

# **Novel Green Synthesis of Polyfunctionally Substituted Phthalazines Promoted by Visible Light, DFT Studies and Molecular Docking with Antimicrobial and Antibiofilm Potency**

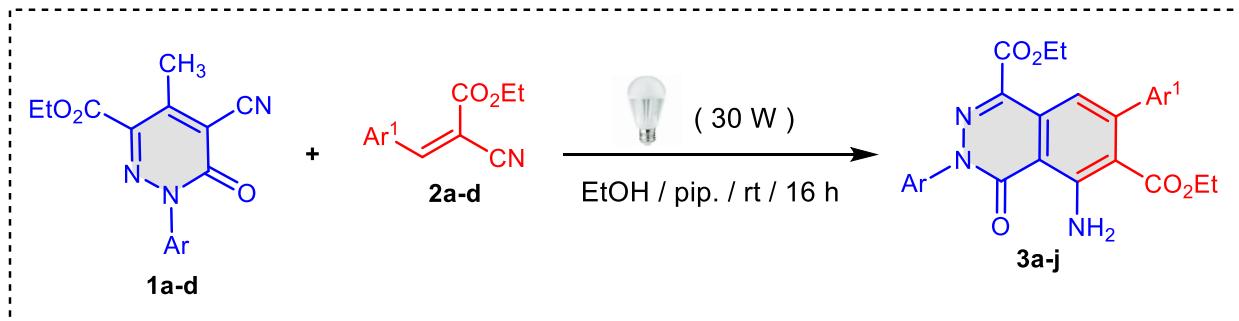
**Ramadan A. Mekheimer, Basma A. Khalifa, Zeinab Shawky Hashem, Samar M. R. Allam, Kamal Usef Sadek, Mohamed R. Eletmany**

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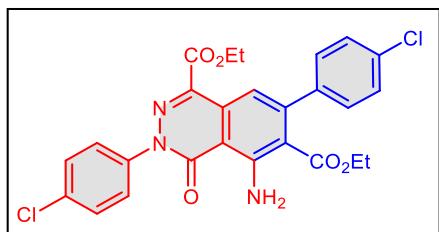
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## 1. Methodology

### 1.1. General Equation

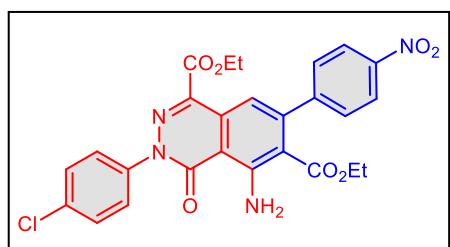


**Diethyl 5-amino-3,7-bis(4-chlorophenyl)-4-oxo-3,4-dihydrophthalazine-1,6-dicarboxylate (3a).**



Yellow crystals; yield (93%); mp: 238-240°C. IR (KBr):  $\nu$  = 3452, 3328 (NH<sub>2</sub>), 2960 (aliph. CH), 1720, 1704 (ester C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz):  $\delta_H$  = 0.76 (t, 3H,  $J$  = 7.2 Hz, CH<sub>3</sub>), 1.29 (t, 3H,  $J$  = 7.2 Hz, CH<sub>3</sub>), 4.0 (q, 2H,  $J$  = 7.2 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 7.2 Hz, CH<sub>2</sub>), 7.34 (d,  $J$  = 8.8 Hz, 2 Ar-H), 7.40 (s, 1 Ar-H), 7.56 (d,  $J$  = 8.4 Hz, 2 Ar-H), 7.60-7.66 (m, 4 Ar-H), 7.98 (br, 2H, NH<sub>2</sub>). <sup>13</sup>C NMR (100 MHz):  $\delta_C$  = 13.08 (CH<sub>3</sub>), 13.91 (CH<sub>3</sub>), 60.94 (CH<sub>2</sub>), 61.93 (CH<sub>2</sub>), 110.20 (C-6), 112.35 (C-4a), 115.25 (C-8), 128.34 (2 Ar-C), 128.48 (2 Ar-C), 128.75 (2 Ar-C), 129.38 (2 Ar-C), 129.84 (1 Ar-C), 132.73 (1 Ar-C), 132.98 (1 Ar-C), 136.61 (C-1), 139.68 (C-7), 139.78 (1 Ar-C), 146.74 (C-8a), 149.90 (C-5), 160.34 (C-4), 162.58 (C=O), 166.87 (C=O). MS: (EI) *m/z* %: 526 (M<sup>+</sup>, 100), 451 (9), 372 (14), 302 (5), 282 (32), 256 (21), 199 (14), 111 (45). Anal. Calcd. for C<sub>26</sub>H<sub>21</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>5</sub> (526.37): C, 59.33; H, 4.02; Cl, 13.47; N, 7.98%; Found: C, 59.44; H, 4.17; Cl, 13.37; N, 8.05%.

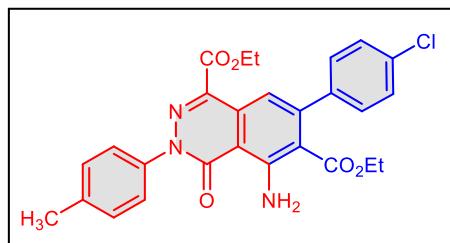
**Diethyl 5-amino-3-(4-chlorophenyl)-7-(4-nitrophenyl)-4-oxo-3,4-dihydrophthalazine-1,6-dicarboxylate (3b).**



Pale yellow crystals; yield (92%); mp: 225-227°C. IR (KBr):  $\nu$  = 3456, 3324 (NH<sub>2</sub>), 3108 (arom. CH), 2984 (aliph. CH), 1728, 1701 (ester C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.76 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.29 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 3.97 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.43 (s, 1 Ar-H), 7.60-7.66 (m, 6 Ar-H), 8.16 (br, 2H, NH<sub>2</sub>), 8.34 (d,  $J$  = 8.4 Hz, 2 Ar-H).

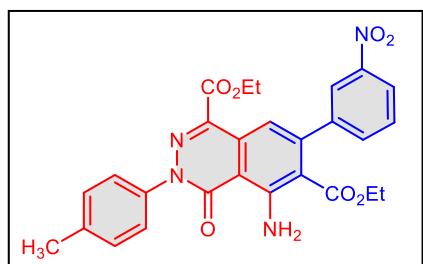
<sup>13</sup>C NMR (150 MHz):  $\delta_C$  = 13.51 (CH<sub>3</sub>), 14.41 (CH<sub>3</sub>), 61.57 (CH<sub>2</sub>), 62.47 (CH<sub>2</sub>), 111.39 (C-6), 112.78 (C-4a), 114.49 (C-8), 124.08 (2 Ar-C), 128.83 (2 Ar-C), 129.27 (2 Ar-C), 129.49 (2 Ar-C), 130.65 (1 Ar-C), 133.31 (1 Ar-C), 136.86 (C-1), 140.24 (C-7), 146.78 (C-8a), 147.52 (1 Ar-C), 148.33 (1 Ar-C), 150.92 (C-5), 160.80 (C-4), 163.03 (C=O), 167.03 (C=O). Anal. Calcd. for C<sub>26</sub>H<sub>21</sub>ClN<sub>4</sub>O<sub>7</sub> (536.93): C, 58.16; H, 3.94; Cl, 6.60; N, 10.43%; Found: C, 58.03; H, 4.05; Cl, 6.50; N, 10.60%.

**Diethyl 5-amino-7-(4-chlorophenyl)-4-oxo-3-(*p*-tolyl)-3,4-dihydropthalazine-1,6-dicarboxylate**



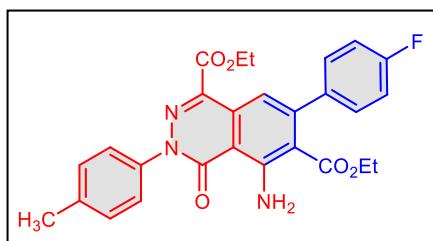
**(3c).** Yellow crystals; yield (91%); mp: 230-232°C. IR (KBr):  $\nu$  = 3422, 3304 (NH<sub>2</sub>), 3148 (arom. CH), 2982 (aliph. CH), 1707 (ester C=O), 1645 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.82 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.28 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 2.40 (s, 3H, CH<sub>3</sub>), 3.99 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.34-7.36 (m, 4 Ar-H), 7.41 (s, 1 Ar-H), 7.46 (d, 2H,  $J$  = 6.0 Hz, Ar-H), 7.56 (d, 2H,  $J$  = 6.0 Hz, Ar-H), 8.0 (br, 2H, NH<sub>2</sub>). <sup>13</sup>C NMR (150 MHz):  $\delta_C$  = 13.58 (CH<sub>3</sub>), 14.42 (CH<sub>3</sub>), 21.21 (CH<sub>3</sub>), 61.41 (CH<sub>2</sub>), 62.36 (CH<sub>2</sub>), 110.82 (C-6), 112.75 (C-4a), 115.39 (C-8), 126.82 (2 Ar-C), 128.92 (1 Ar-C), 129.69 (2 Ar-C), 129.89 (1 Ar-C), 130.42 (1 Ar-C), 130.84 (1 Ar-C), 133.44 (1 Ar-C), 136.67 (C-1), 138.45 (1 Ar-C), 139.11 (C-7), 140.29 (C-8a), 147.19 (1 Ar-C), 150.45 (1 Ar-C), 151.17 (C-5), 163.19 (C-4), 164.23 (C=O), 167.43 (C=O). Anal. Calcd. for C<sub>27</sub>H<sub>24</sub>ClN<sub>3</sub>O<sub>5</sub> (505.96): C, 64.10; H, 4.78; Cl, 7.01; N, 8.31%; Found: C, 64.17; H, 4.91; Cl, 7.10; N, 8.25%.

**Diethyl 5-amino-7-(3-nitrophenyl)-4-oxo-3-(*p*-tolyl)-3,4-dihydropthalazine-1,6-dicarboxylate**



**(3d).** Yellow crystals; yield (90%); mp: 238-240°C. IR (KBr):  $\nu$  = 3423, 3304 (NH<sub>2</sub>), 3149 (arom. CH), 2983 (aliph. CH), 1708 (ester C=O), 1645 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (400 MHz):  $\delta_H$  = 0.82 (t, 3H,  $J$  = 6.8 Hz, CH<sub>3</sub>), 1.29 (t, 3H,  $J$  = 7.2 Hz, CH<sub>3</sub>), 2.40 (s, 3H, CH<sub>3</sub>), 3.99 (q, 2H,  $J$  = 7.2 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 7.2 Hz, CH<sub>2</sub>), 7.34 (d, 2H,  $J$  = 8.8 Hz, Ar-H), 7.40 (s, 1H, Ar-H), 7.56 (d, 2H,  $J$  = 8.8 Hz, Ar-H), 7.60-7.65 (m, 4 Ar-H), 7.97 (s, 2H, NH<sub>2</sub>). Anal. Calcd for C<sub>27</sub>H<sub>24</sub>N<sub>4</sub>O<sub>7</sub>: C, 62.79; H, 4.68; N, 10.85%; Found: C, 62.91; H, 4.51; N, 11.00%.

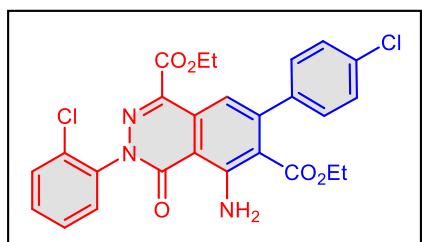
**Diethyl 5-amino-7-(4-fluorophenyl)-4-oxo-3-(p-tolyl)-3,4-dihydrophthalazine-1,6-dicarboxylate**



**(3e).** Yellow crystals; yield (90%); mp: 218-220°C. IR (KBr):  $\nu$  = 3421, 3304 (NH<sub>2</sub>), 3148 (arom. CH), 2982 (aliph. CH), 1708 (ester C=O), 1644 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.81 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.28 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 2.40 (s, 3H, CH<sub>3</sub>), 3.99 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0

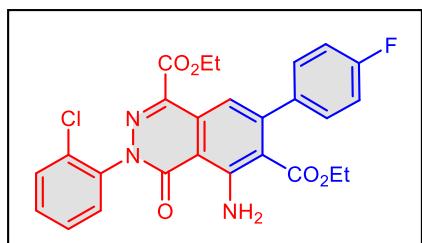
Hz, CH<sub>2</sub>), 7.34-7.36 (m, 4 Ar-H), 7.41 (s, 1 Ar-H), 7.46 (d, 2H,  $J$  = 6.0 Hz, Ar-H), 7.56 (d, 2H,  $J$  = 6.0 Hz, Ar-H), 8.0 (br, 2H, NH<sub>2</sub>). Anal. Calcd. for C<sub>27</sub>H<sub>24</sub>FN<sub>3</sub>O<sub>5</sub> (489.50): C, 66.25; H, 4.94; F, 3.88; N, 8.58%; Found: C, 66.19; H, 5.02; F, 3.77; N, 8.70%.

**Diethyl 5-amino-3-(2-chlorophenyl)-7-(4-chlorophenyl)-4-oxo-3,4-dihydrophthalazine-1,6-dicarboxylate (3f).**



**(3f).** Yellow crystals; yield (92%); mp: 235-237°C. IR (KBr):  $\nu$  = 3422, 3305 (NH<sub>2</sub>), 3150 (arom. CH), 2982 (aliph. CH), 1708 (ester C=O), 1644 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.83 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.28 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 4.01 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.37 (d, 2H,  $J$  = 8.0 Hz, Ar-H), 7.42 (s, 1 Ar-H), 7.46-7.58 (m, 4 Ar-H), 7.72 (s, 2 Ar-H), 7.93 (br, 2H, NH<sub>2</sub>). Anal. Calcd. for C<sub>26</sub>H<sub>21</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>5</sub> (526.37): C, 59.33; H, 4.02; Cl, 13.47; N, 7.98%; Found: C, 59.25; H, 4.14; Cl, 13.60; N, 7.92%.

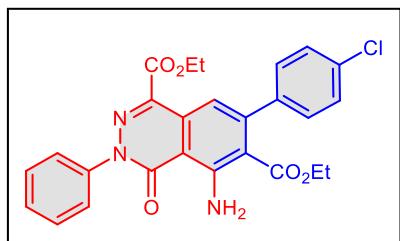
**Diethyl 5-amino-3-(2-chlorophenyl)-7-(4-fluorophenyl)-4-oxo-3,4-dihydrophthalazine-1,6-dicarboxylate (3g).**



**(3g).** Yellow crystals; yield (90%); mp: 248-250°C. IR (KBr):  $\nu$  = 3454, 3332 (NH<sub>2</sub>), 2997, 2902 (aliph. CH), 1722 (ester C=O), 1708 (ester C=O), 1682 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.77 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.27 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 3.98 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.35 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.46 (s, 1 Ar-H), 7.58-7.60 (m, 2 Ar-H), 7.64 (d, 2H,  $J$  = 6.0 Hz, Ar-H), 7.72-7.74 (m, 2 Ar-H), 8.13 (br, 2H, NH<sub>2</sub>), 8.35 (d, 2H,  $J$  = 12.0 Hz, Ar-H). <sup>13</sup>C NMR (150 MHz):  $\delta_C$  = 13.53 (CH<sub>3</sub>), 14.39 (CH<sub>3</sub>), 61.56 (CH<sub>2</sub>), 62.60 (CH<sub>2</sub>), 110.92 (C-6), 113.12 (C-4a), 114.94 (C-8), 124.11 (3 Ar-C), 128.92 (1 Ar-C), 129.53 (2 Ar-C), 130.43 (1 Ar-C), 130.62 (1 Ar-C), 131.48 (1 Ar-C), 131.61 (1 Ar-C), 137.35 (C-1), 138.87 (C-7), 147.02 (1 Ar-C), 147.58 (C-8a), 148.24 (1 Ar-C), 150.77 (C-5), 160.52

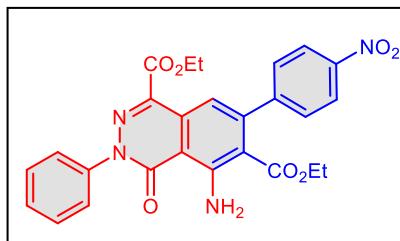
(C-4), 162.87 (C=O), 166.90 (C=O). Anal. Calcd. for  $C_{26}H_{21}ClFN_3O_5$  (509.92): C, 61.24; H, 4.15; Cl, 6.95; F, 3.73; N, 8.24%; Found: C, 61.15; H, 4.31; Cl, 6.85; F, 3.61; N, 8.17%.

**Diethyl 5-amino-7-(4-chlorophenyl)-4-oxo-3-phenyl-3,4-dihydropthalazine-1,6-dicarboxylate**



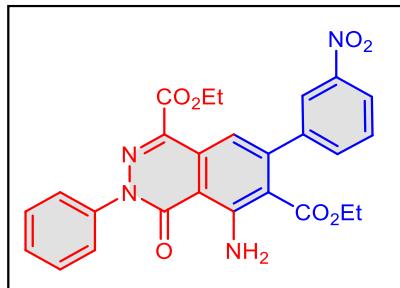
**(3h).** Yellow crystals; yield (90%); mp: 222-224°C. IR (KBr):  $\nu$  = 3459, 3334 (NH<sub>2</sub>), 2991 (aliph. CH), 1722 (ester C=O), 1705 (ester C=O), 1644 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.82 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.29 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 4.0 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.35 (d, 2H,  $J$  = 12.0 Hz, Ar-H), 7.42 (s, 1 Ar-H), 7.47-7.60 (m, 8H, 6 Ar-H and NH<sub>2</sub>), 7.65 (d, 1H,  $J$  = 6.0 Hz, Ar-H). Anal. Calcd. for  $C_{26}H_{22}ClN_3O_5$  (491.93): C, 63.48; H, 4.51; Cl, 7.21; N, 8.54%; Found: C, 63.38; H, 4.59; Cl, 7.08; N, 8.65%.

**Diethyl 5-amino-7-(4-nitrophenyl)-4-oxo-3-phenyl-3,4-dihydropthalazine-1,6-dicarboxylate**



**(3i).** Pale yellow crystals; yield (93%); mp: 226-228°C. IR (KBr):  $\nu$  = 3422, 3305 (NH<sub>2</sub>), 3150 (arom. CH), 2982 (aliph. CH), 1708 (ester, C=O), 1644 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR (600 MHz):  $\delta_H$  = 0.76 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.28 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 3.98 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.45 (s, 1 Ar-H), 7.49 (s, 1 Ar-H), 7.55-7.62 (m, 8H, 6 Ar-H and NH<sub>2</sub>), 8.35 (d, 2H,  $J$  = 6.0 Hz, 2 Ar-H). <sup>13</sup>C NMR (150 MHz):  $\delta_C$  = 13.52 (CH<sub>3</sub>), 14.42 (CH<sub>3</sub>), 61.48 (CH<sub>2</sub>), 62.43 (CH<sub>2</sub>), 111.49 (C-6), 112.72 (C-4a), 114.30 (C-8), 124.09 (2 Ar-C), 127.06 (2 Ar-C), 128.95 (1 Ar-C), 129.30 (2 Ar-C), 129.50 (2 Ar-C), 130.72 (1 Ar-C), 136.58 (C-1), 141.49 (C-7), 146.71 (C-8a), 147.51 (1 Ar-C), 148.42 (1 Ar-C), 150.97 (C-5), 160.88 (C-4), 163.12 (C=O), 167.06 (C=O). Anal. Calcd. for  $C_{26}H_{22}N_4O_7$  (502.48): C, 62.15; H, 4.41; N, 11.15%; Found: C, 62.24; H, 4.35; N, 11.29%.

**Diethyl 5-amino-7-(3-nitrophenyl)-4-oxo-3-phenyl-3,4-dihydropthalazine-1,6-dicarboxylate**



**(3j).** Yellow crystals; yield (93%); mp: 242-244°C. IR (KBr):  $\nu$  = 3423, 3317 (NH<sub>2</sub>), 3100 (arom. CH), 1742 (ester, C=O), 1727 (ester, C=O) 1662 (amide C=O) cm<sup>-1</sup>. <sup>1</sup>H NMR:  $\delta_H$  = 0.82 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 1.29 (t, 3H,  $J$  = 6.0 Hz, CH<sub>3</sub>), 3.99 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 4.36 (q, 2H,  $J$  = 6.0 Hz, CH<sub>2</sub>), 7.34-7.36 (m, 4 Ar-H), 7.4

(s, 1 Ar-H), 7.46 (m, 2 Ar-H), 7.56 (d, 2H,  $J$  = 12 Hz, Ar-H), 7.66 (s, 1 Ar-H), 8.0 (br, 2H, NH<sub>2</sub>). Anal. Calcd. for C<sub>26</sub>H<sub>22</sub>N<sub>4</sub>O<sub>7</sub> (502.48): C, 62.15; H, 4.41; N, 11.15%. Found: C, 62.08; H, 4.54; N, 11.09%.

## 1.2. Antimicrobial screening

### 1.2.1. Microorganisms, Culture Conditions, and Compound Preparation

Antimicrobial testing was performed against clinically multi-resistant isolated strains of *S. aureus*, *K. pneumoniae*, *P. aeruginosa*, and *C. albicans*, obtained from the Microbiology Laboratory, Faculty of Pharmacy, Minia University. Bacterial strains were cultured in tryptic soy broth (TSB), and *C. albicans* was cultured in Sabouraud dextrose broth (SDB). Cultures were incubated at 37°C for 24 hours and adjusted to a 0.5 McFarland standard. A series of synthesized polyfunctionally substituted phthalazine derivatives (**3a-j**) were evaluated. Each compound was dissolved in dimethyl sulfoxide (DMSO) to prepare 2 mg/mL stock solutions and subsequently filtered using 0.22 µm filters. Working concentrations were prepared by serial dilution in sterile distilled water prior to testing.

### 1.2.2. Antimicrobial Susceptibility Testing

The agar well diffusion method was performed to assess the antimicrobial activity. Mueller-Hinton agar (MHA) plates for bacteria and Sabouraud dextrose agar (SDA) plates for fungi were inoculated with standardized microbial inocula. Six millimetres wells were punched and filled with 20 µL of each compound. Ciprofloxacin (20 µg/mL) and fluconazole (20 µg/mL) served as reference antibiotics for bacterial and fungal strains, respectively, while DMSO was used as a negative control. The plates were incubated overnight at 37°C, and inhibition zone diameters were measured.

### 1.2.3. Determination of Minimum Inhibitory Concentration (MIC) and Minimum Bactericidal/Fungicidal Concentration (MBC/MFC)

The Minimum Inhibitory Concentration (MIC) values were determined using the broth microdilution method. Two hundred microliters of each phthalazine derivative solutions were added to the first column of a 96-well microtiter plate, followed by serial two-fold dilutions across the row, resulting in final concentrations ranging from 200 to 0.82 µg/mL. Each well received 10 µL of bacterial or fungal inoculum suspension. Columns 11 and 12 served as microbial growth and broth sterility controls respectively. Plates were incubated at 37°C for 24 hours. The MIC was

defined as the lowest concentration of an antimicrobial agent that completely inhibits the visible growth of a microorganism. All data were reported as the mean of three independent experiments, each performed in duplicate. The Minimum Bactericidal and Fungicidal Concentrations (MBC and MFC) were determined polystyrene microplates by streaking samples from each MIC well onto agar plates. The lowest concentration at which no visible microbial growth was observed after incubation, indicating bactericidal or fungicidal activity.

#### **1.2.4. Biofilm Inhibition Assay**

The antibiofilm activity of the phthalazine compounds was investigated using the crystal violet staining method in 96-well flat-bottom [1]. Each well was loaded with 180  $\mu$ L of TSB supplemented with 1% glucose, 10  $\mu$ L of microbial inoculum, and 10  $\mu$ L of test compound. Negative control wells received an equivalent volume of DMSO. Following static incubation at 37°C for 24 hours, non-adherent cells were removed by washing with phosphate-buffered saline (PBS, pH 7.2). Plates were air-dried and biofilms were fixed with 99% methanol for 15 minutes. Wells were then stained with 0.5% crystal violet for 30 minutes, washed with distilled water to remove excess stain, and air-dried. The bound dye was solubilized using 95% ethanol and absorbance was measured at 570 nm using a microplate reader.

The biofilm inhibition percentage was calculated using the following formula:

$$\% \text{ Inhibition} = [(OD \text{ control} - OD \text{ treated}) / OD \text{ control}] \times 100$$

All experiments were performed in triplicate and repeated independently three times.

### **1.3. Computational Methodology**

#### **1.3.1. DFT Strategy**

Calculations were carried out using the 6-311G(d,p) basis set within the Gaussian 09 software suite [2]. Molecular electrostatic potential (MEP) mapping was performed to pinpoint key nucleophilic and electrophilic regions in the optimized structure. The most stable conformer and its electronic excitation properties were visualized using Chemcraft [3] and VMD [4] to elucidate its electronic characteristics. Harmonic vibrational frequencies computed at the 6-311G(d,p) level were scaled using a factor of 0.967, as recommended by NIST (<https://cccbdb.nist.gov/vibscalejustx.asp>) [5], to correct for anharmonic effects.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were computed using the Gauge-Including Atomic Orbital (GIAO) method within the DFT framework [6]. Time-Dependent Density Functional Theory (TD-DFT), a widely adopted

approach for simulating UV/vis absorption spectra and characterizing electronic excitation states in various molecular systems [7], was employed in this study. The TD-DFT calculations were conducted using the Conductor-like Polarizable Continuum Model (CPCM) to simulate solvent effects in DMSO.

Further topological analyses were executed using Multiwfn software [8], including reduced density gradient (RDG) and non-covalent interaction (NCI) analyses, alongside the electron localization function (ELF), to thoroughly characterize the intramolecular interactions and bonding nature within the heterocyclic framework [9].

The quantum chemical reactivity parameters for all the designed systems were computed following geometric optimization. These parameters were evaluated using the HOMO and LUMO energy values, based on the equations provided as the following [10,11].

$$\text{Energy Gap } (\Delta E) = E_{\text{LUMO}} - E_{\text{HOMO}} \quad (1)$$

$$\text{Ionization potential } (I) = - E_{\text{HOMO}} \quad (2)$$

$$\text{Electron affinity } (A) = - E_{\text{LUMO}} \quad (3)$$

$$\text{Hardness } (\eta) = (I - A)/2 \quad (4)$$

$$\text{Chemical potential } (\mu) = - (I + A)/2 \quad (5)$$

$$\text{Softness } (\sigma) = 1/\eta \quad (6)$$

$$\text{Electronegativity } (\chi) = -(E_{\text{HOMO}} + E_{\text{LUMO}})/2 \quad (7)$$

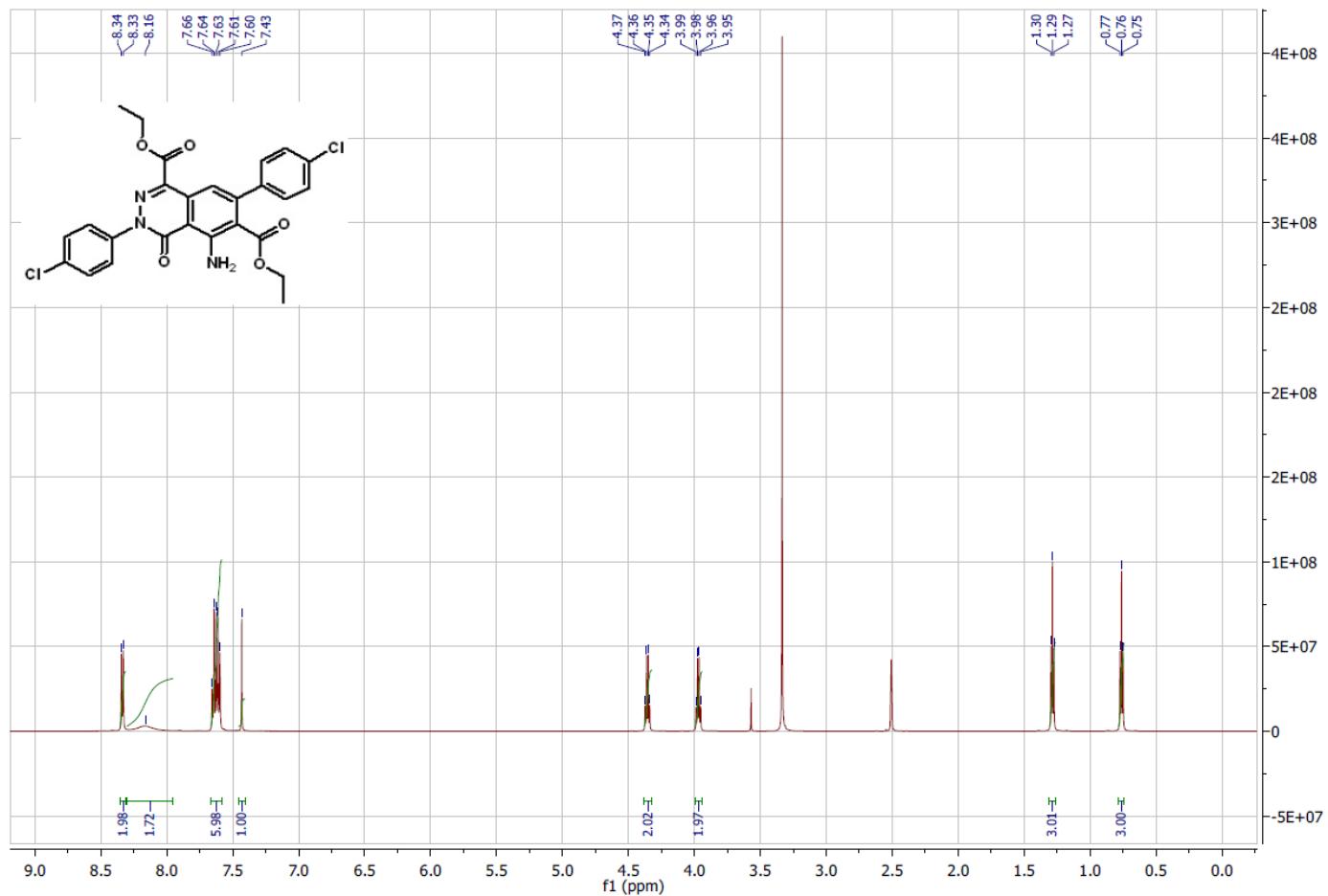
$$\text{Electrophilicity } (\omega) = \mu^2/2\eta \quad (8)$$

### 1.3.2. Molecular Docking methodology

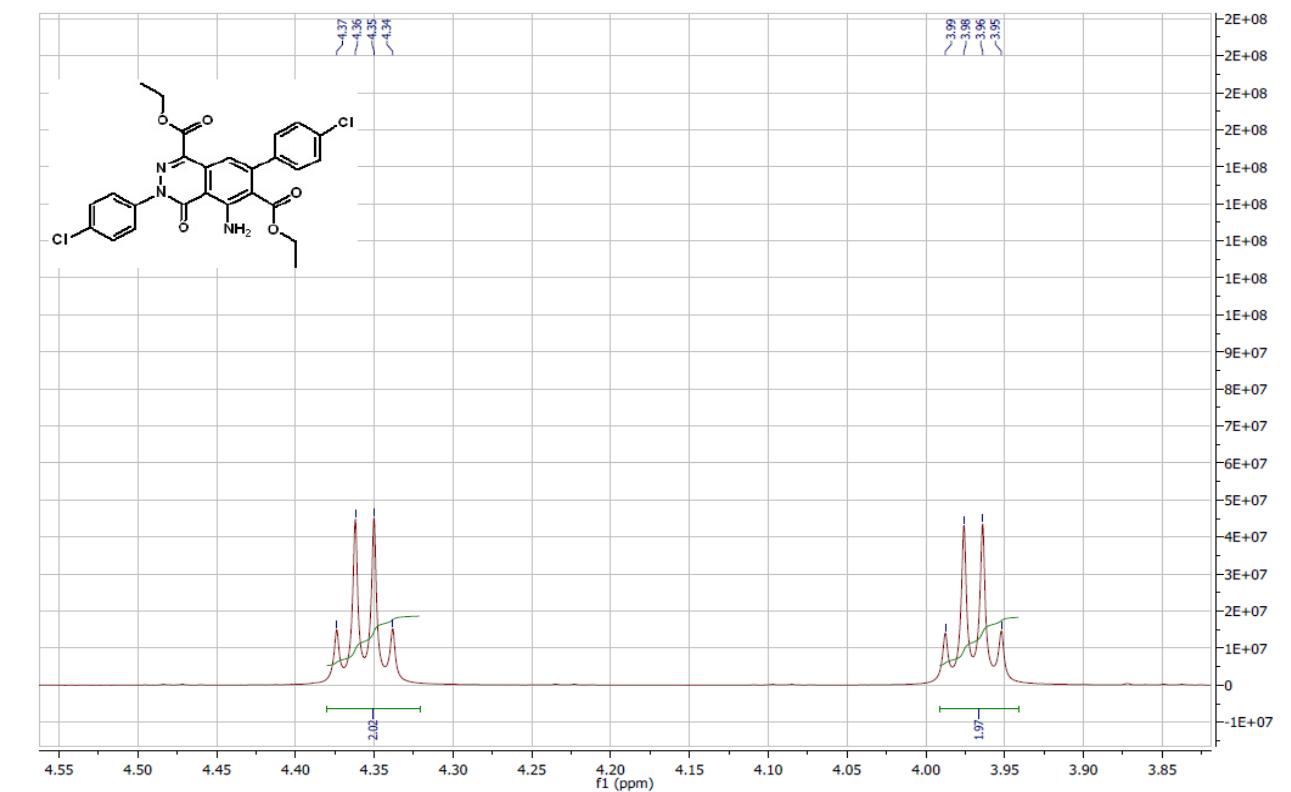
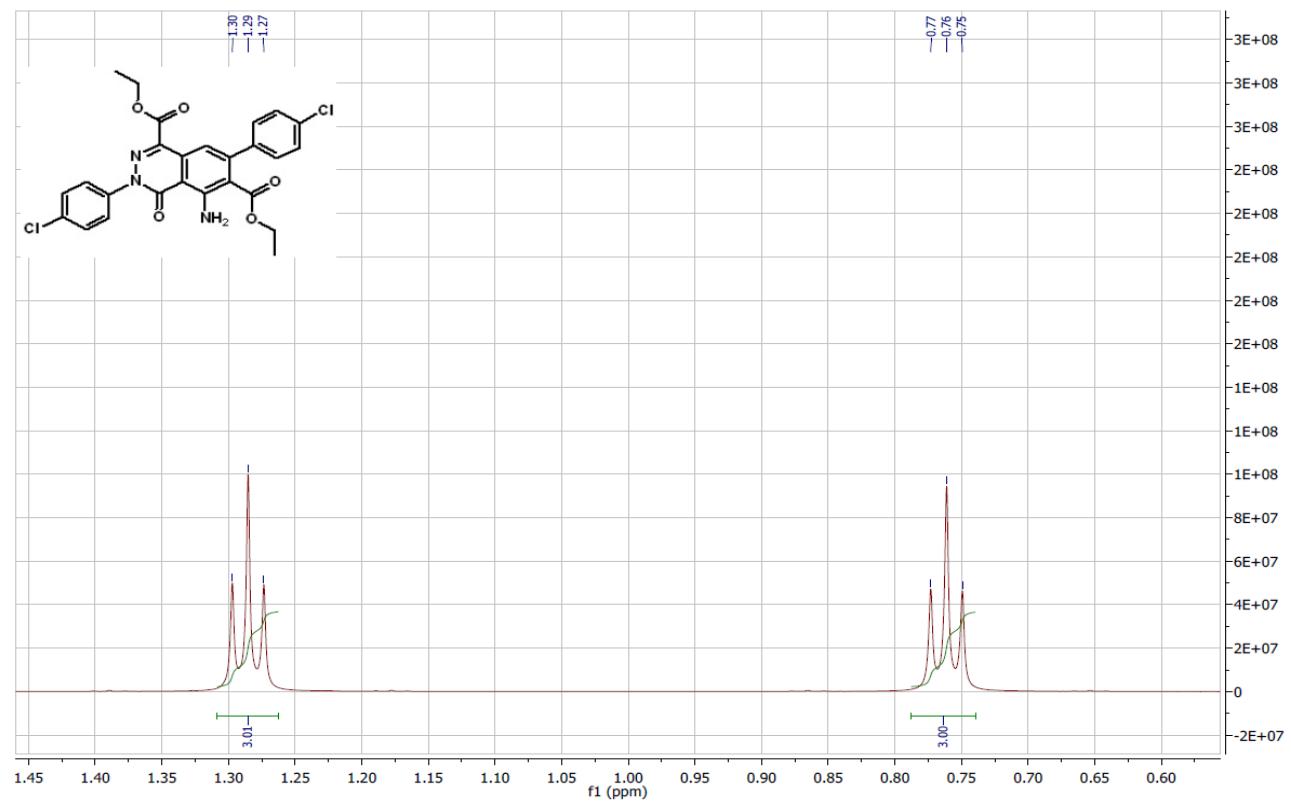
The molecular docking procedure followed a well-established protocol to ensure reliable and reproducible results [12,13]. Docking simulations for the investigated heterocyclic compounds were conducted using AutoDock Vina software [14]. Post-docking analysis and visualization of ligand–protein interactions were carried out using Discovery Studio (<https://www.3ds.com/products-services/biovia/>). The selected receptors *S. aureus* (ID:2XCT) [15] and human CYP51 (ID: 3LD6) [16] were obtained from the Protein Data Bank (<https://www.rcsb.org/>). Protein preparation involved the removal of water molecules and non-essential atoms, adding polar hydrogen atoms, and assigning partial atomic charges. The ligand and protein files were converted into the PDBQT format for docking. Active site coordinates were

determined, and a grid box was defined with dimensions of  $40 \times 40 \times 40 \text{ \AA}$  and a grid spacing of  $0.375 \text{ \AA}$ . The grid centers were set at coordinates ( $x = 21.348, y = 24.336, z = 79.243$ ) for 2XCT and ( $x = 42.348, y = -0.623, z = -1.711$ ) for 3LD6. The Genetic Algorithm (GA) was employed as the docking search method to predict optimal binding conformations [17].

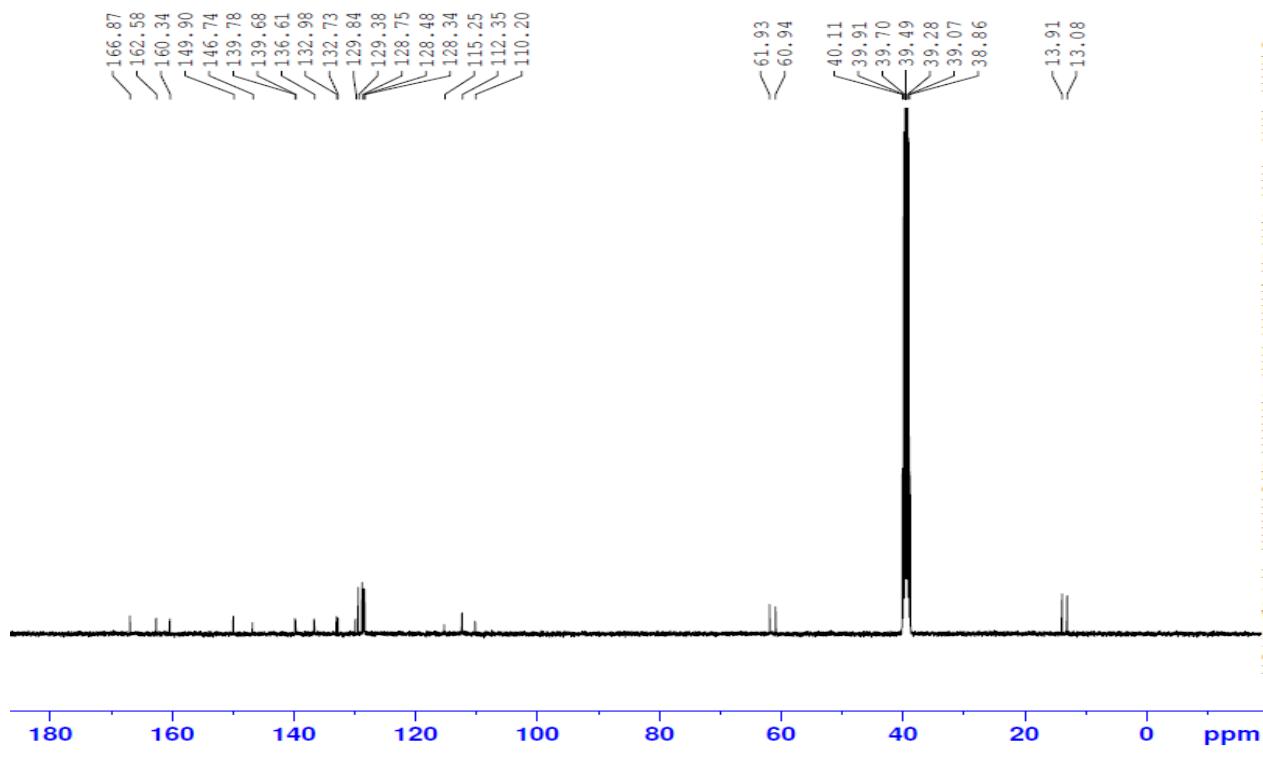
## 2. Spectral Data



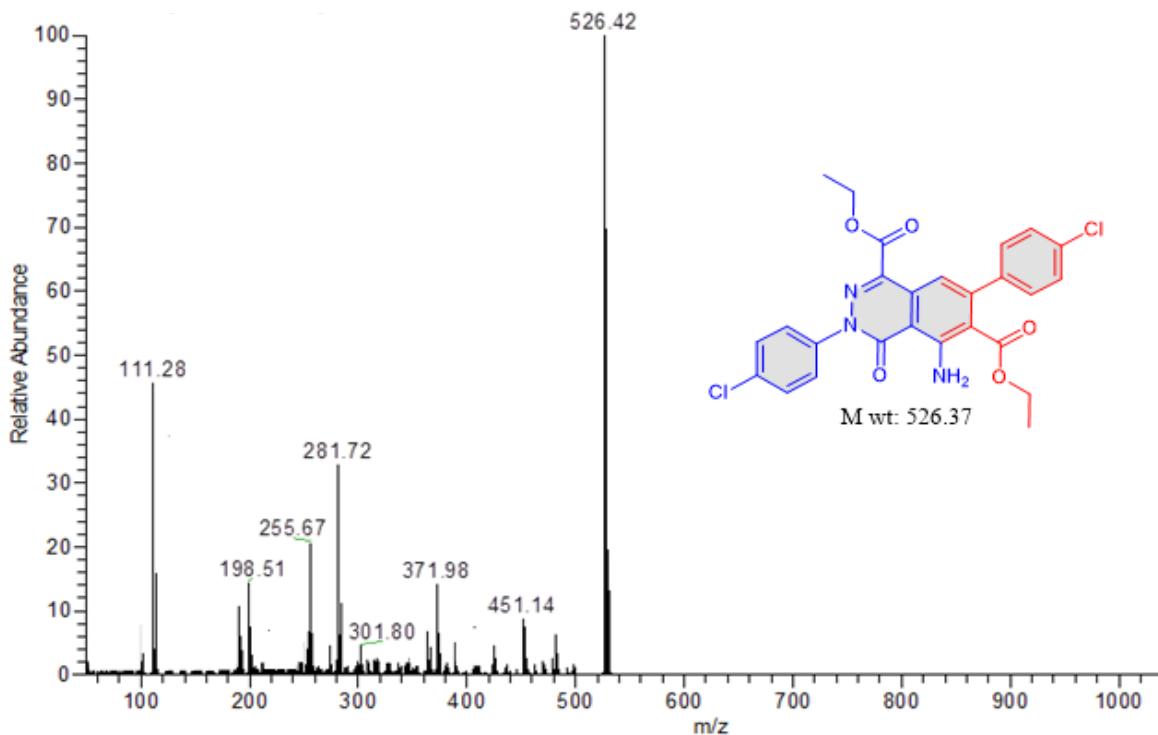
**Figure 1.**  $^1\text{H}$  NMR spectrum of compound 3a.



**Figure 2.** Expanded  $^1\text{H}$  NMR of compound 3a.



**Figure 3.**  $^{13}\text{C}$  NMR spectrum of compound **3a**.



**Figure 4.** Mass spectrum of compound **3a**.

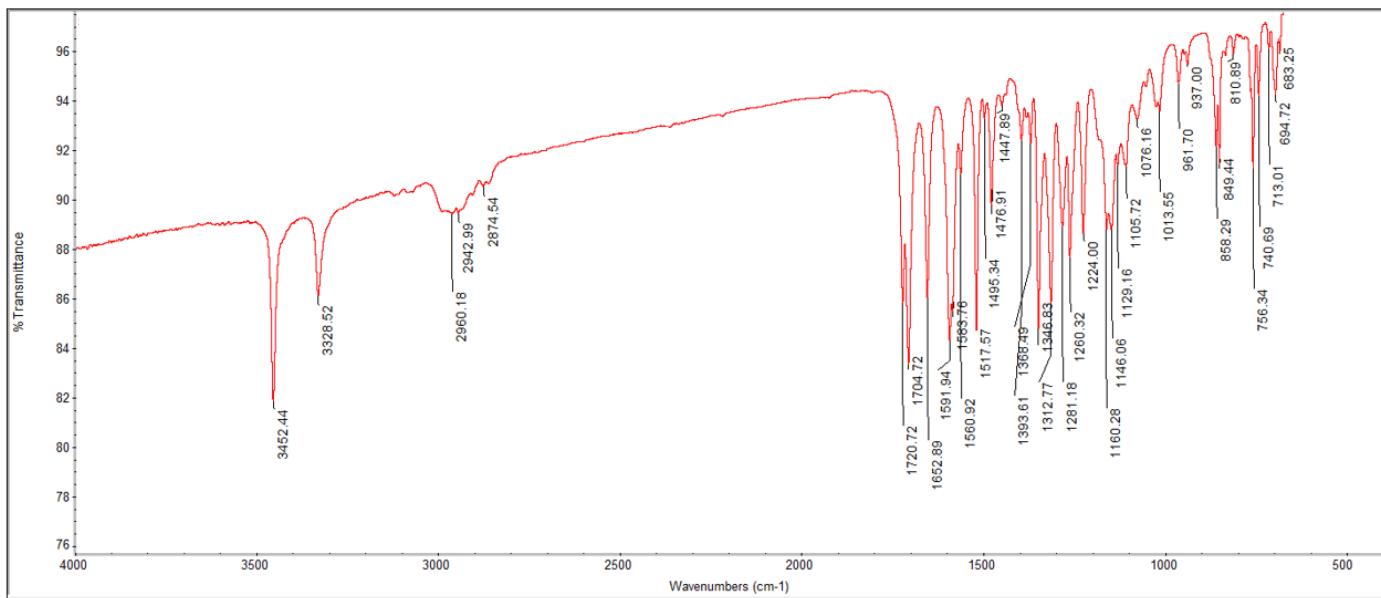


Figure 5. IR spectrum of compound 3a.

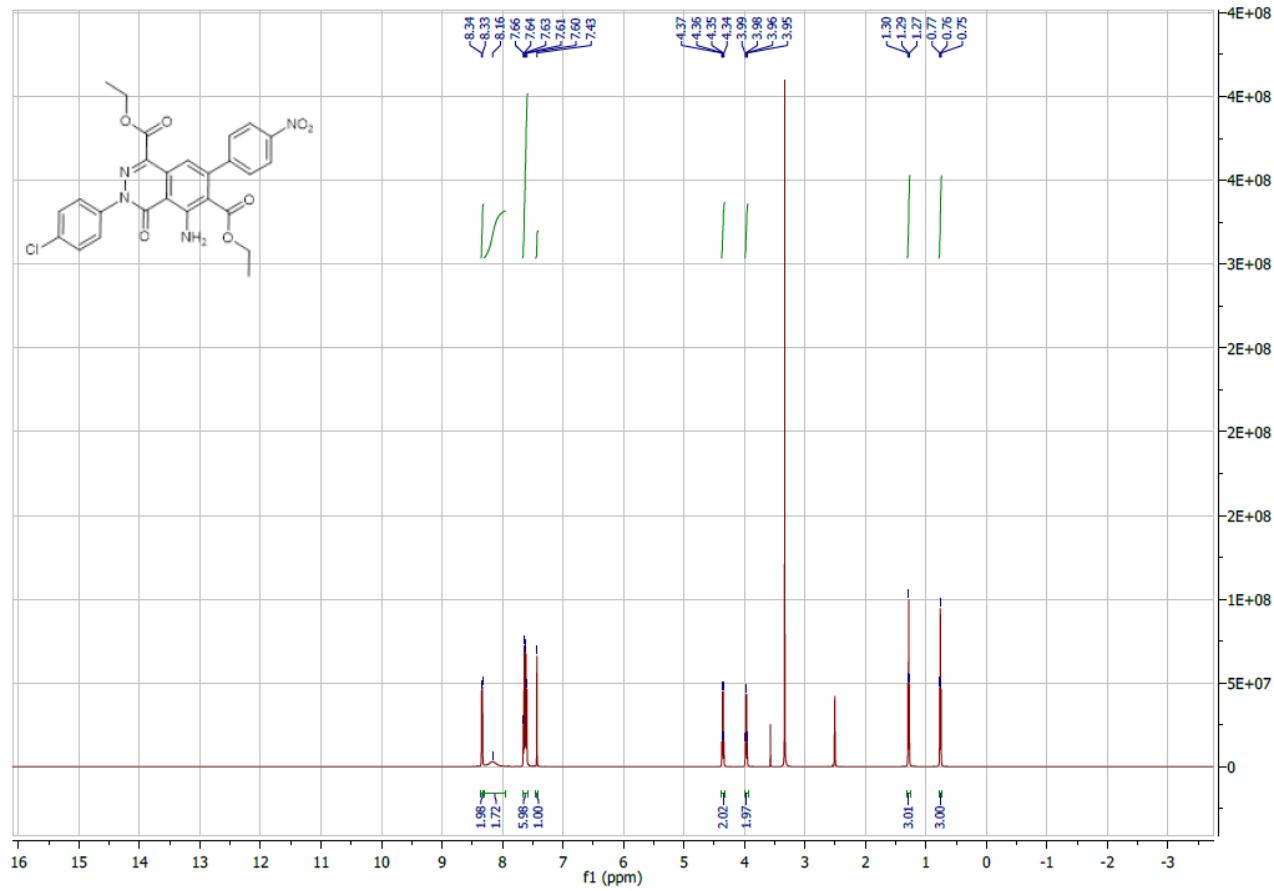
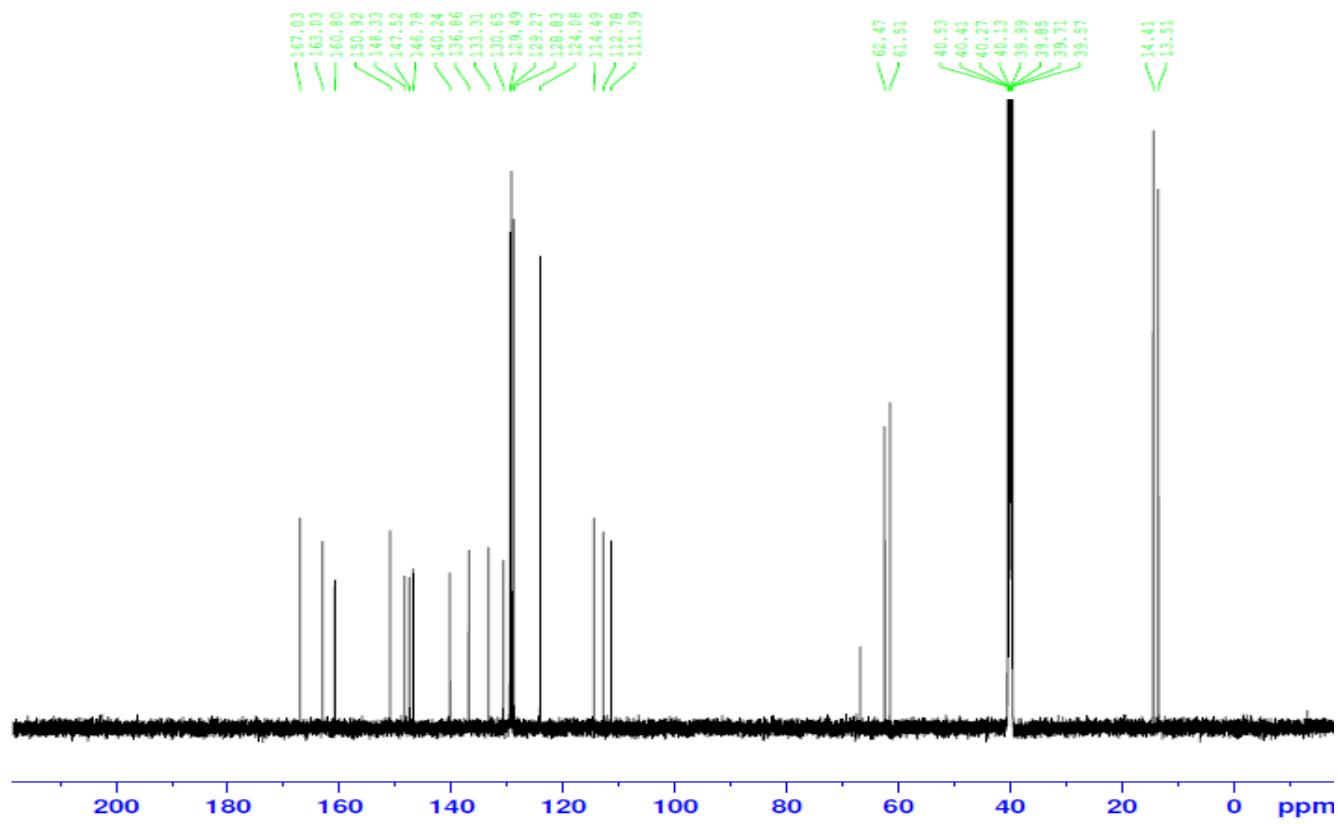
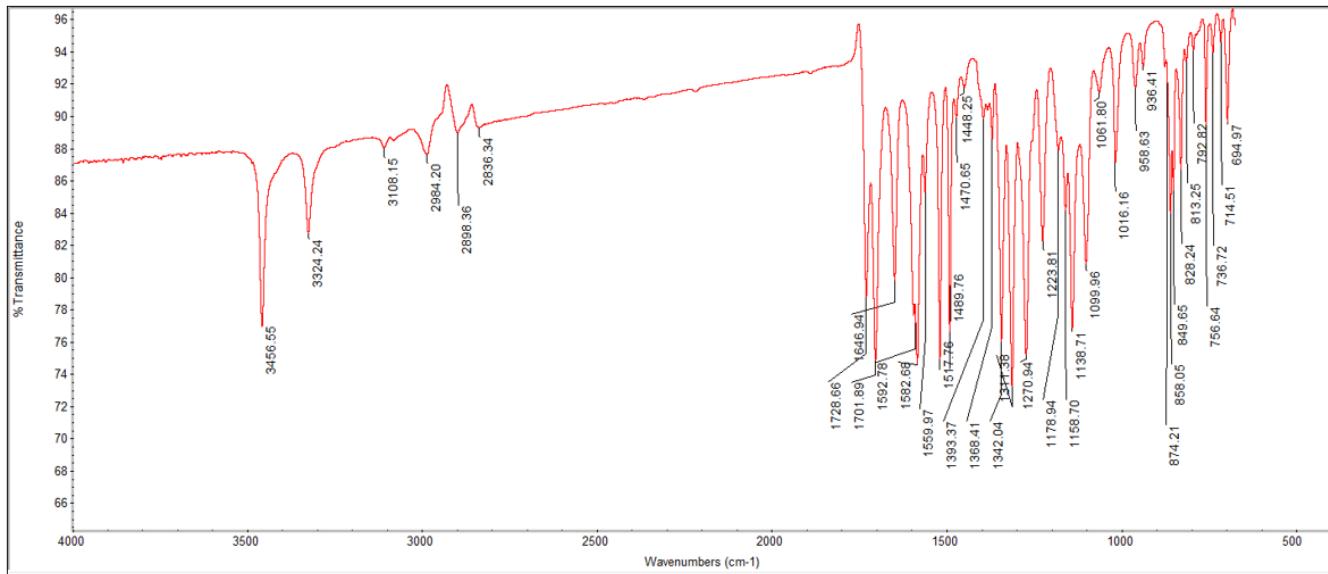


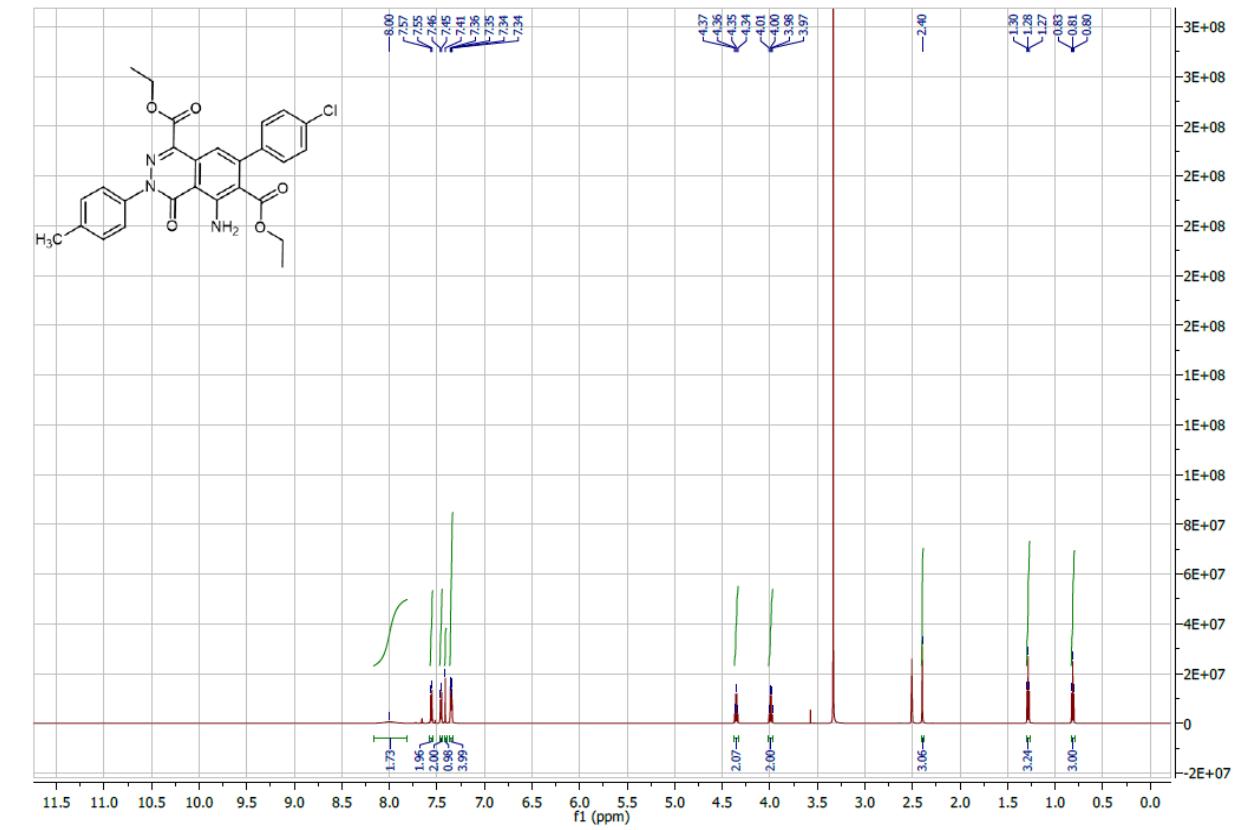
Figure 6.  $^1\text{H}$  NMR spectrum of compound 3b.



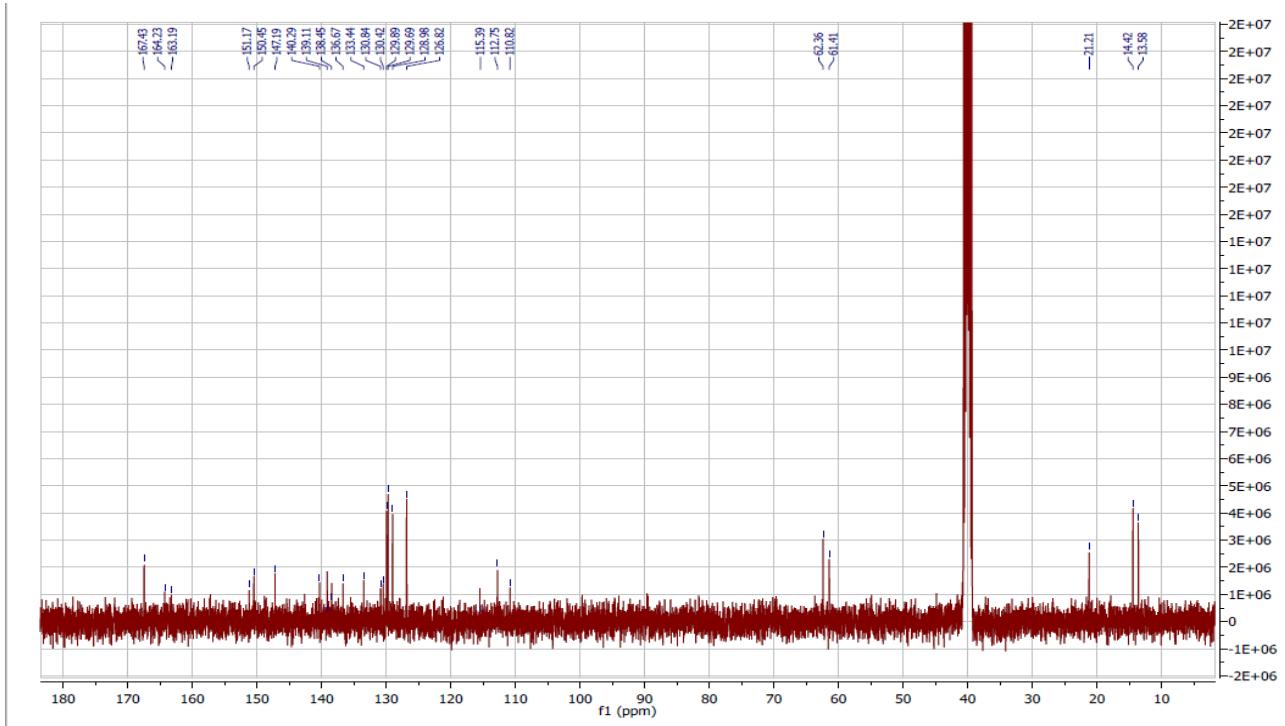
**Figure 7.**  $^{13}\text{C}$  NMR spectrum of compound **3b**.



**Figure 8.** IR spectrum of compound **3b**.



**Figure 9.**  $^1\text{H}$  NMR spectrum of compound **3c**.



**Figure 10.**  $^{13}\text{C}$  NMR spectrum of compound **3c**.

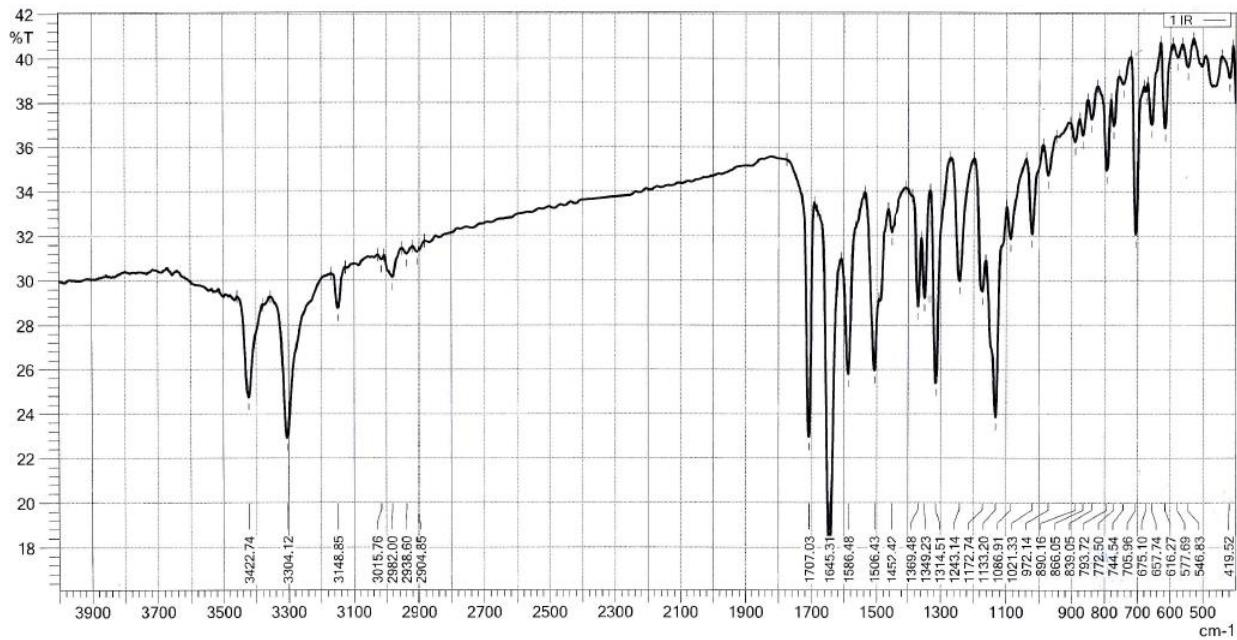


Figure 11. IR spectrum of compound 3c.

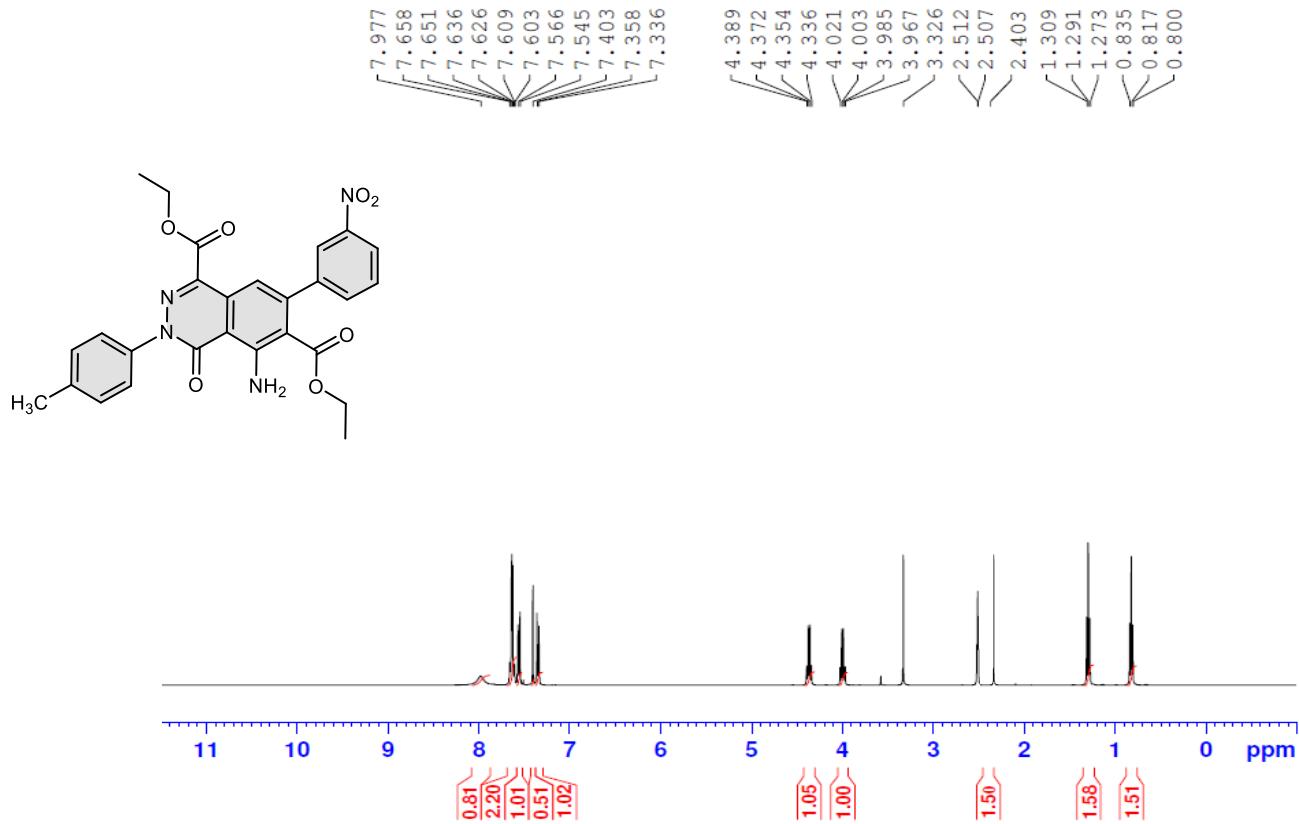
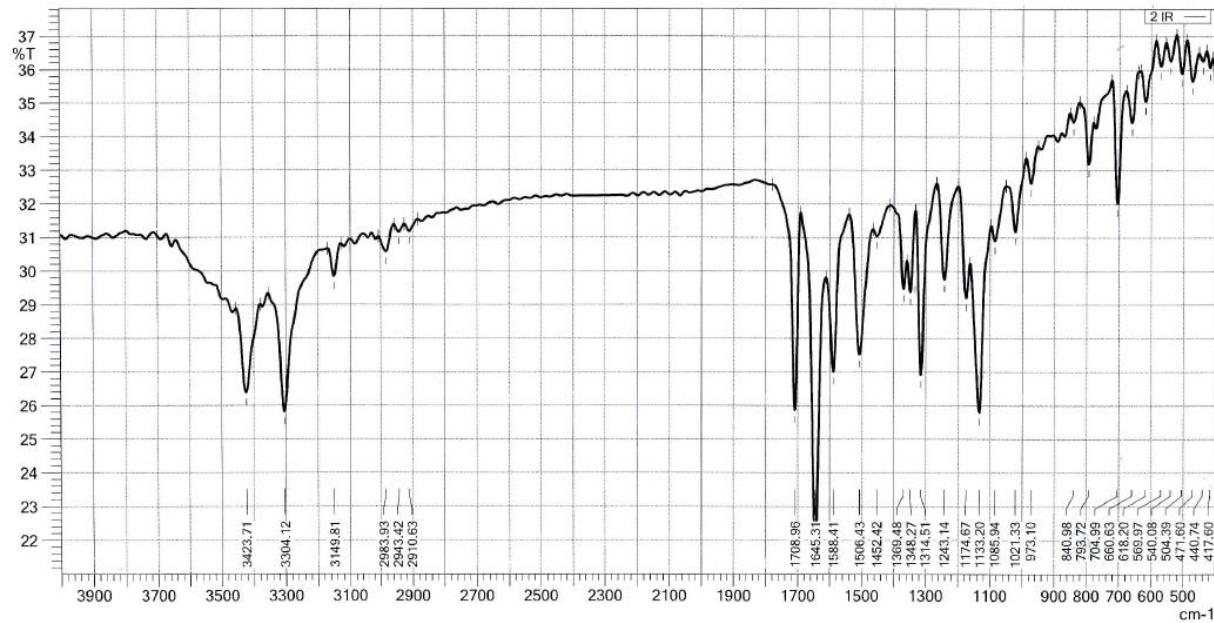
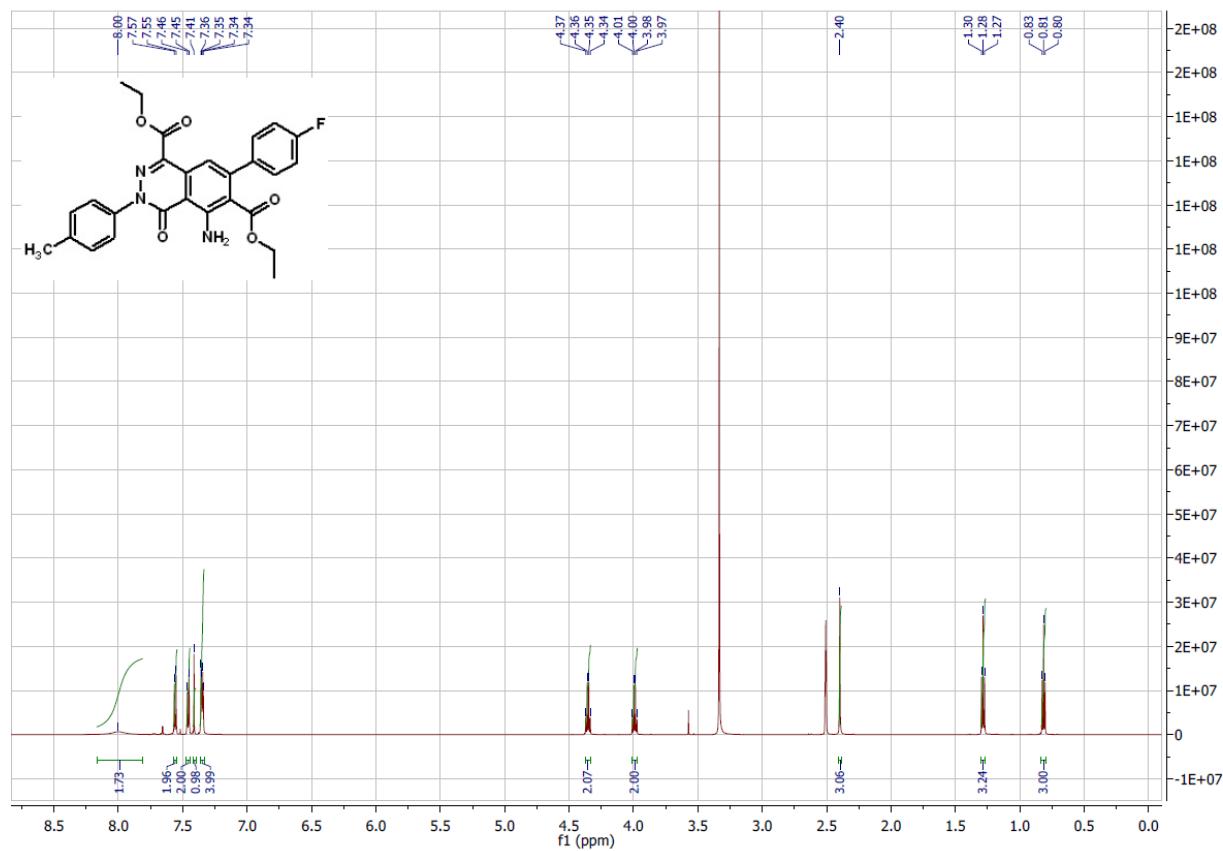


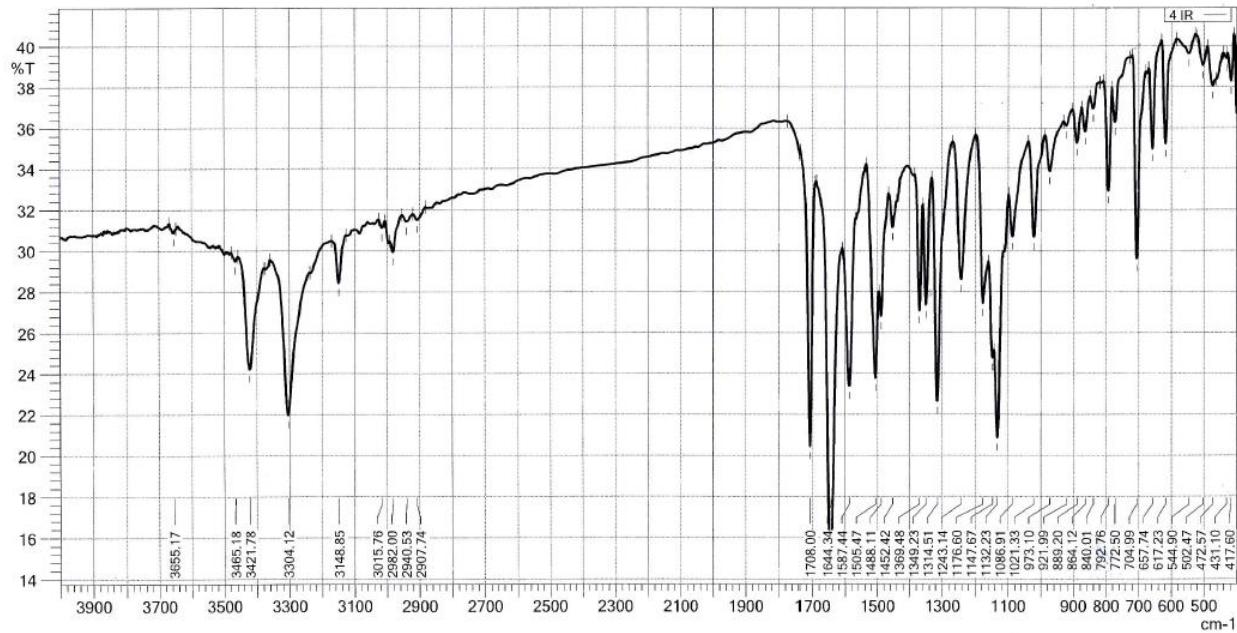
Figure 12. <sup>1</sup>H NMR spectrum of compound 3d.



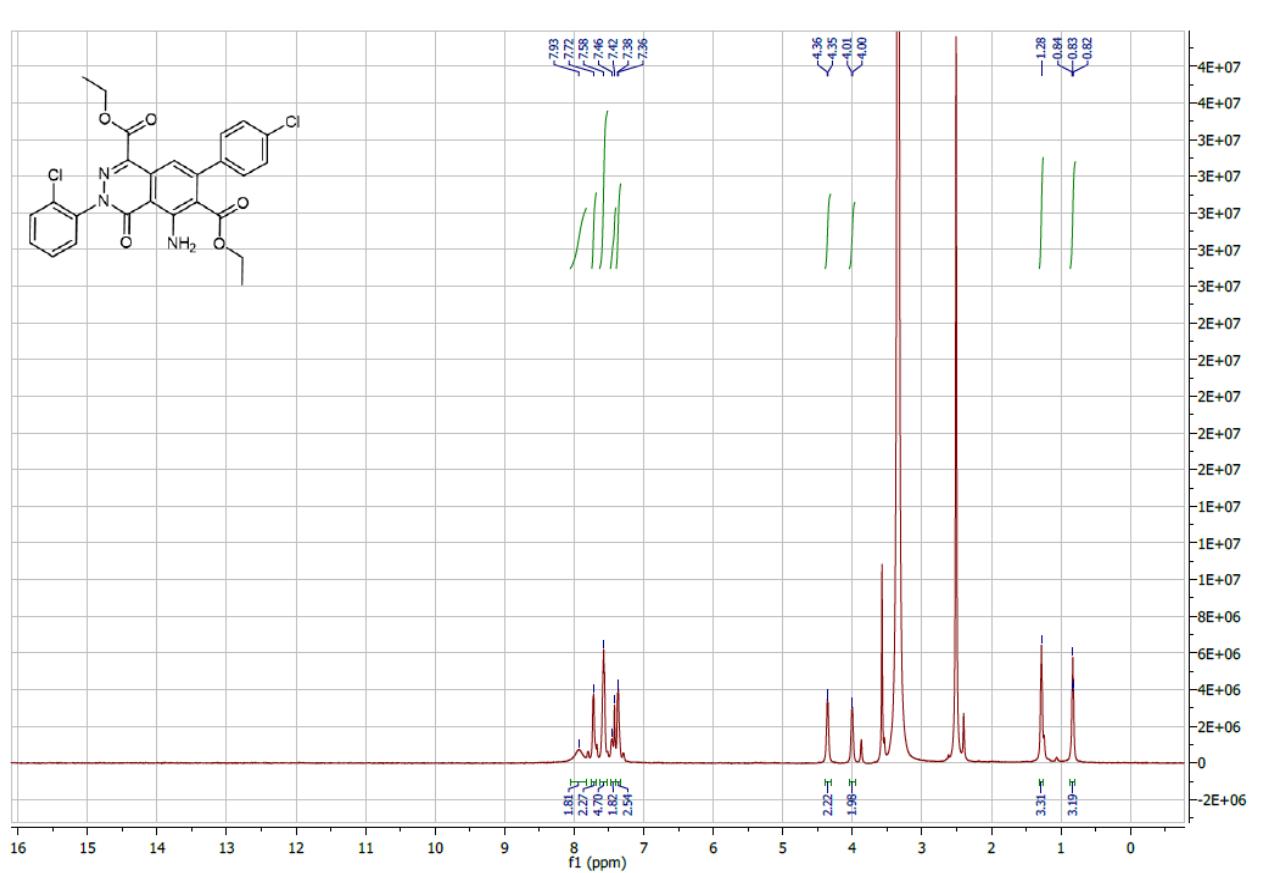
**Figure 13.** IR spectrum of compound **3d**.



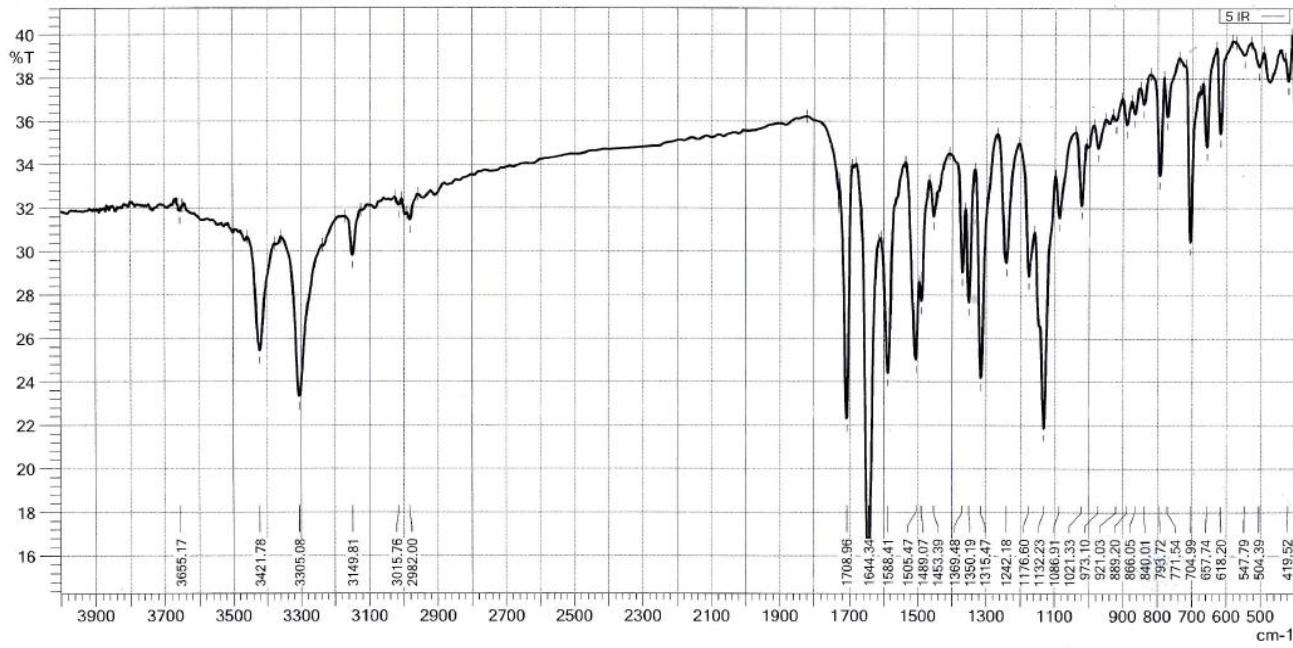
**Figure 14.**  $^1\text{H}$  NMR spectrum of compound **3e**.



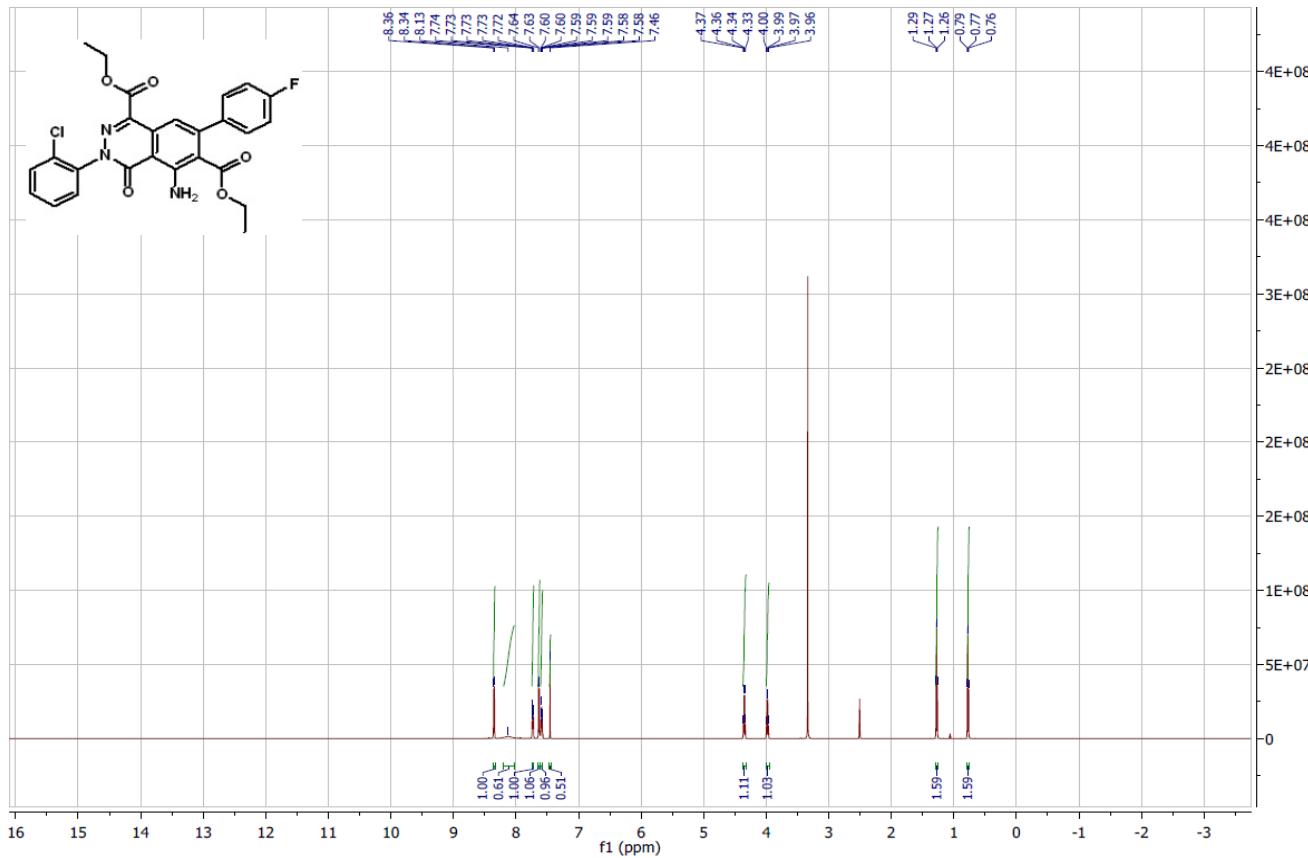
**Figure 15.** IR spectrum of compound **3e**.



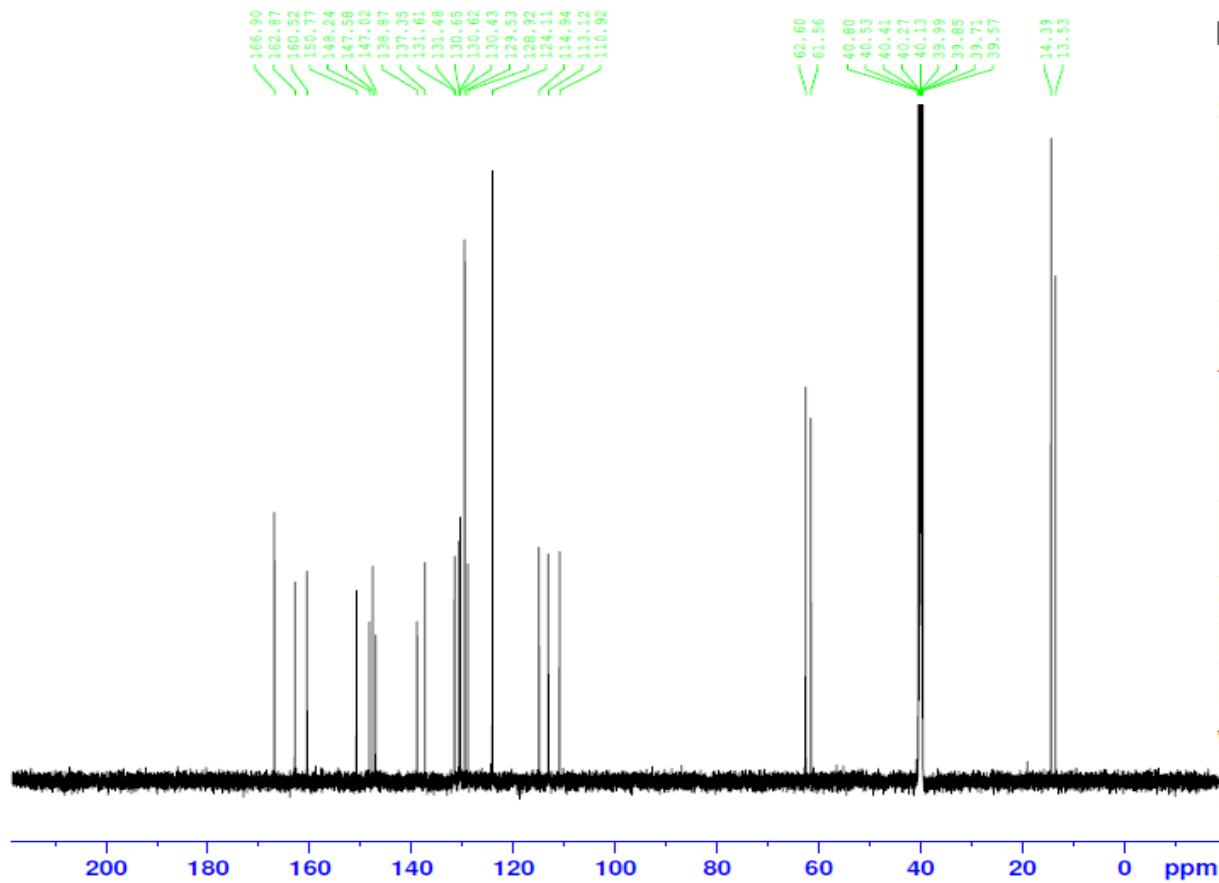
**Figure 16.** <sup>1</sup>H NMR spectrum of compound **3f**.



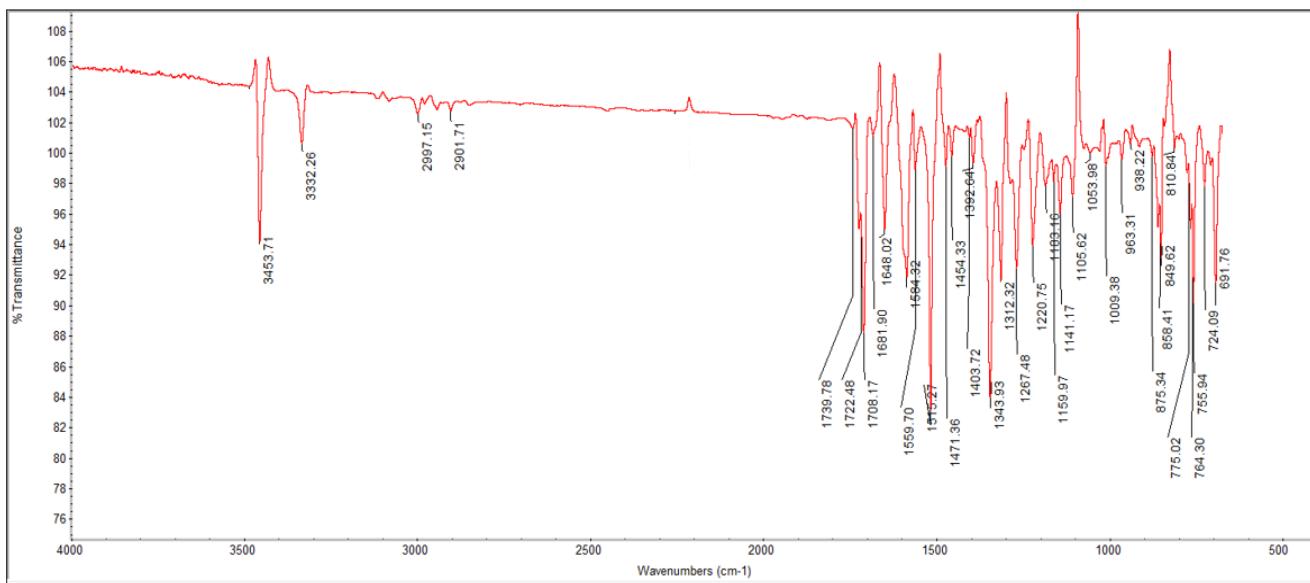
**Figure 17.** IR spectrum of compound **3f**.



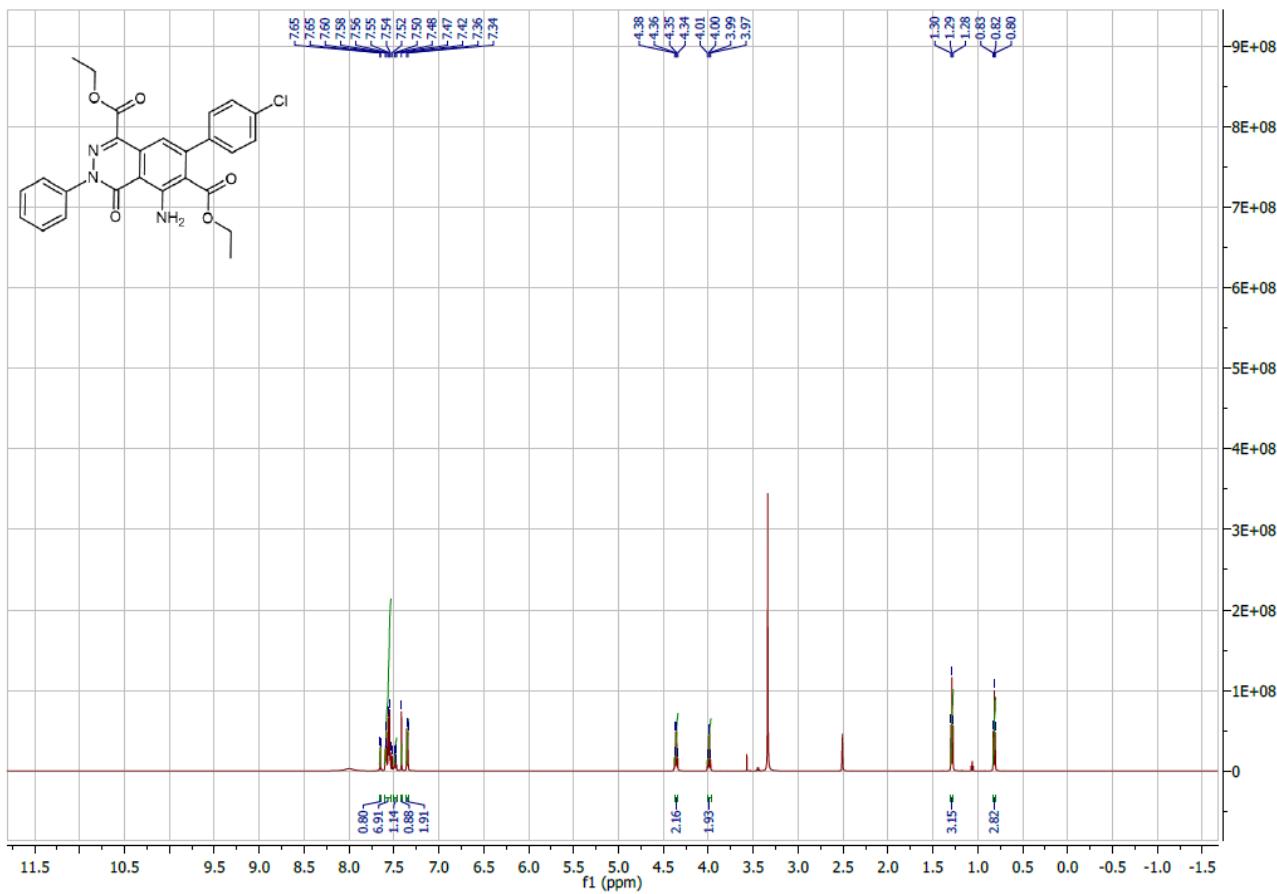
**Figure 18.**  $^1\text{H}$  NMR spectrum of compound 3g.



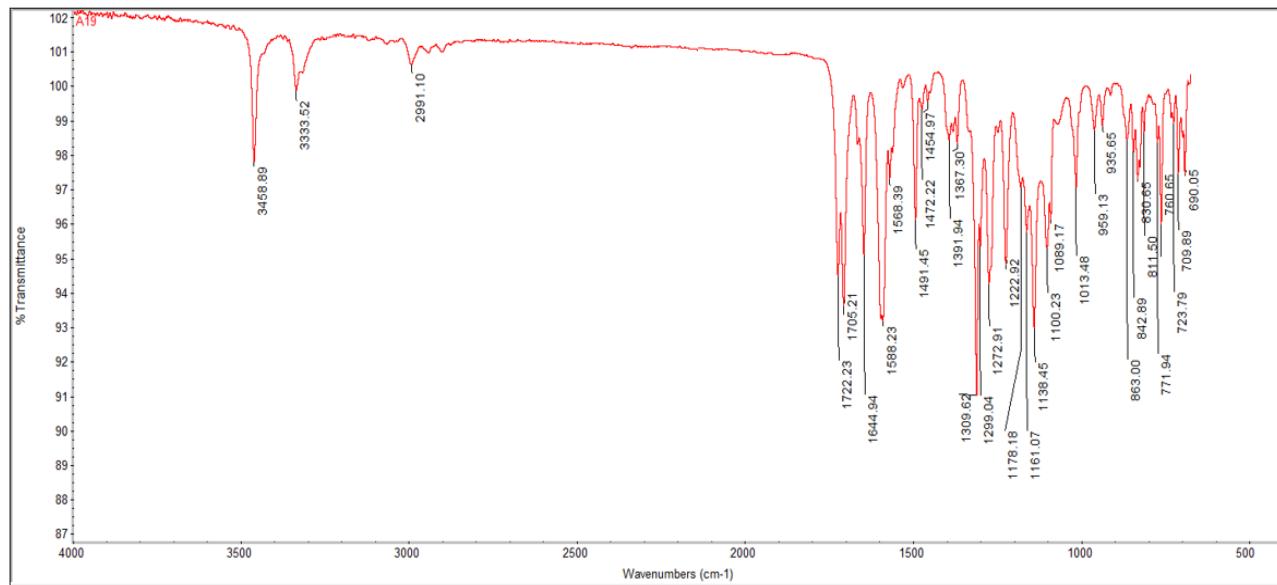
**Figure 19.**  $^{13}\text{C}$  NMR spectrum of compound **3g**.



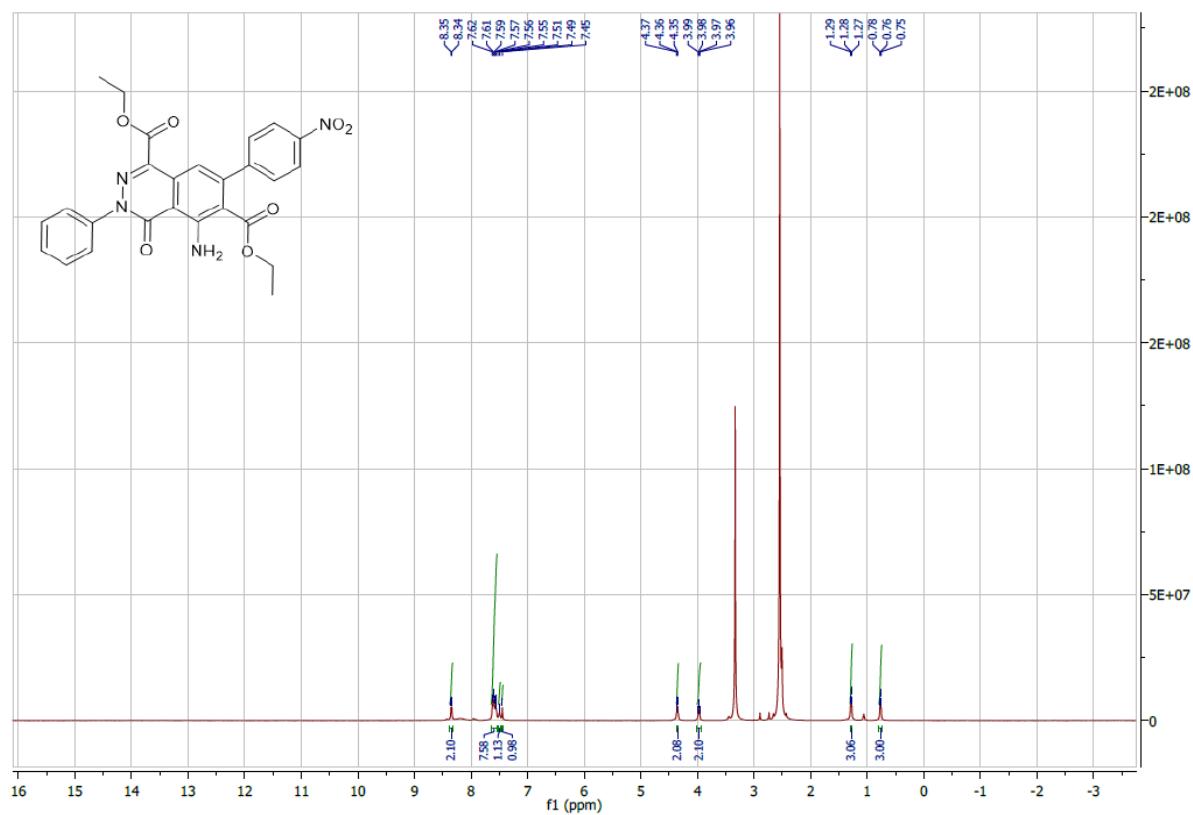
**Figure 20.** IR spectrum of compound **3g**.



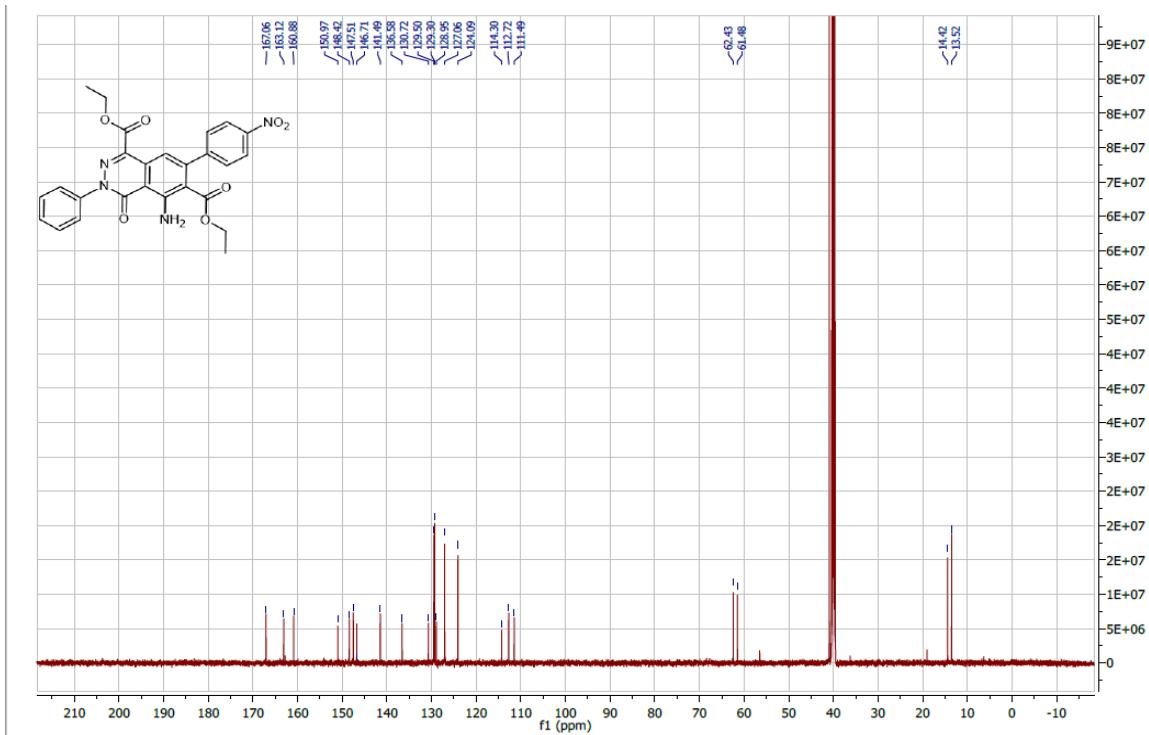
**Figure 21.**  $^1\text{H}$  NMR spectrum of compound 3h.



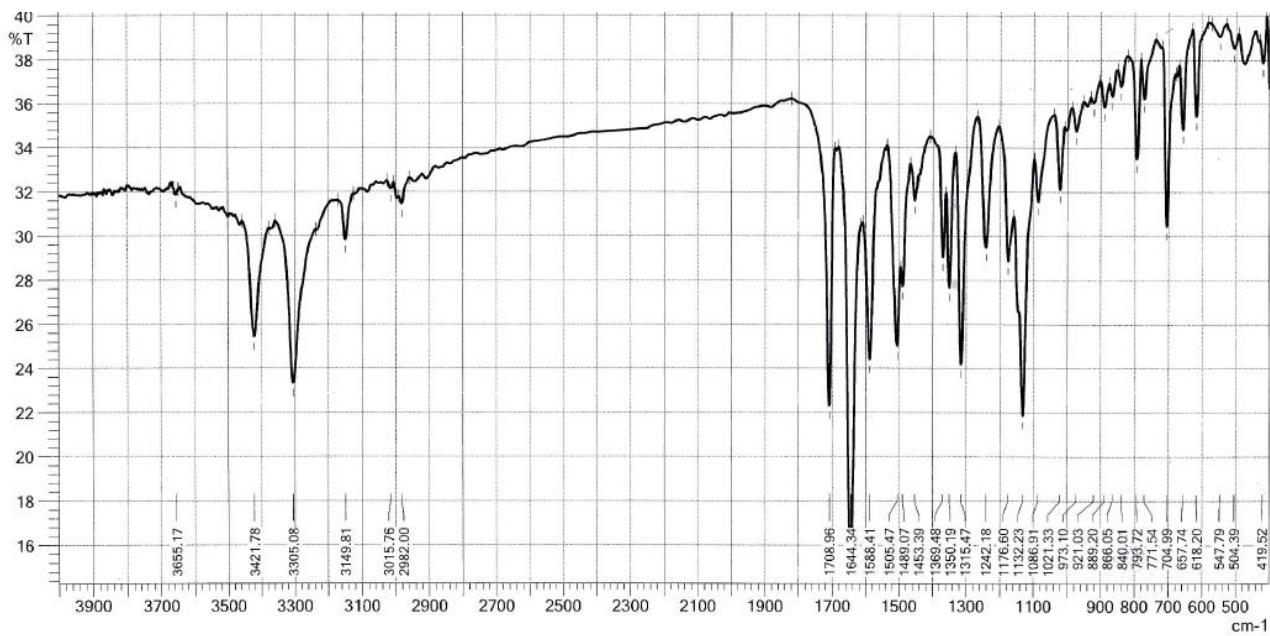
**Figure 22.** IR spectrum of compound **3h**.



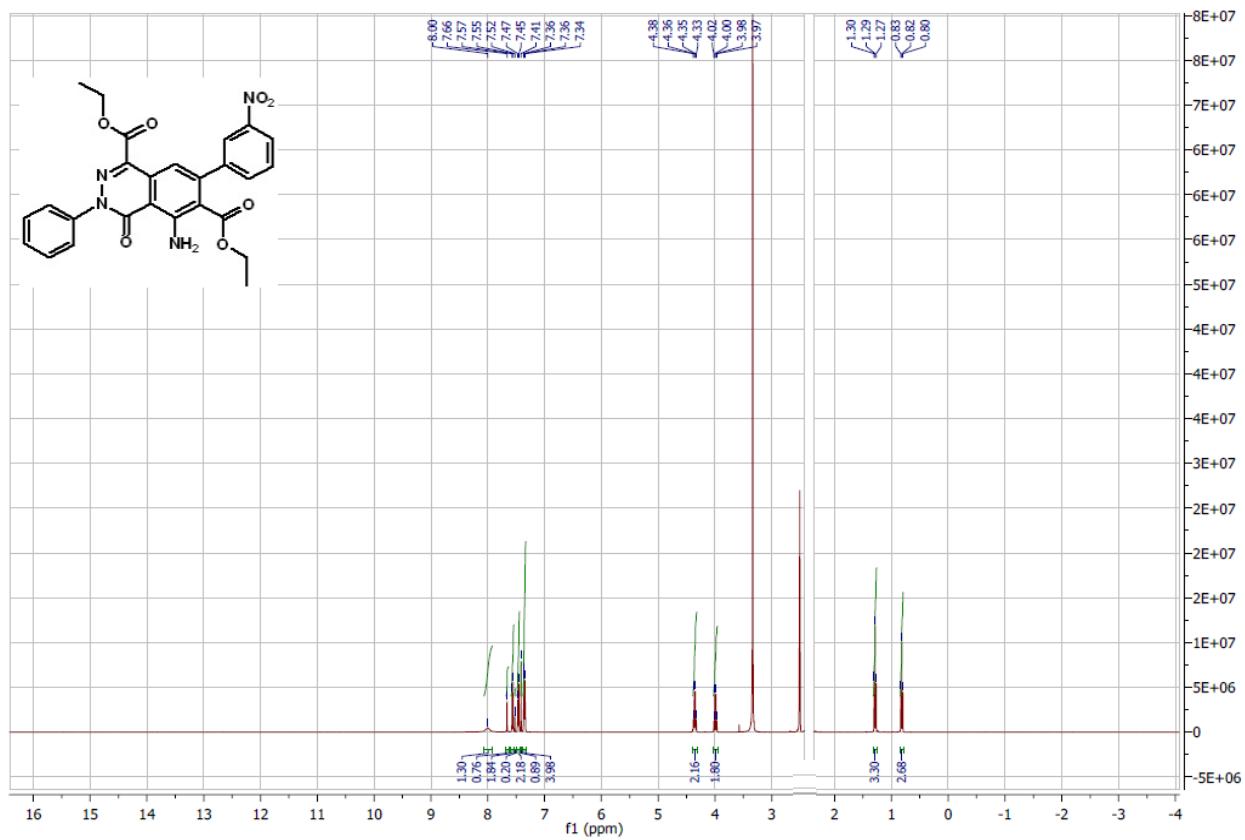
**Figure 23.**  $^1\text{H}$  NMR spectrum of compound 3i.



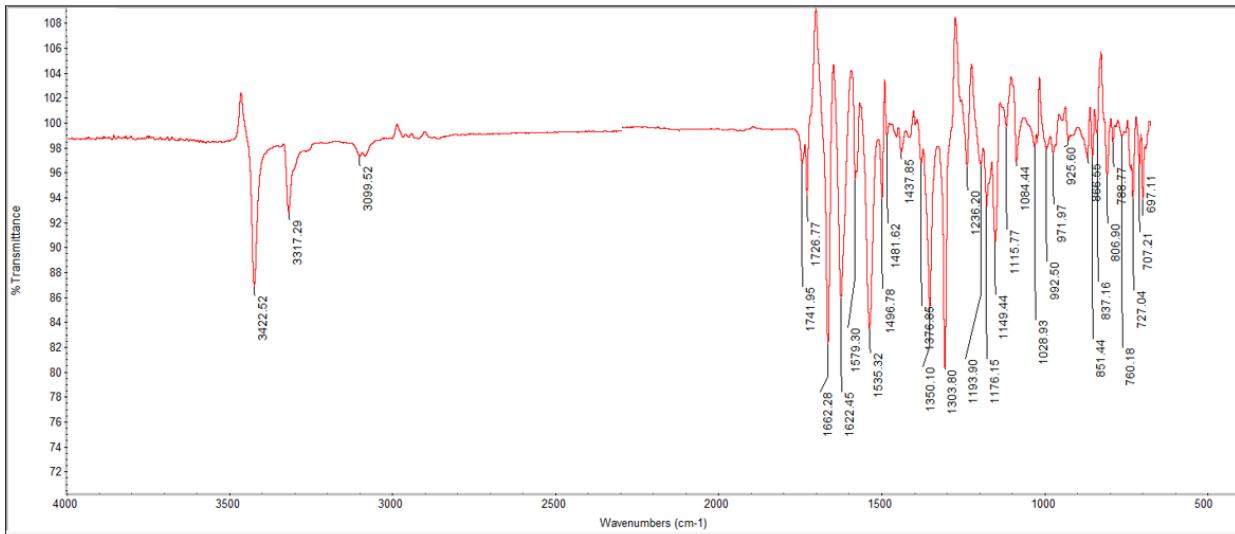
**Figure 24.**  $^{13}\text{C}$  NMR spectrum of compound 3i.



**Figure 25.** IR spectrum of compound **3i**.



**Figure 26.**  $^1\text{H}$  NMR spectrum of compound **3j**.



**Figure 27.** IR spectrum of compound **3j**.

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