S₄.



Supplementary Fig. 96 HAADF-STEM and EDS elemental mappings of W/2D-C₃N₄. a,b, HAADF-STEM images and c, EDS elemental mappings of W/2D-C₃N₄.



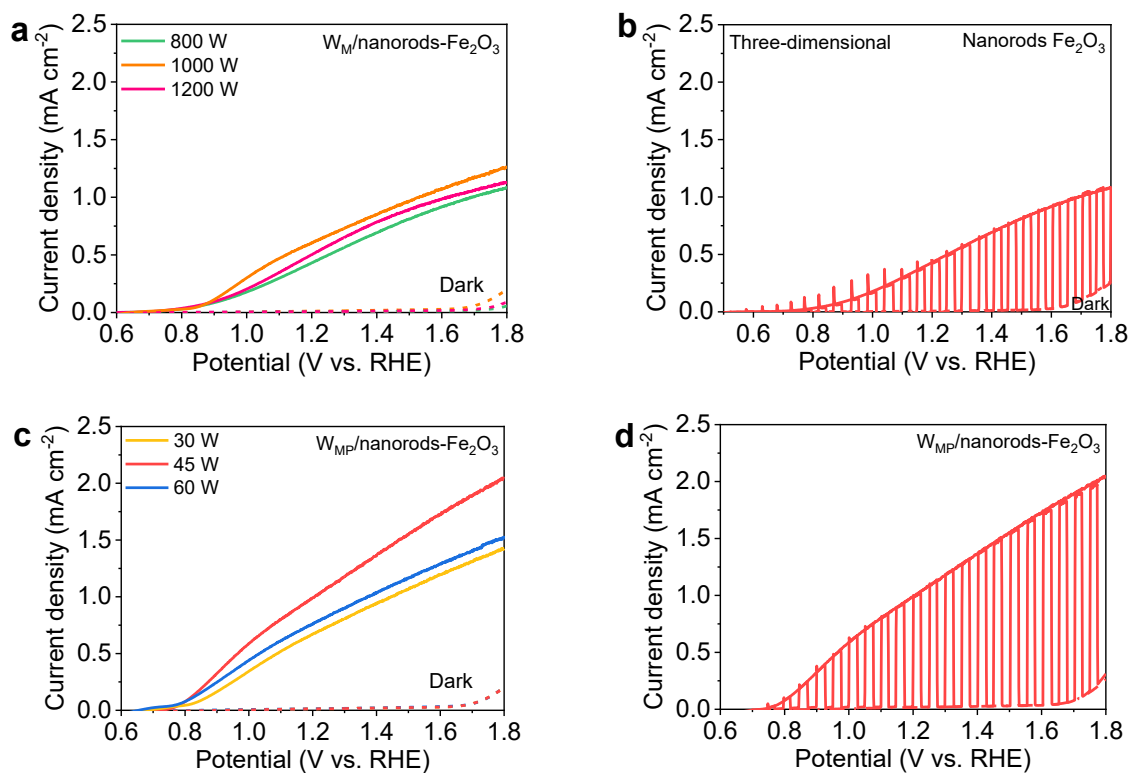
Supplementary Fig. 97 HAADF-STEM and EDS elemental mappings of W/3D-Fe₂O₃. a,b, HAADF-STEM images and c, EDS elemental mappings of W/3D-Fe₂O₃.



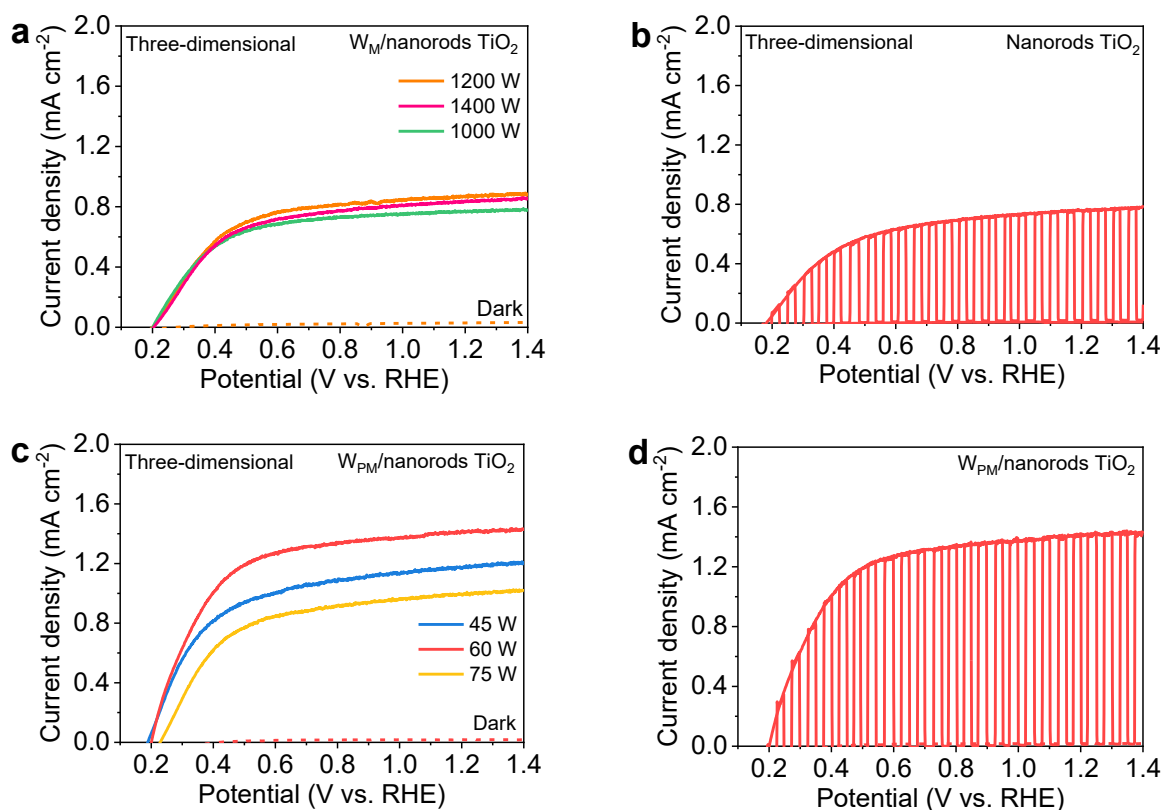
Supplementary Fig. 98 HAADF-STEM and EDS elemental mappings of W/3D-TiO₂. a,b, HAADF-STEM images and c, EDS elemental mappings of W/3D-TiO₂.



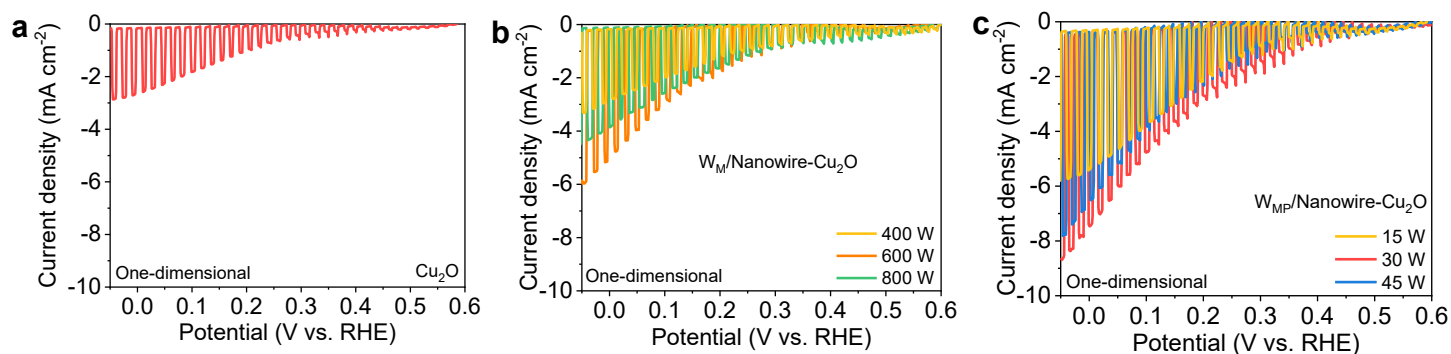
Supplementary Fig. 99 Industrial-scale production capacity. Schematic representation of kilogram-scale ultra-high single-atom-loaded Fe_2O_3 .



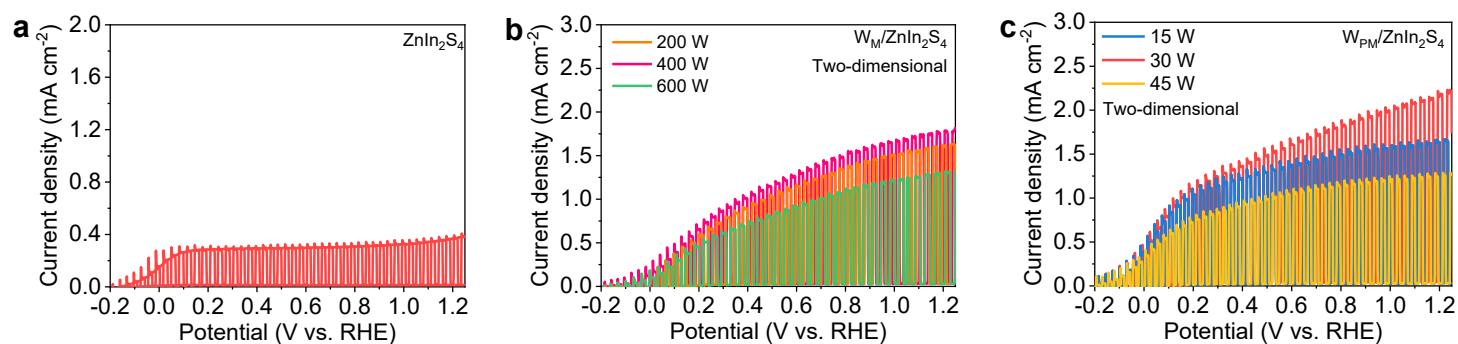
Supplementary Fig. 10 Photoelectrochemical performances of $W_M/3D-Fe_2O_3$ and $W_{PM}/3D-Fe_2O_3$. **a**, J - V curves of $W_M/nanorods-Fe_2O_3$ annealed directly by microwave at various power (800 W, 1000 W, and 1200 W) without plasma pretreatment. **b**, Chopped J - V curve of $W_M/nanorods Fe_2O_3$. **c**, J - V curves of W single atoms anchored on nanorods- Fe_2O_3 via microwave annealing after plasma pretreatment at various powers (30, 45, and 60 W). **d**, Chopped J - V curve of $W_{MP}/nanorods Fe_2O_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



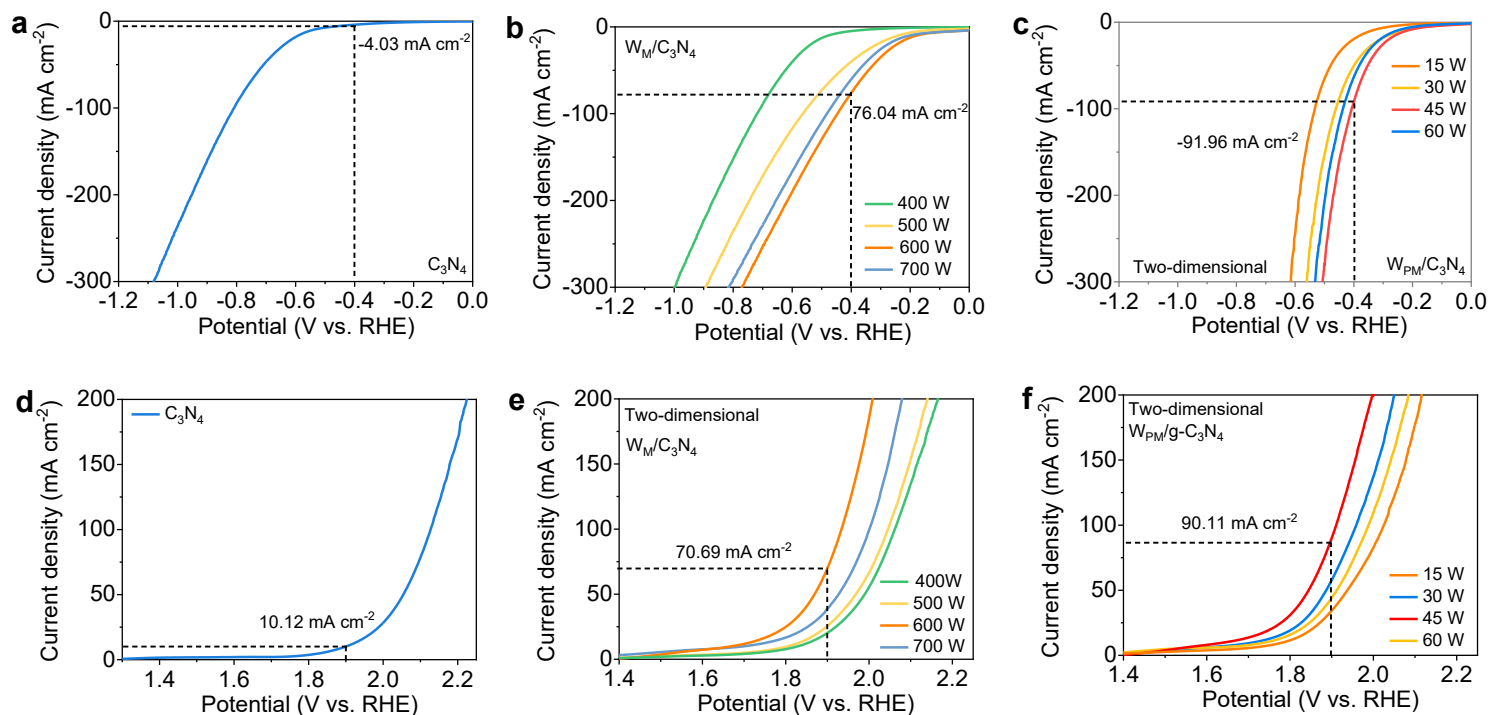
Supplementary Fig. 101 Photoelectrochemical performances of W_M/TiO_2 and W_{PM}/TiO_2 . **a**, $J-V$ curves of $W_M/\text{nanorods-TiO}_2$ annealed directly by microwave at various power (1000 W, 1200 W, and 1400 W) without plasma pretreatment. **b**, Chopped $J-V$ curve of $W_M/\text{nanorods-TiO}_2$. **c**, $J-V$ curves of W single atoms anchored on nanorods- TiO_2 via microwave annealing after plasma pretreatment at various powers (45, 60, and 75 W). **d**, Chopped $J-V$ curve of $W_{MP}/\text{nanorods TiO}_2$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



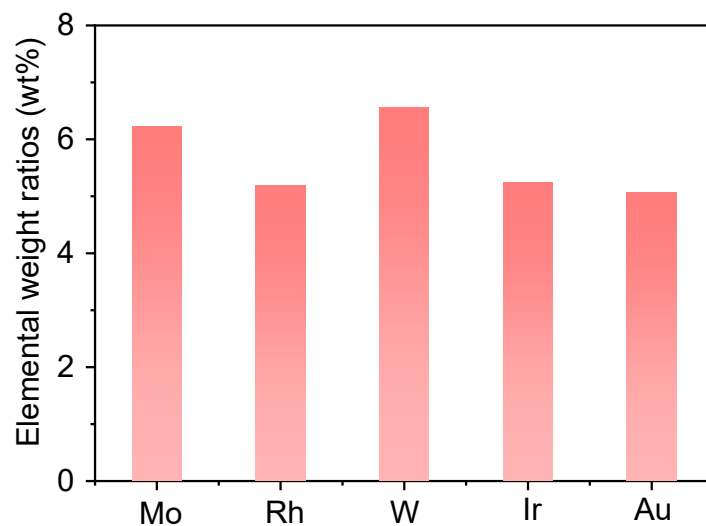
Supplementary Fig. 102 Photoelectrochemical performances of W_M/Cu_2O and W_{PM}/Cu_2O . **a**, Chopped J - V curve of nanowire- Cu_2O . **b**, Chopped J - V curves of W_M /nanowire- Cu_2O annealed directly by microwave at various power (400 W, 600 W, and 800 W) without plasma pretreatment. **c**, J - V curves of W single atoms anchored on nanowire Cu_2O via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



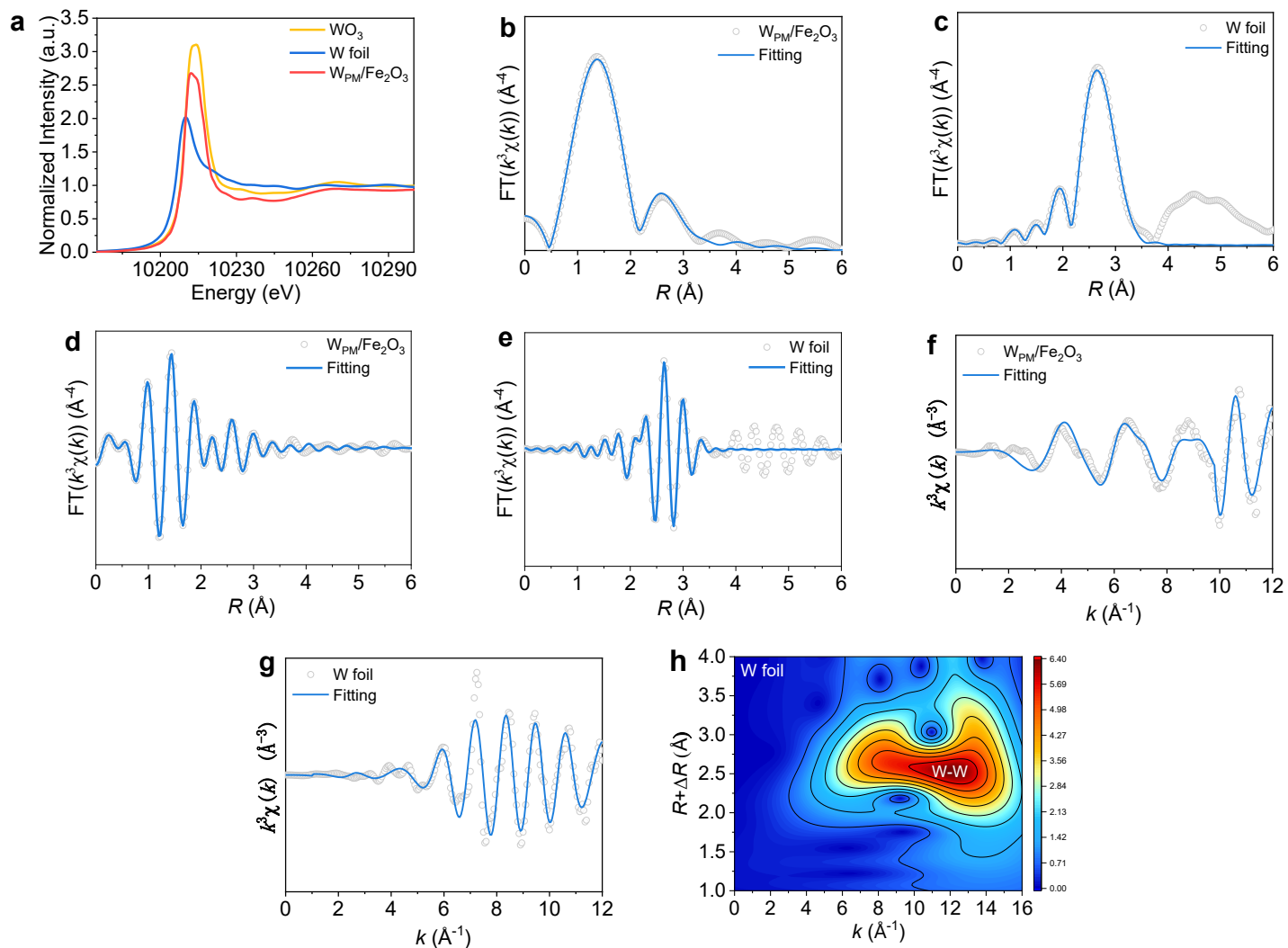
Supplementary Fig. 103 Photoelectrochemical performances of $\text{W}_M/\text{ZnIn}_2\text{S}_4$ and $\text{W}_{PM}/\text{ZnIn}_2\text{S}_4$. **a**, Chopped J - V curve of ZnIn_2S_4 . **b**, Chopped J - V curves of $\text{W}_M/\text{ZnIn}_2\text{S}_4$ annealed directly by microwave at various power (200 W, 400 W, and 600 W) without plasma pretreatment. **c**, J - V curves of W single atoms anchored on ZnIn_2S_4 via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm⁻²).



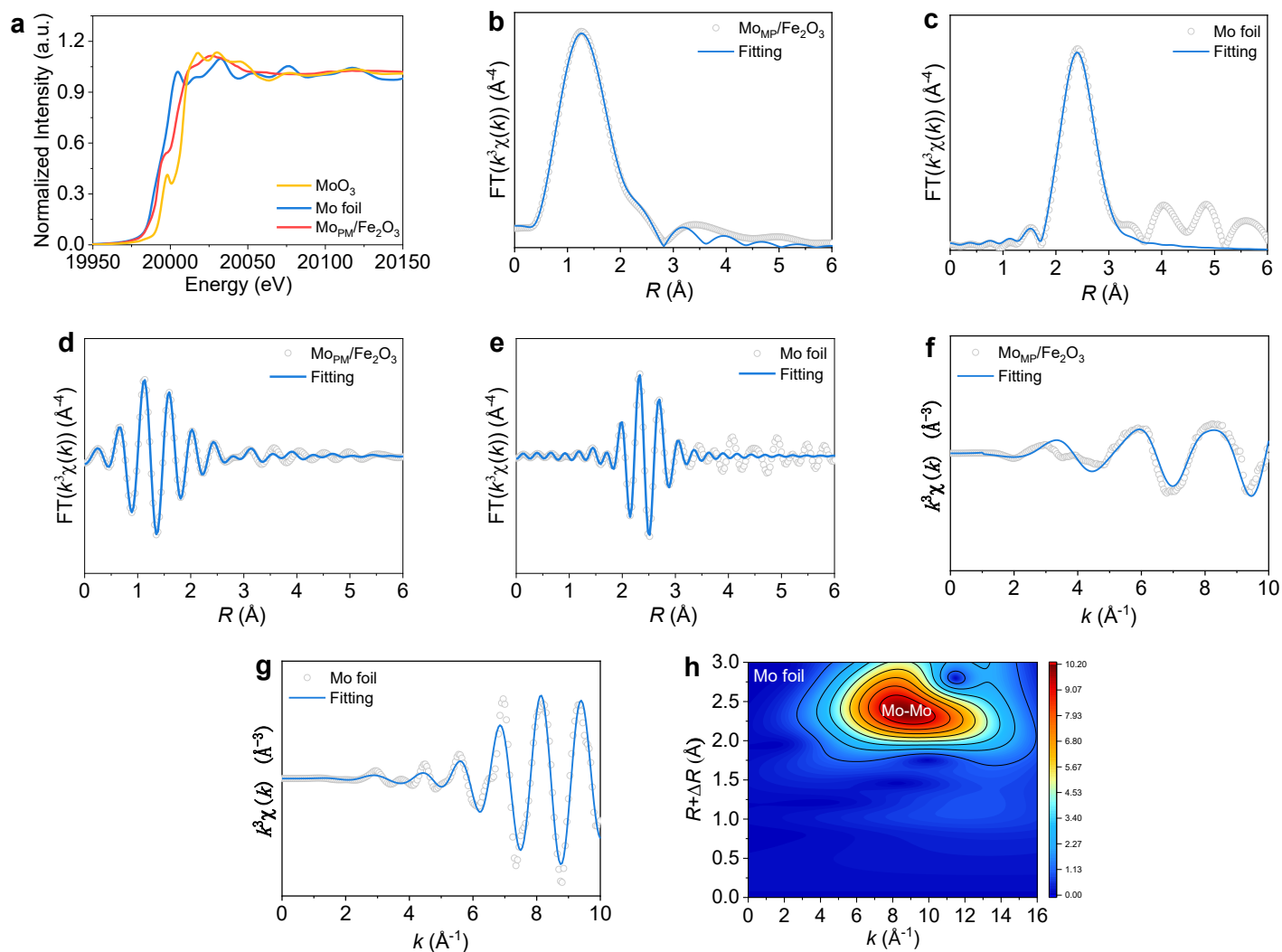
Supplementary Fig. 104 Electrochemical HER and OER performances of W_{PM}/C_3N_4 . **a**, HER performance of C_3N_4 at $-0.4 V_{RHE}$. **b**, HER J - V curves of C_3N_4 annealed directly by microwave at various power (400, 500, 600, and 700 W) without plasma pretreatment. **c**, HER J - V curves of W single atoms anchored on C_3N_4 via microwave annealing after plasma pretreatment at various powers (15, 30, 45, and 60 W). **d**, OER performance of C_3N_4 at $-0.4 V_{RHE}$. **e**, OER J - V curves of C_3N_4 annealed directly by microwave at various power (400, 500, 600, and 700 W) without plasma pretreatment. **f**, OER J - V curves of W single atoms anchored on C_3N_4 via microwave annealing after plasma pretreatment at various powers (15, 30, 45, and 60 W). All measurements were performed in 0.5 M H_2SO_4 .



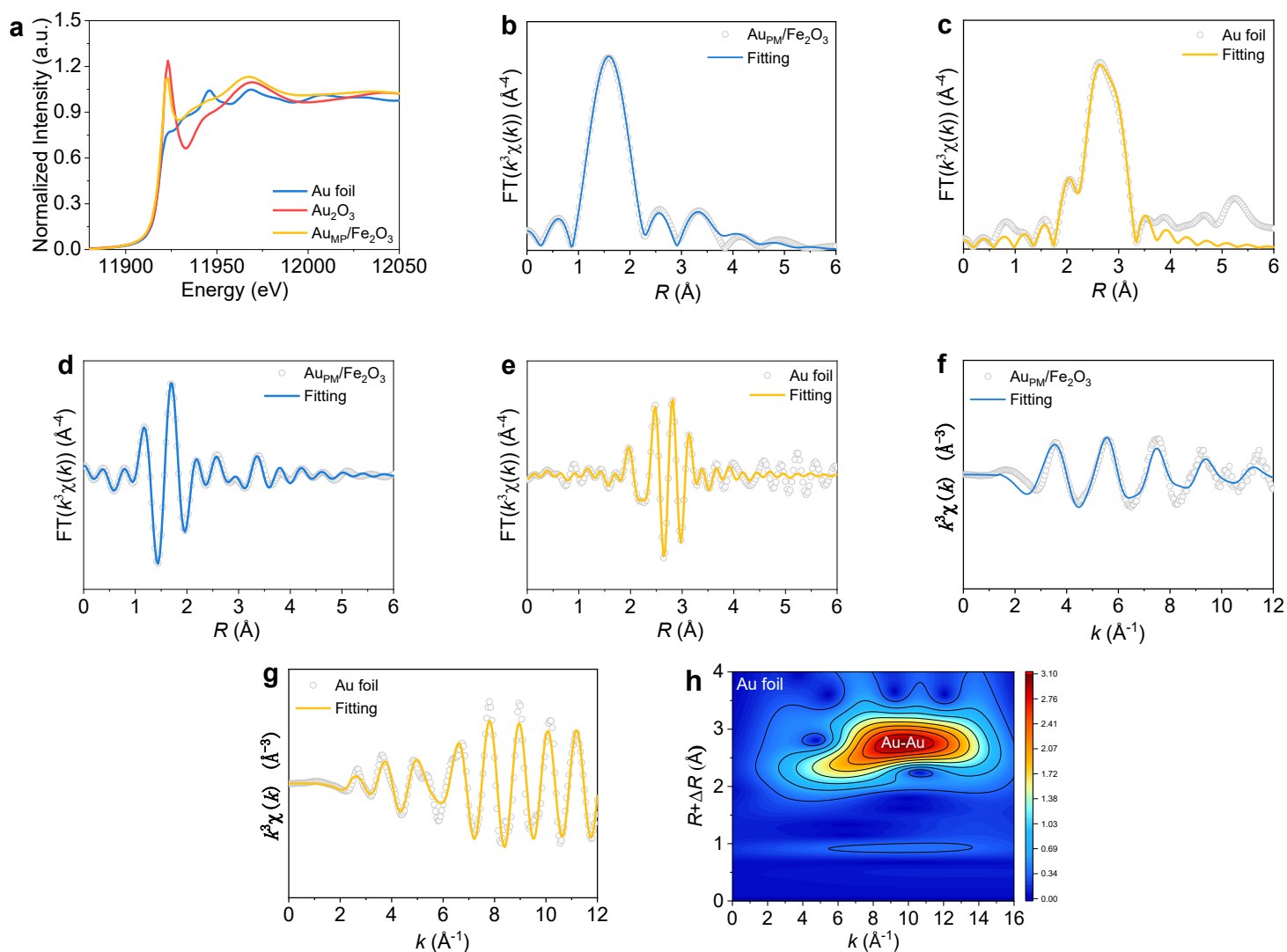
Supplementary Fig. 105 Elemental ratios of the 5-M_{PM}/Fe₂O₃. Surface elemental weight ratios of the 5-M_{PM}/Fe₂O₃ determined by X-ray photoelectron spectroscopy.



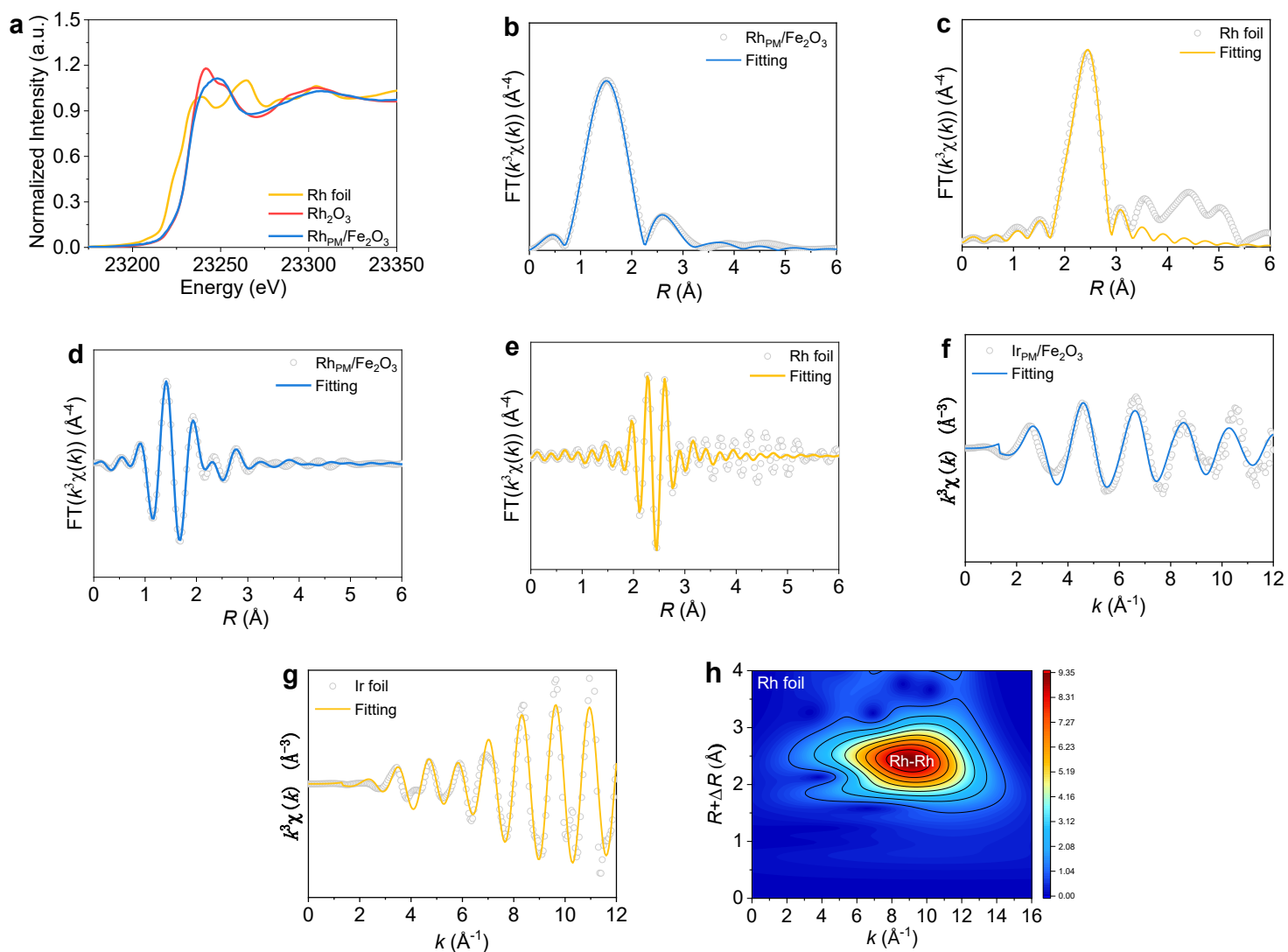
Supplementary Fig. 106 Atom structural characterization of 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, W L_3 -edge XANES spectra of 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$, W foil and WO_3 . **b-e**, W L_3 -edge EXAFS data and fit for 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and W foil in R -space and **f, g**, k -space. **h**, WT-EXAFS plot of W foil.



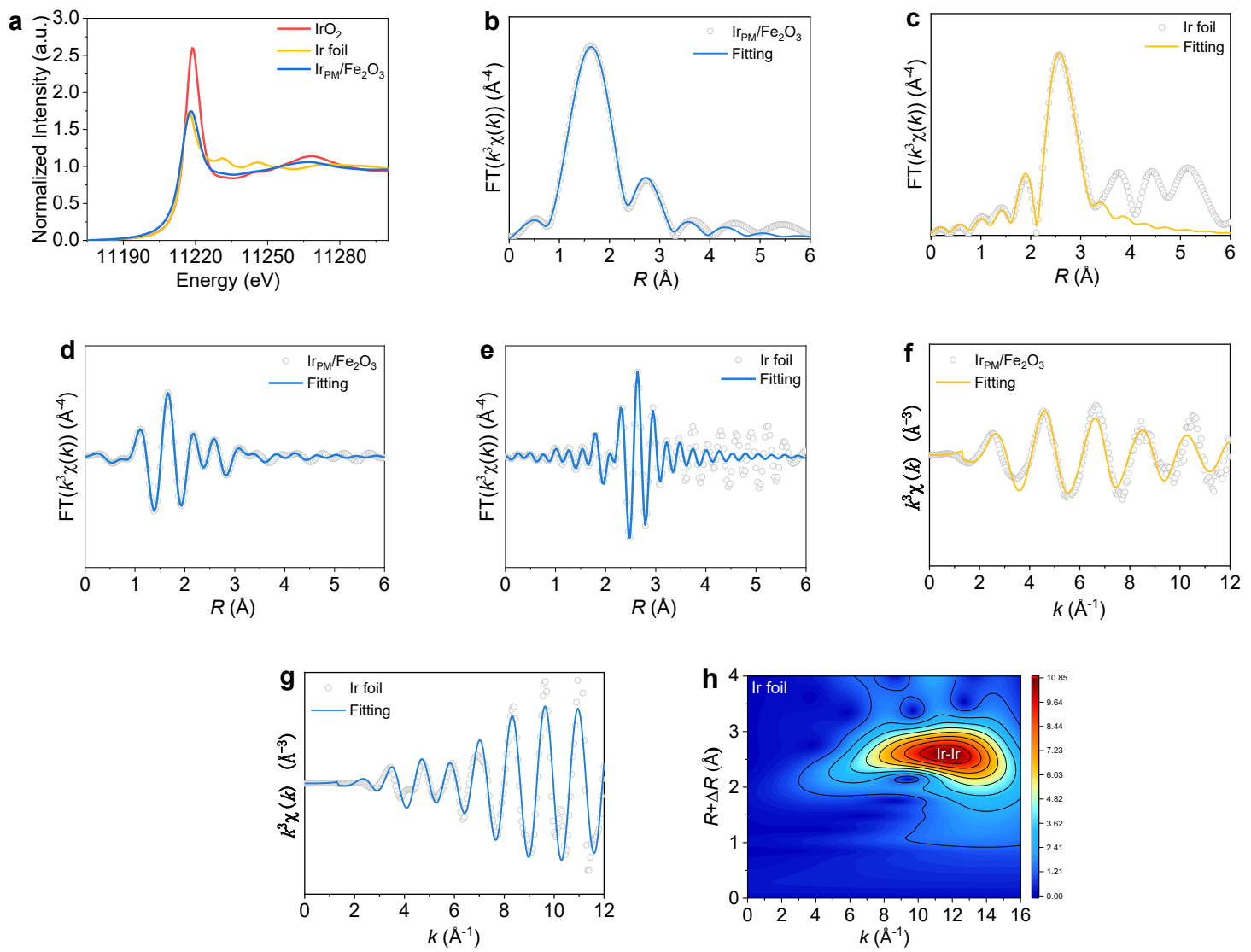
Supplementary Fig. 107 Atom structural characterization of 5-MPM/Fe₂O₃. **a**, Mo L_3 -edge XANES spectra of 5-MPM/Fe₂O₃, Mo foil and MoO₃. **b-e**, Mo L_3 -edge EXAFS data and fit for 5-MPM/Fe₂O₃ and Mo foil in R -space and **f, g**, k -space. **h**, WT-EXAFS plot of Mo foil.



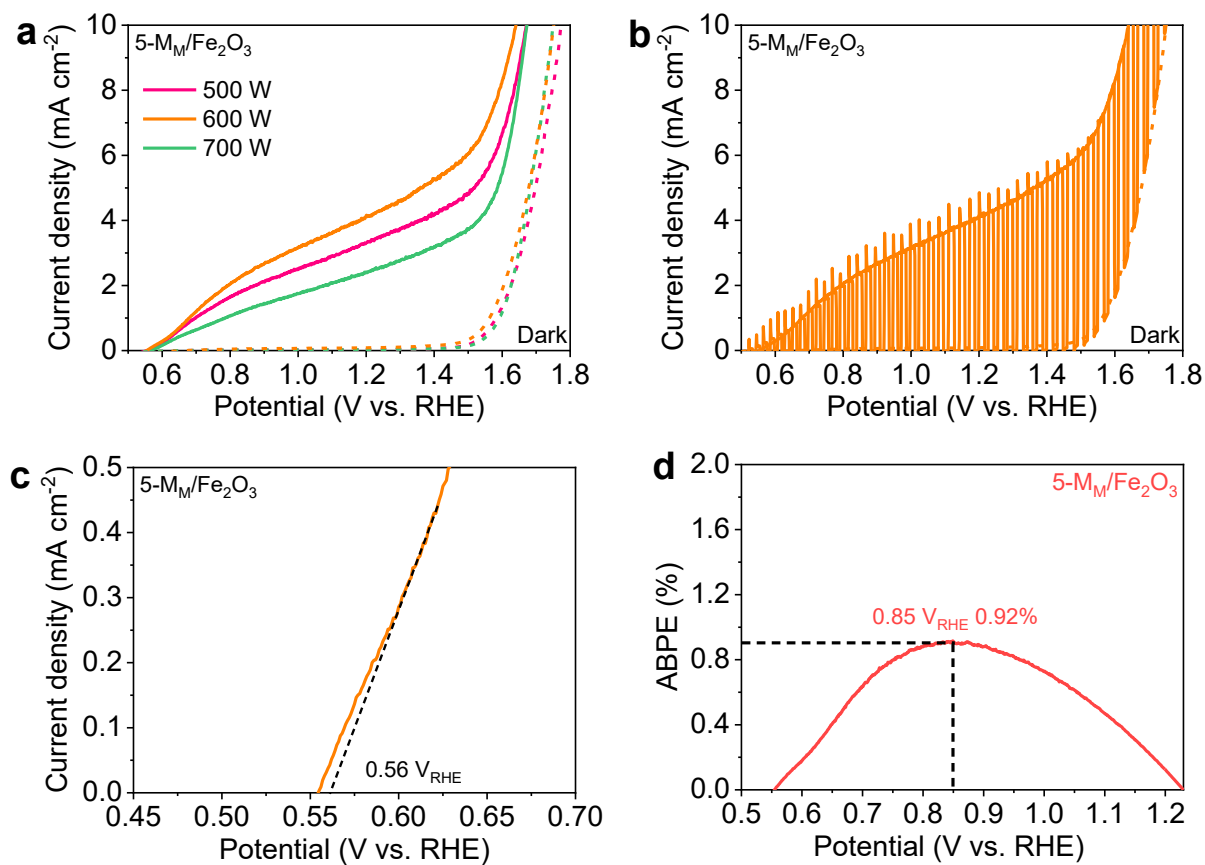
Supplementary Fig. 108 Atom structural characterization of 5-M_{PM}/Fe₂O₃. **a**, Au *K*-edge XANES spectra of 5-M_{PM}/Fe₂O₃, Au foil and Au₂O₃. **b-e**, Au *K*-edge EXAFS data and fit for 5-M_{PM}/Fe₂O₃ and Au foil in *R*-space and **f**, **g**, *k*-space. **h**, WT-EXAFS plot of Au foil.



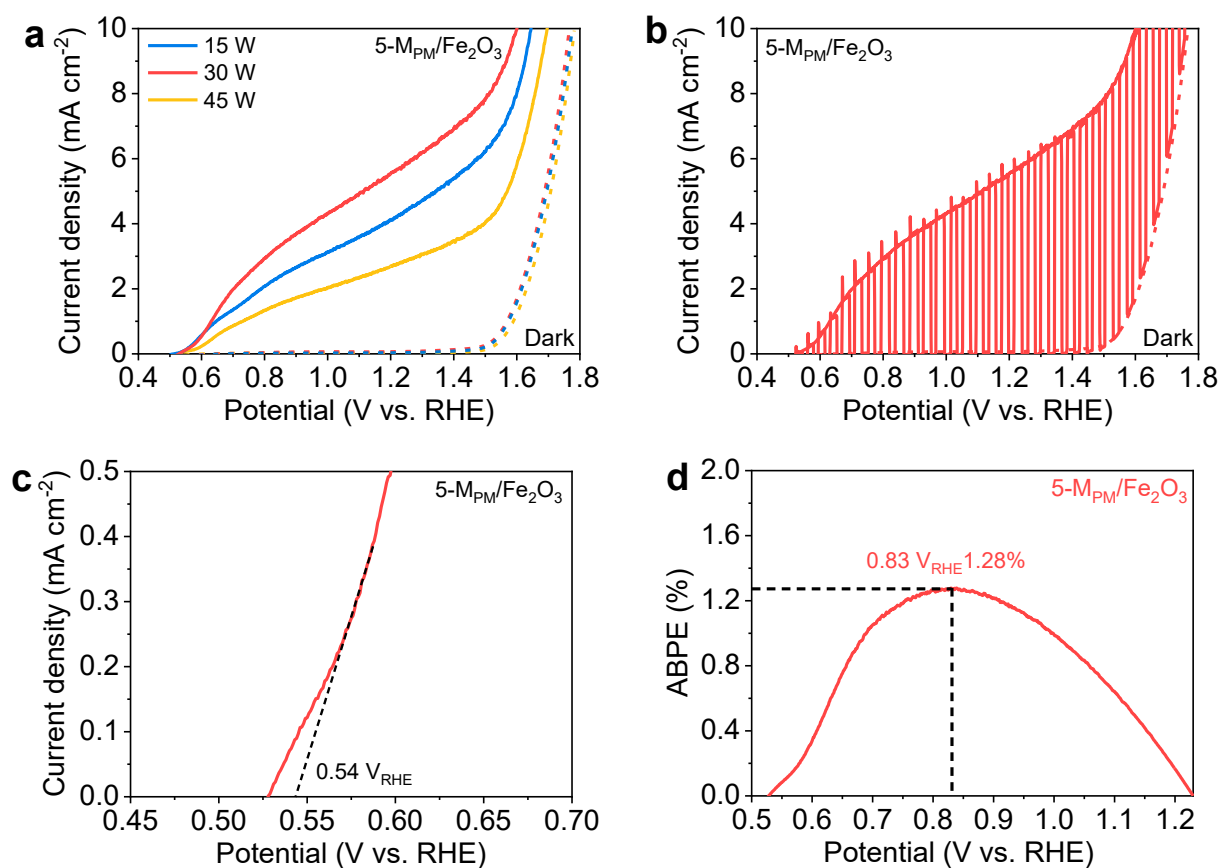
Supplementary Fig. 109 Atom structural characterization of 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, Rh K -edge XANES spectra of 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$, Rh foil and Rh_2O_3 . **b-e**, Rh K -edge EXAFS data and fit for 5- $\text{M}_{\text{PM}}/\text{Fe}_2\text{O}_3$ and Rh foil in R -space and **f, g**, k -space. **h**, WT-EXAFS plot of Rh foil.



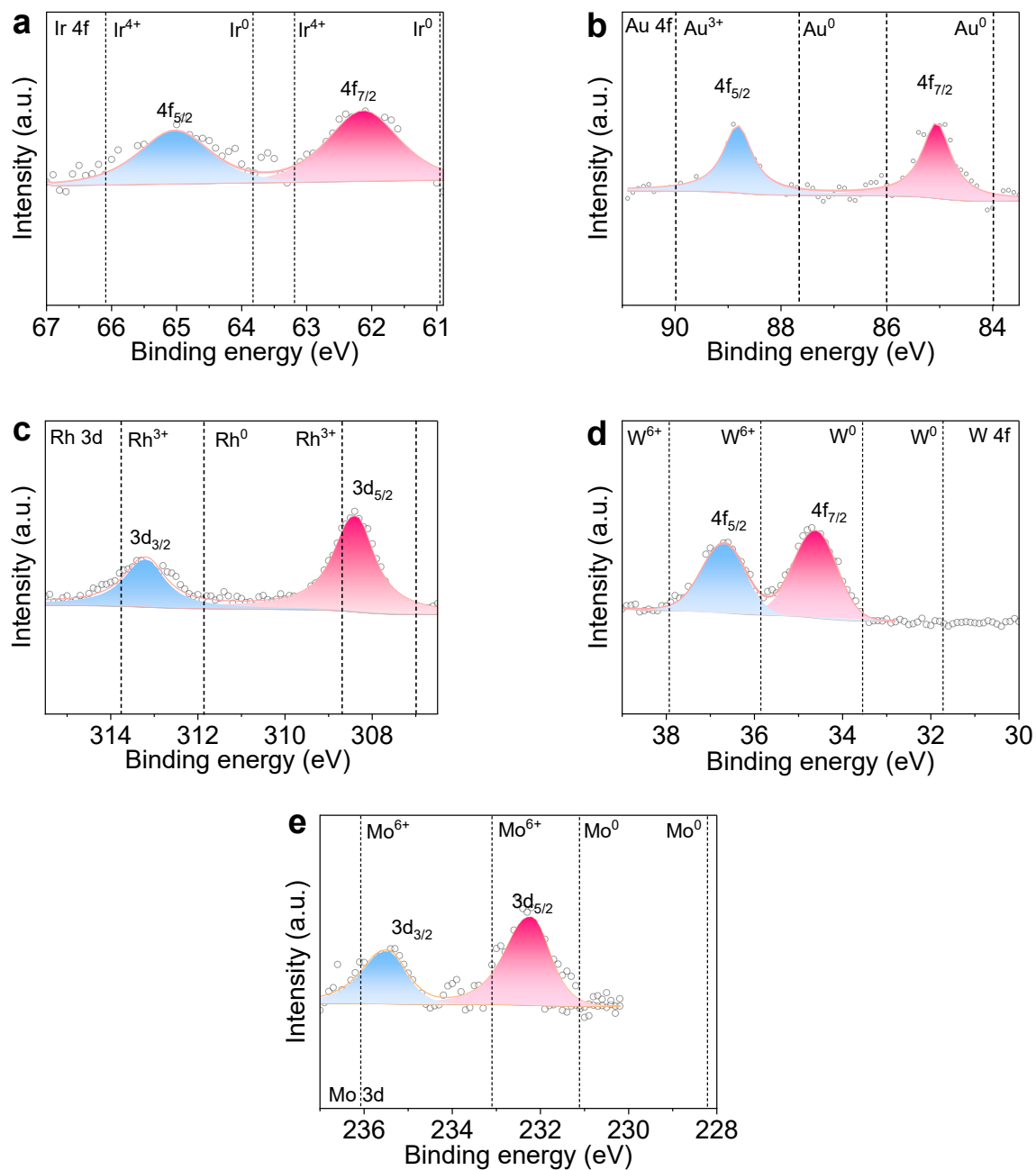
Supplementary Fig. 110 Atom structural characterization of 5-M_{PM}/Fe₂O₃. **a**, Ir L_3 -edge XANES spectra of 5-M_{PM}/Fe₂O₃, Ir foil and Rh₂O₃. **b-e**, Ir L_3 -edge EXAFS data and fit for 5-M_{PM}/Fe₂O₃ and Ir foil in R -space and **f, g**, k -space. **h**, WT-EXAFS plot of Ir foil.



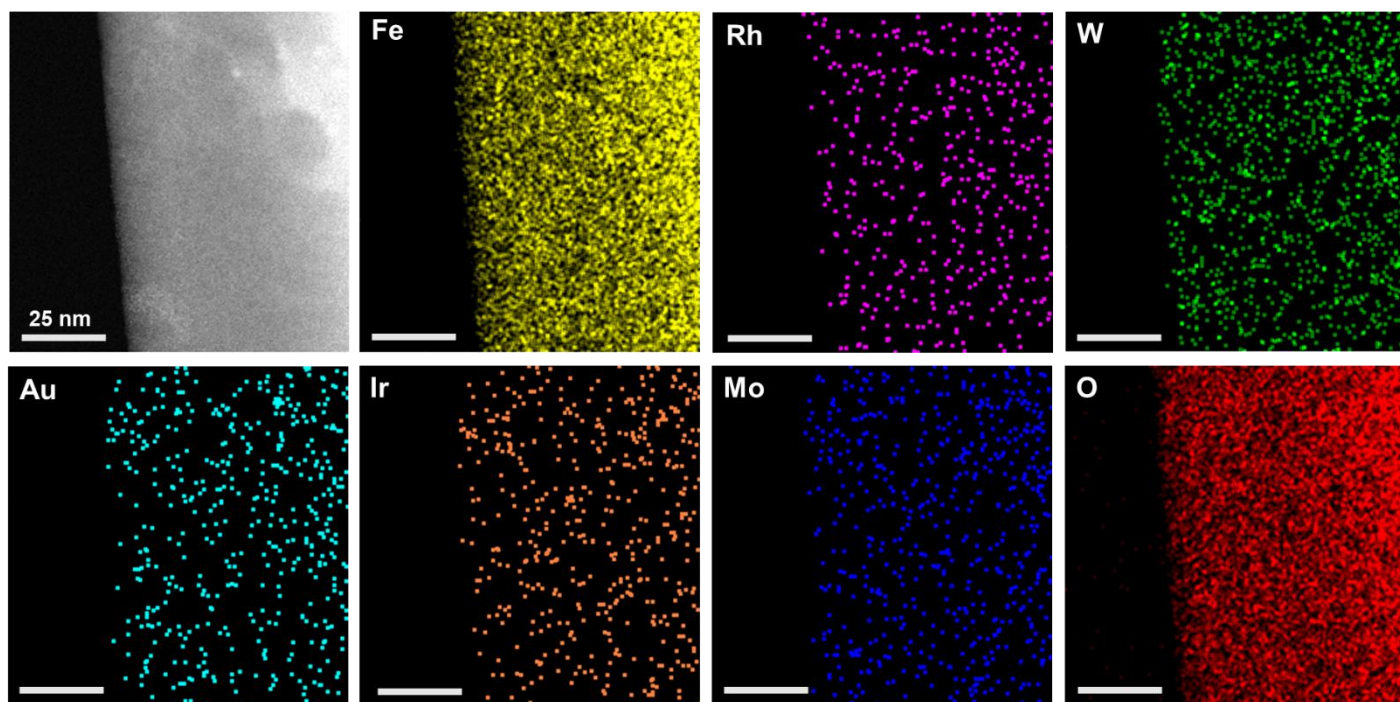
Supplementary Fig. 111 Photoelectrochemical OER performances of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Fe_2O_3 loaded with five metals (W, Rh, Mo, Ir, and Au) annealed directly by microwave at various power (500 W, 600 W, and 700 W) without plasma pretreatment. **b**, Chopped J - V curve of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$. **c**, Onset potential of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$. **d**, Applied bias photon-to current efficiency of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



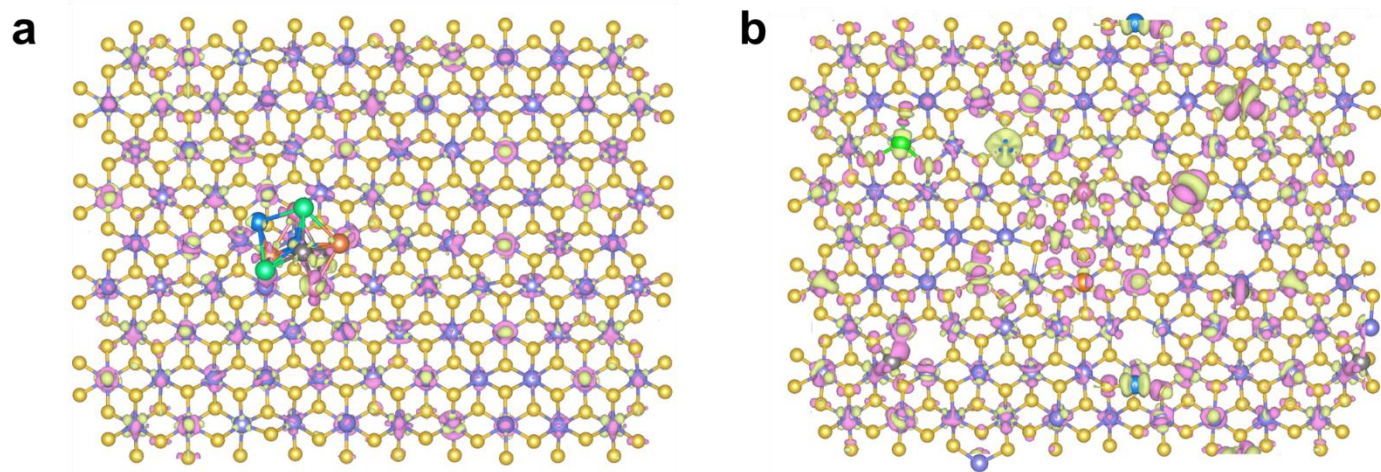
Supplementary Fig. 112 Photoelectrochemical OER performance of $5\text{-M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **a**, J - V curves of Fe_2O_3 loaded with five metals (W, Rh, Mo, Ir, and Au) via microwave annealing after plasma pretreatment at various powers (15, 30, and 45 W). **b**, Chopped J - V curve of $5\text{-M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **c**, Onset potential of $5\text{-M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. **d**, Applied bias photon-to-current efficiency of $5\text{-M}_{\text{PM}}/\text{Fe}_2\text{O}_3$. All measurements were performed in 1 M KOH under AM 1.5G illumination (100 mW cm^{-2}).



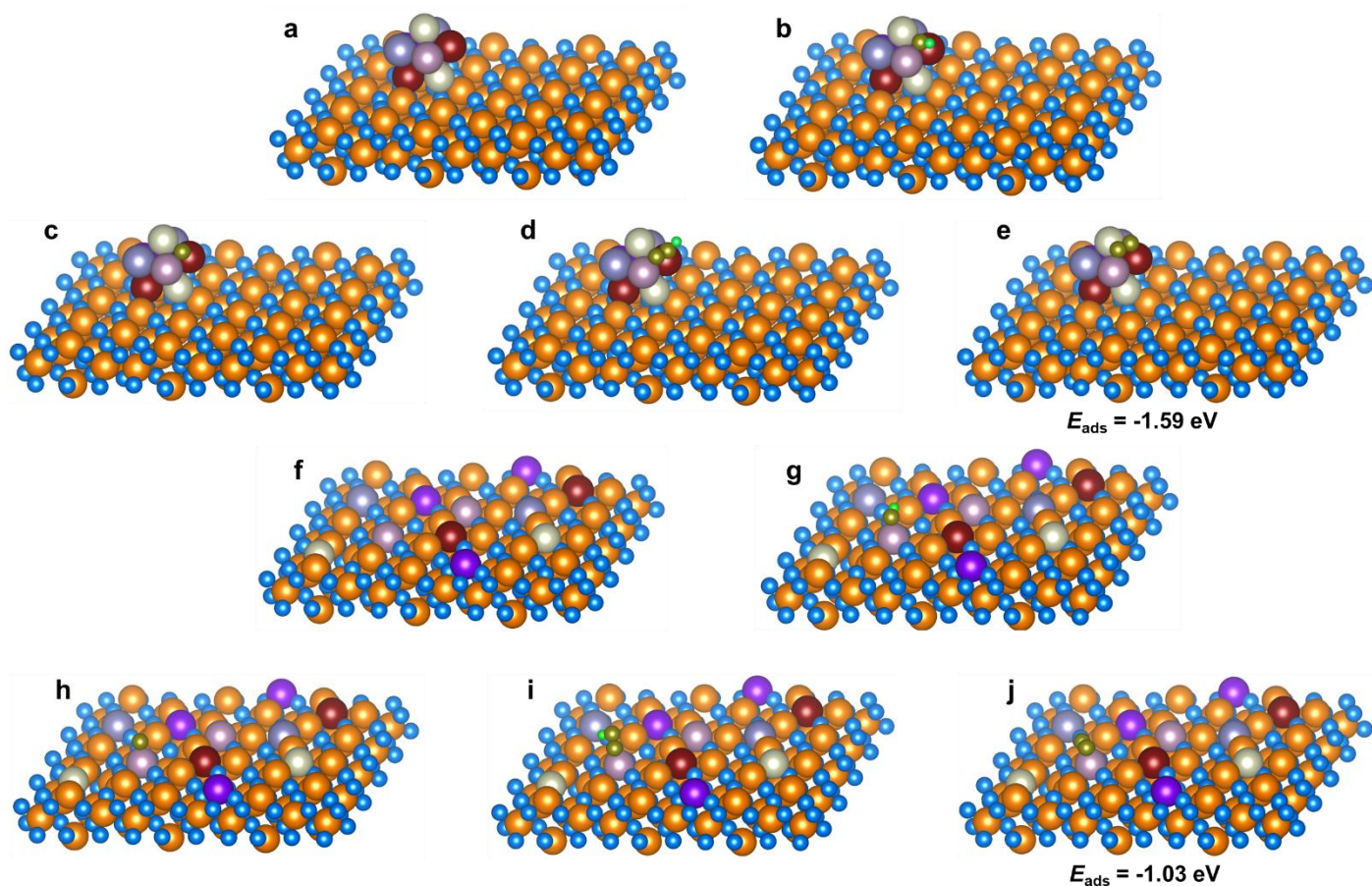
Supplementary Fig. 113 XPS elemental map after stability. XPS spectra of Ir 4f, Au 4f, Rh 3d, W 4f, and Mo 3d for 5-M_{PM}/Fe₂O₃ after stability testing.



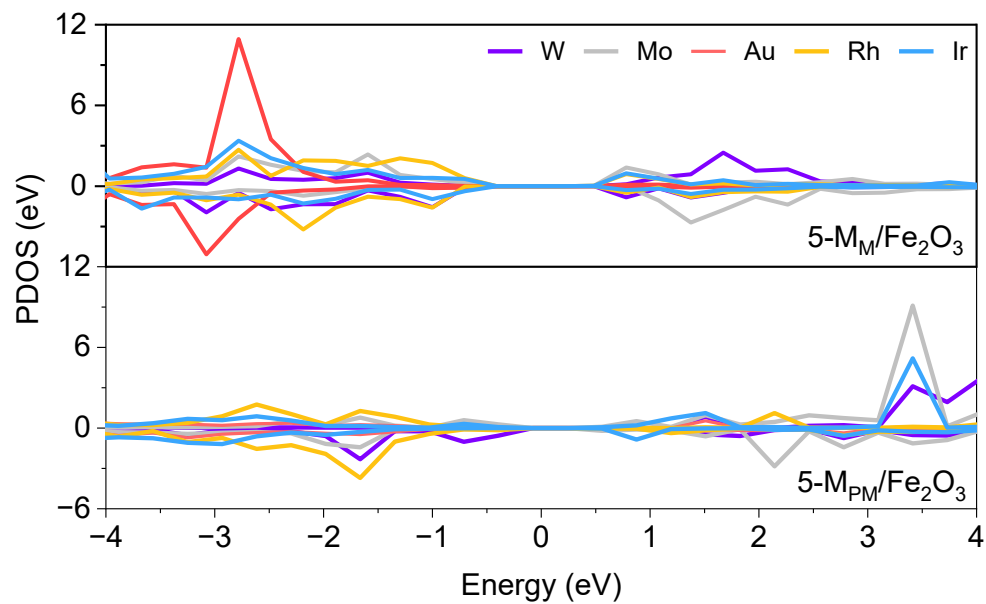
Supplementary Fig. 114 EDS elemental map after stability. EDS elemental mappings of 5-M_{PM}/Fe₂O₃ after stability testing.



Supplementary Fig. 115 Differential charge density. **a,b**, Differential charge density of 5- M_M /Fe₂O₃ (a) and 5- M_{PM} /Fe₂O₃ (b).



Supplementary Fig. 116 Structure of the key steps and intermediates in OER. (a-e) OER proceeding on the surfaces of 5-M_M/Fe₂O₃ and (f-j) 5-M_{PM}/Fe₂O₃. Here, Fe, O, Mo, Au, Ir, W, Rh and H atoms are represented by orange, pink, grey, purple, red, beige and green, respectively. To distinguish adsorbed oxygen, it is shown in brown.



Supplementary Fig. 117 Electronic structures of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$ and $5\text{-M}_\text{PM}/\text{Fe}_2\text{O}_3$. Projected density of states of $5\text{-M}_\text{M}/\text{Fe}_2\text{O}_3$ and $5\text{-M}_\text{PM}/\text{Fe}_2\text{O}_3$.

Supplementary Table 1 Comparison between this work and representative reported UHL-SACs.

Catalysts	Metal-atom loadings	Reference
5-M _{PM} /Fe ₂ O ₃	27.69 wt% (7.75 at.%)	This work.
W _{PM} /Fe ₂ O ₃	21.02 wt% (4.13 at.%)	
Pt SAC	11.4 wt%	<i>Nat. Mater.</i> 21 , 681-688 (2022).
Ni-N-C	15.3 wt% (3.61 at.%)	<i>Nat. Chem.</i> 13 , 887-894 (2021).
Ir-N-C	41.6 wt% (3.8 at.%)	
Pt ₁ /NC	<25 wt%	<i>Nat. Nanotechnol.</i> 17 , 174-181 (2022).
Cu-N-C	0.6 wt%	<i>Nat. Nanotechnol.</i> 17 , 403-407 (2022).
Rh SACs/PCN	23 wt%	<i>Nat. Nanotechnol.</i> 17 , 174-181 (2022).
Pt/CA-CNF	~0.24 wt%	<i>Nat. Nanotechnol.</i> 14 , 851-857 (2019).
SA-Rh/CN	0.92 wt%	<i>Nat. Nanotechnol.</i> 15 , 390–397 (2020).
SA-Pd/CN	0.76 wt%	
SA-Pt/CN	0.57 wt%	
A-Ni-NSG	2.8 wt% (0.47 at.%)	<i>Nat. Energy</i> 3 , 140-147 (2018).
Cu-SAs/N-G	1.3 wt%	<i>Nat. Catal.</i> 1 , 781-786 (2018).
UHD Ni-N-C SACs	23.3 wt%	<i>Nat. Synth.</i> 3 , 1427-1438 (2024).
UHD Cu-N-C SACs	26.9 wt%	
UHD Pt-N-C SACs	33.4 wt%	
Ni SACs	20.8 wt%	<i>Nat. Synth.</i> 2 , 129-139 (2023).
2000-Pt-N-CNT	7.0 wt%	<i>Nat. Commun.</i> 13 , 685 (2022).
Cu-NC SAC	10.5 wt%	<i>Nat. Commun.</i> 10 , 1278 (2019).
Ir/meso_S-C	~10 wt% (~0.77 at.%)	<i>Sci. Adv.</i> 5 , eaax6322 (2019).

Red denotes single-atom-loaded metal oxides, while blue represents the C/N substrate.

Supplementary Table 2 Surface elemental weight ratios of the UHL-Ir/Fe₂O₃ determined by X-ray photoelectron spectroscopy.

	Ir (at.%)	Fe (at.%)	O (at.%)
UHL-Ir/Fe ₂ O ₃	4.12	36.17	59.71
	Ir (wt%)	Fe (wt%)	O (wt%)
UHL-Ir/Fe ₂ O ₃	21.02 wt%	53.62	25.36

Supplementary Table 3 Structural parameters extracted from Ir- L_3 edge $\chi(R)$ space spectra fitting of UHL-Ir/Fe₂O₃.

	S_0^2	shell	CN*	R(Å)	σ^2 (10^{-3} Å ²)	ΔE_0	R factor
Ir foil	0.95	Ir-Ir	12	2.71±0.01	3.5±1.7	9.60±3.88	0.0073
UHL-Ir/Fe ₂ O ₃	0.95	Ir-O	6.2	1.92±0.01	2.3±1.7	-3.35±2.40	0.0087
		Ir-O-Fe	5.2	3.33±0.01	11.0±0.1		

CN: coordination numbers. R : bond distance. σ^2 : Debye-Waller factors. ΔE_0 : the inner potential correction. R factor: goodness of fit.

Supplementary Table 4. Metal precursor salts used in the synthesis of UHL-SACs/Fe₂O₃.

Elements	Atomic numbers	Metal precursors	Solubleness in water
Y	39	YCl ₃ ·6H ₂ O	freely soluble
Zr	40	ZrCl ₄	very soluble
Nb	41	NbCl ₅	decomposes
Mo	42	(NH ₄) ₆ Mo ₇ O ₂₄ ·4H ₂ O	soluble
Ru	44	RuCl ₃	very soluble
Rh	45	RhCl ₃ ·xH ₂ O	very soluble
Pd	46	PdCl ₂	soluble
Ag	47	AgNO ₃	very soluble
Cd	48	Cd(Ac) ₂ ·2H ₂ O	very soluble
In	49	In(Ac) ₃	soluble
Sn	50	SnCl ₄	very soluble
Sb	51	Sb(Ac) ₃	Moderately soluble
Ba	56	Ba(Ac) ₂	freely soluble
La	57	La(Ac) ₃ ·xH ₂ O	soluble
Ce	58	(NH ₄) ₆ Ce(SO ₄) ₅ ·4H ₂ O	soluble
Pr	59	Pr(Ac) ₃ ·xH ₂ O	soluble
Nd	60	Nd(Ac) ₃ ·xH ₂ O	soluble
Sm	62	Sm(Ac) ₃ ·xH ₂ O	soluble
Gd	64	Gd(Ac) ₃ ·4H ₂ O	soluble
Tb	65	Tb(Ac) ₃ ·6H ₂ O	soluble
Dy	66	Dy(Ac) ₃ ·4H ₂ O	soluble
Er	68	Er(Ac) ₃ ·4H ₂ O	soluble
Tm	69	Tm(Ac) ₃ ·H ₂ O	soluble
Yb	70	Yb(Ac) ₃ ·4H ₂ O	soluble

Lu	71	$\text{LuCl}_3 \cdot 6\text{H}_2\text{O}$	soluble
Hf	72	HfCl_4	soluble
W	74	$(\text{NH}_4)_{10}\text{W}_{12}\text{O}_{41} \cdot x\text{H}_2\text{O}$	soluble
Ir	77	$\text{H}_2\text{Cl}_6\text{Ir}$	soluble
Pt	78	$\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}$	highly soluble
Au	79	$\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$	very soluble
Pb	82	$\text{Pb}(\text{Ac})_2 \cdot 3\text{H}_2\text{O}$	freely soluble
Bi	83	$\text{Bi}(\text{NO}_3)_3$	slightly soluble

Supplementary Table 5 Structural parameters extracted from $\chi(R)$ spatial spectrum fitting of 33 kinds of UHL-SACs/Fe₂O₃ prepared by the plasma-microwave method.

	S_0^2	shell	CN*	R(Å)	σ^2 (10^{-3} Å ²)	ΔE_0	R factor
Y foil	0.95	Y-Y	12.00	3.60±0.01	14.8±3.6	5.03±2.51	0.0131
UHL-Y _{PM} /Fe ₂ O ₃	0.95	Y-O	5.97	2.38±0.01	5.3±1.3	-0.16±0.01	0.0085
Zr foil	0.90	Zr-Zr	8.00	3.19±0.01	7.6±0.3	-8.13±0.42	0.0049
		Zr-Zr	6.00	3.61±0.01	14.6±0.1		
UHL-Zr _{PM} /Fe ₂ O ₃	0.90	Zr-O	5.61	2.17±0.01	7.1±1.0	-0.23±0.01	0.0079
Nb foil	0.90	Nb-Nb	8.00	2.85±0.01	6.5±0.3	4.32±1.03	0.0137
		Nb-Nb	6.00	3.28±0.01	6.8±0.6		
UHL-Nb _{PM} /Fe ₂ O ₃	0.90	Nb-O	5.34	1.99±0.01	13.1±1.5	2.65±0.89	0.0188
Mo foil	0.80	Mo-Mo	8.00	2.72±0.01	2.8±0.2	-6.42±0.74	0.0038
		Mo-Mo	6.00	3.13±0.01	2.2±0.3		
UHL-Mo _{PM} /Fe ₂ O ₃	0.80	Mo-O	5.12	1.75±0.01	12.7±0.5	-8.93±0.93	0.0024
Ru foil	0.84	Ru-Ru	6.00	2.68±0.01	1.7±0.1	-3.17±0.85	0.0080
		Ru-Ru	6.00	2.67±0.01	9.0±2.1		
UHL-Ru _{PM} /Fe ₂ O ₃	0.84	Mo-O	4.82	2.07±0.01	5.1±1.2	5.42±1.24	0.0126
Rh foil	0.82	Rh-Rh	12.00	2.68±0.01	3.6±0.2	-6.03±0.62	0.0087
UHL-Rh _{PM} /Fe ₂ O ₃	0.82	Rh-O	4.57	2.03±0.01	4.1±0.3	-3.63±0.30	0.0011
Pd foil	0.84	Pd-Pd	12.00	2.74±0.01	5.6±0.1	3.85±0.29	0.0030
UHL-Pd _{PM} /Fe ₂ O ₃	0.84	Pd-O	4.20	1.96±0.01	5.5±0.8	-8.54±0.94	0.0145
Ag foil	0.78	Ag-Ag	12.00	2.86±0.01	9.0±0.1	2.52±0.24	0.0028
UHL-Ag _{PM} /Fe ₂ O ₃	0.78	Ag-O	3.10	1.92±0.01	13.7±2.0	9.98±0.01	0.0181
Cd foil	0.80	Cd-Cd	6.00	2.95±0.01	14.7±0.5	-2.66±0.01	0.0014
		Cd-Cd	3.00	3.28±0.01	13.6±1.0		
UHL-Cd _{PM} /Fe ₂ O ₃	0.80	Cd-O	4.1	2.25±0.01	3.5±0.6	-1.03±0.01	0.0052
In foil	0.84	In-In	4.00	3.24±0.01	7.4±0.7	-9.31±0.01	0.0030
		In-In	3.00	3.46±0.01	8.3±0.1		
UHL-In _{PM} /Fe ₂ O ₃	0.84	In-O	3.91	2.17±0.01	10.8±0.2	8.83±0.08	0.0001

Sn foil	0.82	Sn-Sn	4.00	3.00±0.01	8.1±0.8	1.89±0.01	0.0028
UHL-Sn _{PM} /Fe ₂ O ₃	0.82	Sn-O	3.80	2.10±0.01	6.9±0.9	9.16±1.23	0.0021
Sb foil	0.78	Sb-Sb	3.00	2.90±0.01	4.4±0.3	7.02±0.54	0.0192
		Sb-Sb	3.00	3.34±0.01	10.6±2.3		
UHL-Sb _{PM} /Fe ₂ O ₃	0.78	Sb-O	3.72	1.98±0.01	2.0±0.8	9.84±0.01	0.0054
UHL-Ba _{PM} /Fe ₂ O ₃	0.85	Ba-O	7.98	2.87±0.01	4.4±0.9	-8.58±0.43	0.0054
UHL-La _{PM} /Fe ₂ O ₃	0.90	La-O	6.60	2.37±0.01	0.4±0.1	9.92±0.01	0.0046
		La-O	1.30	2.59±0.01	14.3±0.1		
UHL-Ce _{PM} /Fe ₂ O ₃	0.72	Ce-O	7.52	2.56±0.01	14.8±0.1	4.53±0.37	0.0064
UHL-Pr _{PM} /Fe ₂ O ₃	0.80	Pr-O	7.33	2.53±0.01	14.0±1.0	2.86±0.29	0.0037
UHL-Nd _{PM} /Fe ₂ O ₃	0.80	Nd-O	7.20	2.51±0.01	10.8±0.9	4.32±0.24	0.0035
UHL-Sm _{PM} /Fe ₂ O ₃	0.78	Sm-O	7.06	2.42±0.01	4.3±1.4	3.01±0.70	0.0100
Gd foil	0.98	Gd-Gd	6.00	2.95±0.01	9.0±0.1	-9.81±0.01	0.0128
UHL-Gd _{PM} /Fe ₂ O ₃	0.98	Gd-O	6.84	2.47±0.01	0.2±0.1	-9.81±0.01	0.0184
UHL-Tb _{PM} /Fe ₂ O ₃	0.90	Tb-O	6.79	2.28±0.01	14.3±4.2	-9.83±0.01	0.0011
UHL-Dy _{PM} /Fe ₂ O ₃	0.85	Dy-O	6.75	2.27±0.01	9.3±0.8	-8.08±0.68	0.0021
UHL-Er _{PM} /Fe ₂ O ₃	0.78	Er-O	6.71	2.25±0.01	0.8±0.1	-6.85±1.24	0.0105
UHL-Tm _{PM} /Fe ₂ O ₃	0.78	Tm-O	6.43	2.24±0.01	5.9±0.1	-8.20±0.46	0.0010
UHL-Yb _{PM} /Fe ₂ O ₃	0.98	Yb-O	6.21	2.48±0.01	4.1±0.7	9.87±0.01	0.0074
UHL-Lu _{PM} /Fe ₂ O ₃	0.80	Lu-O	6.02	2.26±0.01	8.9±0.9	4.61±0.51	0.0046
UHL-Hf _{PM} /Fe ₂ O ₃	0.81	Hf-O	5.82	2.05±0.01	2.8±1.0	0.78±0.01	0.0038
W foil	0.83	W-W	8.00	2.73±0.01	3.6±0.3	5.09±1.62	0.0082
		W-W	6.00	3.14±0.01	3.8±0.7		
UHL-W _{PM} /Fe ₂ O ₃	0.83	W-O	5.61	2.23±0.01	12.1±0.7	-9.73±0.01	0.0003
UHL-Os _{PM} /Fe ₂ O ₃	0.84	Os-O	5.26	1.96±0.01	14.8±0.1	-6.24±0.01	0.0091
Ir foil	0.95	Ir-Ir	12	2.71±0.01	3.5±1.7	9.60±3.88	0.0073
UHL-Ir/Fe ₂ O ₃	0.95	Ir-O	6.2	1.92±0.01	2.3±1.7	-3.35±2.40	0.0087
		Ir-O-Fe	5.2	3.33±0.01	11.0±0.1		
Pt foil	0.97	Pt-Pt	12.00	2.76±0.01	5.4±0.3	4.73±0.58	0.0040

UHL-Pt _{PM} /Fe ₂ O ₃	0.97	Pt-O	4.10	2.02±0.01	9.3±1.7	9.32±0.94	0.0164
Au foil	0.80	Au-Au	12.00	2.86±0.01	7.6±0.2	4.06±0.39	0.0052
UHL-Au _{PM} /Fe ₂ O ₃	0.80	Au-O	3.92	1.98±0.01	3.3±0.6	9.92±0.01	0.0077
Pb foil	0.81	Pb-Pb	2.00	3.41±0.01	12.4±2.0	-2.75±1.61	0.0029
UHL-Pb _{PM} /Fe ₂ O ₃	0.81	Pb-O	3.73	2.19±0.01	6.5±0.9	-1.73±0.74	0.0150
Bi foil	0.95	Bi-Bi	3.00	3.08±0.01	8.2±0.6	-0.04±0.01	0.0147
		Bi-Bi	2.00	3.69±0.01	12.3±3.5		
UHL-Bi _{PM} /Fe ₂ O ₃	0.95	Bi-O	3.61	2.14±0.01	12.5±0.4	-8.71±0.58	0.0049

CN: coordination numbers; *R*: bond distance; σ^2 : Debye-Waller factors; ΔE_0 : the inner potential correction. R factor: goodness of fit.

Supplementary Table 6 PEC performance comparisons between this work and typical Fe₂O₃-based photoanodes.

Photoanode	<i>J</i> (mA cm ⁻²) @1.23 V _{RHE}	ABPE (%)	Onset potential (V _{RHE})	Reference
5-M_{PM}/Fe₂O₃	5.70	1.2	0.54	This work
W_{PM}/Fe₂O₃	5.04	1.05	0.66	
SAs Pt:Fe ₂ O ₃ -O _v	3.65	0.68	0.63	Nat. Commun. 14, 2640 (2023).
Ge:Fe ₂ O ₃	3.5	---	0.70	Nat. Commun. 12, 4309 (2021).
NiFe(OH) _x /Ta:Fe ₂ O ₃ @Fe ₂ O ₃	3.22	0.55	0.55	Nat. Commun. 11, 4622 (2020).
Co-Pi/Ti:Fe ₂ O ₃	3.50	---	0.86	Nat. Commun. 10, 4832 (2019).
NiFeO _x /Dual-regrowth Fe ₂ O ₃	1.30	---	0.45	Nat. Commun. 6, 7447 (2015).
Co-Pi/Si:Fe ₂ O ₃	2.70	---	0.82	J. Am. Chem. Soc. 128, 15714 (2006).
Co-Pi/NPs Ag/Fe ₂ O ₃	4.68	0.55	0.70	Adv. Mater. 28, 6405 (2016).
Co-Pi/Zr:Fe ₂ O ₃	2.20	0.23	0.65	Angew. Chem. Int. Ed. 56, 4150 (2017).
FeOOH/Fe ₂ O ₃	1.21	---	0.65	Angew. Chem. Int. Ed. 55, 10854 (2016).
Ti-SiO _x /CoPi/Ti:Fe ₂ O ₃	3.19	---	0.80	Angew. Chem. Int. Ed. 55, 9922 (2016).
FeNiOOH/Fe ₂ TiO ₅ /Fe ₂ O ₃	2.20	---	0.95	Energy Environ. Sci. 10, 2124 (2017).
Co-Pi/P:Fe ₂ O ₃	3.10	---	0.75	Energy Environ. Sci. 8, 1231 (2015).
FeOOH/SnO ₂ -SiO _x /Ti:Fe ₂ O ₃	1.54	---	0.64	Adv. Energy Mater. 6, 1501840 (2016).
Fe ₂ TiO ₅ /Fe ₂ O ₃ /Pt	1.00	---	0.80	Adv. Funct. Mater. 27, 1703527 (2017).
Co-Pi/Co ₃ O ₄ /Ti:Fe ₂ O ₃	2.70	0.43	0.64	Adv. Funct. Mater. 29, 1801902 (2019).
Co-Mn/Fe ₂ O ₃	2.09	0.25	0.60	Adv. Funct. Mater. 29, 1904622 (2019).
FeOOH/Fe ₂ O ₃	0.85	---	0.80	Adv Funct Mater. 25, 2686 (2015).
Cobalt oxide/Sn:Fe ₂ O ₃	2.20	---	0.80	Nano Lett. 17, 2490 (2017).
Co-Pi/Fe ₂ O ₃ @Fe ₂ TiO ₅	2.60	---	0.90	ACS Nano. 9, 5348 (2015).

Supplementary Table 7 PEC performance comparisons between this work and other metal oxide-based photoanodes.

Photoanode	J (mA cm ⁻²) @1.23 V _{RHE}	ABPE (%)	Onset potential (V _{RHE})	Reference
5-M_{PM}/Fe₂O₃	5.70	1.2	0.54	This work
W_{PM}/Fe₂O₃	5.04	1.05	0.66	
Vw-WO ₃ /Co-Pi	4.8	1.2		Nat. Catal. 8 , 229-238 (2025).
WO ₃ -(PDDA-Aux) ₂	0.6	0.9		J. Am. Chem. Soc. 142 , 21899-21912 (2020).
Y:ZnO NR	0.4	---		Angew. Chem. Int. Ed. 57 , 9749-9753 (2018).
Fe ₂ TiO ₅ -TiO ₂	0.325	0.017		Angew. Chem. Int. Ed. Engl. 59 , 8128-8132 (2020).
CuO/WO ₃	3.2	---	0.6	Angew. Chem. Int. Ed. 60 , 17601(2021).
WO ₃ with dual W and O vacancy	2.8	---		Angew. Chem. Int. Ed. 55 , 11819-11823 (2016)
WO ₃ nanobar	2			Angew. Chem. Int. Ed. 60 , 9357-9361 (2021).
Ag@Ag ₃ (PO ₄) _{1-x} /ZnO	3.1	---	---	Energy Environ. Sci., 5 , 8917-8922 (2012).
m-WO ₃ with oxygen vacancy	2.4	---		Adv. Funct. Mater. 32 , 2204106 (2022).
In ₂ S ₃ -P/WO ₃	1.61	0.29		Adv. Energy Mater. 10 , 1903951 (2020).
WO ₃ porous nanosheet	1.25	---		Adv. Energy Mater. 6 , 1600437 (2016).
CoMnO/BBNO/WO ₃	3.56	---		Adv. Energy Mater. 8 , 1701655 (2018)
Co-Pi/GaN:ZnO	1.19	---		Adv. Energy Mater., 6 , 1600864 (2016).
Ni(OH) ₂ /ZIF-8/ZnO/NF	1.95	---		Adv. Energy Mater. 8 , 1800101 (2018).
N-TiO ₂	1.9	---		Nano Lett. 15 , 4692-4698 (2015).

Supplementary Table 8 Surface elemental weight ratios of the various electrodes determined by X-ray photoelectron spectroscopy.

	W (at.%)	Cu (at.%)	O (at.%)	
W _{PM} /1D-Cu ₂ O	4.02	59.99	35.99	
	W (wt%)	Cu (wt %)	O (wt %)	
W _{PM} /1D-Cu ₂ O	14.42	74.36	11.23	
	W (at.%)	Zn (at.%)	In (at.%)	S (at.%)
W _{PM} /2D-ZnIn ₂ S ₄	3.79	12.74	24.49	58.98
	W (wt%)	Zn (wt%)	In (wt%)	S (wt%)
W _{PM} /2D-ZnIn ₂ S ₄	11.18	13.36	45.13	30.34
	W (at.%)	C (at.%)	N (at.%)	
W _{PM} /2D-C ₃ N ₄	3.88	41.19	54.93	
	W (wt%)	C (wt%)	N (wt%)	
W _{PM} /2D-C ₃ N ₄	36.07	25.01	38.92	
	W (at.%)	Fe (at.%)	O (at.%)	
W _{PM} /2D-Fe ₂ O ₃	4.13	34.35	61.52	
	W (wt%)	Fe (wt%)	O (wt%)	
W _{PM} /2D-Fe ₂ O ₃	20.73	52.39	26.88	
	W (at.%)	Fe (at.%)	O (at.%)	
W _{PM} /3D-Fe ₂ O ₃	4.31	36.28	59.41	
	W (wt%)	Fe (wt%)	O (wt%)	
W _{PM} /3D-Fe ₂ O ₃	21.02	53.76	25.22	
	W (at.%)	Ti (at.%)	O (at.%)	
W _{PM} /3D-TiO ₂	3.84	30.05	66.11	
	W (wt%)	Ti (wt%)	O (wt%)	
W _{PM} /3D-TiO ₂	22.05	44.92	33.03	

Supplementary Table 9 Surface elemental weight ratios of the 5-M_{PM}/Fe₂O₃ electrode determined by X-ray photoelectron spectroscopy.

	Mo (at.%)	Rh (at.%)	W (at.%)	Ir (at.%)	Au (at.%)	Fe (at.%)	O (at.%)
5-M _{PM} /Fe ₂ O ₃	1.75	1.67	1.51	1.44	1.38	37.04	55.21
	Mo (wt%)	Rh (wt%)	W (wt%)	Ir (wt%)	Au (wt%)	Fe (wt%)	O (wt%)
5-M _{PM} /Fe ₂ O ₃	4.08	4.17	6.74	6.72	6.60	50.74	21.57

Supplementary Table 10 Structural parameters extracted from Au-*K* edge, W-*L*₃ edge, Mo-*L*₃ edge, Rh-*K* edge and Ir-*L*₃ edge $\chi(R)$ space spectra fitting of 5-M_{PM}/Fe₂O₃.

	S_0^2	shell	CN*	$R(\text{\AA})$	$\sigma^2 (10^{-3} \text{\AA}^2)$	ΔE_0	R factor
Au foil	0.75	Au-Au	12	2.86±0.01	7.3±0.1	4.54±0.49	0.0039
5-M _{PM} /Fe ₂ O ₃	0.75	Au-O	4.1	1.94±0.01	5.1±1.1	1.25±0.73	0.0036
		Au-O-Fe	2.2	3.61±0.01	13.0±6.4		
Ir foil	0.90	Ir-Ir	12	2.71±0.01	3.5±0.2	9.43±1.07	0.0074
Sample-Ir	0.90	Ir-O	4.1	1.90±0.01	1.6±1.1	-3.15±0.57	0.0039
		Ir-O-Fe	2.0	3.45±0.01	13.7±4.5		
Mo foil	0.75	Mo-Mo	8	2.71±0.01	2.6±0.2	-8.41±0.78	0.0022
		Mo-Mo	6	3.12±0.01	1.4±0.3		
Sample-Mo	0.75	Mo-O	3.6	1.72±0.01	9.4±1.4	9.50±1.56	0.0074
		Mo-O-Fe	1.3	2.79±0.01	14.2±3.5		
Rh foil	0.92	Rh-Rh	12	2.69±0.01	5.0±0.3	6.93±1.43	0.0170
Sample-Rh	0.92	Rh-O	3.6	2.08±0.01	5.2±1.6	7.08±0.89	0.0145
		Rh-O-Fe	1.8	3.50±0.01	14.3±0.1		
W foil	0.80	W-W	8	2.73±0.01	3.8±0.3	4.46±0.70	0.0042
		W-W	6	3.13±0.01	3.8±0.6		
Sample-W	0.80	W-O	4.3	1.72±0.01	2.5±1.2	-3.91±1.11	0.0069
		W-O-Fe	2.1	3.34±0.01	15.0±4.8		

CN: coordination numbers. R : bond distance. σ^2 : Debye-Waller factors. ΔE_0 : the inner potential correction. R factor: goodness of fit.

Supplementary Table 11 The PEC performance of this work is compared with that of C-N-based photoanodes currently reported.

Photoanode	$J @ 1.23 V_{RHE}$	Stability	Reference
5-M_{PM}/Fe₂O₃	5.70 mA cm⁻²	500 h	This work
PEDOT/C ₃ N ₄	14.7 $\mu A cm^{-2}$	150 s	<i>ACS Appl. Nano Mater.</i> 8, 985-996 (2025).
Pt-NCN-CC	$\sim 0.21 \mu A cm^{-2}$	24 h	<i>Int. J. Hydrog. Energy</i> 48, 29942-29951 (2023).
g-C ₃ N ₄ +420-g-C ₃ N ₄	0.16 mA cm ⁻²	400 s	<i>ChemCatChem</i> 11, 729 (2019).
V_C and V_N co-doped g-C ₃ N ₄	1.2 $\mu A cm^{-2}$	50 s	<i>Sol. RRL</i> 3, 1800298 (2019).
SnO ₂ /g-C ₃ N ₄	42 $\mu A cm^{-2}$	80 s	<i>Appl. Surf. Sci.</i> 440, 153-161 (2018).
TNTAs/g-C ₃ N ₄	0.86 mA cm ⁻²	10 h	<i>Nano-Micro Lett.</i> 10, 37 (2018).
C _{PVP} /g-C ₃ N ₄ -60	12.6 $\mu A cm^{-2}$ at 1.2 V	4000 s	<i>Chem. Eur. J.</i> 23, 419 (2017).
Pt/ZnO/g-C ₃ N ₄	30 $\mu A cm^{-2}$ at 1.12 V_{RHE}	3400 s	<i>Int. J. Hydrog. Energy</i> 40, 9080-9087 (2015).
MoS ₂ /NRGO/CNNS	27.76 $\mu A cm^{-2}$ at 0.8 V	1400 s	<i>Adv. Mater.</i> 25, 6291-6297 (2013).

Supplementary Table 12 Bader charge quantification results for 5-M_M/Fe₂O₃ and 5-M_{PM}/Fe₂O₃.

	5-M _M /Fe ₂ O ₃	5-M _{PM} /Fe ₂ O ₃
Ir	9.22 eV	7.48 eV
Ir	9.18 eV	7.54 eV
W	13.33 eV	12.33 eV
W	13.88 eV	12.57 eV
Mo	13.28 eV	12.46 eV
Mo	13.64 eV	12.30 eV
Au	11.47 eV	10.08 eV
Au	11.51 eV	10.62 eV
Rh	14.76 eV	14.69 eV
Rh	14.73 eV	14.60 eV

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