

**Supplementary Information for**

**Dynamic Chirality Beyond Configuration:  
Reversible Exchange of Chirality Type**

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## 1) General Methods

For manipulations that were carried out in an argon-filled glovebox or using standard Schlenk techniques, glassware was dried overnight at 150 °C in an oven or by flame under vacuum. Dry solvents were received using an MBraun Solvent Purification System or distilled from sodium wire and benzophenone or calcium hydride, as appropriate, degassed by freeze-pump-thaw, and stored over activated 3Å molecular sieves prior to use. Thin-layer chromatography was carried out on aluminum-backed silica gel 60 (F254) plates from MERCK (grain-size distribution 60/20 µm) and visualized using UV light. Column chromatography was performed with silica gel (spherical, particle size 60 µm, neutral) purchased from Silicycle or Sigma Aldrich.

NMR spectra were recorded on Bruker (300 MHz, 400 MHz), Varian (500 MHz), or Joel (400 MHz, 500 MHz) spectrometers at 25 °C. <sup>1</sup>H NMR chemical shifts are reported relative to TMS (δ in ppm) and were referenced via residual proton resonances of the corresponding deuterated solvent (CHCl<sub>3</sub>: 7.26 ppm; C<sub>6</sub>D<sub>5</sub>H : 7.16 ppm; DCM-d<sub>1</sub>: 5.32 ppm; THF-d<sub>7</sub>: 3.58 ppm; ACN-d<sub>2</sub>: 1.94 ppm; DMSO-d<sub>5</sub>: 2.50 ppm) whereas <sup>13</sup>C{<sup>1</sup>H} NMR spectra are reported relative to TMS using the natural-abundance carbon resonances (CDCl<sub>3</sub>: 77.16 ppm; C<sub>6</sub>D<sub>6</sub>: 128.06 ppm; DCM-d<sub>2</sub>: 53.84 ppm; THF-d<sub>8</sub>: 67.21 ppm; ACN-d<sub>3</sub>: 1.32 ppm; DMSO-d<sub>6</sub>: 39.52 ppm).

Electron paramagnetic resonance (EPR) measurements were carried out with a Bruker 300 spectrometer (X-band, 100 kHz field modulation).

Preparative chiral HPLC were performed on an Agilent 1260 Infinity unit (pump G1311C, autosampler G1329B, DAD G1365D and fraction collector G1364C), monitored by Agilent OpenLAB CDS Chemstation LC at Aix Marseille University.

| Trade name            | Chiral stationary phase                                              | Seller*    |
|-----------------------|----------------------------------------------------------------------|------------|
| <b>Chiralcel OD-3</b> | Cellulose tris(3,5-dimethylphenylcarbamate) coated on 3 µm silica    | Daicel     |
| <b>Lux-Amylose-1</b>  | Amylose tris(3,5-dimethylphenylcarbamate) immobilized on 5 µm silica | Phenomenex |

\*Phenomenex: Danaher Corporation (Torrance, USA); Daicel: Daicel Corporation (Osaka, Japan)

High Resolution Mass Spectrometry (HRMS) were recorded on a Waters QToF-I spectrometer using ESI at the Centre Régional de Mesures Physiques de l'Ouest (CRMPO), Université de Rennes 1 as well as the UC San Diego Mass Spectrometry Laboratory on an Agilent 6230 Accurate-Mass TOF-MS spectrometer.

Single crystal X-ray diffraction data were collected on Bruker Apex or D8-Venture diffractometers using Mo-Kα radiation (λ = 0.71073 Å) or Cu-Kα radiation (λ = 1.54178 Å).

UV-Visible absorption spectra were recorded on a Jasco V-770 spectrophotometer.

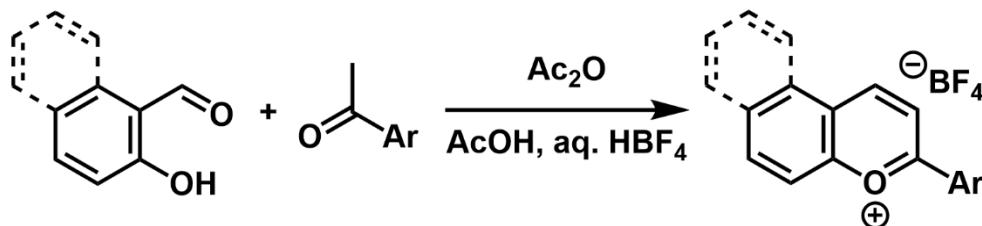
Electronic Circular Dichroism (ECD) were measured on a JASCO J-815 spectrometer equipped with a JASCO Peltier cell holder PTC-423 to maintain the temperature at  $25.0 \pm 0.2^\circ\text{C}$ . A quartz cell with a 1 cm optical path length was used. The CD spectrometer was purged with nitrogen before recording each spectrum, which was baseline subtracted. The baseline was always measured for the same solvent and in the same cell as the samples. Acquisition parameters: 0.5 nm interval, scanning speed 50 nm/min, band width 2 nm, and 3 accumulations per sample. Spectra are presented without smoothing or further data processing.

Mesityl acetophenone<sup>1</sup> and chiral CAAC ***rac-2***<sup>2</sup> were prepared according to known literature procedures. All other starting materials were purchased from commercial sources and used without further purification.



## 2) Synthetic Procedures

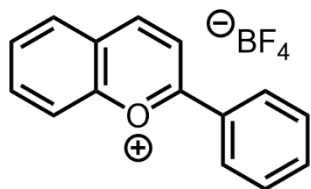
### Preparation of chromenylium tetrafluoroborate salts 1a-b



### General Procedure

A round-bottom flask was charged with ketone (1 eq.), 2-hydroxy aryl aldehyde (1 eq.), and a magnetic stir bar. AcOH (1.5 mL per mmol) was added, followed by 50% aq. HBF<sub>4</sub> (0.3 mL per mmol). Acetic anhydride (1.5 mL per mmol) was slowly added via dropping funnel over 30 minutes. After the addition was complete, the dropping funnel was replaced with an air condenser. The reaction mixture was stirred at the indicated temperature for the indicated time, then poured into 500 mL of Et<sub>2</sub>O. After stirring for 30 minutes, the suspension was filtered and the solid washed with 500 mL of Et<sub>2</sub>O. Drying the solid under vacuum at 80 °C for 2 days afforded dry product, which was used without further purification.

### 2-phenylchromenylium tetrafluoroborate [1a]:



Prepared according to the general procedure from acetophenone (2.41 g, 20.1 mmol) and salicylaldehyde (2.45 g, 20.1 mmol). Room temperature, 16 h. Deep blue-green powder, 2.07 g (7.04 mmol, 35% yield).

**<sup>1</sup>H NMR (500 MHz, 25°C, ACN-d<sub>3</sub>)** δ = 9.50 (d, J = 8.9 Hz, 1H), 8.72 (d, J = 8.9 Hz, 1H), 8.59 – 8.55 (m, 2H), 8.40 – 8.36 (m, 3H), 8.04 (ddd, J = 8.2, 5.7, 2.6 Hz, 1H), 7.97 (t, J = 7.5 Hz, 1H), 7.84 – 7.80 (m, 2H).

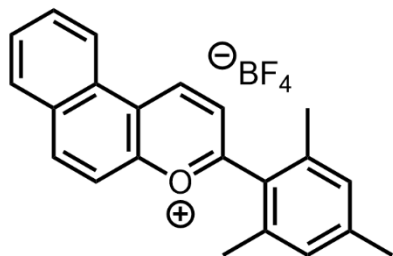
**<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, 25°C, ACN-d<sub>3</sub>)** δ = 176.82, 158.90, 157.66, 141.28, 138.78, 131.71, 131.63, 131.45, 131.32, 129.67, 125.75, 120.33, 118.95.

**<sup>19</sup>F NMR (282 MHz, 25°C, ACN-d<sub>3</sub>)** δ = -151.10 (small), -151.15.

**HRMS:** calc. for M<sup>+</sup> C<sub>15</sub>H<sub>11</sub>O<sup>+</sup> m/z: 207.08044; found 207.0805 (0 ppm).

Spectral data matched those reported previously.<sup>3</sup>

**3-mesitylbenzo[*f*]chromenylium tetrafluoroborate [1b]:**



Prepared according to the general procedure from mesityl acetophenone (2.52 g, 14.6 mmol) and 2-hydroxynaphthaldehyde (2.48 g, 15.3 mmol). Reflux, 5 h. Brown-orange powder, 2.77 g (7.17 mmol, 49% yield).

**<sup>1</sup>H NMR (300 MHz, 25°C, ACN-d<sub>3</sub>)** δ = 10.21 (d, J = 8.7 Hz, 1H), 8.97 (d, J = 8.3 Hz, 1H), 8.87 (d, J = 9.3 Hz, 1H), 8.45 (d, J = 8.7 Hz, 1H), 8.35 (dd, J = 8.0, 1.2 Hz, 1H), 8.24 (dd, J = 9.2, 0.8 Hz, 1H), 8.13 (ddd, J = 8.3, 7.2, 1.4 Hz, 1H), 8.05 (ddd, J = 8.0, 7.3, 1.2 Hz, 1H), 7.04 (s, 2H), 2.42 (s, 3H), 2.38 (s, 6H).

**<sup>13</sup>C{<sup>1</sup>H} NMR (101 MHz, 25°C, ACN-d<sub>3</sub>)** δ = 176.39, 162.11, 152.74, 145.40, 145.33, 139.93, 132.59, 132.33, 131.67, 131.33, 130.97, 128.63, 127.81, 125.07, 124.98, 124.27, 118.16, 21.41, 20.65.

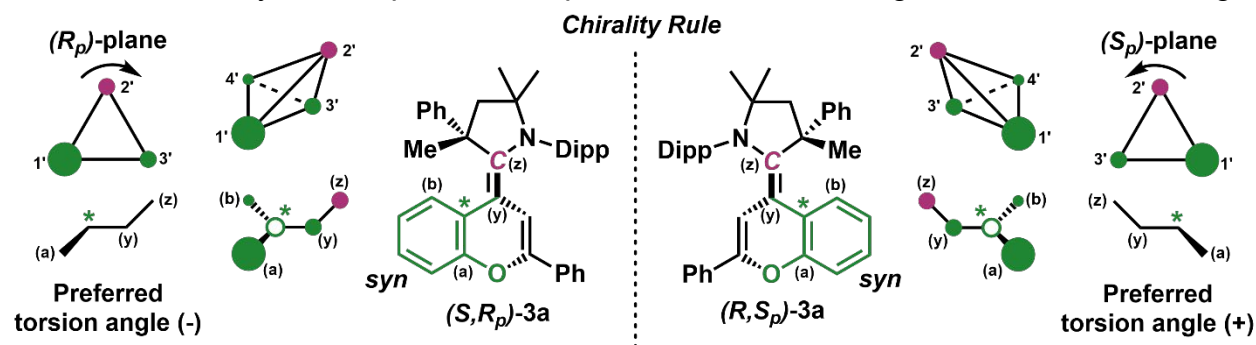
**<sup>19</sup>F NMR (282 MHz, 25°C, ACN-d<sub>3</sub>)** δ = -151.10 (small), -151.15.

**HRMS:** calc. for M<sup>+</sup> C<sub>22</sub>H<sub>19</sub>O<sup>+</sup> m/z: 299.14304; found 299.1430 (0 ppm).

## Assignment of Absolute Configuration – Planar Chirality

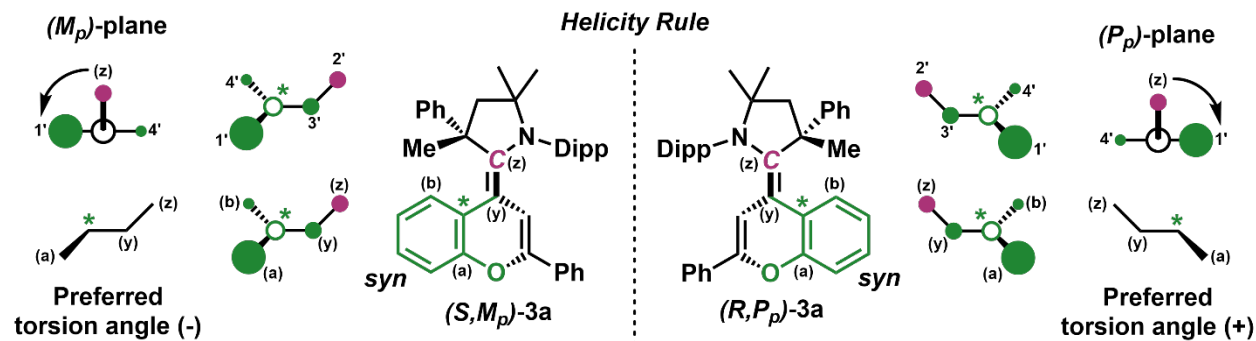
The absolute configurations of planar chiral overcrowded alkenes were assigned according to the Cahn-Ingold-Prelog priority system.<sup>4,5</sup> Below, we derive both chirality and helicity descriptors, although helicity descriptors are preferred and used throughout the main text. Readers are referred to the “Blue Book,” P-92.1.1-3, as well as section 2.5.2 of Prelog and Helmchen, *ACIE*, **1982**, 21, 567-583 for further details.

**Chirality rule:** Using **3a** as an example, we derive the enantiomorphous five-point figures,  $C_s$ -tetrahedra, and the corresponding preferred torsion angles (Figure S1). Note, because the relative priorities of (a),  $C^*$ , (y), and (z) are the same in both **3a** and **3b**, the relationship between chirality descriptor and preferred torsion angle does not change.



**Figure S1.** Determining chirality descriptors for planar chiral overcrowded alkenes.

**Helicity rule:** Viewing the five-point figures along the '(y)-\*' axis (Figure S2), the sense of helicity is determined by rotating (z) towards the highest priority substituent, (a). Counterclockwise rotation indicates a left-handed helix,  $M_p$ , with a negative preferred torsion angle. Conversely, clockwise rotation indicates a right-handed helix,  $P_p$ , with a positive preferred torsion angle. As for the chirality rule, the relationship between helicity descriptor and preferred torsion angle does not change for **3b**.



**Figure S2.** Determining helicity descriptors for planar chiral overcrowded alkenes.

From these derivations, the equivalence between  $R_p$  and  $M_p$ ,  $S_p$  and  $P_p$ , is clear. Note, in the context of the overcrowded alkenes from this work, a correlation between the sense

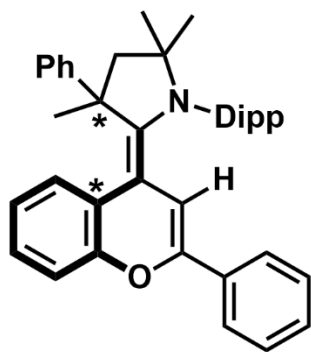
of helicity and the orientation (*syn* or *anti*) of the chromene ring relative to the phenyl substituent of the stereogenic  $\alpha$ -carbon center is observed: When the stereogenic  $\alpha$ -carbon is *S*, *syn* = *R<sub>p</sub>* = *M<sub>p</sub>*, while *anti* = *S<sub>p</sub>* = *P<sub>p</sub>*. A change in orientation is equivalent to configurational inversion of the stereogenic plane.

## Preparation of planar chiral overcrowded alkenes *rac*-3a-b

### General Procedure

In a glovebox, a THF solution of chiral CAAC *rac*-2 (2 eq.) was prepared at room temperature according to a previously described procedure <sup>2</sup>. In one portion, the appropriate chromenylium tetrafluoroborate (1 eq.) was added, resulting in a deeply colored suspension. The reaction vessel was sealed and stirred at room temperature overnight. After 16 h, the flask was removed from the glovebox and unsealed, and the reaction was vigorously stirred while diluting five-fold with hexanes. The resulting suspension was filtered through a pad of dry silica (ht. = 4 cm, Ø = 5 cm), eluting with Et<sub>2</sub>O. Removal of volatiles under vacuum yielded the alkene, which was further purified by flash column chromatography if necessary.

### Alkene *rac*-3a:



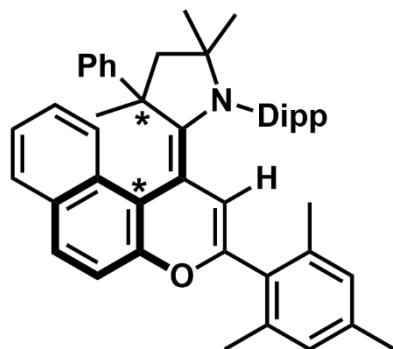
Prepared according to the general procedure from **1a** (738 mg, 2.51 mmol). Gradient of 0% to 20% Et<sub>2</sub>O in hexanes. Yellow-orange foam, 1.25 g (2.26 mmol, 90% yield). Single crystals for XRD analysis were obtained by slow evaporation of a solution in hexanes.

**<sup>1</sup>H NMR (400 MHz, 25°C, C<sub>6</sub>D<sub>6</sub>)**  $\delta$  = 7.74 (d, *J* = 8.3 Hz, 2H), 7.24 (t, *J* = 7.9 Hz, 2H), 7.22 – 7.18 (m, 5H), 7.10 – 7.05 (m, 2H), 7.04 (dd, *J* = 8.0, 1.5 Hz, 1H) 6.99 (dd, *J* = 6.3, 3.1 Hz, 1H), 6.79 (ddd, *J* = 8.0, 7.2, 1.5 Hz, 1H), 6.55 (ddd, *J* = 8.6, 7.2, 1.5 Hz, 1H), 6.24 (s, 1H), 3.86 (sept, *J* = 6.8 Hz, 1H), 3.12 (sept, *J* = 6.8 Hz, 1H), 2.23 (s, 2H), 1.91 (s, 3H), 1.36 (d, *J* = 6.8 Hz, 3H), 1.31 (d, *J* = 6.8 Hz, 3H), 1.24 (d, *J* = 6.8 Hz, 3H), 1.13 (d, *J* = 6.8 Hz, 3H), 1.12 (s, 3H), 0.73 (s, 3H). *One aryl peak overlaps with solvent.*

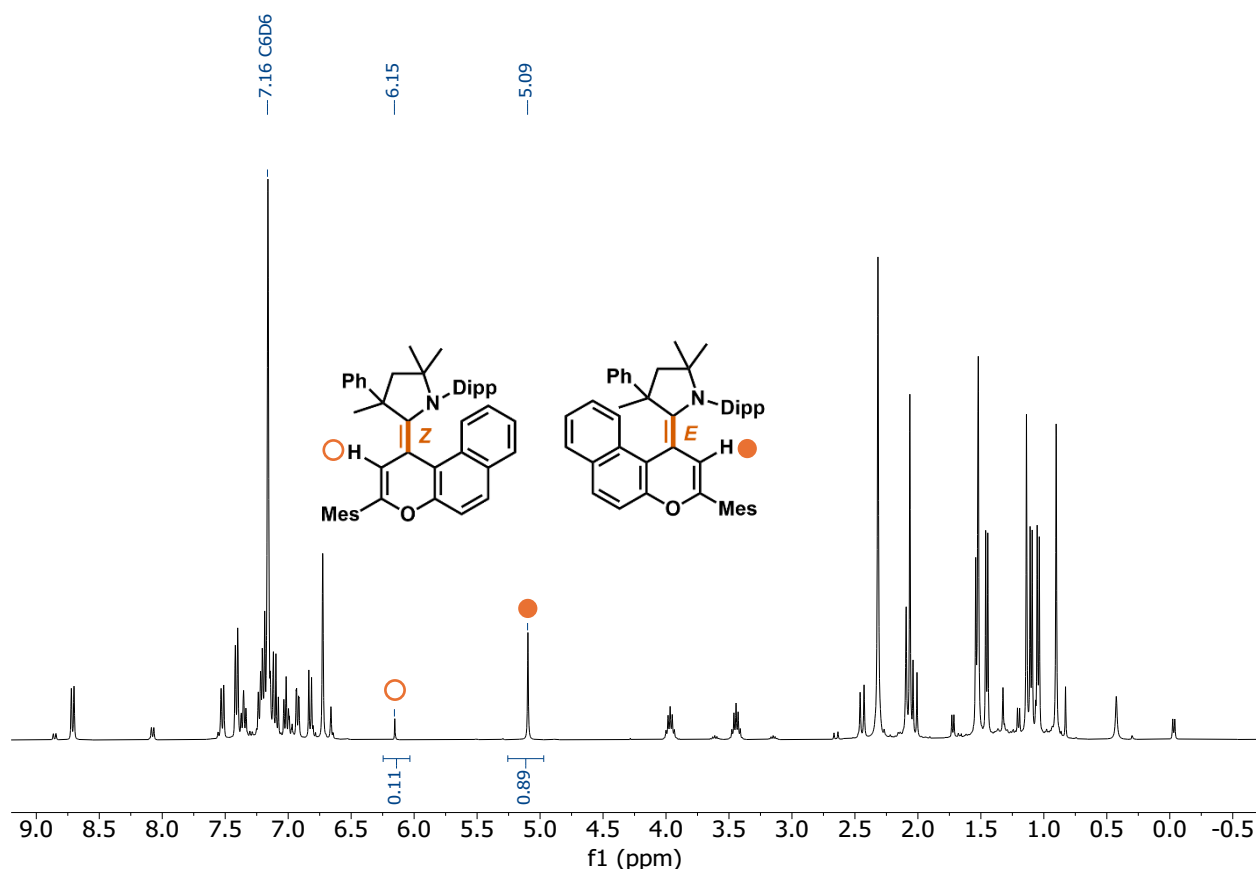
**<sup>13</sup>C{<sup>1</sup>H} NMR (101 MHz, 25°C, C<sub>6</sub>D<sub>6</sub>)**  $\delta$  = 155.20, 150.20, 148.47, 147.92, 146.86, 146.61, 141.25, 135.13, 129.41, 128.66, 128.57, 127.54, 126.50, 126.36, 126.00, 125.80, 125.13, 124.82, 122.13, 116.59, 107.47, 99.10, 64.10, 62.39, 51.16, 29.97, 29.32, 28.88, 28.65, 28.45, 25.59, 24.88, 24.40, 23.43.

**HRMS:** calc. for M<sup>+</sup> C<sub>40</sub>H<sub>43</sub>NO<sup>+</sup> *m/z*: 553.33392; found 553.3336 (1 ppm)

**Alkene *rac*-3b:**



Prepared according to the general procedure from **1c** (1.158 g, 3.00 mmol). Gradient of 0% to 10% Et<sub>2</sub>O in petroleum ether. Yellow-orange foam, 1.328 g (2.06 mmol, 69% yield). NMR analysis indicates a 9:1 mixture of E:Z isomers (Figure S3). Heating a slurry in hexanes at 60°C overnight, followed by removal of volatiles under vacuum, gave the E isomer in quantitative yield. Single crystals for XRD analysis were obtained by slow evaporation of a solution in petroleum ether.



**Figure S3.** <sup>1</sup>H NMR (400 MHz, 25°C, C<sub>6</sub>D<sub>6</sub>) spectrum of initially obtained 9:1 E:Z mixture of *rac*-3b.

**<sup>1</sup>H NMR (400 MHz, 25°C, THF-*d*<sub>8</sub>)** δ = 8.50 (d, *J* = 8.4 Hz, 1H), 7.60 (d, *J* = 7.2 Hz, 1H), 7.36 – 7.29 (m, 5H), 7.27 – 7.22 (m, 2H), 7.10 – 7.06 (m, 3H), 7.00 (t, *J* = 7.3 Hz, 1H), 6.72 (s, 2H), 6.68 (d, *J* = 8.7 Hz, 1H), 4.85 (s, 1H), 3.87 (sept, *J* = 6.6 Hz, 1H), 3.32 (sept, *J* = 6.7 Hz, 1H), 2.64 (d, *J* = 12.5 Hz, 1H), 2.29 (d, *J* = 12.5 Hz, 1H), 2.18 (s, 3H), 2.11 (s,

6H), 1.46 (d, J = 6.6 Hz, 3H), 1.38 (d, J = 6.5 Hz, 3H), 1.34 (s, 3H), 1.27 (s, 3H), 1.12 (d, J = 6.5 Hz, 3H), 0.97 (s, 3H), 0.83 (d, J = 6.6 Hz, 3H).

**$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz, 25°C, THF- $d_8$ )**  $\delta$  = 155.36, 153.52, 151.85, 149.88, 149.62, 148.30, 138.48, 137.31, 136.89, 134.77, 132.50, 132.29, 129.45, 128.86, 128.38, 128.33, 127.83, 127.24, 126.58, 126.34, 126.29, 126.12, 124.35, 122.78, 117.77, 115.03, 96.12, 66.06, 58.84, 53.18, 31.72, 30.23, 30.18, 29.88, 29.43, 26.92, 25.49, 21.93, 21.25.

**HRMS:** calc. for  $\text{M}^+ \text{C}_{47}\text{H}_{51}\text{NO}^+$  m/z: 645.39652; found 645.3970 (1 ppm).

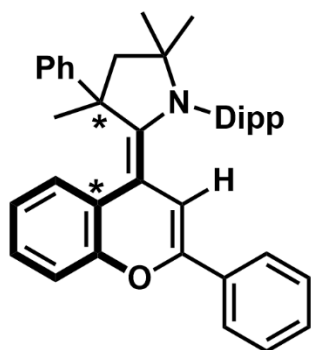
## Enantiomeric Resolution of Alkenes by Chiral HPLC

Absolute configuration was assigned based on the ECD spectra of received products, which was analyzed in conjunction with computationally simulated ECD spectra. Specifically, the signs of the experimentally observed and computationally simulated exciton couplets were compared (See Figures S33 – S34).

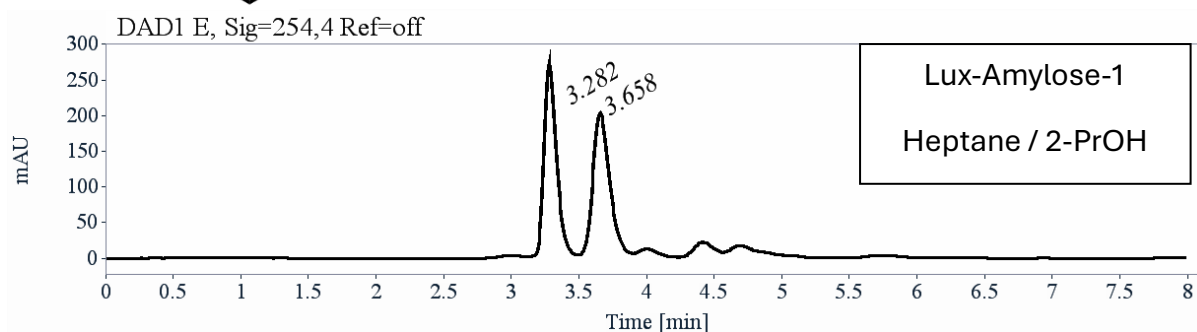
| Order | AbsConfig                     | Mass (mg) | Yield (%) | ee (%) |
|-------|-------------------------------|-----------|-----------|--------|
| 1st   | ( <i>S,M<sub>p</sub></i> )-3a | 70        | 35        | >99.5  |
| 2nd   | ( <i>R,P<sub>p</sub></i> )-3a | 67        | 34        | >98.5  |
| 1st   | ( <i>R,M<sub>p</sub></i> )-3b | 62        | 32        | >99.5  |
| 2nd   | ( <i>S,P<sub>p</sub></i> )-3b | 53        | 27        | >99.5  |

**Table S1.** Obtained masses, yields, and enantiopurities for **3a-b** following preparative chiral HPLC.

## Analytical chiral HPLC separation for compound *rac*-3a



The sample is dissolved in heptane, injected on the chiral column, and detected with a UV detector at 254 nm. The flow-rate is 1 mL/min.



| RT [min] | Area | Area%  | Capacity Factor | Enantioselectivity | Resolution (USP) |
|----------|------|--------|-----------------|--------------------|------------------|
| 3.28     | 1724 | 50.26  | 0.11            |                    |                  |
| 3.66     | 1707 | 49.74  | 0.24            | 2.13               | 1.93             |
| Sum      | 3431 | 100.00 |                 |                    |                  |

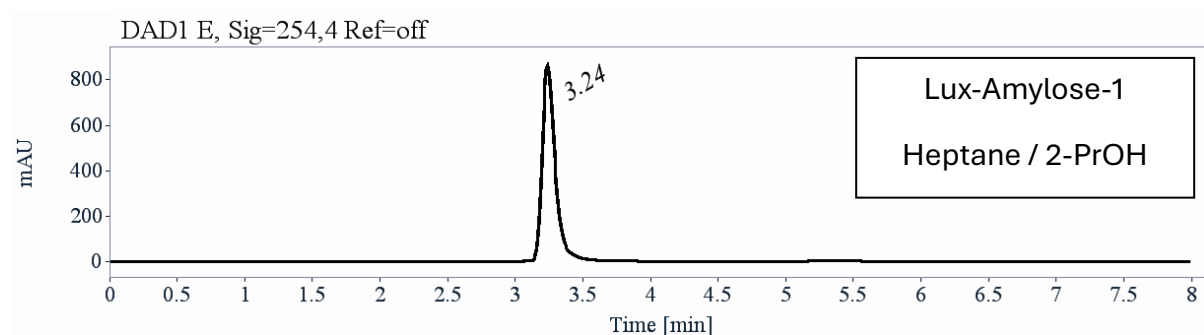
## Preparative chiral HPLC separation for *rac*-3a

Sample preparation: About 200 mg of compound ***rac*-3a** are dissolved in 55 mL of hexane.

Chromatographic conditions: Lux-Amylose-1 (250 x 10 mm), hexane / 2-PrOH (99/1) as mobile phase, flow-rate = 5 mL/min, UV detection at 254 nm.

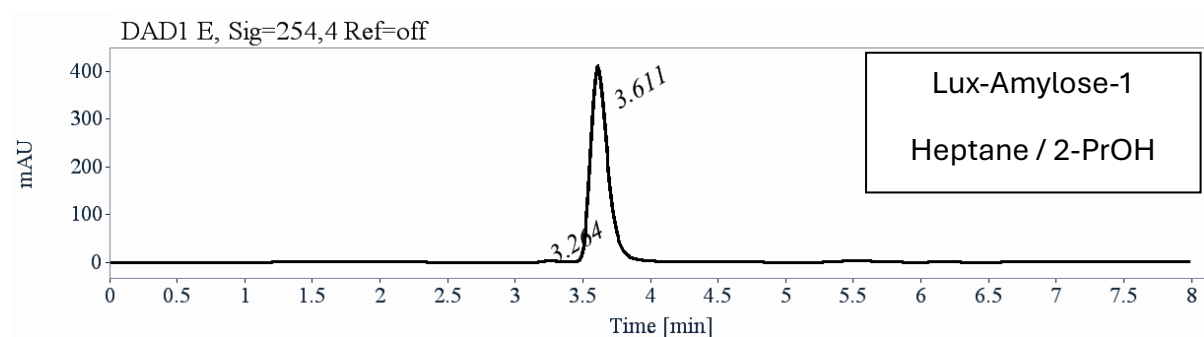
Injections (stacked): 275 times 200  $\mu$ L, every 2.4 minutes.

First fraction: 70 mg of the first eluted enantiomer with ee > 99.5 % (***(S,M<sub>p</sub>)*-3a**: 35% yield).



| RT [min] | Area | Area%  |
|----------|------|--------|
| 3.24     | 5824 | 100.00 |
| Sum      | 5824 | 100.00 |

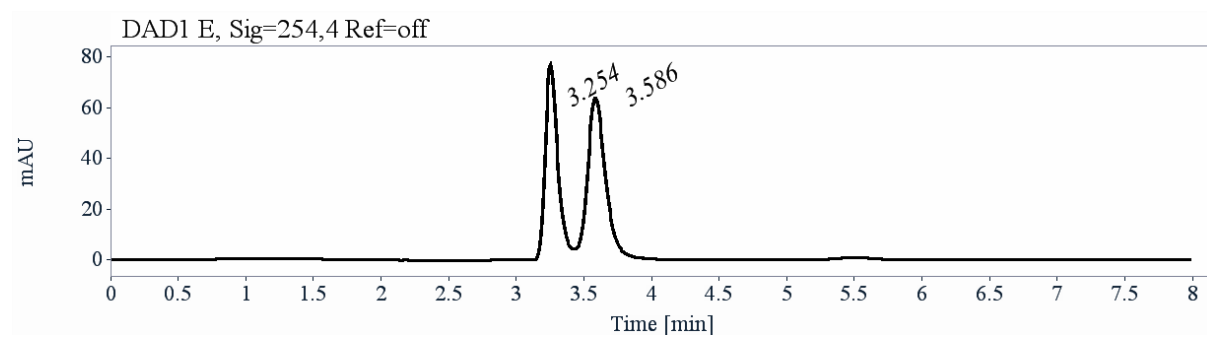
Second fraction: 67 mg of the second eluted enantiomer with ee > 98.5 % (***(R,P<sub>p</sub>)*-3a**: 34% yield).



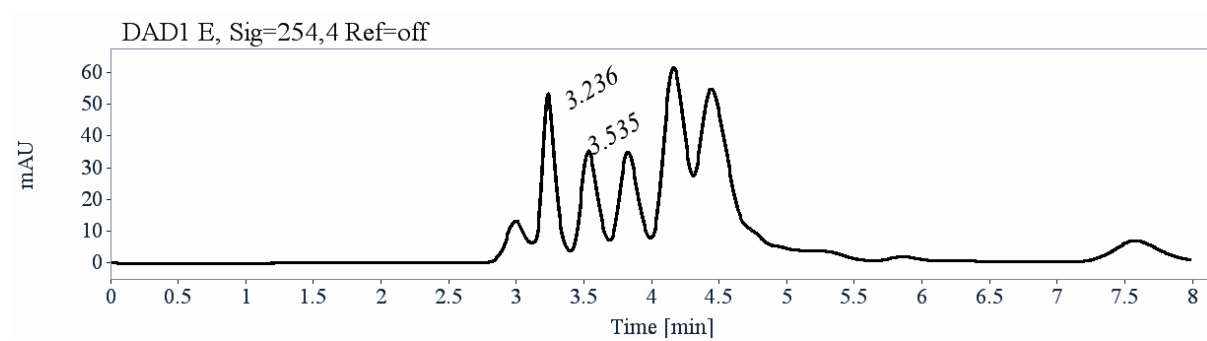
| RT [min] | Area | Area%  |
|----------|------|--------|
| 3.26     | 22   | 0.60   |
| 3.61     | 3647 | 99.40  |
| Sum      | 3669 | 100.00 |



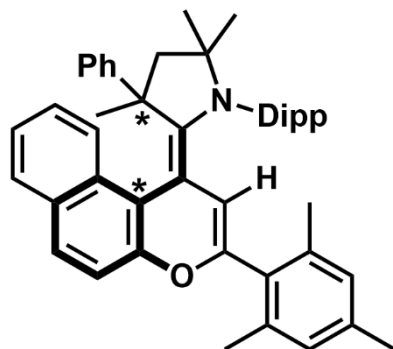
Mixture of the first and the second enantiomers: 14 mg



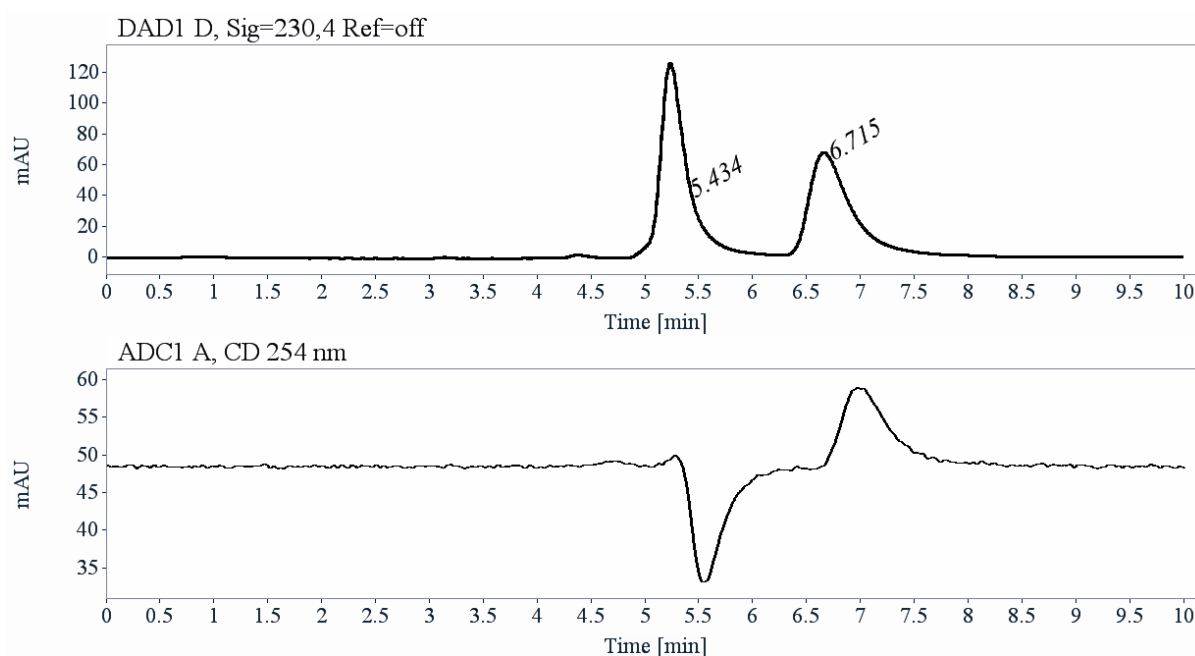
Intermediate: 49 mg



## Analytical chiral HPLC separation for compound *rac-3b*



The sample is dissolved in heptane, injected on the chiral column, and detected with a UV detector at 230 nm and a circular dichroism detector at 254 nm. The flow-rate is 1 mL/min.



| RT [min] | Area | Area%  | Capacity Factor | Enantioselectivity | Resolution (USP) |
|----------|------|--------|-----------------|--------------------|------------------|
| 5.43     | 2142 | 52.13  | 0.84            |                    |                  |
| 6.72     | 1967 | 47.87  | 1.28            | 1.52               | 0.81             |
| Sum      | 4109 | 100.00 |                 |                    |                  |

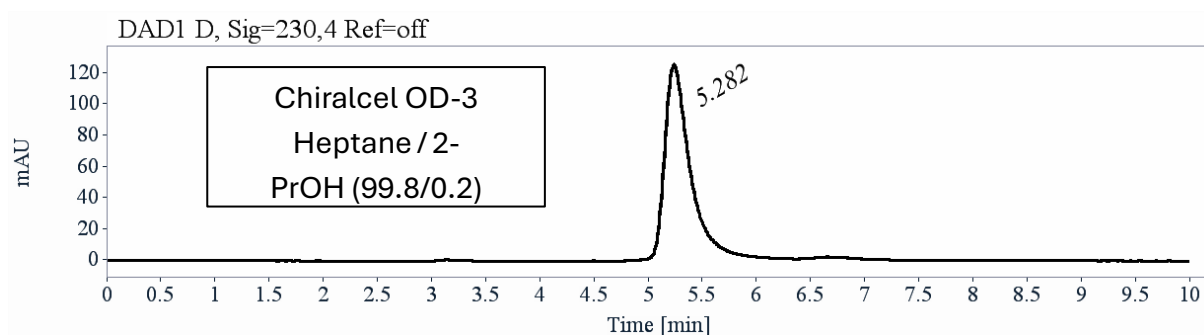
## Preparative chiral HPLC separation for compound *rac-3b*

Sample preparation: About 195 mg of compound *rac-3b* are dissolved in 160 mL of hexane.

Chromatographic conditions: Chiralcel OD-3 (250 x 10 mm), hexane / 2-PrOH (99.8/0.2) as mobile phase, flow-rate = 5 mL/min, UV detection at 290 nm.

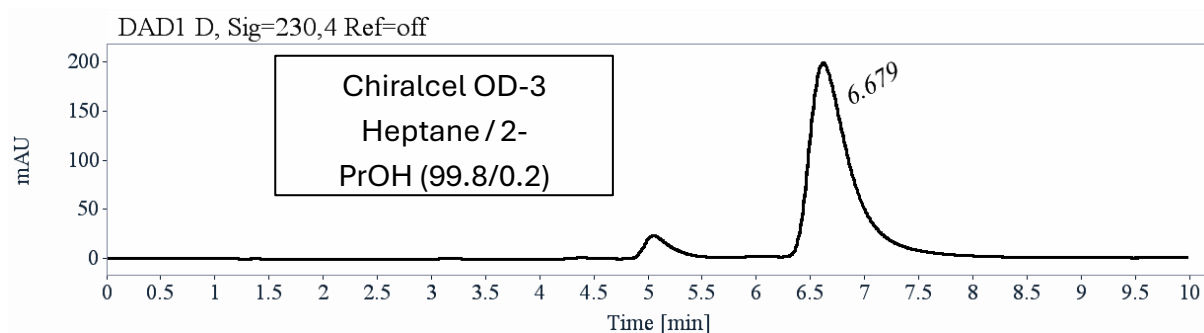
Injections (stacked): 230 times 700  $\mu\text{L}$ , every 3.5 minutes.

The first eluted fraction (78 mg of the first eluted enantiomer with ee = 90 %) is purified again in the same chromatographic conditions (67 stacked injections of 600  $\mu\text{L}$  every 3.5 minutes) to obtain 62 mg of the first eluted enantiomer with ee > 99.5 % ((*R,M<sub>p</sub>*)-**3b**: 32% yield).



| RT [min] | Area | Area%  |
|----------|------|--------|
| 5.28     | 2248 | 100.00 |
| Sum      | 2248 | 100.00 |

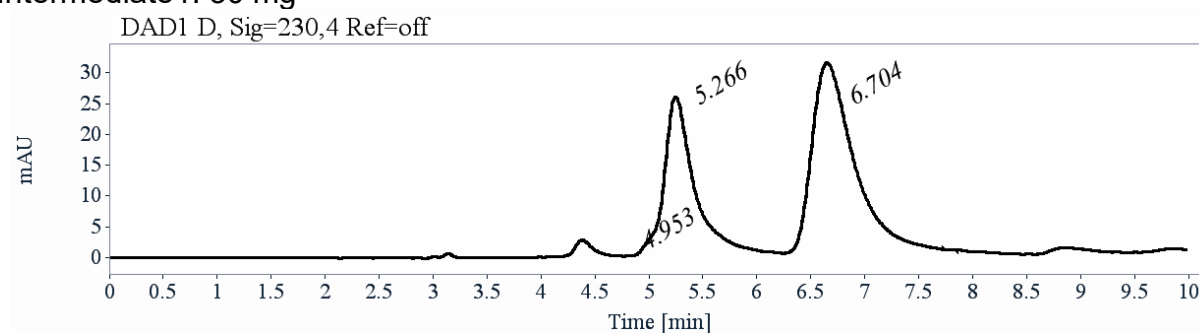
The second eluted fraction (73 mg of the second eluted enantiomer with ee =82%) is purified again in the same chromatographic conditions (63 stacked injections of 400  $\mu\text{L}$  every 3.7 minutes) to obtain 53 mg of the second eluted enantiomer with ee > 99.5 % ((*S,P<sub>p</sub>*)-**3b**: 27% yield).



The peak at 5 minutes is not the first enantiomer but an impurity.

| RT [min] | Area | Area%  |
|----------|------|--------|
| 6.68     | 5354 | 100.00 |
| Sum      | 5354 | 100.00 |

Intermediate1: 30 mg



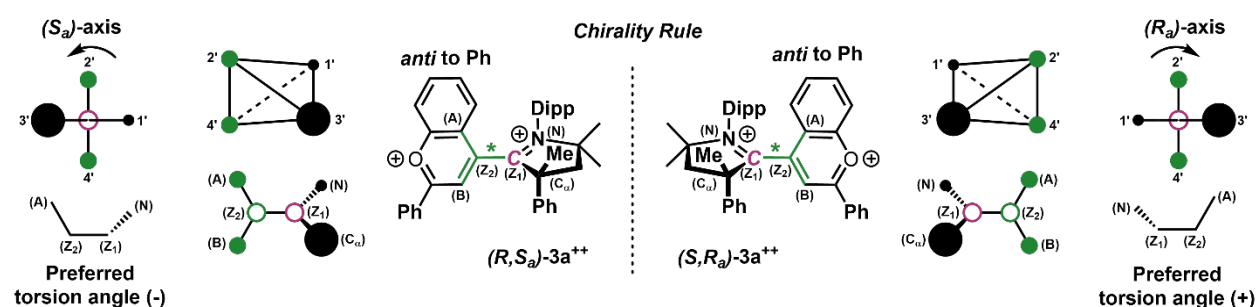
22 mg of the second eluted enantiomer with ee = 30 %

Intermediate2: 16 mg

## Assignment of Absolute Configuration – Axial Chirality

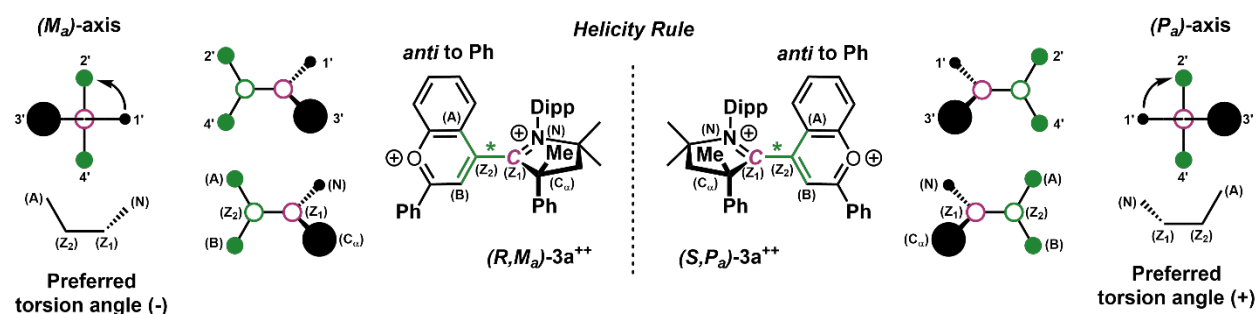
The absolute configuration of axially chiral dications was assigned according to the Cahn-Ingold-Prelog priority system.<sup>4,5</sup> Both chirality and helicity descriptors are derived, although helicity descriptors are preferred. Readers are referred to the “Blue Book,” P-92.1.1-3, as well as section 2.5.2 of Prelog and Helmchen, *ACIE*, **1982**, 21, 567-583 for further details.

**Chirality rule:** Using **3a<sup>++</sup>** as an example, we derive the enantiomorphous six-point figures, C<sub>s</sub>-tetrahedra, and the corresponding preferred torsion angles (Figure S4). Note, because the relative priorities of (a), (b), (C<sub>α</sub>), and (N) are the same in **3b<sup>++</sup>**, the relationship between chirality descriptor and preferred torsion angle does not change.



**Figure S4.** Determining chirality descriptors for axially chiral dications.

**Helicity rule:** Viewing the six-point figures along the ‘(Z<sub>2</sub>)-(Z<sub>1</sub>)’ axis (Figure S5), the sense of helicity is determined by rotating the highest-priority substituent on the near side towards the highest priority substituent of the far side, (A) or (B). Counterclockwise rotation indicates a left-handed helix,  $M_a$ , with a negative preferred torsion angle. Conversely, clockwise rotation indicates a right-handed helix,  $P_a$ , with a positive preferred torsion angle. As for the chirality rule, the helicity assignment based on the preferred torsion angle does not change for **3b<sup>++</sup>**.



**Figure S5.** Determining helicity descriptors for axially chiral dications.

From these derivations, the equivalence between  $R_a$  and  $P_a$ ,  $S_a$  and  $M_a$ , is clear. Note, in the context of the specific dications from this work, a correlation between the sense of

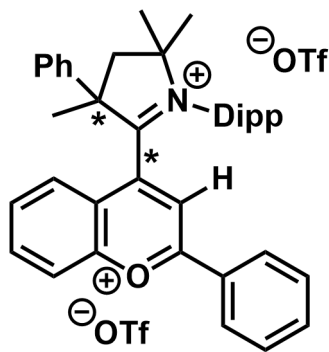
helicity and the orientation (*syn* or *anti*) of the chromene ring relative to the phenyl substituent of the  $\alpha$ -carbon is observed: When the stereogenic  $\alpha$ -carbon center is **S**, *anti* = **R<sub>a</sub>** = **P<sub>a</sub>**, while *syn* = **S<sub>a</sub>** = **M<sub>a</sub>**. A change in orientation is equivalent to a configurational inversion of the stereogenic axis.

## Preparation of axially chiral dications

### General Procedure

In a glovebox, a Schlenk flask was charged with the appropriate alkene (1 eq.), AgOTf (2 eq.), and a magnetic stir bar. The flask was sealed, transferred to a Schlenk line, then equipped with a septum. DCM (5 mL per mmol) was added via syringe at room temperature. A tenebrous solution formed immediately. After stirring for 1 h, the solution was cannula-filtered into another Schlenk flask, washing the remaining solids with DCM as needed to affect quantitative transfer. Volatiles were removed under vacuum at room-temperature, yielding the products as deeply colored residues.

### Dication [*rac*-3a<sup>++</sup>][OTf]<sub>2</sub>



Prepared according to the general procedure from **rac-3a** (100 mg, 0.18 mmol). Dark yellow, 128 mg (0.15 mmol, 83% yield).

In DCM-d<sub>2</sub>, a 3:2 ratio of atropisomers is observed. Due to peak overlap and complex splitting, not all peaks in the aryl regions of the <sup>1</sup>H and <sup>13</sup>C spectra could be unambiguously assigned between major and minor atropisomers.

In THF-d<sub>8</sub>, a 4:1 ratio of atropisomers is observed, enabling the unambiguous assignment of all peaks for the major atropisomer.

**<sup>1</sup>H NMR (500 MHz, 25°C, DCM-d<sub>2</sub>)**  $\delta$  = 8.93 (d, J = 7.4 Hz, 2H, *MAJOR*), 8.87 (d, J = 7.0 Hz, 1H, *minor*), 8.87 (s, 1H, *MAJOR*), 8.44 – 8.38 (m, 3H, *minor*), 8.22 (d, J = 8.6 Hz, 1H, *MAJOR*), 8.10 (t, J = 8.7 Hz, 1H, *MAJOR*), 8.01 (t, J = 7.5 Hz, 1H, *MAJOR*), 7.94 (t, J = 7.4 Hz, 1H, *minor*), 7.89 (d, J = 7.3 Hz, 2H, *minor*), 7.83 (t, J = 8.0 Hz, 1H, *MAJOR*), 7.76 – 7.67 (m), 7.55 – 7.53 (m, 2H, *MAJOR*), 7.48 (t, J = 7.2 Hz, 2H, *minor*), 7.46 – 7.41 (m), 7.33 – 7.29 (m), 7.23 – 7.17 (m), 6.95 (d, J = 8.3 Hz, 1H, *MAJOR*), 6.66 (s, 1H, *minor*), 3.92 (d, J = 14.2 Hz, 1H, *MAJOR*), 3.67 (d, J = 14.7 Hz, 1H, *minor*), 3.28 (d, J = 14.7 Hz, 1H, *minor*), 3.21 (sept, J = 6.3 Hz, 1H, *MAJOR*), 3.05 (sept, J = 6.4 Hz, 1H, *minor*), 3.01 (d, J = 14.3 Hz, 1H, *MAJOR*), 2.74 (sept, J = 6.4 Hz, 1H, *MAJOR*), 2.55 (sept, J = 6.4 Hz,

1H, *minor*), 2.53 (s, 3H, *MAJOR*), 2.30 (s, 3H, *minor*), 2.21 (s, 3H, *minor*), 1.99 (s, 3H, *MAJOR*), 1.93 (s, 3H, *MAJOR*), 1.82 (s, 3H, *minor*), 1.53 (d, J = 6.3 Hz, 3H, *MAJOR*), 1.47 (d, J = 6.3 Hz, 3H, *minor*), 1.42 (d, J = 6.3 Hz, 3H, *MAJOR*), 1.40 (d, J = 6.3 Hz, 3H, *minor*), 1.13 (d, J = 6.4 Hz, 3H, *MAJOR*), 0.85 (d, J = 6.4 Hz, 3H, *minor*), 0.32 (d, J = 6.4 Hz, 3H, *MAJOR*), 0.19 (d, J = 6.6 Hz, 3H, *minor*).

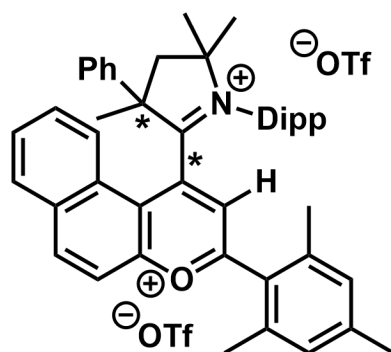
**<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, 25°C, DCM-d<sub>2</sub>)** δ = 192.63 (*MAJOR*), 191.59 (*minor*), 176.67 (*MAJOR*), 174.16 (*minor*), 157.11 (*minor*), 155.50 (*MAJOR*), 149.88 (*MAJOR*), 149.55 (*minor*), 145.47 (*MAJOR*), 145.37 (*minor*), 144.93 (*MAJOR*), 144.63 (*minor*), 141.41 (*MAJOR*), 141.02 (*MAJOR*), 140.53 (*minor*), 137.24 (*minor*), 136.12 (*MAJOR*), 133.95, 133.71, 133.22, 133.01, 131.64, 131.60, 131.34, 131.29, 131.25, 130.79, 130.46, 130.35, 129.85, 129.56, 128.85, 128.72, 128.45, 128.23, 128.11, 128.08, 127.77, 127.75, 127.57, 127.06, 121.42 (*minor*), 121.21 (*minor*), 120.98 (q, J = 320.3 Hz), 120.46 (*MAJOR*), 120.30 (*MAJOR*), 118.28 (*MAJOR*), 117.94 (*minor*), 89.80 (*minor*), 89.30 (*MAJOR*), 62.81 (*minor*), 61.91 (*MAJOR*), 47.88 (*minor*), 46.59 (*MAJOR*), 32.40 (*minor*), 30.56 (*MAJOR*), 30.28 (*MAJOR*), 30.17 (*MAJOR*), 30.14 (*MAJOR*), 30.03 (*minor*), 29.87 (*MAJOR*), 29.15 (*minor*), 28.83 (*minor*), 28.68 (*MAJOR*), 28.23 (*minor*), 28.15 (*MAJOR*), 27.45 (*minor*), 27.20 (*minor*), 25.23 (*minor*), 25.10 (*MAJOR*), 24.93 (*MAJOR*), 24.89 (*minor*).

**<sup>19</sup>F NMR (471 MHz, 25°C, DCM-d<sub>2</sub>)** δ = -76.40

**<sup>1</sup>H NMR (500 MHz, 25°C, THF-d<sub>8</sub>)** δ = 9.16 (d, J = 7.7 Hz, 2H), 9.02 (s, 1H), 8.49 (d, J = 8.6 Hz, 1H), 8.15 (t, J = 7.9 Hz, 1H), 8.00 (t, J = 7.5 Hz, 1H), 7.83 (t, J = 7.9 Hz, 2H), 7.69 (t, J = 8.4 Hz, 2H), 7.65 (d, J = 7.9 Hz, 1H), 7.53 – 7.41 (m, 4H), 7.18 (t, J = 7.8 Hz, 2H), 7.08 (t, J = 7.4 Hz, 1H), 4.18 (d, J = 14.0 Hz, 1H), 3.39 (sept, J = 6.5 Hz, 1H), 3.12 – 2.95 (m, 3H), 2.60 (s, 3H), 2.04 (s, 3H), 1.96 (s, 3H), 1.60 (d, J = 6.1 Hz, 3H), 1.45 (d, J = 6.1 Hz, 3H), 1.29 (d, J = 6.4 Hz, 3H), 0.32 (d, J = 6.4 Hz, 3H). *Only peaks for the major atropisomer are reported.*

**<sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, 25°C, THF-d<sub>8</sub>)** δ = 194.53, 177.97, 156.08, 150.38, 147.38, 146.21, 141.52, 140.85, 137.46, 134.42, 133.34, 132.60, 131.66, 130.55, 130.44, 129.93, 129.76, 129.41, 128.42, 128.23, 121.46, 121.11, 120.85 (q, J = 353.45 Hz), 120.68, 90.16, 67.92, 62.71, 45.93, 31.64, 30.92, 30.49, 30.40, 30.00, 28.98, 28.57. *Only peaks for the major atropisomer are reported. Two peaks overlap with the solvent signal at 25.46 ppm.*

## Dication [*rac*-3b<sup>++</sup>][OTf]<sub>2</sub>



Prepared according to the general procedure from **rac-3b** (100 mg, 0.16 mmol). Dark red, 132 mg (0.14 mmol, 87% yield).

For SbF<sub>6</sub><sup>-</sup> salt, the procedure was reproduced using AgSbF<sub>6</sub> in place of AgOTf. Vapor diffusion of Et<sub>2</sub>O into a DCM solution yielded diffraction quality crystals.

In DCM-d<sub>2</sub>, a 5:2 ratio of atropisomers is observed. Due to peak overlap and complex splitting, unambiguous assignment of all aryl peaks between major and minor atropisomers was not possible.

**<sup>1</sup>H NMR (400 MHz, 25°C, DCM-d<sub>2</sub>)** δ = 9.01 (d, *J* = 9.2 Hz, 1H, *minor*), 8.74 (d, *J* = 9.2 Hz, 1H, *MAJOR*), 8.56 (d, *J* = 3.5 Hz, 2H, *minor*), 8.44 (d, *J* = 7.9 Hz, 1H, *minor*), 8.28 (d, *J* = 9.2 Hz, 1H, *minor*), 8.18 (t, *J* = 8.1 Hz, 1H, *minor*), 8.18 (d, *J* = 9.2 Hz, 1H, *MAJOR*), 8.08 (d, *J* = 6.6 Hz, 1H, *MAJOR*), 8.00 (s, 1H), 7.83 (dd, *J* = 8.4, 2.2 Hz, 1H, *MAJOR*), 7.79 (t, *J* = 7.6 Hz, 1H, *MAJOR*), 7.65 (td, *J* = 7.9, 1.5 Hz, 1H, *MAJOR*), 7.58 (d, *J* = 7.6 Hz, 1H, *MAJOR*), 7.53 (d, *J* = 8.4 Hz, 1H, *MAJOR*), 7.42 (d, *J* = 4.3 Hz, 1H, *minor*), 7.40 (d, *J* = 7.8 Hz, 1H, *minor*), 7.39 (d, *J* = 7.9 Hz, 1H, *MAJOR*), 7.24 (dd, *J* = 7.9, 1.6 Hz, 1H, *MAJOR*), 7.17 (s, 2H, *MAJOR*), 7.11 (dd, *J* = 7.8, 1.5 Hz, 1H, *minor*), 7.06 (dd, *J* = 7.8, 1.5 Hz, 1H, *minor*), 7.01 (dd, *J* = 7.9, 1.6 Hz, 2H, *MAJOR*), 6.97 (s, 1H, *minor*), 6.97 (s, 2H, *minor*), 6.93 (t, *J* = 7.3 Hz, 1H, *MAJOR*), 4.25 (d, *J* = 14.5 Hz, 1H, *MAJOR*), 4.07 (d, *J* = 15.3 Hz, 1H, *minor*), 3.53 (d, *J* = 15.4 Hz, 1H, *minor*), 3.34 (d, *J* = 14.4 Hz, 1H, *MAJOR*), 2.89 (sept, *J* = 6.3 Hz, 1H, *MAJOR*), 2.53 (s, 3H, *MAJOR*), 2.49 (sept, *J* = 6.5 Hz, 1H, *minor*), 2.43 (s, 3H, *MAJOR*), 2.40 (broad, 6H, *MAJOR*), 2.33 (s, 3H, *minor*), 2.29 (sept, *J* = 6.4 Hz, 1H, *MAJOR*), 2.15 (s, 3H, *minor*), 2.10 (s, 3H, *minor*), 2.02 (s, 3H, *MAJOR*), 2.01 (s, 3H, *minor*), 1.98 (s, 3H, *MAJOR*), 1.78 (broad, 6H, *MAJOR*), 1.45 (d, *J* = 6.1 Hz, 3H, *MAJOR*), 1.22 (d, *J* = 6.3 Hz, 3H, *minor*), 0.99 (d, *J* = 6.4 Hz, 3H, *minor*), 0.95 (d, *J* = 6.4 Hz, 3H, *MAJOR*), 0.92 (d, *J* = 6.6 Hz, 3H, *MAJOR*), 0.43 (d, *J* = 6.3 Hz, 3H, *minor*), -0.67 (d, *J* = 6.4 Hz, 3H, *minor*), -0.78 (d, *J* = 6.4 Hz, 3H, *MAJOR*).

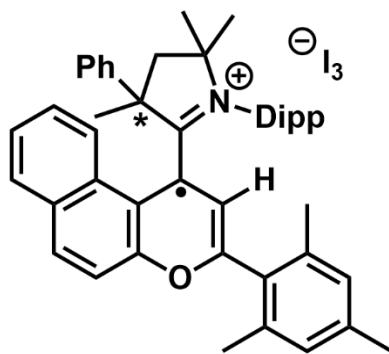
**<sup>13</sup>C{<sup>1</sup>H} NMR (101 MHz, 25°C, DCM-d<sub>2</sub>)** δ = 197.28 (*MAJOR*), 194.65 (*minor*), 175.55 (*MAJOR*), 174.46 (*minor*), 164.49 (*minor*), 161.91 (*MAJOR*), 148.82 (*minor*), 148.03 (*MAJOR*), 147.36 (*MAJOR*), 147.23 (*minor*), 147.02 (*MAJOR*), 146.83 (*minor*), 146.09 (*MAJOR*), 145.88 (*minor*), 145.84 (*minor*), 145.30 (*MAJOR*), 140.76 (*MAJOR*), 138.59 (*minor*), 138.57 (*minor*), 136.98 (*MAJOR*), 133.61 (*minor*), 133.52 (*minor*), 133.37 (*minor*), 133.32 (*minor*), 133.12 (*minor*), 132.53 (*MAJOR*), 132.04 (*MAJOR*), 131.85 (*MAJOR*), 131.76 (*MAJOR*), 131.63 (*minor*), 131.08, 131.00 (*MAJOR*), 130.66 (*MAJOR*), 130.24 (*minor*), 130.22 (*minor*), 129.84 (*minor*), 128.78, 128.59 (*MAJOR*), 128.52 (*minor*), 128.36, 128.07, 127.95, 127.90, 127.81, 127.07 (*MAJOR*), 125.38, 125.20, 124.99,



124.57, 123.62, 123.35 (*minor*), 121.16, 120.85 (*MAJOR*), 120.71 (q,  $J = 320.5$  Hz), 119.78 (*minor*), 119.18 (*MAJOR*), 88.89 (*MAJOR*), 88.86 (*minor*), 66.53 (*minor*), 65.10 (*MAJOR*), 45.26 (*minor*), 45.01 (*MAJOR*), 36.14 (*MAJOR*), 32.56 (*minor*), 31.12 (*MAJOR*), 30.33 (*MAJOR*), 30.05 (*MAJOR*), 29.61 (*minor*), 29.61 (*minor*), 28.45 (*minor*), 28.31 (*MAJOR*), 27.78 (*minor*), 26.88 (*MAJOR*), 26.70 (*minor*), 25.84 (*MAJOR*), 24.94 (*minor*), 24.92 (*minor*), 24.75 (*MAJOR*), 24.70 (*MAJOR*), 24.10 (*minor*), 21.97 (*MAJOR*), 21.62 (*minor*), 21.35 (*MAJOR*), 20.22 (*minor*).

$^{19}\text{F}$  NMR (471 MHz, 25°C, DCM- $d_2$ )  $\delta = -78.39$

### Radical cation [*rac*-3b $^{+\bullet}$ ][I $_3$ ]:



Under an argon atmosphere, a Schlenk flask was charged with ***rac*-3b** (100 mg, 0.15 mmol, 1 eq.), I $_2$  (80 mg, 0.32 mmol, 2.1 eq.) and a magnetic stir bar. DCM (2 mL) was slowly added via syringe at room temperature, producing a deep black-purple solution. After stirring for 1 h, the solution was concentrated under vacuum at room-temperature, leaving a black residue, which was washed with toluene till the washings were colorless. Crystalline material was obtained by vapor diffusion of Et $_2$ O into a DCM solution.

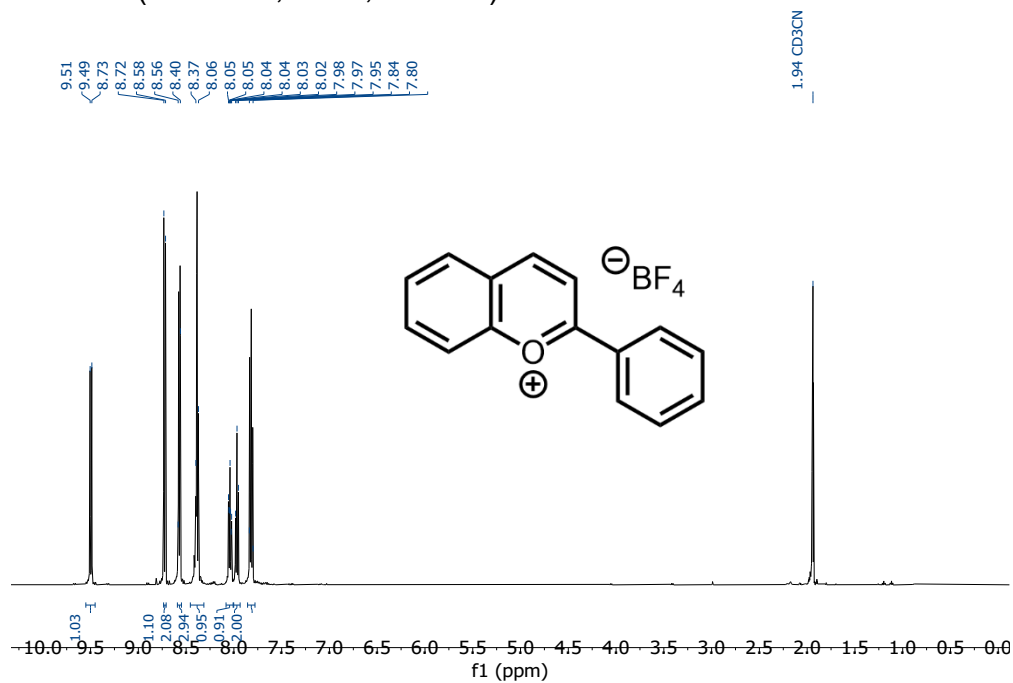
Black rods, 78 mg, (0.076 mmol, 51% yield).

**X-Band EPR:**  $g = 2.002$  (1xN: 12.2 MHz; 1xH: 6.4 MHz; 1xH: 4.3 MHz; 1xH: 2.4 MHz; 3xH: 0.6 MHz; 1xH: 0.5 MHz).

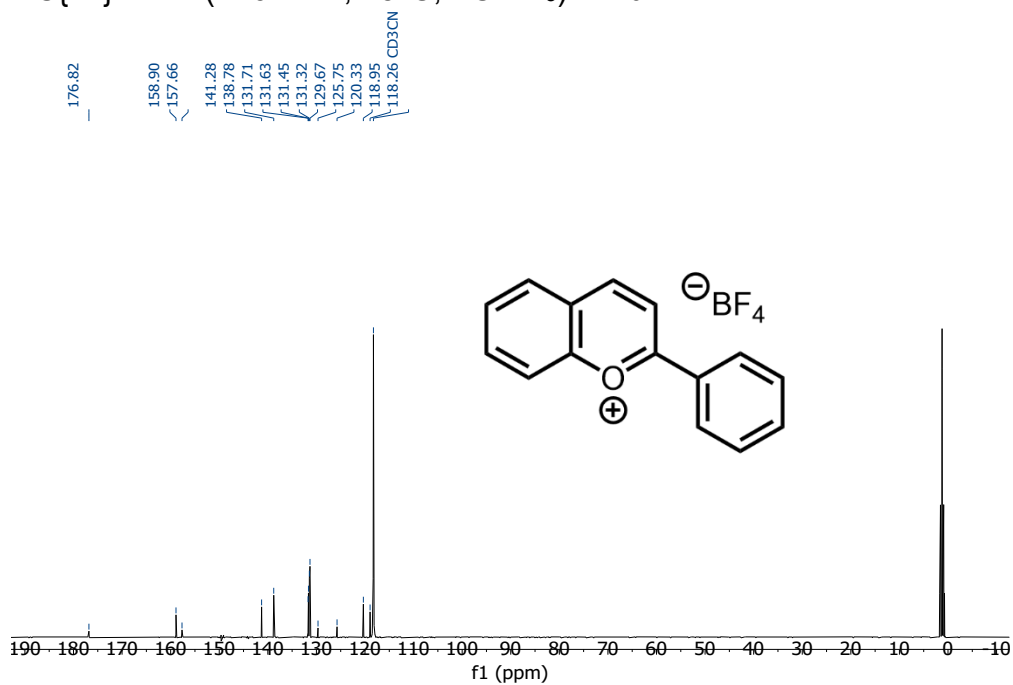
### 3) NMR Characterization

#### 1D NMR Spectra

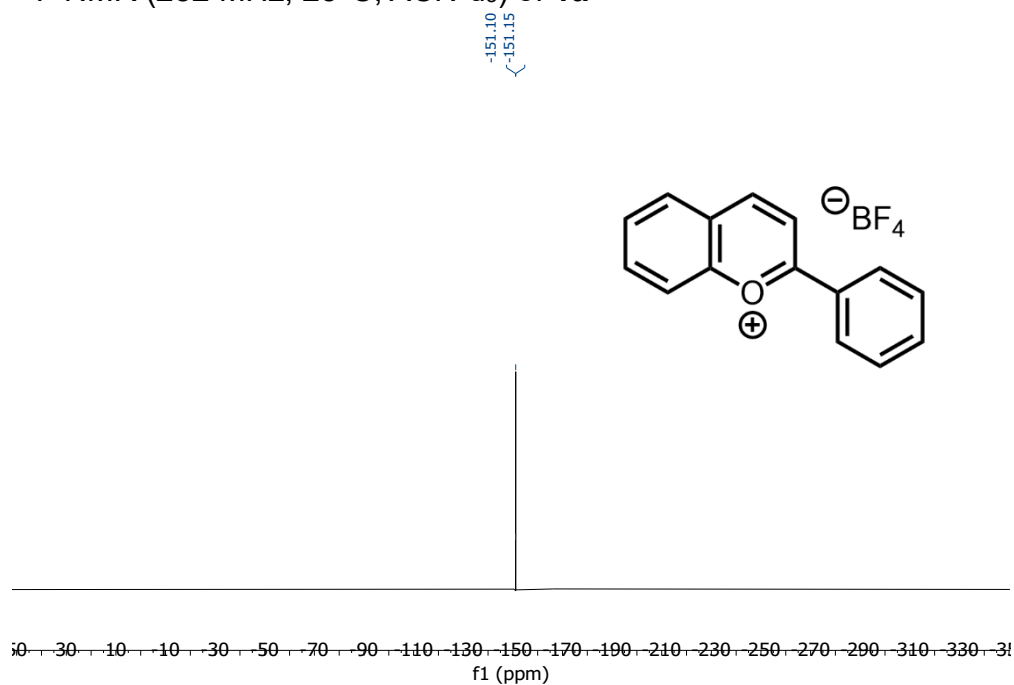
$^1\text{H}$  NMR (500 MHz, 25°C,  $\text{ACN-d}_3$ ) of **1a**



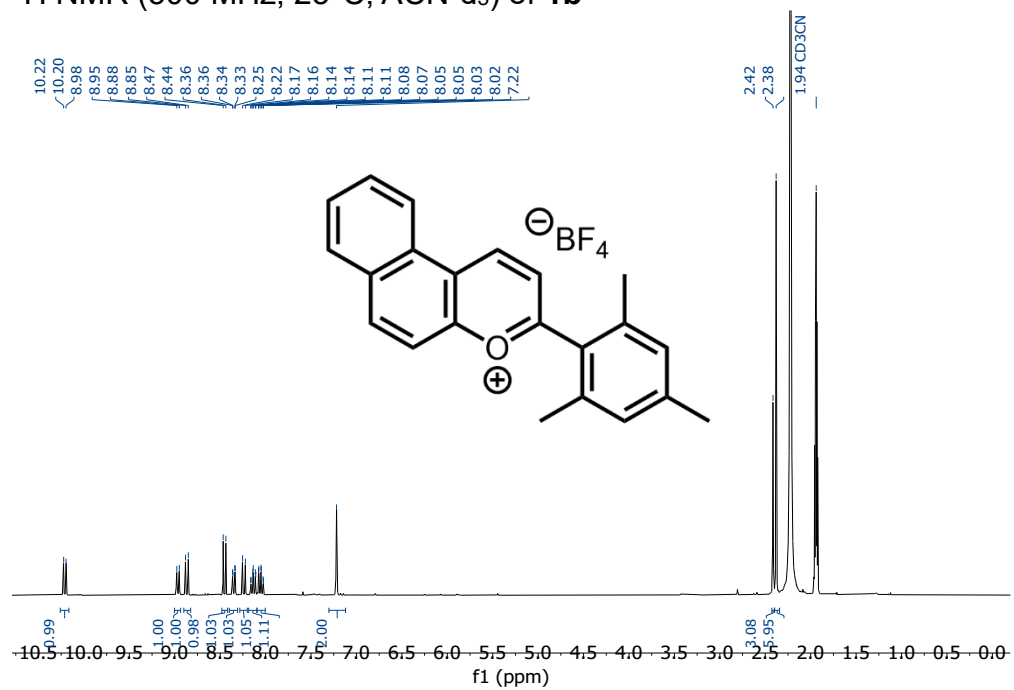
$^{13}\text{C}\{^1\text{H}\}$  NMR (126 MHz, 25°C,  $\text{ACN-d}_3$ ) of **1a**



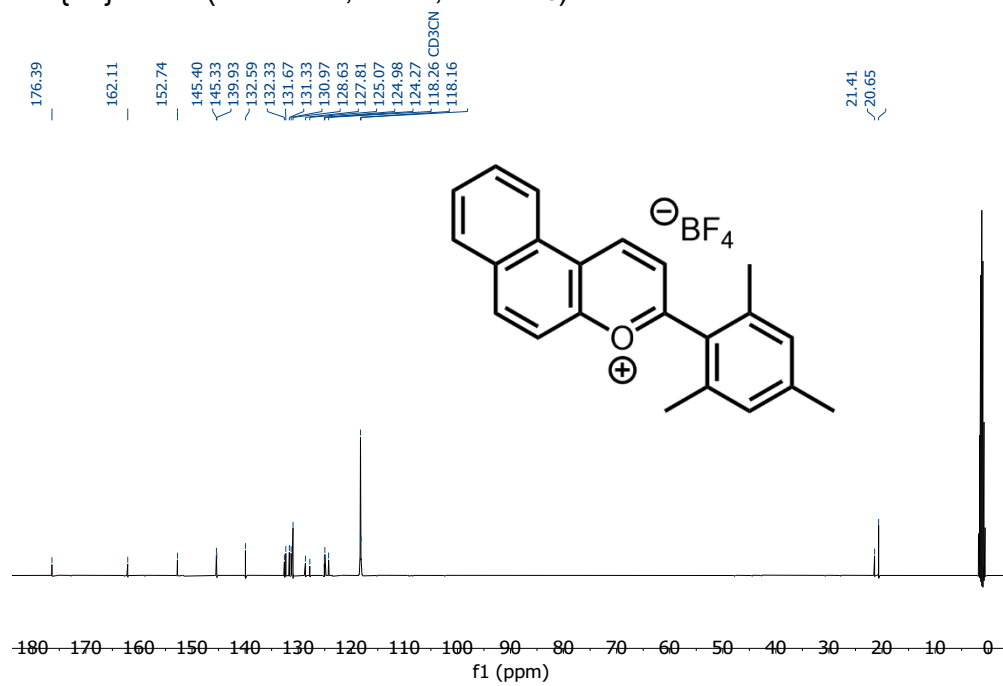
$^{19}\text{F}$  NMR (282 MHz, 25°C, ACN- $\text{d}_3$ ) of **1a**



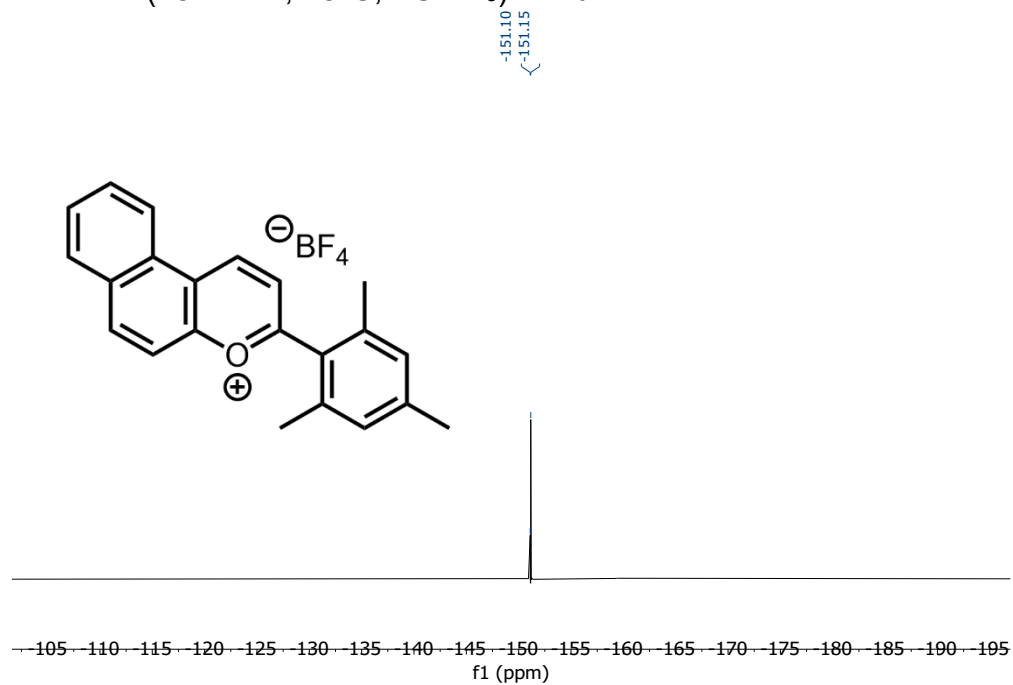
$^1\text{H}$  NMR (500 MHz, 25°C, ACN- $\text{d}_3$ ) of **1b**



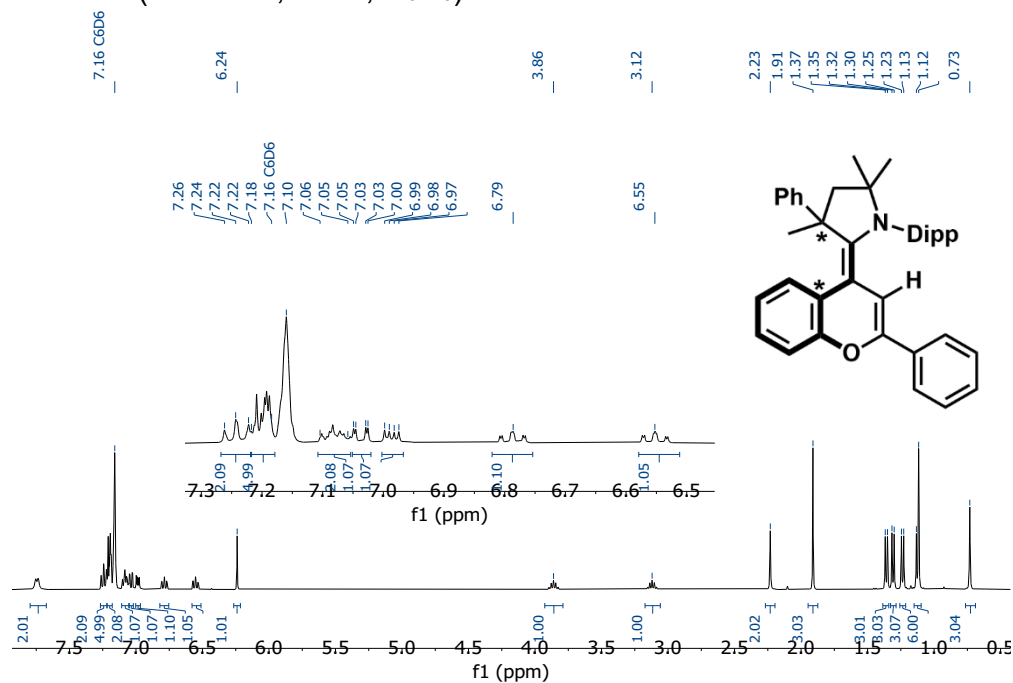
$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz, 25°C, ACN- $\text{d}_3$ ) of **1b**



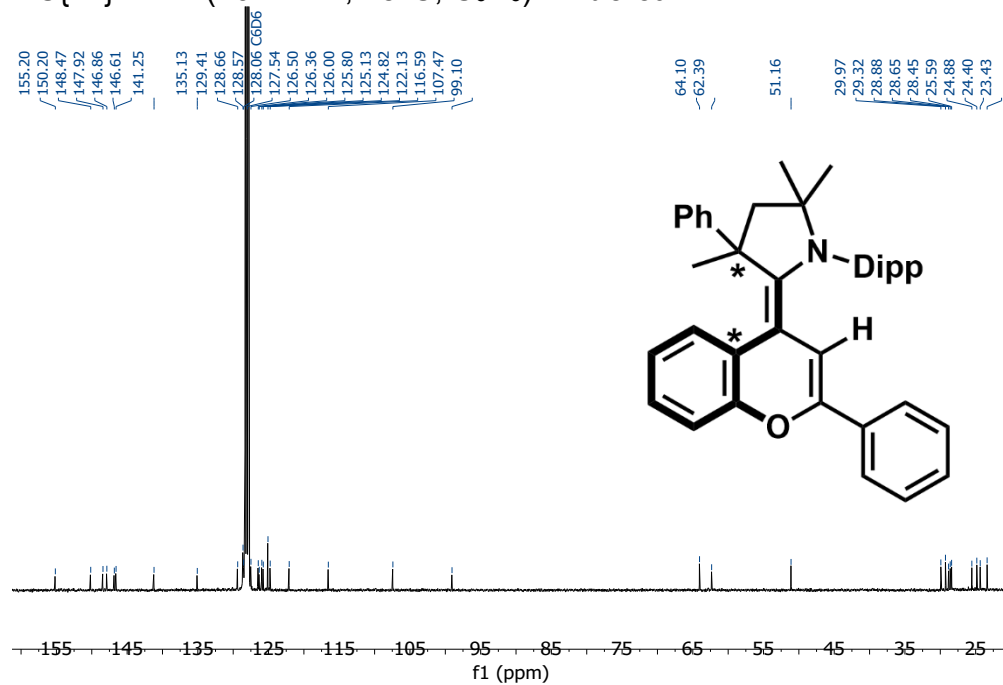
$^{19}\text{F}$  NMR (282 MHz, 25°C, ACN- $\text{d}_3$ ) of **1b**



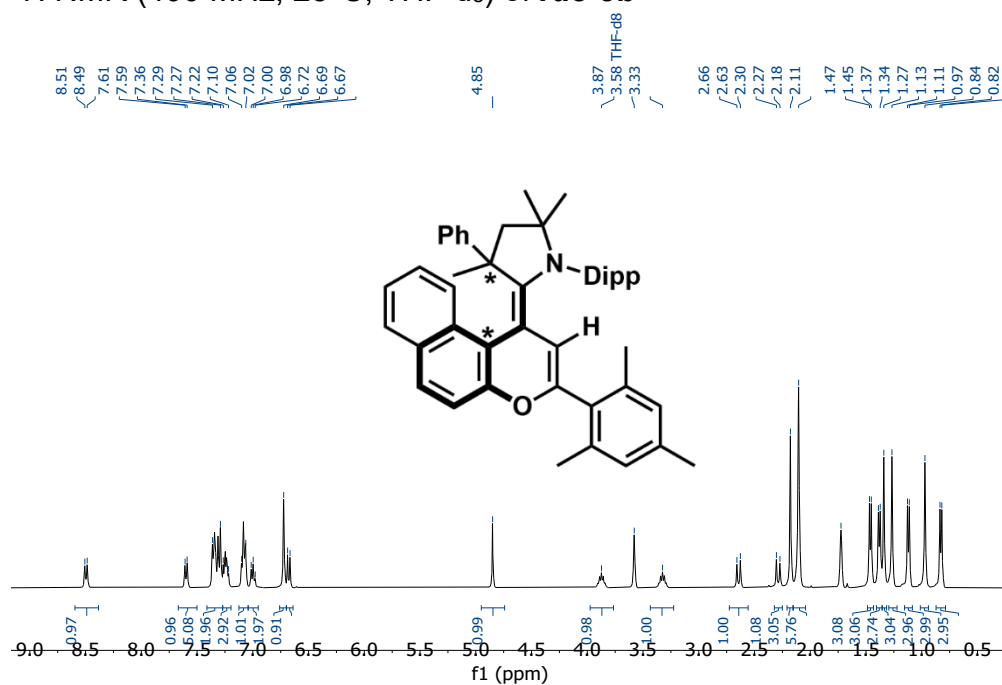
$^1\text{H}$  NMR (400 MHz,  $25^\circ\text{C}$ ,  $\text{C}_6\text{D}_6$ ) of *rac*-**3a**



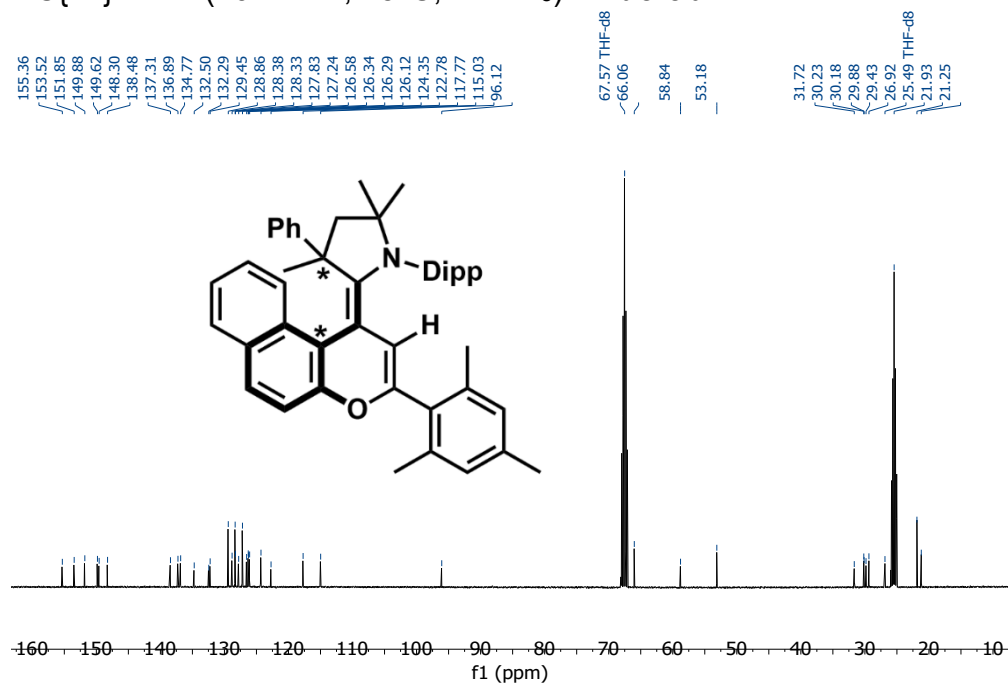
$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz,  $25^\circ\text{C}$ ,  $\text{C}_6\text{D}_6$ ) of *rac*-**3a**



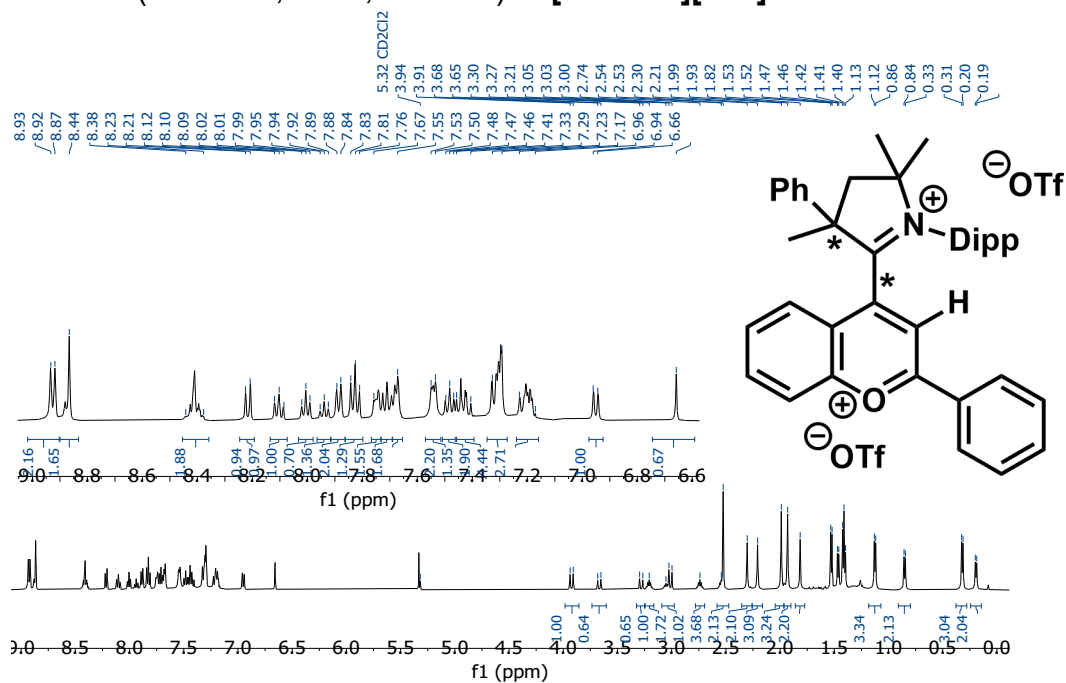
$^1\text{H}$  NMR (400 MHz, 25°C, THF- $\text{d}_8$ ) of *rac*-**3b**



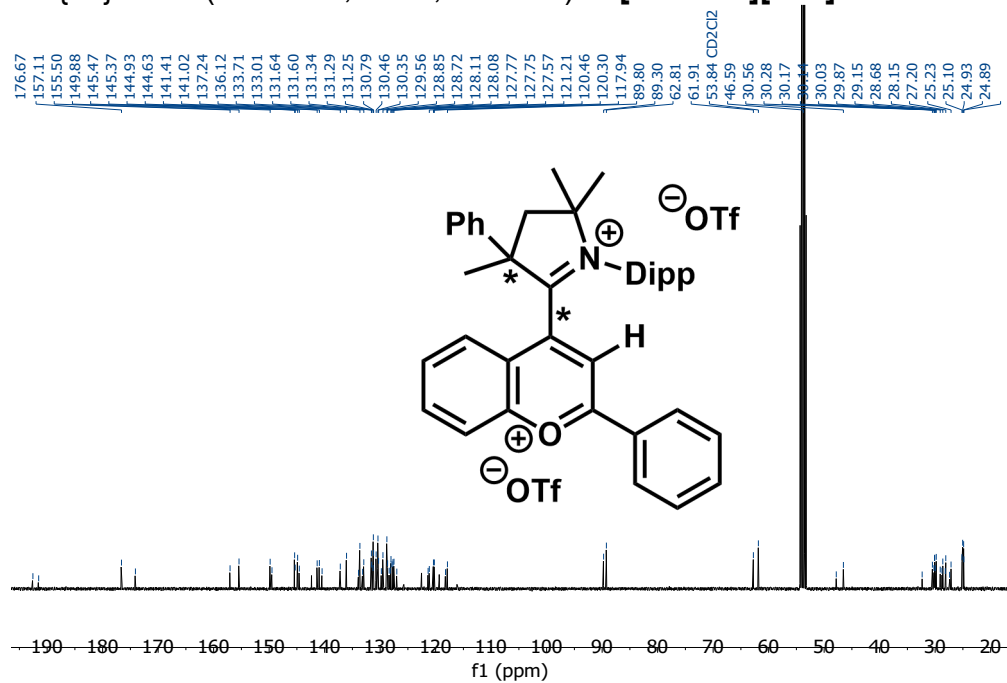
$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz, 25°C, THF- $\text{d}_8$ ) of *rac*-**3b**



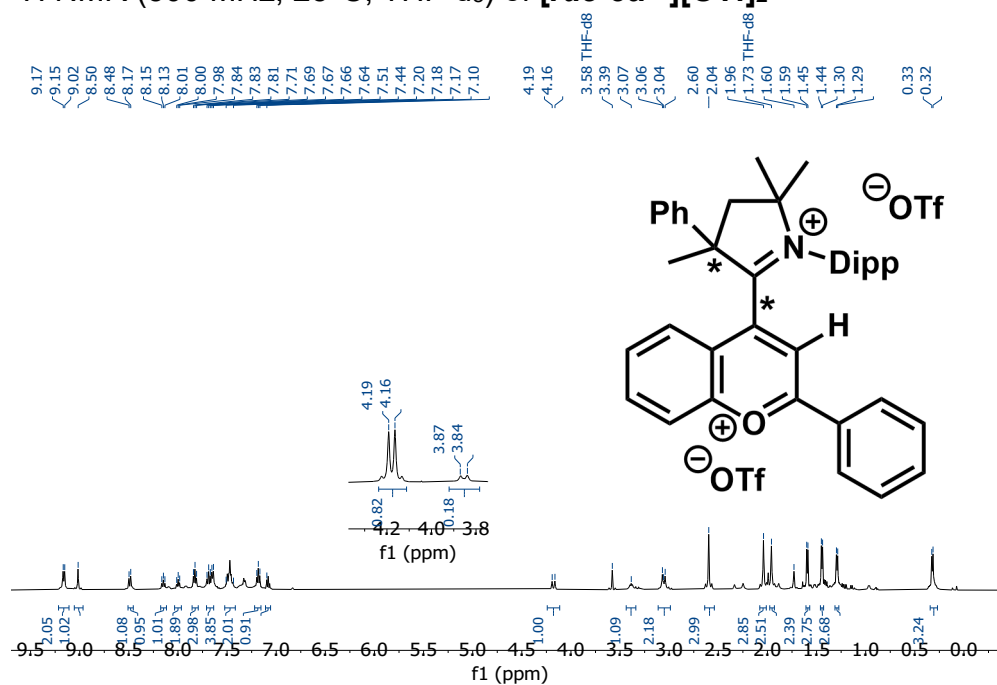
$^1\text{H}$  NMR (500 MHz,  $25^\circ\text{C}$ ,  $\text{DCM-d}_2$ ) of  $[\text{rac-3a}^{++}][\text{OTf}]_2$



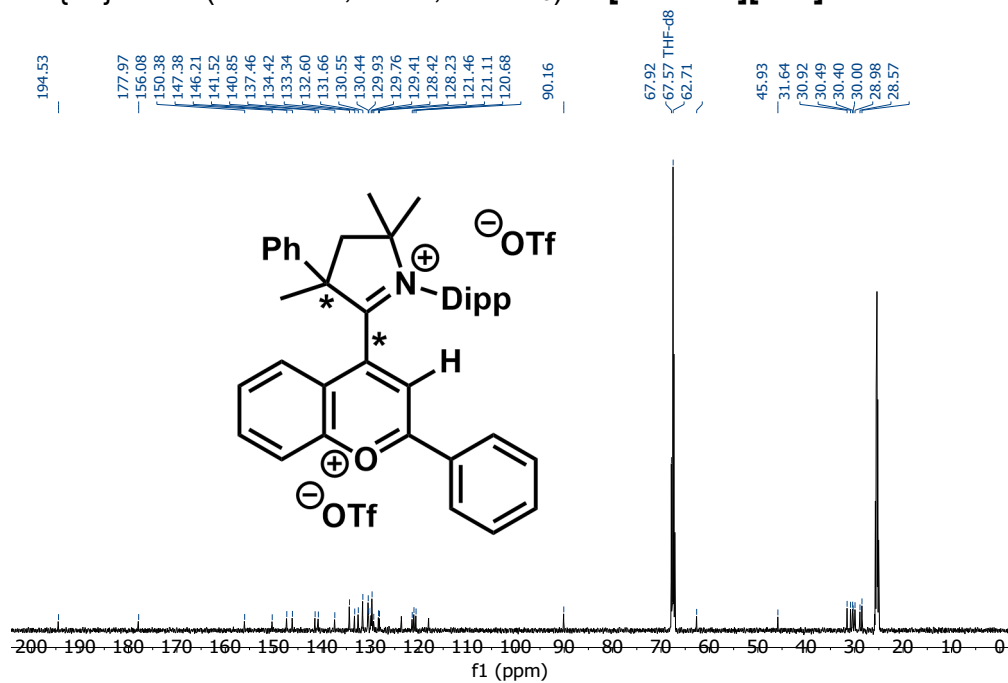
$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz,  $25^\circ\text{C}$ ,  $\text{DCM-d}_2$ ) of  $[\text{rac-3a}^{++}][\text{OTf}]_2$



$^1\text{H}$  NMR (500 MHz, 25°C, THF- $d_8$ ) of  $[\text{rac-3a}^{++}][\text{OTf}]_2$

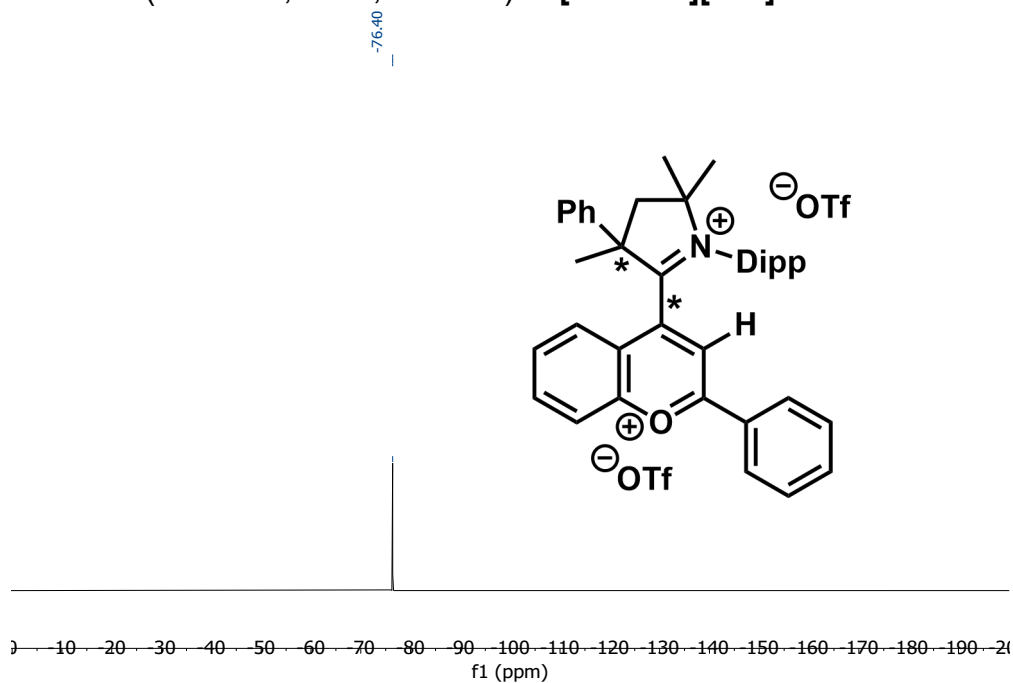


$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz, 25°C, THF- $d_8$ ) of  $[\text{rac-3a}^{++}][\text{OTf}]_2$

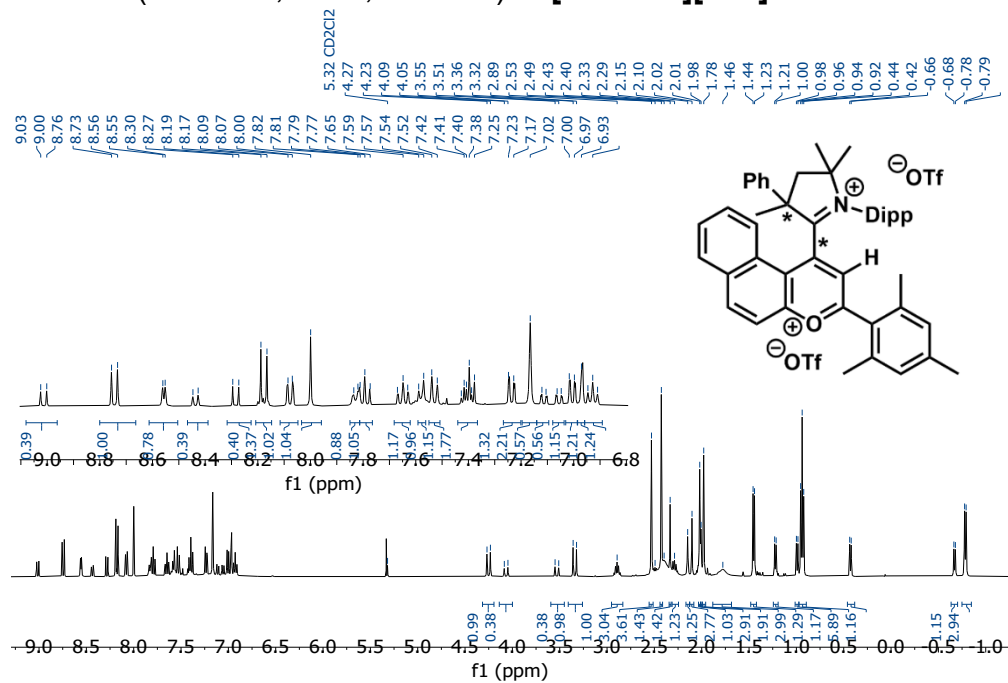




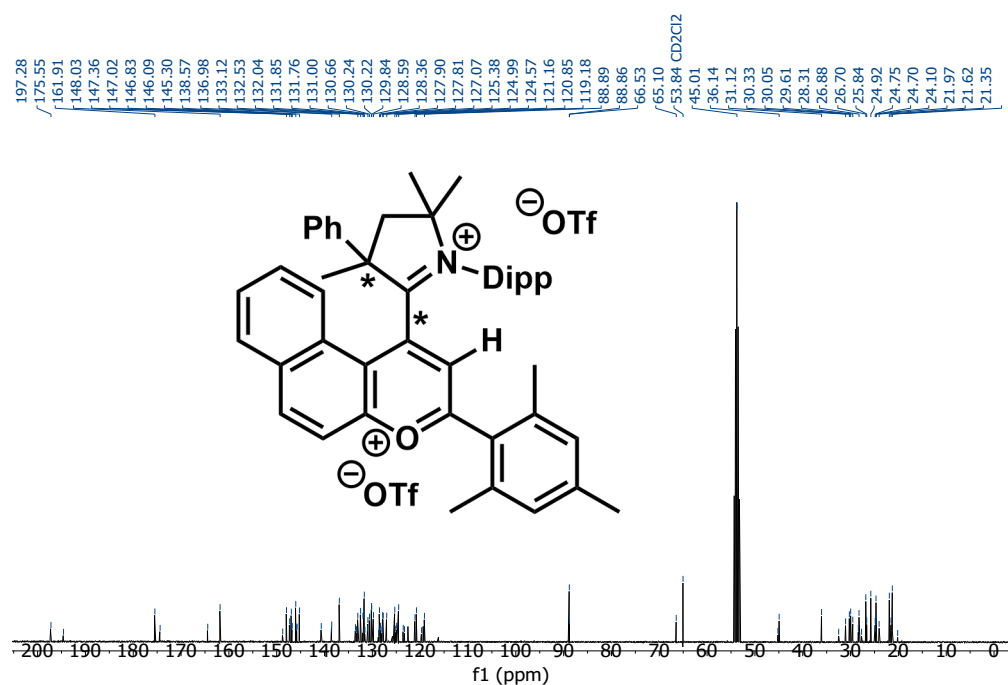
$^{19}\text{F}$  NMR (500 MHz, 25°C, DCM-d<sub>2</sub>) of [*rac*-3a<sup>++</sup>][OTf]<sub>2</sub>



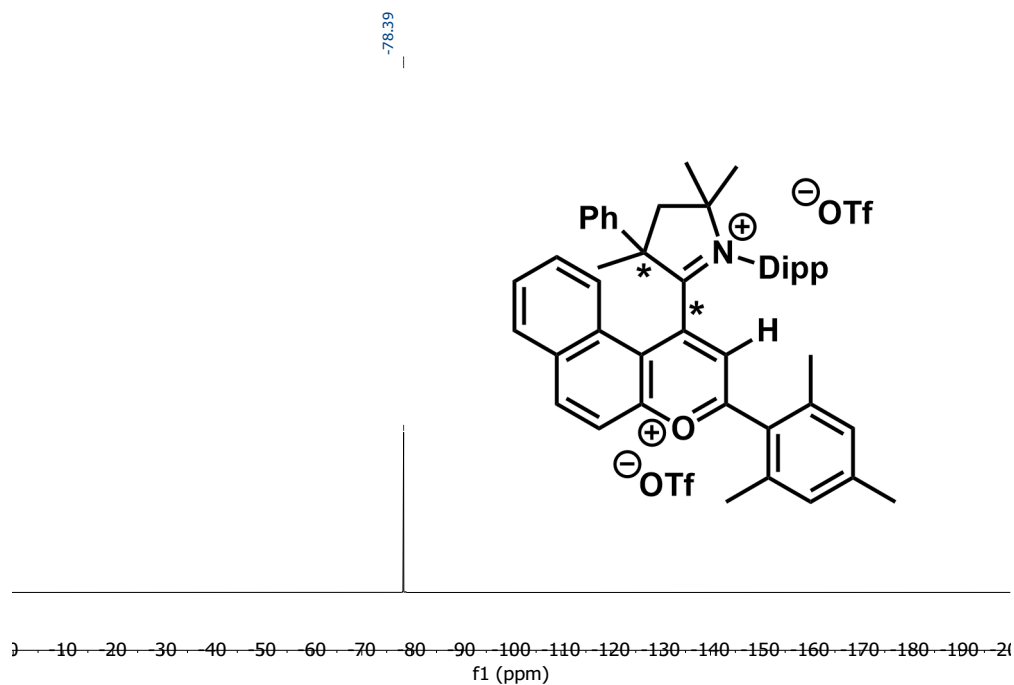
$^1\text{H}$  NMR (400 MHz, 25°C, DCM-d<sub>2</sub>) of [*rac*-3b<sup>++</sup>][OTf]<sub>2</sub>



$^{13}\text{C}\{^1\text{H}\}$  NMR (101 MHz, 25°C, DCM-d<sub>2</sub>) of [*rac*-3b<sup>++</sup>][OTf]<sub>2</sub>

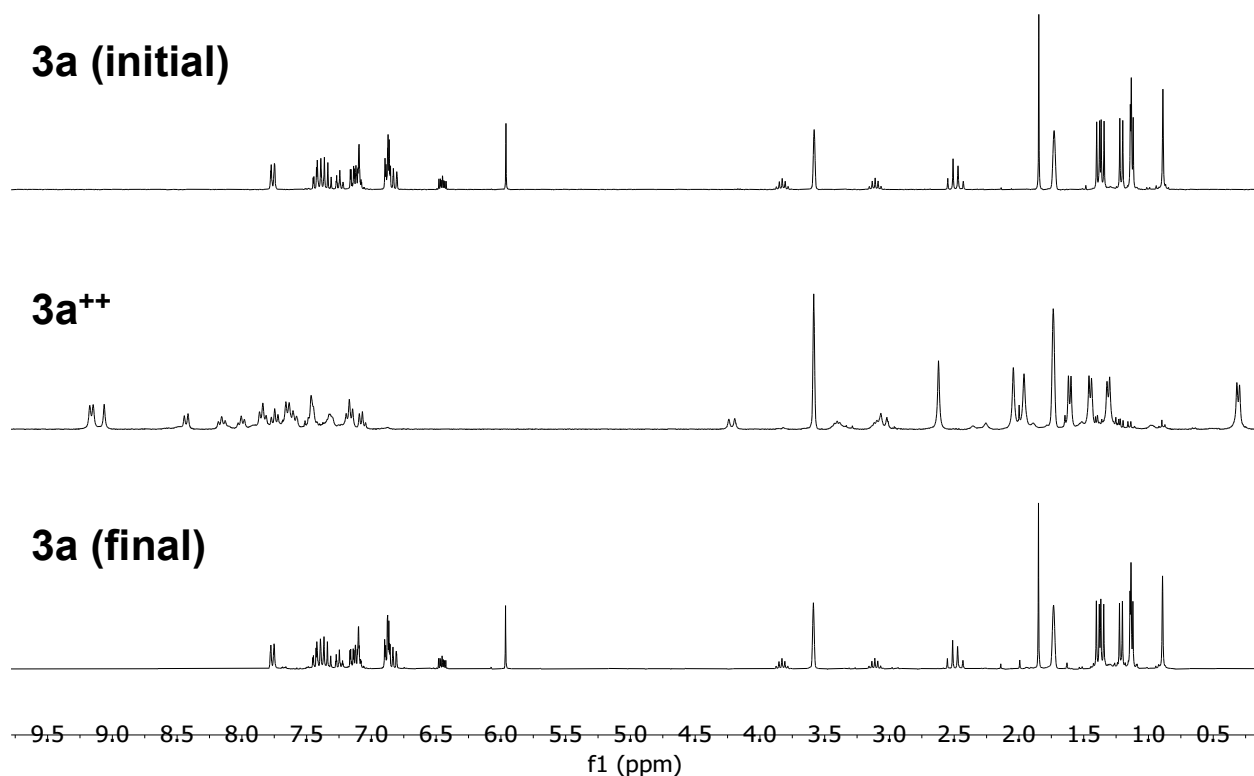


$^{19}\text{F}$  NMR (471 MHz, 25°C, DCM-d<sub>2</sub>) of [*rac*-3b<sup>++</sup>][OTf]<sub>2</sub>



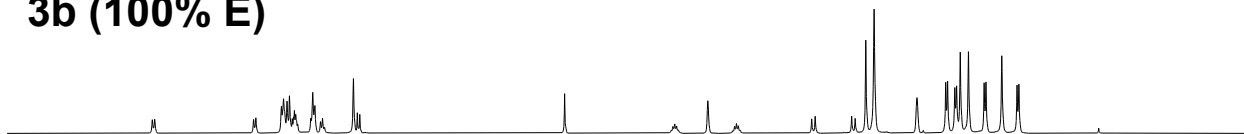
## Chemical Switching Experiments

A solution of the appropriate alkene was prepared in THF- $d_8$ . An initial  $^1\text{H}$  NMR spectrum was taken. 2.2 eq. of AgOTf were added the sample was sonicated until the color change was complete (ca. 10 minutes). Solids were allowed to settle, and a second  $^1\text{H}$  NMR spectrum was taken. Then, 2 eq. of  $\text{Zn}^0$  powder were added. The sample was sonicated until the color had fully reverted (ca. 1 h for **3a**, overnight for **3b**), after which solids were allowed to settle and the final  $^1\text{H}$  NMR spectrum was taken.

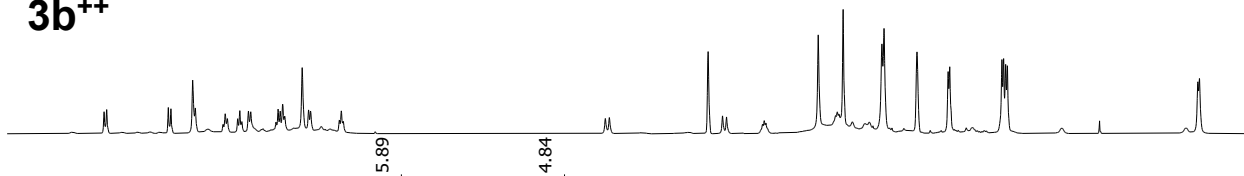


**Figure S5.**  $^1\text{H}$  NMR (300 MHz, 25°C, THF- $d_8$ ) of **3a**  $\rightarrow$  **3a<sup>++</sup>**  $\rightarrow$  **3a**

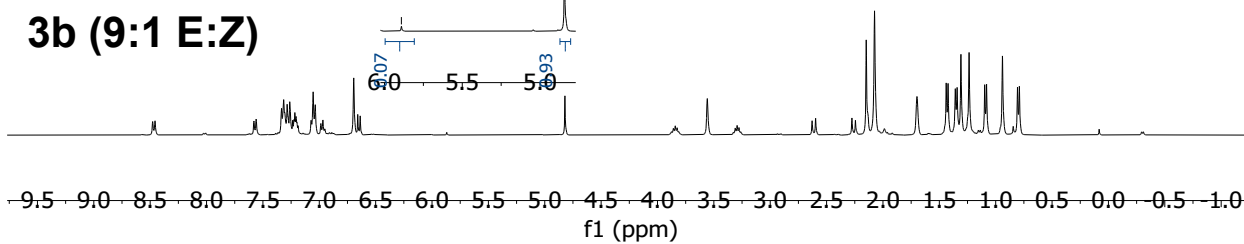
**3b (100% E)**



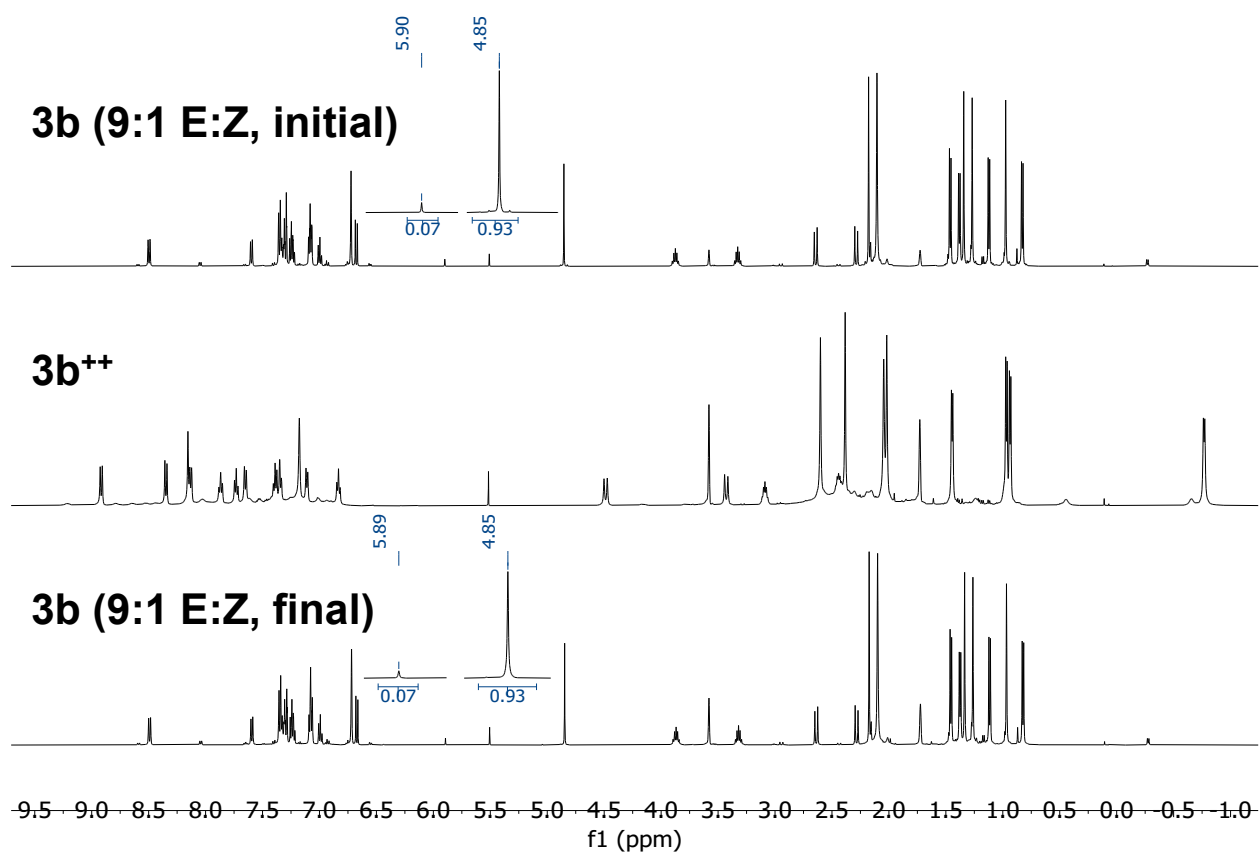
**3b<sup>++</sup>**



**3b (9:1 E:Z)**



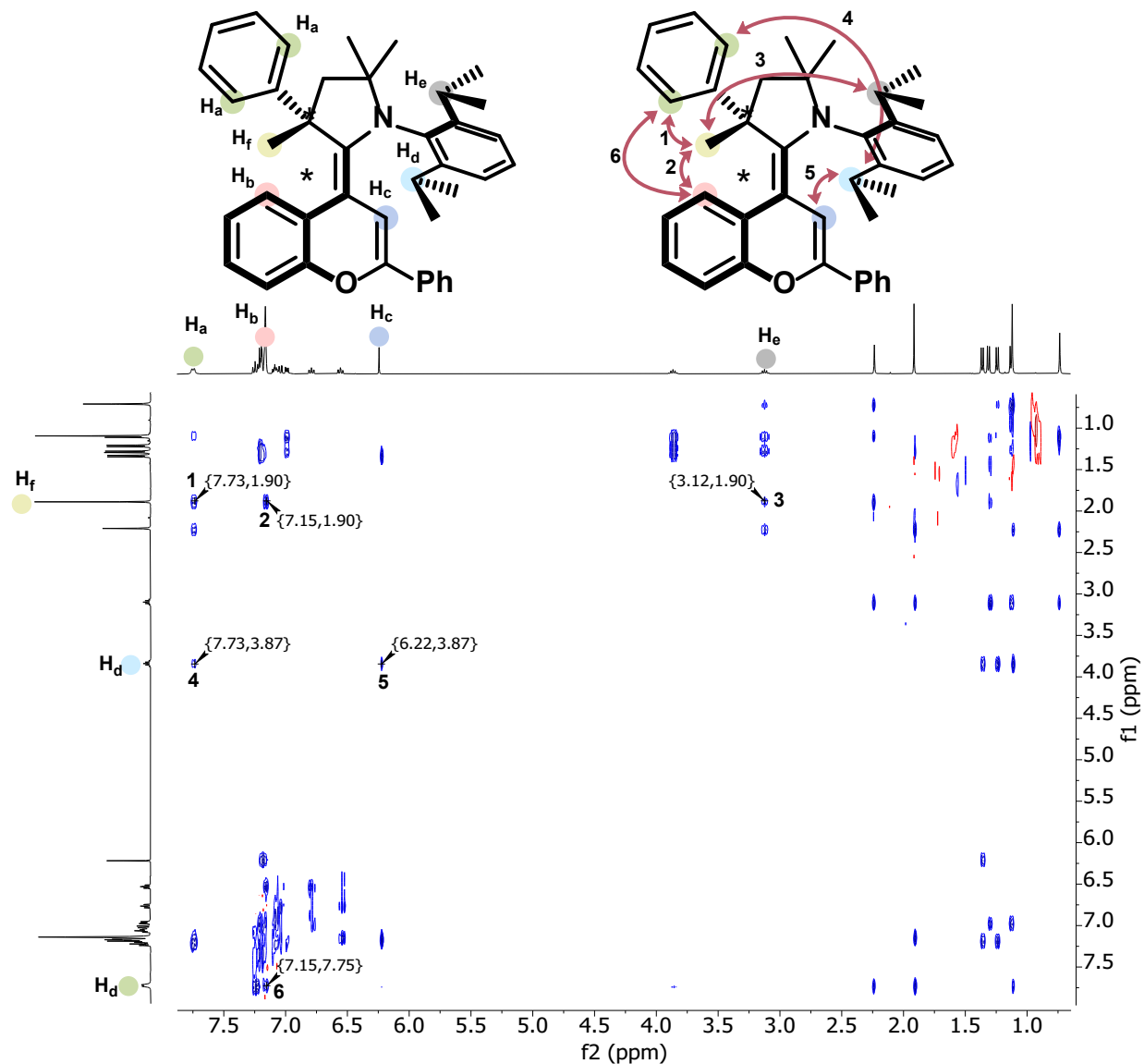
**Figure S6.** <sup>1</sup>H NMR (400 MHz, 25°C, THF-d<sub>8</sub>) of **3b (100% E)** → **3b<sup>++</sup>** → **3b (9:1 E:Z)**



**Figure S7.**  $^1\text{H}$  NMR (500 MHz, 25°C, THF- $\text{d}_8$ ) of **3b (9:1 E:Z)**  $\rightarrow$  **3b<sup>++</sup>**  $\rightarrow$  **3b (9:1 E:Z)**

## 2D NMR Analysis and Spectra

### Determining the olefinic and planar stereochemistry of **3a** by $^1\text{H}$ - $^1\text{H}$ NOESY analysis



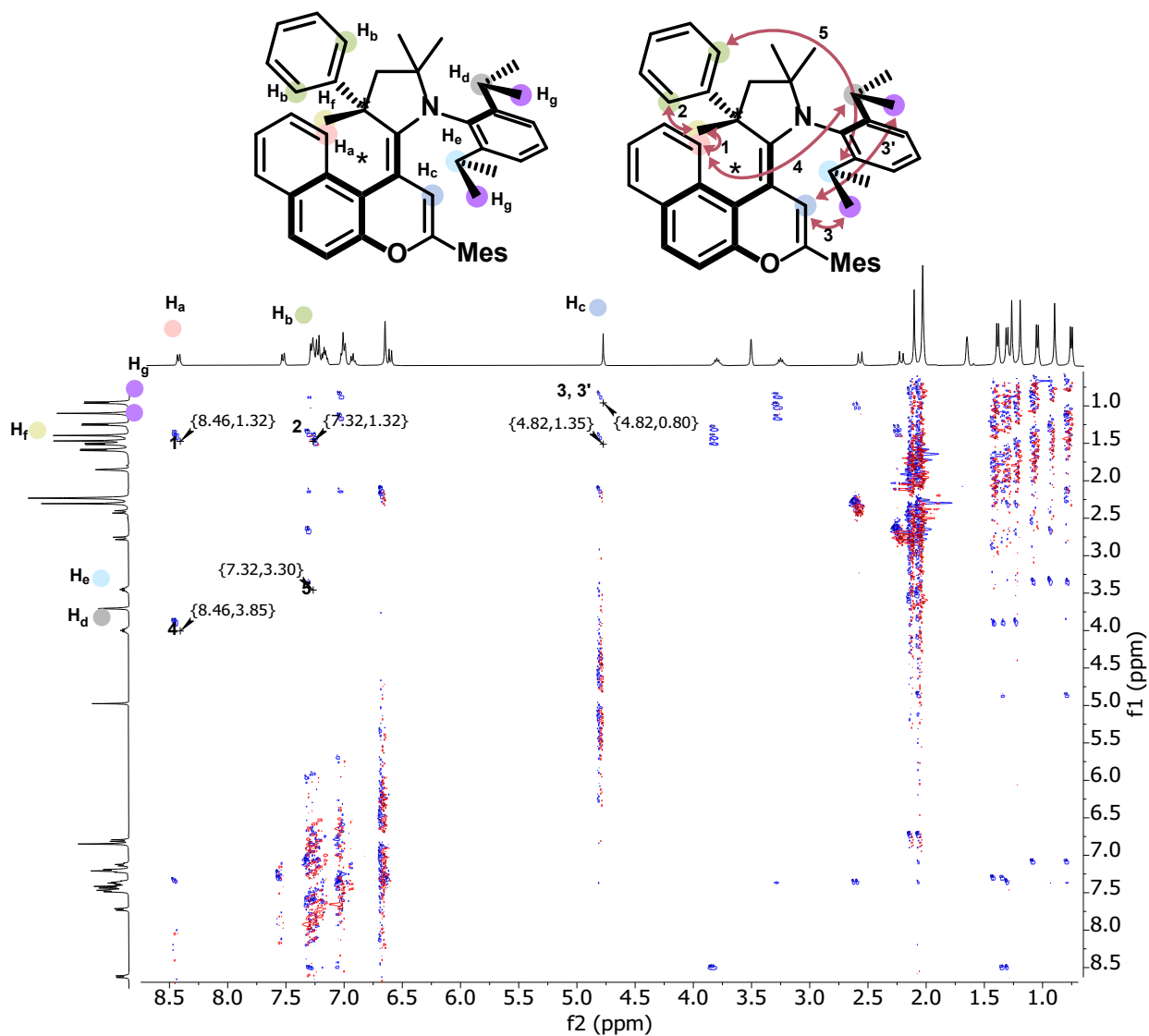
**Figure S8.**  $^1\text{H}$ - $^1\text{H}$  NOESY NMR (400 MHz, 25°C,  $\text{C}_6\text{D}_6$ ) of **3a**

The E-configuration of the central olefinic bond is unambiguously determined by the correlation (5) between the olefinic proton,  $\text{H}_c$  (6.22 ppm), and  $\text{H}_a$  (3.87 ppm), one of the two tertiary protons of the N-Dipp isopropyl groups. This configuration is corroborated by the absence of any correlations between  $\text{H}_c$  and  $\text{H}_f$  (1.90 ppm), the methyl substituent of the stereogenic carbon.  $\text{H}_f$  is unambiguously identified as such by the correlation (3) with  $\text{H}_a$  (7.73 ppm), the o-protons of the phenyl substituent on the stereogenic carbon.

Correlations between the substituents on the stereogenic carbon and the two isopropyl septets allows clear differentiation of the corresponding isopropyl groups as being either *syn* or *anti* to the methyl substituent on the stereogenic carbon. Specifically, the correlation (4) between **H<sub>a</sub>** and **H<sub>d</sub>** reveals **H<sub>d</sub>** as *anti*, while the correlation (3) between **H<sub>f</sub>** and **H<sub>e</sub>** (3.12 ppm) reveals **H<sub>f</sub>** as *syn*.

The relative configuration of the stereogenic plane is determined by the absolute configuration of the stereogenic carbon center. The interior aryl C-H proton, **H<sub>b</sub>** (7.15 ppm), is poorly resolved in the 1D spectrum. However, the well-resolved correlation (2) between **H<sub>f</sub>** and another aryl signal besides **H<sub>a</sub>** enables **H<sub>b</sub>** to be located in the 2D spectrum; besides **H<sub>a</sub>**, **H<sub>b</sub>** (7.15 ppm) is the only aryl proton which could be expected to correlate with **H<sub>f</sub>**. Based on the correlations of **H<sub>b</sub>** to both **H<sub>a</sub>** (6) and **H<sub>f</sub>** (2), we propose that the solution state conformation of **4a** resembles that which is observed in the crystal state. Furthermore, the olefinic proton, **H<sub>c</sub>**, only correlates (5) to one of the two isopropyl septets. In the crystal structure of **3a**, **H<sub>c</sub>** is tilted towards **H<sub>d</sub>** and away from **H<sub>e</sub>**. This is corroborated by computational modeling of **3a** in the solution state, which predicts the conformation observed in the crystal structure to be thermodynamically preferred in the solution state. Notably, in the higher energy planar configuration, **H<sub>c</sub>** is directed *away* from **H<sub>e</sub>**. Thus, the absolute configurations of the enantiomers present in racemic **3a** are (**S**,**M<sub>p</sub>**) and (**R**,**P<sub>p</sub>**).

## Determining the olefinic and planar stereochemistry of **3b** by $^1\text{H}$ - $^1\text{H}$ NOESY analysis



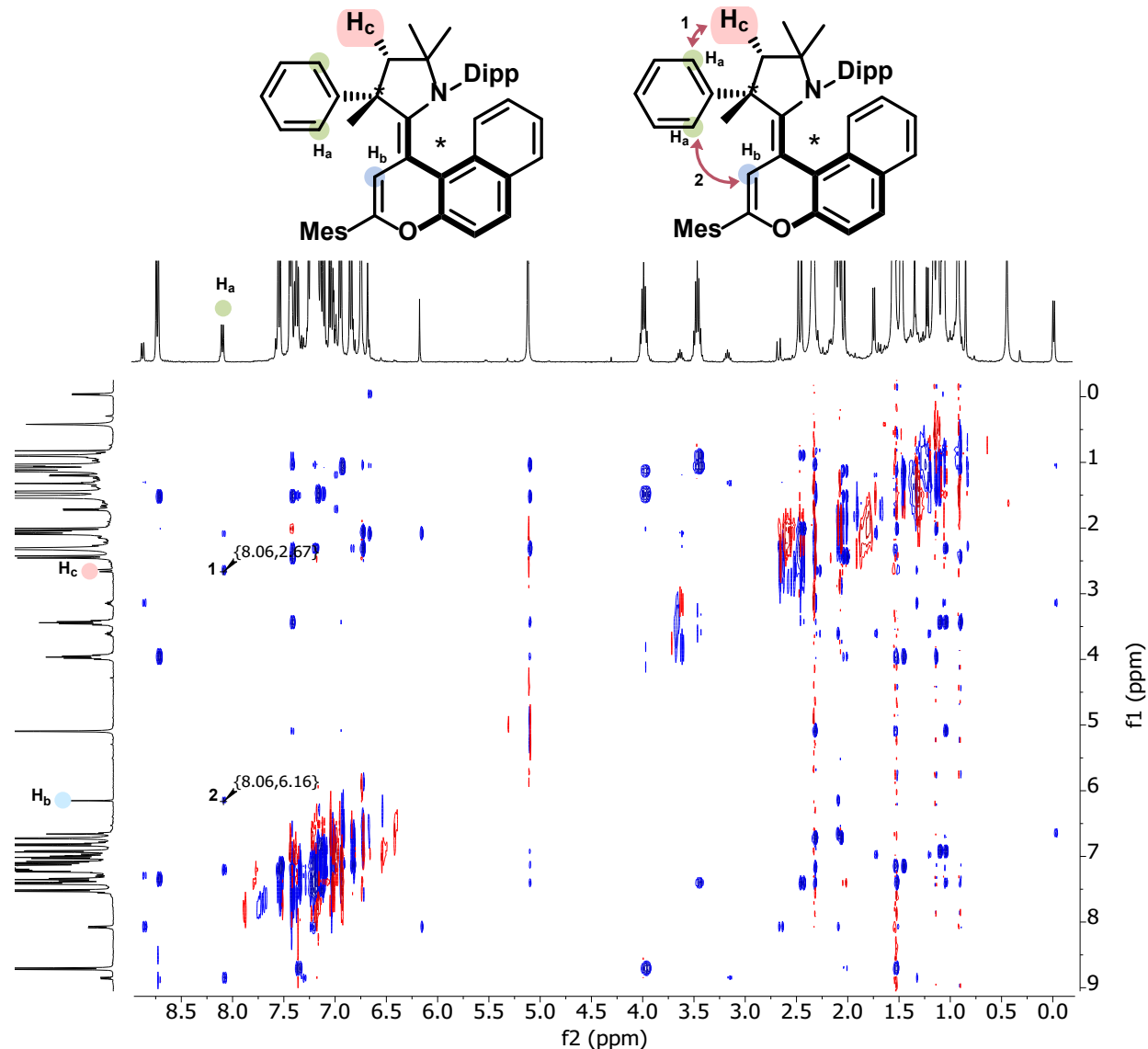
**Figure S9.**  $^1\text{H}$ - $^1\text{H}$  NOESY NMR (400 MHz, 25°C,  $\text{THF-d}_8$ ) of **3b**

The E-configuration of the central olefinic bond is unambiguously determined by correlations (**3**, **3'**) between the olefinic proton, **H<sub>c</sub>** (4.82 ppm), and one of the isopropyl methyls **H<sub>g</sub>** (1.35 ppm, 0.80 ppm). This configuration is corroborated by the absence of any correlations between **H<sub>c</sub>** and **H<sub>f</sub>** (1.32 ppm), the methyl substituent of the stereogenic carbon. **H<sub>f</sub>** is unambiguously identified as such by the correlation (**2**) with **H<sub>b</sub>** (7.32 ppm), the o-protons of the phenyl substituent on the stereogenic carbon. Furthermore, the “bay position” aryl C-H proton, **H<sub>a</sub>** (8.46 ppm), correlates (**1**) with **H<sub>f</sub>**, supporting this assignment.



The relative configuration of the stereogenic plane is determined by the absolute configuration of the stereogenic carbon center. **H<sub>a</sub>** correlates (1) with **H<sub>f</sub>** as well as (4) **H<sub>d</sub>** (3.85 ppm), one of the two isopropyl septets. The other isopropyl septet, **H<sub>e</sub>** (3.30 ppm), correlates (5) with **H<sub>b</sub>**. Given that **H<sub>a</sub>** and **H<sub>b</sub>** correlate with distinct isopropyl septets, **H<sub>a</sub>** and **H<sub>b</sub>** must lie on opposing sides of the stereogenic carbon center. Thus, we propose that the solution state conformation of **3b** resembles that which is observed in the crystal state; the absolute configurations of the enantiomers present in racemic **3b** are (**S**,**P<sub>p</sub>**) and (**R**,**M<sub>p</sub>**).

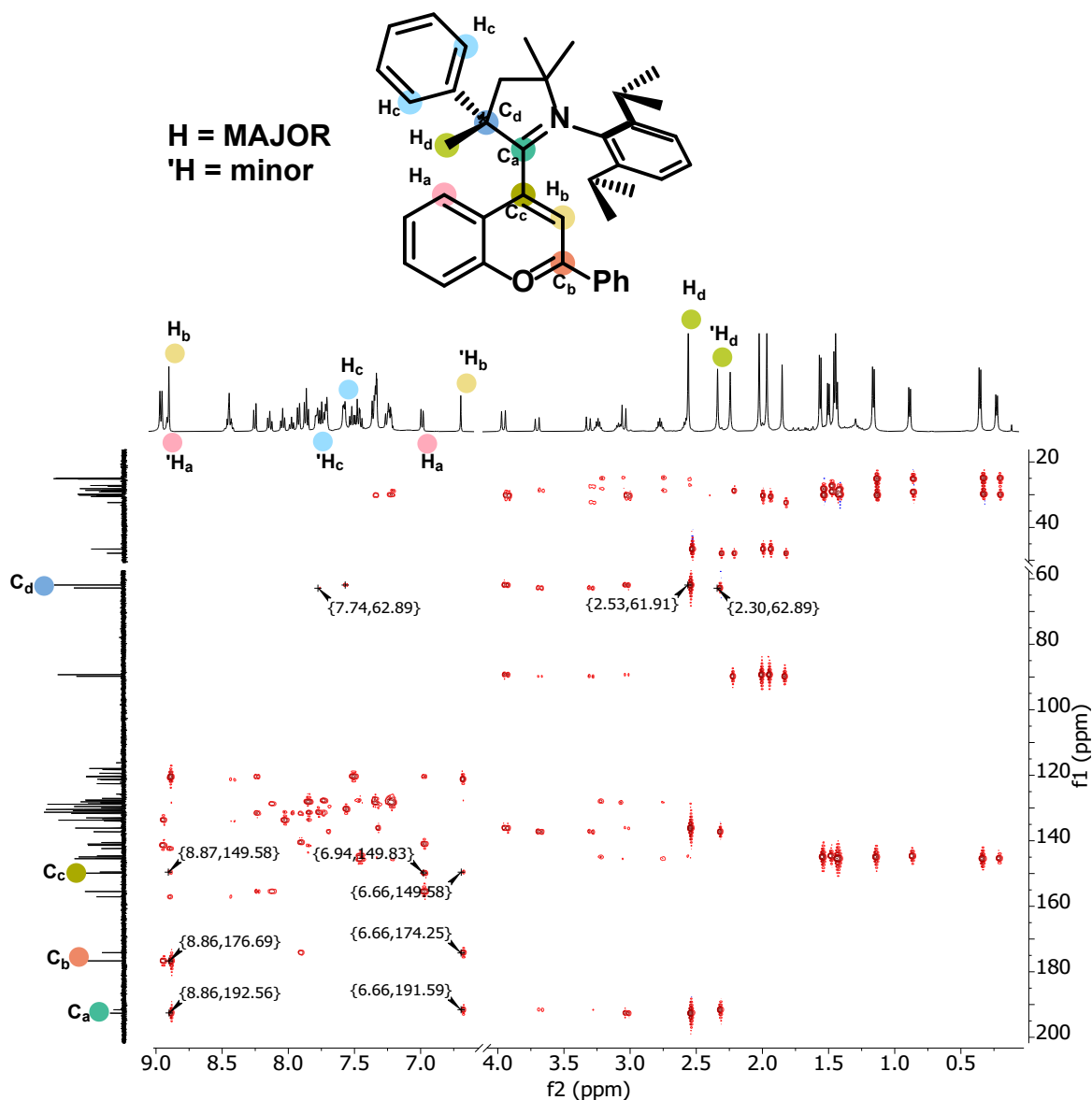
## Identifying the minor Z-olefin stereoisomer of **3b** by $^1\text{H}$ - $^1\text{H}$ NOESY analysis



**Figure S10.**  $^1\text{H}$ - $^1\text{H}$  NOESY NMR (400 MHz, 25°C,  $\text{C}_6\text{D}_6$ ) of 9:1 E:Z mixture of **3b**

The Z-configuration of the central olefinic bond in the minor stereoisomer of **3b** is unambiguously determined by the correlation (2) between the olefinic proton,  $\text{H}_b$  (6.16 ppm), and the o-phenyl protons  $\text{H}_a$  (8.06 ppm). The assignment of  $\text{H}_a$  is supported by the correlation (1) between this doublet and one of the two diastereotopic methylene protons,  $\text{H}_c$  (2.67) ppm. Correlation 2 requires that  $\text{H}_b$  and  $\text{H}_a$  be *cis* in reference to the central olefinic bond.

## Determining the axial stereochemistry of atropisomeric $3a^{++}$ by $^1\text{H}$ - $^1\text{H}$ NOESY analysis



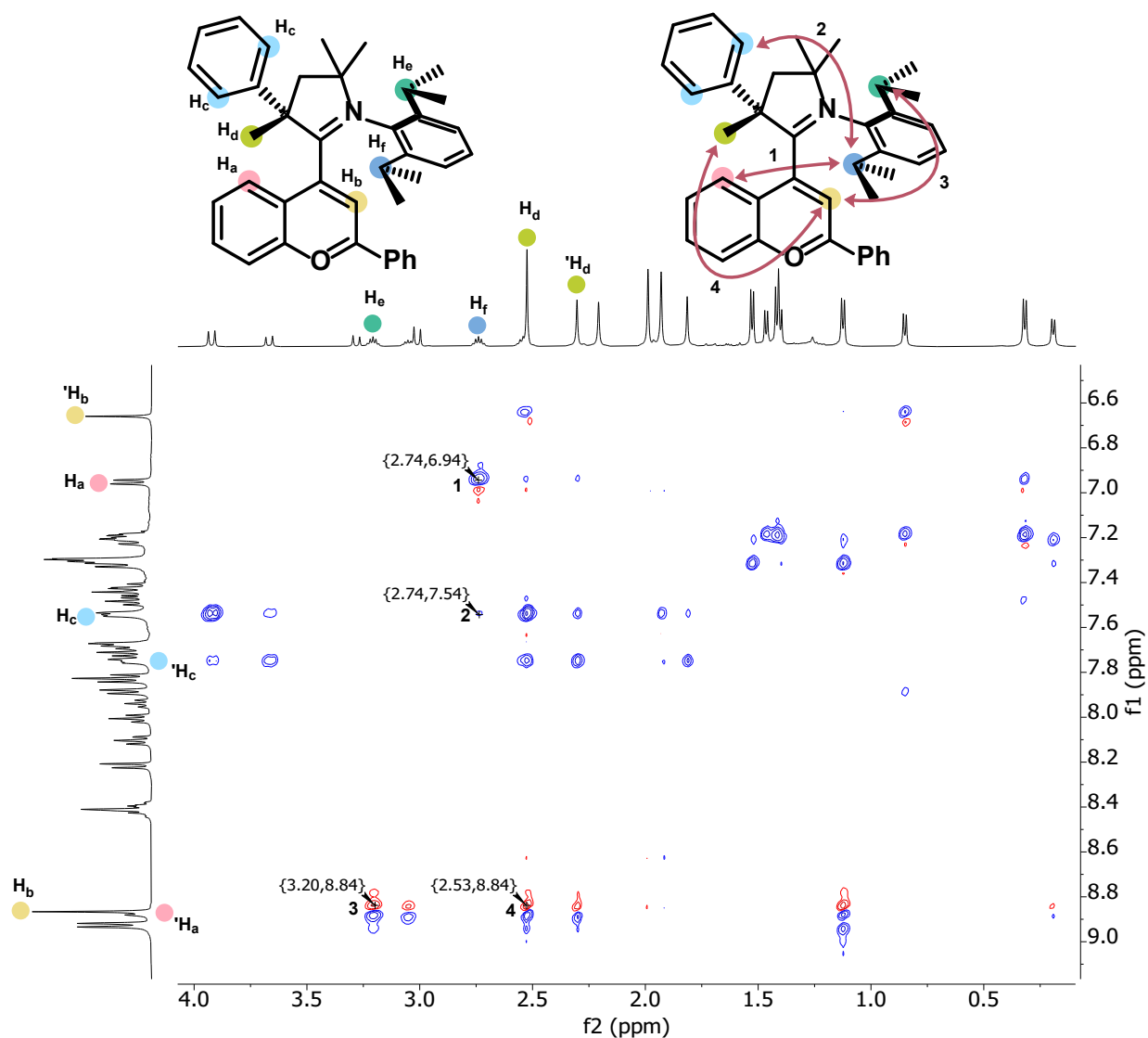
**Figure S11.**  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR (500 MHz, 25°C, DCM- $d_2$ ) of  $[3a^{++}][\text{OTf}]_2$ ; formal charges and anions omitted for clarity.

HMBC was used to assign key proton resonances for both the major and minor atropisomers:

- The *ortho* protons, H<sub>c</sub> (7.74 ppm, minor; 7.53 ppm, MAJOR) of the phenyl substituent of the stereogenic carbon C<sub>d</sub> (61.91 ppm, MAJOR; 62.89 ppm, minor),
- The protons for the methyl substituent of the same carbon, H<sub>d</sub> (2.53 ppm, MAJOR; 2.30 ppm, minor).

- The “interior” aryl protons of the chromenylium ring:  $H_a$  (8.87 ppm, MAJOR; 6.94 ppm, minor), which presents as a doublet, and  $H_b$  (8.86 ppm, MAJOR; 6.66 ppm, minor), which presents as a singlet.

With these resonances located, it is possible to assign the configuration of the central stereogenic axis using NOESY.



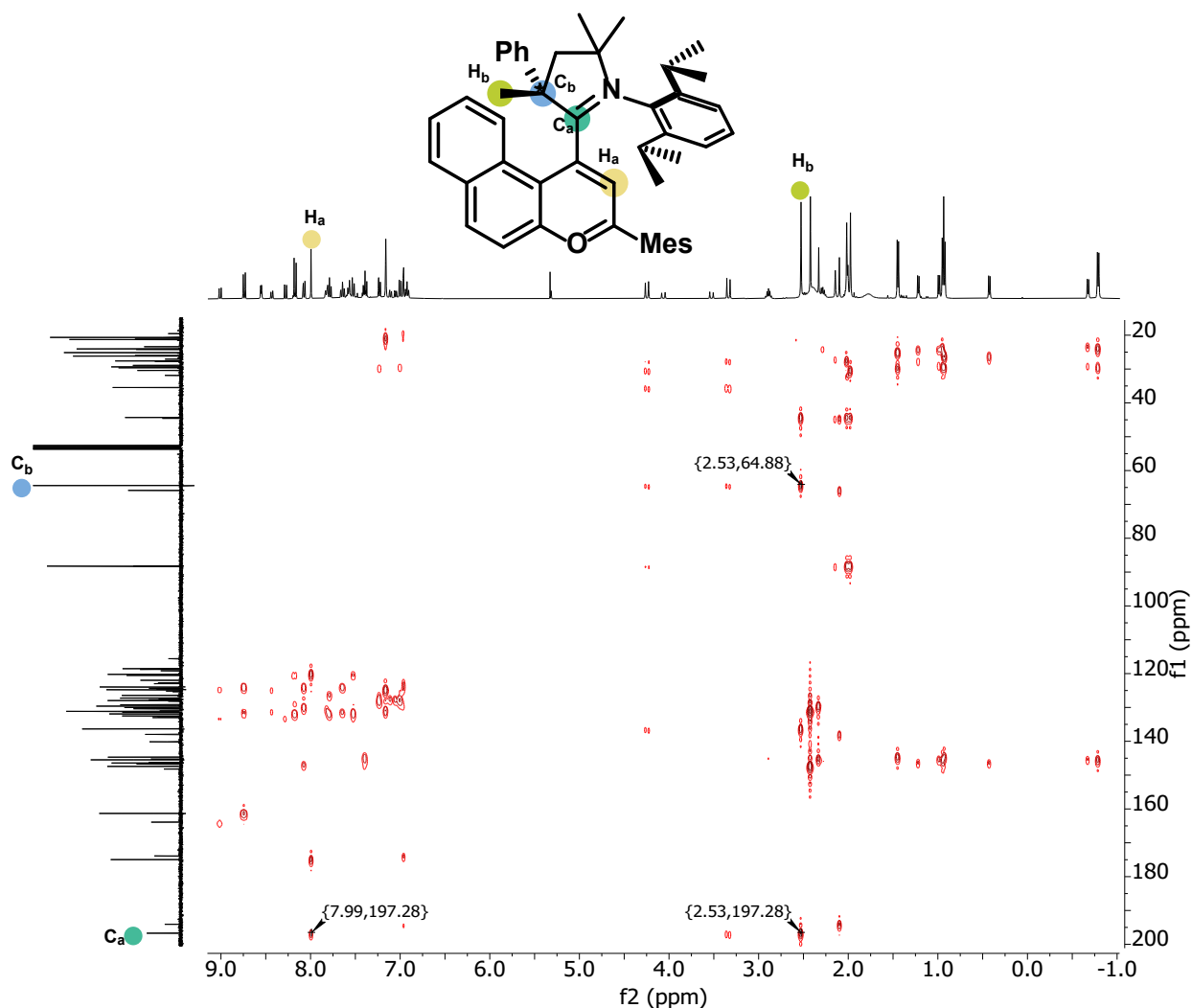
**Figure S12.**  $^1\text{H}$ - $^1\text{H}$  NOESY NMR (400 MHz, 25°C, DMC- $d_2$ ) of  $[3a^{++}][\text{OTf}]_2$ ; formal charges and anions omitted for clarity.

The *syn* and *anti* isopropyl protons were assigned based on the correlation (2) between  $H_f$  (2.74 ppm) and  $H_a$ . Thus,  $H_f$  is *syn* to the phenyl substituent of the stereogenic carbon center and *anti* to the corresponding methyl. Consequently, the fact that  $H_a$  correlates (1) to  $H_f$  necessitates that these, too, are *syn*, and therefore that the chromenylium ring is

directed towards the phenyl substituent. This is corroborated by correlations (**3**, **4**) of the methyl substituent protons, **H<sub>a</sub>**, and the other isopropyl proton, **H<sub>e</sub>** (3.20 ppm) with **H<sub>b</sub>**, which lay on the opposite side of the chromenylium ring as **H<sub>a</sub>**. Based upon this analysis, the absolute configurations of the major atropisomer are (**S**,**M<sub>a</sub>**) and (**R**,**P<sub>a</sub>**), while the minor are (**S**,**P<sub>a</sub>**) and (**R**,**M<sub>a</sub>**).

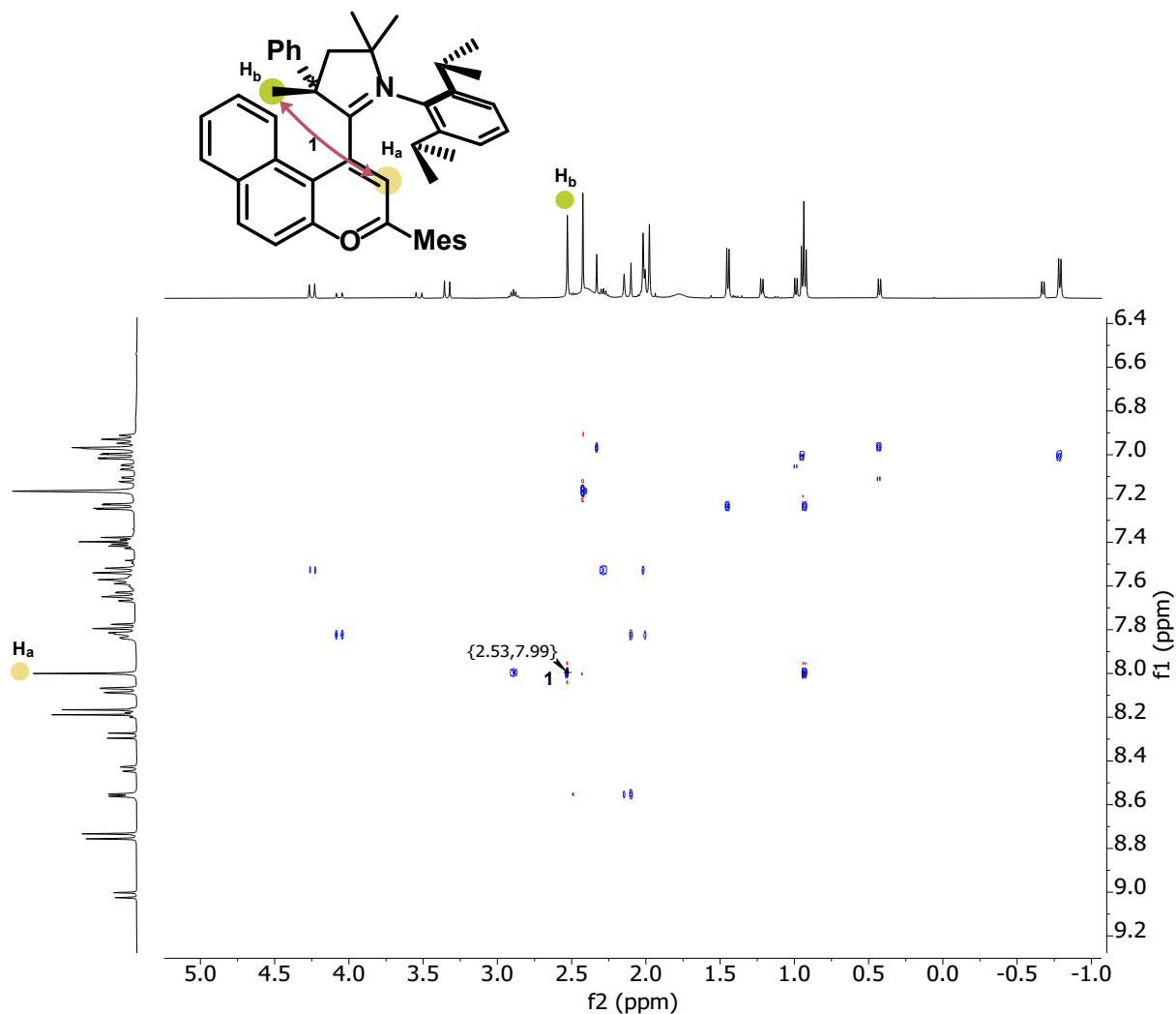
### Determining the axial stereochemistry of atropisomeric **3b<sup>++</sup>** by <sup>1</sup>H-<sup>1</sup>H NOESY analysis

HMBC was used to assign key proton resonances for the major atropisomer. Namely, the interior aryl proton singlet, **H<sub>a</sub>** (7.99 ppm), and the methyl substituent of the stereogenic carbon center, **H<sub>b</sub>** (2.53 ppm).



**Figure S13.** <sup>1</sup>H-<sup>13</sup>C HMBC NMR (500 MHz, 25°C, DCM-d<sub>2</sub>) of [**3b<sup>++</sup>**][OTf]<sub>2</sub>; formal charges and anions omitted for clarity.

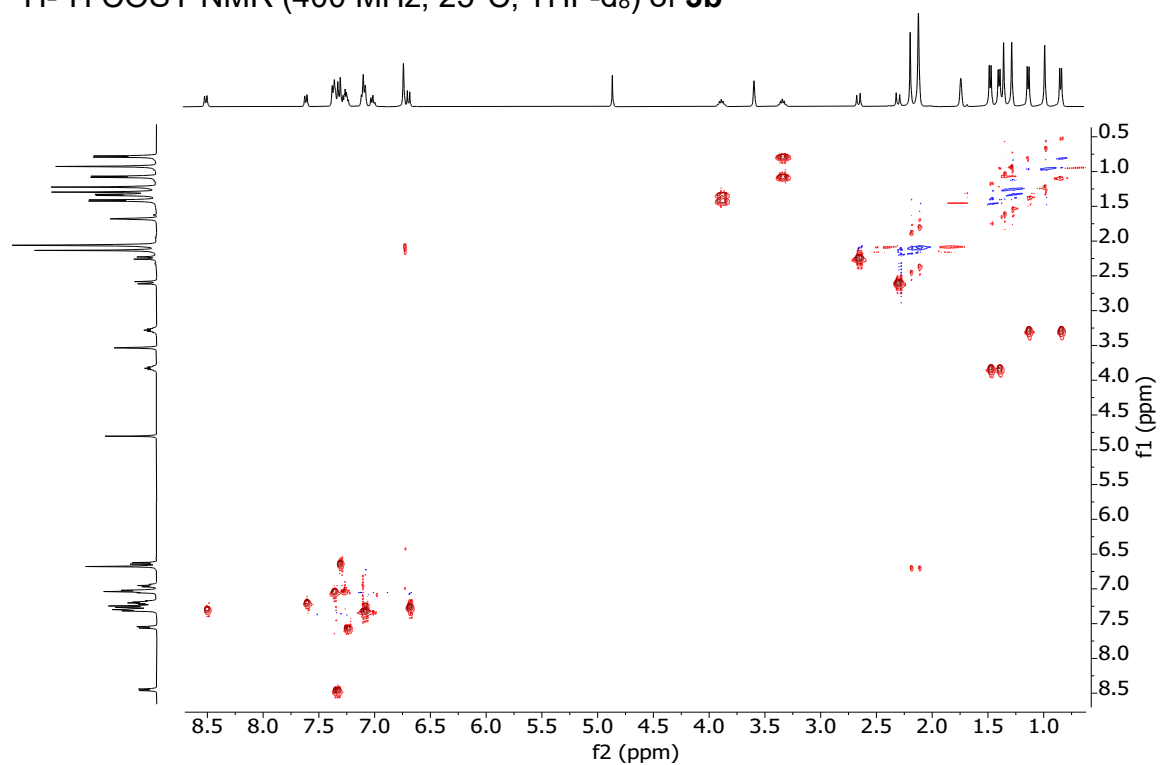
With these resonances located, it is possible to assign the configuration of the central stereogenic axis using NOESY.



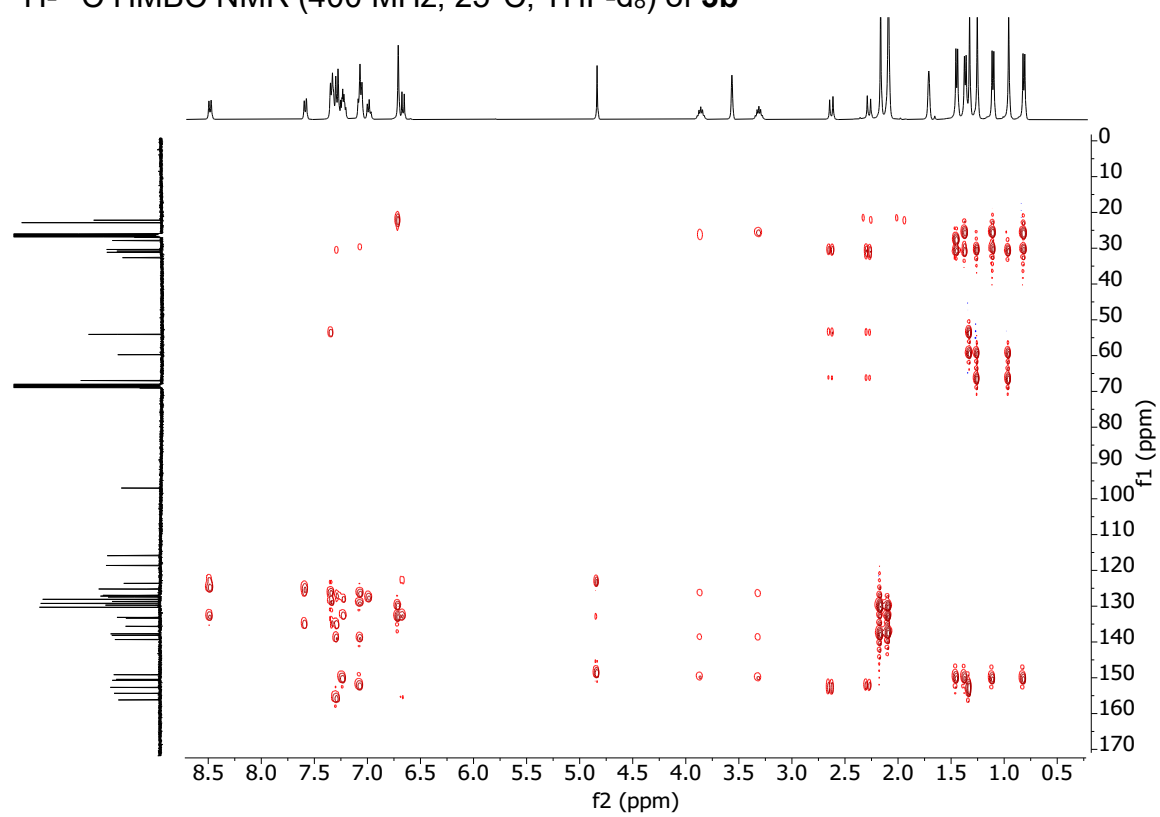
**Figure S14.**  $^1\text{H}$ - $^1\text{H}$  NOESY NMR (400 MHz, 25°C, DCM- $d_2$ ) of  $[\mathbf{3b}^{++}][\text{OTf}]_2$ ; formal charges and anions omitted for clarity.

The correlation (**1**) between  $\text{H}_a$  and  $\text{H}_b$  necessitates that these are *syn*. Therefore, the benzochromenylium ring is directed towards the phenyl substituent. This configuration of the stereogenic axis is like that which is observed in the solid state. Based upon this analysis, the absolute configurations of the major atropisomer are (*S*,*M<sub>a</sub>*) and (*R*,*P<sub>a</sub>*), while the minor are (*S*,*P<sub>a</sub>*) and (*R*,*M<sub>a</sub>*).

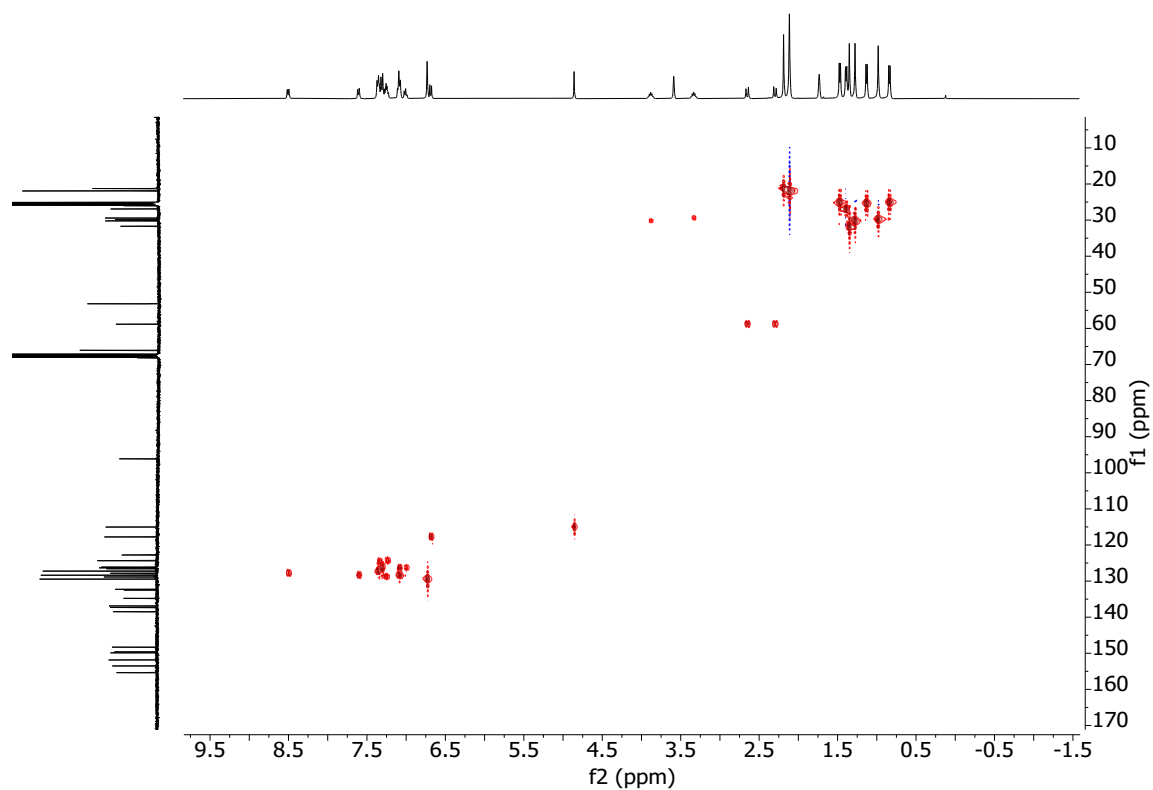
$^1\text{H}$ - $^1\text{H}$  COSY NMR (400 MHz, 25°C, THF- $\text{d}_8$ ) of **3b**



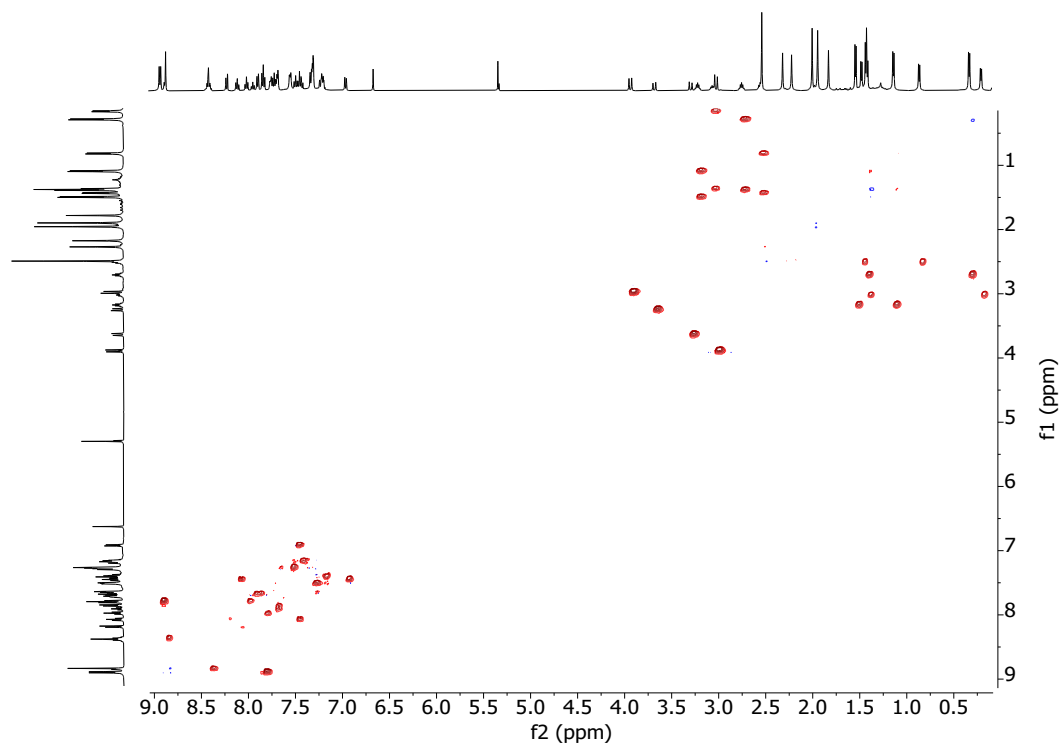
$^1\text{H}$ - $^{13}\text{C}$  HMBC NMR (400 MHz, 25°C, THF- $\text{d}_8$ ) of **3b**



$^1\text{H}$ - $^{13}\text{C}$  HMQC NMR (400 MHz, 25°C, THF- $\text{d}_8$ ) of **3b**

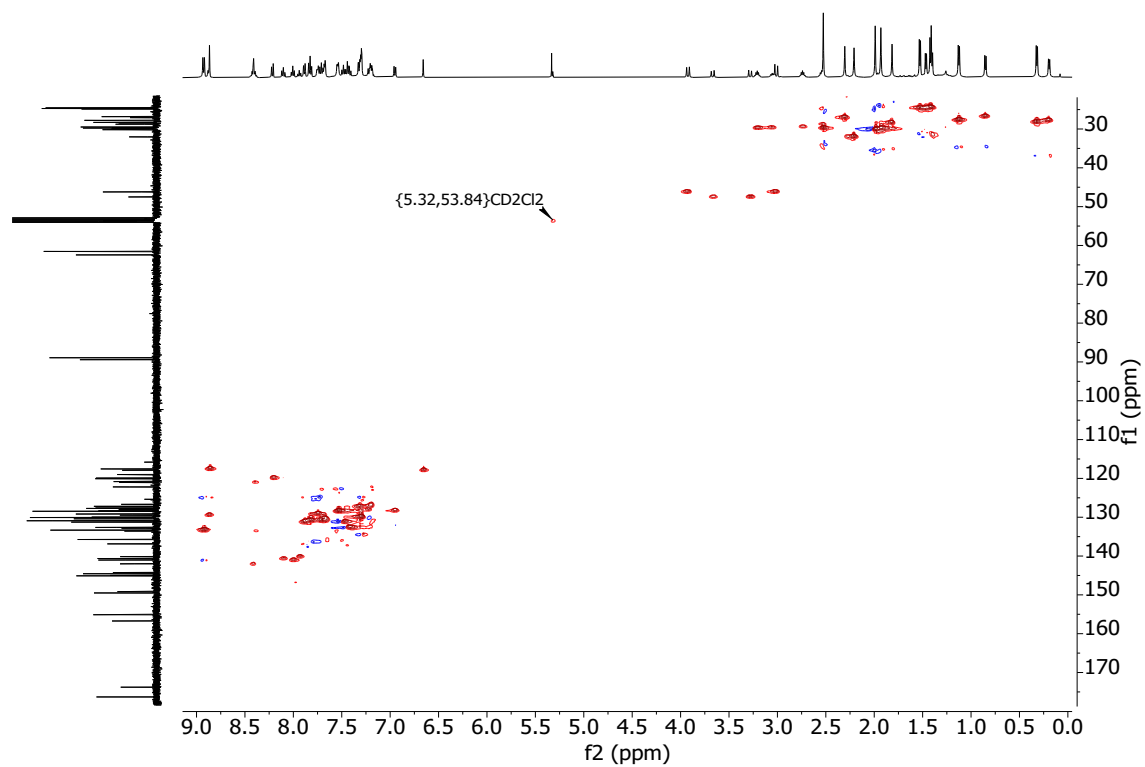


$^1\text{H}$ - $^1\text{H}$  COSY NMR (400 MHz, 25°C, DCM- $\text{d}_2$ ) of **[3a $^{++}$ ][OTf] $_2$**

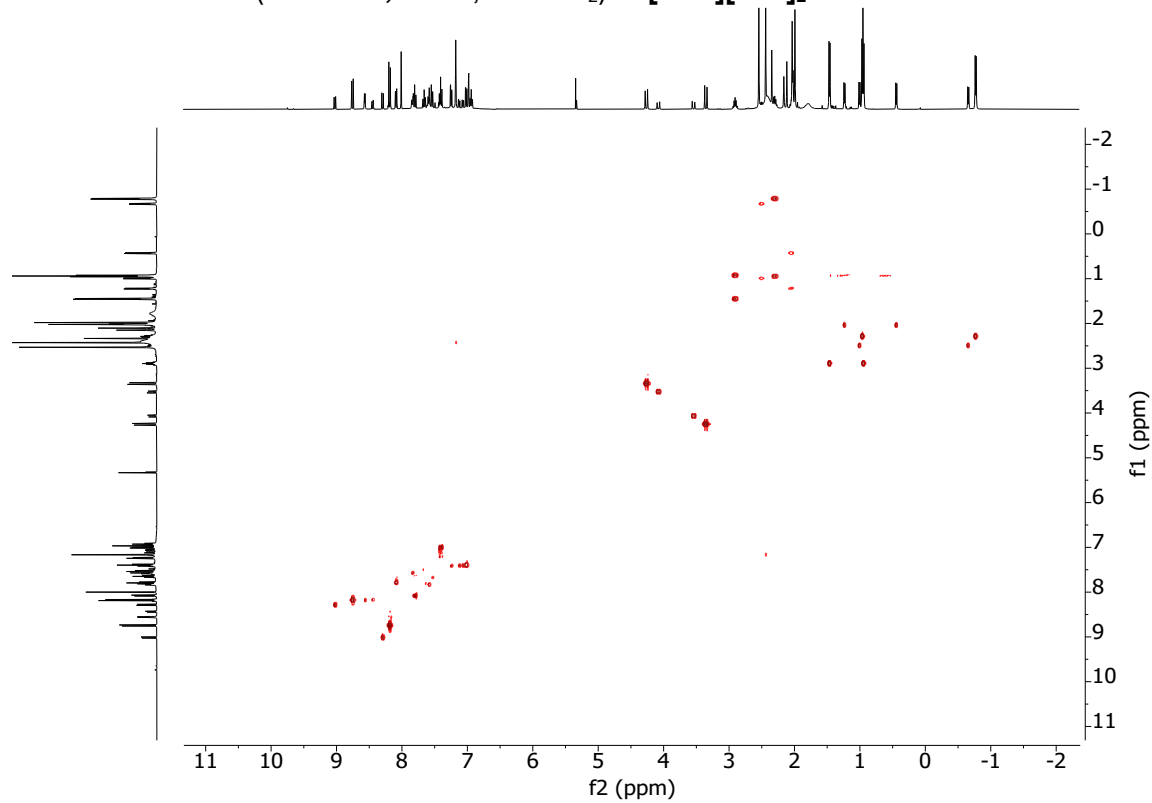




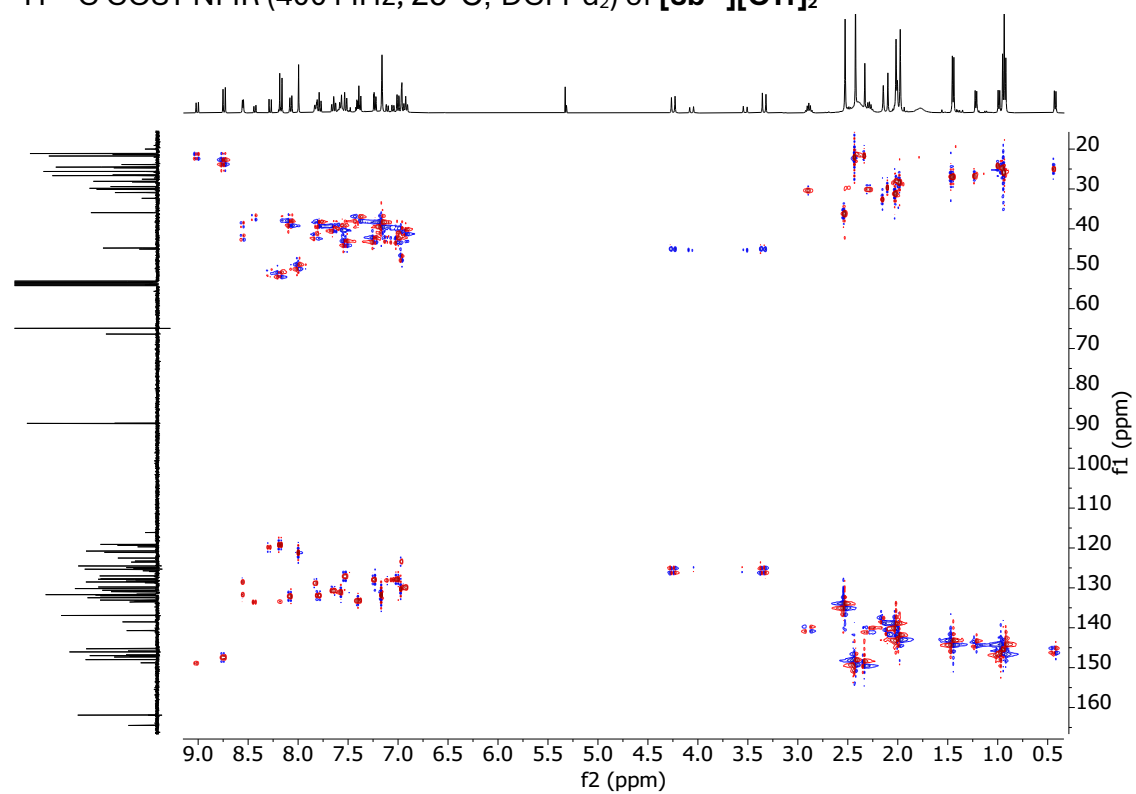
$^1\text{H}$ - $^{13}\text{C}$  HSQC NMR (400 MHz, 25°C,  $\text{DCM-d}_2$ ) of  $[\mathbf{3a}^{++}][\text{OTf}]_2$



$^1\text{H}$ - $^1\text{H}$  COSY NMR (400 MHz, 25°C,  $\text{DCM-d}_2$ ) of  $[\mathbf{3b}^{++}][\text{OTf}]_2$

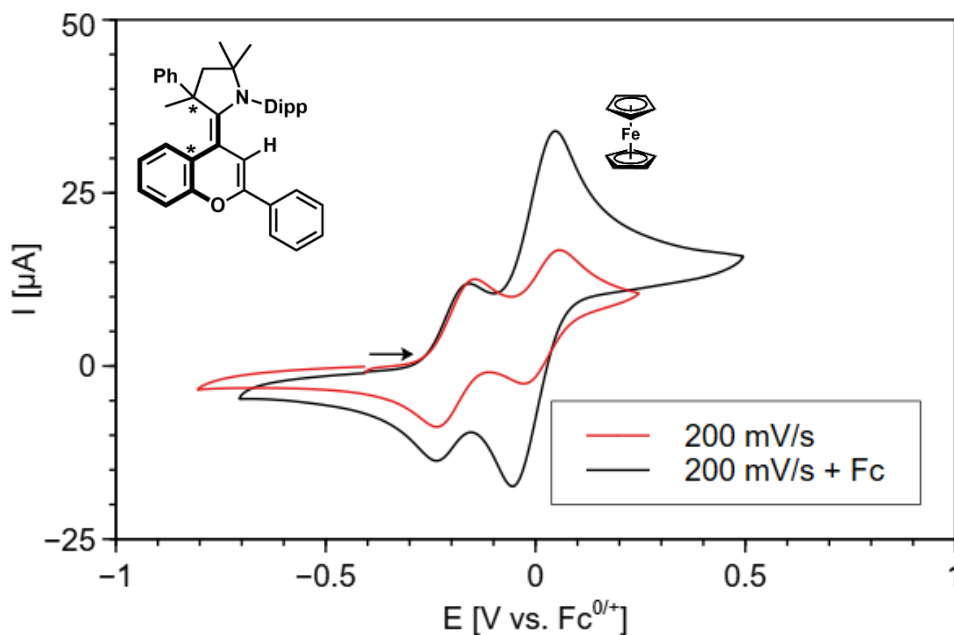


$^1\text{H}$ - $^{13}\text{C}$  COSY NMR (400 MHz, 25°C, DCM- $\text{d}_2$ ) of  $[\mathbf{3b}^{++}][\text{OTf}]_2$

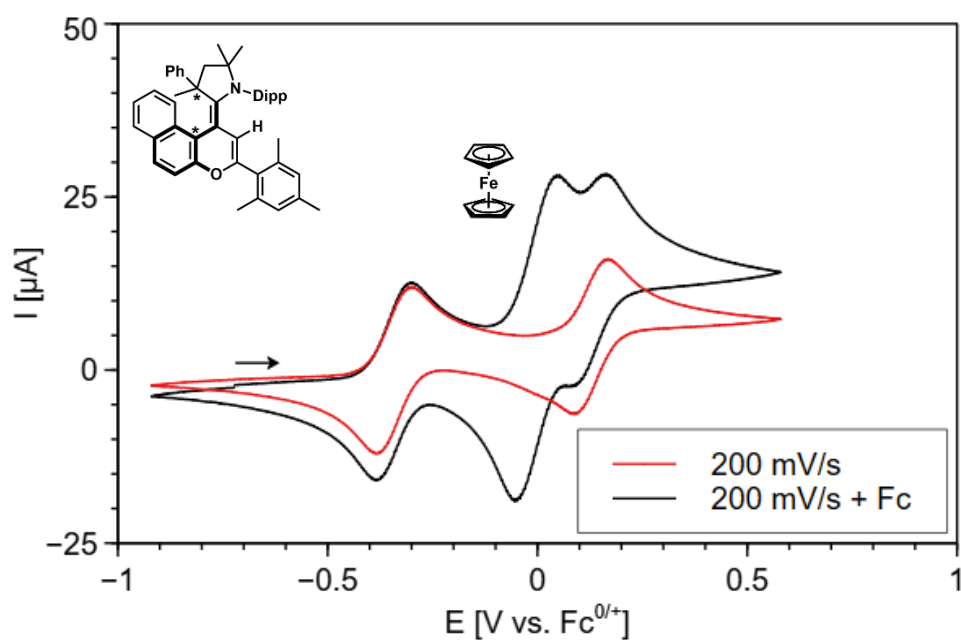


#### 4) Cyclic Voltammetry

Electrochemical measurements were performed with a potentiostat-galvanostat AutoLab PGSTAT 302N controlled by resident GPES (General Purpose Electrochemical System 4.9) software using a conventional single-compartment three-electrode cell. The working and auxiliary electrodes were platinum electrodes and the reference electrode was the saturated potassium chloride calomel electrode (SCE). The supporting electrolyte was 0.1 M Bu<sub>4</sub>NPF<sub>6</sub> (tetrabutylammonium hexafluorophosphate) in dichloromethane and solutions were purged with argon before the measurements. All potentials are quoted relative to Fc<sup>0/+</sup>.



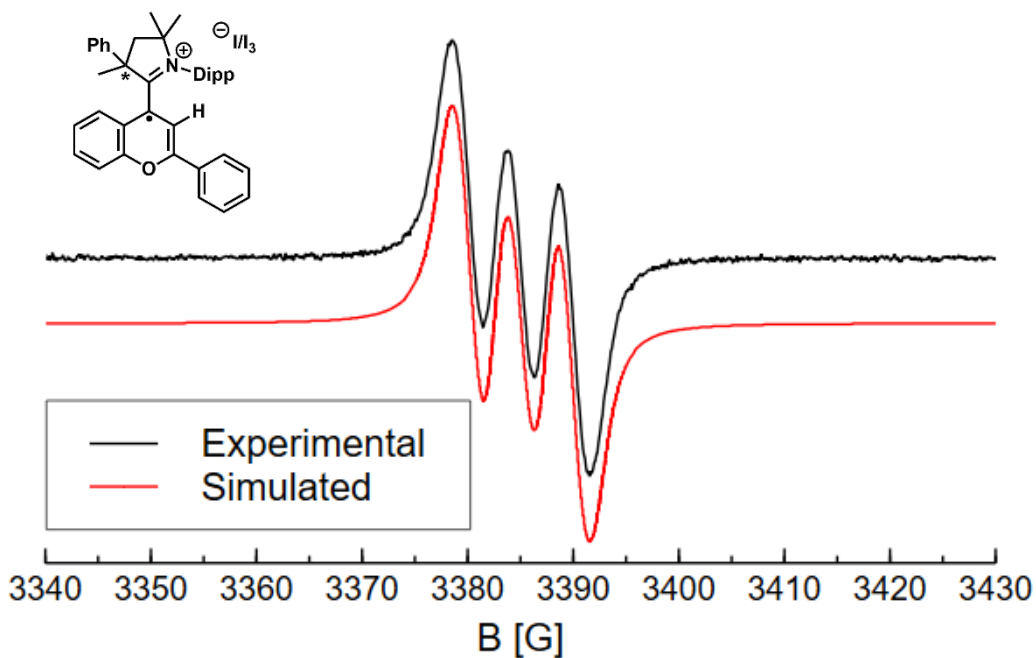
**Figure S15.** Cyclic voltammograms of **3a** (~0.5 mg/mL) in DCM (0.1 M [NBu<sub>4</sub>][PF<sub>6</sub>]) at room temperature with and without ferrocene as an internal standard. Scan rate of 200 mV/s (arrow indicates scanning direction).



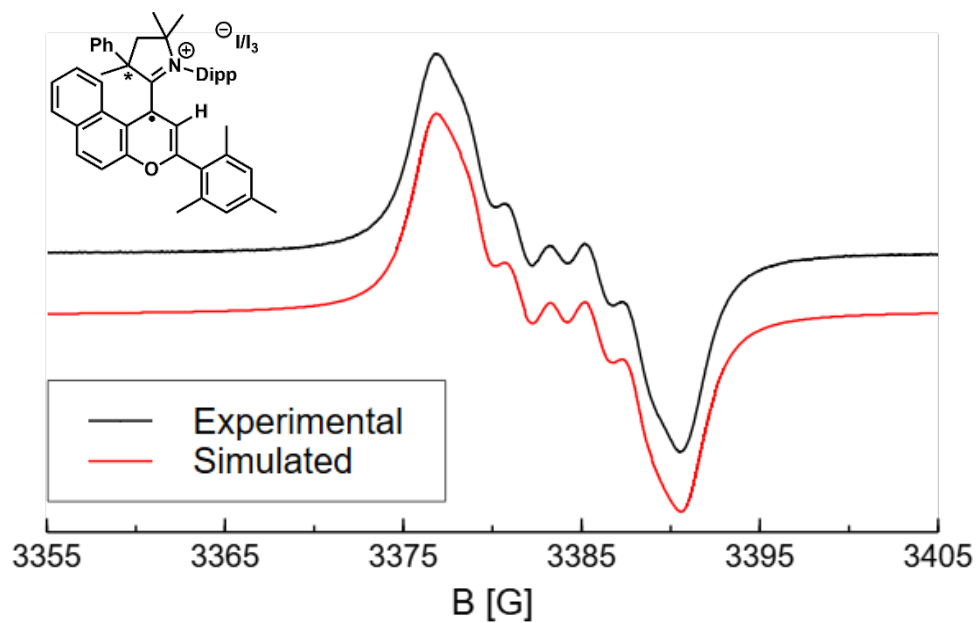
**Figure S16.** Cyclic voltammograms of **3b** (~0.5 mg/mL) in DCM (0.1 M  $[\text{NBu}_4][\text{PF}_6]$ ) at room temperature with and without ferrocene as an internal standard. Scan rate of 200 mV/s (arrow indicates scanning direction).

## 5) EPR Measurements

Samples for EPR analysis were prepared by *in situ* oxidation of the corresponding neutral hybrids using iodine in DCM (~ 1  $\mu$ M). To interpret the experimental spectra, simulated spectra were obtained by parameter fitting using WinSim2002. Input values for hyperfine coupling constants were obtained from density functional theory calculations (See “Hyperfine Coupling Constants and Spin Densities”).



**Figure S17.** X-Band EPR spectrum of **3a<sup>++</sup>** in DCM (1  $\mu$ M).  $g = 2.0037$ ;  $LW = 1.603$ ; Hyperfine coupling constants:  $1xN$ : 13.5 MHz;  $1xH$ : 4.0 MHz;  $1xH$ : 3.0 MHz;  $1xH$ : 2.2 MHz;  $1xH$ : 2.1 MHz;  $1xH$ : 1.5 MHz.



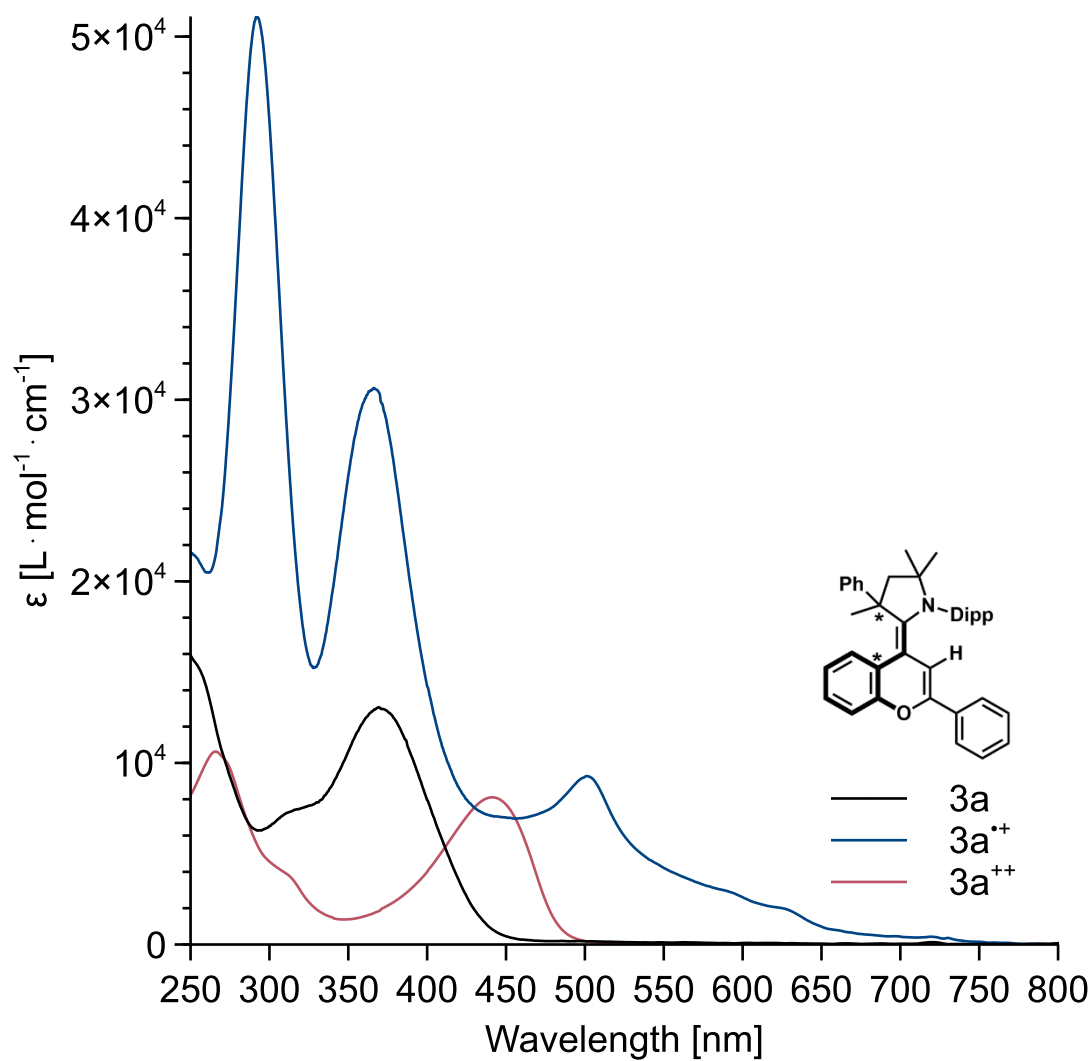
**Figure S18.** X-Band EPR spectrum of **3b**<sup>•+</sup> in DCM (1  $\mu$ M).  $g = 2.0019$ ;  $LW = 1.442$ ; Hyperfine coupling constants: 1xN: 12.2 MHz; 1xH: 6.4 MHz; 1xH: 4.3 MHz; 1xH: 2.4 MHz; 3xH: 0.6 MHz; 1xH: 0.5 MHz.

## 6) UV-Visible Spectroscopy

Radical cations were prepared by *in situ* oxidation with 1 equivalent of iodine, delivered as a solution in DCM. Dications were prepared as solutions with the same concentration in DCM by oxidation with a slight excess of AgSbF<sub>6</sub>, then filtration through a 0.45  $\mu$ m PTFE syringe filter to remove precipitated solids before analysis. Filters should be dried in a desiccator prior to use, to limit decomposition due to hydrolysis from ambient water.

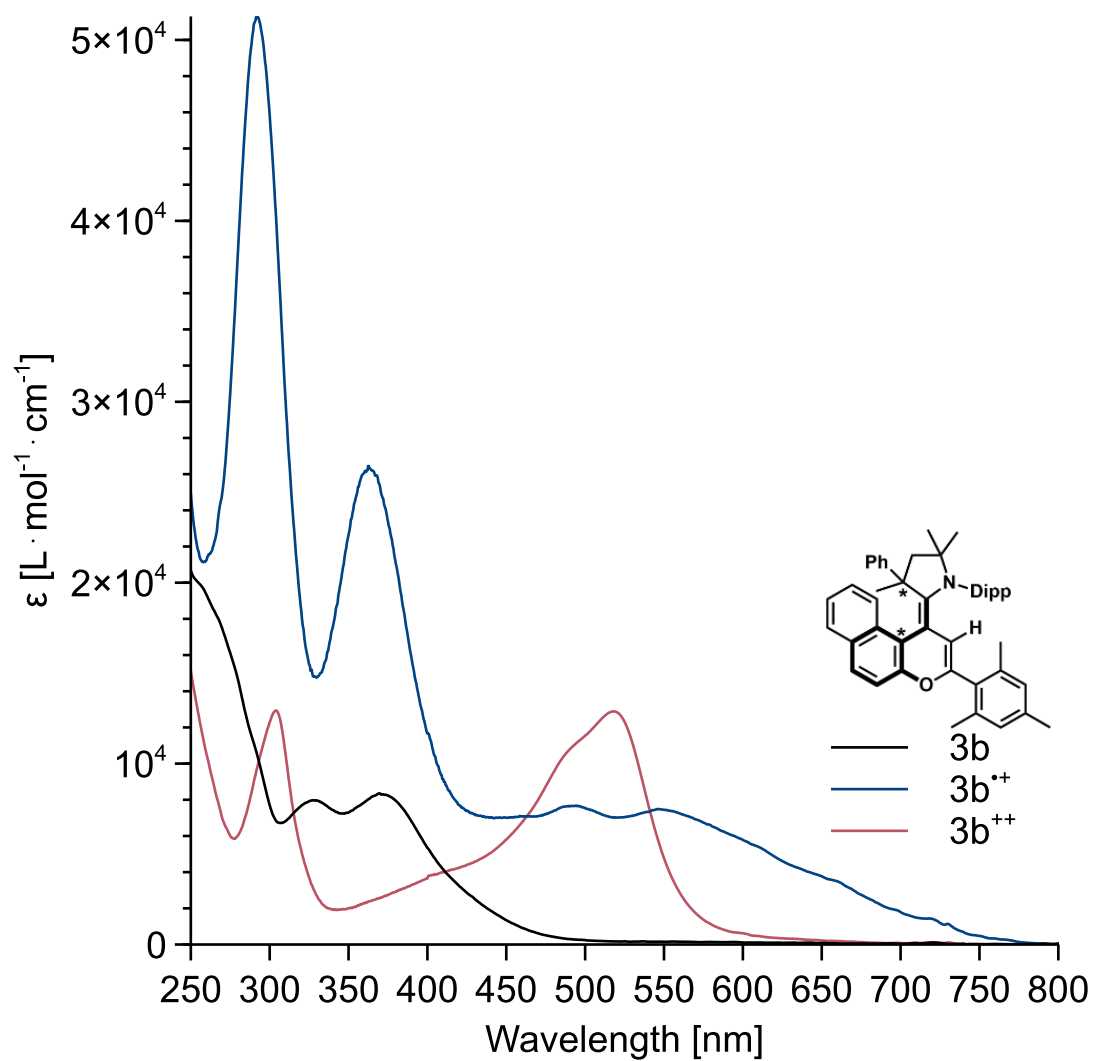
| Compound               | $\lambda_{\text{abs}}$ (nm) | $\epsilon$ (M <sup>-1</sup> ·cm <sup>-1</sup> ) |
|------------------------|-----------------------------|-------------------------------------------------|
| <b>3a</b>              | 370                         | 13055                                           |
| <b>3a<sup>•+</sup></b> | 501                         | 9273                                            |
|                        | 366                         | 30649                                           |
|                        | 292                         | 51109                                           |
| <b>3a<sup>++</sup></b> | 440                         | 8099                                            |
|                        | 266                         | 10629                                           |
|                        |                             |                                                 |
| <b>3b</b>              | 370                         | 8341                                            |
|                        | 328                         | 7977                                            |
| <b>3b<sup>•+</sup></b> | 546                         | 7489                                            |
|                        | 494                         | 7673                                            |
|                        | 363                         | 26350                                           |
|                        | 292                         | 51257                                           |
| <b>3b<sup>++</sup></b> | 518                         | 12885                                           |
|                        | 304                         | 12935                                           |

**Table S2.** Tabulated spectroscopic data for all compounds in DCM.



**Figure S19.** UV-vis spectra of **3a** (black line), **3a<sup>+</sup>** (blue line), and **3a<sup>++</sup>** (red line) in DCM.





**Figure S20.** UV-vis spectra of **3b** (black line), **3b<sup>•+</sup>** (blue line), and **3b<sup>++</sup>** (red line) in DCM.

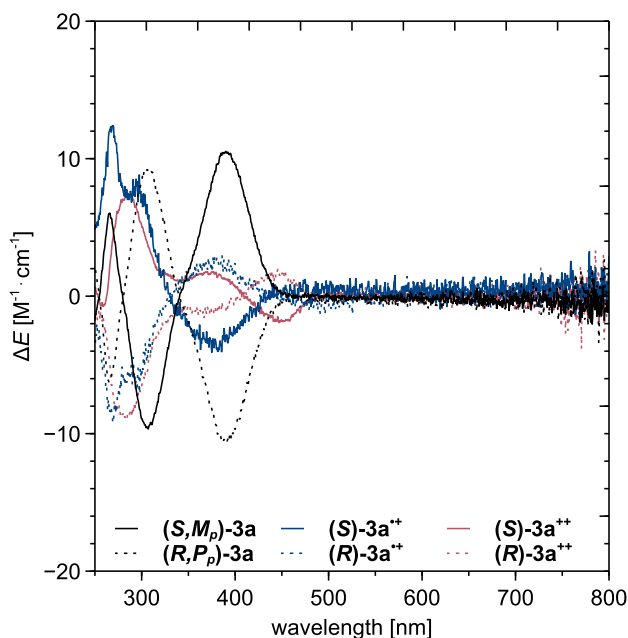
## 7) Electronic Circular Dichroism Spectroscopy

Radical cations were prepared by *in situ* oxidation with 1 equivalent of iodine, delivered as a solution in DCM. Dications were prepared as solutions in DCM by oxidation with a slight excess of  $\text{AgSbF}_6$ , then filtration through a 0.45  $\mu\text{m}$  PTFE syringe filter to remove precipitated solids before analysis. Filters should be dried in a desiccator prior to use to avoid decomposition due to hydrolysis by ambient water.

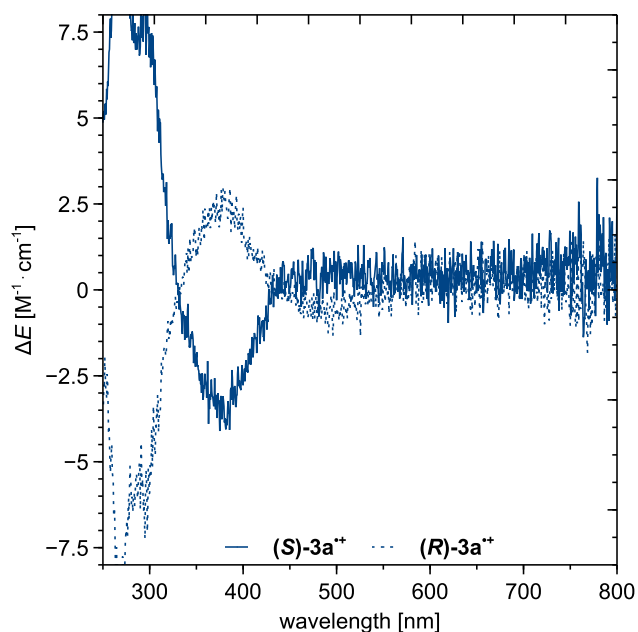
| Compound               | $ g_{\text{abs}} $   | $(\lambda_{\text{abs}})$ |
|------------------------|----------------------|--------------------------|
| <b>3a</b>              | $9.7 \times 10^{-4}$ | (390 nm)                 |
| <b>3a<sup>•+</sup></b> | $1.0 \times 10^{-4}$ | (500 nm)                 |
| <b>3a<sup>2+</sup></b> | $8.8 \times 10^{-5}$ | (450 nm)                 |
| <b>3b</b>              | $2.3 \times 10^{-3}$ | (420 nm)                 |
| <b>3b<sup>•+</sup></b> | $7.0 \times 10^{-4}$ | (665 nm)                 |
| <b>3b<sup>2+</sup></b> | $2.1 \times 10^{-4}$ | (510 nm)                 |

**Table S3.** Experimental absolute absorption dissymmetry factors ( $|g_{\text{abs}}|$ ) for all compounds in DCM.

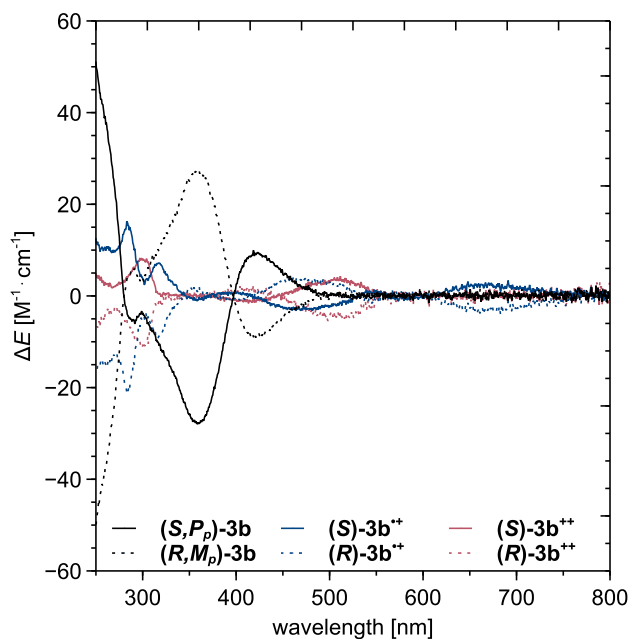
Note, the  $g_{\text{abs}}$  spectra for both enantiomers of **3a<sup>•+</sup>** (Figure S25) shows a weak peak around 500 nm, which is difficult to discern on account of noise. The ECD spectra of the enantiomers of **3a<sup>•+</sup>** alone are included for clarity (Figure S22).



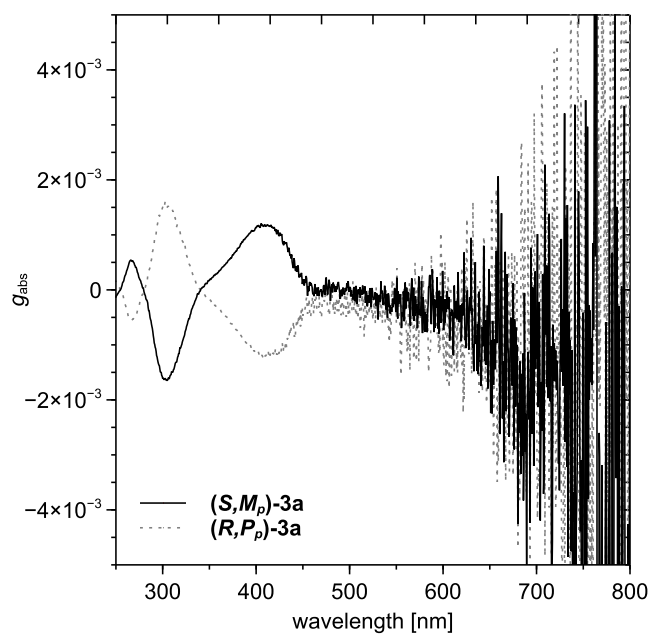
**Figure S21.** Electronic circular dichroism spectra of **(S,M<sub>p</sub>)-3a** (black solid lines), **(R,P<sub>p</sub>)-3a** (black dotted lines), and the corresponding redox series (radical cations and dications, blue and red lines, respectively) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



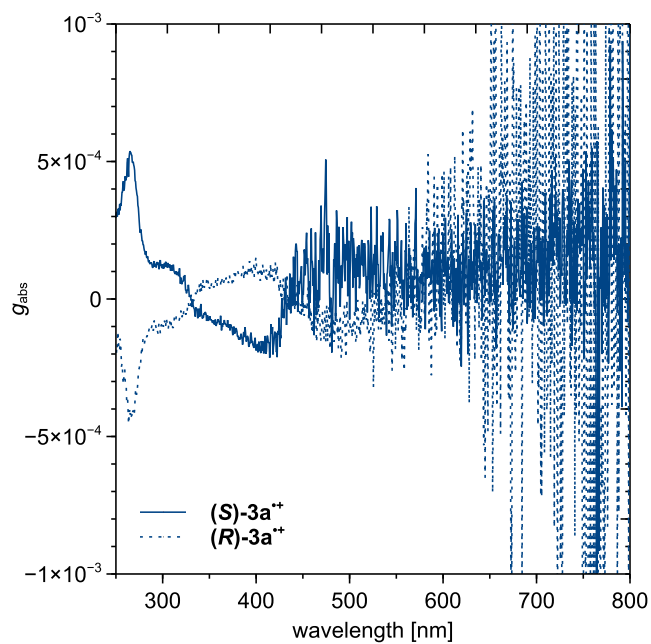
**Figure S22.** Electronic circular dichroism spectra of **(S,M<sub>p</sub>)-3a<sup>++</sup>** (blue solid lines) and **(R,P<sub>p</sub>)-3a<sup>++</sup>** (blue dotted lines), emphasizing the weak transition at ~ 500 nm. Recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



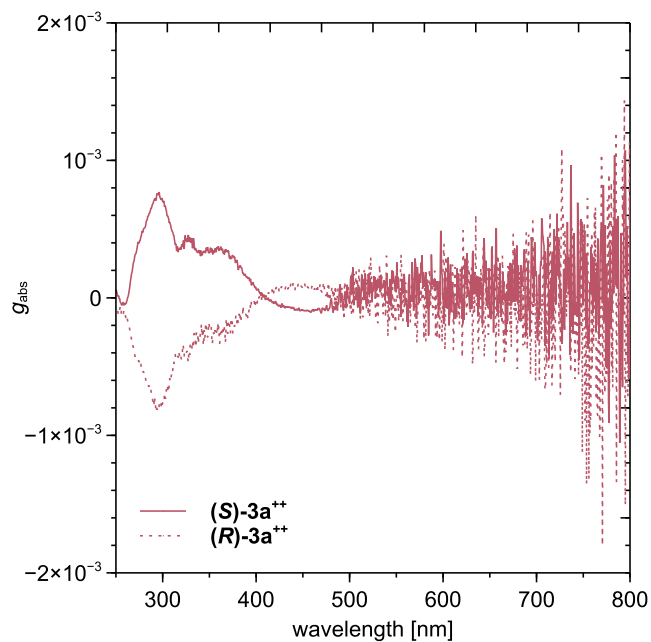
**Figure S23.** Electronic circular dichroism spectra of **(S,P<sub>p</sub>)-3b** (black solid lines), **(R,M<sub>p</sub>)-3b** (black dotted lines), and the corresponding redox series (radical cations and dications, blue and red lines, respectively) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



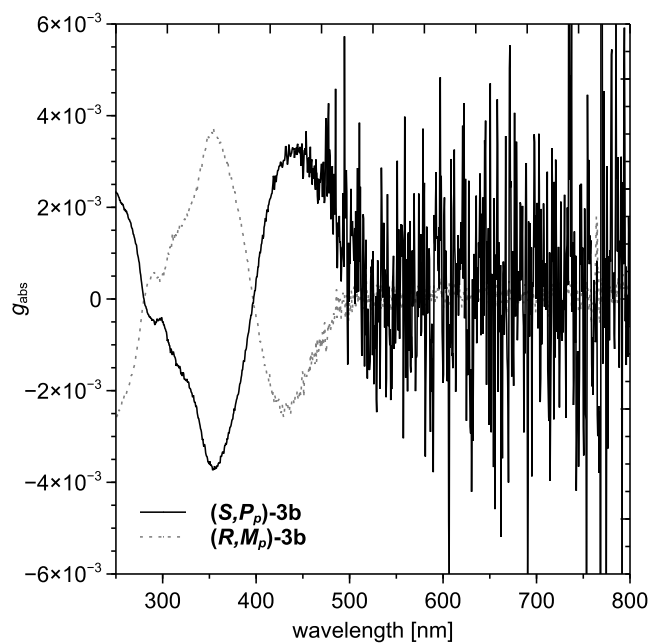
**Figure S24.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(*S*,*M*<sub>p</sub>)-3a** (black solid lines) and **(*R*,*P*<sub>p</sub>)-3a** (black dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



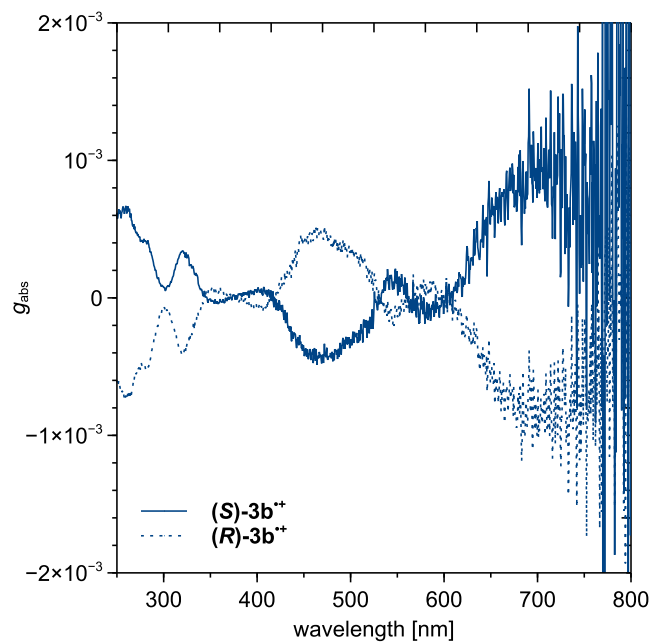
**Figure S25.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(*S*)-3a<sup>+</sup>** (blue solid lines) and **(*R*)-3a<sup>+</sup>** (blue dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



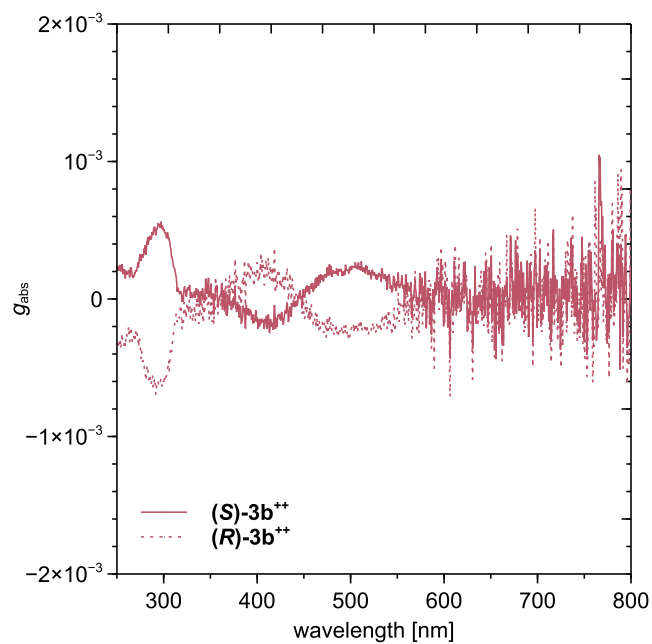
**Figure S26.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(S)-3a<sup>++</sup>** (red solid lines) and **(R)-3a<sup>++</sup>** (red dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



**Figure S27.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(S, $P_\rho$ )-3b** (black solid lines) and **(R, $M_\rho$ )-3b** (black dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.

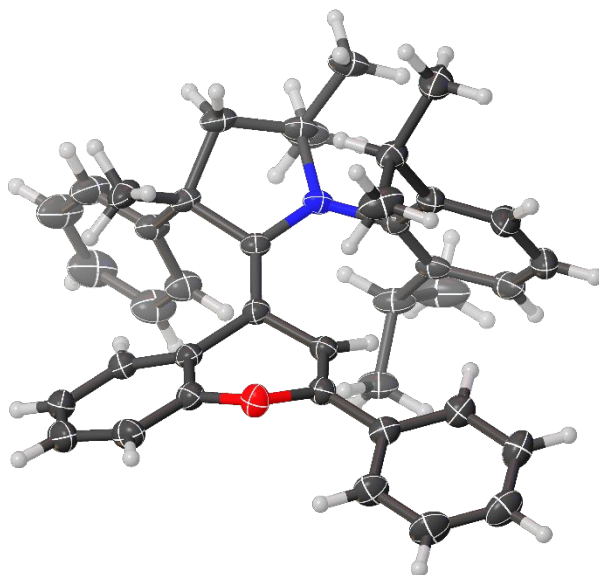


**Figure S28.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(S)-3b<sup>++</sup>** (blue solid lines) and **(R)-3b<sup>++</sup>** (blue dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.



**Figure S29.** Absorption dissymmetry factor ( $g_{\text{abs}}$ ) spectra of **(S)-3b<sup>++</sup>** (red solid lines) and **(R)-3b<sup>++</sup>** (red dotted lines) recorded in DCM ( $\sim 10^{-6}$  M) at 298 K.

## 8) XRD Analysis

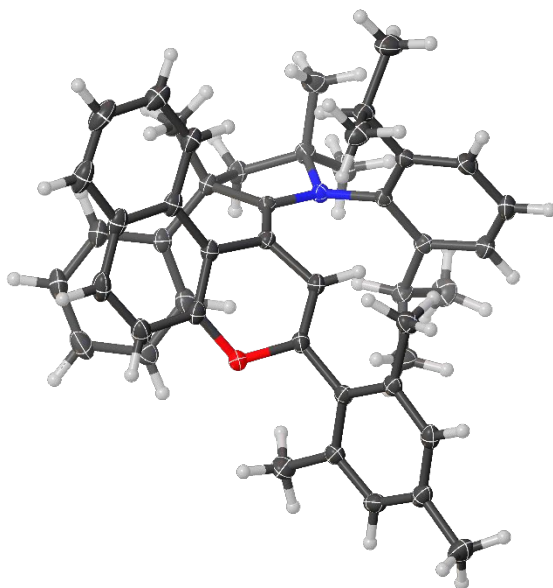


### Crystal data and structure refinement features for *rac*-3a

|                                           |                                                                        |
|-------------------------------------------|------------------------------------------------------------------------|
| CCDC Deposition Number                    | 2488689                                                                |
| Empirical formula                         | C <sub>40</sub> H <sub>43</sub> NO                                     |
| Formula weight                            | 553.75                                                                 |
| Temperature                               | 100                                                                    |
| Radiation type                            | Mo-K $\alpha$ ( $\lambda$ = 0.71073 Å)                                 |
| Crystal system                            | monoclinic                                                             |
| Space group                               | P-21/c                                                                 |
| a (Å)                                     | 9.8476(4)                                                              |
| b (Å)                                     | 16.5956(6)                                                             |
| c (Å)                                     | 19.5216(7)                                                             |
| $\alpha$ (°)                              | 90                                                                     |
| $\beta$ (°)                               | 97.4790(10)                                                            |
| $\gamma$ (°)                              | 90                                                                     |
| Volume (Å <sup>3</sup> )                  | 3163.2(2)                                                              |
| Z                                         | 4                                                                      |
| $\rho_{\text{calc}}$ (g/cm <sup>3</sup> ) | 1.163                                                                  |
| $\mu$ (mm <sup>-1</sup> )                 | 0.068                                                                  |
| F(000)                                    | 1192.0                                                                 |
| Crystal size (mm)                         | 0.1 x 0.1 x 0.1                                                        |
| Crystal color                             | yellow                                                                 |
| Crystal description                       | prism                                                                  |
| 2 $\theta$ range for data collection (°)  | 4.84 to 59.22                                                          |
| Index ranges                              | -13 $\leq$ h $\leq$ 13, -21 $\leq$ k $\leq$ 21, -27 $\leq$ l $\leq$ 26 |
| Reflections collected                     | 42468                                                                  |
| Unique reflections                        | 8603 [ $R_{\text{int}}$ = 0.0976]                                      |
| Data / restraints / parameters            | 8603 / 0 / 386                                                         |

|                                                     |                                  |
|-----------------------------------------------------|----------------------------------|
| Goodness-of-fit                                     | 1.003                            |
| Final R indices [ $I \geq 2\sigma(I)$ ]             | $R_1 = 0.0542$ , $wR_2 = 0.1102$ |
| Final R indices [all data]                          | $R_1 = 0.1324$ , $wR_2 = 0.1413$ |
| Largest diff. hole and peak $e\cdot\text{\AA}^{-3}$ | -0.22 and 0.20                   |

-



### Crystal data and structure refinement features for *rac*-3b

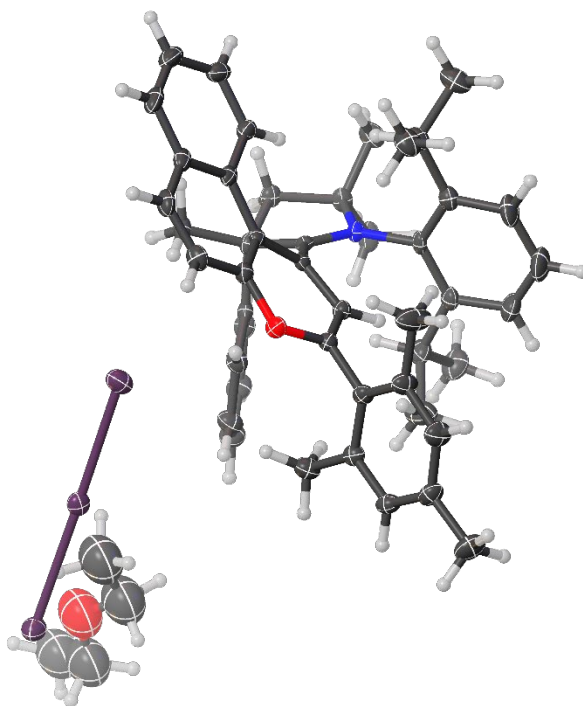
|                                           |                                        |
|-------------------------------------------|----------------------------------------|
| CCDC Deposition Number                    | 2488690                                |
| Empirical formula                         | $C_{47}H_{51}NO$                       |
| Formula weight                            | 645.89                                 |
| Temperature                               | 150(2) K                               |
| Radiation type                            | Mo-K $\alpha$ ( $\lambda = 0.71073$ Å) |
| Crystal system                            | monoclinic                             |
| Space group                               | P-21/c                                 |
| <i>a</i> (Å)                              | 16.8363(18)                            |
| <i>b</i> (Å)                              | 10.7483(12)                            |
| <i>c</i> (Å)                              | 20.356(2)                              |
| $\alpha$ (°)                              | 90                                     |
| $\beta$ (°)                               | 98.788(4)                              |
| $\gamma$ (°)                              | 90                                     |
| Volume (Å <sup>3</sup> )                  | 3640.4(7)                              |
| <i>Z</i>                                  | 4                                      |
| $\rho_{\text{calc}}$ (g/cm <sup>3</sup> ) | 1.178                                  |
| $\mu$ (mm <sup>-1</sup> )                 | 0.069                                  |

S60



|                                                       |                                                                  |
|-------------------------------------------------------|------------------------------------------------------------------|
| F(000)                                                | 1392                                                             |
| Crystal size (mm)                                     | 0.27 x 0.21 x 0.14                                               |
| Crystal color                                         | yellow                                                           |
| Crystal description                                   | prism                                                            |
| 2 $\theta$ range for data collection (°)              | 4.4 to 54.958                                                    |
| Index ranges                                          | -21 $\leq h \leq$ 20, -13 $\leq k \leq$ 13, -26 $\leq l \leq$ 24 |
| Reflections collected                                 | 28155                                                            |
| Unique reflections                                    | 8307 [ $R_{\text{int}} = 0.0508$ ]                               |
| Data / restraints / parameters                        | 8307 / 0 / 452                                                   |
| Goodness-of-fit                                       | 1.040                                                            |
| Final R indices [ $I \geq 2\sigma(I)$ ]               | $R_1 = 0.0496$ , $wR_2 = 0.1091$                                 |
| Final R indices [all data]                            | $R_1 = 0.0654$ , $wR_2 = 0.1170$                                 |
| Largest diff. hole and peak $e \cdot \text{\AA}^{-3}$ | -0.21 and 0.26                                                   |

-

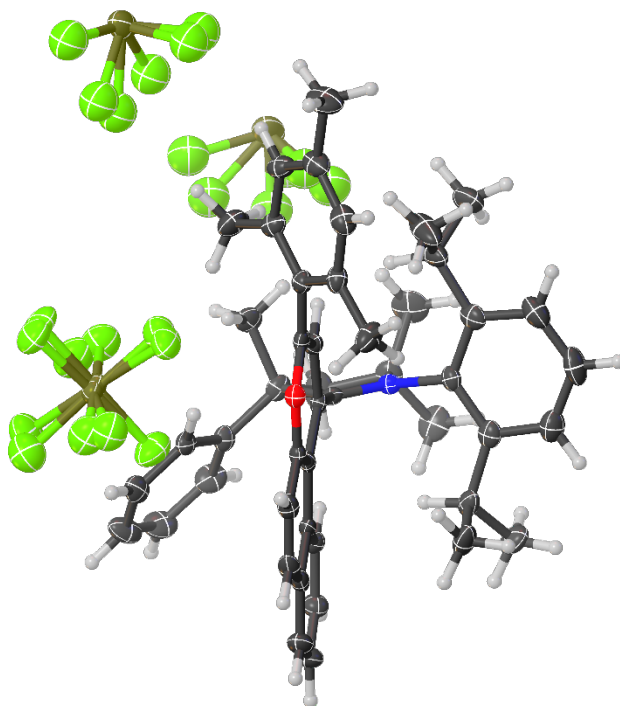


### Crystal data and structure refinement features for *rac*-3b<sup>+</sup>

|                        |                                                             |
|------------------------|-------------------------------------------------------------|
| CCDC Deposition Number | 2488692                                                     |
| Empirical formula      | $\text{C}_{98}\text{H}_{112}\text{I}_6\text{N}_2\text{O}_3$ |
| Formula weight         | 2127.29                                                     |
| Temperature            | 150(2) K                                                    |
| Radiation type         | Mo-K $\alpha$ ( $\lambda = 0.71073 \text{ \AA}$ )           |
| Crystal system         | triclinic                                                   |
| Space group            | P-1                                                         |
| $a$ (Å)                | 10.7490(12)                                                 |
| $b$ (Å)                | 13.7879(14)                                                 |

S61

|                                                            |                                                                        |
|------------------------------------------------------------|------------------------------------------------------------------------|
| c (Å)                                                      | 16.0621(16)                                                            |
| $\alpha$ (°)                                               | 78.407(3)                                                              |
| $\beta$ (°)                                                | 75.739(3)                                                              |
| $\gamma$ (°)                                               | 89.266(4)                                                              |
| Volume (Å <sup>3</sup> )                                   | 2258.4(4)                                                              |
| Z                                                          | 1                                                                      |
| $\rho_{\text{calc}}$ (g/cm <sup>3</sup> )                  | 1.564                                                                  |
| $\mu$ (mm <sup>-1</sup> )                                  | 2.113                                                                  |
| F(000)                                                     | 1056                                                                   |
| Crystal size (mm)                                          | 0.13 x 0.2 x 0.28                                                      |
| Crystal color                                              | black                                                                  |
| Crystal description                                        | prism                                                                  |
| 2 $\theta$ range for data collection (°)                   | 3.912 to 51.362                                                        |
| Index ranges                                               | -13 $\leq$ h $\leq$ 13, -16 $\leq$ k $\leq$ 16, -17 $\leq$ l $\leq$ 19 |
| Reflections collected                                      | 30649                                                                  |
| Unique reflections                                         | 8547 [R(int) = 0.0286]                                                 |
| Data / restraints / parameters                             | 8547 / 38 / 500                                                        |
| Goodness-of-fit                                            | 1.041                                                                  |
| Final R indices [ $I \geq 2\sigma(I)$ ]                    | R <sub>1</sub> = 0.0304, wR <sub>2</sub> = 0.0746                      |
| Final R indices [all data]                                 | R <sub>1</sub> = 0.0358, wR <sub>2</sub> = 0.0803                      |
| Largest diff. hole and peak e <sup>-</sup> Å <sup>-3</sup> | -1.37 and 2.43                                                         |



**Crystal data and structure refinement features for [rac-3b<sup>++</sup>][SbF<sub>6</sub>]<sub>2</sub>**

|                        |                                                                   |
|------------------------|-------------------------------------------------------------------|
| CCDC Deposition Number | 2488693                                                           |
| Empirical formula      | C <sub>47</sub> H <sub>51</sub> F <sub>12</sub> NOSb <sub>2</sub> |
| Formula weight         | 1117.39                                                           |
|                        | S62                                                               |

|                                                       |                                                                        |
|-------------------------------------------------------|------------------------------------------------------------------------|
| Temperature                                           | 150(2) K                                                               |
| Radiation type                                        | Mo-K $\alpha$ ( $\lambda$ = 0.71073 Å)                                 |
| Crystal system                                        | triclinic                                                              |
| Space group                                           | P-1                                                                    |
| a (Å)                                                 | 9.6256(10)                                                             |
| b (Å)                                                 | 11.6000(13)                                                            |
| c (Å)                                                 | 22.777(3)                                                              |
| $\alpha$ (°)                                          | 86.362(4)                                                              |
| $\beta$ (°)                                           | 78.497(4)                                                              |
| $\gamma$ (°)                                          | 75.573(4)                                                              |
| Volume (Å <sup>3</sup> )                              | 2413.3(5)                                                              |
| Z                                                     | 2                                                                      |
| $\rho_{\text{calc}}$ (g/cm <sup>3</sup> )             | 1.538                                                                  |
| $\mu$ (mm <sup>-1</sup> )                             | 1.200                                                                  |
| F(000)                                                | 1116                                                                   |
| Crystal size (mm)                                     | 0.23 x 0.2 x 0.05                                                      |
| Crystal color                                         | orange                                                                 |
| Crystal description                                   | prism                                                                  |
| 2 $\theta$ range for data collection (°)              | 4.036 to 51.362                                                        |
| Index ranges                                          | -11 $\leq$ h $\leq$ 11, -13 $\leq$ k $\leq$ 14, -27 $\leq$ l $\leq$ 27 |
| Reflections collected                                 | 33009                                                                  |
| Unique reflections                                    | 9156 [ $R_{\text{int}}$ = 0.0455; $R_{\text{sigma}}$ = 0.0430]         |
| Data / restraints / parameters                        | 9156 / 293 / 641                                                       |
| Goodness-of-fit                                       | 1.051                                                                  |
| Final R indices [ $I \geq 2\sigma(I)$ ]               | $R_1$ = 0.0726, $wR_2$ = 0.1934                                        |
| Final R indices [all data]                            | $R_1$ = 0.0849, $wR_2$ = 0.2038                                        |
| Largest diff. hole and peak e $\cdot$ Å <sup>-3</sup> | -2.71 and 3.71                                                         |

-

## 9) Computational Section

### Computational Methods

All calculations were performed using Orca 6.0.1.<sup>6</sup> Unless stated otherwise, all default parameters were used. For all calculations, options DefGrid2 and TightSCF were employed. All calculations using the Global Optimizer Algorithm (GOAT), as implemented in ORCA, used Grimme's semi-empirical GFN2-XTB<sup>7</sup> method for downhill steps and the GFN-FF<sup>8</sup> method for uphill steps, with the ALPB solvation model.<sup>9</sup> Minima and transition states were verified by the presence of zero or one imaginary vibrational mode, respectively. Final reported Gibbs free energies were obtained by adding the Gibbs free energy minus the electronic energy from frequency calculations to the relevant single point electronic energies. Cartesian coordinates (Å), canonical orbitals, spin surfaces, and transition densities were visualized using ChemCraft. For consistency, all molecules are presented with the stereogenic carbon center of the <sup>Chi</sup>CAAC fragment in the (**S**) configuration.

Hyperfine coupling constants were assessed at the PBE0/N07D/CPCM(DCM)<sup>10,11</sup> level of theory using geometries for the lowest energy conformer of a given absolute configuration. N07D is a modified 6-31G basis set specifically designed for modeling radicals.<sup>12,13</sup> Spin-orbit coupling (SOC) was treated using the spin-orbit mean-field (SOMF(1X)) approach,<sup>14</sup> with the resolution of identity (RI) used to approximate the Coulombic portion;<sup>15</sup> the necessary auxiliary basis set was automatically generated.<sup>16</sup> Cubes for the total spin density were generated in ChemCraft.

Simplified TD-DFT (sTD-DFT)<sup>17,18</sup> calculations were performed using the  $\omega$ B97x-D3<sup>19</sup> and CAM-B3LYP<sup>20</sup> functionals, as well as the def2-TZVPP<sup>21</sup> basis set and CPCM solvation with parameters for DCM. Further, the RIJCOSX approximation was employed,<sup>22,23</sup> using the def2/J auxiliary basis set.<sup>24</sup> Ultimately, the CAM-B3LYP functional was determined to be in best agreement with the experimental data (Figures S34 – S35), yet both approaches are in qualitative agreement with the experiment and reproduced the shapes of the experimental spectra. Spectral convolution was performed in ChemCraft.

Full TD-DFT calculations, *without* the Tamm-Dancoff approximation, were performed using the CAM-B3LYP functional, maintaining other parameters for the level of theory as in the sTD-DFT calculations. Transition density cube files were generated with `orca_plot` using a grid interval of 120. From these calculations, the transition electric- ( $|\mu|$ ) and transition magnetic ( $|m|$ ) dipole moment vector norms, the angle between the corresponding vectors ( $\theta$ ), and ultimately the dissymmetry factors ( $g_{\text{abs}}$ ), were calculated as described in much detail by Kubo *et al.*<sup>25</sup> Specifically,  $g_{\text{abs}}$  was calculated according to equation S1.

$$g_{\text{abs}} = \frac{4 \cos \theta_{\mu, m} |\mu| |m|}{|\mu|^2 + |m|^2} \quad (\text{equation S1})$$

With: 1 a.u. =  $2.54 \times 10^{-18}$  cgs (electric dipole) (equation S2)

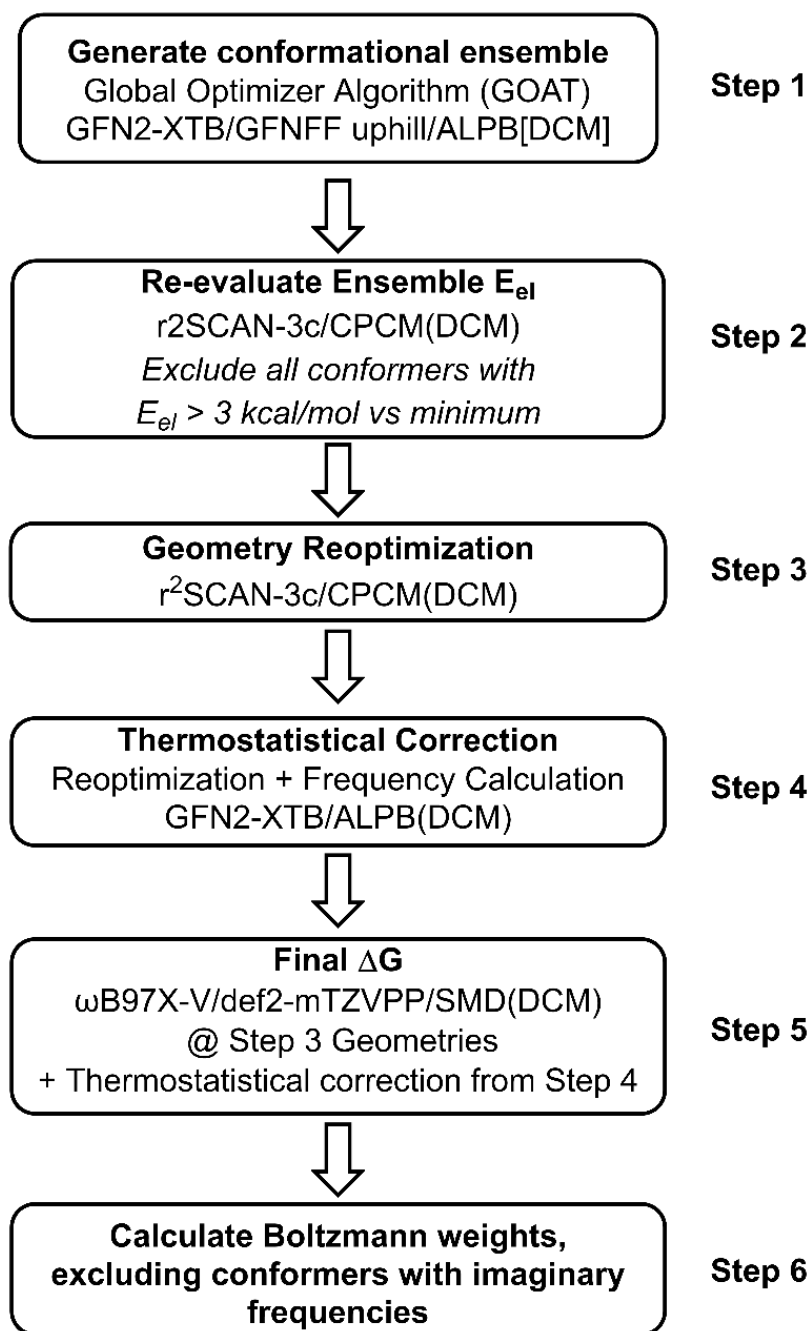
And: 1 a.u. =  $1.857 \times 10^{-20}$  cgs (magnetic dipole) (equation S3)

All output files are available for download, free of charge, from the UCSD Library Digital Collections.<sup>26</sup>

To generate conformational ensembles for **3a-b** and the respective radical cations and dications, a multistep computational workflow was applied, an adaptation of the *CREST/CENSO* workflow developed by Grimme and coworkers.<sup>27</sup>

- Step 1:** For each stereoisomer, an initial conformational ensemble was generated using GOAT.
- Step 2:** The electronic energies of these conformers were re-evaluated at the r<sup>2</sup>SCAN-3c/CPCM(DCM) level of theory.<sup>28</sup> For each stereoisomer, all conformers with an electronic energy 3 kcal/mol or greater than the minimum were excluded.
- Step 3:** Geometries were reoptimized at the r<sup>2</sup>SCAN-3c/CPCM(DCM) level of theory.
- Step 4:** Thermostatistical corrections to the electronic energies were obtained by a reoptimization and frequency calculation at the GFN2-XTB level of theory with the ALPB solvation model for DCM.
- Step 5:** Final Gibbs free energies were obtained by recalculating the electronic energies at the  $\omega$ B97X-V/def2-mTZVPP/SMD(DCM) level of theory using Step 3 geometries.<sup>29</sup> Thermostatistical corrections obtained from Step 4 were applied to the resulting electronic energies.
- Step 6:** Using the Gibbs free energies obtained in Step 5, the Boltzmann populations for each conformer were obtained, assuming no kinetic barriers between states. States associated with an imaginary frequency from Step 4 were excluded.

The workflow is represented graphically in Figure S30. The resulting data are tabulated in Table S4. Select cartesian coordinates, specifically the lowest energy conformer for a given absolute configuration, for each compound, are provided in “Cartesian Coordinates.”

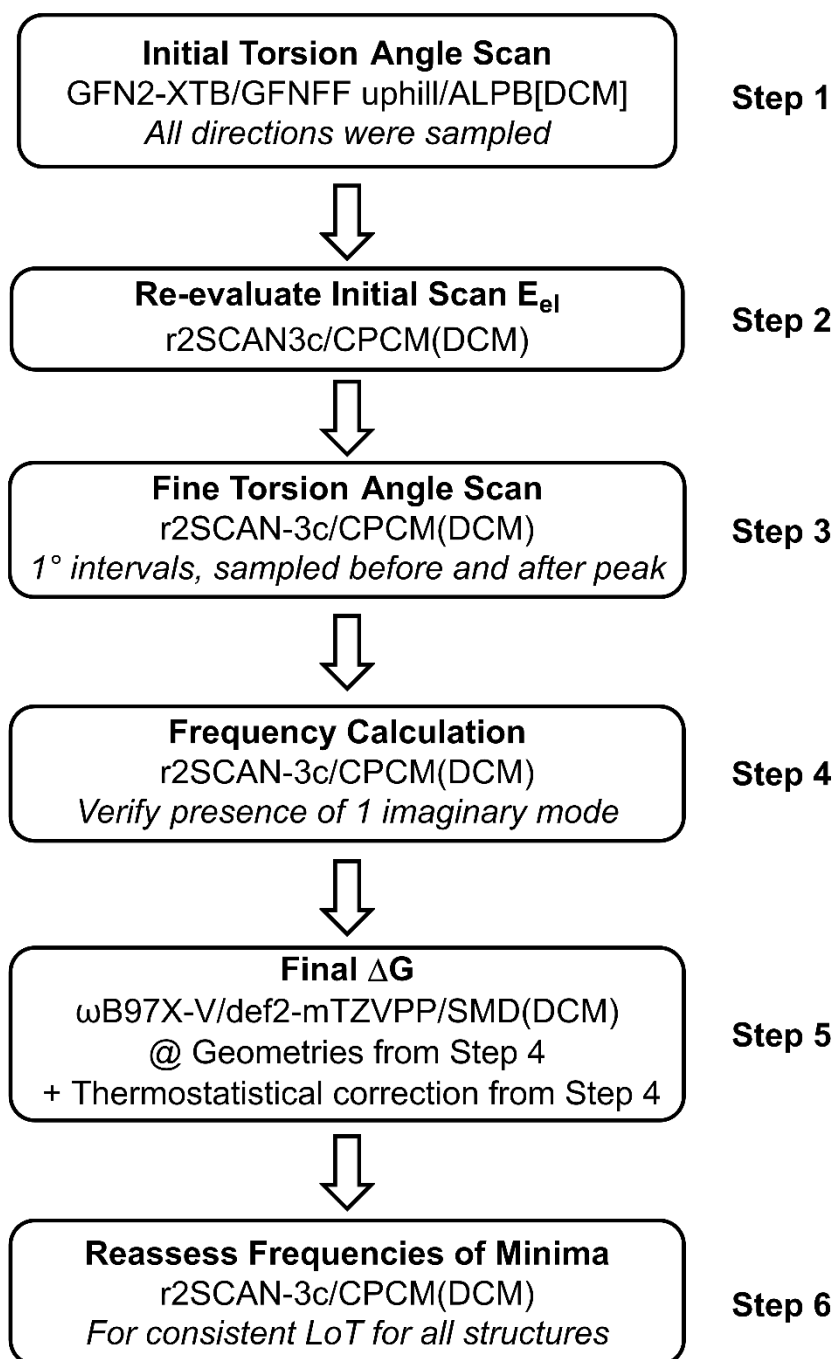


**Figure S30.** Graphical depiction of computational workflow for modeling the conformation ensembles of **3a-b** and the respective radical cations and dications.

To assess the barriers for atropisomerization, the following workflow was applied.

- Step 1:** For a given pair of stereoisomers, initial relaxed coordinate scans were performed at the GFN2-XTB/ALPB(DCM) level of theory along key torsion angles which affected epimerization in both directions.
- Step 2:** The electronic energies for the coordinate scans from Step 1 were reevaluated at the r<sup>2</sup>SCAN-3c/CPCM(DCM) level of theory.
- Step 3:** Relaxed coordinate scans with fine intervals (step = 1°) were performed across the highest “peak” of each coordinate scan r<sup>2</sup>SCAN-3c/CPCM(DCM).
- Step 4:** Frequency calculations at the r<sup>2</sup>SCAN-3c/CPCM(DCM) level of theory were performed for the highest energy conformation from Step 3, verifying the presence of a single imaginary mode.
- Step 5:** In the case of a transition state (one imaginary mode), final Gibbs free energies were obtained by recalculating the electronic energy at the ωB97X-V/def2-mTZVPP/SMD(DCM) level of theory. To the resulting electronic energy, the thermostistical corrections obtained from Step 4 were applied. The true barrier was taken as the lowest barrier obtained from all coordinate scans, regardless of direction.<sup>30</sup>
- Step 6:** Thermostistical corrections for the lowest energy conformers of each stereoisomer were reassessed at the r<sup>2</sup>SCAN-3c/CPCM(DCM), then used to calculate the final Gibbs free energies as in Step 5.

The workflow is represented graphically in Figure S31. The resulting data are tabulated in Table S5. Cartesian coordinates for the minima and transition states are provided in “Cartesian Coordinates”



**Figure S31.** Graphical depiction of computational workflow for obtaining the barriers to atropisomerization of **3a**, **3b**, and the respective radical cations and dications.



## Conformational Ensembles

**Table S4.** Gibbs free energy and Boltzmann weight for conformers of all compounds.

| Compound | Conf. # | AbsConfig | E(el) (H)<br>Step 5 | G-E(el) (H)<br>Step 4 | G (H)<br>Step 5 | Boltzmann<br>Weight |
|----------|---------|-----------|---------------------|-----------------------|-----------------|---------------------|
| 3a       | 1       | S,Mp      | -1680.252043        | 0.647216              | -1679.604827    | 0.87048716          |
|          | 2       | S,Pp      | -1680.249461        | 0.647607              | -1679.601854    | 0.03808033          |
|          | 3       | S,Pp      | -1680.249202        | 0.647278              | -1679.601924    | 0.04097865          |
|          | 4       | S,Pp      | -1680.249929        | 0.647808              | -1679.602121    | 0.05045387          |
|          |         |           |                     |                       | M-plane         | 0.87048716          |
|          |         |           |                     |                       | P-plane         | 0.12951284          |
|          |         |           |                     |                       |                 |                     |
| 3a+      | 1       | S,Mp      | -1680.088784        | 0.647730              | -1679.441055    | 0.06867850          |
|          | 2       | S,Mp      | -1680.088321        | 0.647794              | -1679.440527    | 0.03939043          |
|          | 3       | S,Pp      | -1680.088212        | 0.648057              | -1679.440155    | 0.02663822          |
|          | 4       | S,Pp      | -1680.090191        | 0.647497              | -1679.442694    | 0.38579619          |
|          | 5       | S,Mp      | -1680.088049        | 0.647480              | -1679.440569    | 0.04120685          |
|          | 6       | S,Mp      | -1680.089056        | 0.646919              | -1679.442137    | 0.21466347          |
|          | 7       | S,Pp      | -1680.089302        | 0.647701              | -1679.441601    | 0.12205663          |
|          | 8       | S,Pp      | -1680.088578        | 0.647229              | -1679.441350    | 0.09367170          |
|          | 9       | S,Pp      | -1680.085328        | 0.647546              | -1679.437782    | 0.00219152          |
|          | 10      | S,Mp      | -1680.084385        | 0.646854              | -1679.437531    | 0.00168249          |
|          | 11      | S,Pp      | -1680.086295        | 0.647936              | -1679.438359    | 0.00402399          |
|          |         |           |                     |                       | M-plane         | 0.36562174          |
|          |         |           |                     |                       | P-plane         | 0.63437826          |
|          |         |           |                     |                       |                 |                     |
| 3a++     | 1       | S,Ma      | -1679.911895        | 0.647284              | -1679.264610    | 0.23081752          |
|          | 2       | S,Ma      | -1679.910493        | 0.646964              | -1679.263529    | 0.07395955          |
|          | 3       | S,Pa      | -1679.909160        | 0.647418              | -1679.261742    | 0.01127160          |
|          | 4       | S,Pa      | -1679.909507        | 0.648138              | -1679.261369    | 0.00761509          |
|          | 5       | S,Ma      | -1679.908827        | 0.646814              | -1679.262013    | 0.01499334          |
|          | 6       | S,Pa      | -1679.913292        | 0.647961              | -1679.265331    | 0.49293156          |
|          | 7       | S,Pa      | -1679.909253        | 0.646610              | -1679.262643    | 0.02911028          |
|          | 8       | S,Ma      | -1679.910631        | 0.646501              | -1679.264131    | 0.13930105          |
|          |         |           |                     |                       | M-axis          | 0.45907146          |
|          |         |           |                     |                       | P-axis          | 0.54092854          |
|          |         |           |                     |                       |                 |                     |
| 3b       | 1       | S,Pp      | -1951.851480        | -133.747486           | -2085.598966    | 0.99651838          |
|          | 2       | S,Pp      | -1951.848973        | -133.744618           | -2085.593591    | 0.00347793          |
|          | 3       | S,Mp      | -1951.843412        | -133.743096           | -2085.586508    | 0.00000201          |
|          | 4       | S,Mp      | -1951.843625        | -133.742558           | -2085.586183    | 0.00000143          |
|          | 5       | S,Mp      | -1951.841652        | -133.741139           | -2085.582790    | 0.00000004          |
|          | 6       | S,Mp      | -1951.844429        | -133.739951           | -2085.584380    | 0.00000021          |

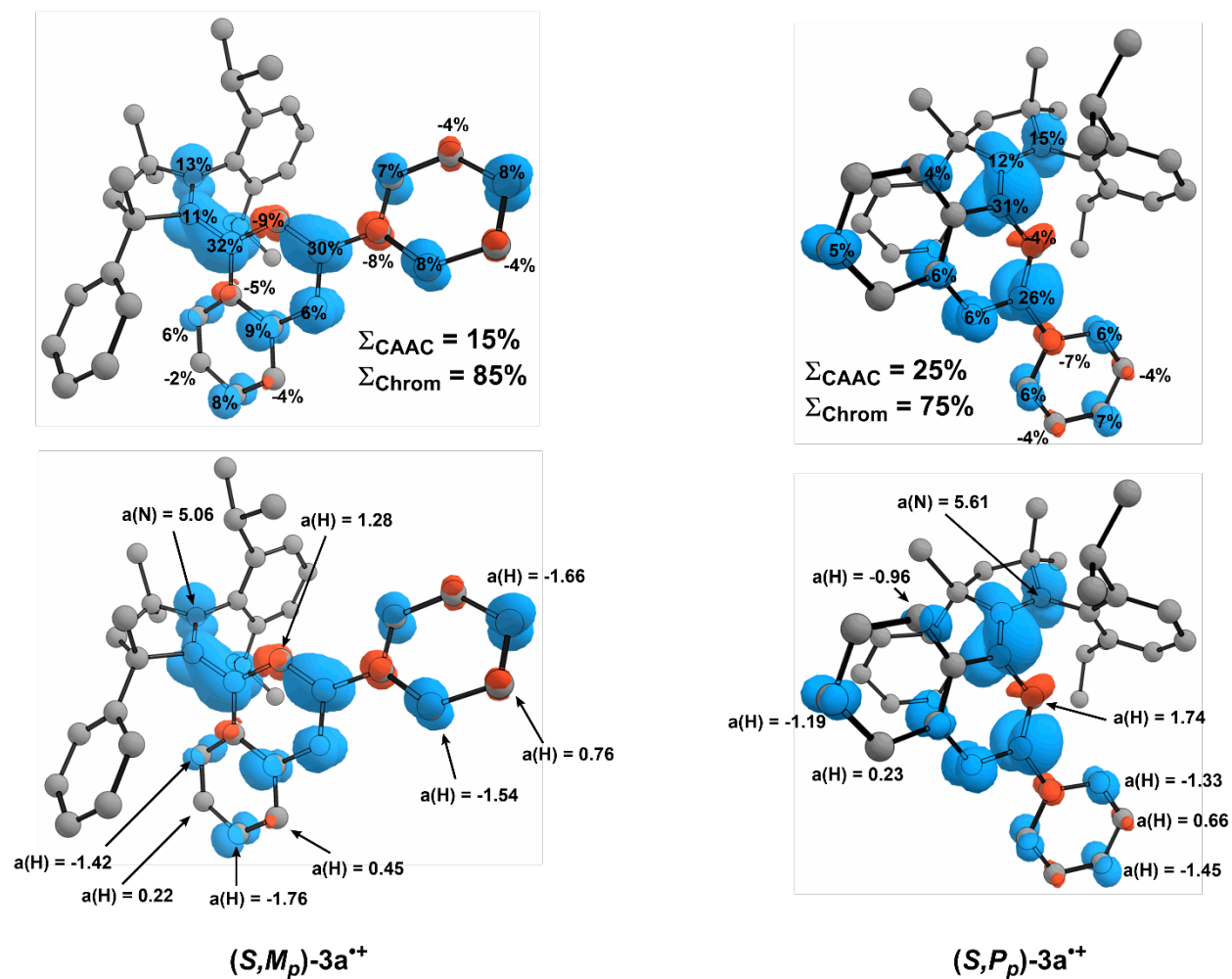
|             |          |             |              |             |                |            |
|-------------|----------|-------------|--------------|-------------|----------------|------------|
|             |          |             |              |             | <b>M-plane</b> | 0.00000369 |
|             |          |             |              |             | <b>P-plane</b> | 0.99999631 |
|             |          |             |              |             |                |            |
| <b>3b+</b>  | <b>1</b> | <b>S,Pp</b> | -1951.693383 | -133.408642 | -2085.102025   | 0.66832672 |
|             | <b>2</b> | <b>S,Mp</b> | -1951.692081 | -133.407928 | -2085.100009   | 0.08002500 |
|             | <b>3</b> | <b>S,Pp</b> | -1951.693420 | -133.407671 | -2085.101091   | 0.24998100 |
|             | <b>4</b> | <b>S,Mp</b> | -1951.689370 | -133.406961 | -2085.096331   | 0.00166729 |
|             |          |             |              |             | <b>M-plane</b> | 0.08169228 |
|             |          |             |              |             | <b>P-plane</b> | 0.91830772 |
|             |          |             |              |             |                |            |
| <b>3b++</b> | <b>1</b> | <b>S,Ma</b> | -1951.517085 | -133.034219 | -2084.551304   | 0.99973615 |
|             | <b>2</b> | <b>S,Pa</b> | -1951.511360 | -133.031853 | -2084.543213   | 0.00020016 |
|             | <b>3</b> | <b>S,Pa</b> | -1951.511879 | -133.028895 | -2084.540774   | 0.00001536 |
|             | <b>4</b> | <b>S,Pa</b> | -1951.509898 | -133.027801 | -2084.537700   | 0.00000060 |
|             | <b>5</b> | <b>S,Pa</b> | -1951.512282 | -133.028825 | -2084.541107   | 0.00002181 |
|             | <b>6</b> | <b>S,Pa</b> | -1951.512372 | -133.028899 | -2084.541271   | 0.00002592 |
|             |          |             |              |             | <b>M-axis</b>  | 0.99973615 |
|             |          |             |              |             | <b>P-axis</b>  | 0.00026385 |

## Atropisomerization Barriers

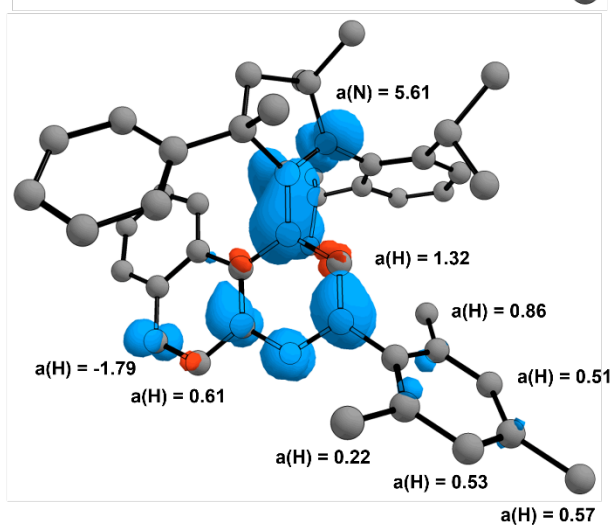
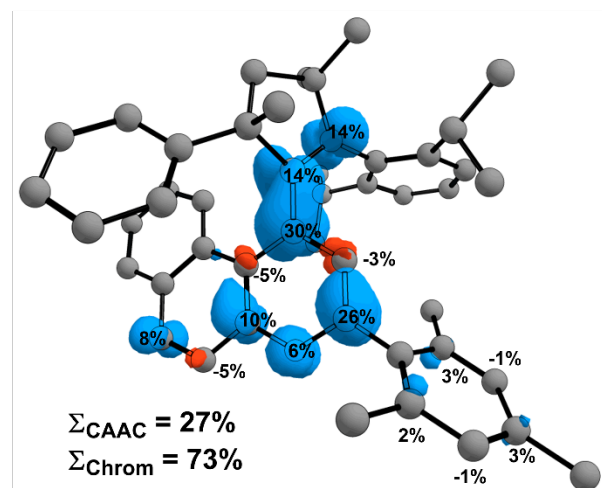
**Table S5.** Absolute contributions to the Gibbs free energies for assessing barriers to atropisomerization

| Structure                             | E <sub>el</sub> (H) | H (H)        | -T*S (H)  | G (H)        | G-E <sub>el</sub> (H) | $\nu$ (cm <sup>-1</sup> ) | E <sub>el</sub> SPC (H) | G <sub>final</sub> | G <sub>rel</sub><br>(kcal/mol) |
|---------------------------------------|---------------------|--------------|-----------|--------------|-----------------------|---------------------------|-------------------------|--------------------|--------------------------------|
| (S,M <sub>p</sub> )-3a                | -1679.742536        | -1678.981026 | -0.100318 | -1679.081343 | 0.661193              | 17.46                     | -1680.252043            | -1679.590850       | 0.00                           |
| TS-3a                                 | -1679.729697        | -1678.969400 | -0.099009 | -1679.068409 | 0.661288              | -21.46                    | -1680.241259            | -1679.579971       | 6.83                           |
| (S,P <sub>p</sub> )- 3a               | -1679.739012        | -1678.977398 | -0.100596 | -1679.077994 | 0.661019              | 11.65                     | -1680.249852            | -1679.588834       | 1.27                           |
| (S,M <sub>p</sub> )- 3a <sup>++</sup> | -1679.579736        | -1678.816601 | -0.100874 | -1678.917475 | 0.662261              | 21.52                     | -1680.089056            | -1679.426795       | 0.42                           |
| TS-3a <sup>++</sup>                   | -1679.568758        | -1678.806770 | -0.099858 | -1678.906628 | 0.662130              | -27.03                    | -1680.081964            | -1679.419834       | 4.79                           |
| (S,P <sub>p</sub> )- 3a <sup>++</sup> | -1679.580798        | -1678.817148 | -0.100928 | -1678.918076 | 0.662722              | 13.29                     | -1680.090191            | -1679.427470       | 0.00                           |
| (S,M <sub>a</sub> )- 3a <sup>++</sup> | -1679.395757        | -1678.630093 | -0.099534 | -1678.729627 | 0.666131              | 16.96                     | -1679.911895            | -1679.245764       | 0.82                           |
| TS-3a <sup>++</sup>                   | -1679.366582        | -1678.602462 | -0.097365 | -1678.699827 | 0.666754              | -13.93                    | -1679.876979            | -1679.210225       | 23.12                          |
| (S,P <sub>a</sub> )- 3a <sup>++</sup> | -1679.396622        | -1678.630868 | -0.099526 | -1678.730394 | 0.666229              | 17.3                      | -1679.913292            | -1679.247063       | 0.00                           |
|                                       |                     |              |           |              |                       |                           |                         |                    |                                |
| (S,M <sub>p</sub> )-3b                | -1951.246496        | -1950.349020 | -0.113592 | -1950.462612 | 0.783883              | 21.84                     | -1951.843411            | -1951.059528       | 5.10                           |
| TS-3b                                 | -1951.211025        | -1950.314847 | -0.112226 | -1950.427073 | 0.783953              | -28.56                    | -1951.799073            | -1951.015121       | 32.96                          |
| (S,P <sub>p</sub> )- 3b               | -1951.254727        | -1950.357190 | -0.113708 | -1950.470898 | 0.783829              | 22.74                     | -1951.851480            | -1951.067651       | 0.00                           |
| (S,M <sub>p</sub> )- 3b <sup>++</sup> | -1951.097672        | -1950.198627 | -0.115155 | -1950.313782 | 0.783891              | 22.13                     | -1951.692081            | -1950.908190       | 0.29                           |
| TS-3b <sup>++</sup>                   | -1951.061217        | -1950.162757 | -0.113159 | -1950.275916 | 0.785301              | -28.63                    | -1951.658876            | -1950.873575       | 21.72                          |
| (S,P <sub>p</sub> )- 3b <sup>++</sup> | -1951.099564        | -1950.200185 | -0.114650 | -1950.314835 | 0.784729              | 18.17                     | -1951.693383            | -1950.908654       | 0.00                           |
| (S,M <sub>a</sub> )- 3b <sup>++</sup> | -1950.917323        | -1950.015846 | -0.113068 | -1950.128914 | 0.788409              | 20.29                     | -1951.517085            | -1950.728676       | 0.00                           |
| TS-3b <sup>++</sup>                   | -1950.875665        | -1949.975983 | -0.110781 | -1950.086764 | 0.788901              | -5.73                     | -1951.467311            | -1950.678410       | 31.54                          |
| (S,P <sub>a</sub> )- 3b <sup>++</sup> | -1950.912127        | -1950.011417 | -0.113977 | -1950.125393 | 0.786734              | 16.13                     | -1951.511360            | -1950.724626       | 2.54                           |

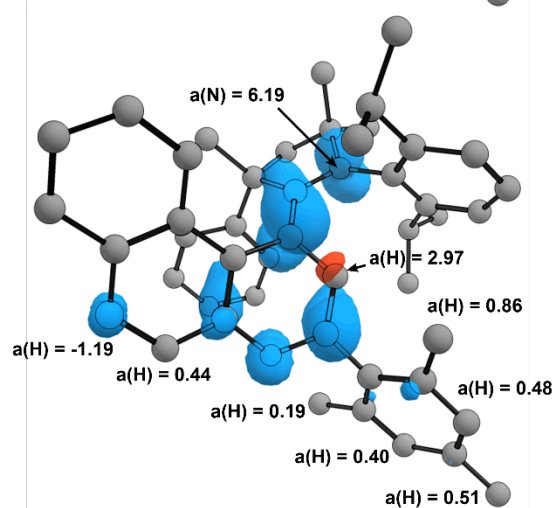
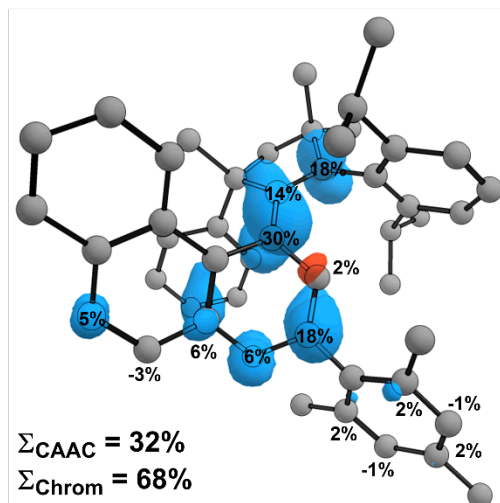
## Hyperfine Coupling Constants and Spin Densities



**Figure S32.** Comparison of total spin density surfaces, Mulliken spin densities, and hyperfine coupling constants (Gauss) for atropisomers of  $3a^{*+}$ .



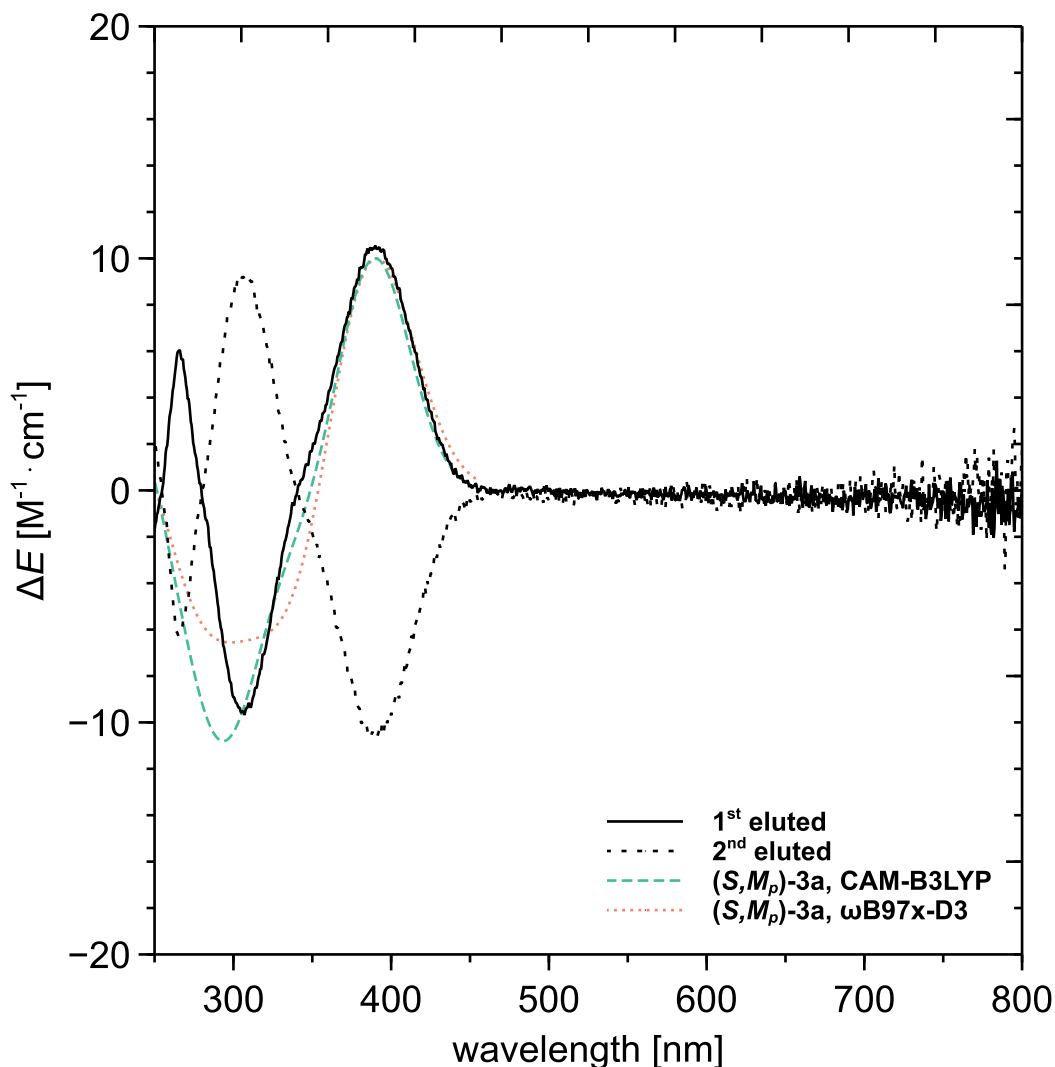
**(*S,M<sub>p</sub>*)-3b<sup>\*+</sup>**



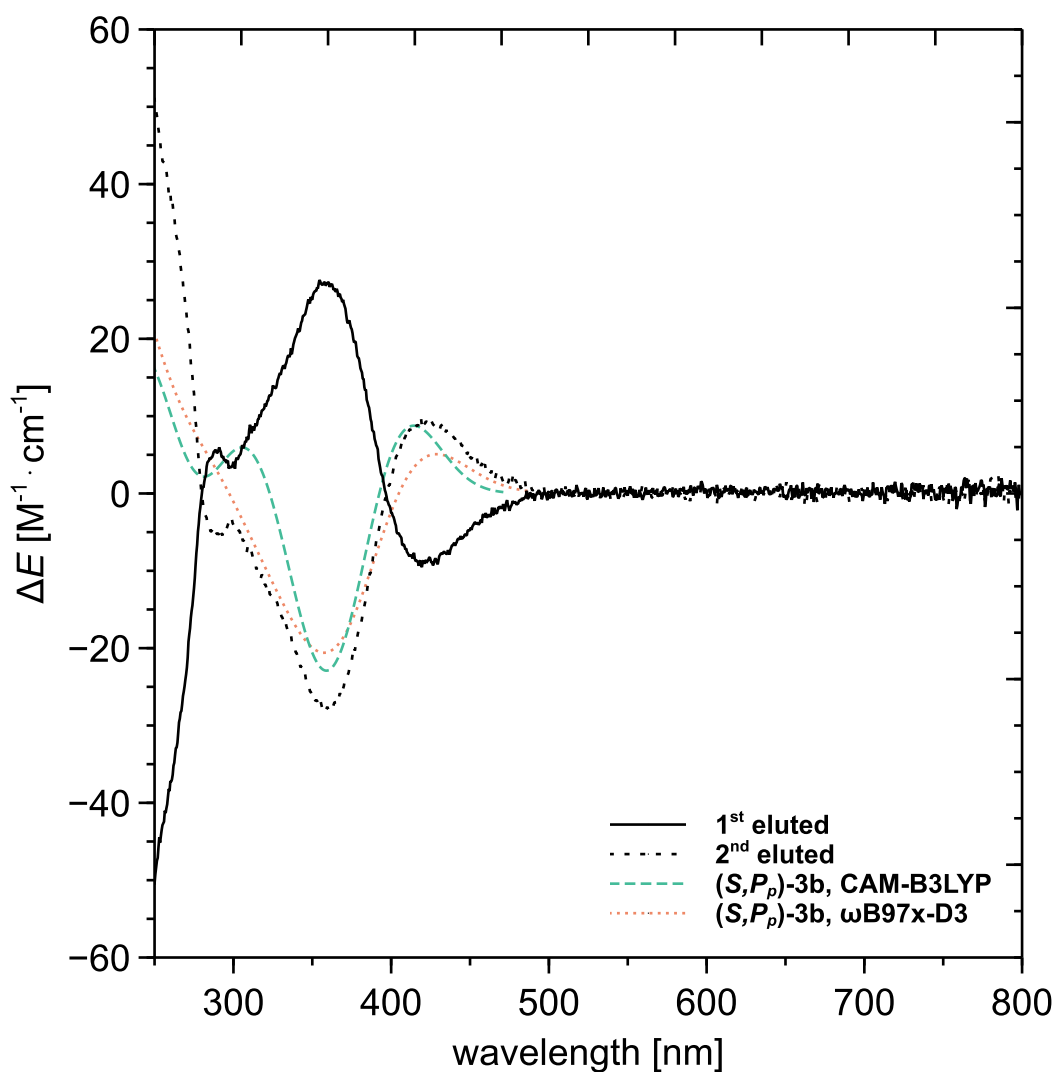
**(*S,P<sub>p</sub>*)-3b<sup>\*+</sup>**

**Figure S33.** Comparison of total spin density surfaces, Mulliken spin densities, and hyperfine coupling constants (Gauss) for atropisomers of **3b<sup>\*+</sup>**.

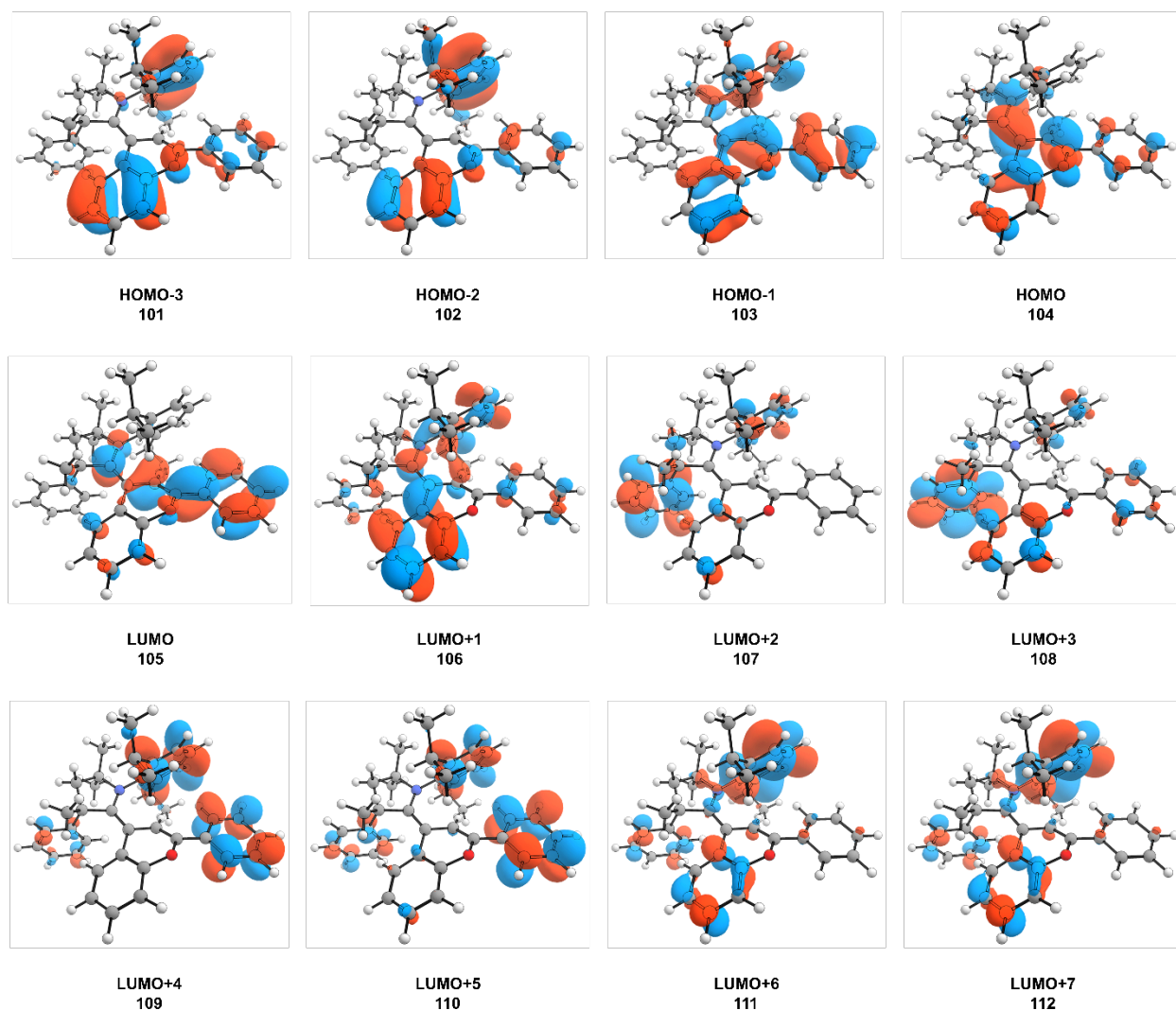
## TD-DFT Calculations



**Figure S34.** Comparison of experimental CD spectra (black, solid and dotted) of chiral HPLC products with computed spectra (Gaussian broadening, FWHM = 50 nm) for (S,M<sub>p</sub>)-3a. Relative intensities ( $\Delta E$ ) are adjusted for clarity. Computed data obtained using CAM-B3LYP (green, large dash) are offset by  $-800 \text{ cm}^{-1}$ , while computed data obtained using  $\omega$ B97X-D3 (orange, small dash) are offset by  $-2900 \text{ cm}^{-1}$ .



**Figure S35.** Comparison of experimental CD spectra (black, solid and dotted) of chiral HPLC products with computed spectra (Gaussian broadening, FWHM = 50 nm) for (**S,P<sub>p</sub>**)-**3b**. Relative intensities ( $\Delta E$ ) are adjusted for clarity. Computed data obtained using CAM-B3LYP (green, large dash) are offset by  $-500 \text{ cm}^{-1}$ , while computed data obtained using  $\omega\text{B97X-D3}$  (orange, small dash) are offset by  $-3500 \text{ cm}^{-1}$ .

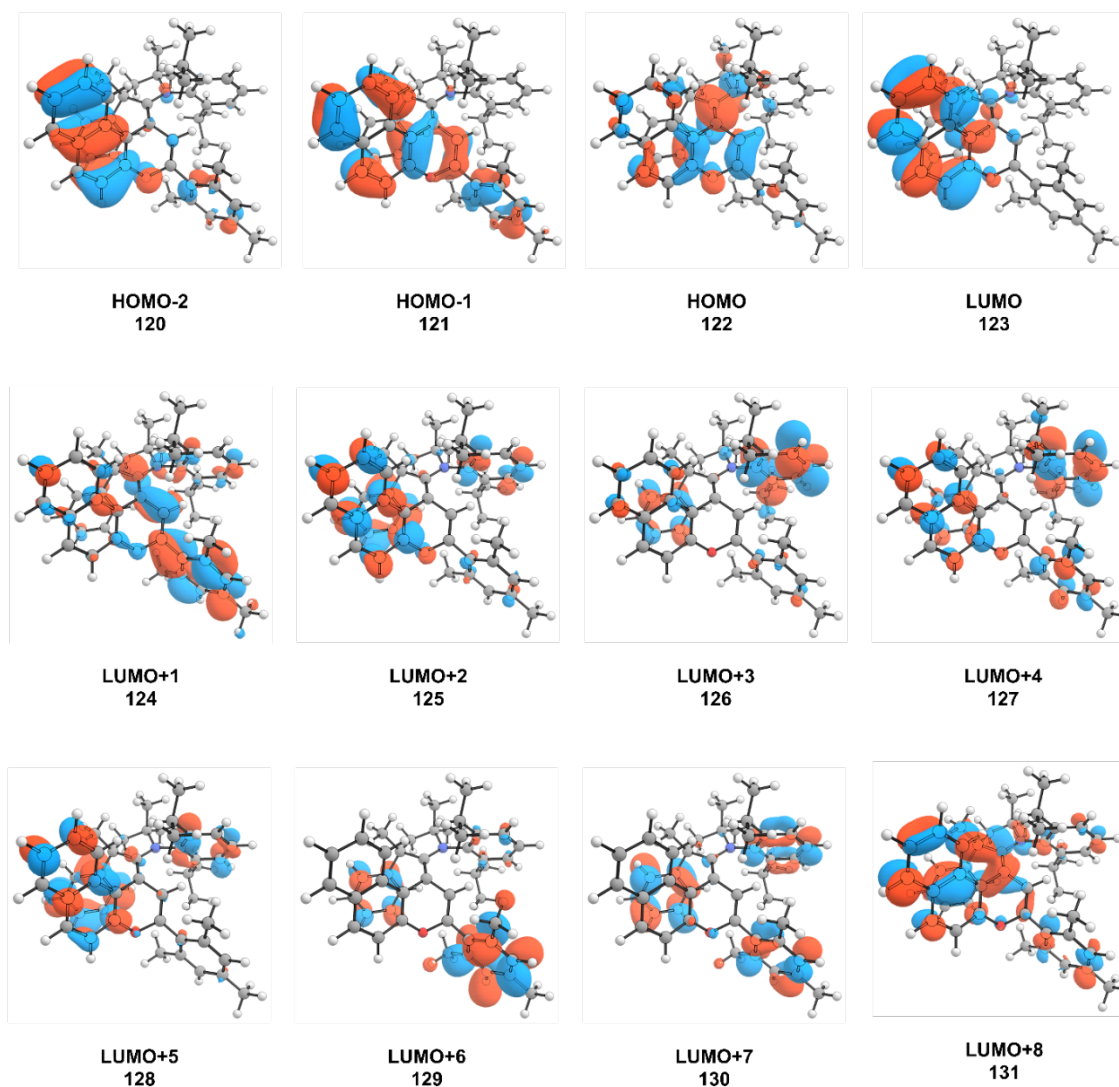


**Figure S36.** Canonical orbitals of **(*S,M<sub>p</sub>*)-3a** obtained via sTD-DFT using the CAM-B3LYP functional

| Transition | eV    | nm    | Amplitudes       |                  |                  |
|------------|-------|-------|------------------|------------------|------------------|
| <b>S1</b>  | 3.287 | 377.2 | 0.92 (104-> 105) | 0.04 (104-> 106) | 0.01 (104-> 112) |
| <b>S2</b>  | 3.729 | 332.5 | 0.92 (104-> 106) | 0.04 (104-> 105) | 0.01 (104-> 114) |
| <b>S3</b>  | 4.086 | 303.4 | 0.44 (104-> 109) | 0.34 (104-> 107) | 0.15 (104-> 110) |
| <b>S4</b>  | 4.156 | 298.3 | 0.54 (104-> 108) | 0.28 (104-> 111) | 0.06 (104-> 107) |
| <b>S5</b>  | 4.255 | 291.4 | 0.24 (104-> 107) | 0.24 (104-> 109) | 0.24 (104-> 108) |
| <b>S6</b>  | 4.376 | 283.4 | 0.39 (104-> 111) | 0.28 (104-> 107) | 0.19 (104-> 110) |
| <b>S7</b>  | 4.389 | 282.5 | 0.52 (104-> 110) | 0.25 (104-> 109) | 0.07 (104-> 111) |
| <b>S8</b>  | 4.51  | 274.9 | 0.69 (104-> 112) | 0.17 (104-> 108) | 0.08 (104-> 111) |
| <b>S9</b>  | 4.932 | 251.4 | 0.96 (103-> 105) | 0.00 (103-> 114) | 0.00 (99-> 105)  |

**Table S6.** Nature of selected electronic transitions for **(*S,M<sub>p</sub>*)-3a** obtained via sTD-DFT using the CAM-B3LYP functional





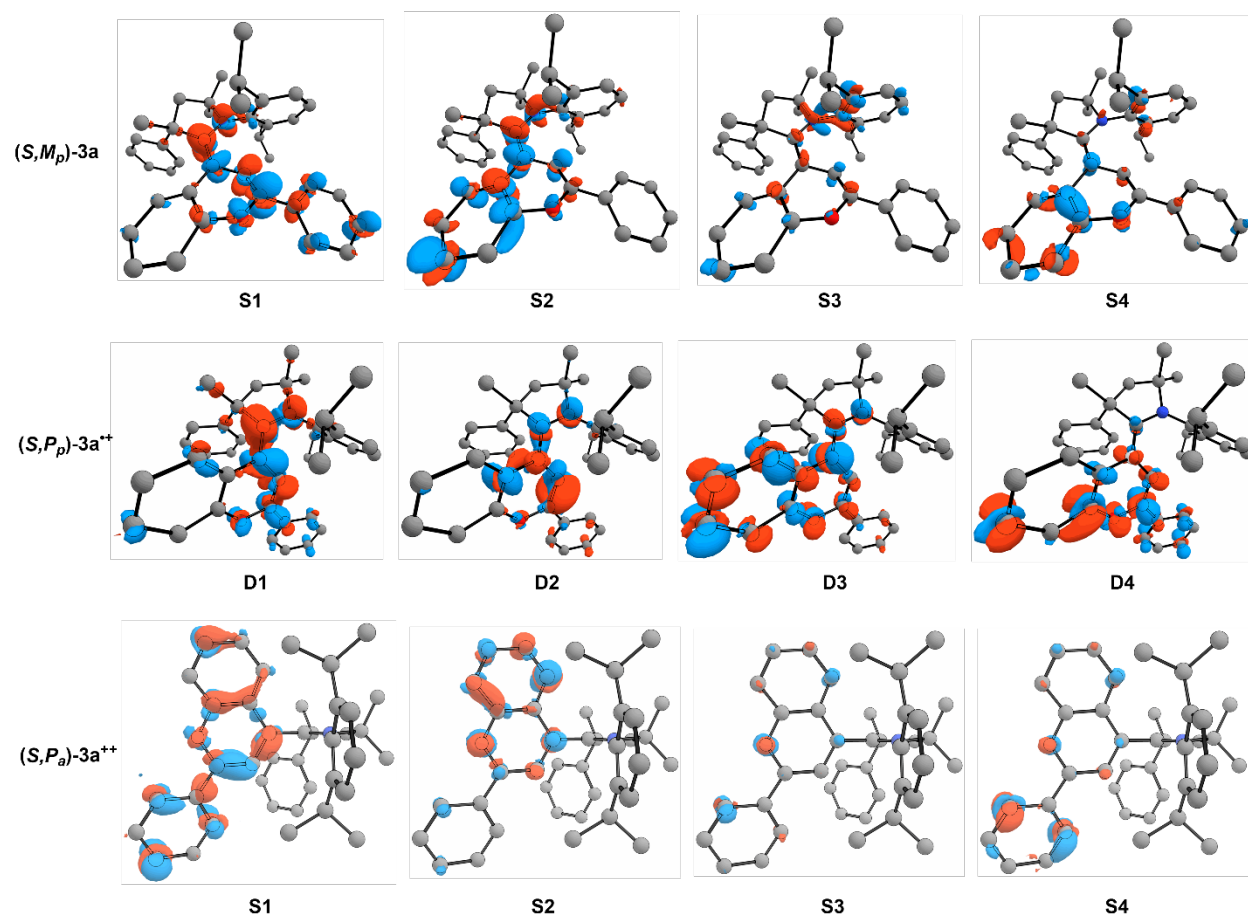
**Figure S37.** Canonical orbitals of (*S,P<sub>p</sub>*)-**3b** obtained via sTD-DFT using the CAM-B3LYP functional

| Transition | eV    | nm    | Amplitudes       |                  |                  |
|------------|-------|-------|------------------|------------------|------------------|
| <b>S1</b>  | 3.101 | 399.8 | 0.93 (122-> 123) | 0.05 (122-> 124) | 0.01 (122-> 134) |
| <b>S2</b>  | 3.507 | 353.6 | 0.91 (122-> 124) | 0.04 (122-> 123) | 0.01 (122-> 127) |
| <b>S3</b>  | 3.931 | 315.4 | 0.90 (122-> 125) | 0.03 (120-> 123) | 0.03 (122-> 126) |
| <b>S4</b>  | 4.101 | 302.3 | 0.88 (122-> 126) | 0.07 (122-> 127) | 0.02 (122-> 125) |
| <b>S5</b>  | 4.142 | 299.3 | 0.58 (122-> 127) | 0.24 (122-> 128) | 0.06 (120-> 123) |
| <b>S6</b>  | 4.164 | 297.8 | 0.69 (122-> 128) | 0.24 (122-> 127) | 0.02 (122-> 129) |
| <b>S7</b>  | 4.358 | 284.5 | 0.68 (122-> 130) | 0.18 (122-> 129) | 0.07 (121-> 123) |
| <b>S8</b>  | 4.47  | 277.4 | 0.75 (121-> 123) | 0.06 (122-> 129) | 0.04 (122-> 130) |
| <b>S9</b>  | 4.517 | 274.5 | 0.65 (122-> 129) | 0.22 (122-> 130) | 0.03 (122-> 127) |
| <b>S10</b> | 4.589 | 270.2 | 0.84 (122-> 131) | 0.04 (121-> 124) | 0.02 (122-> 134) |
| <b>S11</b> | 4.712 | 263.1 | 0.42 (120-> 123) | 0.11 (121-> 125) | 0.09 (121-> 124) |

**Table S7.** Nature of selected electronic transitions in (*S,P<sub>p</sub>*)-**3b** obtained via sTD-DFT using the CAM-B3LYP functional

**Table S8.** Transition energies (E), rotatory strengths (R), transition electric dipole moments ( $\mu$ ), transition magnetic dipole moments ( $m$ ), the angle between these two vectors ( $\theta$ ), and the theoretical dissymmetry factors ( $g_{\text{abs}}$ ) for the first four states obtained via TD-DFT calculations using the CAM-B3LYP functional.

| Compound                                 | State | E (eV)   | E (cm <sup>-1</sup> ) | $\lambda$ (nm) | R (10 <sup>-40</sup> cgs) | $\mu_x$ (a.u.) | $\mu_y$ (a.u.) | $\mu_z$ (a.u.) | $m_x$ (a.u.) | $m_y$ (a.u.) | $m_z$ (a.u.) | $\theta_{\mu,m}$ (°) | $g_{\text{abs}}$ |
|------------------------------------------|-------|----------|-----------------------|----------------|---------------------------|----------------|----------------|----------------|--------------|--------------|--------------|----------------------|------------------|
| <b>(S,M<sub>p</sub>)-3a</b>              | S1    | 3.390334 | 27344.9               | 365.7          | 116.56492                 | -1.49668       | 1.82179        | 0.01709        | 0.21688      | 0.04233      | 0.01389      | 118.2694             | -0.0013          |
|                                          | S2    | 3.82939  | 30886.1               | 323.8          | -24.13426                 | -0.64794       | 0.66339        | -1.36415       | 0.27677      | 0.33226      | -0.00741     | 85.88475             | 0.00055          |
|                                          | S3    | 4.334803 | 34962.5               | 286            | 6.46994                   | -0.21534       | -0.02777       | 0.01375        | 0.0552       | 0.01742      | -0.0984      | 123.5418             | -0.00847         |
|                                          | S4    | 4.412152 | 35586.4               | 281            | 14.20285                  | -0.40487       | 0.3747         | 0.32051        | 0.31613      | 0.24839      | 0.01496      | 96.74004             | -0.00216         |
| <b>(S,P<sub>p</sub>)-3a<sup>++</sup></b> | D1    | 2.294203 | 18504                 | 540.4          | -6.183                    | -0.78476       | 1.41522        | -0.58353       | 0.04225      | 0.02373      | -0.02173     | 81.7512              | 0.000129         |
|                                          | D2    | 2.717158 | 21915.4               | 456.3          | -9.28767                  | 0.36679        | -1.32092       | -0.26694       | -0.33104     | -0.09977     | 0.11262      | 92.22275             | -0.00029         |
|                                          | D3    | 3.017292 | 24336.1               | 410.9          | -9.02212                  | 0.16756        | -0.79667       | -0.04445       | 0.01728      | -0.0211      | 0.01272      | 38.73156             | 0.000841         |
|                                          | D4    | 3.089717 | 24920.3               | 401.3          | -28.11756                 | 0.09067        | 0.86058        | 1.0658         | -0.13411     | 0.06415      | 0.01557      | 73.10413             | 0.000924         |
| <b>(S,P<sub>a</sub>)-3a<sup>++</sup></b> | S1    | 3.185394 | 25691.9               | 389.2          | 18.51664                  | 1.87814        | 2.2383         | -0.33872       | 0.05003      | -0.13118     | -0.47349     | 91.54933             | -0.00013         |
|                                          | S2    | 3.532097 | 28488.3               | 351            | -2.50839                  | 0.35202        | -0.00869       | 0.00231        | 0.01332      | -0.05062     | 0.08339      | 81.17092             | 0.001253         |
|                                          | S3    | 3.590279 | 28957.6               | 345.3          | -3.87974                  | 0.07115        | 0.12887        | 0.04314        | 0.07974      | -0.03567     | 0.1658       | 73.36568             | 0.010212         |
|                                          | S4    | 3.68746  | 29741.4               | 336.2          | -3.69457                  | 0.19256        | -0.46192       | -0.21191       | -0.1178      | 0.01767      | -0.18253     | 86.20785             | 0.000775         |
| <b>(S,P<sub>p</sub>)-3b</b>              | S1    | 3.2843   | 26489.7               | 377.5          | 97.54772                  | 0.48826        | -0.08789       | -1.06284       | -0.19914     | -0.45108     | 0.1405       | 110.1254             | -0.00439         |
|                                          | S2    | 3.585589 | 28919.7               | 345.8          | -243.08054                | 1.50635        | 0.32318        | -0.89326       | 0.26201      | 0.17669      | -0.07145     | 26.6696              | 0.004748         |
|                                          | S3    | 4.136262 | 33361.2               | 299.7          | -24.5065                  | -0.73196       | 0.04979        | -0.2413        | -0.00961     | 0.33469      | -0.11719     | 79.06402             | 0.002545         |
|                                          | S4    | 4.32852  | 34911.9               | 286.4          | 61.04354                  | 0.78139        | 0.32932        | 0.16429        | -0.0605      | -0.15551     | -0.18868     | 126.5258             | -0.00507         |
| <b>(S,P<sub>p</sub>)-3b<sup>++</sup></b> | D1    | 2.295058 | 18510.9               | 540.2          | 5.69577                   | -0.37035       | -0.37767       | -1.0412        | 0.04357      | 0.30529      | -0.11463     | 91.80205             | -0.00026         |
|                                          | D2    | 2.452194 | 19778.3               | 505.6          | 9.87623                   | -1.07258       | 0.16025        | 1.0187         | 0.02038      | 0.013        | -0.00115     | 125.5732             | -0.00028         |
|                                          | D3    | 2.684708 | 21653.6               | 461.8          | -60.65573                 | -0.80667       | -0.48892       | -0.61068       | -0.11471     | -0.08201     | 0.0065       | 35.79502             | 0.002976         |
|                                          | D4    | 2.942269 | 23731                 | 421.4          | -36.07463                 | -0.94683       | -0.22457       | -0.21246       | -0.07843     | -0.03646     | 0.02791      | 32.29481             | 0.002253         |
| <b>(S,M<sub>a</sub>)-3b<sup>++</sup></b> | S1    | 2.779537 | 22418.5               | 446.1          | -2.41583                  | 2.00602        | 0.78408        | 2.25753        | -0.13925     | 0.53213      | -0.05963     | 89.89124             | 9.83E-06         |
|                                          | S2    | 3.337907 | 26922                 | 371.4          | -2.56416                  | 0.073          | -0.36187       | 0.27254        | -0.27803     | 0.19821      | 0.35387      | 88.8773              | 0.000613         |
|                                          | S3    | 3.358582 | 27088.8               | 369.2          | -9.51193                  | -0.39448       | -0.17061       | -0.26206       | -0.23756     | 0.25021      | 0.11237      | 83.21608             | 0.002487         |
|                                          | S4    | 3.634184 | 29311.7               | 341.2          | -9.04586                  | 0.51118        | 0.21455        | 0.03947        | 0.04872      | -0.02827     | 0.01158      | 52.86085             | 0.001824         |



**Figure S38.** Transition densities for the first four transitions of **(S,M<sub>p</sub>)-3a**, **(S,P<sub>p</sub>)-3a<sup>++</sup>**, and **(S,P<sub>a</sub>)-3a<sup>++</sup>** obtained via TD-DFT using the CAM-B3LYP functional. Isosurface = 0.002.

### S1

148a -> 149a : 0.933169  
148a -> 158a : 0.017006

### S2

148a -> 150a : 0.912072  
148a -> 151a : 0.018115

### S3

148a -> 151a : 0.288073  
148a -> 152a : 0.056422  
148a -> 153a : 0.145276  
148a -> 154a : 0.044313  
148a -> 155a : 0.394915

**S4**

|                |          |
|----------------|----------|
| 148a -> 151a : | 0.036492 |
| 148a -> 152a : | 0.448664 |
| 148a -> 153a : | 0.167684 |
| 148a -> 154a : | 0.089354 |
| 148a -> 155a : | 0.037594 |
| 148a -> 156a : | 0.141415 |

**Figure S39.** Nature of selected electronic transitions in (**S**,**M<sub>p</sub>**)-**3a** obtained via TD-DFT using the CAM-B3LYP functional

**D1**

|                |          |
|----------------|----------|
| 148a -> 149a : | 0.863092 |
| 147b -> 148b : | 0.033875 |

**D2**

|                |          |
|----------------|----------|
| 146a -> 149a : | 0.015579 |
| 148a -> 149a : | 0.044363 |
| 148a -> 150a : | 0.209108 |
| 140b -> 150b : | 0.014033 |
| 141b -> 148b : | 0.016702 |
| 145b -> 148b : | 0.034210 |
| 147b -> 148b : | 0.487864 |
| 147b -> 149b : | 0.016612 |

**D3**

|                |          |
|----------------|----------|
| 146a -> 149a : | 0.016598 |
| 148a -> 150a : | 0.212183 |
| 139b -> 148b : | 0.019822 |
| 141b -> 148b : | 0.017982 |
| 143b -> 148b : | 0.010301 |
| 144b -> 148b : | 0.181562 |
| 145b -> 148b : | 0.312317 |
| 146b -> 148b : | 0.055598 |
| 147b -> 148b : | 0.014617 |
| 147b -> 149b : | 0.025375 |

**D4**

|                |          |
|----------------|----------|
| 146a -> 150a : | 0.011265 |
| 148a -> 150a : | 0.338583 |
| 148a -> 157a : | 0.010569 |

|                |          |
|----------------|----------|
| 144b -> 148b : | 0.044720 |
| 145b -> 148b : | 0.052298 |
| 146b -> 148b : | 0.044425 |
| 147b -> 148b : | 0.289498 |

**Figure S40.** Nature of selected electronic transitions in **(S,M<sub>p</sub>)-3a<sup>++</sup>** obtained via TD-DFT using the CAM-B3LYP functional

#### **S1**

|                |          |
|----------------|----------|
| 141a -> 148a : | 0.023078 |
| 144a -> 148a : | 0.110470 |
| 145a -> 148a : | 0.078281 |
| 146a -> 148a : | 0.747922 |

#### **S2**

|                |          |
|----------------|----------|
| 141a -> 148a : | 0.388336 |
| 142a -> 148a : | 0.042965 |
| 143a -> 148a : | 0.038722 |
| 144a -> 148a : | 0.219422 |
| 145a -> 148a : | 0.026697 |
| 146a -> 148a : | 0.022328 |
| 147a -> 148a : | 0.237230 |

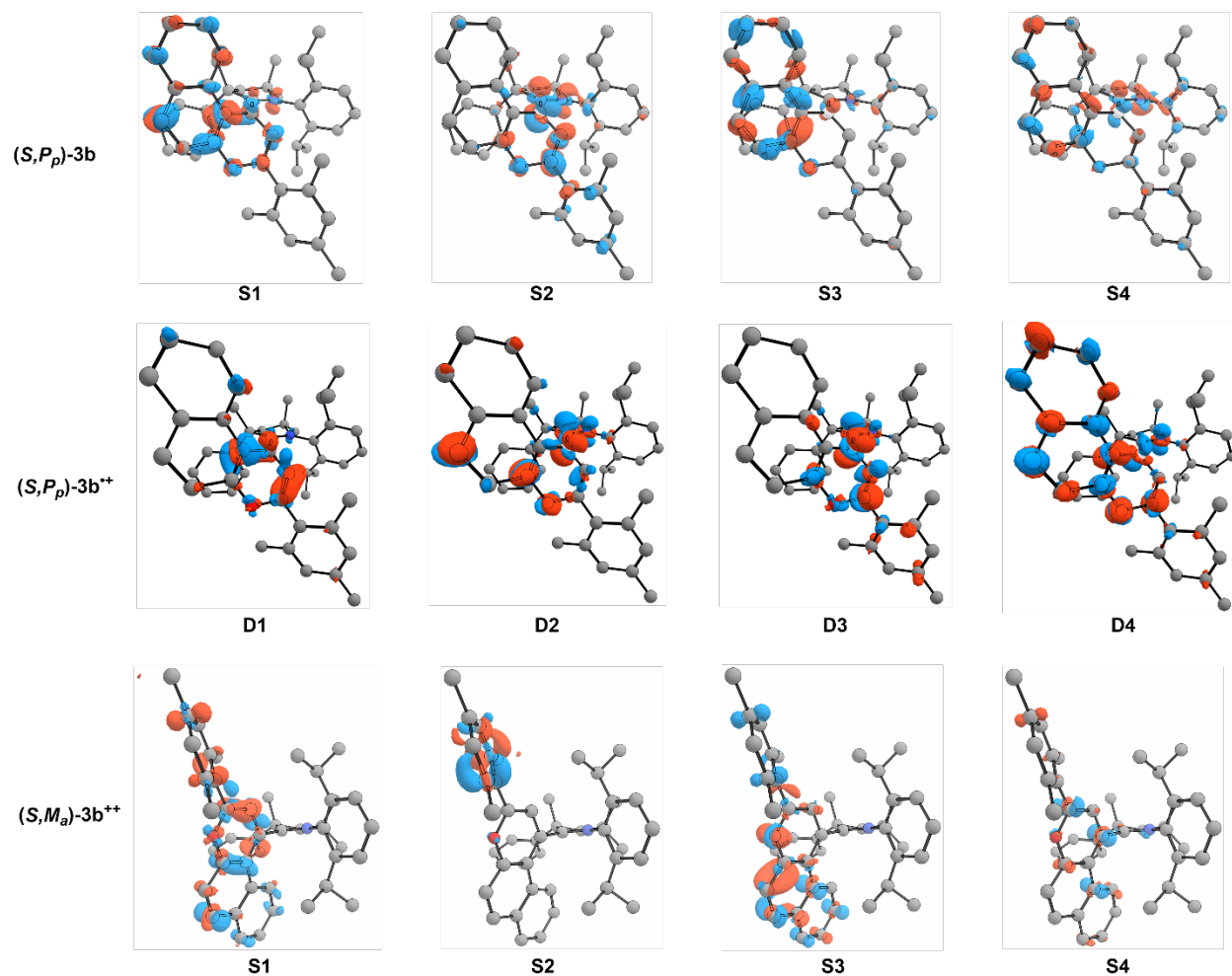
#### **S3**

|                |          |
|----------------|----------|
| 141a -> 148a : | 0.058174 |
| 142a -> 148a : | 0.078587 |
| 143a -> 148a : | 0.023395 |
| 144a -> 148a : | 0.062689 |
| 145a -> 148a : | 0.035558 |
| 146a -> 148a : | 0.012720 |
| 147a -> 148a : | 0.714152 |

#### **S4**

|                |          |
|----------------|----------|
| 141a -> 148a : | 0.228591 |
| 142a -> 148a : | 0.169550 |
| 142a -> 150a : | 0.012735 |
| 143a -> 148a : | 0.010243 |
| 145a -> 148a : | 0.448697 |
| 146a -> 148a : | 0.070239 |
| 147a -> 148a : | 0.033523 |

**Figure S41.** Nature of selected electronic transitions in **(S,M<sub>p</sub>)-3a<sup>++</sup>** obtained via TD-DFT using the CAM-B3LYP functional



**Figure S42.** Transition densities for the first four transitions of **(S,P<sub>r</sub>)-3b**, **(S,P<sub>r</sub>)-3b<sup>++</sup>**, and **(S,M<sub>a</sub>)-3b<sup>++</sup>**, obtained via TD-DFT using the CAM-B3LYP functional. Isosurface = 0.002.

### S1

173a -> 174a : 0.903573  
 173a -> 175a : 0.027863  
 173a -> 182a : 0.011726  
 173a -> 185a : 0.011461

### S2

|                |          |
|----------------|----------|
| 173a -> 174a : | 0.027578 |
| 173a -> 175a : | 0.887889 |
| 173a -> 182a : | 0.011601 |
| 173a -> 185a : | 0.021491 |

### S3

|                |          |
|----------------|----------|
| 171a -> 174a : | 0.090683 |
| 172a -> 175a : | 0.022148 |
| 172a -> 176a : | 0.011024 |
| 173a -> 176a : | 0.686901 |
| 173a -> 177a : | 0.028433 |
| 173a -> 178a : | 0.010811 |
| 173a -> 179a : | 0.073495 |

### S4

|                |          |
|----------------|----------|
| 171a -> 174a : | 0.058699 |
| 172a -> 175a : | 0.011837 |
| 172a -> 176a : | 0.014483 |
| 172a -> 179a : | 0.011099 |
| 173a -> 176a : | 0.029796 |
| 173a -> 177a : | 0.371919 |
| 173a -> 178a : | 0.302446 |
| 173a -> 179a : | 0.070524 |
| 173a -> 181a : | 0.032660 |
| 173a -> 182a : | 0.025123 |

**Fig. S39.** Nature of selected electronic transitions in **(S,P<sub>p</sub>)-3b** according to TD-DFT calculations.

### D1

|                |          |
|----------------|----------|
| 172a -> 174a : | 0.040264 |
| 172a -> 175a : | 0.012574 |
| 173a -> 174a : | 0.099131 |
| 173a -> 175a : | 0.068137 |
| 164b -> 173b : | 0.018037 |
| 172b -> 173b : | 0.567992 |
| 172b -> 174b : | 0.064031 |

### D2

|                |          |
|----------------|----------|
| 172a -> 174a : | 0.010642 |
| 172a -> 175a : | 0.043938 |
| 173a -> 174a : | 0.519019 |
| 164b -> 173b : | 0.015395 |

172b -> 173b : 0.225384  
172b -> 174b : 0.060793

**D3**

172a -> 174a : 0.034067  
173a -> 174a : 0.127335  
173a -> 175a : 0.398991  
168b -> 173b : 0.032328  
169b -> 173b : 0.029673  
170b -> 173b : 0.035340  
170b -> 174b : 0.011972  
172b -> 173b : 0.111546  
172b -> 174b : 0.045327

**D4**

170a -> 174a : 0.011086  
172a -> 175a : 0.011147  
173a -> 174a : 0.011722  
173a -> 175a : 0.251884  
164b -> 173b : 0.032017  
166b -> 173b : 0.011312  
168b -> 173b : 0.179565  
169b -> 173b : 0.168717  
170b -> 173b : 0.136311  
171b -> 173b : 0.019504  
172b -> 174b : 0.011322

**Figure S43.** Nature of selected electronic transitions in **(S,P<sub>p</sub>)-3b<sup>++</sup>** according to TD-DFT calculations.

**S1**

172a -> 173a : 0.942109

**S2**

170a -> 173a : 0.882544  
170a -> 175a : 0.030252  
170a -> 177a : 0.011244  
171a -> 173a : 0.056594

**S3**

165a -> 173a : 0.049066



167a -> 173a : 0.121281  
 168a -> 173a : 0.195025  
 169a -> 173a : 0.523738  
 170a -> 173a : 0.016183  
 171a -> 173a : 0.025370  
 172a -> 175a : 0.030845

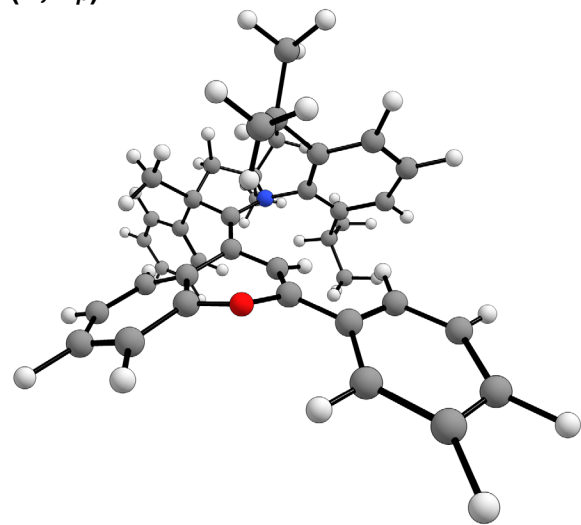
#### S4

165a -> 173a : 0.105317  
 166a -> 173a : 0.019447  
 168a -> 173a : 0.551096  
 169a -> 173a : 0.227522  
 171a -> 173a : 0.026804  
 172a -> 173a : 0.013202

**Figure S44.** Nature of selected electronic transitions in **(S,P<sub>p</sub>)-3b<sup>++</sup>** according to TD-DFT calculations.

#### Cartesian Coordinates

##### (S,M<sub>p</sub>)-3a

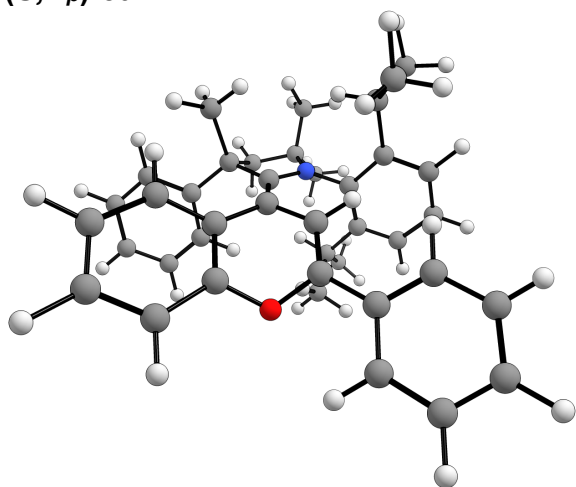


|   |              |              |              |
|---|--------------|--------------|--------------|
| O | 10.517543947 | 13.470808079 | 15.403087937 |
| N | 6.998181392  | 16.410217849 | 13.555022756 |
| C | 9.035427215  | 16.581536614 | 12.175733674 |
| C | 9.712669561  | 13.309113853 | 14.284352593 |
| C | 8.667086018  | 14.595623116 | 16.515689170 |
| C | 8.149328802  | 14.976266322 | 15.180716819 |
| C | 10.253519360 | 12.346263961 | 13.327532439 |
| C | 7.734234223  | 16.063033616 | 12.368945642 |
| C | 9.870398896  | 13.863654413 | 16.557790728 |
| C | 7.387036484  | 16.078313024 | 14.856036651 |

|   |              |              |              |
|---|--------------|--------------|--------------|
| C | 6.835847774  | 17.144749822 | 15.817325471 |
| C | 8.577052009  | 14.022363119 | 14.168337963 |
| H | 7.965051086  | 13.858840520 | 13.295946379 |
| C | 8.042039671  | 14.794884446 | 17.758016044 |
| H | 7.067895134  | 15.262895359 | 17.792158417 |
| C | 7.140049107  | 15.238749091 | 11.388527335 |
| C | 9.809146143  | 12.328574798 | 11.994099016 |
| H | 9.077388800  | 13.055204824 | 11.654919528 |
| C | 9.722358780  | 17.503504318 | 13.168199365 |
| H | 8.987096226  | 17.785654919 | 13.925758175 |
| C | 10.317103285 | 11.406859133 | 11.090342288 |
| H | 9.959706315  | 11.412866677 | 10.063914648 |
| C | 6.057464003  | 17.586408767 | 13.515569356 |
| C | 10.480668211 | 13.477965043 | 17.744877275 |
| H | 11.420698477 | 12.936209359 | 17.695862412 |
| C | 11.244954488 | 11.429370245 | 13.713694730 |
| H | 11.607177786 | 11.431404650 | 14.736094834 |
| C | 9.870401922  | 13.771206204 | 18.958532272 |
| H | 10.343105712 | 13.475010396 | 19.890332155 |
| C | 5.625573059  | 16.655042146 | 16.625078834 |
| C | 7.924590094  | 17.756276437 | 16.730372294 |
| H | 8.181753047  | 17.131525853 | 17.584273141 |
| H | 7.579154355  | 18.723525771 | 17.108345776 |
| H | 8.833131368  | 17.933666687 | 16.146853521 |
| C | 8.633382395  | 14.409669569 | 18.956599336 |
| H | 8.116118761  | 14.606877087 | 19.891214409 |
| C | 9.720611655  | 16.243910261 | 11.006060593 |
| H | 10.726973366 | 16.624444913 | 10.852864351 |
| C | 5.023639232  | 15.414285210 | 16.397254399 |
| H | 5.444446474  | 14.747851926 | 15.648648430 |
| C | 6.388097868  | 18.264657049 | 14.844113544 |
| H | 5.546981065  | 18.837354380 | 15.244049204 |
| H | 7.226078392  | 18.955385267 | 14.690555729 |
| C | 5.852906680  | 13.059938125 | 11.542543257 |
| H | 6.230709312  | 12.696340099 | 10.580529884 |

|   |              |              |              |
|---|--------------|--------------|--------------|
| H | 4.852338130  | 12.638514880 | 11.690995661 |
| H | 6.501123754  | 12.665294791 | 12.331535632 |
| C | 11.755225697 | 10.512845952 | 12.799796279 |
| H | 12.518522752 | 9.808540085  | 13.119512797 |
| C | 5.776340509  | 14.595414373 | 11.569003361 |
| H | 5.399044347  | 14.886730549 | 12.553631341 |
| C | 7.851697482  | 14.964926544 | 10.218133689 |
| H | 7.400541300  | 14.334909434 | 9.455839767  |
| C | 5.070800016  | 17.484157500 | 17.606945143 |
| H | 5.506441235  | 18.459542007 | 17.804773066 |
| C | 9.135340401  | 15.455289557 | 10.025648832 |
| H | 9.681657899  | 15.217259785 | 9.116914724  |
| C | 6.343403145  | 18.515770020 | 12.339320993 |
| H | 7.377437608  | 18.867003250 | 12.344949349 |
| H | 5.684946100  | 19.386705624 | 12.422456595 |
| H | 6.142709658  | 18.028486610 | 11.379987545 |
| C | 10.226059701 | 18.795232301 | 12.503605145 |
| H | 11.077017129 | 18.597075545 | 11.843079519 |
| H | 10.562459005 | 19.497914777 | 13.273956049 |
| H | 9.447234078  | 19.284197854 | 11.909985053 |
| C | 11.293809817 | 10.491296853 | 11.486202045 |
| H | 11.695601024 | 9.775936054  | 10.774408496 |
| C | 10.889545151 | 16.805781452 | 13.882680358 |
| H | 11.657234124 | 16.503587189 | 13.160584479 |
| H | 10.552946064 | 15.917191603 | 14.420604658 |
| H | 11.352682022 | 17.491140103 | 14.601925351 |
| C | 4.579477344  | 17.168912622 | 13.457760437 |
| H | 4.316616287  | 16.767814454 | 12.477404015 |
| H | 3.959006439  | 18.055246840 | 13.630278201 |
| H | 4.331588456  | 16.428385640 | 14.220372728 |
| C | 3.908966546  | 15.010154417 | 17.127267784 |
| H | 3.463566374  | 14.038667170 | 16.930389404 |
| C | 4.787605829  | 15.076787762 | 10.495204600 |
| H | 5.068662588  | 14.687748201 | 9.509881877  |
| H | 4.760696065  | 16.168756382 | 10.424860849 |
| H | 3.776623367  | 14.718860380 | 10.720720939 |
| C | 3.958789688  | 17.082138830 | 18.341344146 |
| H | 3.549738914  | 17.743487547 | 19.100516418 |
| C | 3.371274114  | 15.841239195 | 18.105789936 |
| H | 2.504381892  | 15.525972552 | 18.679729972 |

(*S,P*)-3a

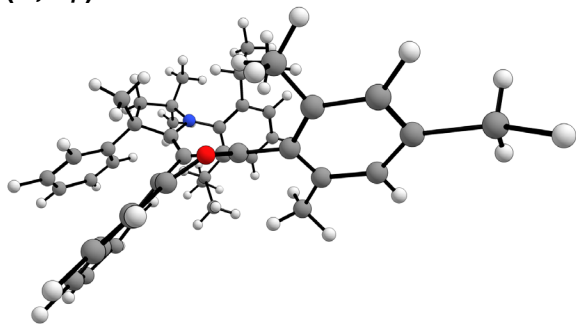


|   |               |               |               |
|---|---------------|---------------|---------------|
| O | -10.005651156 | -13.440053208 | -15.618622390 |
| N | -6.704424509  | -16.353935999 | -13.369227213 |
| C | -8.846651197  | -16.524474098 | -12.177527677 |
| C | -9.544359796  | -13.343205633 | -14.311800725 |

|   |               |               |               |
|---|---------------|---------------|---------------|
| C | -7.761850644  | -14.085702875 | -16.293796865 |
| C | -7.556702713  | -14.644998797 | -14.936526286 |
| C | -10.477701503 | -12.653974093 | -13.425288084 |
| C | -7.526096153  | -16.025350581 | -12.240194603 |
| C | -9.026887480  | -13.542656338 | -16.586174051 |
| C | -6.786039361  | -15.751570260 | -14.620319215 |
| C | -5.844804245  | -16.536405270 | -15.558560921 |
| C | -8.383861476  | -13.934860799 | -13.968574924 |
| H | -8.087533927  | -13.871769746 | -12.935204572 |
| C | -6.813345969  | -13.985589584 | -17.320229441 |
| H | -5.797515474  | -14.292726438 | -17.129085536 |
| C | -7.017709264  | -15.189048586 | -11.224363634 |
| C | -10.145808301 | -12.377852916 | -12.085846973 |
| H | -9.169155607  | -12.647860414 | -11.696635029 |
| C | -9.461652604  | -17.407294573 | -13.251701920 |
| H | -8.674750360  | -17.674844030 | -13.968574603 |
| C | -11.054069074 | -11.756866434 | -11.242090508 |
| H | -10.771793416 | -11.553866440 | -10.212400998 |
| C | -5.830737013  | -17.574701038 | -13.321925414 |
| C | -9.373618769  | -13.071932657 | -17.846574912 |
| H | -10.378818153 | -12.691986408 | -18.004043705 |
| C | -11.750418372 | -12.272528224 | -13.884161867 |
| H | -12.031654881 | -12.469621821 | -14.912464474 |
| C | -8.428916042  | -13.080907590 | -18.866786077 |
| H | -8.694185951  | -12.720254087 | -18.56222190  |
| C | -6.409791867  | -16.865353899 | -16.935793464 |
| C | -4.462085836  | -15.833777593 | -15.613695229 |
| H | -3.656434059  | -16.566695762 | -15.723564405 |
| H | -4.378720530  | -15.132687743 | -16.444726940 |
| H | -4.284391928  | -15.265750714 | -14.698524322 |
| C | -7.135219755  | -13.512280052 | -18.587895064 |
| H | -6.367836813  | -13.481804069 | -19.356072134 |
| C | -9.635600854  | -16.172129817 | -11.080065759 |
| H | -10.658888812 | -16.533248092 | -11.022580152 |
| C | -5.568992891  | -17.041415195 | -18.036183228 |
| H | -4.500137259  | -16.876286921 | -17.939743517 |
| C | -5.715032124  | -17.888024935 | -14.810984772 |
| H | -4.786468284  | -18.405529816 | -15.068085903 |
| H | -6.555558354  | -18.528969710 | -15.099502423 |
| C | -5.770169725  | -13.008180143 | -11.502827942 |
| H | -6.267668138  | -12.757836826 | -12.444979534 |
| H | -6.332949319  | -12.546206447 | -10.683609246 |
| H | -4.770913364  | -12.558416931 | -11.521841828 |
| C | -12.656138538 | -11.648620942 | -13.031059278 |
| H | -13.634950583 | -11.365419298 | -13.408938248 |
| C | -5.651464022  | -14.528931657 | -11.303002829 |
| H | -5.133483071  | -14.929760220 | -12.177142598 |
| C | -7.841955629  | -14.883531296 | -10.137879750 |
| H | -7.468884431  | -14.232168213 | -9.351889944  |
| C | -7.777001100  | -17.112618247 | -17.100980810 |
| H | -8.446954767  | -16.988500939 | -16.253690592 |
| C | -9.139806616  | -15.368224383 | -10.062981158 |
| H | -9.771722950  | -15.107155042 | -9.218203701  |
| C | -6.480848605  | -18.738570416 | -12.570246939 |
| H | -5.783509688  | -19.582511722 | -12.580045180 |
| H | -6.683186608  | -18.481836086 | -11.525256412 |
| H | -7.406620309  | -19.064087318 | -13.043981048 |
| C | -10.041124850 | -18.699986700 | -12.653096901 |
| H | -10.954022594 | -18.491698179 | -12.084481883 |
| H | -10.304266397 | -19.397773023 | -13.455776898 |
| H | -9.334727988  | -19.195557087 | -11.980359876 |
| C | -12.318212724 | -11.387540656 | -11.706683548 |
| H | -13.026358460 | -10.899965047 | -11.042942193 |
| C | -10.558355202 | -16.680103591 | -14.044986920 |
| H | -10.999677026 | -17.365535133 | -14.777979794 |
| H | -11.359578024 | -16.332607044 | -13.381756149 |
| H | -10.157638616 | -15.815964525 | -14.577320000 |
| C | -4.467035450  | -17.323571704 | -12.658001099 |

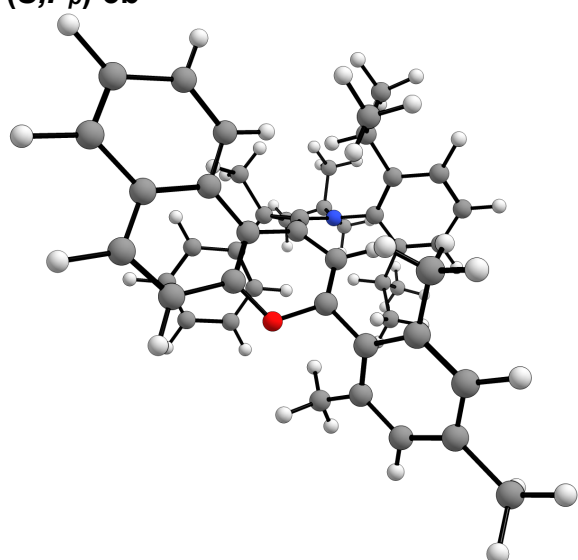
|   |              |               |               |
|---|--------------|---------------|---------------|
| H | -4.589716945 | -17.131105043 | -11.589187160 |
| H | -3.859459937 | -18.228489841 | -12.763941119 |
| H | -3.916855690 | -16.494040204 | -13.101958139 |
| C | -6.084327965 | -17.411899222 | -19.277882443 |
| H | -5.411243891 | -17.533067317 | -20.122523997 |
| C | -4.805428286 | -14.818891312 | -10.054184097 |
| H | -3.776627872 | -14.475158458 | -10.208450229 |
| H | -5.204039037 | -14.291836724 | -9.180146934  |
| H | -4.781078985 | -15.886864941 | -9.816358042  |
| C | -8.295275787 | -17.474107659 | -18.338146129 |
| H | -9.364175524 | -17.638760531 | -18.445722879 |
| C | -7.450346543 | -17.619099122 | -19.438538447 |
| H | -7.854245967 | -17.896217555 | -20.408314540 |

**(S,M<sub>p</sub>)-3b**



|   |             |              |             |
|---|-------------|--------------|-------------|
| O | 5.513178606 | 3.578108543  | 6.377981651 |
| N | 9.386093102 | 4.504373849  | 3.636025490 |
| C | 5.573299695 | 3.438627717  | 4.985062778 |
| C | 6.481634236 | 4.137414412  | 4.283609230 |
| H | 6.517511982 | 3.979136457  | 3.213732249 |
| C | 7.407171428 | 5.030152543  | 4.985692665 |
| C | 6.720252174 | 5.643054455  | 6.147773794 |
| C | 5.829940309 | 4.835285294  | 6.841355136 |
| C | 6.715684793 | 7.032901768  | 6.496431022 |
| C | 6.024613552 | 7.469774769  | 7.675632445 |
| C | 5.306907772 | 6.530129102  | 8.453423423 |
| H | 4.807313678 | 6.861273521  | 9.360285003 |
| C | 5.171915035 | 5.237065814  | 8.016495637 |
| H | 4.551516423 | 4.516228607  | 8.541016295 |
| C | 7.300938806 | 8.019658438  | 5.670025869 |
| H | 7.798894024 | 7.699782494  | 4.763213876 |
| C | 7.232612888 | 9.353255314  | 5.990125488 |
| H | 7.678835966 | 10.088764552 | 5.326468186 |
| C | 6.603292197 | 9.775160696  | 7.180440568 |
| H | 6.574243625 | 10.831363745 | 7.434036742 |
| C | 6.015254290 | 8.847698639  | 8.005563579 |
| H | 5.503700774 | 9.159193431  | 8.913609778 |
| C | 4.601087018 | 2.471988876  | 4.451365652 |
| C | 4.514674815 | 1.177349754  | 5.017725427 |
| C | 3.621568338 | 0.256575803  | 4.477727966 |
| H | 3.574299427 | -0.742567060 | 4.908167736 |
| C | 2.799140138 | 0.572969224  | 3.394420887 |
| C | 2.895733319 | 1.850021185  | 2.850272454 |
| H | 2.256973315 | 2.118869892  | 2.010898796 |
| C | 3.779286251 | 2.809170609  | 3.354857068 |
| C | 5.391533397 | 0.745962227  | 6.164186720 |
| H | 5.454974102 | -0.345126943 | 6.199655247 |
| H | 4.992807674 | 1.093192518  | 7.123489410 |
| H | 6.403686124 | 1.152365429  | 6.075221790 |
| C | 1.830545319 | -0.439291773 | 2.846954931 |
| H | 1.020944200 | -0.630501901 | 3.561592512 |
| H | 2.326595022 | -1.397923009 | 2.659898530 |

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| C | 8.763465301  | 5.063132079  | 4.744751379  |
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| C | 10.439443984 | 7.820507840  | 5.575963677  |
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| C | 10.672601231 | 9.090724226  | 6.085852000  |
| H | 11.042193861 | 9.873251833  | 5.428279915  |
| C | 10.432306358 | 9.359540100  | 7.431067725  |
| H | 10.617133336 | 10.350114964 | 7.837609449  |
| C | 9.944094067  | 8.343208526  | 8.243926631  |
| H | 9.740499654  | 8.534734139  | 9.294304852  |
| C | 9.711788862  | 7.070654673  | 7.724759027  |
| H | 9.325959395  | 6.305869860  | 8.389629150  |
| C | 11.339893265 | 5.728514950  | 2.654181793  |
| H | 12.426211543 | 5.841773332  | 2.728857832  |
| H | 10.880953419 | 6.697807254  | 2.849909980  |
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| H | 11.391731477 | 3.080176003  | 2.212557535  |
| H | 11.361640299 | 2.529416061  | 3.911536258  |
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| C | 8.030195702  | 2.621241681  | 0.707391565  |
| H | 7.928000144  | 1.594167606  | 0.367952703  |
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| H | 7.860531081  | 0.083999927  | 3.876159494  |
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| H | 6.261304436  | 6.595474537  | 3.023086408  |
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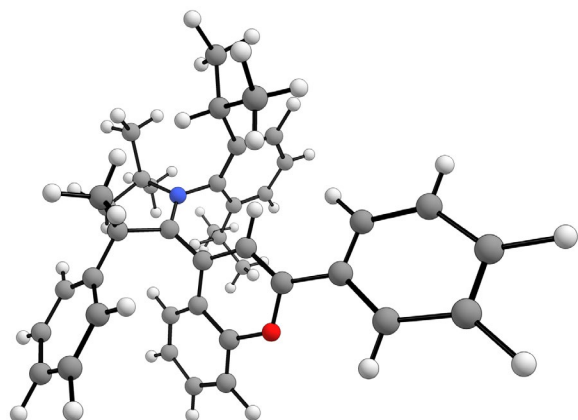
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| C | 7.792485825  | 4.630541076 | 5.185807250  |
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| C | 6.047085003  | 5.275876404 | 6.764292194  |
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| C | 6.321799246  | 5.547488748 | 9.131888354  |
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| C | 5.539419440  | 5.677776747 | 8.011721008  |
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| H | 9.616244815  | 3.374076734 | 6.765530618  |
| C | 9.919675330  | 3.366303991 | 8.880137606  |
| H | 10.802307002 | 2.734820964 | 8.822814473  |
| C | 9.471593978  | 3.834721081 | 10.134243085 |
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| H | 7.946372919  | 4.922911658 | 11.173811013 |
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| C | 3.333328097  | 4.990858925 | 3.447978655  |
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| H | 2.629639305  | 1.240131188 | 2.766824384  |
| C | 3.971798752  | 2.643609719 | 3.677548733  |
| C | 3.580028315  | 6.462921460 | 3.652113731  |
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| H | 3.016348409  | 7.045245972 | 2.918004965  |
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| H | 0.609507789  | 1.765823496 | 1.637674970  |
| H | 0.779764630  | 3.228601862 | 0.651555408  |
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| H | 5.292020102  | 1.766495438 | 5.153486143  |
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| H | 9.904025923  | 7.677440827  | 3.539774456  |
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| H | 11.928621776 | 4.421576937  | 4.237066602  |
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| C | 9.364901218  | 2.535791647  | 2.858831952  |
| C | 9.020177420  | 1.493077852  | 1.994443754  |
| H | 9.305881628  | 0.476659521  | 2.251053845  |
| C | 8.319144055  | 1.724895616  | 0.820903784  |
| H | 8.069342971  | 0.899128784  | 0.160124915  |
| C | 7.917556540  | 3.015422958  | 0.510944537  |
| H | 7.332502281  | 3.189374707  | -0.387864696 |
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| C | 9.120147777  | 1.327899876  | 5.039996154  |
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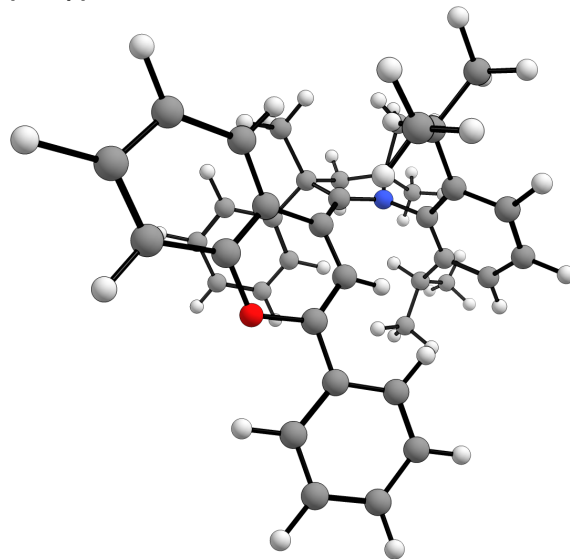
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| C | 10.630441205 | 12.530268116 | 14.219284627 |
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| C | 6.506647065  | 18.131108727 | 13.409553730 |
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| H | 10.803978418 | 13.783849641 | 12.457116660 |
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| H | 3.701871347  | 13.344825644 | 16.696263647 |
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| H | 7.134169413  | 14.791345558 | 19.664070703 |
| C | 5.251803714  | 15.817581769 | 19.439360968 |
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**(S,P<sub>p</sub>)-3a<sup>++</sup>**

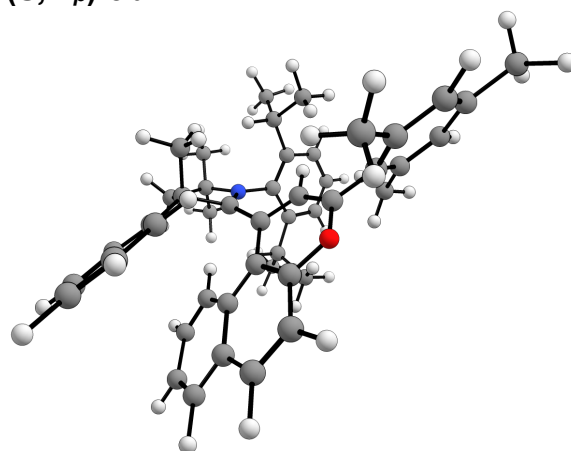


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(S,M<sub>p</sub>)-3b<sup>+</sup>

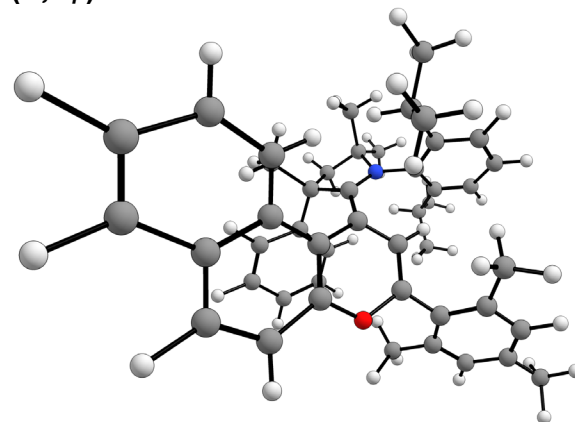


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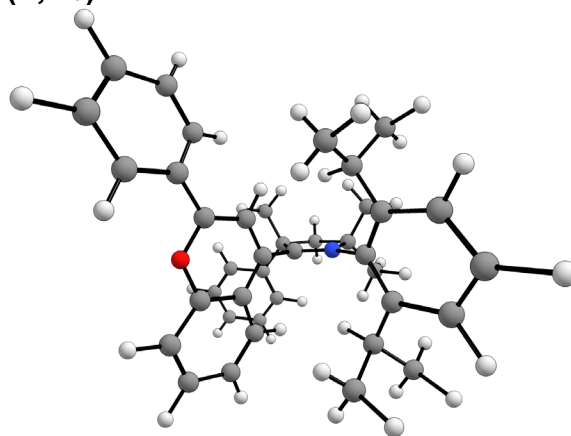
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(S,M<sub>a</sub>)-3a<sup>++</sup>

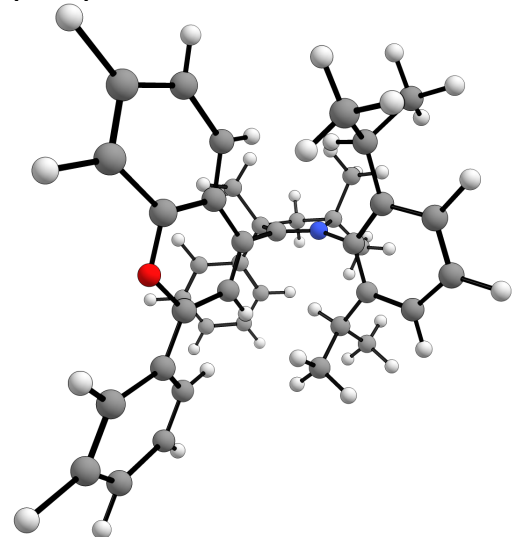


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| H | 8.442614539  | 16.262165634 | 9.275640616  |
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**(S,P<sub>a</sub>)-3a<sup>++</sup>**

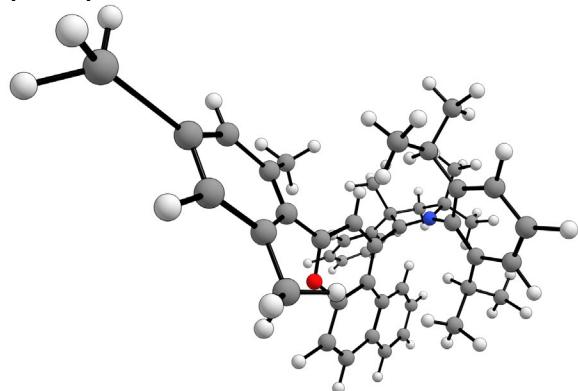


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| C | 5.913831659  | 15.899272640 | 10.677054208 |
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| H | 8.780261020  | 15.007776085 | 17.957267748 |
| C | 6.769179395  | 16.378215907 | 9.699909189  |
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| H | 10.715832996 | 16.525964826 | 10.137908394 |
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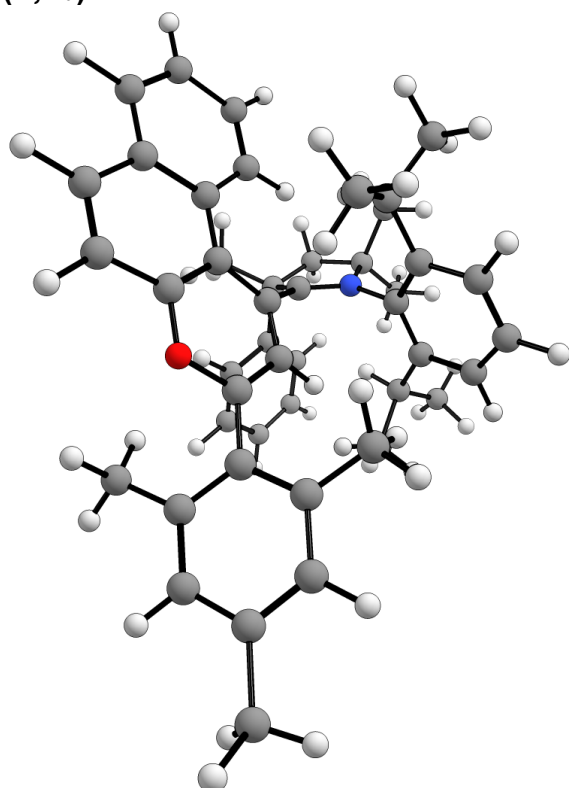
(S,M<sub>a</sub>)-3b<sup>++</sup>



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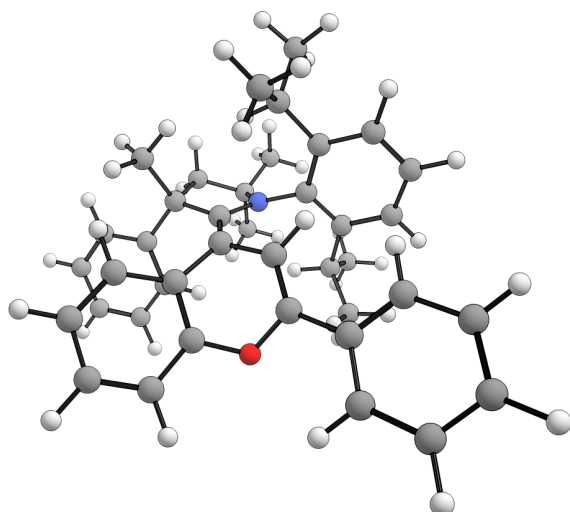
**(S,P<sub>a</sub>)-3b<sup>++</sup>**



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| H | 6.693091858  | 4.425688067 | 3.428146147  |
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| H | 4.811180597  | 3.283642859 | 8.701975520  |
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| H | 8.556177328  | 4.625019508 | 11.243587417 |
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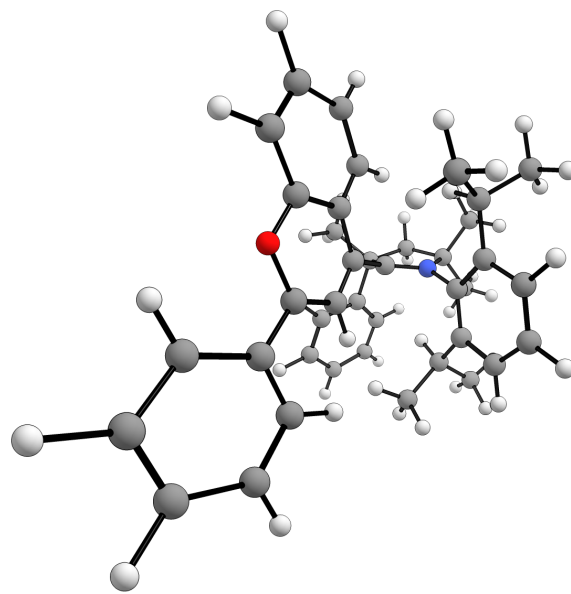
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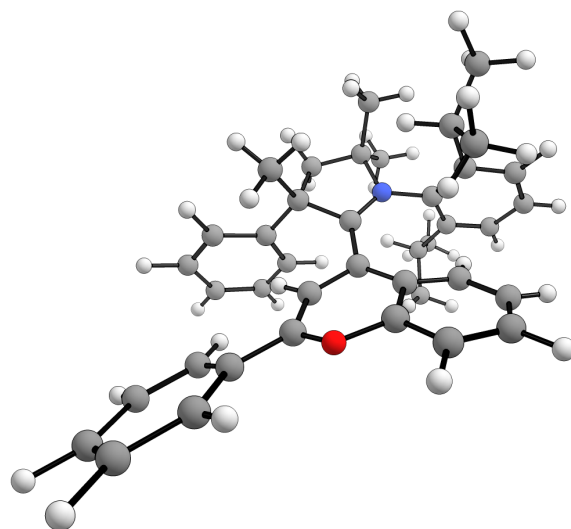
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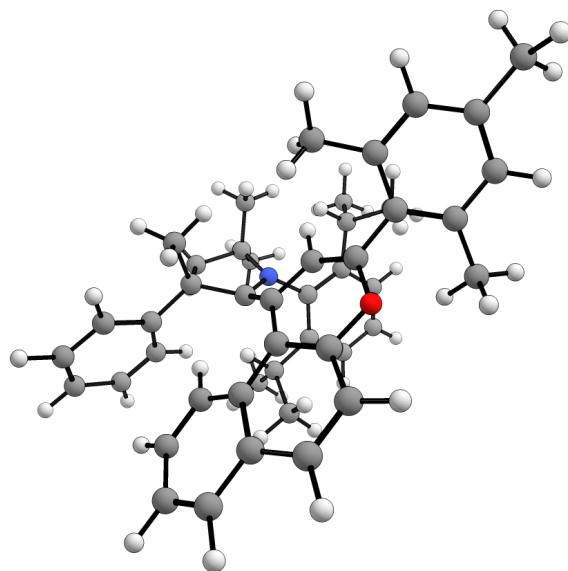
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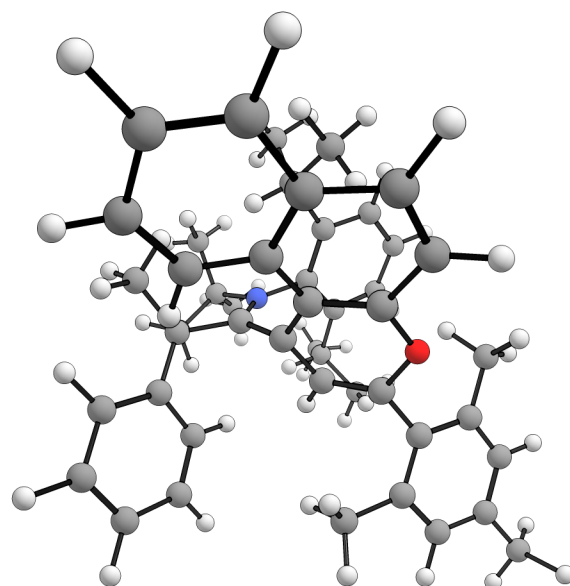
TS-3b



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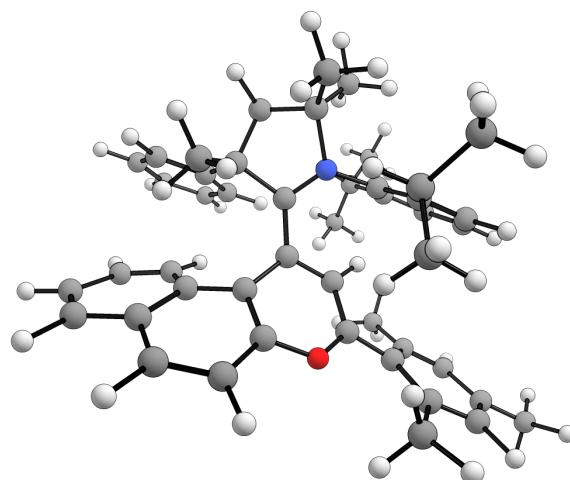
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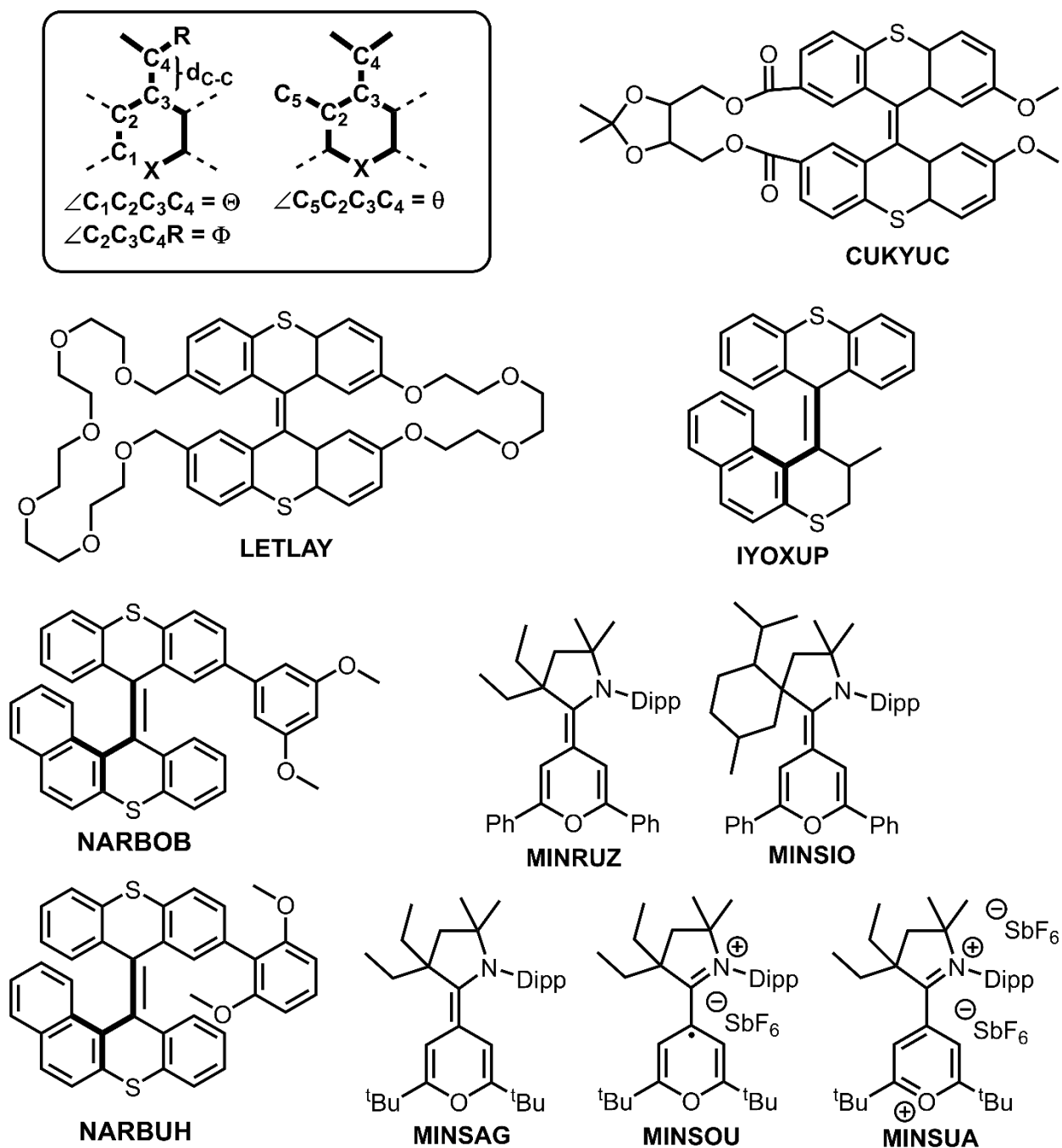
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| C | -9.259257073  | -6.940199442 | -5.116647703 |
| C | -10.583884929 | -6.874681865 | -4.304485360 |
| H | -11.329418328 | -6.325369449 | -4.886665815 |
| H | -10.975337003 | -7.874764417 | -4.110253307 |



|   |               |              |               |
|---|---------------|--------------|---------------|
| C | -10.288513531 | -6.113136480 | -3.030141856  |
| C | -8.428967374  | -8.208260849 | -4.774188366  |
| H | -7.822595219  | -8.528697402 | -5.620140279  |
| H | -7.760548780  | -8.041189335 | -3.925438782  |
| H | -9.106984133  | -9.022865781 | -4.510150342  |
| C | -9.724704351  | -6.887859685 | -6.567004966  |
| C | -10.029346871 | -8.073834320 | -7.233472173  |
| H | -9.819329067  | -9.034342602 | -6.773503464  |
| C | -10.597209495 | -8.040072662 | -8.504475414  |
| H | -10.824264081 | -8.972949899 | -9.012713650  |
| C | -10.865467062 | -6.824028736 | -9.125565556  |
| H | -11.298478046 | -6.801035562 | -10.121302939 |
| C | -10.583264372 | -5.635557962 | -8.455161547  |
| H | -10.793511160 | -4.678407381 | -8.923806427  |
| C | -10.035730699 | -5.670561065 | -7.178071463  |
| H | -9.828851241  | -4.732315112 | -6.669173911  |
| C | -9.837619477  | -7.039950681 | -1.898172123  |
| H | -10.669608475 | -7.728622901 | -1.722955023  |
| H | -8.958304413  | -7.634795826 | -2.148820331  |
| H | -9.663399867  | -6.493936212 | -0.971657163  |
| C | -11.464777139 | -5.282851807 | -2.539790138  |
| H | -11.183717759 | -4.627881227 | -1.710151302  |
| H | -11.920665105 | -4.696707862 | -3.336923717  |
| H | -12.218175846 | -5.985547245 | -2.171695848  |
| C | -8.843772856  | -3.977738174 | -2.830788546  |
| C | -9.474740960  | -2.818354646 | -3.337594500  |
| C | -9.153441138  | -1.603189571 | -2.733018054  |
| H | -9.613860017  | -0.692314623 | -3.103709986  |
| C | -8.254277766  | -1.534029336 | -1.677490603  |
| H | -8.008628680  | -0.572990013 | -1.235348632  |
| C | -7.685256766  | -2.693460298 | -1.177190208  |
| H | -6.999578877  | -2.631569435 | -0.337575644  |
| C | -7.968734209  | -3.945815126 | -1.729282978  |
| C | -10.456944050 | -2.806522477 | -4.501082862  |
| H | -10.702236846 | -3.838111389 | -4.768066934  |
| C | -9.868941605  | -2.140548133 | -5.755525540  |
| H | -10.599054849 | -2.190245832 | -6.570493888  |
| H | -9.643692455  | -1.085469702 | -5.568044943  |
| H | -8.951083590  | -2.634119291 | -6.092677235  |
| C | -11.769183151 | -2.104511417 | -4.114964941  |
| H | -12.515772120 | -2.268409616 | -4.898854999  |
| H | -12.171100171 | -2.476324027 | -3.168126243  |
| H | -11.622221978 | -1.024250356 | -4.017996773  |
| C | -7.317408507  | -5.164874803 | -1.103527795  |
| H | -7.632624993  | -6.054947588 | -1.649733766  |
| C | -5.788118685  | -5.126006191 | -1.184237268  |
| H | -5.454271258  | -5.213747426 | -2.218988742  |
| H | -5.378976776  | -4.203097318 | -0.760008234  |
| H | -5.368396646  | -5.971321932 | -0.628967507  |
| C | -7.757511063  | -5.302653439 | 0.365017731   |
| H | -7.275392113  | -4.538336987 | 0.983679078   |
| H | -8.839456921  | -5.190103464 | 0.483702146   |
| H | -7.461909353  | -6.283584546 | 0.751406499   |



**Figure S45.** Select structures from the CCDC database, with CSD refcodes, for comparison with solid-state structures from this work. Values are tabulated in Table S9.

| Compound | CSD Refcode   | Exp/Calc | d <sub>c-c</sub> (Å) | Θ (°)   | θ (°)  | Φ (°)   |
|----------|---------------|----------|----------------------|---------|--------|---------|
| 3a       | ---           | Exp      | 1.374                | -146.83 | 37.9   | 179.86  |
|          | ---           | Calc     | 1.379                | -150.8  | 34.36  | 176.233 |
| 3b       | ---           | Exp      | 1.364                | 127.58  | -61.2  | 178.85  |
|          | ---           | Calc     | 1.376                | 128.45  | -61.15 | 179.08  |
| 3b+      | ---           | Exp      | 1.442                | 157     | -29.73 | 137.77  |
|          | ---           | Calc     | 1.433                | 139.5   | -46.59 | 165.88  |
| 3b++     | ---           | Exp      | 1.489                | -178.51 | -0.45  | -103.58 |
|          | ---           | Calc     | 1.484                | 177.33  | -2.81  | -100.98 |
| ---      | <b>IYOXUP</b> | ---      | 1.349                | -126.12 | 61.42  | -175.48 |
| ---      | <b>CUKYUK</b> | ---      | 1.359                | -129.53 | 47.08  | -179.38 |
|          |               |          |                      | 34.80   | -46.34 | 179.08  |
|          |               |          |                      | 131.25  | 50.57  |         |
|          |               |          |                      | -136.67 | -50.63 |         |
| ---      | <b>LETLAY</b> | ---      | 1.347                | -127.28 | 54.75  | -178.83 |
|          |               |          |                      | 127.17  | -53.58 | -173.57 |
|          |               |          |                      | -135.62 | -49.61 |         |
|          |               |          |                      | 135.54  | 48.85  |         |
| ---      | <b>NARBOB</b> | ---      | 1.354                | -126.78 | 57.29  | 176.42  |
| ---      | <b>NARBUH</b> | ---      | 1.356                | 125.85  | -59.85 | 178.59  |
| ---      | <b>MINSAG</b> | ---      | 1.373                | 173.76  | ---    | -179.95 |
|          |               |          |                      | -175.52 |        |         |
| ---      | <b>MINSOU</b> | ---      | 1.433                | 175.46  | ---    | 171.96  |
|          |               |          |                      | -173.28 |        |         |
| ---      | <b>MINSUA</b> | ---      | 1.485                | 172.21  | ---    | 91.88   |
|          |               |          |                      | -172.08 |        | -96.10  |
| ---      | <b>MINRUZ</b> | ---      | 1.378                | -179.03 | ---    | -178.16 |
|          |               |          |                      | 176.85  |        |         |
| ---      | <b>MINSIO</b> | ---      | 1.373                | 178.87  | ---    | -171.5  |
|          |               |          |                      | -179.99 |        |         |

**Table S9.** Key structural parameters for comparison of solid-state structures obtained in this work, computational models from this work, and experimental data for related molecules from the literature. Depending on molecular symmetry, multiple values are listed for a given parameter. See Figure S45 for more details.

## 10) References

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