

Supplementary information

Understanding the deformation of IRMOF-1 during
carbon dioxide and argon adsorption from hybrid
Monte Carlo/molecular dynamics simulations

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Force Field

The force field proposed by Dubbeldam et al. [1] for IRMOF-1 (Fig. 1) represents the Zn_4O cluster using only Lennard-Jones and Coulombic potentials between the individual atoms. In contrast, linker molecules are described using a combination of general force field parameters, including DREIDING [2] and CVFF [3], for bond, bend, and torsion constants:

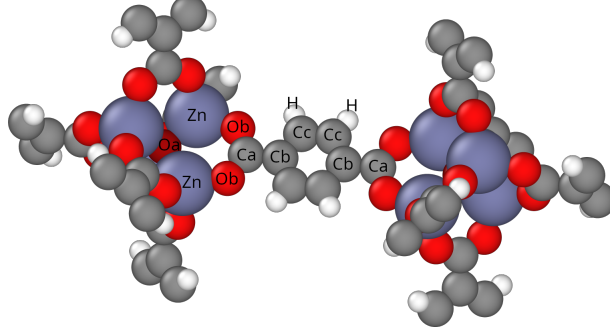


Fig. 1 Definition of the different atom types present in IRMOF-1.

The bond stretching between two bonded atoms is represented by a harmonic potential:

$$V_b(r_{ij}) = \frac{1}{2}k_b(r_{ij} - r_0)^2 \quad (1)$$

The bond-angle vibration between a triplet of atoms is also described by a harmonic potential:

$$V_a(\theta_{ijk}) = \frac{1}{2}k_\theta(\theta_{ijk} - \theta_0)^2 \quad (2)$$

Proper and improper dihedrals are represented by a periodic function:

$$V_d(\phi_{ijkl}) = k_\phi(1 + \cos(m\phi - \phi_s)) \quad (3)$$

where ϕ is the angle between the ijk and the jkl planes. The only difference here compared to what Dubbeldam et al. [1] described is that in the definition of improper dihedral angles of atoms $i-j-k-l$, i is the central atom. Additionally, the Lennard-Jones and Coulomb 1-4 interactions were included (with full values).

Table 1 Bond parameters [1]

<i>i</i>	<i>j</i>	k_b (kJ mol ⁻¹ nm ⁻²)	r_0 (nm)
Ob	Ca	452173.73	0.127
Ca	Cb	294124.11	0.144
Cb	Cc	401932.76	0.1365
Cc	Cc	401932.76	0.1365
Cc	H	304309.33	0.095

Table 2 Bend parameters [1]

<i>i</i>	<i>j</i>	<i>k</i>	k_θ (kJ mol ⁻¹ rad ⁻²)	θ_0 (deg)
Ob	Ca	Ob	1130.43	132
Ob	Ca	Cb	456.31	114
Ca	Cb	Cc	290.399	120
Cc	Cb	Cc	753.62	120
Cc	Cc	Cb	753.62	120
Cb	Cc	H	309.82	120
Cc	Cc	H	309.82	120

Table 3 Torsion parameters [1]

<i>i</i>	<i>j</i>	<i>k</i>	<i>l</i>	k_ϕ (kJ mol ⁻¹)	ϕ_s (deg)	<i>m</i>
Ob	Ca	Cb	Cc	10.467	180	2
Ca	Cb	Cc	H	12.560	180	2
Ca	Cb	Cc	Cc	12.560	180	2
Cb	Cc	Cc	H	12.560	180	2
Cb	Cc	Cc	Cb	12.560	180	2
Cc	Cb	Cc	Cc	12.560	180	2
H	Cc	Cb	Cc	12.560	180	2
H	Cc	Cc	H	12.560	180	2

Table 4 Improper torsion parameters [1]

<i>i</i>	<i>j</i>	<i>k</i>	<i>l</i>	k_ϕ (kJ mol ⁻¹)	ϕ_s	<i>m</i>
Ca	Cb	Ob	Ob	41.872	180	2
Cb	Cc	Cc	Ca	41.88	180	2
Cc	Cb	Cc	H	1.549	180	2

Table 5 Non-bonded parameters for IRMOF-1 [1]

Atom type	ϵ/k_B (K)	σ (Å)	Charge
Zn	0.42	2.7	1.275
Oa	700	2.98	-1.5
Ob	70.5	3.11	-0.6
Ca	47	3.74	0.475
Cb	47.86	3.47	0.125
Cc	47.86	3.47	-0.15
H	7.65	2.85	0.15

For unlike LJ interactions, the standard Lorentz-Bethelot combining rules are used.

Table 6 Non-bonded parameters for CO₂ [4] and Ar [5]

Atom type	ϵ/k_B (K)	σ (Å)	Charge
C _{CO₂}	27.0	2.80	0.7
O _{CO₂}	70.0	3.05	-0.35
Ar	119.8	3.40	—

For unlike LJ interactions, the standard Lorentz-Bethelot combining rules are used.

Additional results

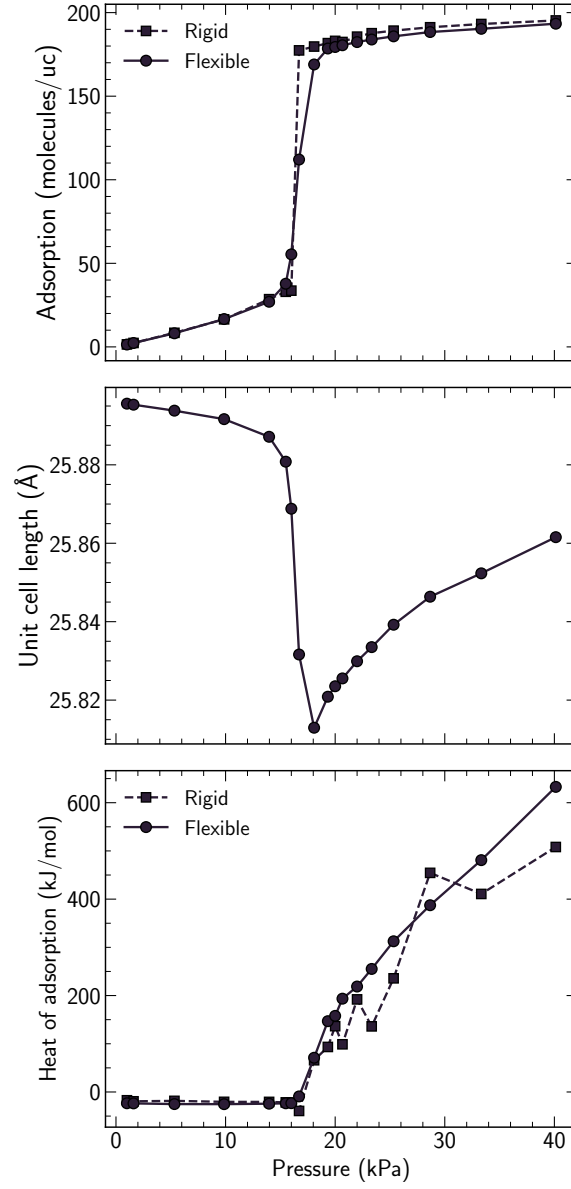


Fig. 2 Adsorption isotherms, strain isotherms, and heats of adsorption of CO₂ on IRMOF-1 at 195 K for $1 \times 1 \times 1$ cells and 12 Å of cutoff radius.

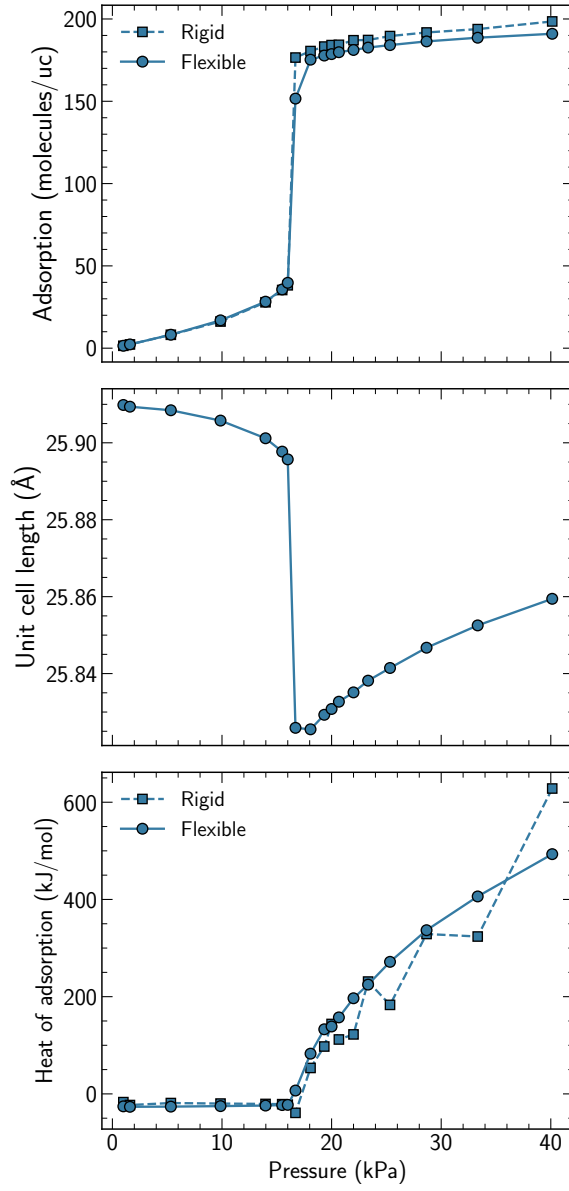


Fig. 3 Adsorption isotherms, strain isotherms, and heats of adsorption of CO₂ on IRMOF-1 at 195 K for $2 \times 2 \times 2$ cells and 12 Å of cutoff radius.

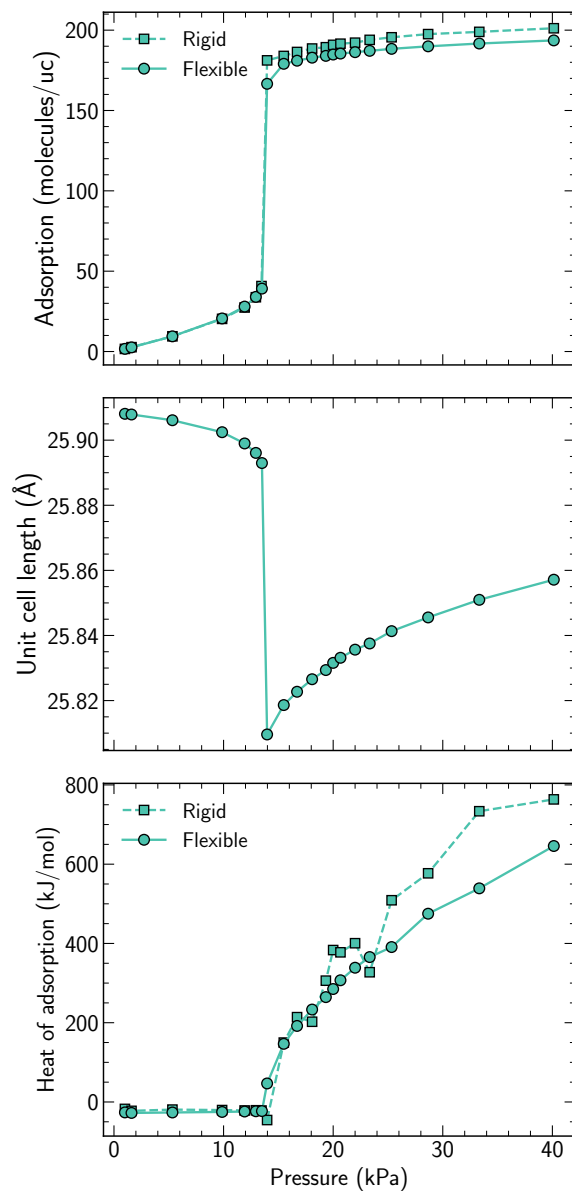


Fig. 4 Adsorption isotherms, strain isotherms, and heats of adsorption of CO₂ on IRMOF-1 at 195 K for $2 \times 2 \times 2$ cells and 14 Å of cutoff radius.

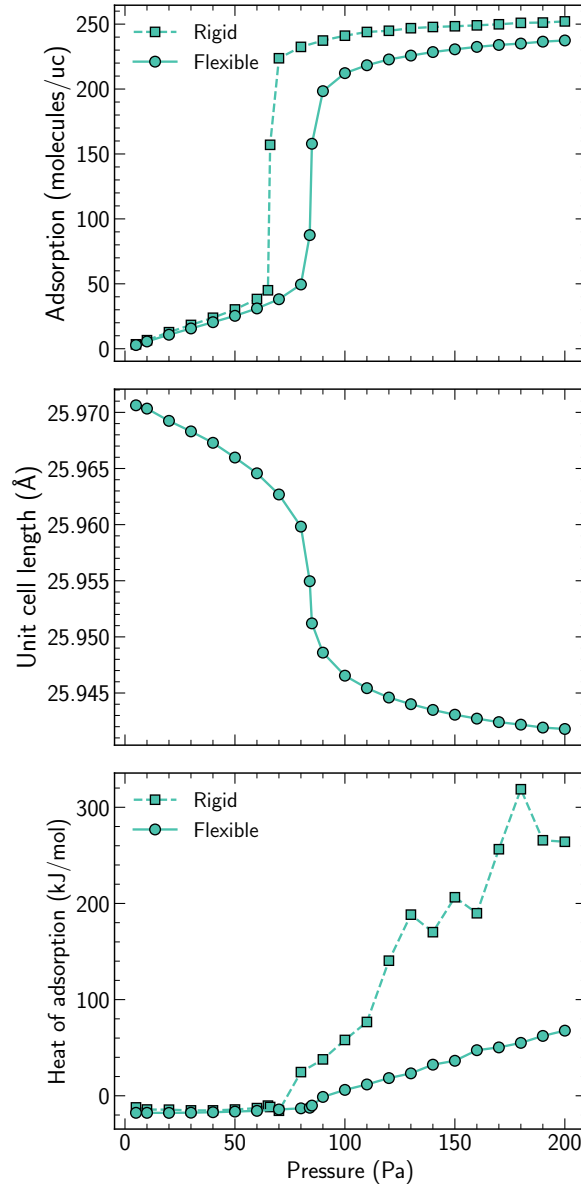


Fig. 5 Adsorption isotherms, strain isotherms, and heats of adsorption of Ar on IRMOF-1 at 78 K for $2 \times 2 \times 2$ cells and 14 Å of cutoff radius.

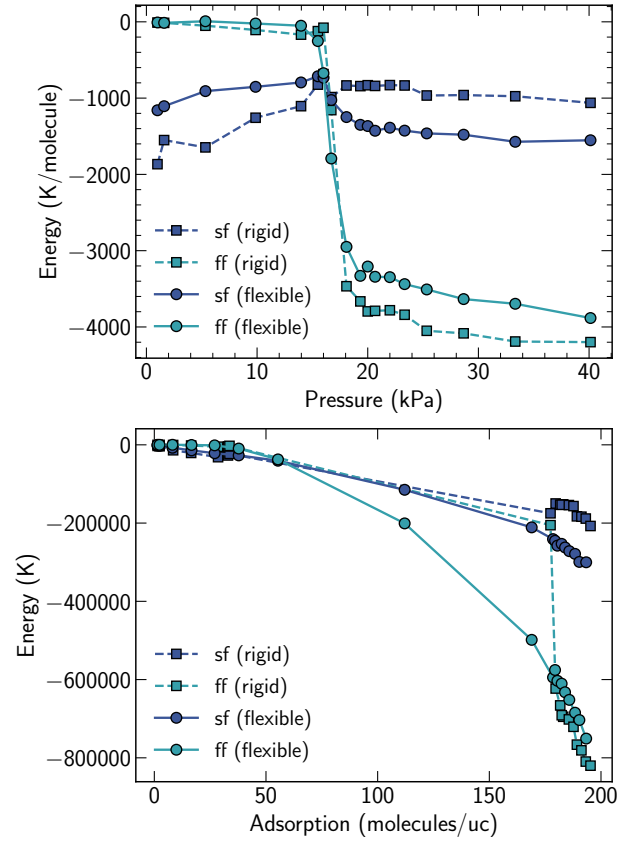


Fig. 6 Fluid-fluid and solid-fluid energy contributions for CO₂ adsorption on IRMOF-1 at 195 K ($1 \times 1 \times 1$ cells and 12 Å of cutoff radius).

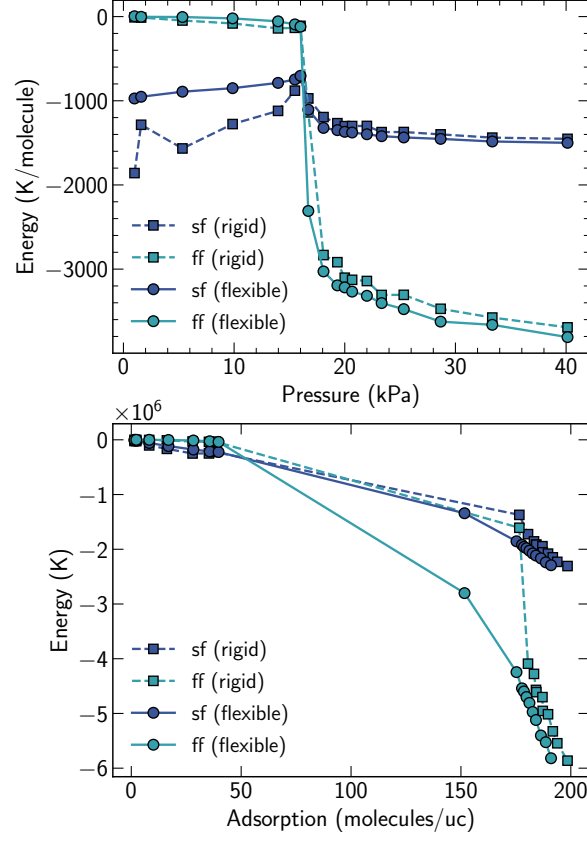


Fig. 7 Fluid-fluid and solid-fluid energy contributions for CO₂ adsorption on IRMOF-1 at 195 K ($2 \times 2 \times 2$ cells and 12 Å of cutoff radius).

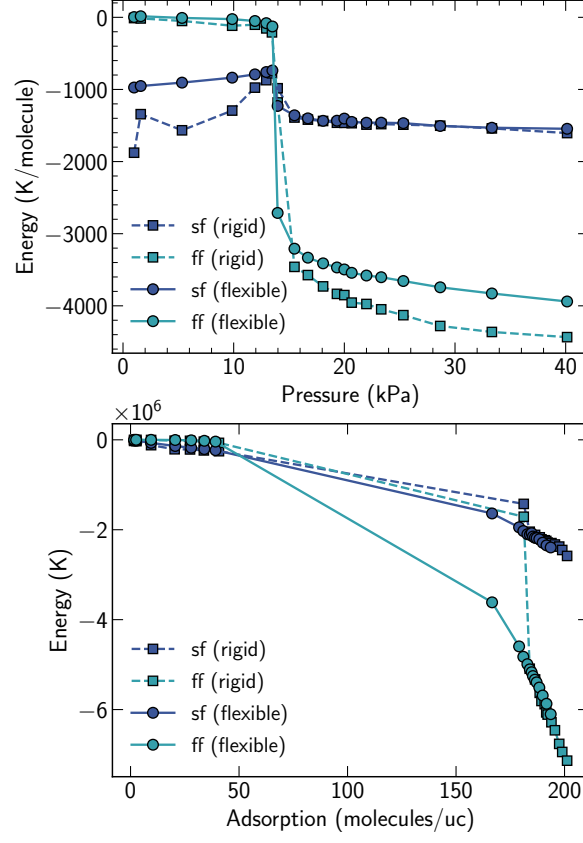


Fig. 8 Fluid-fluid and solid-fluid energy contributions for CO₂ adsorption on IRMOF-1 at 195 K ($2 \times 2 \times 2$ cells and 14 Å of cutoff radius).

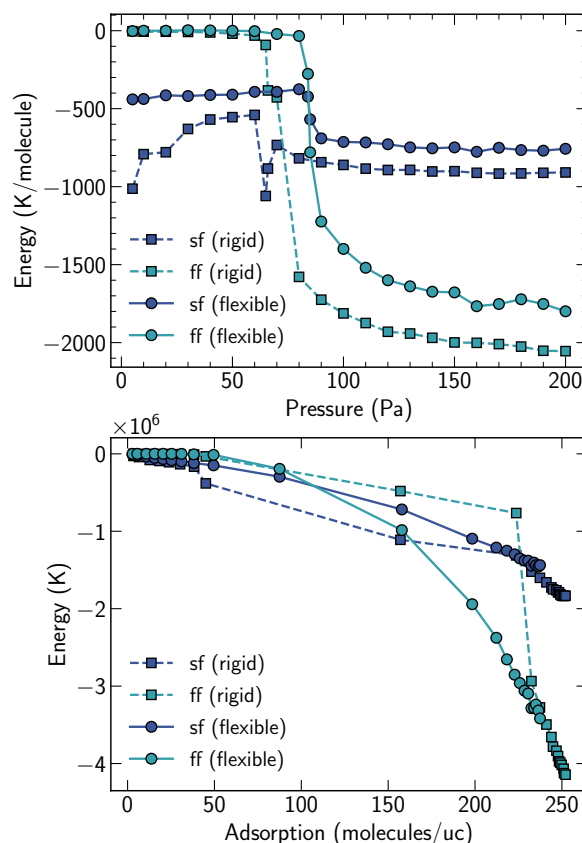


Fig. 9 Fluid-fluid and solid-fluid energy contributions for Ar adsorption on IRMOF-1 at 78 K ($2 \times 2 \times 2$ cells and 14 Å of cutoff radius).

References

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