

Supplementary Information for:

Programmable Assembly of Multistranded Helices in Water

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1. Methods and General Information

All reagents and solvents were purchased from commercial sources and used without further purification – except triethylamine, which was distilled on potassium hydroxide under inert atmosphere before being used. Reagents were purchased from Sigma-Aldrich [(triisopropylsilyl)acetylene, trimethylsilylacetylene, triethylamine, bis(triphenylphosphine)palladium(II) dichloride, tetrakis(triphenylphosphine)palladium(0), methyl trifluoromethanesulfonate, potassium perfluorobutanesulfonate, potassium trifluoromethanesulfonate] and Tokyo Chemical Industry Co [3,5-dibromopyridine, 3-ethynylpyridine, copper(I) iodide, tetrabutylammonium fluoride solution (1 M in tetrahydrofuran), 3-bromiodobenzene, 1,3-diiodobenzene, 1,3-diethynylbenzene, sodium 1-butanesulfonate, 4,4'-biphenyldisulfonic acid].

NMR analyses. Spectra were measured on a JEOL 400 MHz NMR ECX-400 spectrometer equipped with a 5-mm broadband 40TH5AT/FG2 autotunable universal probe, a 400 MHz Bruker Avance III NMR spectrometer equipped with a 5-mm BBFO probehead, a Bruker Avance III 500 MHz NMR spectrometer equipped with a 5-mm nitrogen-cooled ^1H , ^{13}C , ^{15}N Prodigy probehead and a Bruker Avance III HD 800 MHz NMR spectrometer, equipped with a cryogenically cooled, 5 mm ^1H , ^{13}C , ^{15}N triple-resonance TCI probehead. All signals were internally referenced to the solvent residue (CDCl_3 : 7.26 ppm, D_2O : 4.79 ppm, CD_3CN : 1.96 ppm) or tetramethylsilane. NMR solvents (CDCl_3 with or without 0.03% of tetramethylsilane, D_2O and CD_3CN) were purchased from Sigma-Aldrich and Eurisotop.

HR-MS analyses. HR-MS analyses were performed on an Agilent 6560 ESI-IM-QTOF mass spectrometer equipped with AJS ESI ion source.

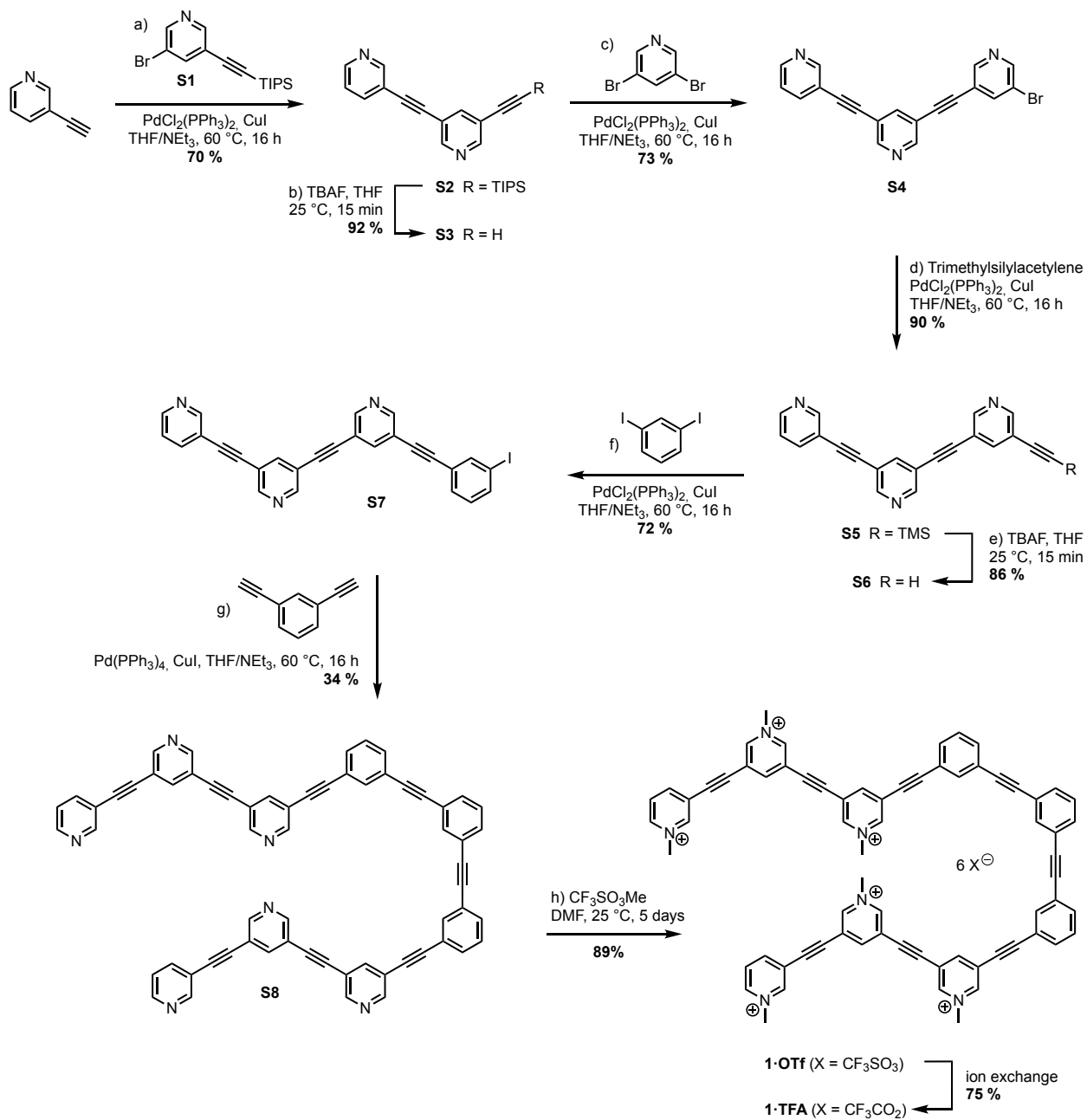
UHPLC-MS analyses. UHPLC-MS analyses were performed on an Agilent 6530-QTOF mass spectrometer (ionisation mode: ESI+) equipped with an Agilent 1290 UHPLC inlet, UV detector and Autosampler.

Melting Point. Melting points were measured on a Büchi Melting point B-540/B-545.

Preparative HPLC. Water soluble compounds were purified by HPLC Shimadzu LC-8A system equipped with a Shimadzu array detector SPD-M20A using a Gemini 10 μm C18 column 110 Å, 100 x 21.20 mm 10 micron from Phenomenex. Eluents: solution A (99.9% water, 0.1 % trifluoroacetic acid), solution B (100% acetonitrile). Gradient: 0 min, 5% B; 1 min, 5% B; 15 min, 100% B. Flow rate: 12 mL/min.

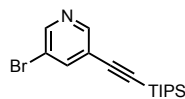
List of abbreviations. DOSY: Diffusion-ordered spectroscopy; DMF: N,N-Dimethylformamide, equiv.: equivalents; ESI-IM-QTOF: Electrospray Ionisation – Ion Mobility – Quadrupole Time-of-Flight; EXSY: Exchange spectroscopy; HR-MS: High-Resolution Mass Spectrometry; NMR: Nuclear Magnetic Resonance; OTf: Triflate; ROESY: rotating frame Overhauser effect spectroscopy; TBAF: Tetra-*n*-butylammonium fluoride; TFA: Trifluoroacetic acid; TIPS: Triisopropylsilyl; TMS: Trimethylsilyl; UHPLC: Ultra-High Performance Liquid Chromatography; UV: Ultraviolet.

2. Synthesis of strand 1



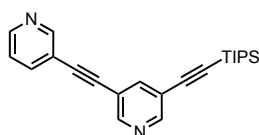
Scheme S1 | General procedure for the synthesis of strand 1.

Compound S1



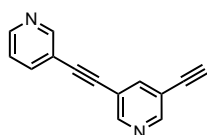
The synthesis of **S1** was adapted from a published procedure.^{S1} 3,5-Dibromopyridine (1.0 g, 4.2 mmol), (triisopropylsilyl)acetylene (1.0 mL, 4.7 mmol), PdCl₂(PPh₃)₂ (190 mg, 0.06 mmol) and CuI (23 mg, 0.03 mmol) were dissolved in triethylamine (10 mL) under nitrogen atmosphere. The mixture was stirred 16 h at 50 °C and concentrated by rotary evaporation. The crude product was redissolved in dichloromethane, washed with a saturated solution of Na₂CO₃, dried over Na₂SO₄, filtered, and concentrated by rotary evaporation. The residue was purified by silica gel column chromatography (gradient pentane/ethyl acetate = 100/0 to 30/70) to yield **S1** (1.4 g, 97 %) as a yellow oil. Spectral data closely match those previously reported in the literature.^{S1}

Compound S2



3-Ethynylpyridine (242 mg, 2.4 mmol), **S1** (783 mg, 2.3 mmol), PdCl₂(PPh₃)₂ (84 mg, 0.1 mmol) and CuI (12 mg, 0.06 mmol) were dissolved in triethylamine (3 mL) and tetrahydrofuran (3 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred 16 h at 60 °C, concentrated by rotary evaporation, and the resulting residue was purified by silica gel column chromatography (gradient pentane/ethyl acetate = 100/0 to 50/50) to yield **S2** (581.0 mg, 70 %) as a colorless oil. Melting point: 67-68 °C. ¹H NMR (400 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K): δ 8.78 (br d, *J* = 1.7 Hz, 1H), 8.67 (d, *J* = 2.1 Hz, 1H), 8.64 (d, *J* = 2.1 Hz, 1H), 8.59 (dd, *J* = 4.9, 1.7 Hz, 1H), 7.90 (t, *J* = 2.1 Hz, 1H), 7.82 (dt, *J* = 8.1, 1.7 Hz, 1H), 7.32 (ddd, *J* = 8.1, 4.9, 0.8 Hz, 1H), 1.13 (m, 21H). ¹³C NMR (100 MHz, CDCl₃ + 0.03 % TMS, 293 K): δ 152.4, 151.9, 150.9, 149.3, 141.3, 138.7, 123.3, 120.5, 119.6, 119.4, 102.4, 96.2, 89.8, 88.5, 18.7, 11.3. HR-MS (ESI+) *m/z* calculated for C₂₃H₂₈N₂Si⁺ [M]⁺ = 361.2024, found 360.2022, Δ = -0.58 ppm.

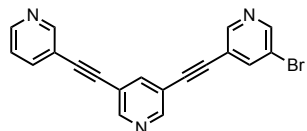
Compound S3



Tetrabutylammonium fluoride (1 M in tetrahydrofuran, 720 μL) was added to a solution of **S2** (795 mg, 2.2 mmol) in tetrahydrofuran (150 mL). The reaction mixture was stirred for 15 min at 25 °C and concentrated by rotary evaporation. The residue was purified by silica gel column chromatography (gradient pentane/ethyl acetate = 100/0 to 60/40) to yield **S3** (403 mg, 92 %) as a white solid. Melting point: 107-108 °C. ¹H NMR (400 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K): δ 8.78 (br s, 1H), 8.70 (d, *J* = 2.0 Hz, 1H), 8.66 (d, *J* = 2.0 Hz, 1H), 8.60 (br d, *J* = 5.0 Hz, 1H), 7.91 (t, *J* = 2.0 Hz, 1H), 7.82 (dt, *J* = 7.8, 2.0 Hz, 1H), 7.32 (dd, *J* = 7.8, 5.0 Hz, 1H), 3.26 (s, 1H). ¹³C NMR (100 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K): δ 152.4, 151.9, 151.5, 149.4, 141.4, 138.7, 123.3, 119.5, 119.2, 90.0, 88.2, 81.6, 79.5. HR-MS (ESI+) *m/z* calculated

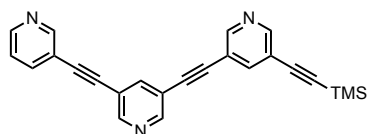
for $C_{14}H_9N_2$ $[M]^+ = 204.0687$, found 204.0695, $\Delta = -3.76$ ppm.

Compound S4



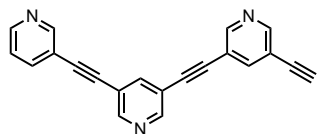
Compound **S3** (127 mg, 0.6 mmol), 3,5-dibromopyridine (966 mg, 4.1 mmol), $PdCl_2(PPh_3)_2$ (72 mg, 0.1 mmol) and CuI (11 mg, 0.03 mmol) were dissolved in dry triethylamine (10 mL) and tetrahydrofuran (10 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred for 8 h at 25 °C and 16 h at 60 °C before being concentrated by rotary evaporation. The crude product was redissolved in dichloromethane, washed with a saturated solution of Na_2CO_3 , dried over Na_2SO_4 , filtered, and concentrated by rotary evaporation. The residue was purified by silica gel column chromatography (gradient pentane/ethyl acetate = 100/0 to 0/100) to yield **S4** (481 mg, 73 %) as a white solid. Melting point: 189-190 °C. 1H NMR (400 MHz, $CDCl_3$ + 0.03 % tetramethylsilane, 293 K): δ 8.79 (br s, 1H), 8.74 (d, $J = 2.0$ Hz, 1H), 8.72 (d, $J = 2.0$ Hz, 1H), 8.68 (d, $J = 2.0$ Hz, 1H), 8.66 (d, $J = 2.0$ Hz, 1H), 8.61 (br d, $J = 4.9$ Hz, 1H), 7.98 (t, $J = 2.0$ Hz, 1H), 7.97 (t, $J = 2.0$ Hz, 1H), 7.83 (dt, $J = 7.9, 2.0$ Hz, 1H), 7.33 (dd, $J = 7.9, 4.9$ Hz, 1H). ^{13}C NMR (100 MHz, $CDCl_3$ + 0.03 % tetramethylsilane, 293 K): δ 152.5, 151.7, 151.4, 150.6, 150.4, 149.5, 140.9, 138.7, 123.3, 120.9, 120.3, 119.7, 119.5, 119.2, 90.3, 89.5, 88.2. HR-MS (ESI+) m/z calculated for $C_{14}H_{11}BrN_3$ $[M]^+ = 359.0058$, found 359.0062, $\Delta = -1.19$ ppm.

Compound S5



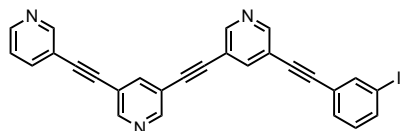
Compound **S4** (1.90 g, 5.27 mmol, 1.00 equiv.), $PdCl_2(PPh_3)_2$ (0.185 g, 0.263 mmol, 0.05 equiv.), CuI (50 mg, 0.263 mmol, 0.05 equiv.) in dry degassed tetrahydrofuran (25 mL) at 0 °C was added trimethylsilylacetylene (1.12 mL, 7.91 mmol, 1.5 equiv.) and Et_3N (2.5 mL, 15.82 mmol, 3.00 equiv.). The resulting mixture was then allowed to stir overnight at 60 °C and monitored by thin-layer chromatography. After completion, the reaction mixture was filtered through celite. The resulting organic phase was diluted in dichloromethane (50 mL) and washed with a saturation solution of NH_4Cl . The combined organic layers were washed with brine (100 mL), dried (Na_2SO_4), filtered and concentrated. The residue was purified by flash chromatography (hexane/ethyl acetate = 30/70) to give product **S5** (1.79 g, 90%) as a white solid. Melting point: 188-190 °C. 1H NMR (500 MHz, $CDCl_3$, 298 K): δ 8.80 (s, 1H), 8.74 (dd, $J = 2.0, 2.0$ Hz, 2H), 8.68 (dd, $J = 2.0$ Hz, 2H), 8.62 (d, $J = 4.8$ Hz, 1H), 7.98 (br d, $J = 2.0$ Hz, 1H), 7.91 (br d, $J = 2.0$ Hz, 1H), 7.86-7.84 (m, 1H), 7.34 (dd, $J = 4.7, 4.9$ Hz, 1H), 0.28 (s, 9H). ^{13}C NMR (126 MHz, $CDCl_3$, 298 K): δ 152.37, 151.86, 151.42, 151.28, 150.94, 149.36, 141.18, 140.77, 138.62, 123.16, 120.14, 119.59, 119.44, 119.33, 118.97, 100.29, 99.64, 90.08, 89.17, 88.73, 88.16, -0.25. HR-MS (ESI+) m/z calculated for $C_{24}H_{19}N_3Si$ $[M+H]^+ = 378.1426$, found 378.1424, $\Delta = -1.56$ ppm.

Compound S6



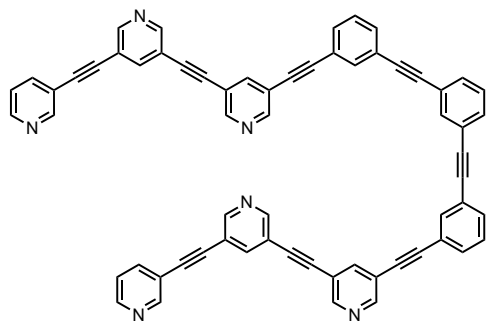
To a stirred solution of substrate **S5** (1.75 g, 4.63 mmol, 1.00 equiv.) in tetrahydrofuran/methanol (1:2, 30 mL) was added K_2CO_3 (1.27 g, 9.27 mmol, 2.00 equiv.). The resulting mixture was stirred at room temperature for 15 min at 25 °C and monitored by thin-layer chromatography. After completion, the reaction mixture was filtered, concentrated and subjected to column chromatography (hexane/ethyl acetate = 20/80) to give product **S6** as a white solid (1.216 g, 86%). Melting point: 189-190 °C. 1H NMR (500 MHz, $CDCl_3$, 298 K): δ 8.81 (br d, J = 2.0 Hz, 1H), 8.76-8.74 (m, 3H), 8.70 (d, J = 2.0 Hz, 1H), 8.62 (dd, J = 2.0, 2.0 Hz, 1H), 8.0 (br t, J = 2.0 Hz, 1H), 7.94 (br t, J = 2.0 Hz, 1H), 7.87-7.85 (m, 1H), 7.35 (dd, J = 4.0, 5.0 Hz, 1H), 3.29 (s, 1H). ^{13}C NMR (126 MHz, $CDCl_3$, 298 K): δ 152.38, 152.06, 151.49, 151.45, 151.30, 149.38, 141.37, 140.81, 138.62, 123.17, 119.61, 119.43, 119.24, 119.15, 90.12, 88.95, 88.14, 81.66, 79.34. HR-MS (ESI+) m/z calculated for $C_{21}H_{11}N_3^+$ $[M+H]^+$ = 306.1031, found 306.1028, Δ = 0.38 ppm.

Compound S7



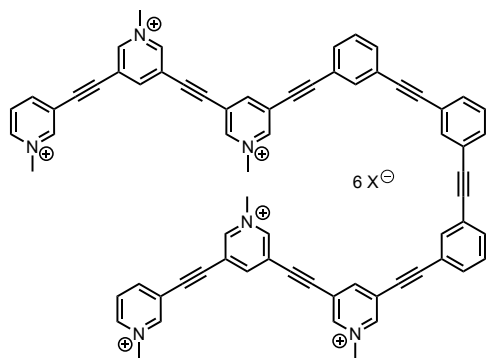
Compound **S6** (1.0 g, 3.27 mmol, 1.00 equiv.), $Pd(PPh_3)_4$ (0.189 g, 0.163 mmol, 0.05 equiv.), CuI (31 mg, 0.163 mmol, 0.05 equiv.) in dry degassed THF (25 mL) at 0 °C was added 1,3-diiodobenzene (1.18 mL, 3.60 mmol, 1.1 equiv.) and Et_3N (1.4 mL, 9.82 mmol, 3.00 equiv.). The resulting mixture was then allowed to stir overnight at 60 °C and monitored by thin-layer chromatography. After completion, the reaction mixture was filtered through celite. The resulting organic phase was diluted in dichloromethane (50 mL) and washed with a saturated solution of NH_4Cl . The combined organic layers were washed with brine (100 mL), dried (Na_2SO_4), filtered and concentrated. The residue was purified by flash chromatography (hexane/ethyl acetate = 20/80) to give product **S7** (1.2 g, 72 %) as a white solid. Melting point: 186-188 °C. 1H NMR (500 MHz, $CDCl_3$, 298 K): δ 8.82 (br d, J = 2.0, 1H), 8.76-8.73 (m, 4H), 8.63 (dd, J = 2.0, 2.0 Hz, 1H), 8.0 (br t, J = 2.0 Hz, 1H), 7.97 (br t, J = 2.0 Hz, 1H), 7.93 (br t, J = 2.0 Hz, 1H), 7.87-7.85 (m, 1H), 7.75-7.73 (m, 1H), 7.54-7.52 (m, 1H), 7.35 (dd, J = 5.0, 5.0 Hz, 1H), 7.14 (t, J = 8.0 Hz, 1H). ^{13}C NMR (126 MHz, $CDCl_3$, 298 K): δ 152.37, 151.49, 151.31, 151.10, 149.37, 140.82, 140.75, 140.30, 138.64, 138.11, 130.86, 130.02, 124.21, 123.19, 119.88, 119.63, 119.44, 119.30, 119.20, 93.78, 91.80, 90.12, 89.12, 88.93, 88.16, 86.16. HR-MS (ESI+) m/z calculated for $C_{27}H_{14}N_3I^+$ $[M+H]^+$ = 508.0310, found 508.0305, Δ = -0.46 ppm.

Compound S8



Compound **S7** (1.80 g, 3.56 mmol, 2.00 equiv.), Pd(PPh₃)₄ (0.205 g, 0.178 mmol, 0.1 equiv.), CuI (33 mg, 0.178 mmol, 0.1 equiv.) in dry degassed tetrahydrofuran (25 mL) at 0 °C was added 1,3-diethynylbenzene (0.237 mL, 1.78 mmol, 1.0 equiv.) and Et₃N (1.5 mL, 10.68 mmol, 6.00 equiv.). The resulting mixture was then allowed to stir overnight at 60 °C and monitored by thin-layer chromatography. After completion, the reaction mixture concentrated by rotary evaporation. The crude product was diluted in dichloromethane (50 mL) and washed with a saturated solution of NH₄Cl. The combined organic layers were washed with brine (100 mL), dried (Na₂SO₄), filtered and concentrated. The residue was purified by flash chromatography (dichloromethane/methanol = 95/5) to give product **S8** (568 g, 34 %) as a pale-yellow solid. Melting point: >400 °C (decomposition). ¹H NMR (500 MHz, CDCl₃, 298 K): δ 8.82 (br t, *J* = 3.0, 2H), 8.77-8.73 (m, 8H), 8.63 (br d, *J* = 5.0 Hz, 2H), 8.0 (br t, *J* = 6.0 Hz, 4H), 7.86 (br t, *J* = 6.0 Hz, 2H), 7.76 (br t, *J* = 3.0 Hz, 2H), 7.58-7.53 (m, 5H), 7.43-7.39 (m, 3H), 7.37-7.33 (m, 4H). ¹³C NMR (126 MHz, CDCl₃, 298 K): δ 152.37, 151.54, 151.45, 151.31, 150.99, 149.35, 140.78, 138.64, 138.63, 135.41, 134.75, 133.17, 132.43, 132.16, 131.62, 128.70, 123.48, 123.18, 122.58, 120.07, 119.60, 119.44, 119.31, 119.16, 90.11, 89.19, 88.89, 88.18. HR-MS (ESI+) *m/z* calculated for C₆₄H₃₃N₆⁺ [M+H]⁺ = 885.2762, found 885.2732, Δ = 3.39 ppm.

Strand 1



Methyl triflate (120 μL, 1.1 mmol) was added to a solution of **S7** (40 mg, 0.05 mmol) in dry N,N-dimethylformamide (2 mL) under argon atmosphere. The mixture was stirred 24 h at 60 °C, and reaction progress was monitored by LC-MS. Upon completion, water (10 mL) was added, and the mixture was lyophilized to obtain a crude solid. The residue was redissolved in water (5 mL), cooled to 0 °C, and centrifuged. The resulting precipitate was collected and washed with dichloromethane (**1·OTf**, 75 mg, 89% yield).

Ion exchange was performed on preparative HPLC to afford salt of **1·CF₃CO₂** as a beige solid (50 mg, 75% yield). Melting point: >400 °C (decomposition). ¹H NMR (500 MHz, CD₃CN, 298 K): δ 9.08 (br t, *J* = 2.0, 2H), 9.04 (br t, *J* = 2.0, 2H), 9.01-8.99 (m, 4H), 8.93 (br t, *J* = 1.0 Hz, 2H), 8.83 (br t, *J* = 2.0 Hz, 2H), 8.74 (br t, *J* = 2.0 Hz, 2H), 8.72-8.70 (m, 2H), 8.65 (br t, *J* = 2.0 Hz, 1H), 8.63 (br t, *J* = 2.0 Hz, 1H), 8.12-8.09 (m, 2H), 7.83 (t, *J* = 2.0 Hz, 2H), 7.77 (t, *J* = 1.0 Hz, 1H), 7.73-7.68 (m, 4H), 7.63 (dd, *J* = 1.0, 2.0 Hz, 2H), 7.57-7.48 (m, 3H), 4.40 (s, 6H), 4.37 (s, 6H), 4.36 (s, 6H). ¹³C NMR (126 MHz, CD₃CN, 298 K): δ 160.6, 160.3, 150.2, 149.6, 149.1, 149.0, 148.8, 148.5, 147.8, 147.0, 135.5, 135.1, 134.5, 133.2, 132.9, 130.5, 130.3, 129.4, 124.5,

124.0, 123.6, 123.4, 123.1, 122.9, 122.0, 120.7, 116.0, 90.2, 89.5, 89.3, 88.8, 82.4, 50.2, 50.0, 49.7. HR-MS (ESI+) m/z calculated for $C_{80}H_{50}F_{15}N_6O_{10}^+$ $[M+5CF_3CO_2]^+ = 1539.3344$, found 1539.3328, $\Delta = -1.04$ ppm.

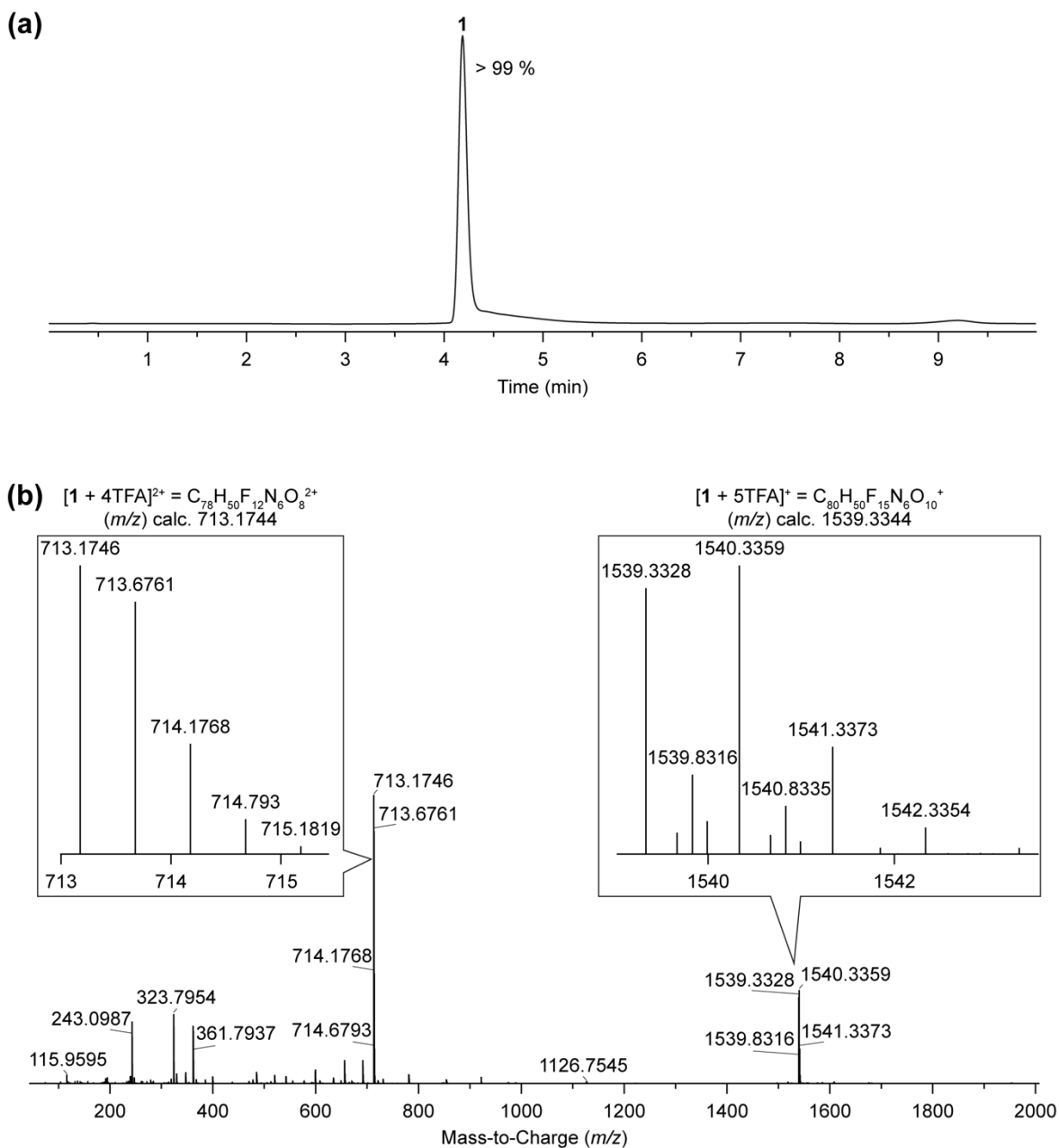


Figure S1 | UHPLC-MS characterization of strand 1. Tandem **(a)** Reverse-phase UHPLC chromatogram and **(b)** MS spectrum of strand 1. Absorbance was recorded at 254 nm.

3. NMR characterization of unfolded strand 1 in CD₃CN

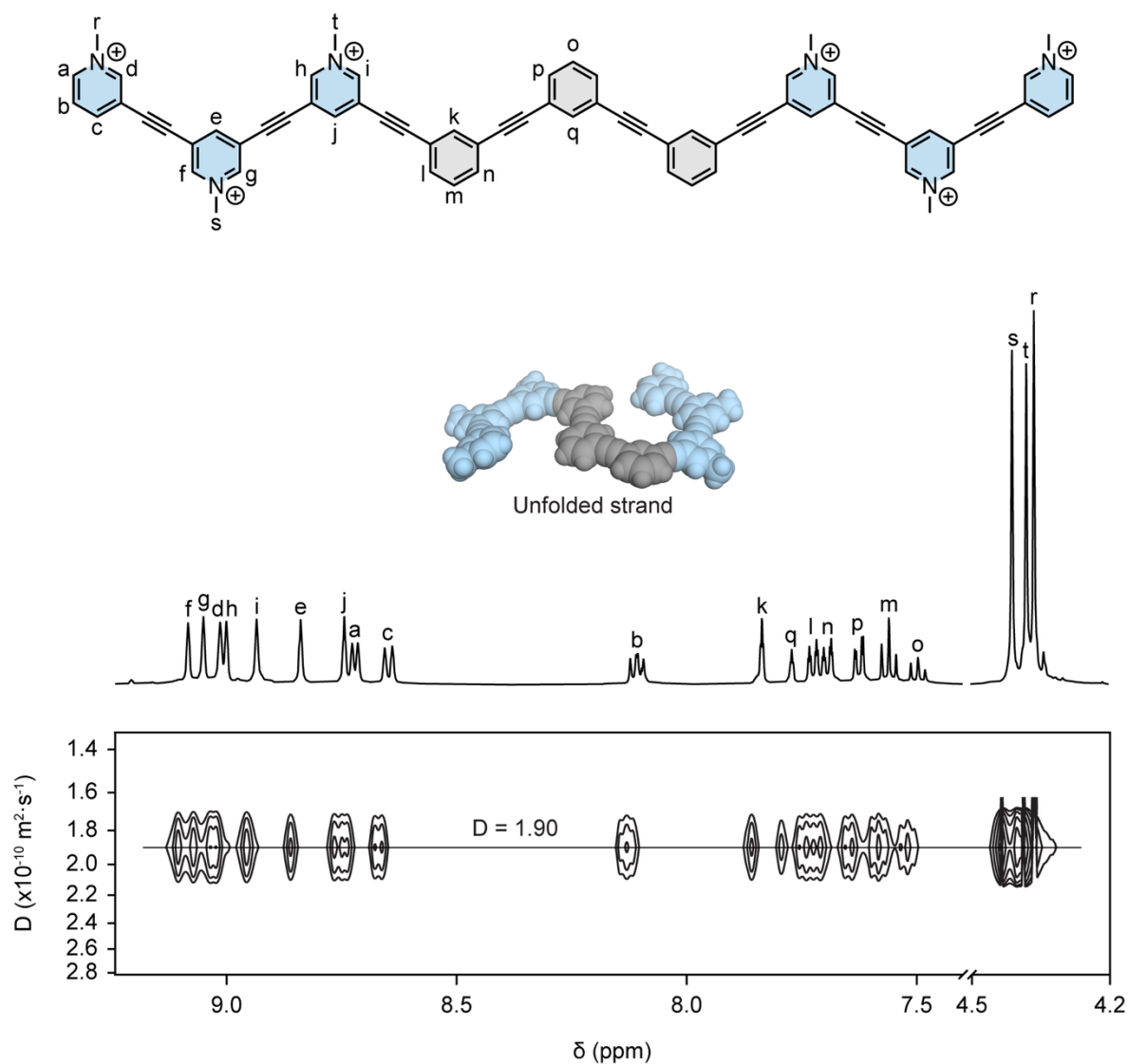


Figure S2 | Expansion of ¹H and DOSY NMR spectra of unfolded strand 1·TFA in CD₃CN (5 mM, 500 MHz, 298 K). Hydrodynamic radius $r_H = 12.0 \text{ \AA}$ was calculated using the Stokes-Einstein equation.

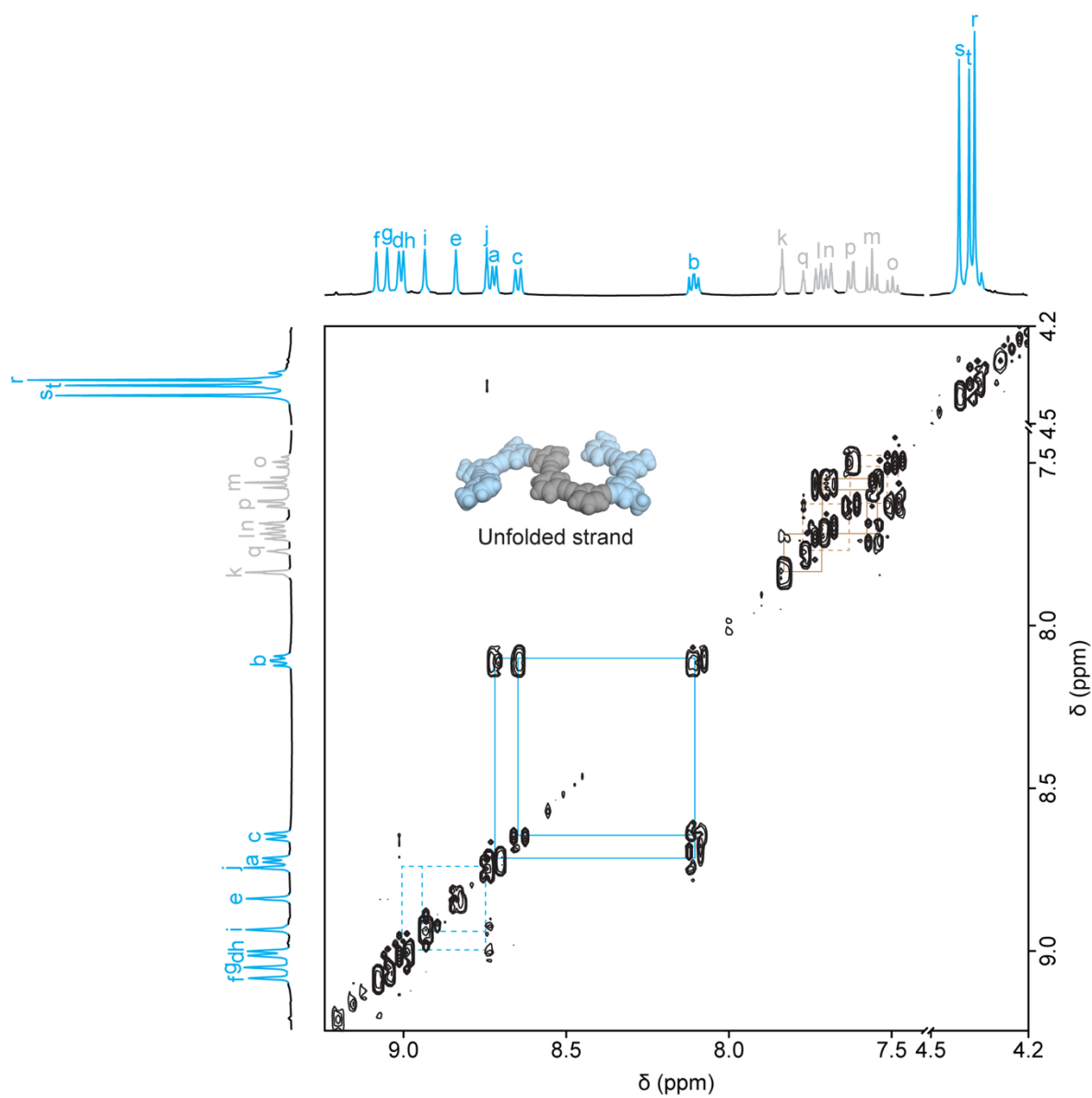


Figure S3 | Expansion of 2D ^1H - ^1H COSY NMR spectrum of unfolded strand 1·TFA in CD_3CN (5 mM, 500 MHz, 298 K).

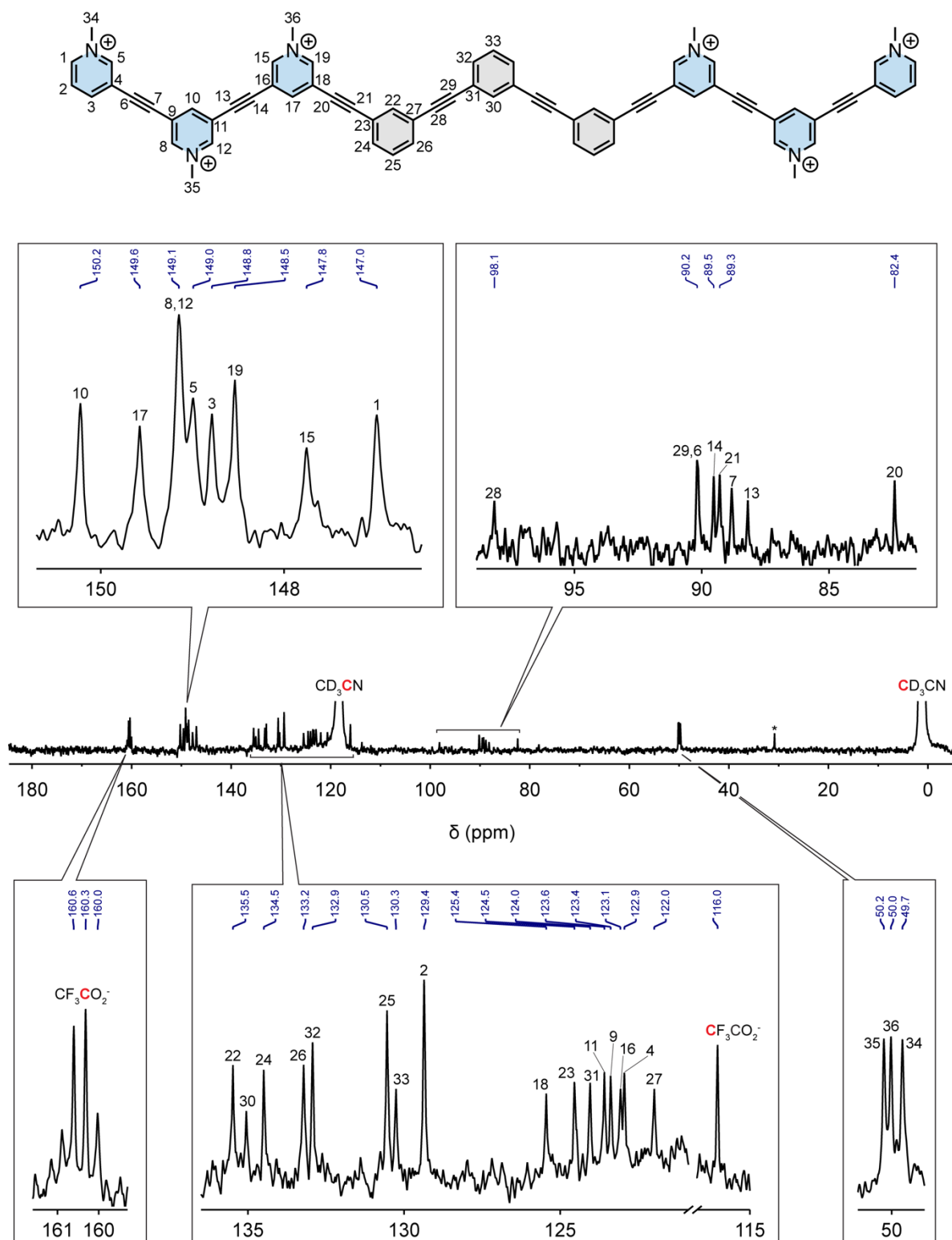


Figure S4 | ^{13}C NMR spectrum of unfolded strand 1·TFA in CD_3CN (5 mM, 126 MHz, 298 K). Assigned ^{13}C NMR chemical shifts are denoted onto the 1·TFA structure.

4. NMR characterization of double helix (1)₂ in D₂O

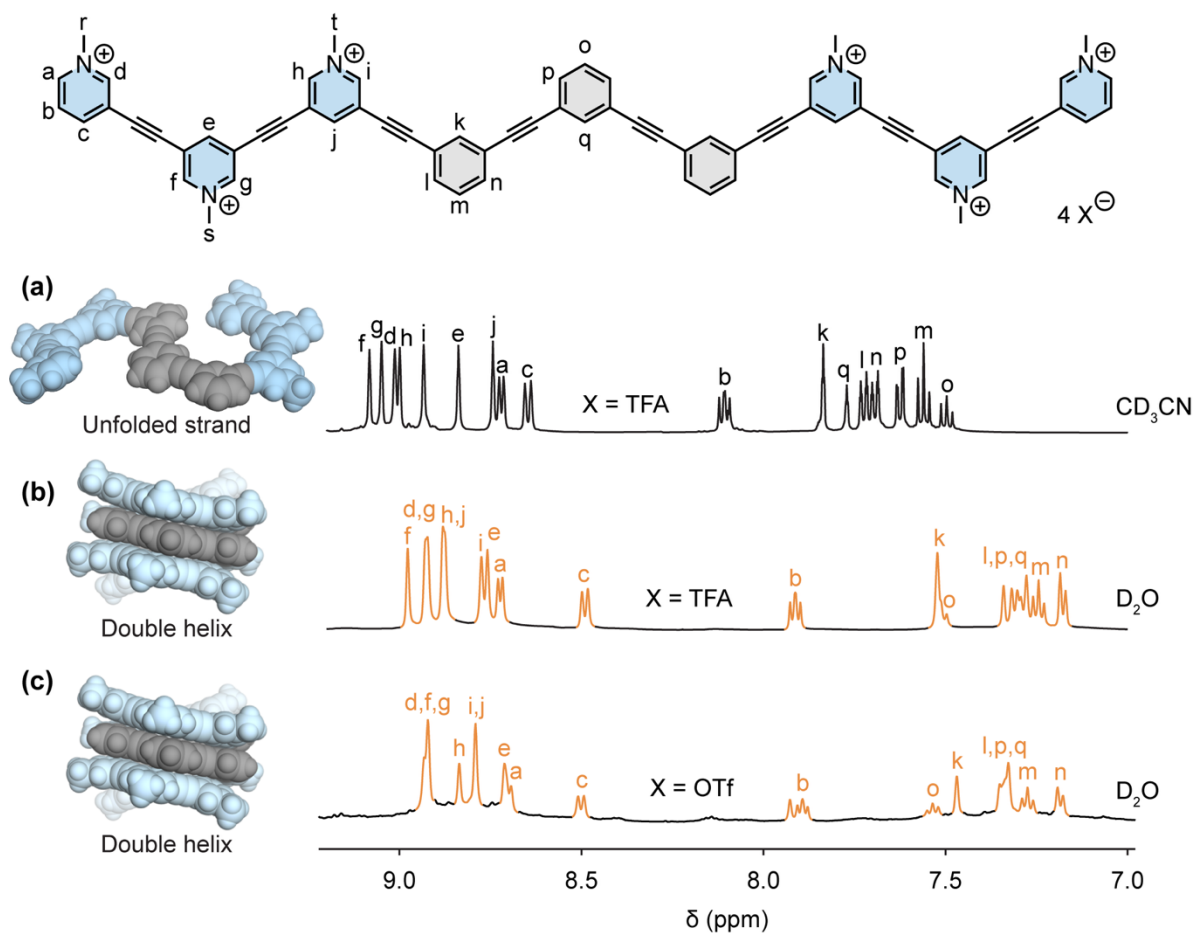


Figure S5 | Solvent-dependent ¹H NMR spectra of strand 1. Comparison between the ¹H NMR spectra of **a**, unfolded strand 1·TFA in CD₃CN (5 mM); **b**, double helix (1)₂·TFA in D₂O (10 mM); and **c**, double helix (1)₂·OTf in D₂O (3 mM). All the spectra were recorded at 500 MHz and 298 K.

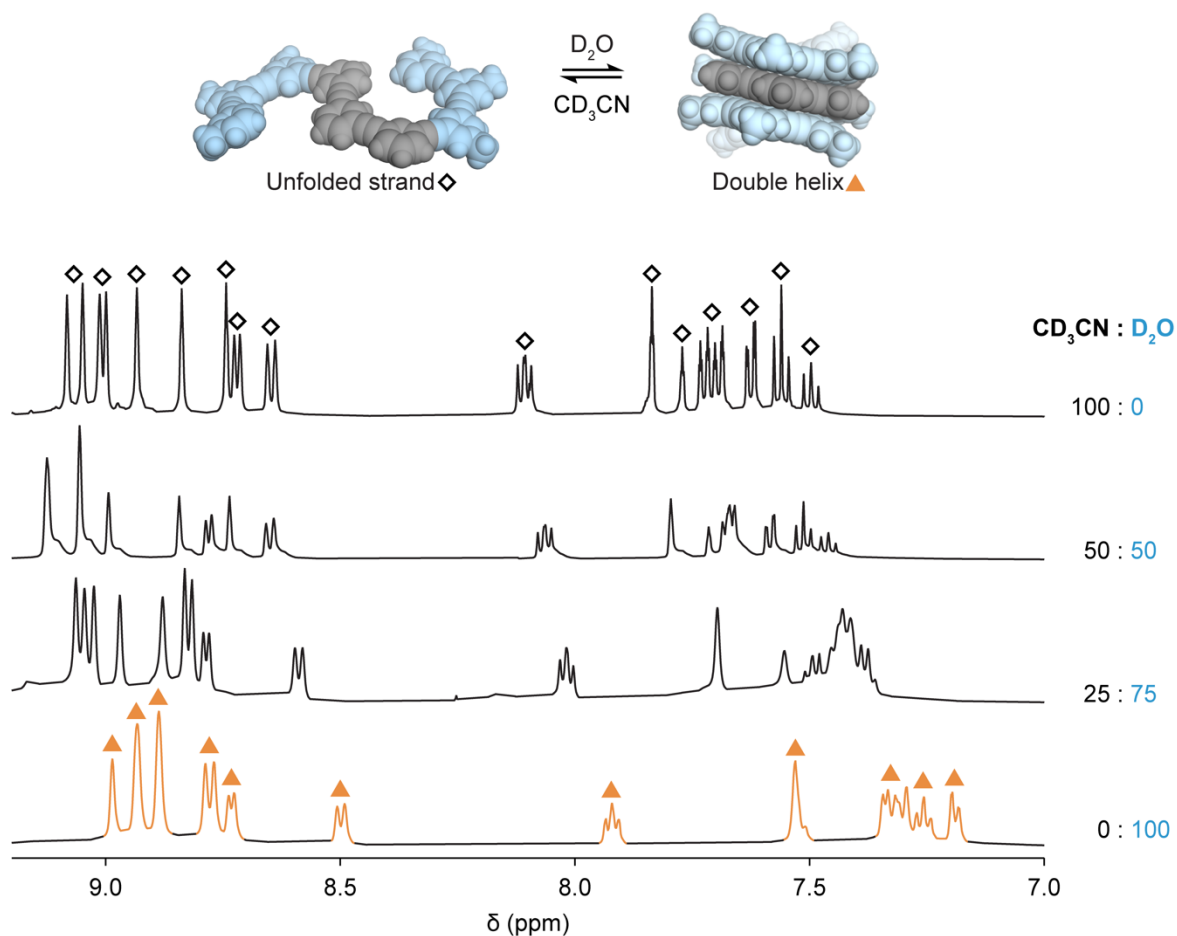


Figure S6 | Expansion of aromatic ^1H NMR spectral region of 1·TFA in $\text{CD}_3\text{CN} / \text{D}_2\text{O}$ mixtures (5 mM, 500 MHz, 298 K). The signals corresponding to the unfolded strand and double helix are labelled with empty diamonds and orange triangles, respectively.

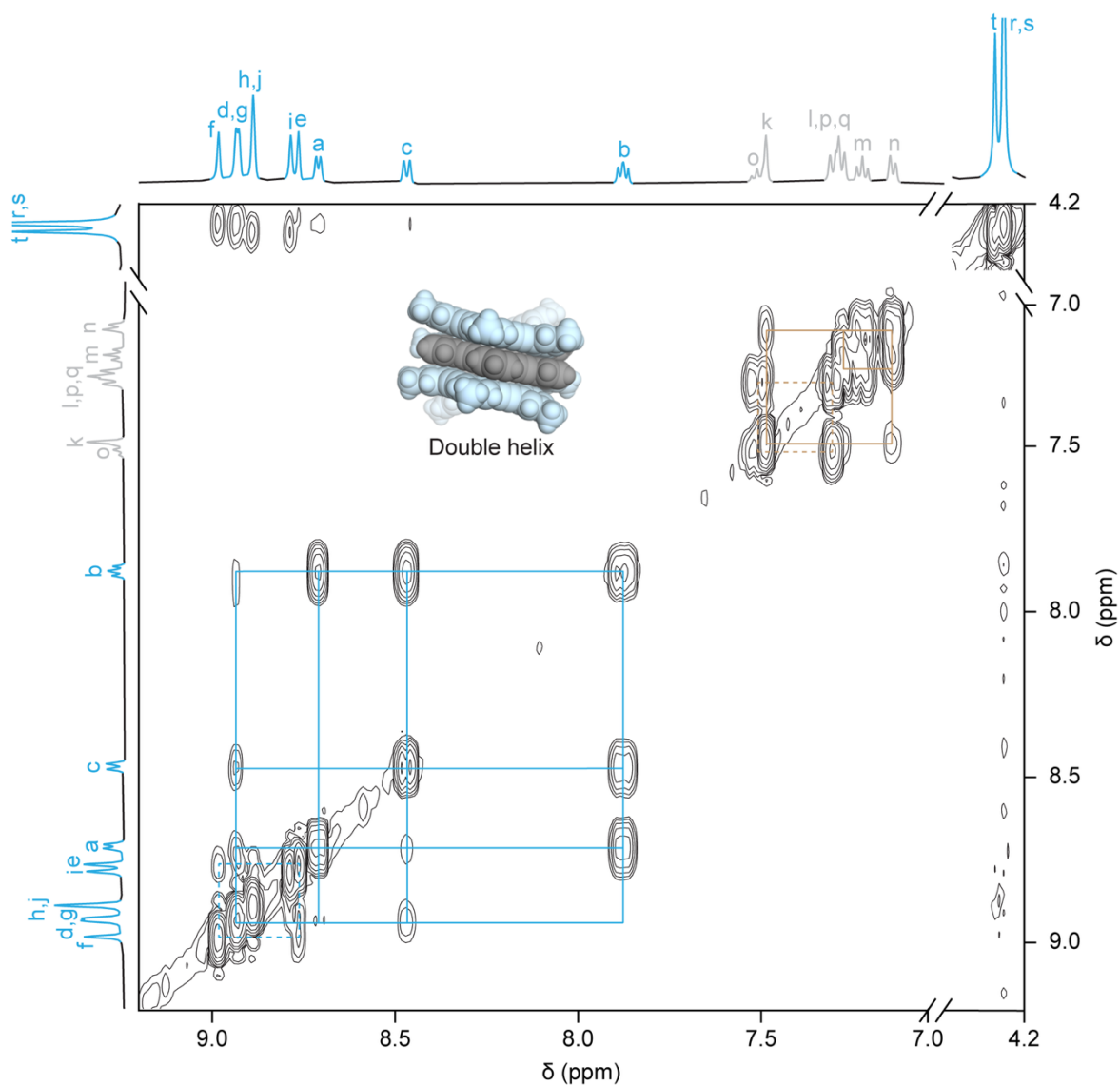


Figure S7 | Expansion of selected regions in 2D ^1H - ^1H COSY NMR spectrum of double helix (1) $_2$ ·TFA in D $_2$ O (10 mM, 500 MHz, 278 K).

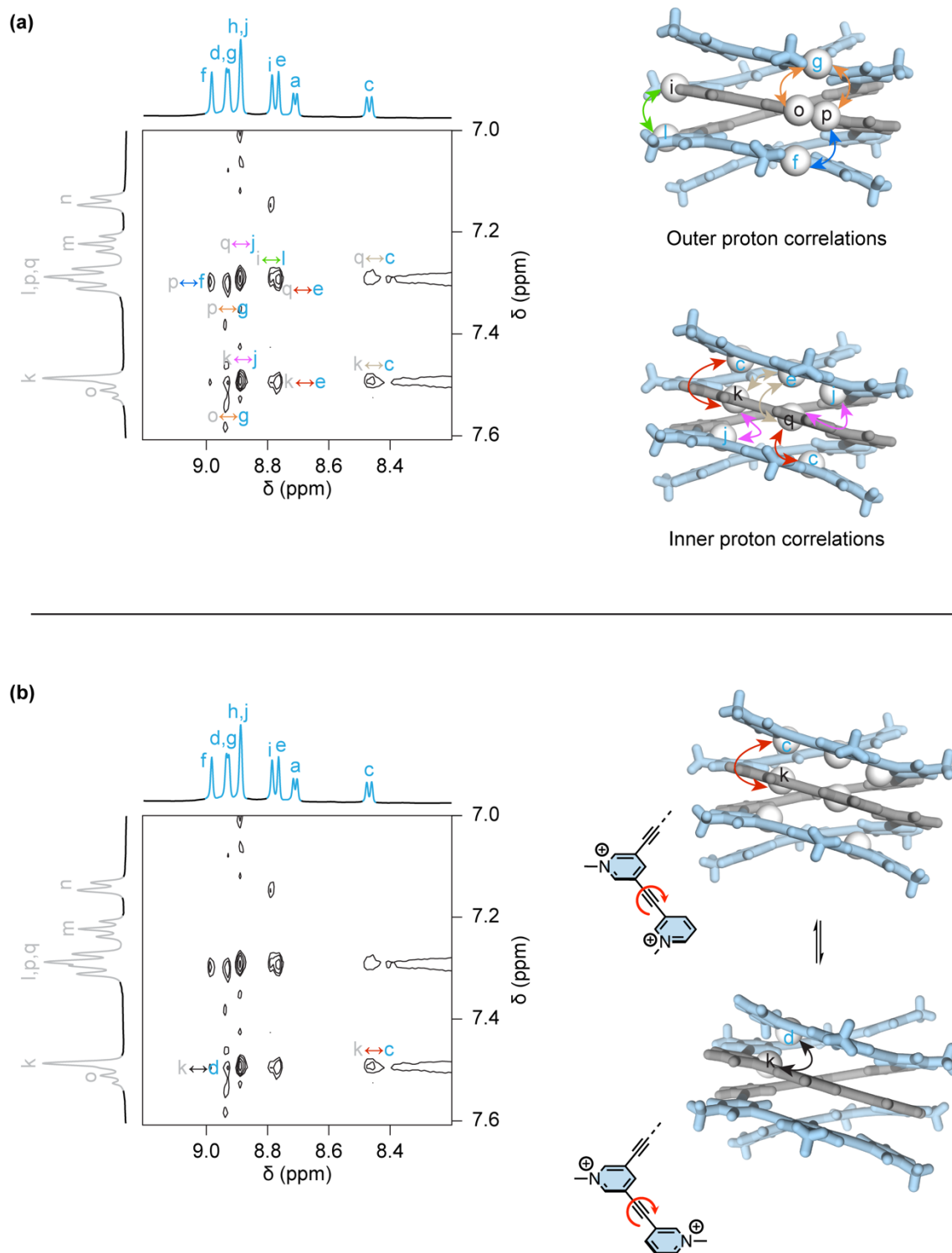


Figure S8 | Expansion of 2D ^1H - ^1H ROESY NMR spectrum of double helix $(\mathbf{1})_2\cdot\text{TFA}$ in D_2O (10 mM, 500 MHz, 278 K, 100 ms mixing time). **a, Key ROE correlations between protons of stacked pyridinium (colored in blue) and phenylene (colored in gray) residues. **b**, Rotation of the terminal pyridinium residue around the acetylene bond results in the appearance of additional ROE correlations between stacked pyridinium and phenylene residues.**

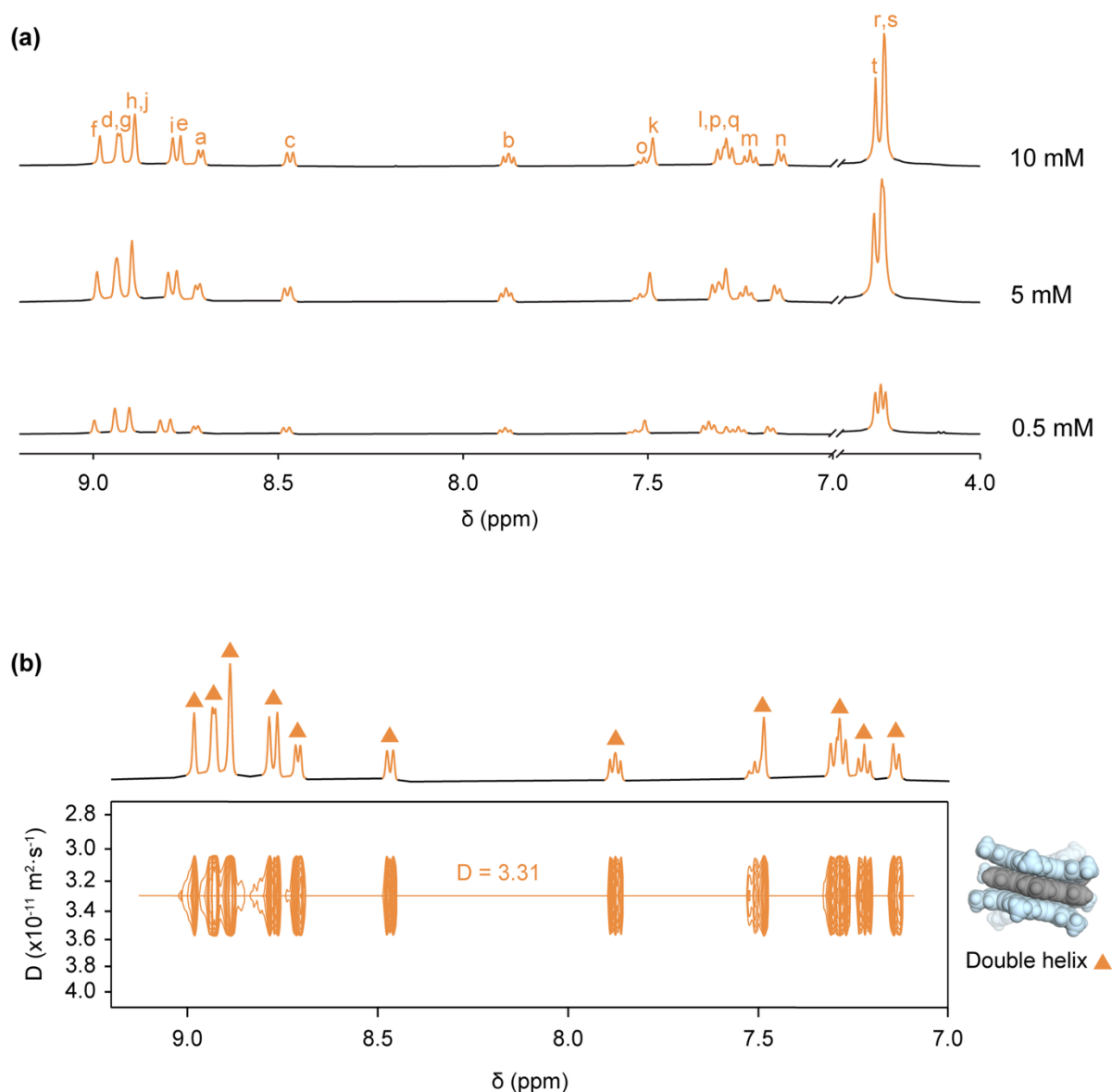


Figure S9 | Concentration-dependent ¹H and DOSY NMR spectra of double helix (1)₂·TFA. **a**, Concentration-dependent ¹H NMR spectra of 1·TFA between 0.5 mM and 10 mM in D₂O (500 MHz, 278 K). **b**, ¹H DOSY NMR spectrum (500 MHz, D₂O, 278 K) of the double helix assembled from 1·TFA (10 mM) in D₂O. The signals corresponding to the double helix are labelled with orange triangles. Its hydrodynamic radius $r_H = 8.4 \text{ \AA}$ was calculated using the Stokes-Einstein equation.

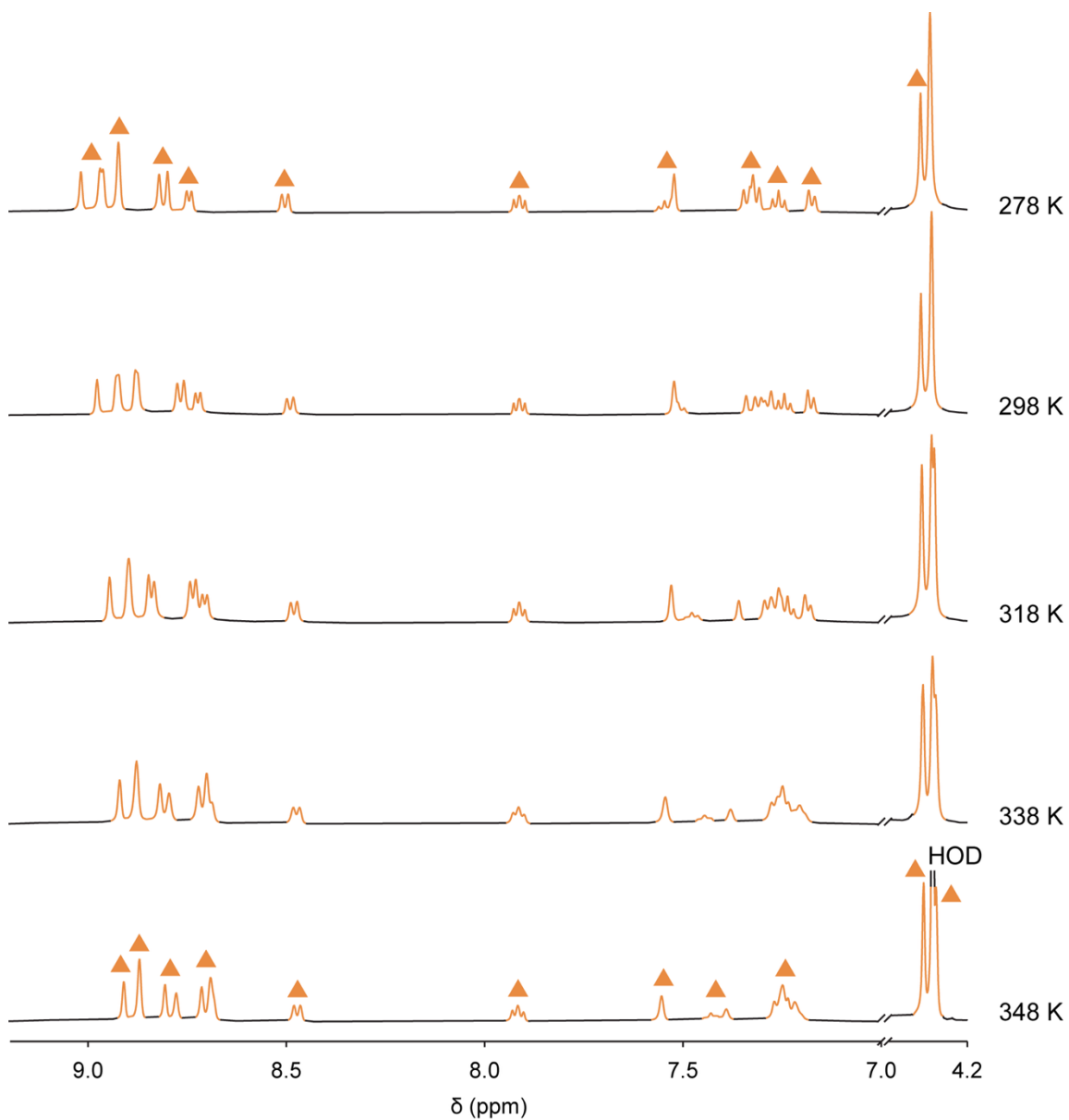


Figure S10 | Variable temperature ^1H NMR spectra (500 MHz) of a 10 mM solution of 1-TFA in D_2O between 278 K and 348 K. The signals corresponding to the double helix are labelled with orange triangles.

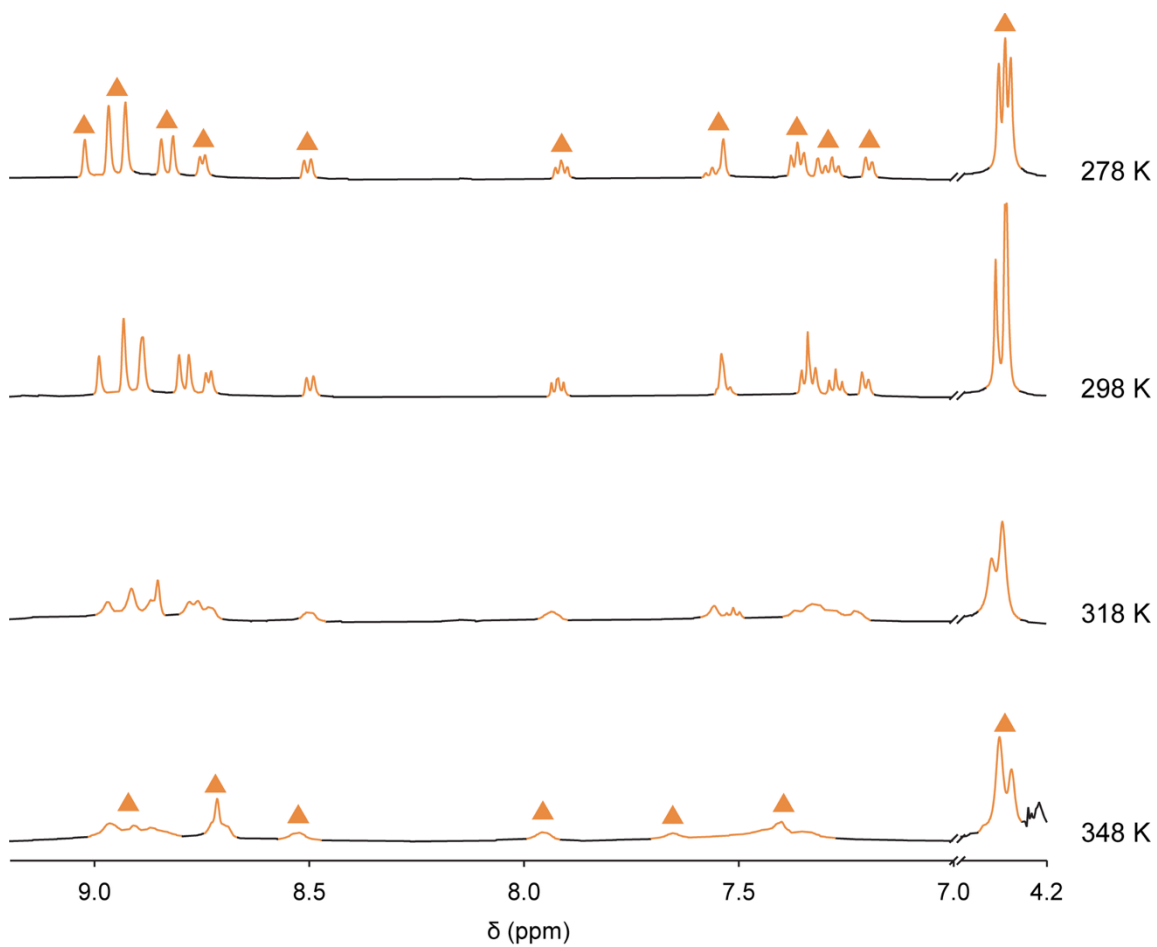


Figure S11 | Variable temperature ^1H NMR spectra (500 MHz) of a 0.5 mM solution of 1·TFA in D_2O between 278 K and 348 K. The signals corresponding to the double helix are labelled with orange triangles.

5. Host-guest studies with double helix (1)₂·TFA

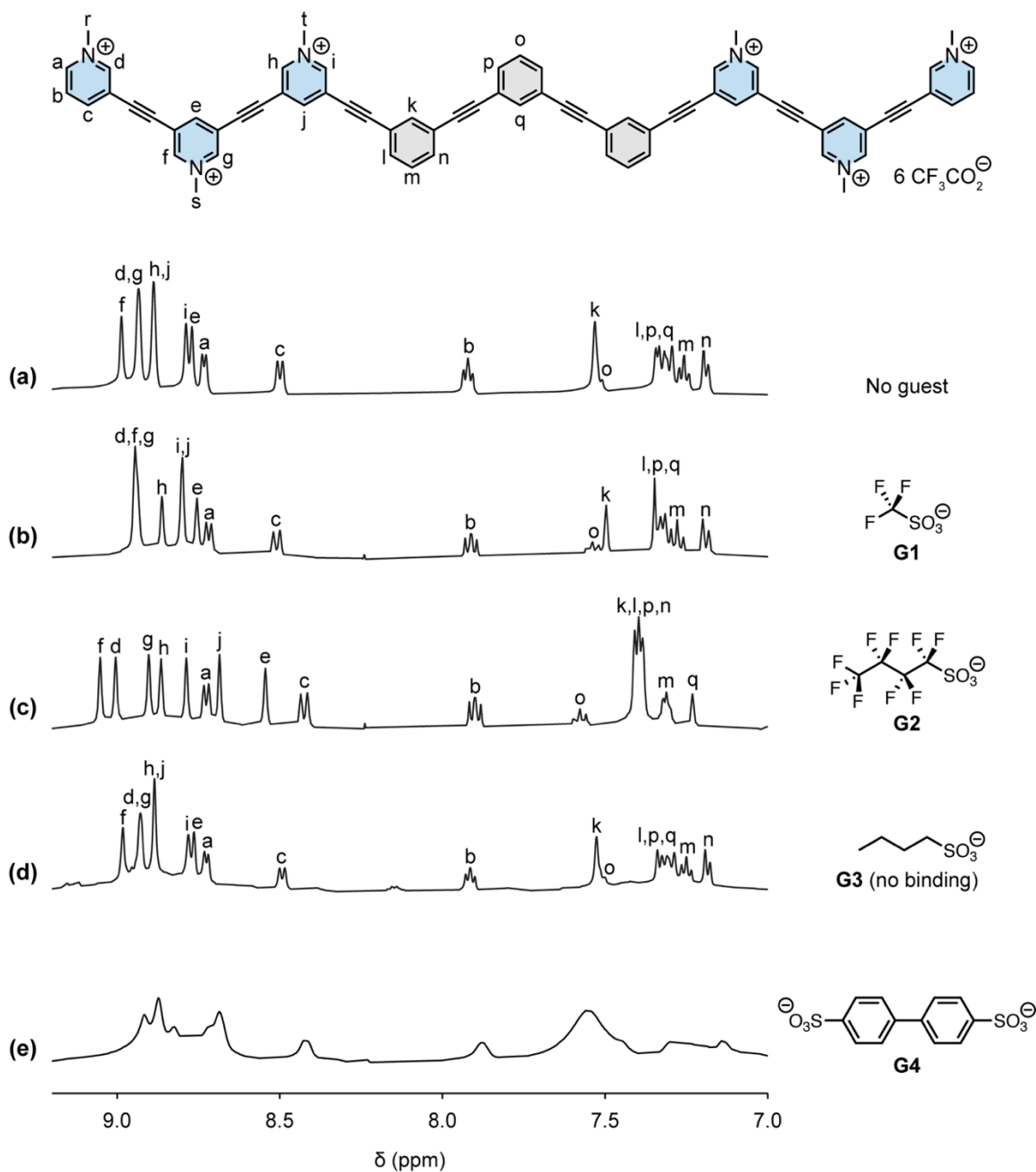


Figure S12 | ¹H NMR spectra of 1·TFA aromatic region (5 mM in D₂O, 500 MHz, 298 K) after addition of different guests. a, No guest added. b, After addition of 6.0 equiv. of potassium trifluoromethanesulfonate **G1. c, After addition of 3.0 equiv. of potassium perfluorobutanesulfonate **G2**. d, After addition of 3.0 equiv. of sodium butanesulfonate **G3**. e, After addition of 2.0 equiv. of sodium 4,4'-biphenyldisulfonate **G4** (the sample precipitates above 2 equiv. of **G4**).**

5.1. Titration with potassium trifluoromethanesulfonate (G1)

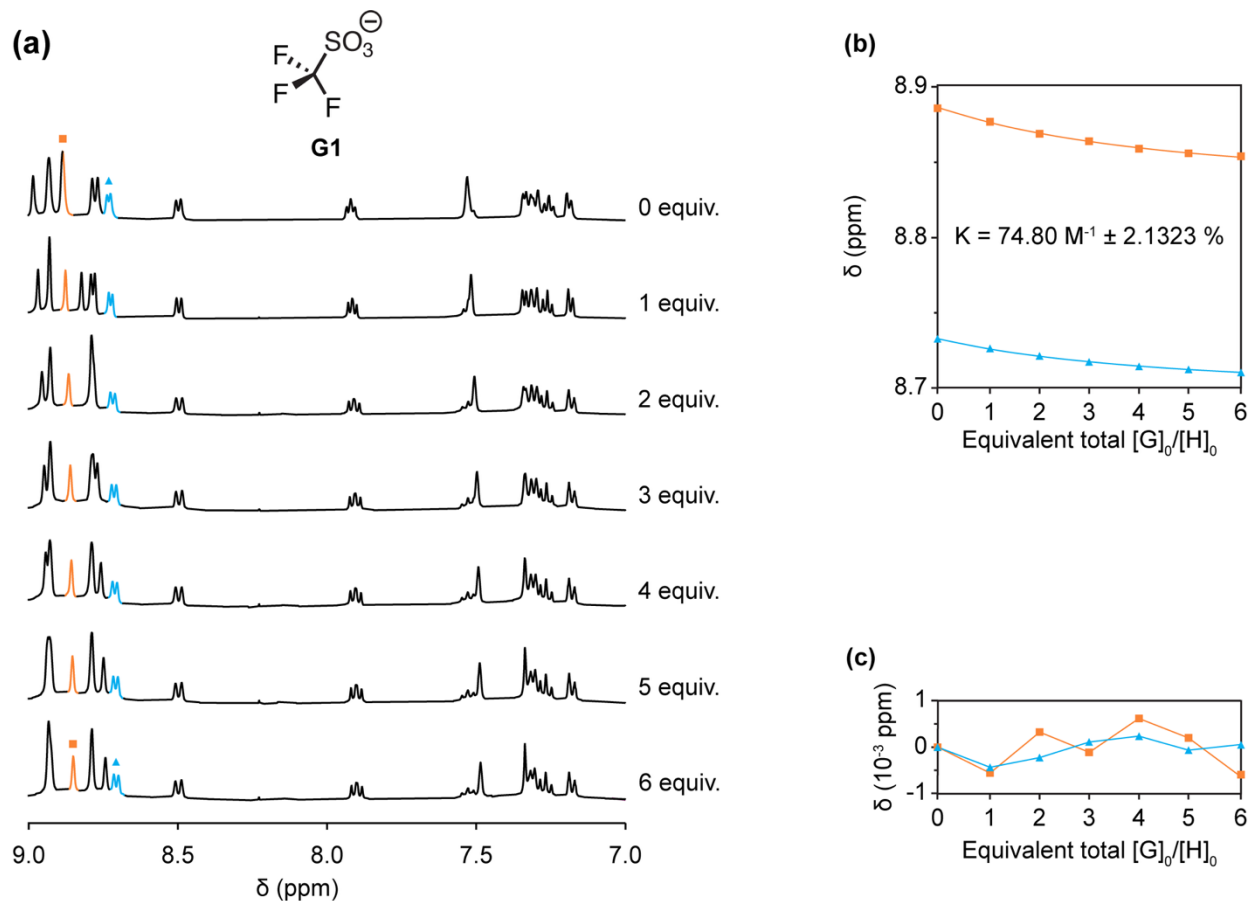


Figure S13 | ^1H NMR titration (400 MHz, 298 K) of potassium trifluoromethanesulfonate G1 into a 5 mM-solution of 1·TFA in D_2O . **a**, Selected region of the ^1H NMR spectra recorded during the titration. **b**, Chemical shift changes recorded as a function of guest equivalents. **c**, Residuals from the curve fitting. Association constant was obtained by fitting the data with *BindFit*.^{S2}

5.2. Titration with potassium perfluorobutanesulfonate (G2)

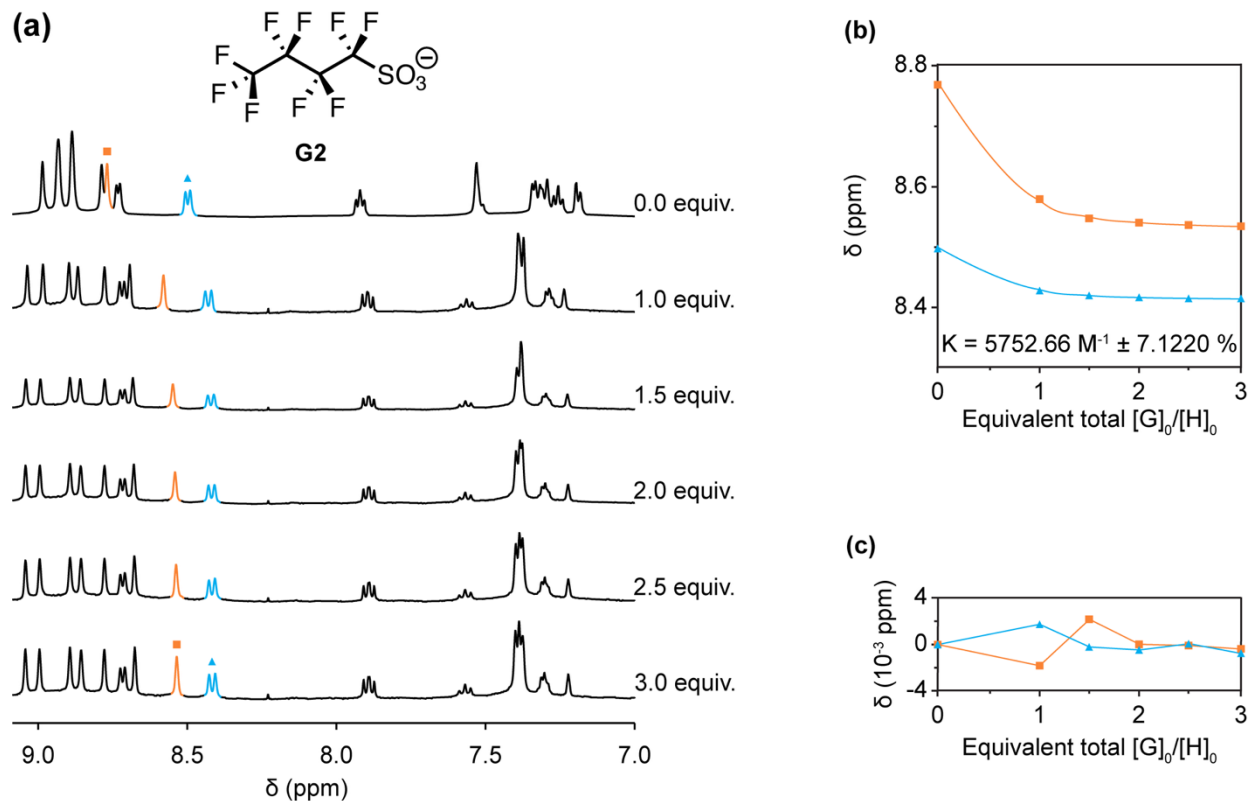
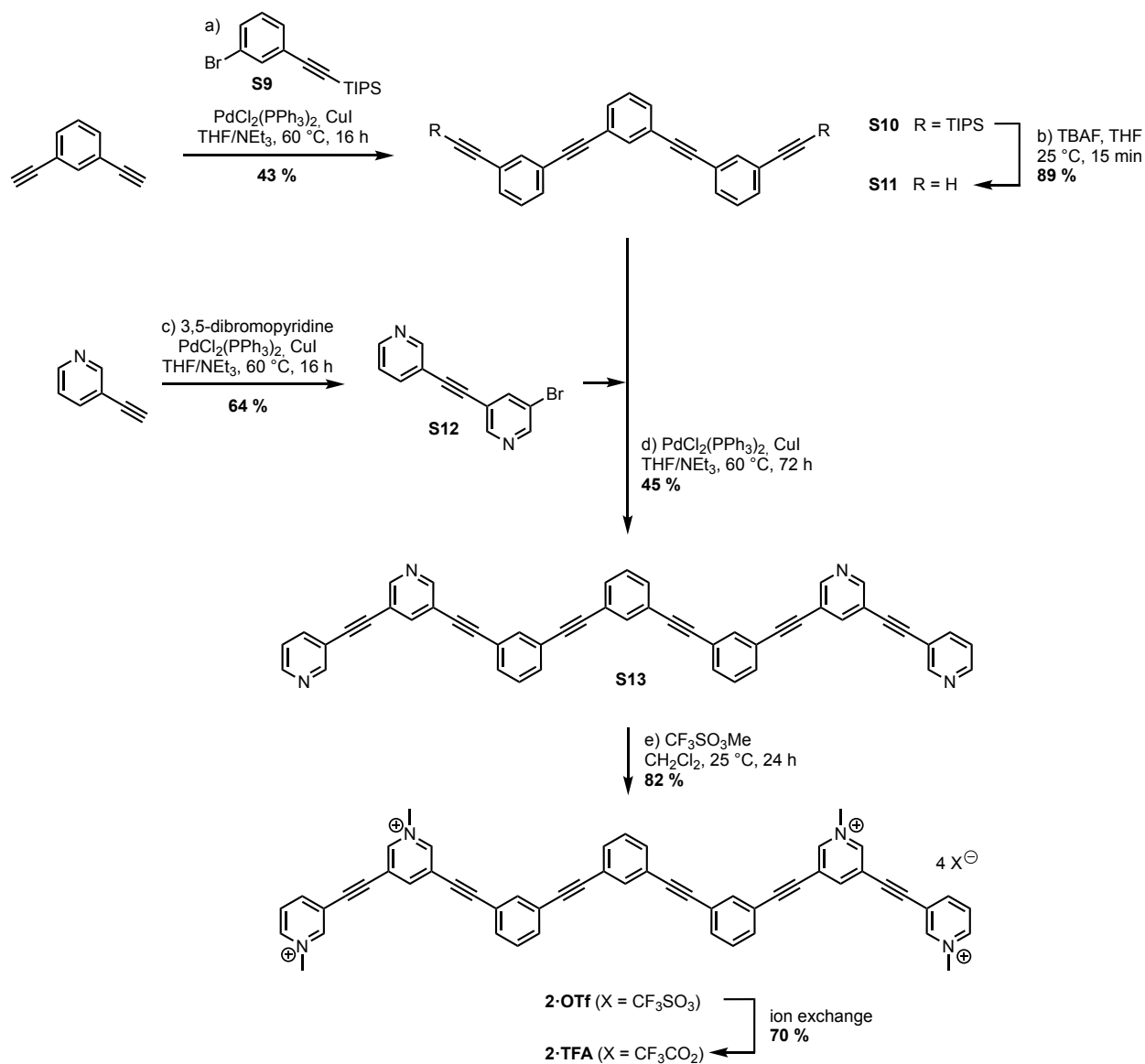


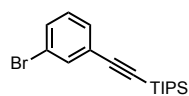
Figure S14 | ^1H NMR titration (400 MHz, 298 K) of potassium perfluorobutanesulfonate G2 into a 5 mM-solution of 1·TFA in D_2O . **a**, Selected region of the ^1H NMR spectra recorded during the titration. **b**, Chemical shift changes recorded as a function of guest equivalents. **c**, Residuals from the curve fitting. Association constant was obtained by fitting the data with *BindFit*.^{S2}

6. Synthesis of strand 2



Scheme S2 | General procedure for the synthesis of strand 2.

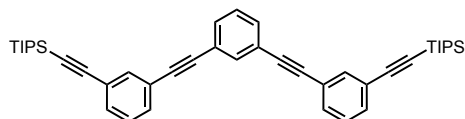
Compound S9



The synthesis of **S9** was adapted from a published procedure.^{S3} 3-bromoiodobenzene (735 μL , 5.7 mmol), (triisopropylsilyl)acetylene (1.4 mL, 6.4 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (206 mg, 0.3 mmol), and CuI (30 mg, 0.2 mmol) were

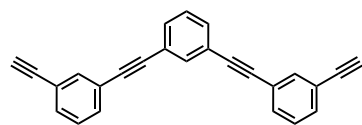
dissolved in dry triethylamine (18 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred 16 h at 25 °C, concentrated by rotary evaporation, and the resulting residue was purified by silica gel column chromatography (100% cyclohexane) to yield **S9** (1.9 g, 97 %) as a colorless oil. Spectral data closely match those previously reported in the literature.^{S3}

Compound S10



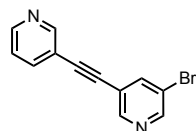
The synthesis of **S10** was adapted from a published procedure.^{S4} 1,3-diethynylbenzene (263 μ L, 2.0 mmol), **S9** (1.5 g, 4.4 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (139 mg, 0.2 mmol), and CuI (20 mg, 0.1 mmol) were dissolved in dry triethylamine (10 mL) and tetrahydrofuran (10 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred 16 h at 60 °C, concentrated by rotary evaporation, and the resulting residue was purified by silica gel column chromatography (gradient hexane/dichloromethane = 100/0 to 90/10) to yield **S10** (551 mg, 43 %) as a colorless oil. Spectral data closely match those previously reported in the literature.^{S4}

Compound S11



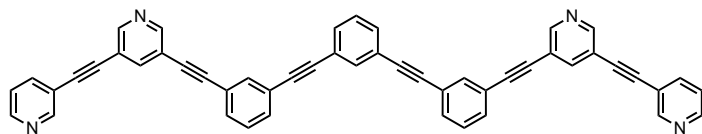
The synthesis of **S11** was adapted from a published procedure.^{S4} Tetrabutylammonium fluoride (1 M in tetrahydrofuran, 4.6 mL) was added to a solution of **S10** (829 mg, 1.3 mmol) in tetrahydrofuran (150 mL). The reaction mixture was stirred for 15 min at 25 °C and concentrated by rotary evaporation. The crude product was first washed with plenty of CH_3OH . The residue was then redissolved in dichloromethane, washed with water, dried over Na_2SO_4 , filtered and concentrated by rotary evaporation to yield **S11** (376 mg, 89 %) as a white solid. Spectral data closely match those previously reported in the literature.^{S4}

Compound S12



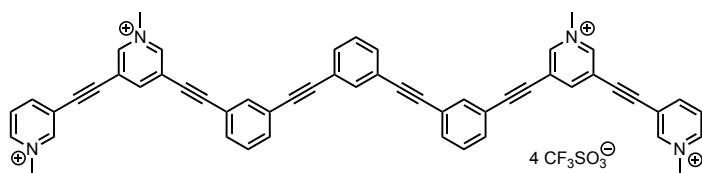
3-Ethynylpyridine (1.0 g, 9.8 mmol), 3,5-dibromopyridine (3.0 g, 12.8 mmol), $\text{PdCl}_2(\text{PPh}_3)_2$ (341 mg, 0.5 mmol) and CuI (53 mg, 0.3 mmol) were dissolved in dry triethylamine (35 mL) and tetrahydrofuran (43 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred 16 h at 60 °C, concentrated by rotary evaporation, and the resulting residue was purified by silica gel column chromatography (gradient hexane/ethyl acetate = 100/0 to 50/50) to yield **S12** (1.6 g, 64 %) as a white solid. Melting point: 91-92 °C. ^1H NMR (400 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K): δ 8.78 (br d, J = 1.5 Hz, 1H), 8.68 (d, J = 2.0 Hz, 1H), 8.65 (d, J = 2.0 Hz, 1H), 8.61 (br dd, J = 4.9, 1.5 Hz, 1H), 7.99 (t, J = 2.0 Hz, 1H), 7.83 (dt, J = 7.9, 2.0 Hz, 1H), 7.33 (ddd, J = 7.9, 4.9, 0.7 Hz, 1H). ^{13}C NMR (100 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K): δ 152.5, 150.4, 150.3, 149.5, 140.9, 138.8, 123.3, 121.2, 120.3, 119.4, 90.5, 87.7. HR-MS (ESI+) m/z calculated for $\text{C}_{12}\text{H}_7\text{BrN}_2^+$ $[\text{M}]^+ = 257.9793$, found 257.9798, Δ = -2.02 ppm.

Compound S13



Compounds **S11** (395 mg, 1.2 mmol), **S12** (624 mg, 2.4 mmol), PdCl₂(PPh₃)₂ (86 mg, 0.1 mmol) and CuI (14 mg, 0.07 mmol) were dissolved in dry triethylamine (10 mL) and tetrahydrofuran (10 mL) at 0 °C under nitrogen atmosphere. The reaction mixture was stirred for 72 h at 60 °C and concentrated by rotary evaporation. The solid residue was dissolved in dichloromethane, washed with a saturated solution of Na₂CO₃, dried over Na₂SO₄, filtered, and concentrated by rotary evaporation. The resulting solid was washed several times with petroleum ether and methanol to yield **S13** (372 mg, 45 %) as a beige solid. Melting point: >400 °C (decomposition). ¹H NMR (400 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K): δ 8.79 (br d, *J* = 1.5 Hz, 2H), 8.71 (d, *J* = 2.0 Hz, 2H), 8.70 (d, *J* = 2.0 Hz, 2H), 8.60 (br dd, *J* = 5.0, 1.5 Hz, 2H), 7.96 (t, *J* = 2.0 Hz, 2H), 7.83 (dt, *J* = 8.0, 2.0 Hz, 2H), 7.73 (t, *J* = 1.5 Hz, 1H), 7.72 (t, *J* = 1.5 Hz, 2H), 7.55-7.49 (m, 6H), 7.38 (t, *J* = 8.0 Hz, 2H), 7.34 (t, *J* = 8.0 Hz, 1H), 7.32 (ddd, *J* = 8.0, 5.0, 1.0 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K): δ 152.5, 151.4, 151.1, 149.4, 140.8, 138.7, 134.9, 134.8, 132.2, 131.7, 128.8, 128.7, 123.7, 123.4, 123.3, 122.7, 120.1, 119.6, 92.7, 89.9, 89.4, 89.0, 88.5, 85.8. HR-MS (ESI+) *m/z* calculated for C₅₀H₂₇N₄⁺ [M+H]⁺ = 683.2231, found 683.2238, Δ = -1,02 ppm.

Strand 2



Methyl triflate (154 μL, 1.4 mmol) was added to a solution of **S13** (169 mg, 0.3 mmol) in dry dichloromethane (120 mL) under nitrogen atmosphere. The mixture was stirred 24 h at 25 °C before being concentrated by rotary evaporation. The crude product was washed with dichloromethane to yield **2·OTf** (270 mg, 82 %) as a beige solid. Melting point: >400 °C (decomposition). ¹H NMR (500 MHz, CD₃CN, 298 K): δ 9.01 (br s, 2H), 8.94 (br s, 2H), 8.91 (br s, 2H), 8.74 (br s, 2H), 8.71 (br d, *J* = 6.3 Hz, 2H), 8.65 (br d, *J* = 8.2 Hz, 2H), 8.99 (br t, *J* = 7.2 Hz, 2H), 7.84 (br t, *J* = 1.7 Hz, 2H), 7.77 (br t, *J* = 1.6 Hz, 1H), 7.71 (br dt, *J* = 7.8, 1.3 Hz, 2H), 7.69 (br dt, *J* = 7.9, 1.4 Hz, 2H), 7.62 (br dd, *J* = 7.8, 1.7 Hz, 2H), 7.55 (t, *J* = 7.8 Hz, 2H), 7.49 (t, *J* = 7.8 Hz, 1H), 4.37 (s, 6H), 4.36 (s, 6H). ¹³C NMR (126 MHz, CD₃CN, 298 K): δ 149.6, 149.1, 148.6, 148.5, 147.7, 146.9, 135.6, 135.2, 134.5, 133.2, 132.9, 130.6, 130.3, 129.4, 125.4, 124.6, 124.1, 123.4, 123.1, 122.1, 98.1, 90.2, 89.8, 89.4, 88.6, 82.6, 50.0, 49.8. HR-MS (ESI+) *m/z* calculated for C₅₇H₃₈F₉N₄O₉S₃⁺ [M+3CF₃SO₃]⁺ = 1189.1652, found 1189.1686, Δ = -2.86 ppm.

Ion exchange was performed on preparative HPLC. Strand **2**·TFA (169 mg, 70 %) was isolated as a beige solid. HR-MS (ESI+) m/z calculated for $C_{60}H_{38}F_9N_4O_6^+ [M+3CF_3CO_2]^+ = 1081.2643$, found 1081.2623, $\Delta = 1.85$ ppm.

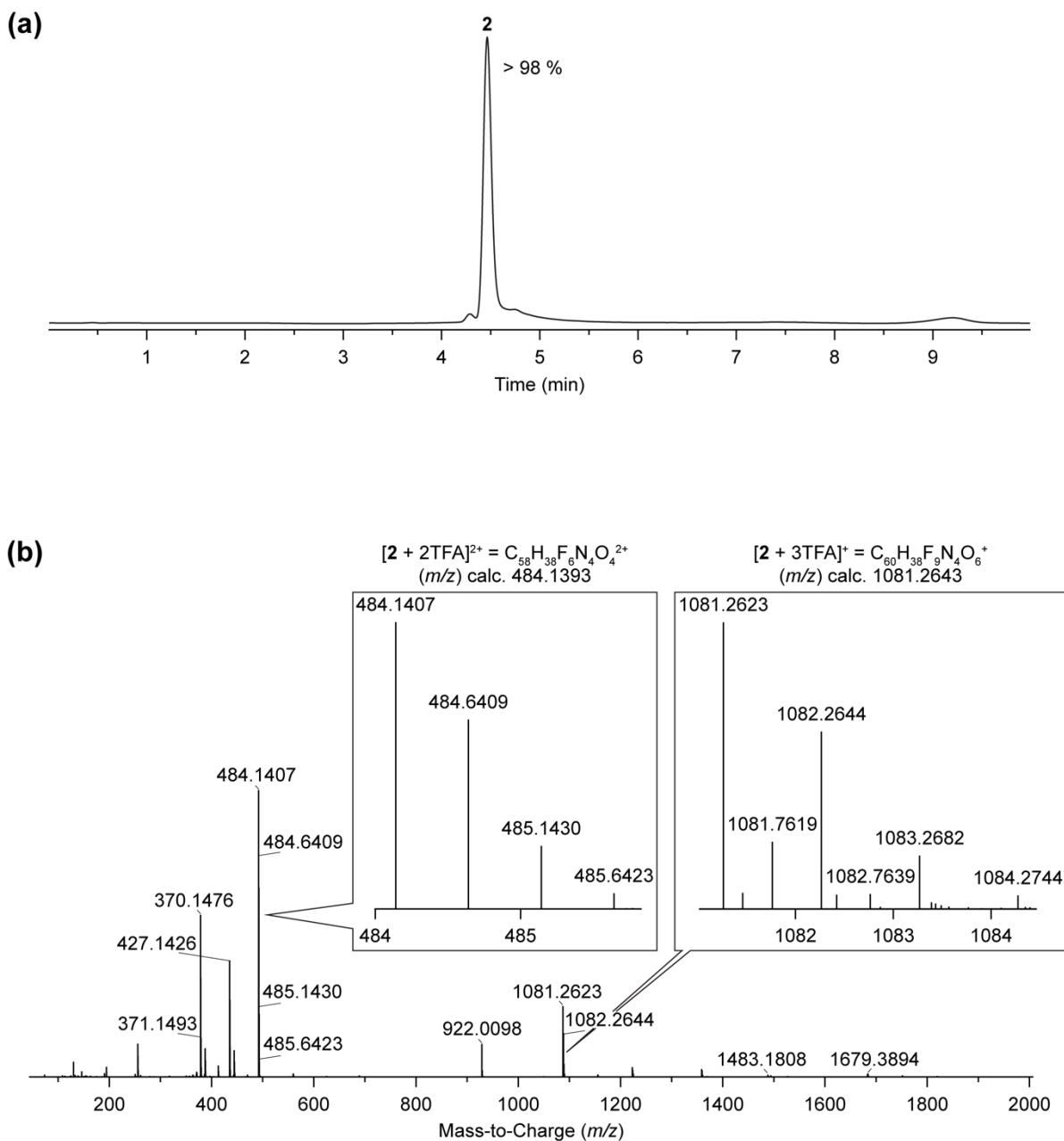


Figure S15 | UHPLC-MS characterization of strand 2. Tandem **a**, Reverse-phase UHPLC chromatogram and **b**, MS spectrum of strand 2. Absorbance was recorded at 254 nm.

7. NMR Characterization of unfolded strand 2 in CD₃CN

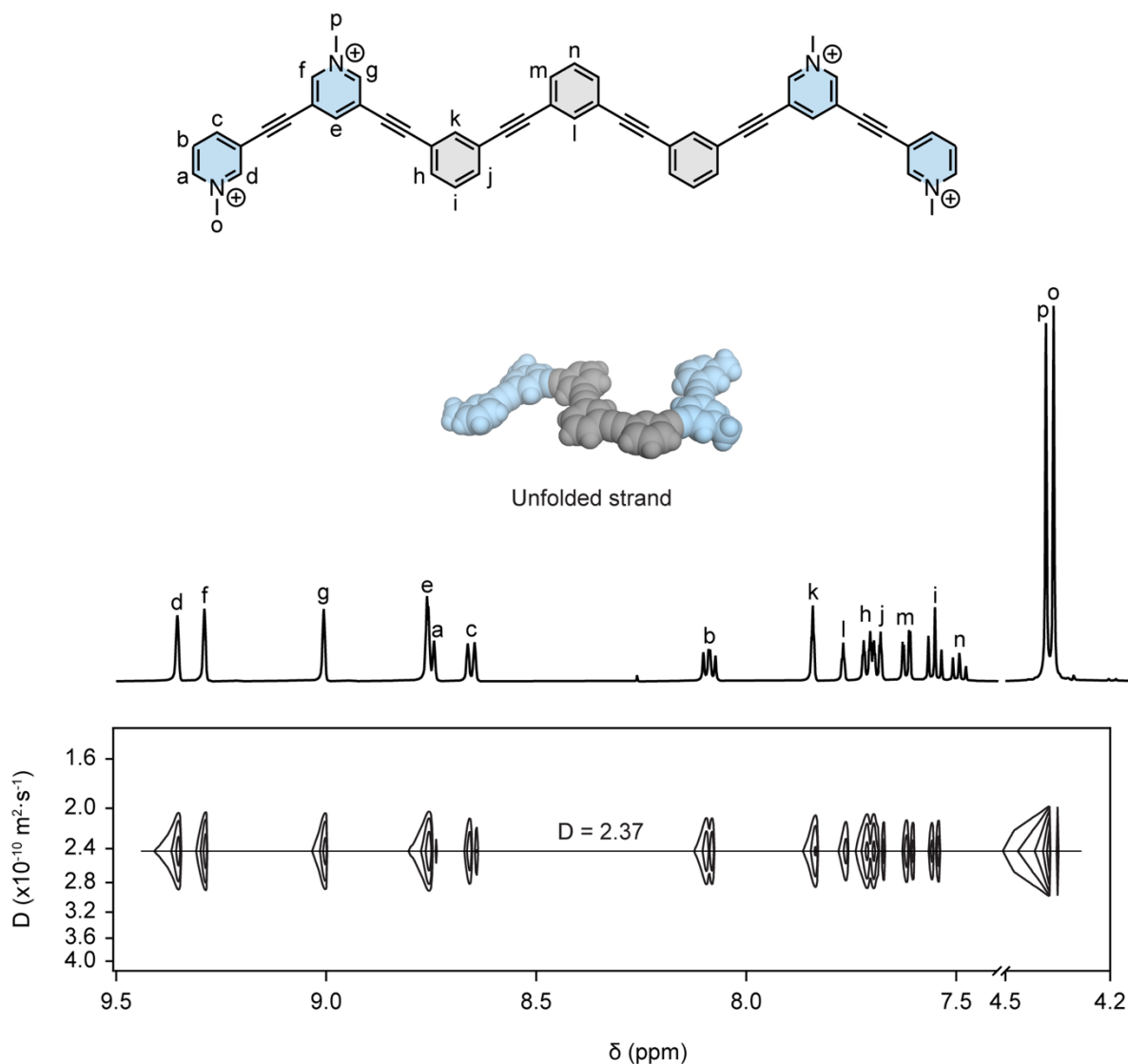


Figure S16 | Expansion of ¹H and DOSY NMR spectra of unfolded strand 2·TFA in CD₃CN (5 mM, 500 MHz, 298 K). Similar experiment was performed on strand 2·OTf. The hydrodynamic radii of unfolded strand 2·OTf and unfolded strand 2·TFA were found to be $r_H = 10.3 \text{ \AA}$ and $r_H = 10.8 \text{ \AA}$, respectively, using the Stokes-Einstein equation.

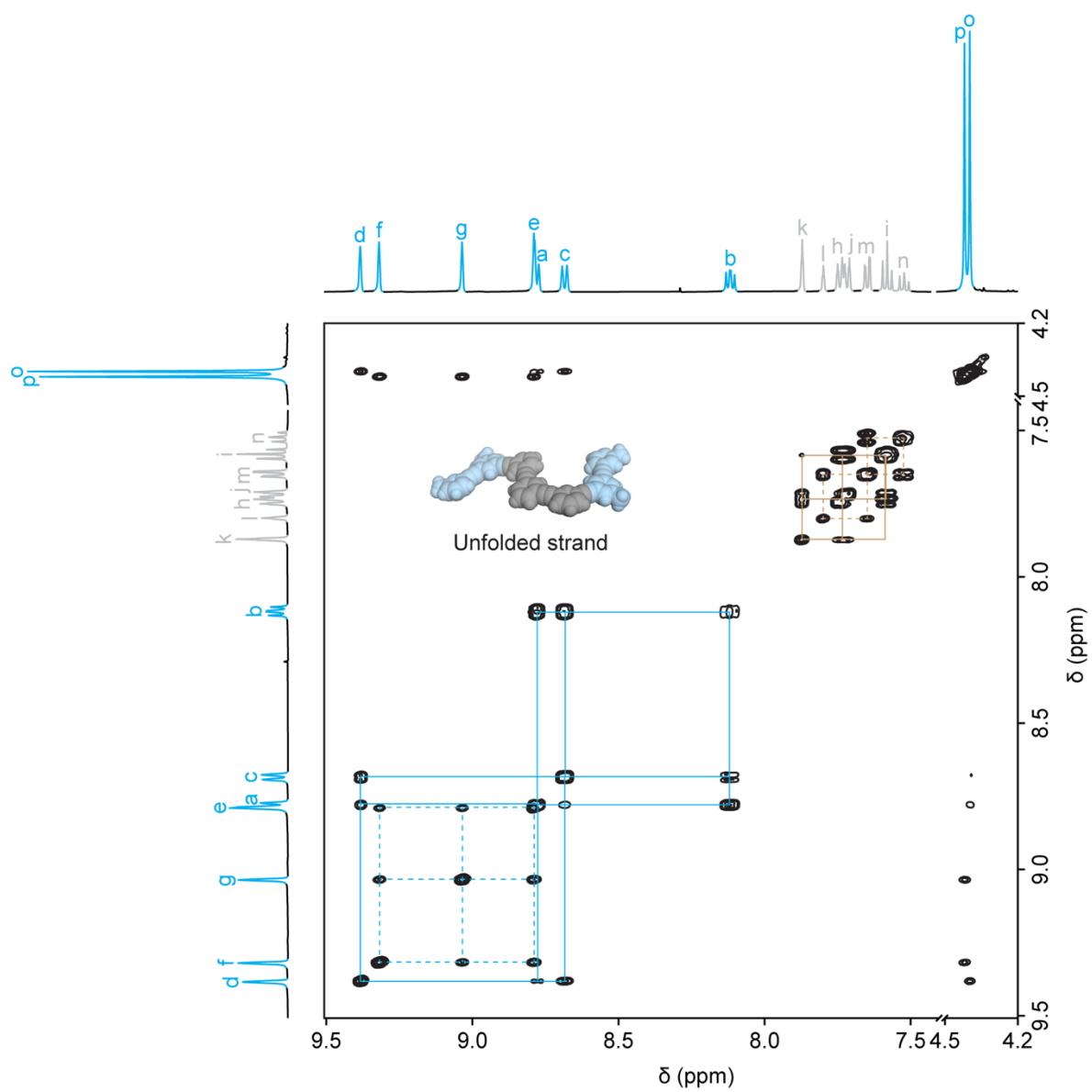


Figure S17 | Expansion of selected regions in 2D ^1H - ^1H COSY NMR spectrum of unfolded strand 2·TFA in CD_3CN (5 mM, 500 MHz, 298 K).

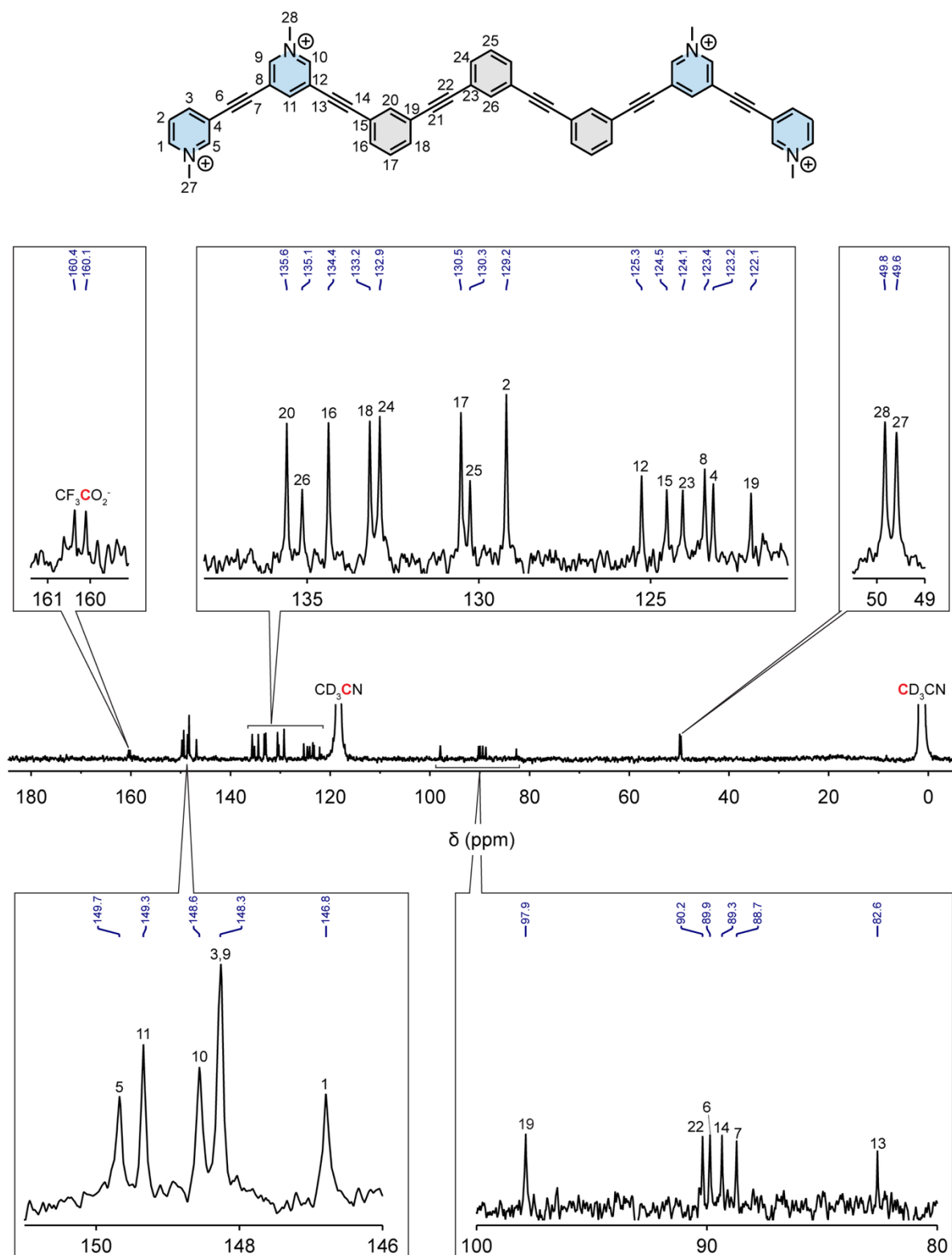


Figure S18 | ^{13}C NMR spectrum of unfolded strand 2·TFA in CD_3CN (5 mM, 126 MHz, 298 K).

8. NMR characterization of double helix (2)₂ and triple helix (2)₃ in D₂O

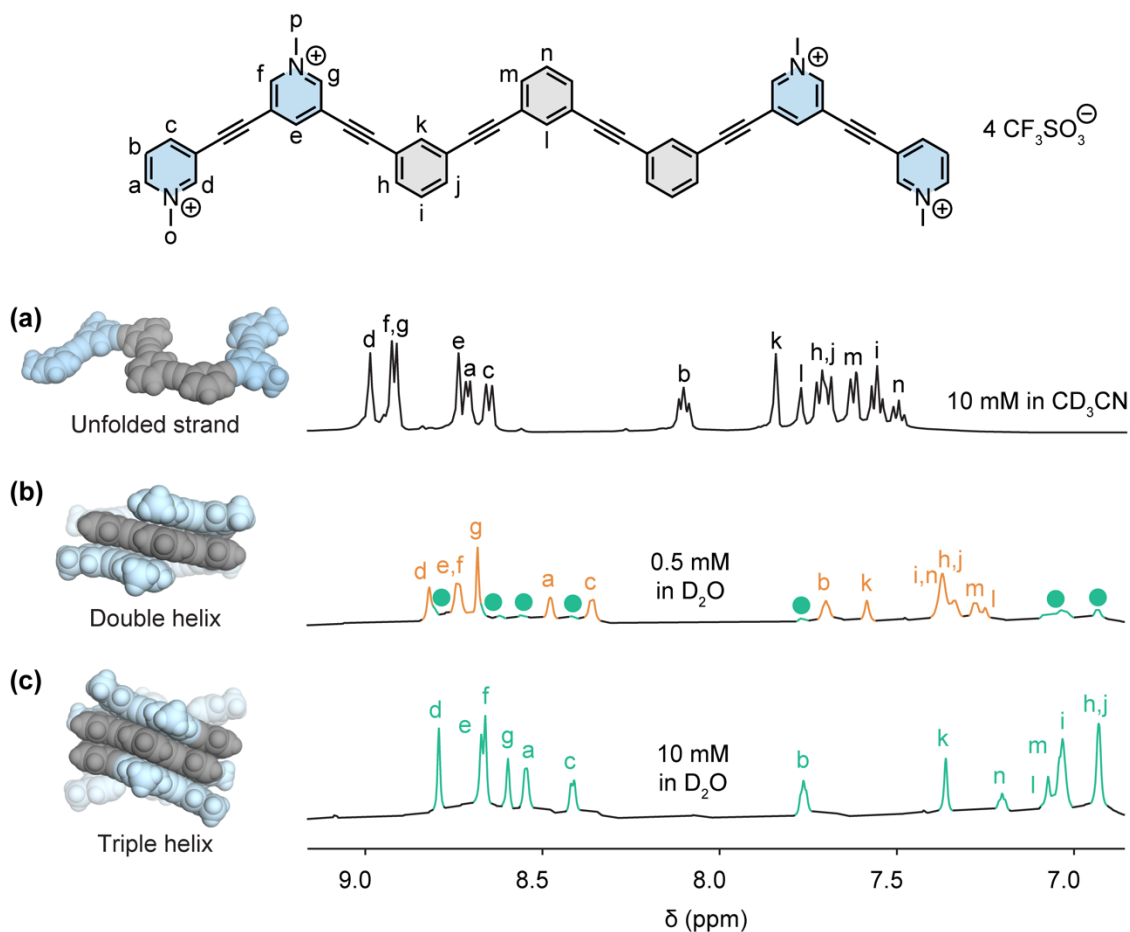


Figure S19 | Solvent-dependent ^1H NMR spectra of strand 2. Comparison between the ^1H NMR spectra of **a**, unfolded strand $2 \cdot \text{OTf}$ in CD_3CN (10 mM, 500 MHz, 298 K); **b**, double helix $(2)_2 \cdot \text{OTf}$ in D_2O (0.5 mM, 800 MHz, 288 K); and **c**, triple helix $(2)_3 \cdot \text{OTf}$ in D_2O (10 mM, 800 MHz, 288 K). In spectrum **b**, the signals corresponding to the triple helix are labelled with green circles.

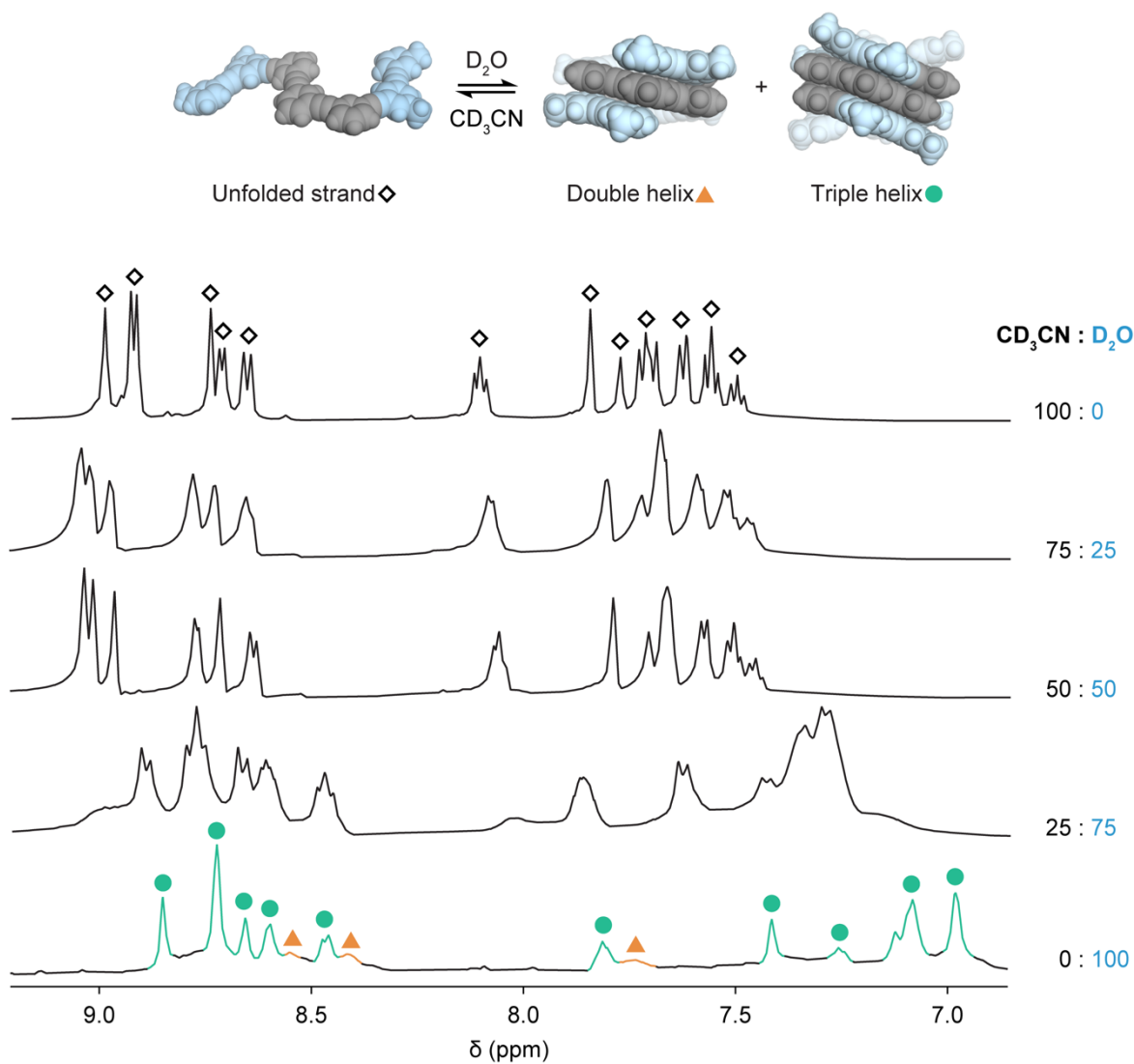


Figure S20 | ^1H NMR spectra of 2·OTf aromatic protons in CD_3CN / D_2O mixtures (5 mM, 500 MHz, 298 K). The signals corresponding to the unfolded strand, double helix and triple helix are labelled with empty diamonds, orange triangles and green circles, respectively.

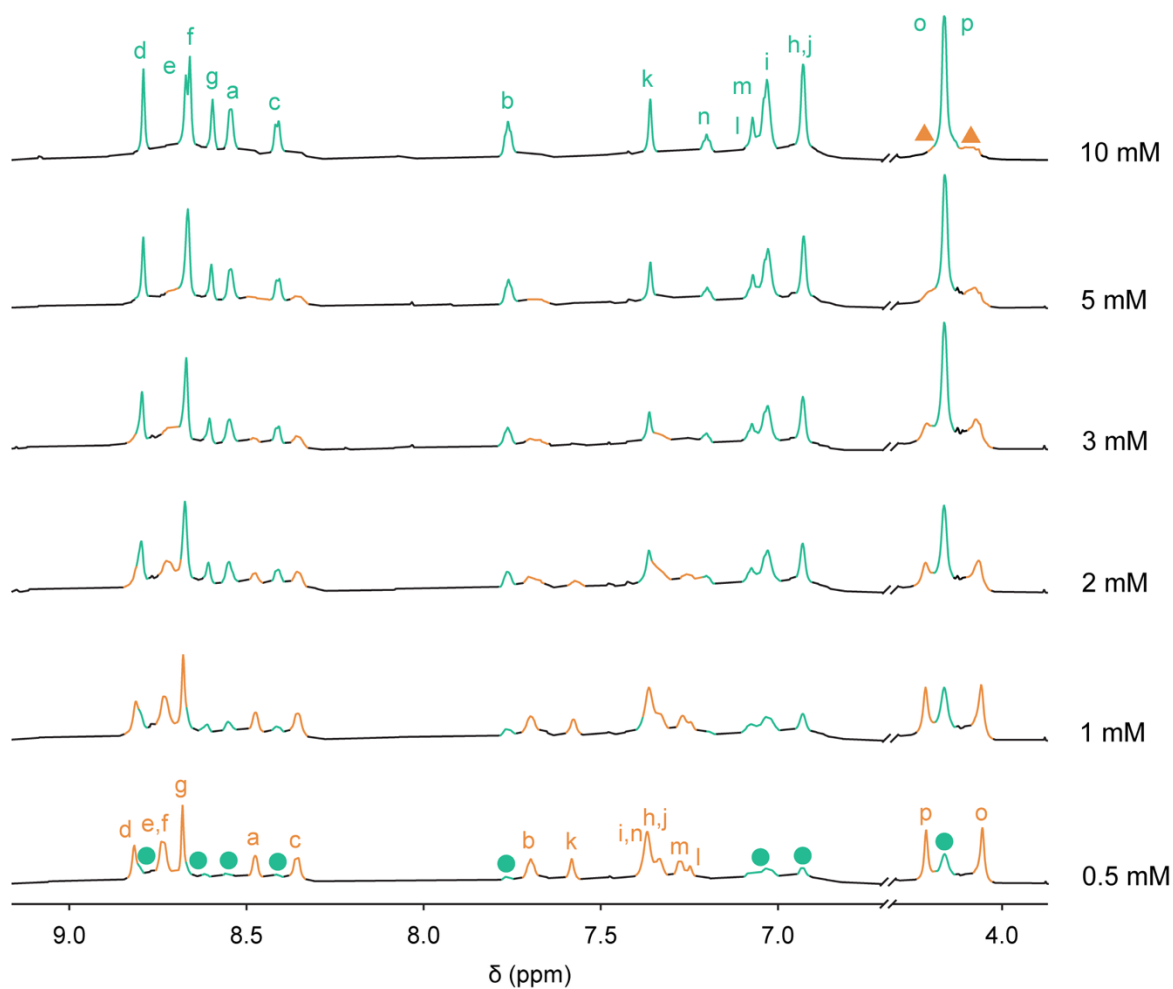


Figure S21 | Concentration-dependent ^1H NMR spectra of 2·OTf between 0.5 mM and 10 mM in D_2O (800 MHz, 288 K). The triple helix is the main species observed at high concentration and dissociates into the double helix upon dilution.

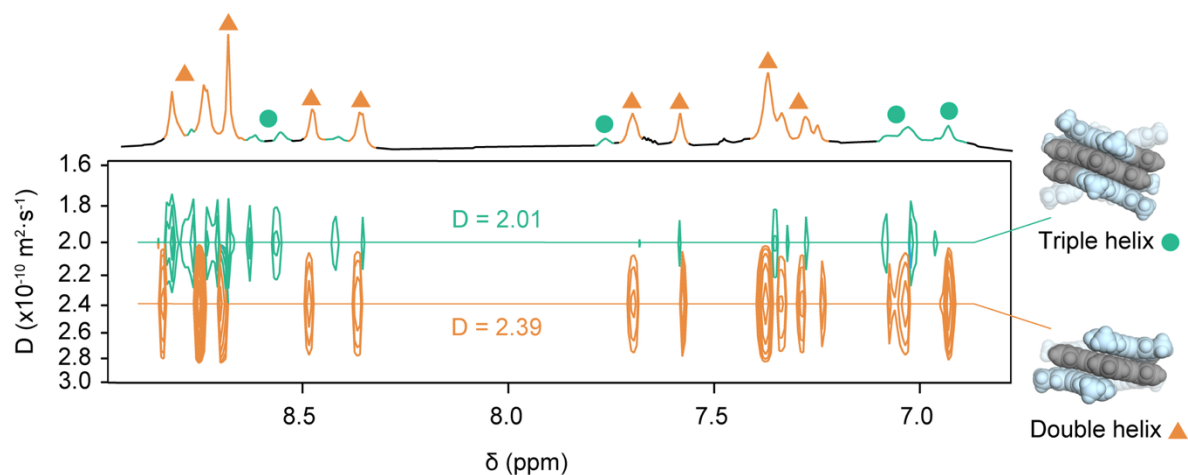


Figure S22 | ^1H DOSY NMR spectrum (800 MHz, D_2O , 288 K) of the double and triple helices assembled from $2\cdot\text{OTf}$ (0.5 mM) in D_2O . The signals corresponding to the double and triple helices are labelled with orange triangles and green circles, respectively. The hydrodynamic radii of the double and triple helices were found to be $r_{\text{H}} = 8.1 \text{ \AA}$ and $r_{\text{H}} = 9.4 \text{ \AA}$, respectively, using the Stokes-Einstein equation.

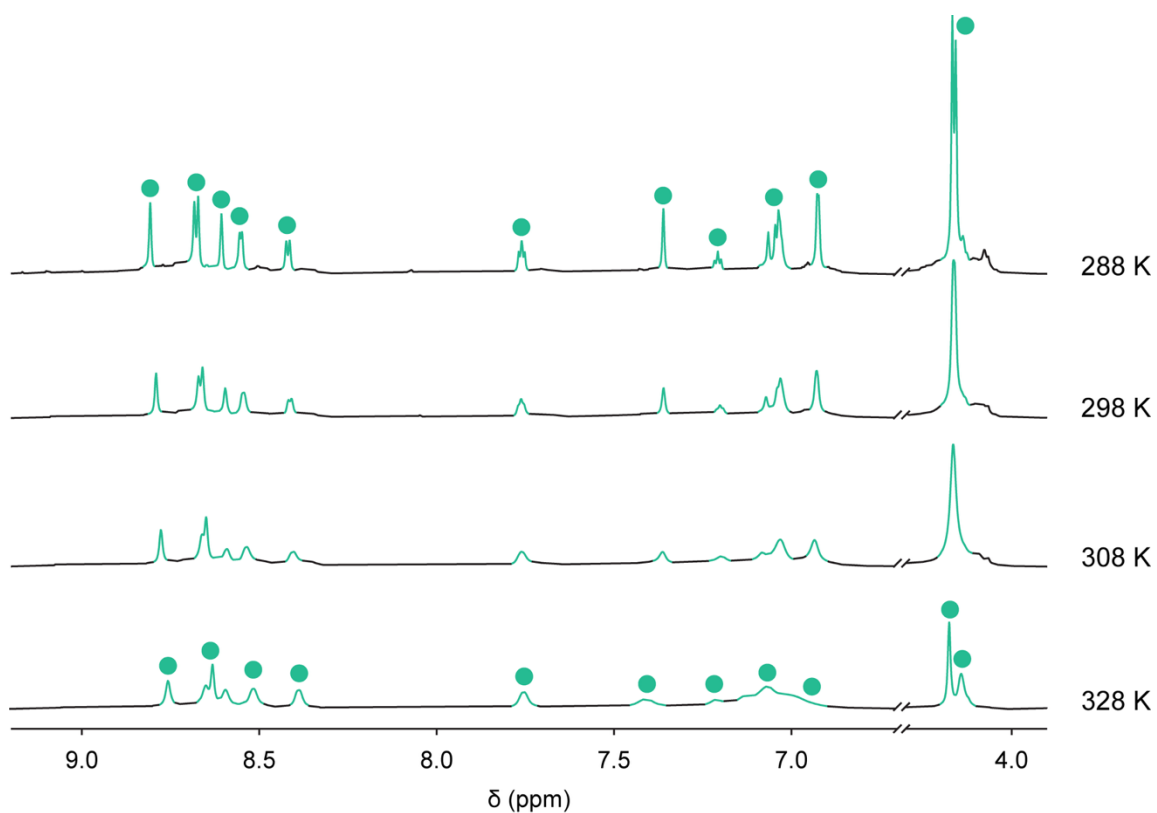


Figure S23 | Variable temperature ^1H NMR spectra of 2·OTf between 288 K and 328 K in D_2O (800 MHz, 10 mM). The triple helix (signals labelled with green circles) remains intact at high temperature when strand concentration is sufficiently high.

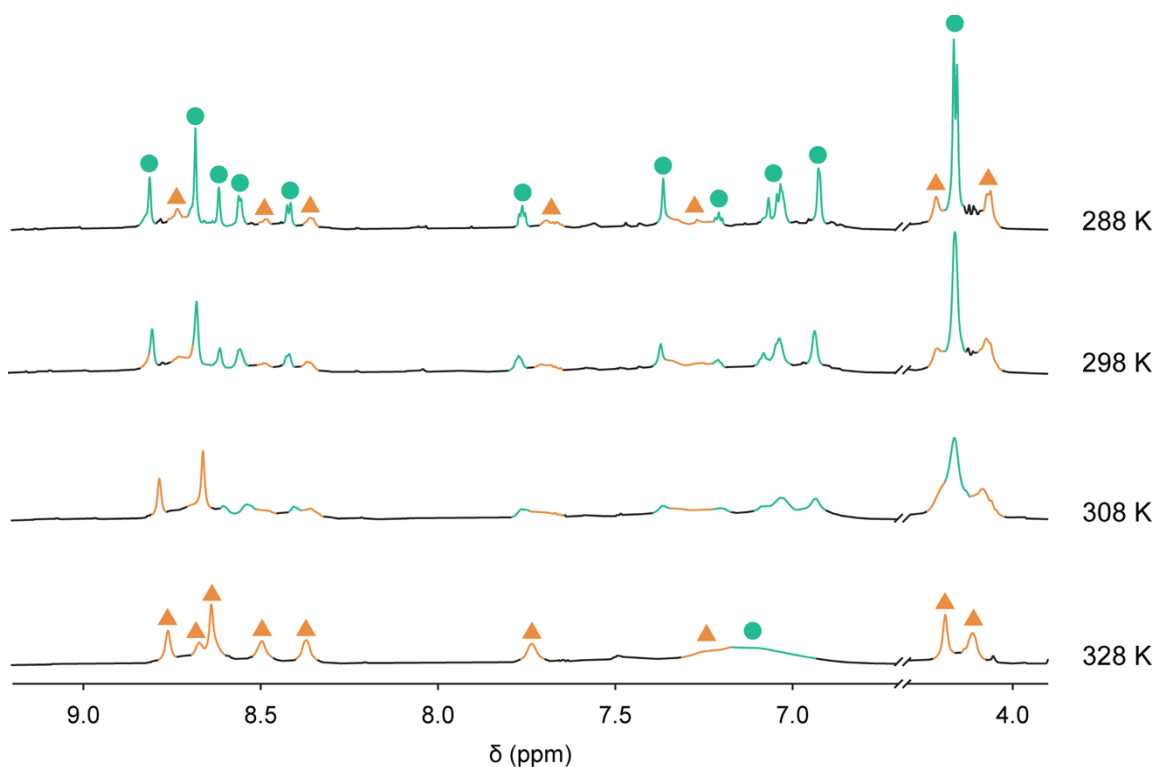


Figure S24 | Variable temperature ^1H NMR spectra of 2·OTf between 288 K and 328 K in D_2O (800 MHz, 3 mM). The spectra show the temperature-dependent interconversion between double and triple helix (signals labelled with orange triangles and green circles, respectively) at low strand concentration.

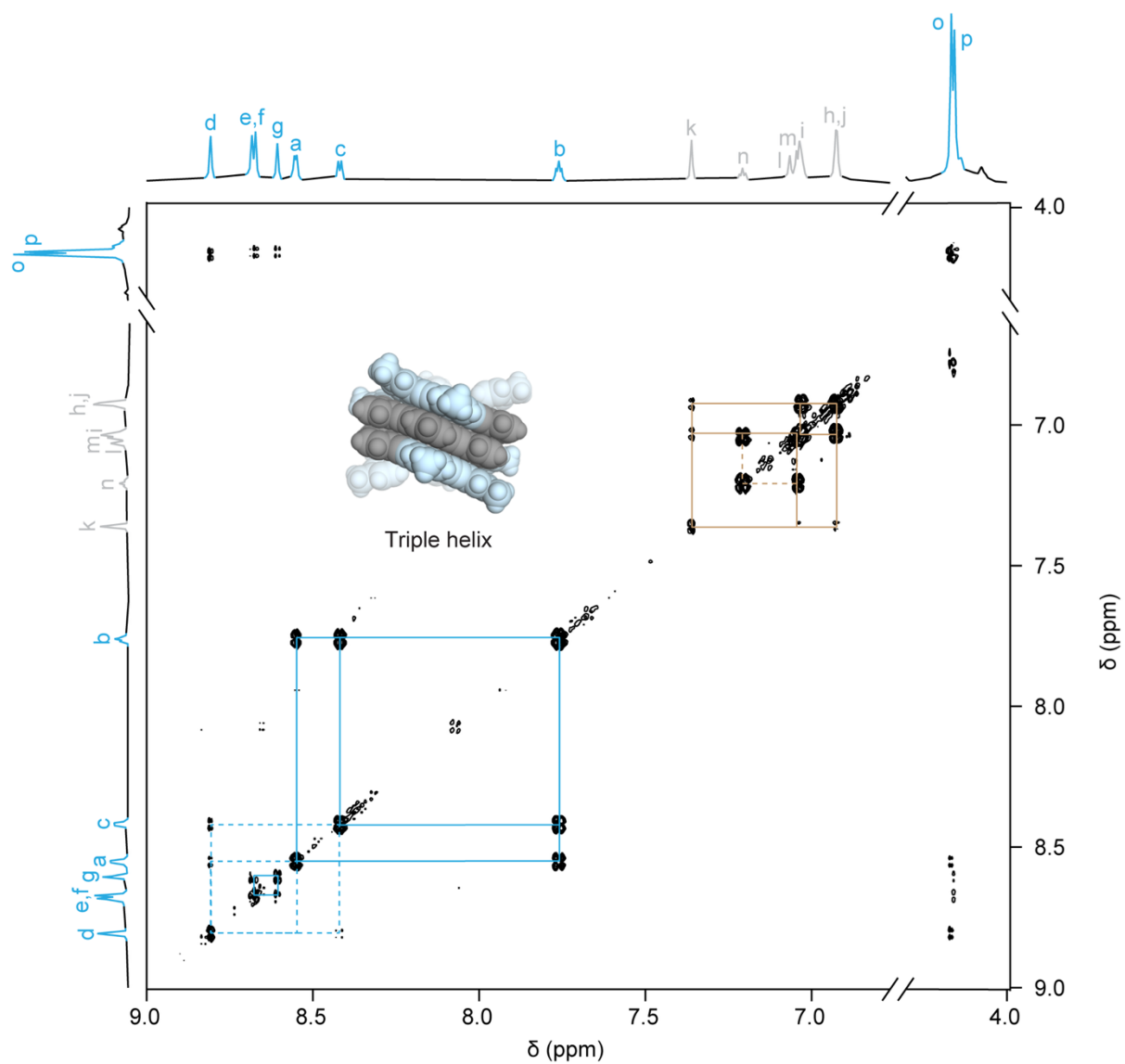


Figure S25 | 2D ^1H - ^1H COSY from selected spectral regions of triple helix $(2)_3 \cdot \text{OTf}$ in D_2O (10 mM, 800 MHz, 288 K).

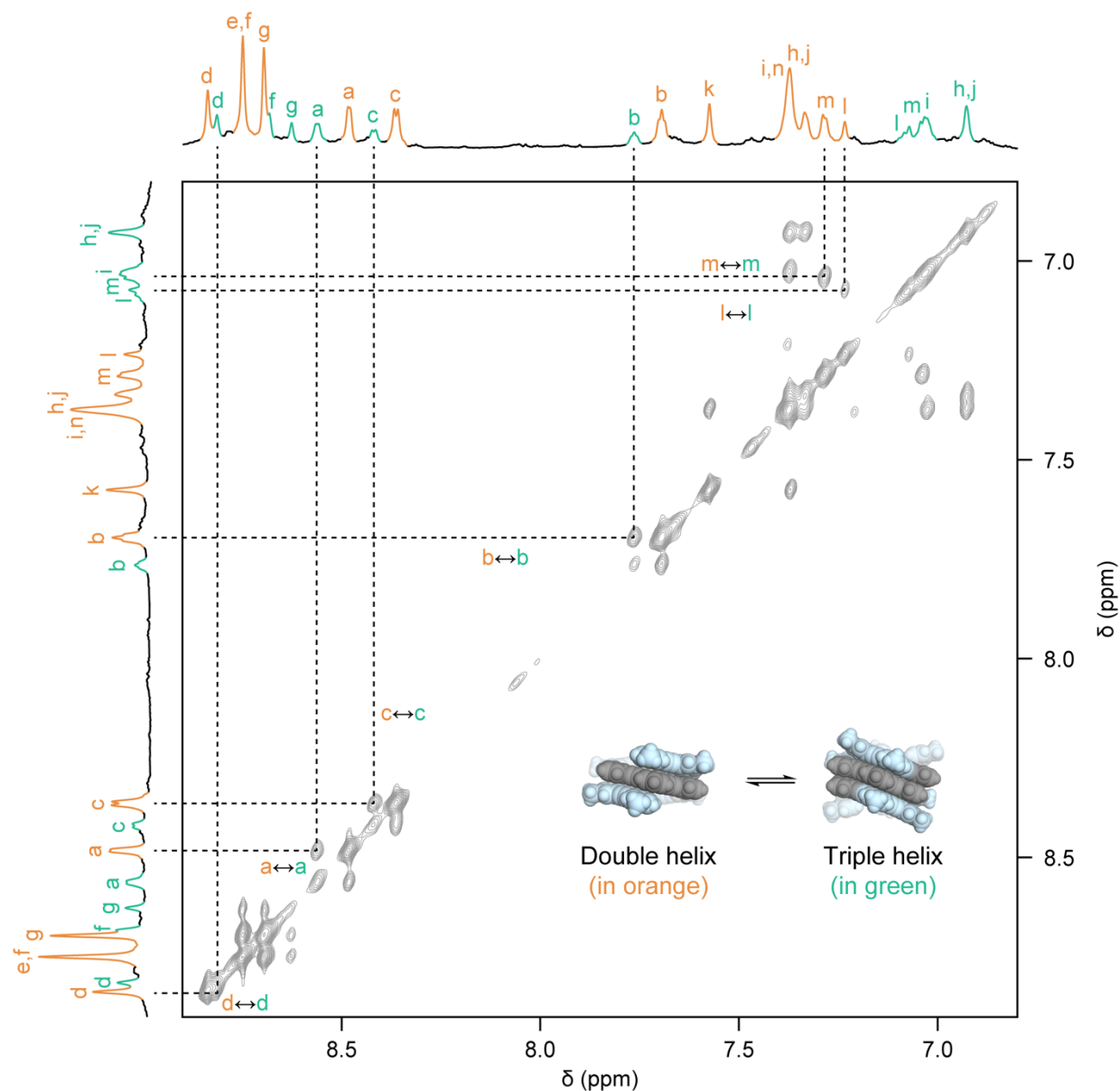


Figure S26 | 2D ^1H - ^1H ROESY NMR spectrum from aromatic region of 2·OTf at 0.5 mM in D_2O (800 MHz, 288 K, 200 ms mixing time). The dotted lines highlight exchange cross-peaks between the protons of the double and triple helices, colored in orange and green, respectively. The ROE cross-peak correlations are shown on the following figures.

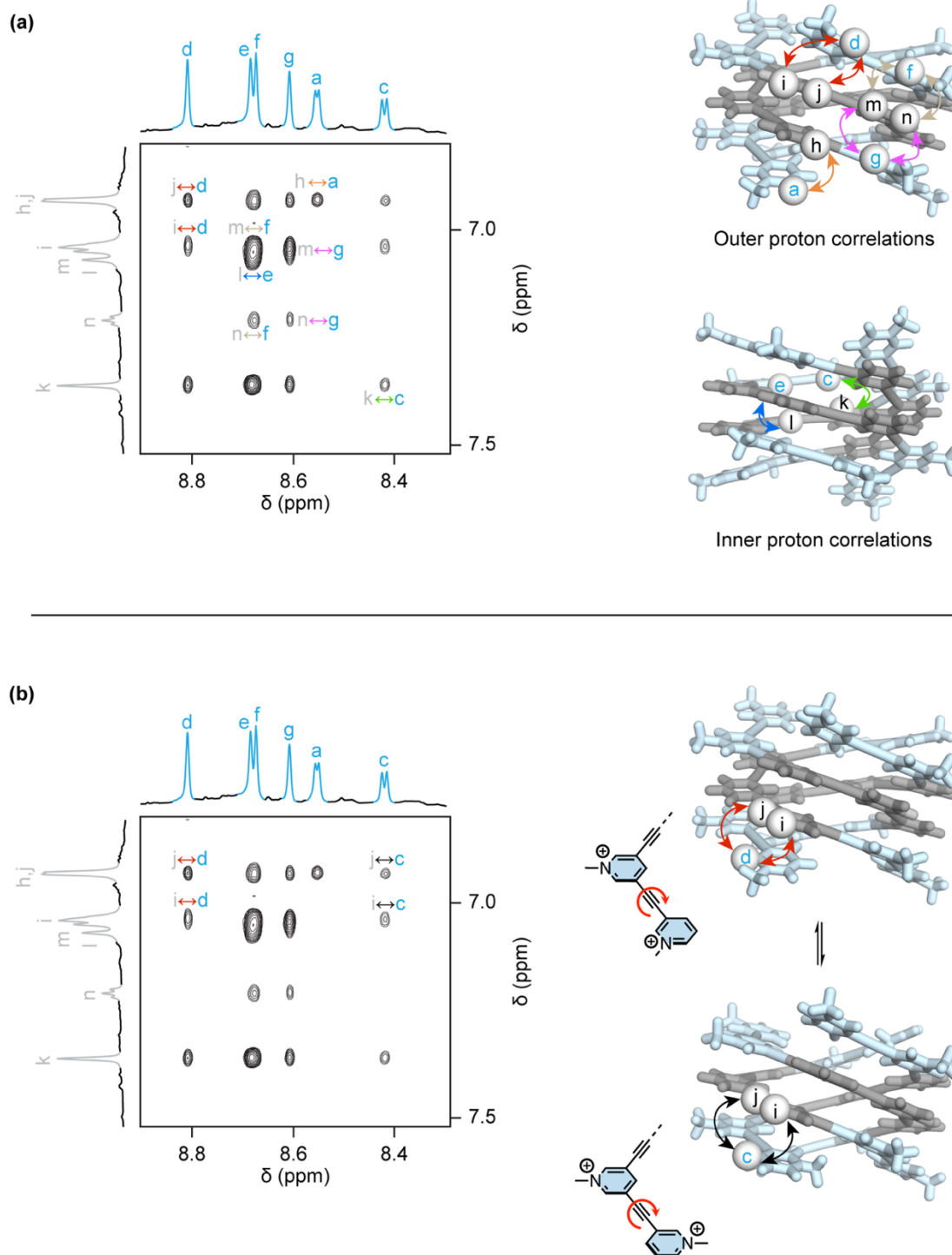


Figure S27 | Expansion of aromatic region in 2D ^1H - ^1H ROESY NMR spectrum of triple helix $(2)_3\cdot\text{OTf}$ in D_2O (10 mM, 800 MHz, 288 K, 200 ms mixing time). **a, Highlighted key correlations between protons of stacked pyridinium (colored in blue) and phenylene (colored in gray) residues. **b**, Rotation of the terminal pyridinium residue around the acetylene bond results in the appearance of additional 2D correlations between stacked pyridinium and phenylene residues.**

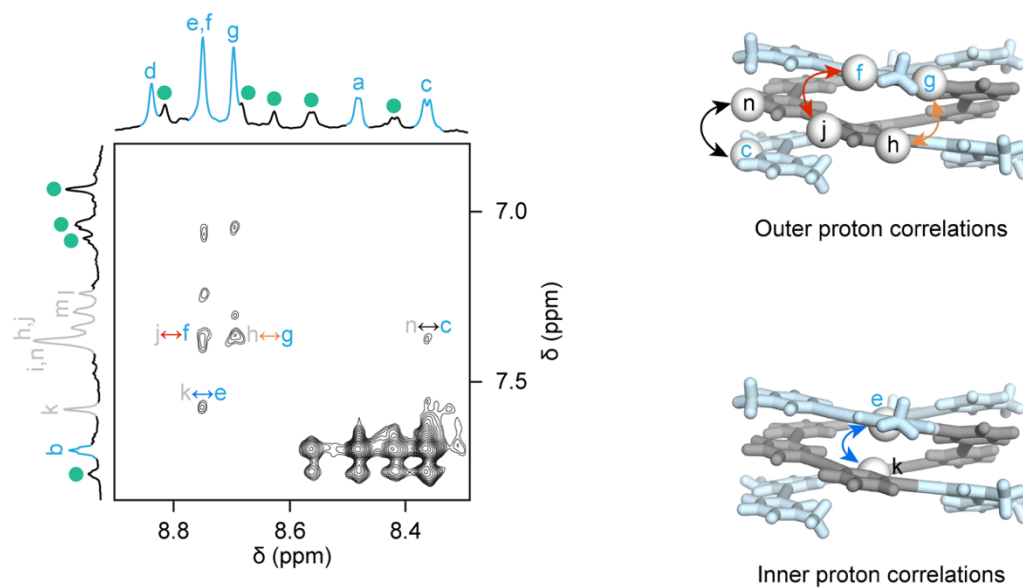


Figure S28 | Aromatic region of 2D ^1H - ^1H ROESY NMR spectrum of double helix $(2)_2\cdot\text{OTf}$ in D_2O (0.5 mM, 800 MHz, 288 K, 200 ms mixing time). Key correlations between protons of stacked pyridinium (colored in blue) and phenylene (colored in gray) residues are highlighted. The signals corresponding to the triple helix are labelled with green circles.

9. Host-guest studies with triple helix (2)₃·TFA

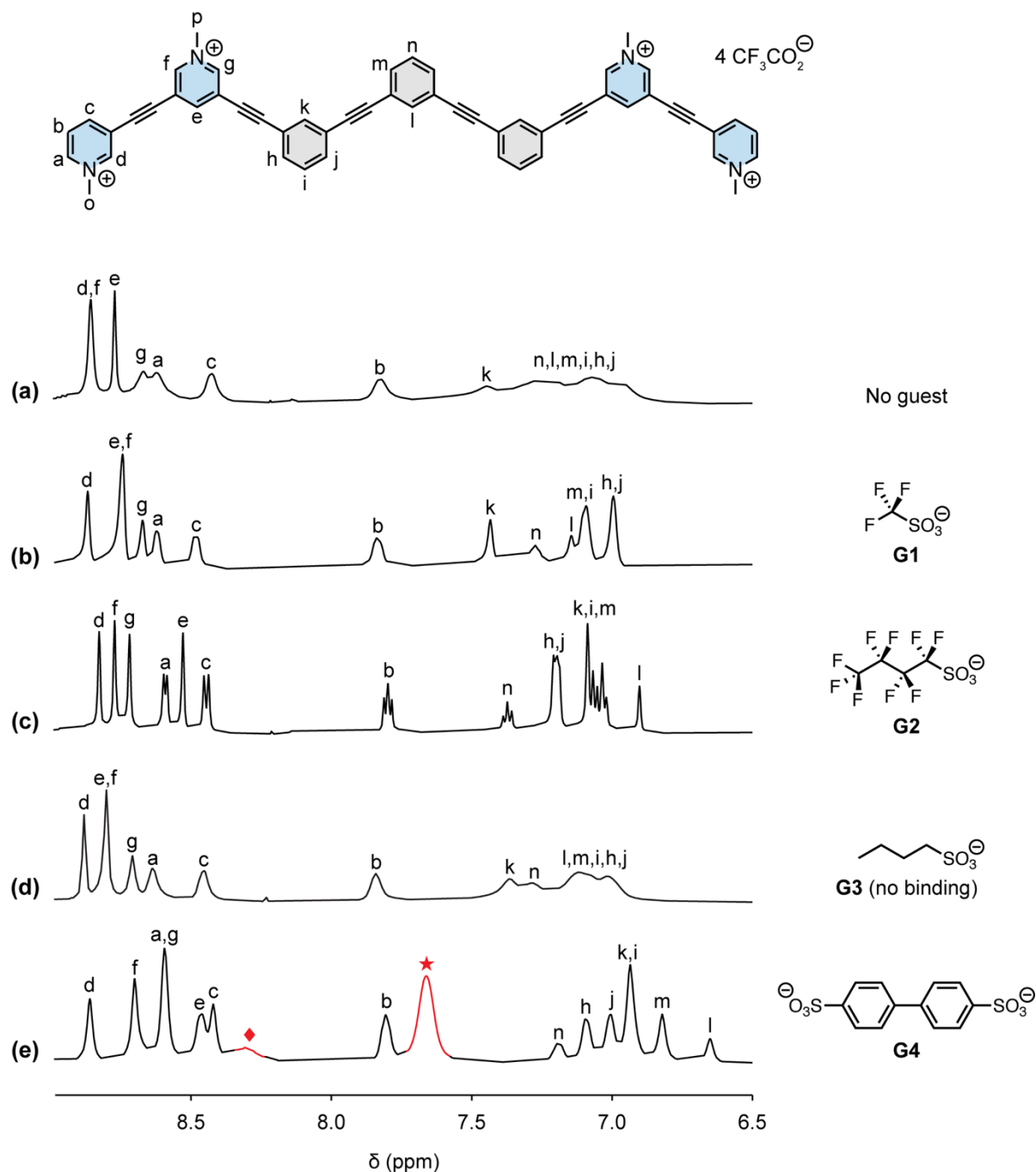


Figure S29 | Aromatic region of ¹H NMR spectra of 2·TFA (5 mM in D₂O, 500 MHz, 298 K) after addition of different guests. a, No guest added. b, After addition of 10.0 equiv. of potassium trifluoromethanesulfonate **G1. c, After addition of 5.0 equiv. of potassium perfluorobutanesulfonate **G2**. d, After addition of 3.9 equiv. of sodium butanesulfonate **G3**. e, After addition of 5.5 equiv. of sodium 4,4'-biphenyldisulfonate **G4**. The signals corresponding to bound and free guest **G4** are labelled with a red diamond and a star, respectively.**

9.1. Titration with potassium trifluoromethanesulfonate (G1)

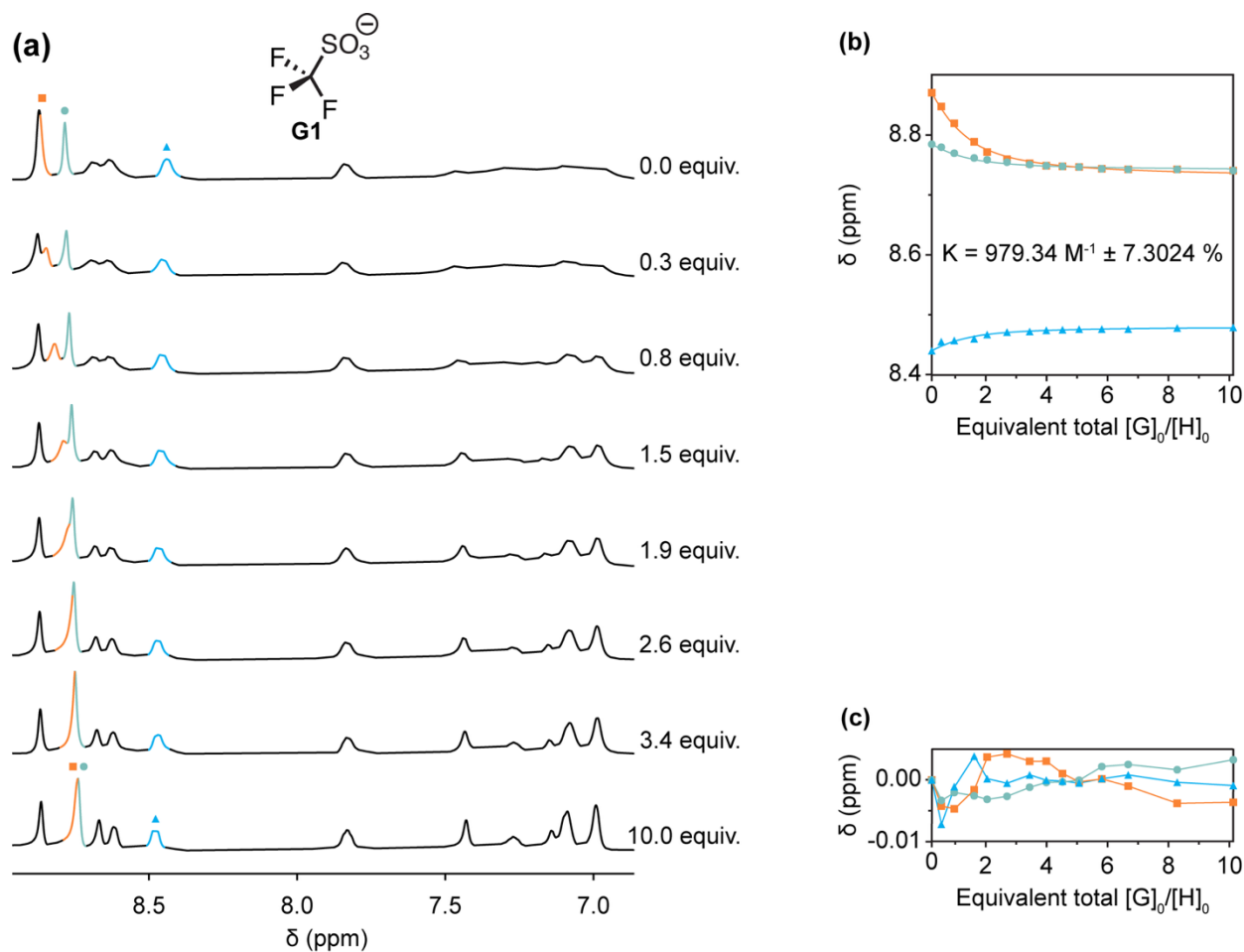


Figure S30 | ^1H NMR titration (500 MHz, 298 K) of potassium trifluoromethanesulfonate G1 into a 5 mM-solution of 2·TFA in D_2O . **a**, Selected region of the ^1H NMR spectra recorded during the titration. **b**, Chemical shift changes recorded as a function of guest equivalents. **c**, Residuals from the curve fitting. Association constant was obtained by fitting the data with *BindFit*.^{S2}

9.2. Titration with potassium perfluorobutanesulfonate (G2)

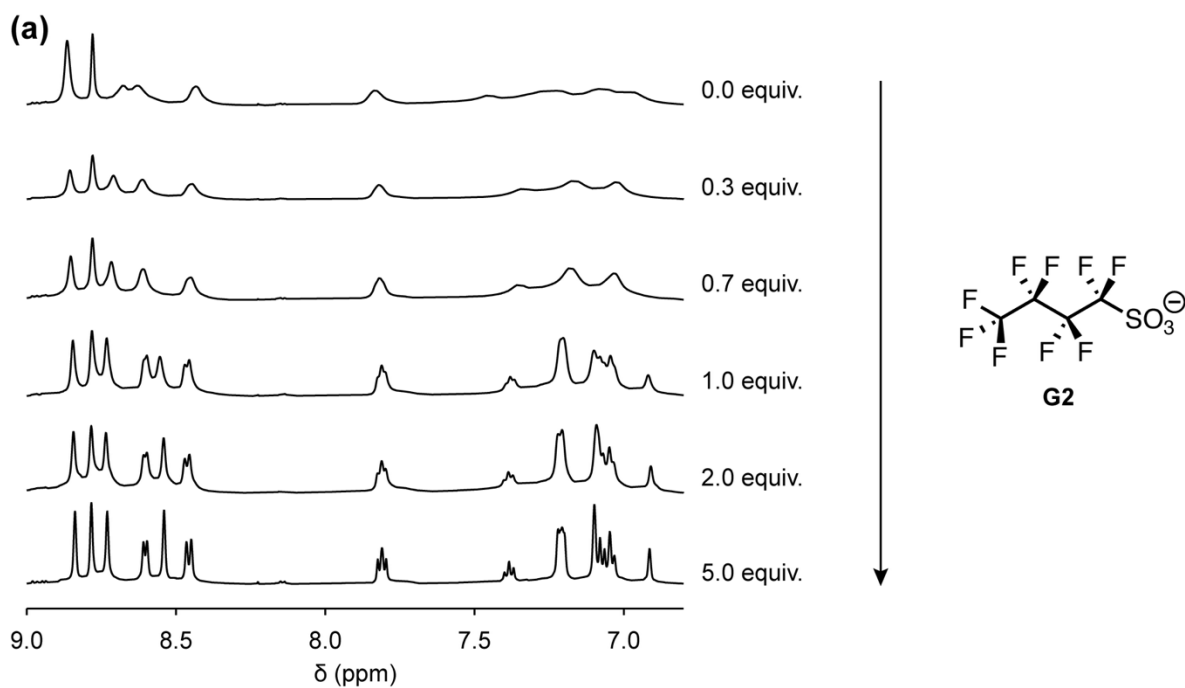


Figure S31 | ^1H NMR titration (500 MHz, 298 K) of potassium perfluorobutanesulfonate G2 into a 5 mM-solution of 2·TFA in D_2O .

1.1. Titration with sodium 4,4'-biphenyldisulfonate (G4)

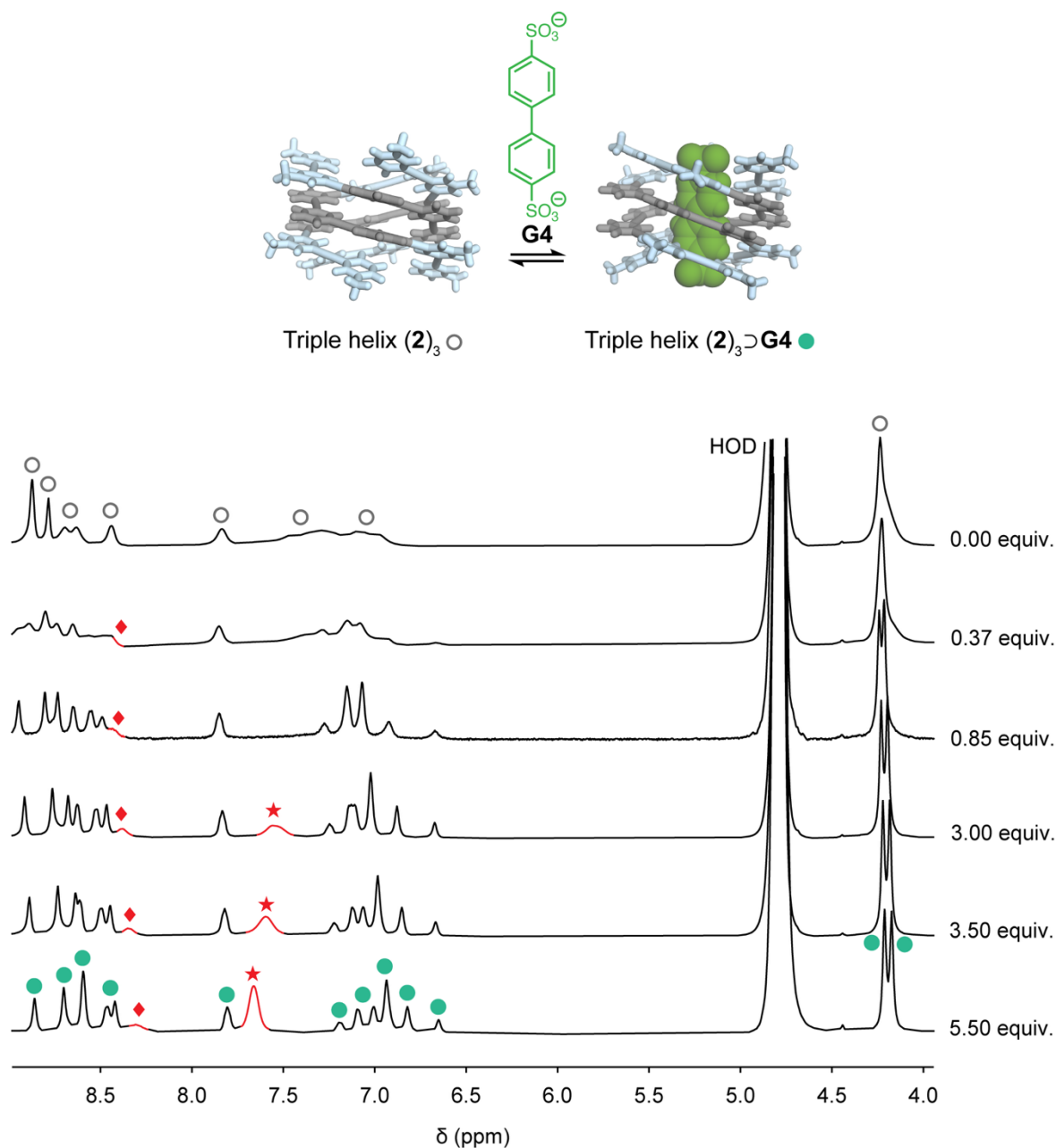


Figure S32 | ¹H NMR titration (500 MHz, 298 K) of sodium 4,4'-biphenyldisulfonate **G4** into a 5 mM-solution **2**·TFA in D₂O. The NMR signals of the resulting complex remain relatively broad, presumably due to dynamic deformations of the triple helix induced by electrostatic attraction between **G4** and the pyridinium units, as evidenced by DFT modeling. This bulky guest appears to occupy the cavity more completely than **G2**, slowing the exchange between free and bound guests on the NMR timescale. The signals corresponding to the bound and free guest are labelled with a red diamond and a red star, respectively.

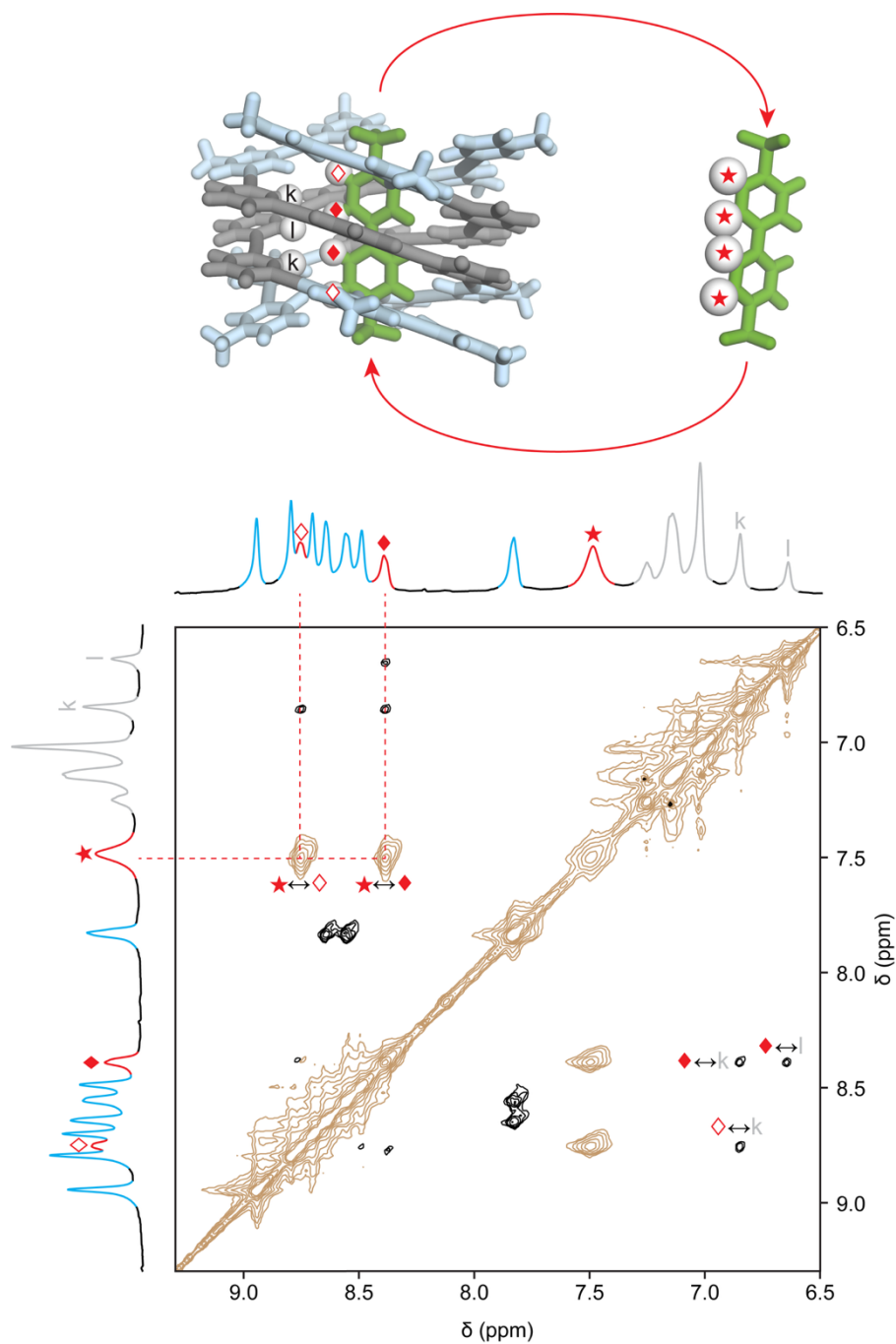


Figure S33 | Expansion of selected spectral regions from 2D ^1H - ^1H ROESY of 2·TFA at 5 mM in D_2O (500 MHz, 100 ms mixing time) recorded at low temperature (278 K) in the presence of 5.5 equiv. of sodium 4,4'-biphenyldisulfonate G4. The dotted lines highlight exchange cross-peaks bound and free guest protons, labelled with red diamonds and star, respectively. ROE correlations between the bound guest and the inner phenylene protons of the triple helix (k and l) are also observable.

10. Concentration- and temperature dependent double-to-triple helix transformation in the presence of guests G1-G4

10.1. Concentration- and temperature dependent double-to-triple helix transformation in the presence of potassium trifluoromethanesulfonate (G1)

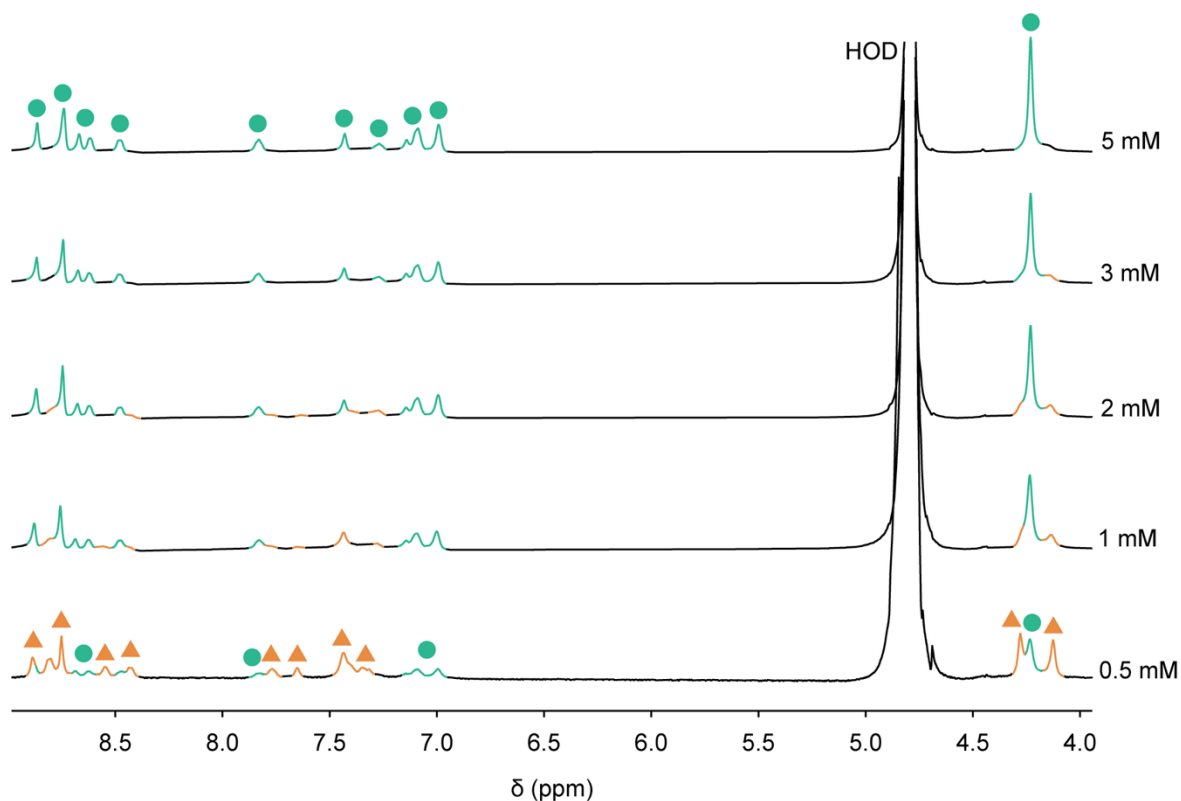


Figure S34 | Concentration-dependent ¹H NMR spectra of 2·TFA between 0.5 mM and 5 mM in D₂O (500 MHz, 298 K) in the presence of 10 equiv. of potassium trifluoromethanesulfonate G1. The signals corresponding to the double and triple helices are labelled with orange triangles and green circles, respectively.

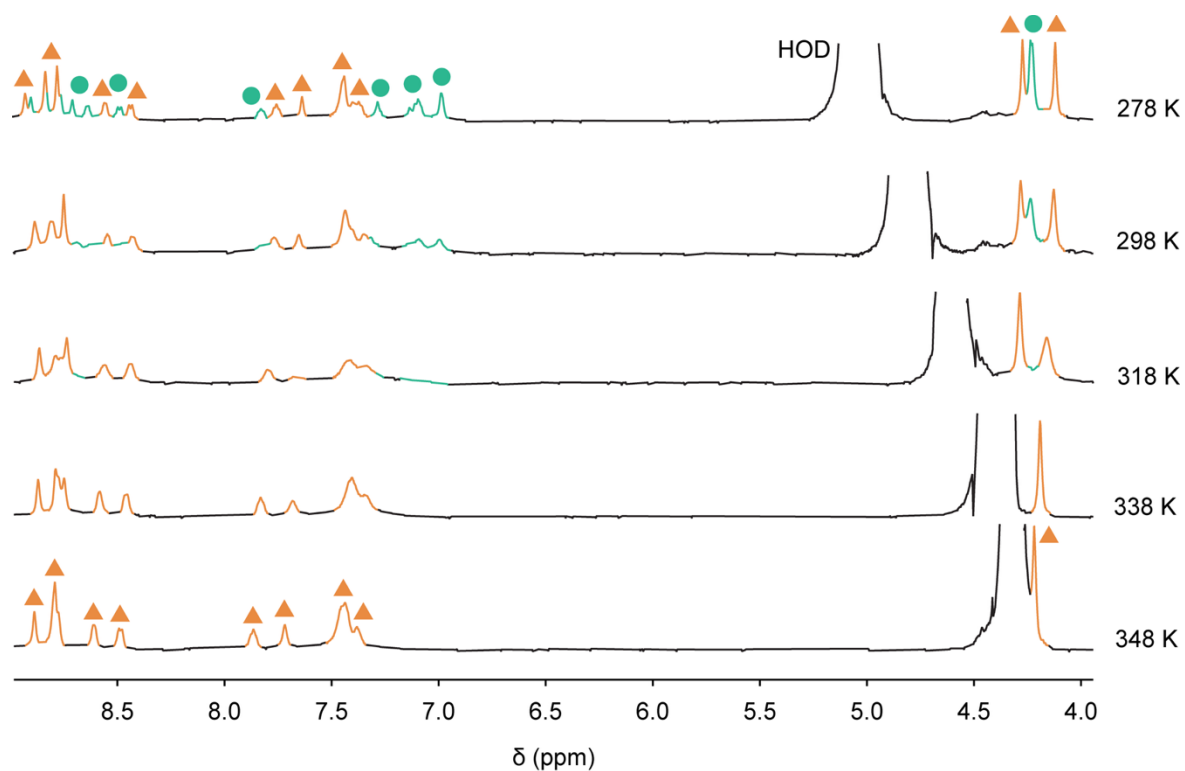


Figure S35 | Variable temperature ^1H NMR spectra of $2\cdot\text{TFA}$ between 278 K and 348 K in D_2O (500 MHz, 0.5 mM) in the presence of 10 equiv. of potassium trifluoromethanesulfonate G1. The signals corresponding to the double and triple helices are labelled with orange triangles and green circles, respectively.

10.2. Concentration- and temperature-dependent double-to-triple helix transformation in the presence of potassium perfluorobutanesulfonate (G2)

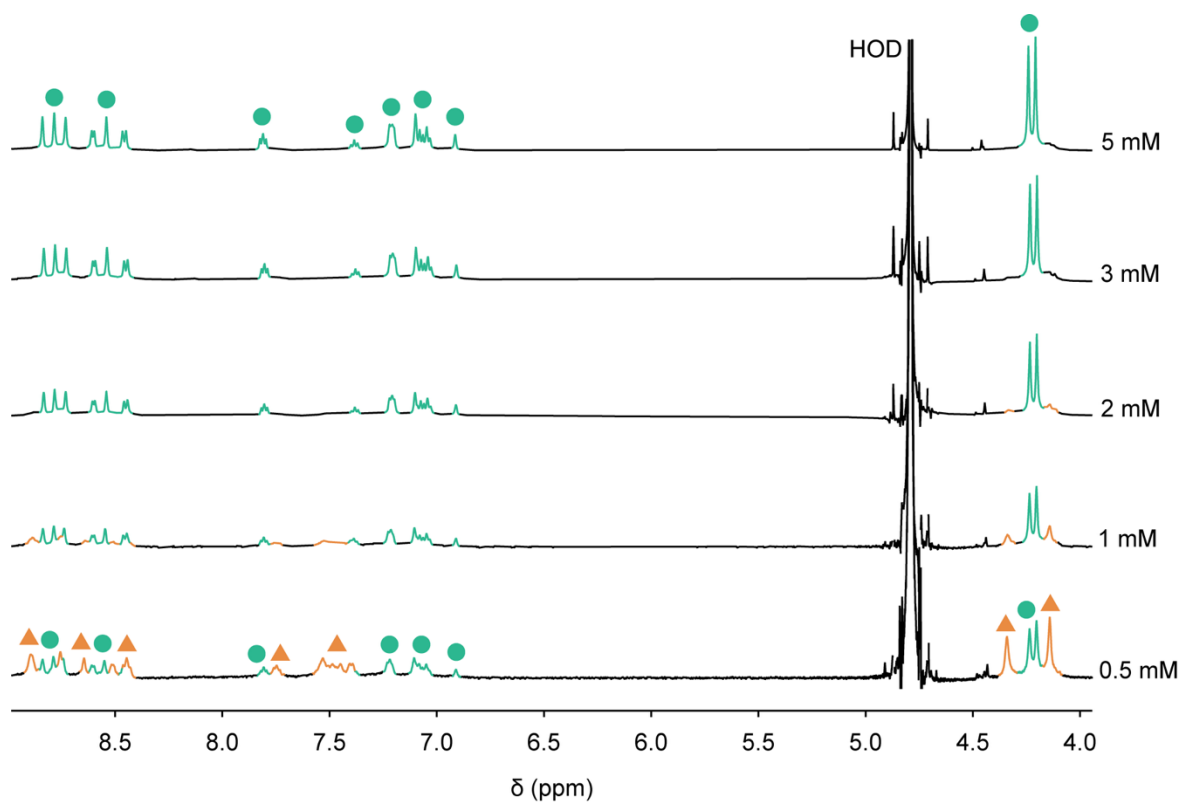


Figure S36 | Concentration-dependent ¹H NMR spectra of 2·TFA between 0.5 mM and 5 mM in D₂O (500 MHz, 298 K) in the presence of 5 equiv. of potassium perfluorobutanesulfonate G2. The signals corresponding to the double and triple helices are labelled with orange triangles and green circles, respectively.

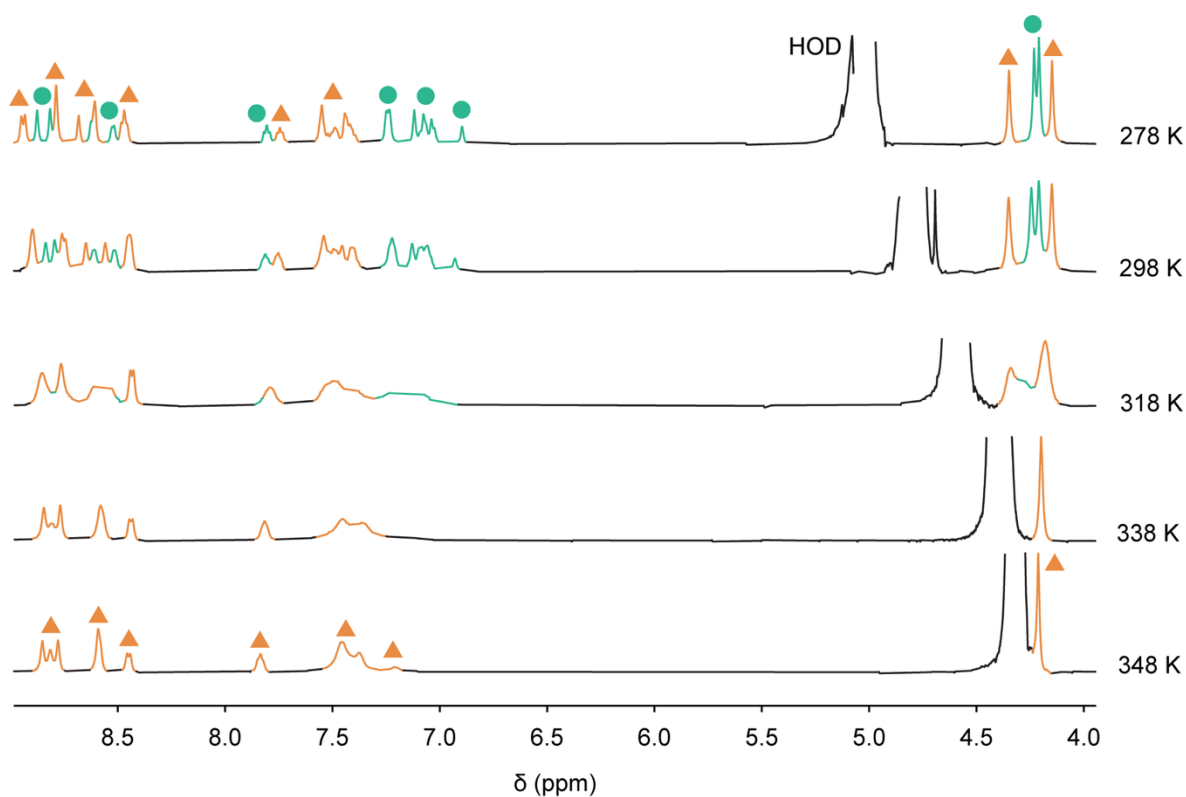


Figure S37 | Variable temperature ^1H NMR spectra of 2·TFA between 278 K and 348 K in D_2O (500 MHz, 0.5 mM) in the presence of 5 equiv. of potassium perfluorobutanesulfonate G2. The signals corresponding to the double and triple helices are labelled with orange triangles and green circles, respectively.

10.3. Concentration- and temperature-dependent double-to-triple helix transformation in the presence of sodium 4,4'-biphenyldisulfonate guest (G4)

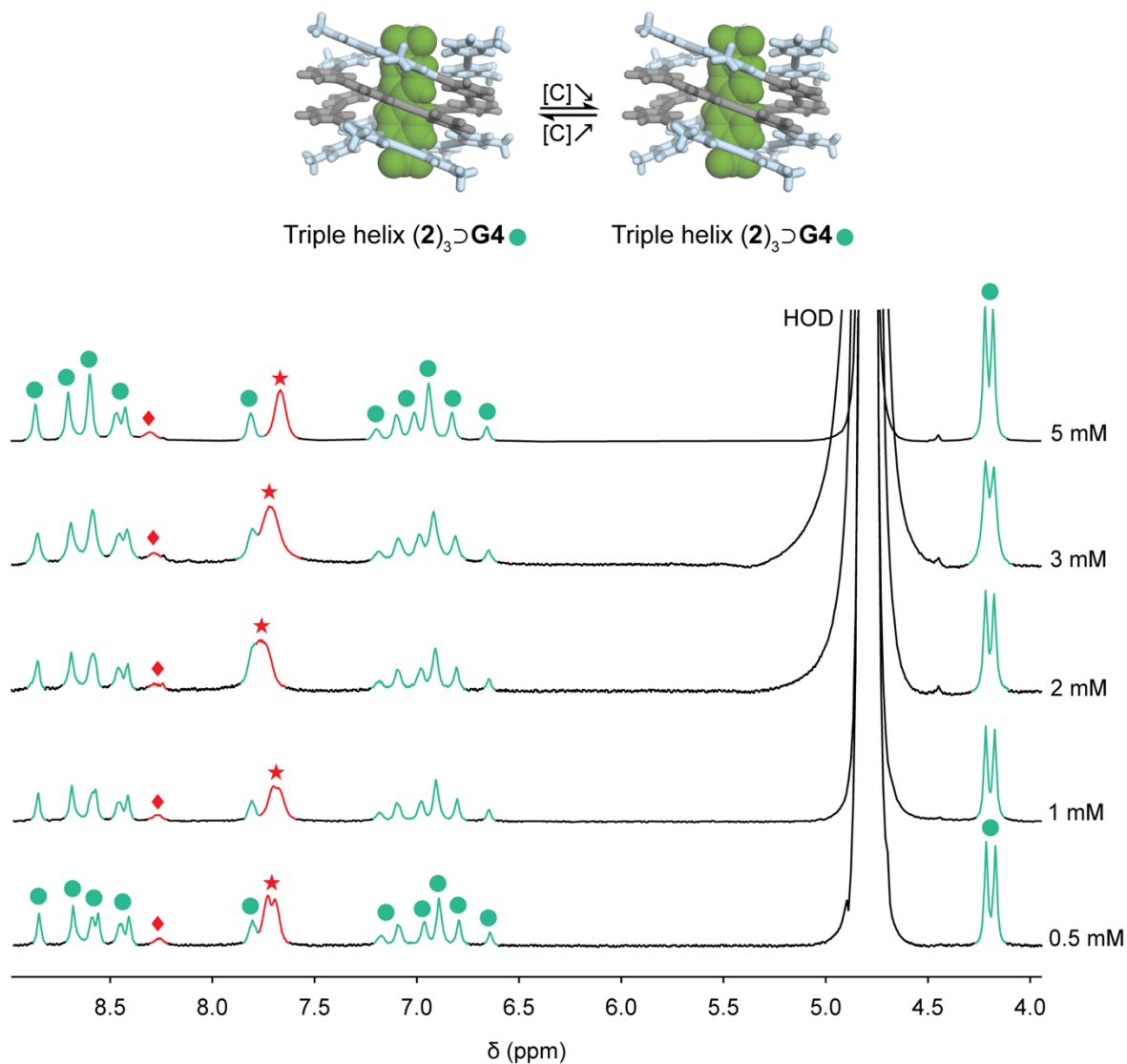


Figure S38 | Concentration-dependent ^1H NMR spectra of $2 \cdot \text{TFA}$ between 0.5 mM and 5 mM in D_2O (500 MHz, 298 K) in the presence of 5.5 equiv. of sodium 4,4'-biphenyldisulfonate G4. No switch was observed in the conditions.

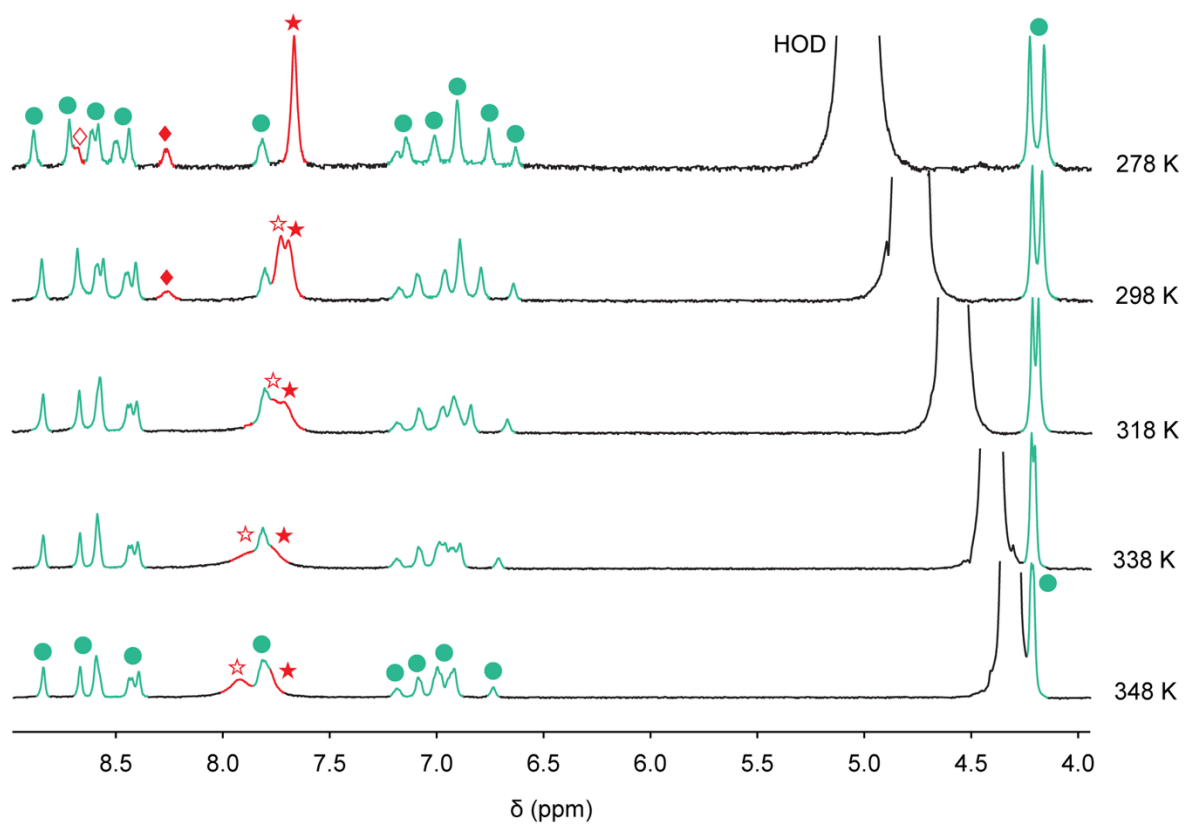


Figure S39 | Variable temperature ¹H NMR spectra of 2·TFA between 278 K and 348 K in D₂O (500 MHz, 0.5 mM) in the presence of 5.5 equiv. of sodium 4,4'-biphenyldisulfonate G4. No switch was observed in the conditions.

11. ^1H and ^{13}C NMR spectra of the synthesized compounds

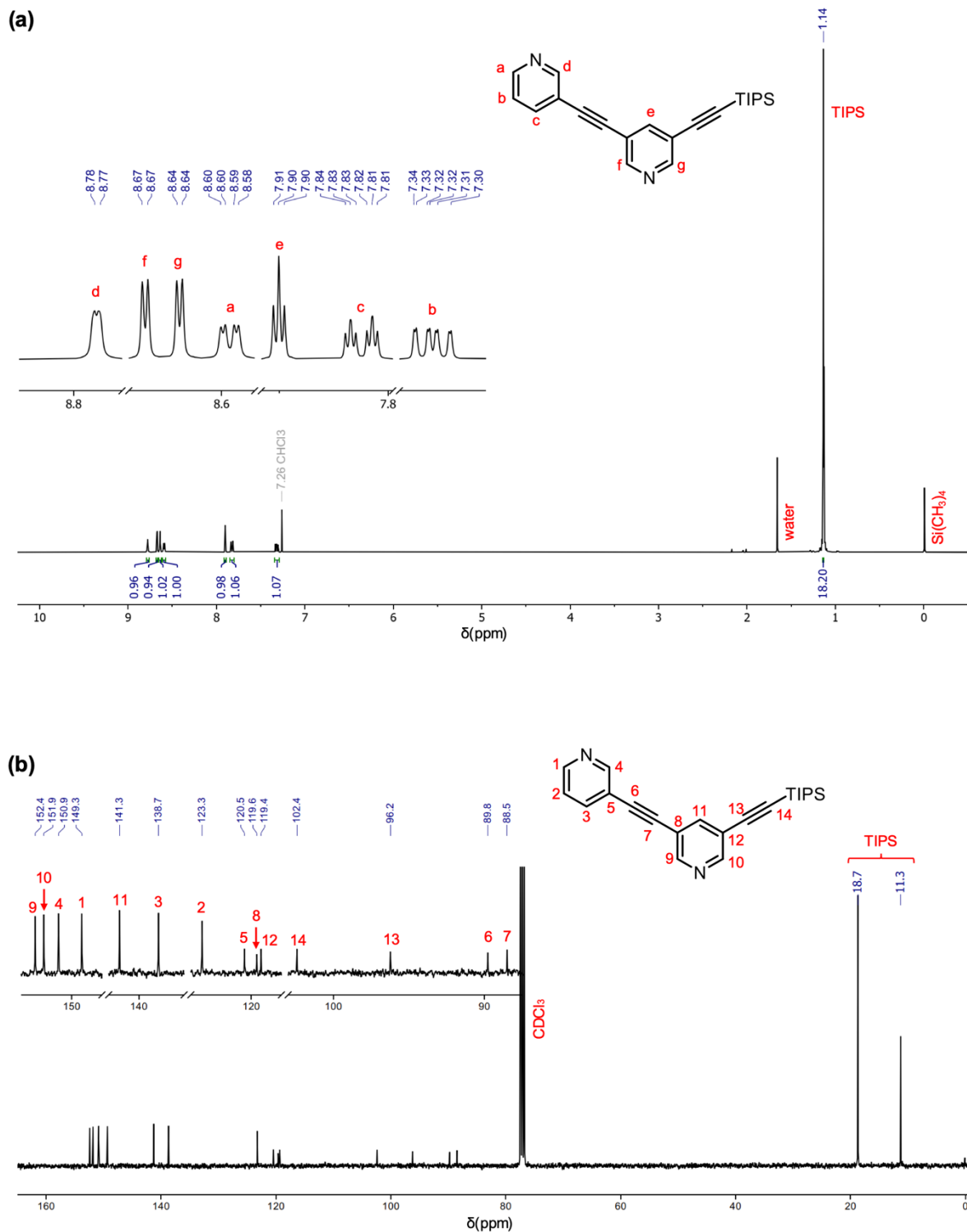


Figure S40 | NMR characterization of S2. a, ^1H NMR (400 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K) and **b,** ^{13}C NMR (100 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K).

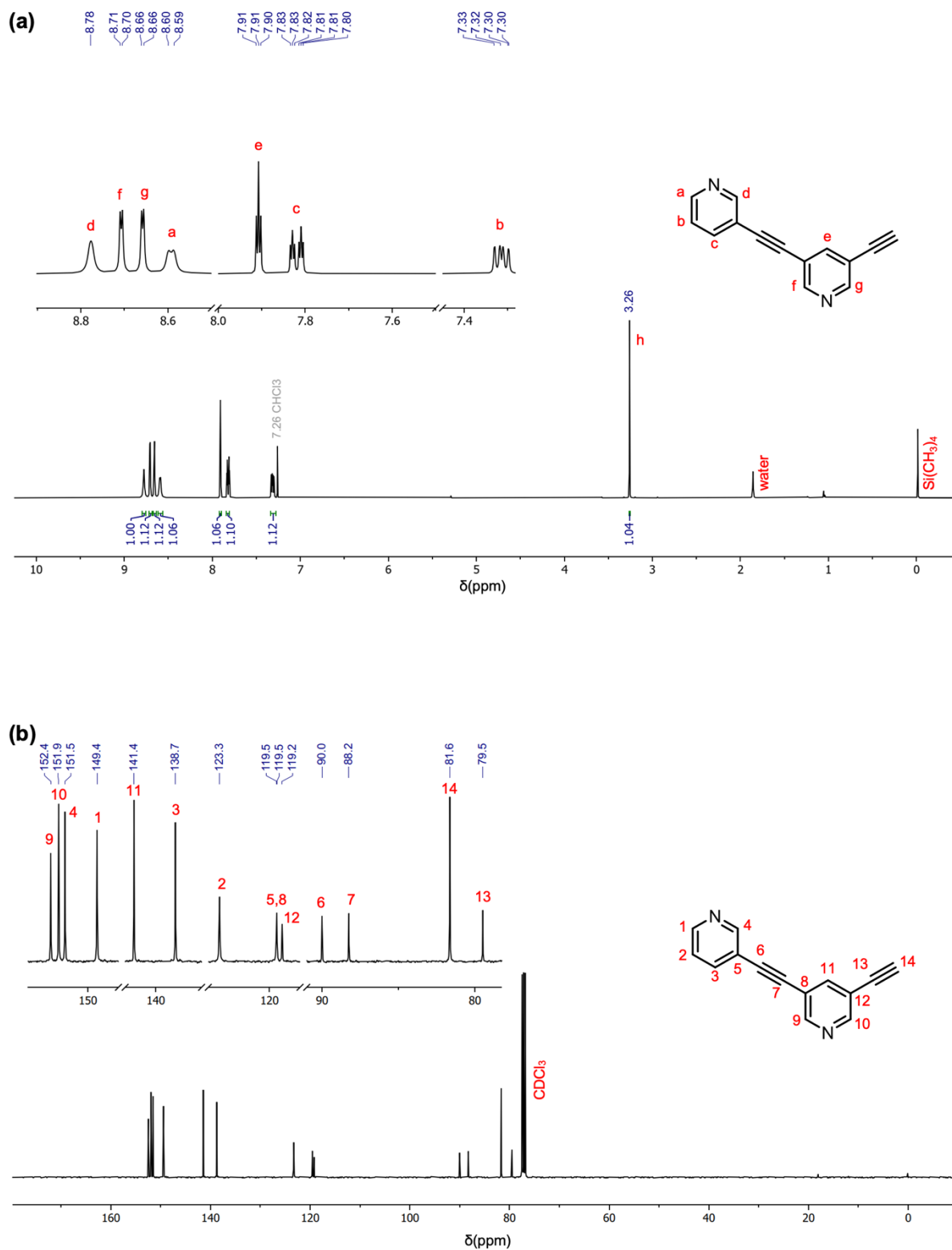


Figure S41. | NMR characterization of S3. a, ^1H NMR (400 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K) and **b,** ^{13}C NMR (100 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K).

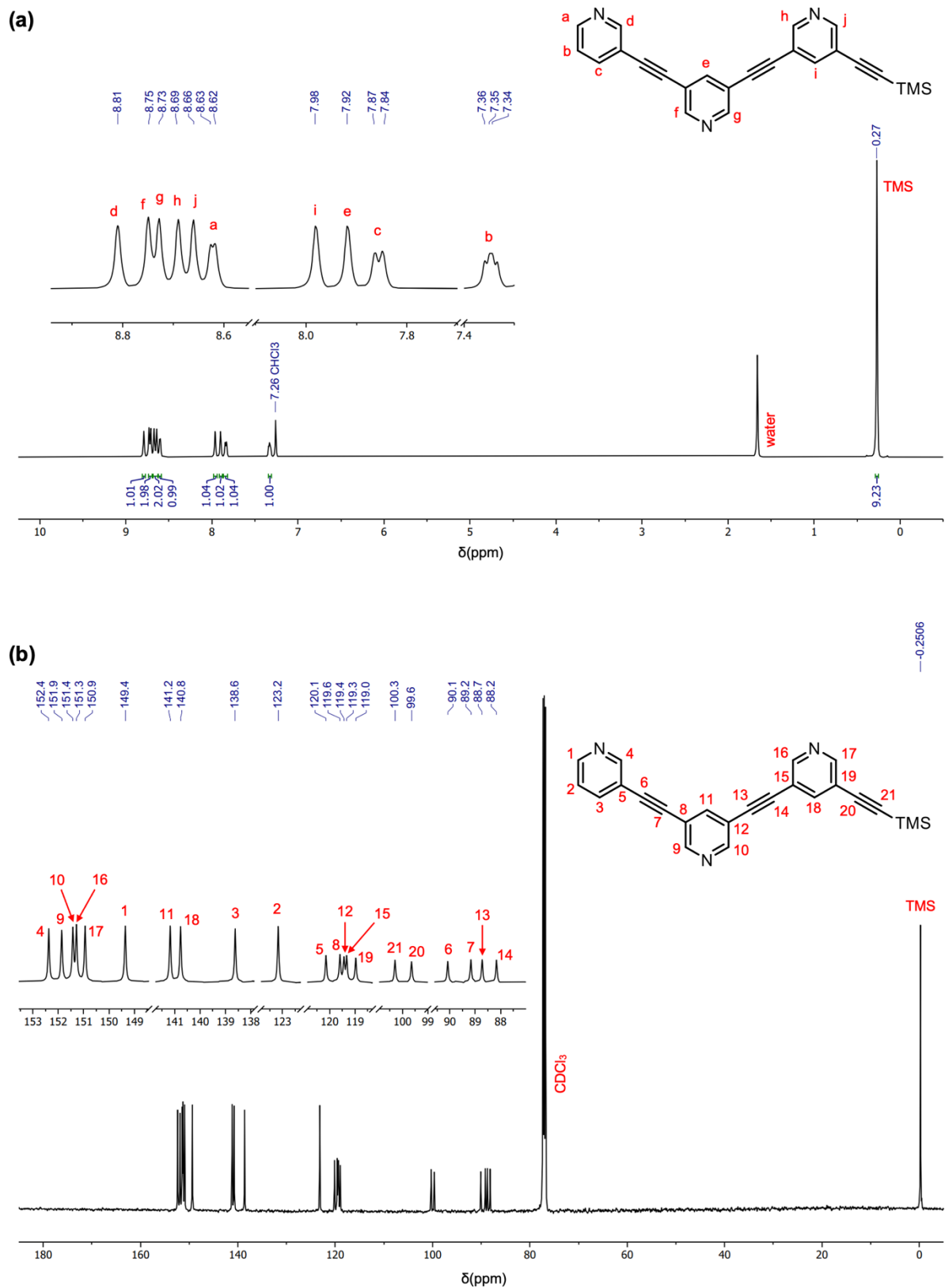


Figure S43 | NMR characterization of S5. a, ^1H NMR (500 MHz, CDCl_3 , 298 K) and **b,** ^{13}C NMR (126 MHz, CDCl_3 , 298 K).

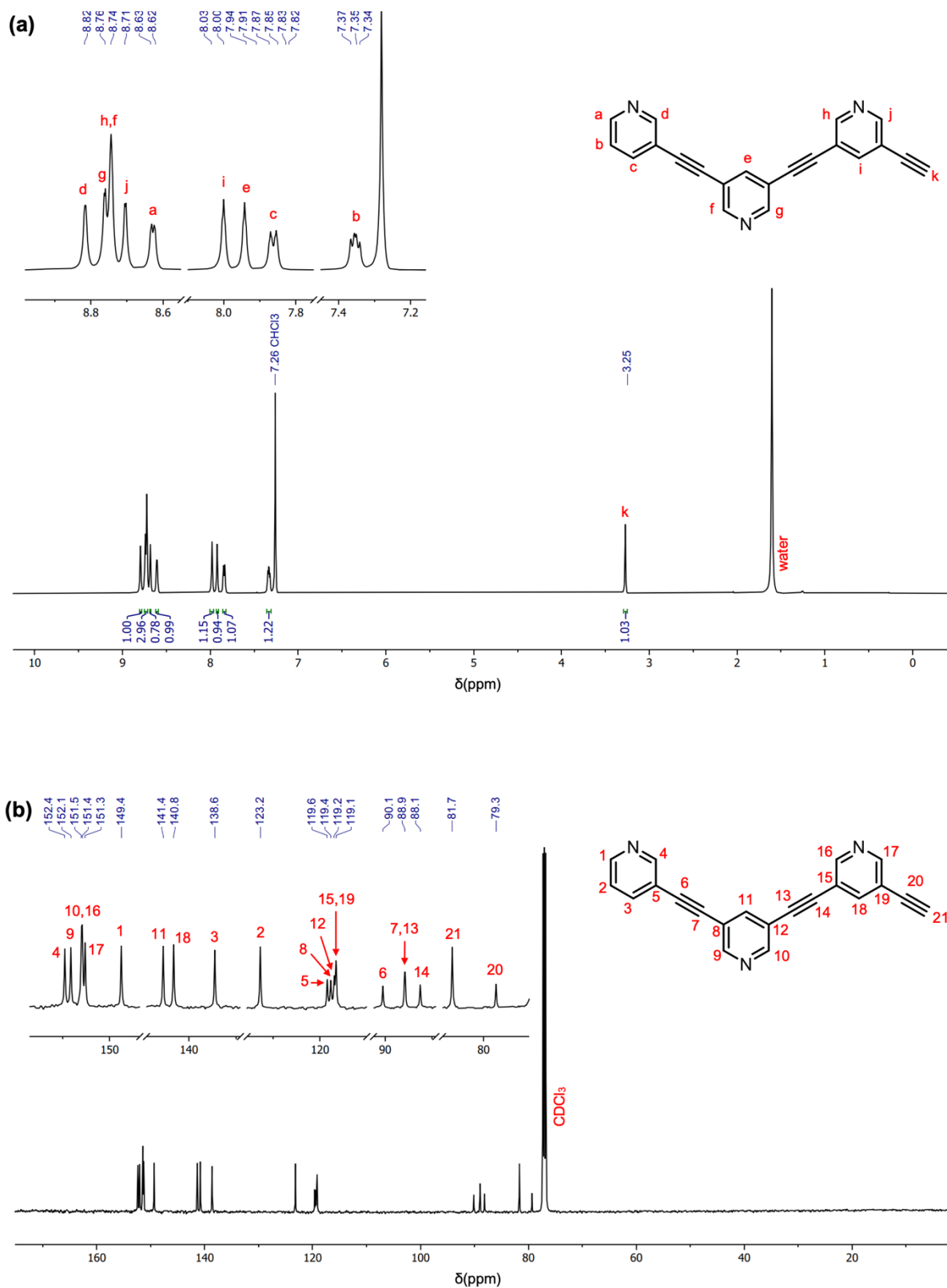


Figure S44 | NMR characterization of S6. a, ¹H NMR (500 MHz, CDCl₃, 298 K) and **b,** ¹³C NMR (126 MHz, CDCl₃, 298 K).

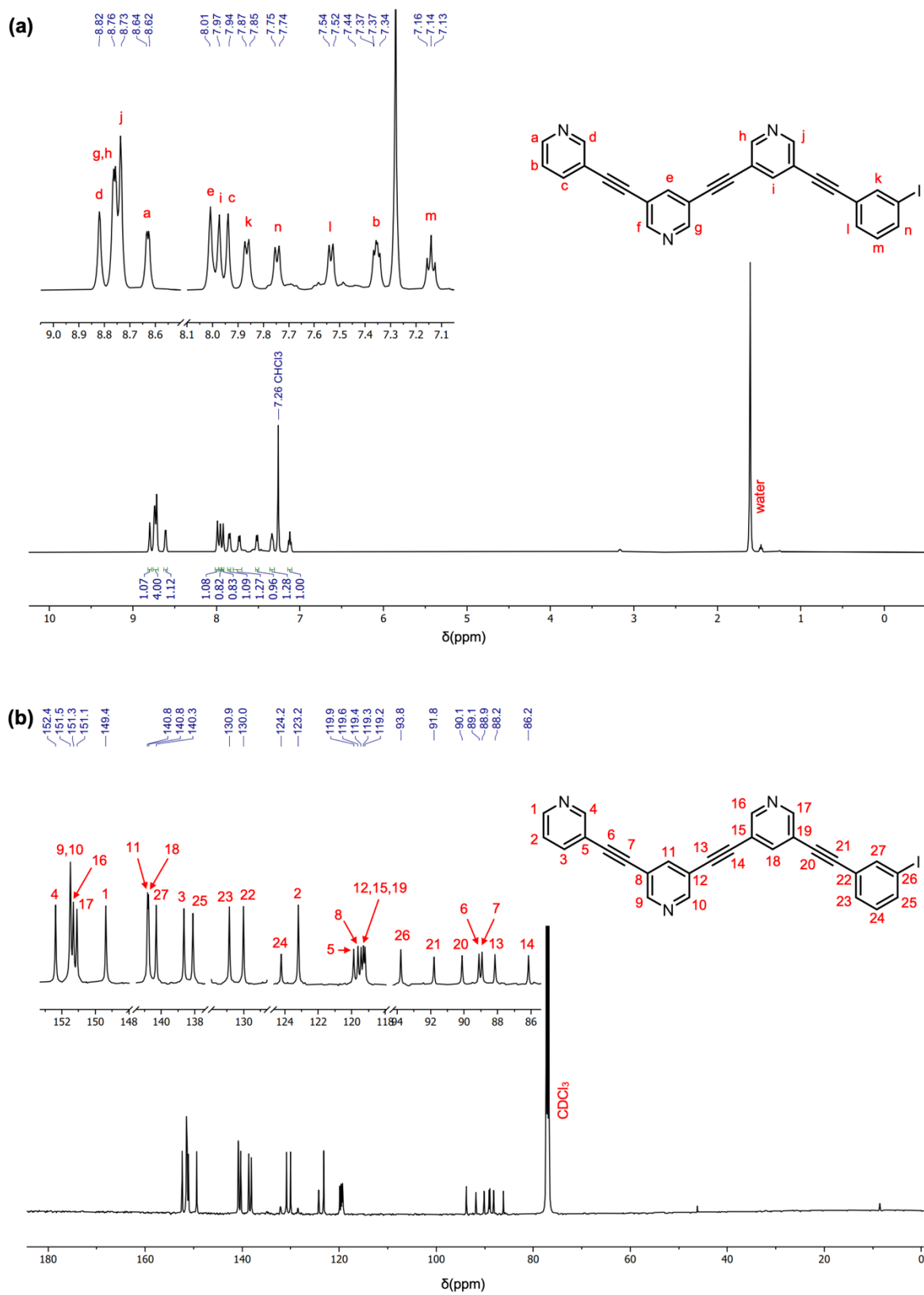


Figure S45 | NMR characterization of S7. a, ^1H NMR (500 MHz, CDCl_3 , 298 K) and **b,** ^{13}C NMR (126 MHz, CDCl_3 , 298 K).

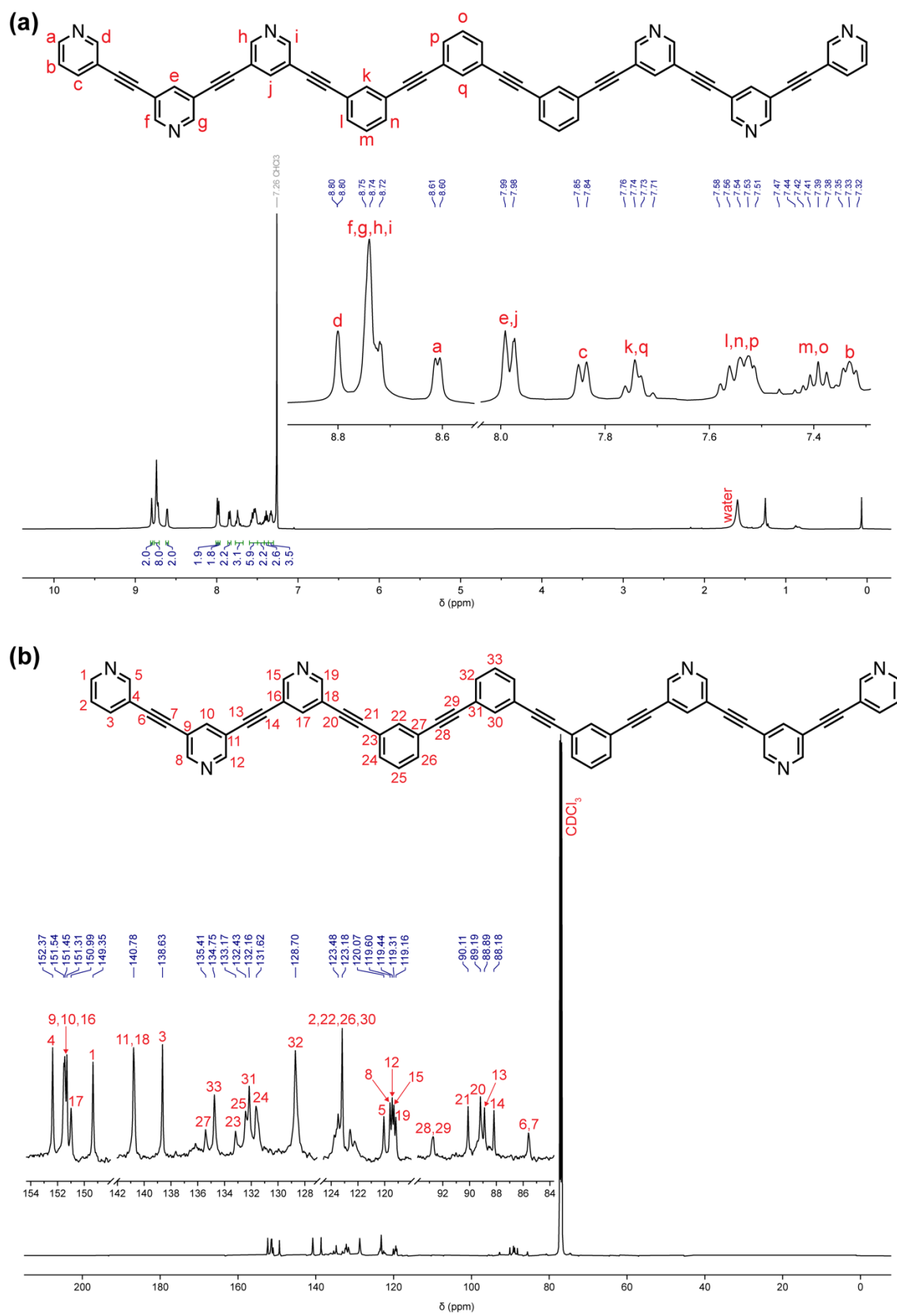


Figure S46 | NMR characterization of S8. a, ^1H NMR (500 MHz, CDCl_3 , 298 K) and **b,** ^{13}C NMR (126 MHz, CDCl_3 , 298 K).

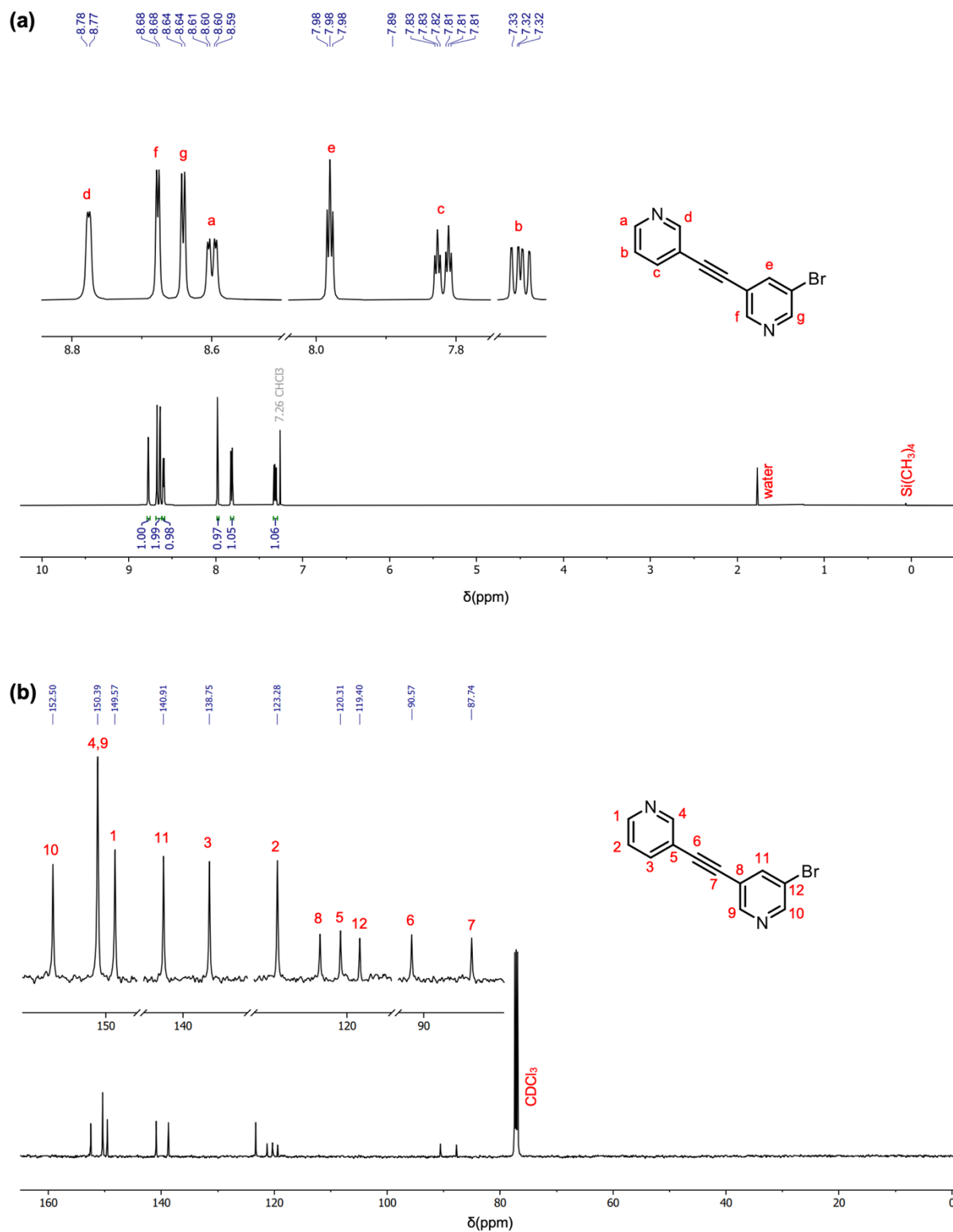


Figure S47 | NMR characterization of S12. a, ¹H NMR (400 MHz, CDCl₃ + 0.03 % tetramethylsilane, 293 K) and **b,** ¹³C NMR (100 MHz, CDCl₃+ 0.03 % tetramethylsilane, 293 K).

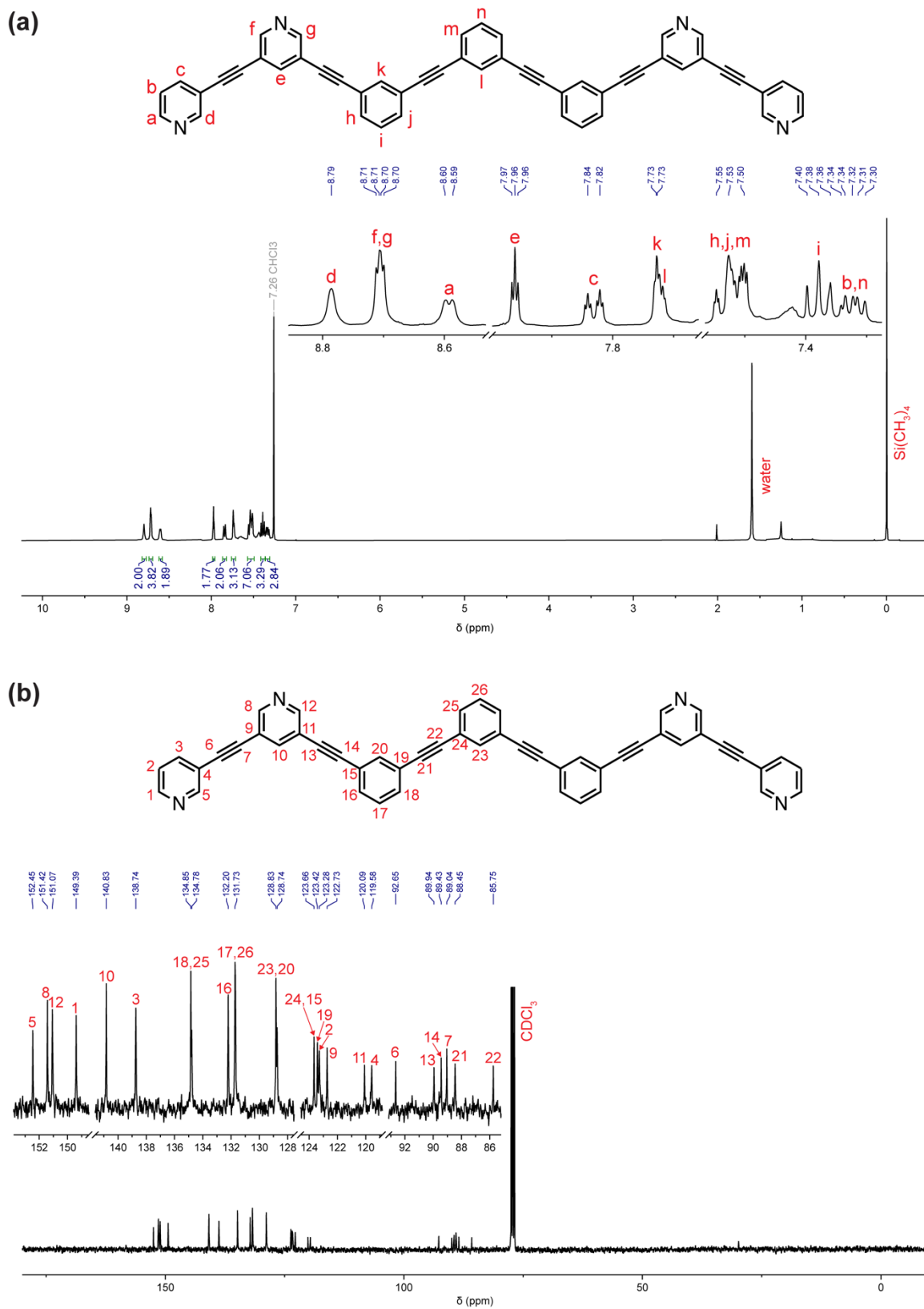


Figure S48 | NMR characterization of S13. a, ^1H NMR (400 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K) and **b,** ^{13}C NMR (100 MHz, CDCl_3 + 0.03 % tetramethylsilane, 293 K).

12. DFT calculations

All geometries of (1)₂, (2)₂, (2)₃, (2)₃⊃G2, and (2)₃⊃G4 were optimized without symmetry constraints. Initial structures were manually constructed and pre-optimized using Grimme's GFN2-xTB program.^{S5} The final re-optimizations were performed at the BP86-D4-COSMO/def2-TZVP^{S6-8} level of theory using the TURBOMOLE 7.7 program suite in a water solvent.^{S9} The solvent effects were modeled using the COSMO model^{S10} with the default TURBOMOLE parameters. The D4 dispersion correction^{S7} was included to account for crucial London dispersion forces, which are essential for the stabilization of these supramolecular complexes, given the dominance of stacking interactions.

Cartesian coordinates for (1)₂

N	-3.0102992	-1.9543833	-7.4652450
C	8.3565201	-0.2134676	1.0538048
N	-4.2485828	2.5810286	7.1616838
C	-5.0608227	-3.7909471	4.0552030
C	-5.9205448	-3.6726462	5.1532185
H	-6.9727803	-3.4232905	5.0479238
N	-5.4662314	-3.8760252	6.4040518
C	-4.1706617	-4.1989022	6.6438743
C	-3.2780584	-4.3281195	5.5941847
C	-3.7131591	-4.1245107	4.2890921
H	-3.0256277	-4.2244503	3.4509478
C	-5.5555725	-3.5629954	2.7508830
C	-5.9647499	-3.3591985	1.6235638
C	-6.4038845	-3.1057599	0.3053870
C	-7.7704479	-2.9823111	0.0332516
H	-8.5349623	-3.0924585	0.7973791
N	-8.1973397	-2.7331915	-1.2235154
C	-7.3399879	-2.5891339	-2.2522059
H	-7.7729958	-2.3911202	-3.2287176
C	-5.9587966	-2.7033083	-2.0452699
C	-5.4879190	-2.9628733	-0.7505607
H	-4.4199381	-3.0548442	-0.5643969
C	-5.0706832	-2.5419224	-3.1327659
C	-4.2959627	-2.3921419	-4.0582302
C	-3.3883618	-2.1970118	-5.1235336
C	-3.8645611	-2.1482843	-6.4380980
H	-4.9131376	-2.2782859	-6.6900458
C	-1.6876143	-1.7917821	-7.2707221
H	-1.0782487	-1.6378995	-8.1568551
C	-1.1454481	-1.8314054	-5.9776891
C	-2.0122995	-2.0370229	-4.8929672
C	0.2429702	-1.6508319	-5.7977127
C	1.4386214	-1.4773079	-5.6395866
C	2.8273769	-1.2678954	-5.4657696
C	3.6802955	-1.2175814	-6.5887230
H	3.2599851	-1.3411185	-7.5863313
C	5.0481547	-1.0162196	-6.4109833
C	5.5817918	-0.8621222	-5.1320882
H	6.6519131	-0.7080196	-4.9995617
C	4.7390216	-0.9018073	-4.0003692
C	3.3598265	-1.1077653	-4.1737133
H	2.7020766	-1.1449328	-3.3070475
C	5.2817632	-0.7290764	-2.7038854
C	5.7682564	-0.5625117	-1.5981238

C	6.3629848	-0.3693002	-0.3275302
C	7.7684654	-0.4008290	-0.1961565
H	8.3834720	-0.5741543	-1.0788995
C	7.5684667	0.0050561	2.1837988
H	8.0331201	0.1481722	3.1584062
C	6.1626175	0.0452619	2.0687143
C	5.5652670	-0.1427707	0.8094115
C	5.3619798	0.2781397	3.2135104
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H	-2.4722523	4.8411624	-8.0760307
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H	4.4812050	-0.1141143	0.7135618
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H	-2.1922065	2.3928714	3.9148139
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N	4.2499759	-2.5840898	7.1612097
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C	5.9143486	3.6735663	5.1505215
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C	4.1634823	4.1986051	6.6403510
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C	5.9604815	3.3612226	1.6208532

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C	-7.7684504	0.4023634	-0.1965151
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C	-7.5638369	-0.0028994	2.1830968
H	-8.0265290	-0.1451848	3.1587502
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C	-1.7710717	-1.2937078	6.3957951
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C	2.9312063	-2.3047193	7.2353079
H	2.5169196	-2.1644192	8.2298918
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C	2.7754043	-2.4435414	4.8323955
C	4.8113975	-2.9720923	3.5597122
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C	6.0469138	-3.4300421	1.2842648
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H	4.2505549	-3.5754657	0.0817235
C	5.3719676	-4.0292057	-2.3373832
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C	2.7314149	-4.6059283	-4.6863261
H	2.1468118	-4.6450769	-3.7690455
C	4.1217214	-4.3878891	-4.6233406
C	4.8484670	-4.3526852	-5.8188682
H	5.9246396	-4.2047672	-5.8439450
H	2.4763906	-4.8447421	-8.0767351
H	-5.7095290	0.9837698	-7.2798072
H	-9.4393395	0.2374059	1.1569290
H	-4.4794624	0.1148520	0.7065180
H	1.6205290	2.0714356	-3.8827157
H	2.1939212	-2.3956354	3.9140299
H	-4.1982412	-0.9121634	8.7644484
H	3.8886982	4.3536951	7.6804958
H	2.2389236	4.5960217	5.8025687
C	-9.5718442	3.7712056	0.1246798
H	-9.8782086	4.6023665	0.7681038
H	-9.9365527	3.9241960	-0.8928238
H	-9.9508263	2.8259957	0.5275246
C	-9.6540690	-2.6380753	-1.4676629
H	-10.0749814	-3.6494862	-1.4588996
H	-10.1054178	-2.0377725	-0.6719406
H	-9.8185972	-2.1683025	-2.4397845
C	5.0249505	-4.5015067	-8.2489268
H	5.2103773	-5.5357284	-8.5601225
H	4.4594792	-3.9739294	-9.0220647
H	5.9735468	-3.9938660	-8.0565334
C	9.5749514	-3.7692963	0.1224064
H	9.9388327	-3.9256447	-0.8949194
H	9.9542848	-2.8228556	0.5219565
H	9.8815452	-4.5984908	0.7682600
C	-6.3956516	-3.7693073	7.5483504
H	-7.3155716	-3.2841879	7.2146697
H	-6.6127611	-4.7773874	7.9184071
H	-5.9175869	-3.1789531	8.3363304
C	-3.5602639	-1.9228984	-8.8389442
H	-2.7566705	-1.6722746	-9.5341661
H	-3.9710481	-2.9107455	-9.0721686
H	-4.3525635	-1.1682751	-8.8838809
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H	3.9698574	2.9094649	-9.0776410
H	4.3538210	1.1679230	-8.8863301
C	-5.0210077	2.6569950	8.4213830
H	-4.9145224	1.7077400	8.9567608
H	-4.6198710	3.4757775	9.0276641
H	-6.0698622	2.8422946	8.1816355
C	6.3886580	3.7696329	7.5456134
H	7.3035983	3.2731107	7.2149078
H	6.6160865	4.7784759	7.9074137
H	5.9057868	3.1903657	8.3387597
C	9.6505819	2.6416192	-1.4694946
H	10.0728458	3.6521978	-1.4458965
H	10.1000351	2.0293351	-0.6817657
H	9.8154079	2.1855842	-2.4480672
C	5.0221608	-2.6598667	8.4210872
H	4.9180249	-1.7095082	8.9549734
H	4.6187517	-3.4763649	9.0289862
H	6.0706300	-2.8481196	8.1816834
C	-5.0214177	4.4997765	-8.2478782
H	-5.2100204	5.5336367	-8.5583465
H	-4.4548096	3.9742677	-9.0215626
H	-5.9685699	3.9892879	-8.0558777

Cartesian coordinates for (2)₂

N	-2.2229435	-2.6413441	-6.9139460
C	9.1283864	0.8672370	0.9814647
N	-3.5687004	2.1793254	7.3053312
H	-4.7999040	-4.0058942	2.0162087
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Cartesian coordinates for $(2)_3$

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H	3.8788304	14.8797843	15.9906128
H	2.4197420	14.1293340	15.2662432
H	4.8839964	14.6185821	4.2345037
C	3.1407498	15.9842782	5.5193964
H	2.4909408	15.4369826	4.8288490
H	3.6984932	16.7553502	4.9774504
H	2.5492810	16.4362746	6.3181512
N	7.8633917	8.0210738	3.1484022
C	16.3398515	6.7485524	14.6891805
C	8.5977214	18.8664068	15.1751038
H	6.2280947	9.5706218	5.6311993
C	7.0142359	9.0013654	5.1410700
C	6.9079040	8.7409516	3.7846201
H	6.0717234	9.0763073	3.1763930
C	8.9386881	7.5457309	3.8063824
H	9.6548812	6.9714798	3.2251422
C	9.0966079	7.7777722	5.1767609
C	8.1104629	8.5256253	5.8503349
C	10.2245308	7.2513283	5.8468072
C	11.1784929	6.7798258	6.4363981
C	12.2808222	6.2278585	7.1269243
C	13.0202288	5.1971323	6.5347410
H	12.7826965	4.7878062	5.5571173
N	14.0711357	4.6557471	7.1853573
C	14.4494610	5.0792244	8.4073612
H	15.3055975	4.5826045	8.8546366
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C	12.6518424	6.6862932	8.4011610
H	12.0913654	7.4884397	8.8764495
C	14.1698340	6.5324302	10.3344563
C	14.5536653	6.8873612	11.4354081
C	15.0082303	7.2523377	12.7242597
C	15.9088249	6.4090138	13.4088363
H	16.2617867	5.4979979	12.9273227
C	15.8906897	7.9174372	15.2991399
H	16.2262644	8.1786440	16.3020183
C	15.0002739	8.7793086	14.6208793
C	14.5623969	8.4406528	13.3298151
C	14.5570077	9.9678977	15.2496316
C	14.1980578	10.9796291	15.8283019
C	13.8010301	12.1377812	16.5396631
C	14.1845456	12.2980813	17.8906504
H	14.7734302	11.5197195	18.3751055
C	13.8066117	13.4397479	18.5929447

C	13.0264540	14.4215584	17.9839317
H	12.7186538	15.3059198	18.5406211
C	12.6156556	14.2675847	16.6422809
C	13.0248002	13.1327183	15.9189925
H	12.7294859	13.0188876	14.8772290
C	11.7731408	15.2345896	16.0397802
C	11.0162309	16.0513725	15.5429774
C	10.1769727	17.0297820	14.9550843
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H	9.3944023	17.7485764	16.8426474
C	8.5896137	19.0400008	13.7930944
H	7.9849383	19.8219602	13.3358439
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C	10.1553319	17.1877392	13.5593362
C	9.3908104	18.3778951	11.5719798
C	9.4223480	18.5439126	10.3648710
C	9.4540424	18.7296980	8.9671172
C	8.5498030	19.6151686	8.3602472
H	7.8111513	20.1752787	8.9263278
N	8.5727260	19.8131186	7.0283148
C	9.4467723	19.1720382	6.2235861
H	9.3984923	19.4031086	5.1633814
C	10.3720658	18.2725835	6.7644544
C	10.3726255	18.0510781	8.1510602
H	11.0859355	17.3569361	8.5902913
C	11.2915108	17.6182122	5.9134595
C	12.0929323	17.0614747	5.1867841
C	15.0576859	15.1420096	4.0062281
H	15.9091604	14.5768816	4.3780604
C	14.9030780	15.3025134	2.6388067
N	13.8568334	16.0090364	2.1473948
C	13.7493008	16.2204231	0.6880660
C	12.9415066	16.5678767	2.9635175
H	12.1374012	17.1239809	2.4892445
C	13.0513093	16.4422074	4.3522350
C	14.1339512	15.7078339	4.8764561
H	14.2402323	15.5937370	5.9539177
H	14.1082198	13.5603137	19.6334593
H	17.0391461	6.0994291	15.2156563
H	13.8762274	9.0972571	12.7977818
H	15.5964110	14.8951329	1.9075646
H	10.7538365	16.5317640	12.9297237
H	7.9870676	19.5138381	15.8041847
H	8.2118007	8.7172492	6.9172137
H	14.1397171	15.3380612	0.1745084
H	14.3396244	17.1038255	0.4189120
H	12.6996059	16.3775442	0.4284280
C	7.6201949	20.7617225	6.4116178
H	6.9607914	21.1548906	7.1880945
H	7.0386760	20.2305367	5.6510198
H	8.1877265	21.5752171	5.9481437
C	7.6908891	7.7039585	1.7145372
H	8.6708743	7.5020683	1.2753513
H	7.2214486	8.5576734	1.2189655
H	7.0499179	6.8194120	1.6277280
C	14.8041804	3.5409944	6.5459315
H	14.8493758	3.7213458	5.4684326
H	14.2662622	2.6080053	6.7476625
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Cartesian coordinates for $(2)_3 \supset G_2$

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C	11.4228480	5.4037790	11.7158605
C	12.2166319	4.4380279	12.3124934
H	12.5044396	3.5128930	11.8190659

C	12.3783222	5.7181341	14.2858316
H	12.7764917	5.7776257	15.2949638
C	11.5916849	6.7285725	13.7244417
C	11.1020321	6.5613692	12.4134242
C	11.3050225	7.9014438	14.4580083
C	11.0364449	8.9427743	15.0252912
C	10.6448208	10.1868958	15.5655295
C	10.7723892	10.4594380	16.9339467
N	10.3473949	11.6389994	17.4326283
C	9.8098707	12.5973426	16.6477386
H	9.5061342	13.5176773	17.1385327
C	9.6658492	12.3907316	15.2685683
C	10.0931843	11.1670939	14.7270088
H	9.9778180	10.9526469	13.6642429
C	9.0859113	13.3890172	14.4584094
C	8.5558024	14.2233842	13.7451407
C	7.8933188	15.1922656	12.9559241
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H	6.8377956	16.0456766	14.6454959
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H	6.0162880	17.8463453	10.8218461
C	7.4421829	16.2154024	10.7965447
C	8.1132078	15.2548933	11.5689174
C	7.6560035	16.3077653	9.3996107
C	7.8000420	16.4276645	8.1942950
C	7.9153435	16.5675181	6.7899125
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C	8.0903469	16.8135409	3.9953044
H	8.1564333	16.9004501	2.9116299
C	8.9067576	15.8799164	4.6690498
C	8.8168143	15.7680271	6.0666289
H	9.4397447	15.0473773	6.5904597
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C	10.5700785	14.3352174	3.3313311
C	11.4802914	13.5180601	2.6189425
C	11.3276817	13.3025983	1.2322930
H	10.4856089	13.7557253	0.7109573
C	13.3331252	11.9461261	1.1990834
H	14.0596472	11.3450610	0.6540405
C	13.5042347	12.1612182	2.5835776
C	12.5656544	12.9310802	3.2896206
C	14.6295782	11.6374731	3.2609405
C	15.6164118	11.2201779	3.8422498
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C	17.0164461	11.0859403	5.8578448
H	16.3712972	11.7948324	6.3724585
C	18.4602016	10.9288070	7.8377961
C	18.7662185	11.2966202	8.9568570
N	20.3758639	11.4032230	12.2509456
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C	19.9666626	12.5988422	12.7447913
C	19.1296193	13.4123693	12.0013056
H	18.8283368	14.3764378	12.4055275
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H	18.0501500	13.6284388	10.1422370
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C	19.9911520	10.9775111	11.0337736
H	20.3775969	10.0170190	10.7038421
H	20.3433546	12.8673447	13.7288420
H	5.6240587	17.7225898	13.2694785
H	8.8050561	14.5594480	11.0972794
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H	21.4993174	9.6455367	12.5283438
H	20.7244238	10.3019206	14.0100259
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H	20.2464285	7.9176003	4.5451872
H	20.1190447	8.9403961	3.0760138
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H	14.1773239	3.1463041	13.4505921
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C	10.4341912	11.8720842	18.8928452
H	11.3145662	11.3539321	19.2805357
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C	18.9031033	7.0757747	11.7098172
H	7.3869322	9.0932563	14.7852623
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C	6.5255716	10.8083743	15.7264728
H	6.6181443	10.4538231	16.7502011
C	5.7579431	12.5476191	14.3318037
H	5.2720931	13.5189952	14.2833853
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C	6.0005778	12.4280679	11.9125824
C	5.9076903	12.8964329	10.7948311
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C	8.8786892	11.6103817	4.2548232
C	8.4033205	11.5723531	2.9262493
H	7.5047869	12.1267657	2.6584989
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C	18.3852285	7.0318256	10.4175597
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H	18.9365958	8.1277416	13.5969233
C	17.6472193	9.1036338	12.1532999
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C	17.0919762	11.1263688	13.7435259
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C	17.4838262	13.2160951	18.1263332
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C	13.3280567	18.5883960	14.7352996
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H	16.4088846	9.8196824	10.5279848
H	12.0398418	20.9898503	12.8551107
H	10.4315977	10.9663223	5.6141290
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H	7.1056824	10.0377853	12.4721724
H	11.2058099	21.0495295	15.0349582
H	12.8710481	21.4006209	15.6020795
H	11.9524838	20.0336664	16.3121070
H	17.5609239	12.1735066	18.4441736
H	16.7793046	13.7590089	18.7619101
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H	5.4808253	12.1552917	17.6056856
H	5.5861790	14.6866629	5.8572901
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H	2.9595840	15.5820902	7.7622927
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H	4.2530689	16.7931605	8.0383132
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C	15.8769590	6.4940769	14.1317100
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C	5.9930395	9.7425654	7.2461816
C	5.4160848	9.8820849	5.9959169
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C	7.3111539	8.9354553	4.9606368
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C	13.2640025	6.6127952	8.4397549
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H	11.6199600	8.0054733	8.7245594
C	13.8404800	6.8434008	9.7058824
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C	15.4959316	6.3020479	12.8056954
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C	14.2734877	8.4095214	12.8616271
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C	14.0023092	10.7902383	15.5044165
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C	14.0319930	12.0573591	17.6043183
H	14.5513249	11.2295547	18.0858279
C	13.7159112	13.2087643	18.3218189
C	13.0238968	14.2574774	17.7168156
H	12.7651244	15.1492809	18.2860789
C	12.6390642	14.1608717	16.3625021
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H	12.7067067	12.9418370	14.5768996
C	11.8816472	15.1905036	15.7525169
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H	9.6262375	17.8456115	16.4772726
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C	10.4658036	17.2309293	13.2242721
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H	16.5049493	5.7543351	14.6274400
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H	8.3558091	19.6810728	15.3856894
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H	14.0475467	14.8596203	-0.2187938
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H	5.1829564	10.6939230	3.4183236
H	4.5103464	9.0383472	3.5584967
C	13.7959880	4.3799736	5.4896938
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H	14.8665705	4.2894302	5.6877575
S	9.5171962	9.8474575	10.6347967
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F	10.0214901	12.8837686	9.2769721
C	12.3368248	12.5835512	8.8043933
C	12.2322424	13.7749419	7.7860898
F	11.6530297	14.8456270	8.3601027
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F	11.5063297	13.4179808	6.7052730
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F	10.6154167	11.1138476	8.0932225
O	9.5756091	9.5261443	12.0678170
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Cartesian coordinates for $(2)_3 \supset G_4$

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C	11.3452885	5.0814124	11.5336593
C	12.2118186	4.2712189	12.2469660
H	12.4717430	3.2604353	11.9424158
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C	11.2038838	9.2278470	14.1130355
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N	10.9181354	12.0525406	16.4380776
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H	13.4825344	13.3795333	-0.5493790
H	13.3969771	15.1260749	-0.9483423
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O	9.9776273	8.7893118	10.3434962

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