Supplementary Information:

Rotor-stator repulsion and medium-induced dephasing enhance and equalise the quantum efficiency of a fluorinated photon-only rotary motor

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Contents

1	Supplementary Note 1. List of acronyms used in the main article	S3					
2	Supplementary Note 2. Synthetic route to fluorine substituted motor	S4					
3	Supplementary Note 3. Computational Methods	S 5					
	SSR method	S5					
	SHXF method	S6					
	Computational details	S8					
4	Supplementary Note 4. Geometry optimization of the ground and excited state stationary species	S10					
5	Supplementary Note 5. Minimum energy paths	S16					
6	Supplementary Note 6. Details of QM/MM simulations	S22					
7	Supplementary Note 7. Details of NAMD simulations	S26					
8	Supplementary Note 8. The results of the gas phase and QM/MM NAMD						
	simulations	S27					
St	applementary References	S42					

1 Supplementary Note 1. List of acronyms used in the main article

CoIn Conical Intersection CCW Counterclockwise

CW Clockwise

DBPI double bond photoisomerisation eDFT ensemble density functional theory

FC Franck-Condon

LDRM Light Driven Rotary Motor MCP 3-methylene-1-cyclopentene

MMTDP *E-1-methyl-3'-(2-methyl-2,3-dihydro-1H-*

benzo[b]cyclopenta[d]thiophen-1-ylidene)pyrrolidin-

2'-one

MTDP *E-3'-*(2-methyl-2,3-dihydro-1H-

benzo[b]cyclopenta[d]thiophen-1-ylidene)pyrrolidin-

2'-one

NAMD nonadiabatic molecular dynamics

OA overcrowded alkene
PES potential energy surface

PO-LDRM Photon Only Light Driven Rotary Motor

QM/MM quantum mechanics / molecular mechanics (multi-scale

modelling technique)

SI Supporting Information

SSR State-interaction State-averaged spin-Restricted

ensemble-referenced Kohn-Sham method

THI Thermal Helix Inversion
TSH trajectory surface-hopping

2 Supplementary Note 2. Synthetic route to fluorine substituted motor

Supplementary Figure 1: Synthetic procedure for obtaining fluorine substituted molecular motor. Reagents: (i) PPA, 80 °C;¹ (ii) LiHMDS, toluene, NFSI, -78 °C;² (iii) LiHMDS, N-Boc pyrrolidin-2-one, BF3O(Et)2, THF, -78 °C; (iv) TFA, DCM.

Starting from benzothiophene and metacrylic acid, two commercially available products, as already shown in previous works¹, it is possible to synthesize the 2-methyl-2,3-dihydro-1H-benzo[b]cyclopenta[d]thiophen-1-one nucleus, **3** in Supplementary Fig. 1. The insertion of a fluorine atom in the alpha position to the carbonyl in **3** can be carried out using N-fluorobenzenesulfonimide (NFSI) in the presence of lithium bis(trimethylsilyl)amide (LiHDMS) as a base in a strongly anhydrous environment in order to obtain the racemic mixture of the fluorinated nucleus **4** by applying the synthetic procedure reported in the literature by Štacko et al..² Finally, similarly to what was done for the synthesis of the non-fluorinated compound, ^{3,4} it is possible to first condense the pyrrolidinone head using LiHDMS as a base to obtain an enantiomeric mixture of carbinol derivatives (5) which, following dehydration with trifluoroacetic acid, allow obtaining the R,S racemic mixture of the desired product (6)

equipped with the isomerizable double bond.

3 Supplementary Note 3. Computational Methods

SSR method

The SSR method $^{5-10}$ employed in this work to obtain the ground and excited electronic states energies, forces on the nuclei (the analytic gradient), and the non-adiabatic coupling (NAC) vector employs ground state eDFT $^{11-16}$ to describe the non-dynamic electron correlation occurring due to multireference character of the ground state and eDFT for ensembles of ground and excited states $^{17-20}$ to obtain excitation energies from a variational time-independent formalism. The ensemble representation of the density and energy of a strongly correlated electronic state results in occurrence of the fractional occupation numbers (FONs) of several frontier KS orbitals. Here, the SSR(2,2) method is used, where two fractionally occupied orbitals accommodate two electrons in total. The energies of the ground S_0 and excited S_1 states are obtained in SSR(2,2) from variational optimization of an ensemble of a perfectly spin-paired singlet (PPS) electronic configuration and an open-shell singlet (OSS) configuration with respect to the KS orbitals and their FONs (the SA-REKS(2,2) 7 orbital optimization), which is followed by solving a 2×2 secular problem

$$\begin{pmatrix} E_0^{PPS} & \Delta_{01}^{SA} \\ \Delta_{01}^{SA} & E_1^{OSS} \end{pmatrix} \begin{pmatrix} a_{00} & a_{01} \\ a_{10} & a_{11} \end{pmatrix} = \begin{pmatrix} E_{S_0} & 0 \\ 0 & E_{S_1} \end{pmatrix} \begin{pmatrix} a_{00} & a_{01} \\ a_{10} & a_{11} \end{pmatrix}$$
(1)

for the actual S_0 and S_1 states.^{8–10} The interstate coupling parameter Δ_{01}^{SA} is calculated simultaneously with the E_0^{PPS} and E_1^{OSS} energies when optimizing the KS orbitals (and their FONs) in the SA-REKS(2,2) calculation.^{8–10}

For the S_0 and S_1 energies, the analytic gradients are available. ^{21–23} The gradients of

the S_0 and S_1 states are related to the gradients of the PPS and OSS states and the coupling element Δ_{01}^{SA} as

$$\nabla E_{S_k} = a_{kk}^2 \nabla E_0^{PPS} + a_{lk}^2 \nabla E_1^{OSS} + 2 a_{kk} a_{lk} \nabla \Delta_{01}^{SA}; \quad l \neq k; \quad l, k = 0, 1$$
 (2)

SHXF method

The SHXF method has been described in detail elsewhere^{24,25} and here only a brief account of its basic features will be given. The SHXF method combines the electronic equations derived from the exact factorization of the electronic-nuclear wavefunction^{26–30} with the conventional TSH formalism.³¹ The exact factorization enables one to seamlessly incorporate the effect of nuclear quantum momentum, which depends on the shape of nuclear distribution, into the classical equations of motion for the nuclei.

SHXF is a mixed quantum-classical method where the electronic degrees of freedom are treated quantum mechanically and the classical description in terms of trajectories is used for the nuclear degrees of freedom. For a given nuclear trajectory $\mathbf{R}^{(I)}$, the time-dependent electronic wavefunction $\Phi_{\mathbf{R}^{(I)}}(\mathbf{r},t)$ is expanded in terms of the wavefunctions of the Born-Oppenheimer (BO) states as $\Phi_{\mathbf{R}^{(I)}}(\mathbf{r},t) = \sum_{l} C_{l}^{(I)}(t) \phi_{l}(\mathbf{r};\mathbf{R}^{(I)}(t))$ and the time evolution of an element $\rho_{lk}^{(I)}(t) = C_{l}^{(I)*}(t) C_{k}^{(I)}(t)$ of the reduced density matrix $\rho^{(I)}(t)$ is described by

$$\frac{d}{dt}\rho_{lk}^{(I)}(t) = \frac{i}{\hbar} \left\{ E_l^{(I)}(t) - E_k^{(I)}(t) \right\} \rho_{lk}^{(I)}(t) - \sum_j \left\{ \sigma_{lj}^{(I)}(t)\rho_{jk}^{(I)}(t) - \rho_{lj}^{(I)}(t)\sigma_{jk}^{(I)}(t) \right\} + \sum_j \left\{ Q_{jl}^{(I)}(t) + Q_{jk}^{(I)}(t) \right\} \rho_{lj}^{(I)}(t)\rho_{jk}^{(I)}(t) \tag{3}$$

where $E_l^{(I)}$ and $\phi_l(\mathbf{r}; \mathbf{R}^{(I)}(t))$ are the energy and the wavefunction of the l-th BO state, $\sigma_{jk}^{(I)}$ is a non-adiabatic coupling matrix element between the jth and kth BO electronic

states, and $Q_{jk}^{(I)}$ is a term that gives quantum correction to the nuclear motion arising from electronic-nuclear correlations;^{28–30} the latter term is derived from the exact factorization of the electronic-nuclear wavefunction.²⁴ In Eq. (3),the superscript (I) indicates a quantity obtained at the nuclear configuration $\mathbf{R}^{(I)}$.

In Eq. (3), the $Q_{jk}^{(I)}$ term yields coupling between the nuclear quantum momenta $i\hbar\nabla_{\nu}|\chi|/|\chi|$ and the electronic phases $\mathbf{f}_{\nu,j}^{(I)}$ and $\mathbf{f}_{\nu,k}^{(I)}$

$$Q_{jk}^{(I)} = \sum_{\nu} \frac{i\hbar}{M_{\nu}} \frac{\nabla_{\nu}|\chi|}{|\chi|} \bigg|_{\mathbf{R}^{(I)}(t)} \cdot \left(\mathbf{f}_{\nu,j}^{(I)} - \mathbf{f}_{\nu,k}^{(I)}\right)$$
(4)

where M_{ν} is a mass of the ν th nucleus. The nuclear quantum momenta are obtained from a number of auxiliary nuclear trajectories generated for electronic states other than the running state l. Although the concept of auxiliary trajectories for decoherence-induced surface hopping dynamics is not new,³² the correction used here is derived from the exact quantum equations and no further renormalisation of the electronic density matrix is needed.

An auxiliary trajectory $\mathbf{R}^{(I)}(t')$ is generated when a nonzero $\rho_{kk}(t')$ is encountered at a time t' and is evolved classically with a uniform velocity obtained from the energy conservation law. The electronic phase term $\mathbf{f}^{(I)}_{\nu,k} = -\int^t \nabla_{\nu} E^{(I)}_k(t') dt'$ is evaluated by time integration of the momentum changes at the kth BO state. To calculate the quantum momentum, a fictitious Gaussian nuclear density $|\chi_k|^2$ with a uniform variance σ is associated with the auxiliary trajectory at each BO state. Then, the nuclear quantum momentum is $\nabla_{\nu}|\chi|/|\chi|(\mathbf{R}^{(\mathbf{I})}(\mathbf{t})) = -\frac{1}{2\sigma^2}(R^{(I)}_{\nu}(t) - \langle R^{(I)}_{\nu}(t)\rangle)$, where $\langle R^{(I)}_{\nu}(t)\rangle = \sum_k \rho^{(I)}_{kk} R^{(I)}_{k\nu}(t)$. The uniform variance σ can be either obtained from the initial distribution of nuclear trajectories or set as a parameter. Here, a uniform value $\sigma=0.1$ a.u. is used.

Nuclear trajectories follow the Newtonian equations of motion on the PES of the running state l, *i.e.*, the force on the ν th nucleus is $\mathbf{F}_{\nu}^{(I)} = -\nabla_{\nu} E_{l}^{(I)}$. When the surface

hop probability becomes greater than a random number, a surface hop is initiated. The surface hop probability from the running state l to another state $k(\neq l)$ at a time interval $[t,t+\Delta t]$ is calculated as

$$P_{l\to k} = \frac{2\Re\left[\rho_{lk}^{(I)}(t)\sigma_{lk}^{(I)}(t)\right]}{\rho_{ll}^{(I)}(t)}\Delta t, \qquad (5)$$

where the same constraints as in the Tully's fewest switches algorithm^{31,33} are imposed on the hopping transition; *i.e.*, when $E_k^{(I)}$ is greater than the total energy, the transition is forbidden and, when a negative hopping probability is obtained (*e.g.*, due to the phase factor of $\rho_{lk}^{(I)}(t)$), the probability is set to zero. Upon a successful surface hop, the nuclear velocities are rescaled to satisfy the total energy conservation, and the running state is switched to the new state.

Computational details

All the quantum chemical computations were carried out using the local version of the GAMESS-US program (2018.v3), ^{34,35} which implements the SSR method and the analytic derivatives formalism. ²¹ All the calculations employed the 6-31G* basis set ³⁶ and the BH&HLYP exchange-correlation density functional. ^{37–39} The geometry optimizations were performed using the DL-FIND module ⁴⁰ interfaced with GAMESS-US.

The QM/MM NAMD simulations were performed by the pyUNIxMD program, 25,41 a standalone code which implements the SHXF method. 24 The energy gradients ∇E_{S_k} (k=0,1) were calculated by the GAMESS-US program. The nonadiabatic couplings were obtained using the Baeck-An method, ${}^{42-45}$ which operates with the energies of the ground and excited states alone. The nuclear equations of motion were integrated using the velocity-Verlet algorithm with the time step of 0.5 fs. The electronic equations of motion (3) were integrated by the 4-th order Runge-Kutta method with the time step of 0.025 fs. When integrating the electronic equations of motion, the electronic energies $E_l^{(I)}$

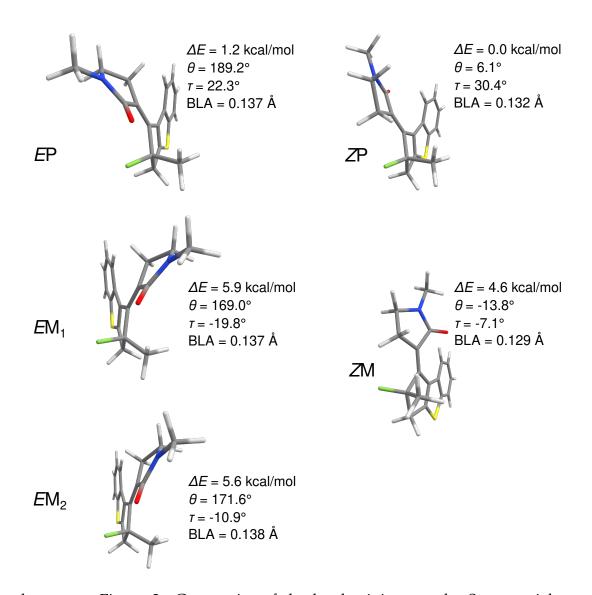
and the nonadiabatic couplings $\sigma_{lk}^{(I)}$ were linearly interpolated between the end points of the integration interval of the nuclear equations of motion.

The QM/MM calculations were performed by the GAMESS-US program interfaced with TINKER program v6.3. 46 The force field calculations of the solvent molecules employed the OPLS-AA force field. 47

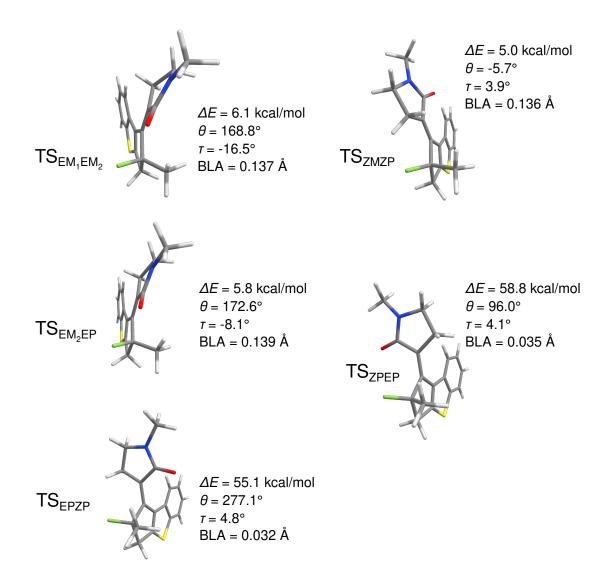
4 Supplementary Note 4. Geometry optimization of the ground and excited state stationary species

As a reference, the reaction path and dynamics of F-MMTDP were first studied under isolated conditions. The geometries of the S_0 and S_1 stationary points have been optimised using the DL-FIND code⁴⁰ interfaced with GAMESS-US. The convergence criteria used in the optimization of the S_0 and S_1 stationary geometries were set to $\Delta E = 10^{-6}$ hartree and $\Delta |\partial E/\partial r_k| = 4.5 \cdot 10^{-4}$ hartree/bohr. The geometries of the S_1/S_0 conical intersections were optimized using the penalty function algorithm as implemented in DL-FIND code,⁴⁰ where the penalty function constants C_1 and C_2 were set to default values (5 kcal mol⁻¹).⁴⁰

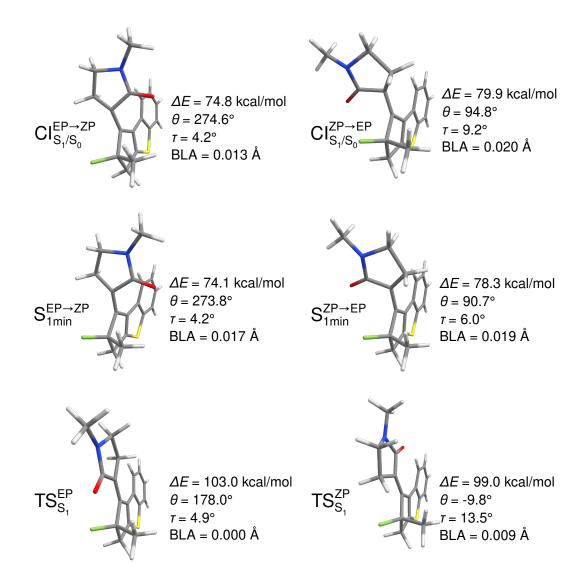
The optimized geometries are shown in the graphics form in Supplementary Figs. 2, 3, and 4. Their relative energies and selected geometric parameters are collected in Supplementary Table 1. According to SSR geometry optimisation (see Supplementary Figs. 2, 3, and 4 and Supplementary Table 1) the most stable S_0 structure is 3-ZP (with θ = 6.1° and τ = 30.4°; its energy is taken as the origin of the energy scale) and 3-EP (θ = 189.2°, τ = 22.3°) is 1.2 kcal/mol less stable. This is different from MTDP and MMTDP, where EP was the most stable structure (by \approx 4 kcal mol⁻¹). 4,48,49 The difference in relative stabilities of the E and E structures is explained by the effect of electrostatic repulsion between the carbonyl oxygen and fluorine atom at the E chiral centre, which was absent in MTDP and MMTDP.



Supplementary Figure 2: Geometries of the local minima on the S_0 potential energy surface of 3 optimised using the SSR-BH&HLYP/6-31G* method. The relative energies (kcal/mol, with respect to the ZP structure) and some important geometric characteristics are given in the insets. See Supplementary Figure 6 for positions of the optimised structures on the global potential energy surface of the motor 3.



Supplementary Figure 3: Geometries of the transition states on the S_0 potential energy surface of 3 optimised using the SSR-BH&HLYP/6-31G* method. The relative energies (kcal/mol, with respect to the ZP structure) and some important geometric characteristics are given in the insets. See Supplementary Figure 6 for positions of the optimised structures on the global potential energy surface of the motor 3.



Supplementary Figure 4: Geometries of the minimum energy S_1/S_0 conical intersections, local minima and transition states on the S_1 potential energy surface of 3 optimised using the SSR-BH&HLYP/6-31G* method. The relative energies (kcal/mol, with respect to the ZP structure) and some important geometric characteristics are given in the insets. See Supplementary Figure 6 for positions of the optimised structures on the global potential energy surface of the motor 3.

Supplementary Table 1: Relative energies (in kcal/mol) and selected geometric parameters of the stationary points on the S_0 and S_1 potential energy surfaces of the motor 3 optimised using the SSR-BH&HLYP/6-31G* method. The central dihedral angle θ , tilting angle τ and Cremer-Pople puckering angle are given in degrees (deg.). The bond length alternation and puckering amplitude are given in Å. See Supplementary Figures 2, 3, and 4 for graphical representation of the structures.

·						pyrrolidinone ^b		cyclopentene	
structure	$\Delta E(S_0)^a$	$\Delta E(S_1)$	θ	τ	BLA	q	φ	q	φ
$EP(S_{0,min})$	1.2	110.0	189.2	22.3	0.1368	0.275	351.7	0.209	358.7
$EM_1(S_{0,min})$	5.9	112.4	169.0	-19.8	0.1373	0.287	354.0	0.209	355.6
$EM_2(S_{0,min})$	5.6	112.1	171.6	-10.9	0.1380	0.244	349.7	0.070	348.6
$ZP(S_{0,min})$	0.0	106.4	6.1	30.4	0.1319	0.217	349.0	0.297	359.0
$ZM(S_{0,min})$	4.6	107.5	-13.8	-7 .1	0.1292	0.286	353.9	0.076	205.0
$TS_{EPZP}(S_0)$	55.1	80.3	277.1	4.8	0.0323	0.021	36.6	0.037	4.4
$TS_{ZPEP}(S_0)$	58.8	83.8	96.0	4.1	0.0351	0.125	330.0	0.114	14.5
$TS_{EM_1EM_2}(S_0)$	6.0	112.1	168.8	-16.5	0.1369	0.279	352.2	0.140	356.1
$TS_{EM_2EP}(S_0)$	5.8	112.4	172.6	-8.1	0.1385	0.217	345.6	0.025	280.8
$TS_{ZMZP}(S_0)$	5.0	110.8	-5.7	3.9	0.1361	0.303	2.6	0.122	3.3
$CI_{c}^{EP\to ZP}$	74.8	74.8	274.6	4.2	0.0128	0.347	342.5	0.125	4.2
CI_{S_1/S_0} $CI_{S_1/S_0}^{ZP \to EP}$	79.9	79.9	94.8	9.2	0.0203	0.374	349.1	0.220	356.8
$C_{S_1/S_0}^{Z_1 \to Z_1}$ $S_{1,min}^{EP \to ZP}$ $S_{2P \to EP}^{ZP \to EP}$	65.9	74.1	273.8	4.2	0.0166	0.322	336.9	0.117	7.2
$S_{1,min}^{I,min}$ EP	66.8	78.3	90.7	6.0	0.0192	0.313	338.5	0.133	4.3
TS ^{EP}	15.5	103.0	178.0	4.9	0.0002	0.141	313.1	0.123	28.4
$TS_{S_1}^{S_1P}$	12.7	99.0	-9.8	13.5	0.0096	0.230	331.3	0.224	7.3

^{a)} All energies are given with respect to the most stable ground state structure, ZP. The energy of the optimised structure is highlighted. The other energy is obtained by a vertical excitation/deexcitation.

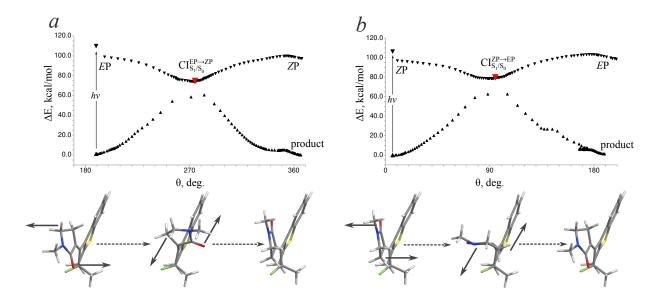
In MTDP and MMTDP, no stable M-helical structures could be obtained in the S_0 state, and the respective regions on the S_0 PES could be best characterised as inflexion points. 4,48,49 However, in F-MMTDP the ground state species with M helicity occur as shoulders on the S_0 PES at ≈ 5 kcal/mol above the respective P-minima, and they correspond to very shallow local minima separated from the P-minima by barriers of

^{b)} Cremer-Pople ring puckering parameters ⁵⁰ of the respective 5-membered rings, the puckering amplitude q and puckering angle ϕ . The puckering coordinates are pivoted at the $C_{4'}$ atom for the pyrrolidinone ring and C_7 atom for the cyclopentene ring. See also atomic numbering in Supplementary Figures 19 and 20.

less than 0.4 kcal/mol; see Supplementary Table 1. With such low barriers, these minima become kinetically unstable because the energy of photoexcitation is on the order of 100 kcal/mol and, during the photodynamics of the motor, this energy is released to the internal vibrational degrees of freedom; thus, heating the motor molecule and enabling easy passage through the shoulders on the S_0 PES. Therefore, F-MMTDP shares, with MTDP and MMTDP, the ability to isomerise without stumbling upon the M-helical structures.

5 Supplementary Note 5. Minimum energy paths

Using the geometries of the stationary species on the S_0 and S_1 potential energy surfaces (PESs), optimisation of minimum energy paths connecting the stationary points has been carried out using the nudged elastic band (NEB) method⁵¹ as implemented in the DL-FIND program.⁴⁰ When optimizing the MEPs, the stationary species were first arranged in the order of increasing central dihedral angle θ (for definition of θ , see Fig. 3c in the main article). Then the MEPs between the stationary points were optimised separately on the S_0 and S_1 PESs using 30 to 50 intermediate images between the stationary geometries.



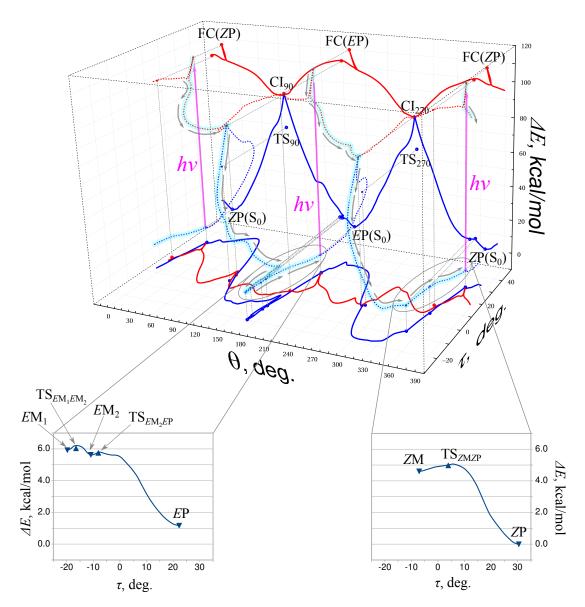
Supplementary Figure 5: Projection of the adiabatic minimum energy paths (MEPs) of the F-MMTDP motor in gas phase onto θ (see Fig. 3c in the main article for the definition of θ) calculated with the SSR-BH&HLYP/6-31G* method. The MEPs optimised for the S₁ state are shown by downward pointing triangles and for the S₀ state by the upward pointing triangles. The structures below the plot show the geometries of the reactants (EP or ZP), the conical intersections (CIs), and the product (ZP or EP). The arrows indicate a CCW motion. Panel (a) shows the EP \rightarrow ZP photoreaction and panel (b) the ZP \rightarrow EP photoreaction. See Supplementary Fig. 6 for a more detailed representation of the MEPs.

Supplementary Fig. 5 shows the potential energy profiles of the S_1 and S_0 states along the minimum energy paths (MEPs) describing the $EP \to ZP$ (Supplementary Fig. 5a) and $ZP \to EP$ (Supplementary Fig. 5b) photochemical steps. The MEPs are computed

using the nudged elastic band (NEB) method⁵¹ starting in the respective ground state equilibrium structures and propagating towards the S_1/S_0 CIs shown in Supplementary Fig. 5 by downward-pointing red triangles. The MEPs in Supplementary Fig. 5 are shown as functions of the central dihedral angle θ alone. A complete view of the MEPs is shown in Supplementary Figs. 6 and 7, where it is seen that, along the MEPs, a substantial variation of the tilting angle τ occurs in the range between -20° and $+30^{\circ}$. However, in the regions on the ground-state PES where M-helical structures occur, the energy profile is essentially flat (see the insets in the Supplementary Fig. 6) and no tall barriers obstruct the passage to the P-helical minima. This is also seen in Supplementary Fig. 5, where the shoulders corresponding to M-helical species occur on the S_0 PES near the P-helical minima.

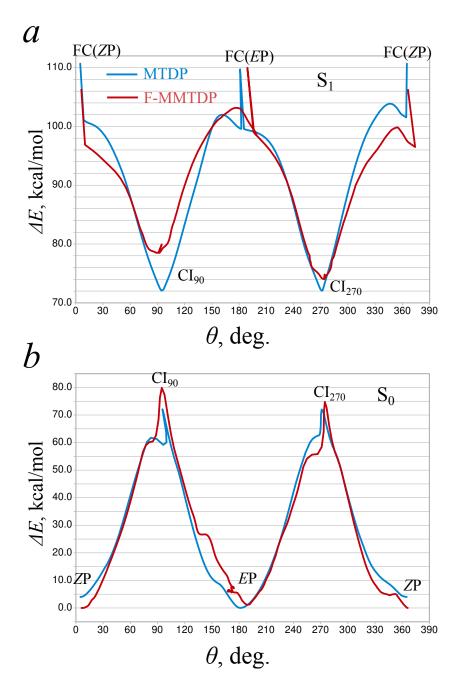
In equilibrium ground state geometries, EP and ZP, photoexcitation to the S_1 state is achieved by a $\pi \to \pi^*$ one-electron transition, which is centred around the $C_6=C_{3'}$ axle and has the vertical excitation energy of 4.72 eV for EP and 4.61 eV for ZP; see Supplementary Table 1. In both structures, EP and ZP, photoexcitation corresponds to a local transition as confirmed by a small variation in the charge separation between the pyrrolidinone and benzocyclopentathiophene units upon the $S_0 \to S_1$ excitation; EP: from $0.05\bar{e}$ (S_0) to $0.13\bar{e}$ (S_1), and ZP: from $0.05\bar{e}$ (S_0) to $0.17\bar{e}$ (S_1).

As a result of the $\pi \to \pi^*$ transition, the $C_6 = C_{3'}$ π -bond is broken and this allows rotation about the axle. However, the first geometric change that occurs along the MEPs is the change in the bond length alternation (BLA) along the $C_{3'} = C_6 - C_{5b} = C_{8a}$ chain (BLA = $R_{C_6 - C_{5b}} - \frac{1}{2}[R_{C_{3'} = C_6} + R_{C_{5b} = C_{8a}}]$). In the S_0 equilibrium geometry, BLA in the EP and ZP structures have values of 0.137 Å and 0.132 Å, respectively, which means that the double bonds along the chain are (on average) shorter than the single bond. However, at the next point along the MEP on the S_1 PES, which differs only slightly in the dihedral angle θ from the Franck-Condon (FC) geometry (by ≈ 4 –6°), the BLA value drops to a near zero value for both structures, EP and ZP. This geometric change is not reflected in the graphs



Supplementary Figure 6: Minimum energy paths (MEPs) optimised using the nudged elastic band (NEB) method 51 for the F-MMTDP motor in the gas phase. The MEPs connect the stationary points, minima, transition states (TSs), and conical intersections (CIs), optimised using the SSR-BH&HLYP/6-31G* method. The MEPs are shown as functions of the central dihedral angle θ and tilting angle τ . The red curves show the MEPs optimised for the S_1 state and the blue curves for the S_0 state. The dashed curves show the 3D representation of the MEPs, and the solid curves show projections of the 3D MEPs onto the respective planes. The photoisomerisation path followed by the motor is highlighted on the 3D MEPs by the cyan colour and grey arrows. The insets zoom in on the flat areas on the MEPs designated by grey ellipses.

in Supplementary Fig. 5, however, it should be kept in mind that the elevation of the S_1 FC energy (the black downward-pointing triangles in Supplementary Fig. 5) above the



Supplementary Figure 7: Comparison between the MEPs optimised for the S_1 (panel a) and S_0 (panel b) states of the F-MMTDP motor (the red curves) and the original unsubstituted MTDP motor (the blue curves) in the gas phase. The MEPs are given as functions of the central dihedral angle θ .

MEP curves is caused by the change in BLA, without much impacting the torsion angle θ .

Upon undergoing the initial drop in the BLA value, the motor 3 occurs on a slope on the S_1 PES inclined towards increasing angle θ . This facilitates torsion in the CCW

direction towards the S_1/S_0 CIs shown in Supplementary Fig. 5 by the red triangles pointing downward. The driving force for the torsion can be evaluated from the slope of the S_1 PES in the vicinity of the FC geometry and amounts to -0.134 kcal mol^{-1} deg⁻¹ ($2.02 \cdot 10^{-4}$ eV rad^{-1}) for the ZP structure and to -0.197 kcal mol^{-1} deg⁻¹ ($2.97 \cdot 10^{-4}$ eV rad^{-1}) for the EP structure. Compared to the parent MTDP motor 1, the driving force at the FC point increases (in absolute value) almost fourfold for the EP structure (from -0.044 kcal mol^{-1} deg⁻¹ to -0.197 kcal mol^{-1} deg⁻¹), which results solely from the $C_{2'}$ = $O \cdot \cdot \cdot F$ - C_7 intramolecular repulsion. There is also an approximately 36% increase in the driving force at the ZP FC geometry, from -0.098 kcal mol^{-1} deg⁻¹ in MTDP to -0.134 kcal mol^{-1} deg⁻¹ in F-MMTDP. Therefore, the motor 3 should possess a stronger propensity to torsion in the desired direction (CCW) than its predecessors, motors 1 and 2.

Sliding down the slope on the S_1 PES the motor 3 reaches the S_1/S_0 CIs, which are located near the torsion angle $\theta \approx 270^\circ$ (for $EP \to ZP$) and $\approx 90^\circ$ (for $ZP \to EP$); see Supplementary Fig. 5. Both CIs occur near local minima on the S_1 PES, $S_{1,min}^{EP\to ZP}$ and $S_{1,min}^{ZP\to EP}$ in the Supplementary Table 1, and are elevated 0.7–1.6 kcal mol⁻¹ above the bottom of the minima. The S_1 state near both the minima and the CIs has charge transfer (CT) characteristics, which are evidenced by the charge separation of \pm 0.55–0.56 \bar{e} between the stator (positively charged) and the rotator (negatively charged) units.

The S_1 minima and the CI have the pyrrolidinone ring noticeably puckered, where the Cremer-Pople puckering amplitude 50 q reaches values of ≈ 0.32 –0.37 Å; see Supplementary Table 1. The pyrrolidinone ring in these geometries attains an envelope conformation (between 1T_5 and 1E) with the puckering angle ≈ 340 –350°. The S_0 state near these structures shows diradical (DIR) characteristics with a very low charge separation of only $\approx 0.1\bar{e}$. Torsional transition state structures occur on the S_0 PES near the torsion angle θ 277° and 96° at energies almost 20 kcal mol⁻¹ below the S_1 local minima and respective CIs. These TSs have almost flat pyrrolidinone rings with

 $q \lesssim 0.1$ Å; see Supplementary Table 1. Therefore, the puckering distortion destabilises the ground electronic state, which results in crossing with the S_1 state and the occurrence of CIs. Consequently, the S_1/S_0 CIs occurring in 3, and 1 and 2, can be described as the torsion–puckering CIs, as opposed to torsion–pyramidalisation and torsion–BLA CIs observed previously in other LDRM designs. $^{4,52-55}$ It is noteworthy that pyrrolidinone pseudo-rotation caused by puckering occurs during the excited state dynamics in all three motors, 1–3, and may affect the outcome of the photoisomerisation. 49 However, a detailed study of this phenomenon goes beyond the scope of the present work and is deferred to a future study.

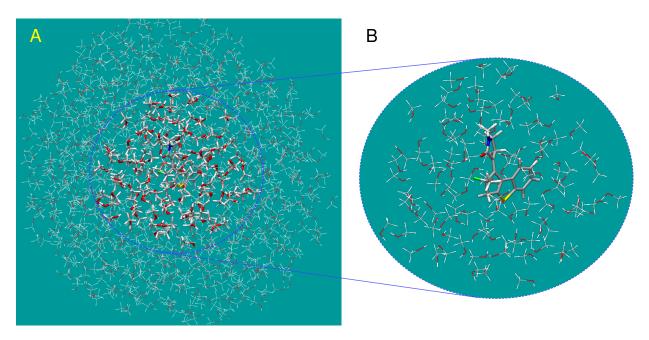
Having crossed the S_1/S_0 CIs and relaxed to the S_0 state, the motor 3 slides down the S_0 PES towards the isomerised structures; see Supplementary Fig. 5. Typically, branching of the reaction paths occurs at the CI structures, where a fraction of the trajectories in the ensemble continue towards the reaction products and another turns back towards the reactant structures. An accurate fraction of the productive (i.e., propagating towards the products) trajectories in the ensemble can only be determined in molecular dynamics simulations because the photoisomerisation occurs on an ultrafast timescale (≈ 1 ps), which requires the application of NAMD formalism.

Let us summarise the main findings in this section: Substitution in 3 results in a steeper slope of the S_1 PES near the FC geometries and a greater driving force in the desired direction of torsion (CCW). This is expected to eliminate a partial loss of the unidirectionality of rotation, which was observed in the previous two-stroke photon-only LDRMs 1 and $2.^{4,48,49}$ At the same time, 3 should retain the ability to perform a full 360° loop by consuming light only because there are no sufficiently stable intermediate structures with the opposite helicity, overcoming which would require thermally activated ground state helix inversion steps.

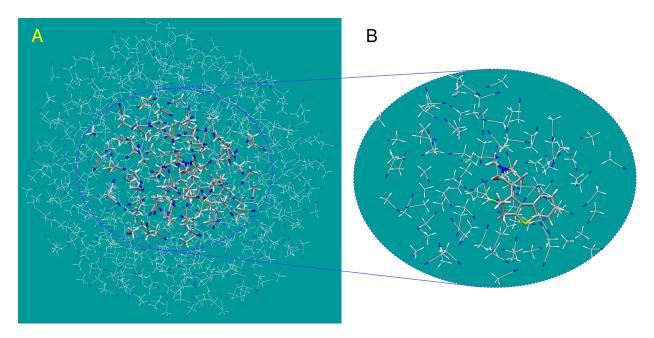
6 Supplementary Note 6. Details of QM/MM simulations

The QM/MM NAMD simulations of F-MMTDP photodecay in three solvents (MeOH, MeCN, and DCM) have been prepared using the protocol described in Ref. 4: After the initial preparation of the target solute molecules, which employed the MP2/6-31G* calculations in connection with the suitably modified OPLS-AA force field, ⁴⁷ a 5000 ps long molecular dynamics (MD) simulation has been performed under the isothermal-isobaric ensemble (NPT) with the pressure set at 1 atm and temperature at 298 K using the GROMACS software package. ⁵⁶ During the first 1 ns, the system was gradually thermalised to 298K and equilibrated and, in the subsequent 4 ns, 400 snapshots were selected with 10 ps interval. Each snapshot comprised an approximately 20 Å solvent sphere, which accommodated 906 solvent molecules in the case of MeOH, 668 molecules in MeCN, and 1044 molecules in DCM, see Supplementary Figs. 8, 9, and 10.

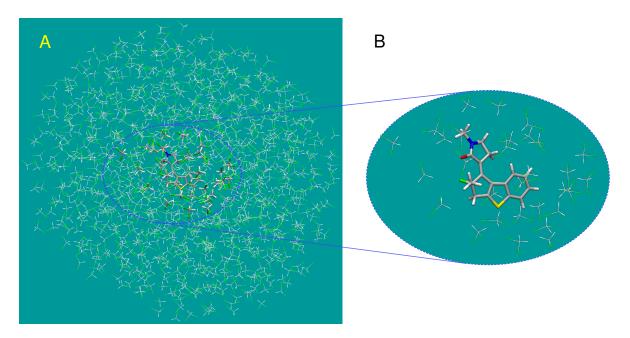
The snapshots were used in the construction of the QM/MM models, which employed frozen boundary conditions. ^{57,58} In the frozen boundary, the solvent molecules have been kept fixed and only the solvent molecules within a sphere of approximately 4 Å radius around the solute molecule have been thawed. For each snapshot, the geometries of the solute molecule and the inner solvent shell have been relaxed in a series of 5 ps QM/MM MD simulations performed in the ground electronic state of the solute molecule at 300K. The MD simulations employed the R-BH&HLYP/6-31G* method in connection with the modified OPLS-AA force field for the solvent molecules. The final snapshots from these MD trajectories were used to initiate the NAMD simulations of F-MMTDP.



Supplementary Figure 8: The solvent box used in the NAMD QM/MM simulations of the motor 3 in MeOH solution. Panel (A) shows a full box of 906 MeOH molecules and 3. The circle in the panel (A) highlights the solvation shell of the motor 3, where the solvent molecules (100 MeOH molecules modelled by MM) and solute molecule (modelled by QM) have been moved during the simulations. The remaining solvent molecules were frozen during the simulations; i.e., the frozen boundary conditions were used. The inset in panel (B) zooms in into the active solvation shell highlighted in panel (A).



Supplementary Figure 9: The solvent box used in the NAMD QM/MM simulations of the motor 3 in MeCN solution. Panel (A) shows a full box of 668 MeCN molecules and 3. The circle in the panel (A) highlights the solvation shell of the motor 3, where the solvent molecules (100 MeCN molecules modelled by MM) and solute molecule (modelled by QM) have been moved during the simulations. The remaining solvent molecules were frozen during the simulations; i.e., the frozen boundary conditions were used. The inset in panel (B) zooms in into the active solvation shell highlighted in panel (A).



Supplementary Figure 10: The solvent box used in the NAMD QM/MM simulations of the motor 3 in DCM solution. Panel (A) shows a full box of 1044 DCM molecules and 3. The circle in the panel (A) highlights the solvation shell of the motor 3, where the solvent molecules (30 DCM molecules modelled by MM) and solute molecule (modelled by QM) have been moved during the simulations. The remaining solvent molecules were frozen during the simulations; i.e., the frozen boundary conditions were used. The inset in panel (B) zooms in into the active solvation shell highlighted in panel (A).

7 Supplementary Note 7. Details of NAMD simulations

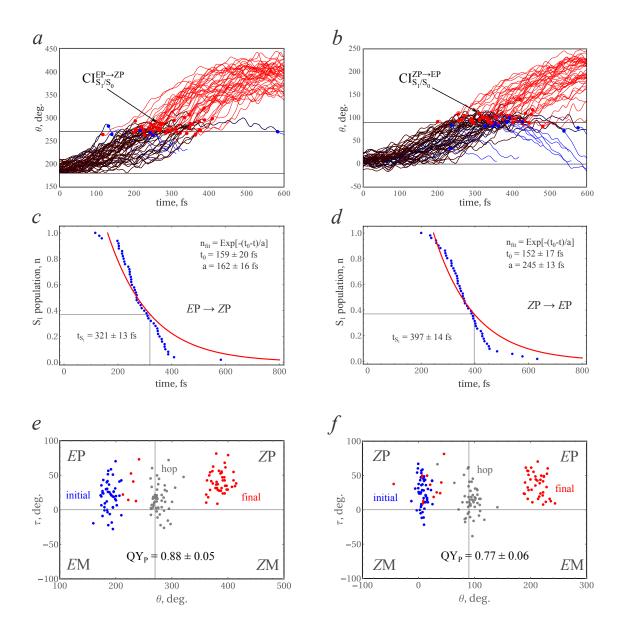
The NAMD simulations, both gas phase and QM/MM, employed the SSR(2,2)-BH&HLYP/6-31G* method for obtaining the electronic structure of the solute molecules in the ground and excited states. The initial geometries and nuclear velocities were taken from the previous ground-state MD simulations. Because the SSR/REKS and traditional KS DFT methods yield very close results for the ground states of molecules, ^{9,10} the use of the initial conditions obtained with KS DFT in lieu of SSR/REKS is justified. The NAMD simulations were carried out using the SHXF method ^{24,25} implemented in the pyUNIxMD program. ^{25,41}

In the gas phase NAMD simulations, 50 NAMD trajectories were initiated in the S_1 state in each ground state stationary structure, *E*Pand *Z*P. The trajectories were propagated until the final structure in the S_0 state has been reached. The final S_0 structure was deemed stable, if the dihedral angles θ and τ remained in the range of values characteristic for the *E*Pand *Z*Pstructures (*i.e.*, $\theta \approx 180^{\circ}$ and $\tau > 0^{\circ}$ for *E*P, and $\theta \approx 0^{\circ}$ and $\tau > 0^{\circ}$ for *Z*P) for approximately 500 fs or longer. Then, the trajectory was terminated. The gas phase simulations have been performed under the microcanonical ensemble (NVE) conditions.

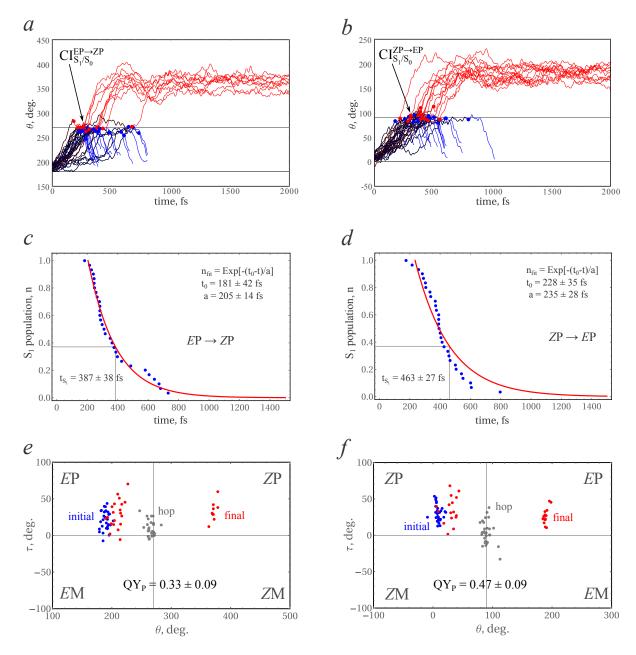
In QM/MM NAMD simulations, 30 trajectories have been initiated in the S_1 state in each ground state stationary structure, EPand ZP, and propagated using the SSR(2,2)-BH&HLYP/6-31G* method for a maximum duration of 1500 fs or until the $S_1 \rightarrow S_0$ transition occurred. After the $S_1 \rightarrow S_0$ surface hop, the trajectory has been allowed to continue on the S_0 PES until a S_1 - S_0 energy gap became $\gtrsim 3$ eV. Then, the trajectory has been continued on the S_0 PES using the conventional RKS-BH&HLYP/6-31G* method. Because the SSR and the traditional RKS methods yield very close energies for the ground states of the closed-shell single-reference molecules, this procedure is justified. Typically, the energy difference between the SSR and RKS methods for the same ground state geometry along a trajectory is $\lesssim 1.5$ kcal mol⁻¹. The RKS-BH&HLYP/6-31G*

QM/MM NAMD simulations have been continued until a stable final structure has been reached. The NAMD and the ground-state MD simulations were performed under the microcanonical ensemble (NVE) conditions.

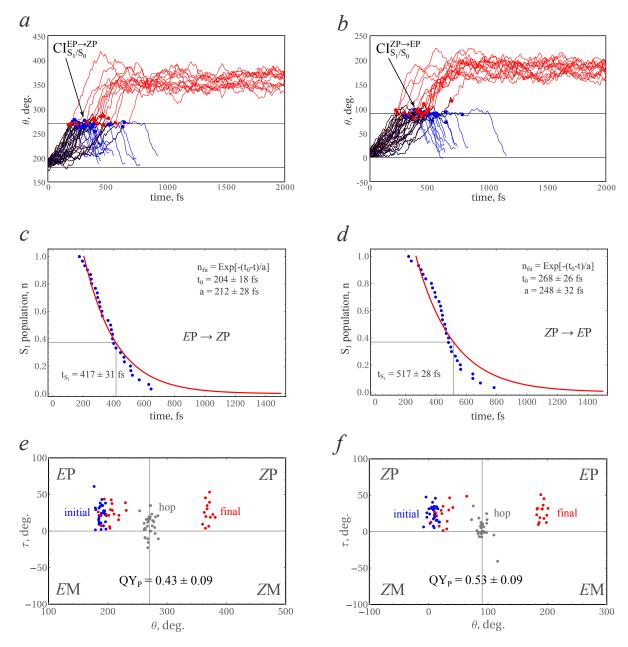
8 Supplementary Note 8. The results of the gas phase and QM/MM NAMD simulations



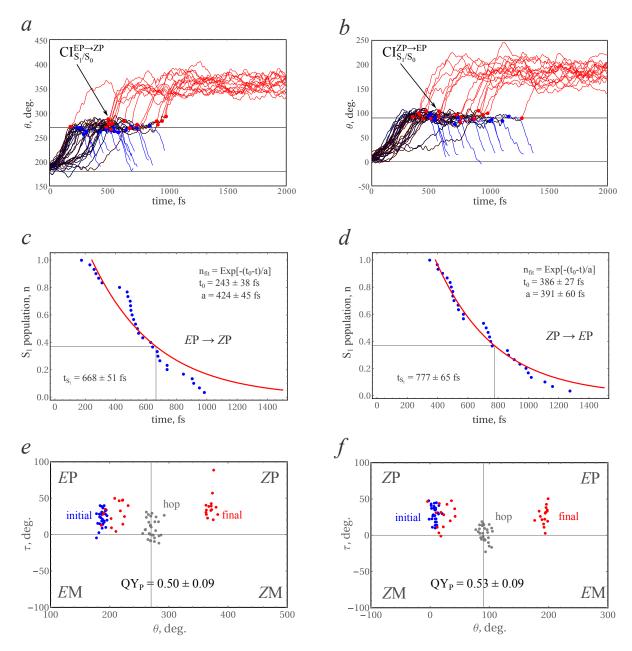
Supplementary Figure 11: Dynamics of the F-MMTDP motor in the gas phase. Panels (a) and (b) show the time evolution of the central dihedral angle θ for the EP \to ZP and ZP \to EP photoreactions, respectively. Panels (c) and (d) show the population dynamics of the S₁ state for the two photoreactions. The red curves are fitted to the computed populations (the blue dots) by a monoexponential function $n_{fit}(t) = e^{-(t_0 - t)/\tau}$, where t_0 is the latency time and τ is the exponential decay constant. The excited-state lifetimes $t_{\rm S_1}$ are shown in the insets. The margins of error of the fitted quantities were obtained by bootstrapping with 10^4 replicas. Fanels (e) and (f) show the distribution of the dihedral angles θ and τ for the EP \to ZP and ZP \to EP photoreactions, respectively. The initial structures are shown by the blue dots, the final structures by the red dots, and the structures at the moment of the S₁ \to S₀ transition (hops) by the grey dots. The quantum yields of photoisomerisation for both photoreactions are shown in the insets.



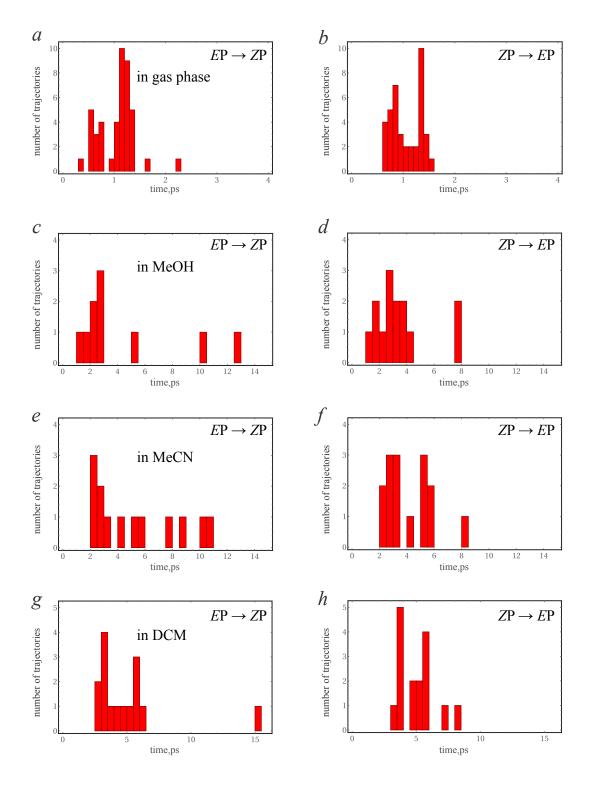
Supplementary Figure 12: Dynamics of the F-MMTDP motor in MeOH. Panels (a) and (b) show the time evolution of the central dihedral angle θ for the EP \to ZP and ZP \to EP photoreactions, respectively. Panels (c) and (d) show the population dynamics of the S₁ state for the two photoreactions. The red curves are fitted to the computed populations (the blue dots) by a monoexponential function $n_{fit}(t) = e^{-(t_0 - t)/\tau}$, where t_0 is the latency time and τ is the exponential decay constant. The excited-state lifetimes $t_{\rm S_1}$ are shown in the insets. The margins of error of the fitted quantities were obtained by bootstrapping with 10⁴ replicas. Fanels (e) and (f) show the distribution of the dihedral angles θ and τ for the EP \to ZP and ZP \to EP photoreactions, respectively. The initial structures are shown by the blue dots, the final structures by the red dots, and the structures at the moment of the S₁ \to S₀ transition (hops) by the grey dots. The quantum yields of photoisomerisation for both photoreactions are shown in the insets.



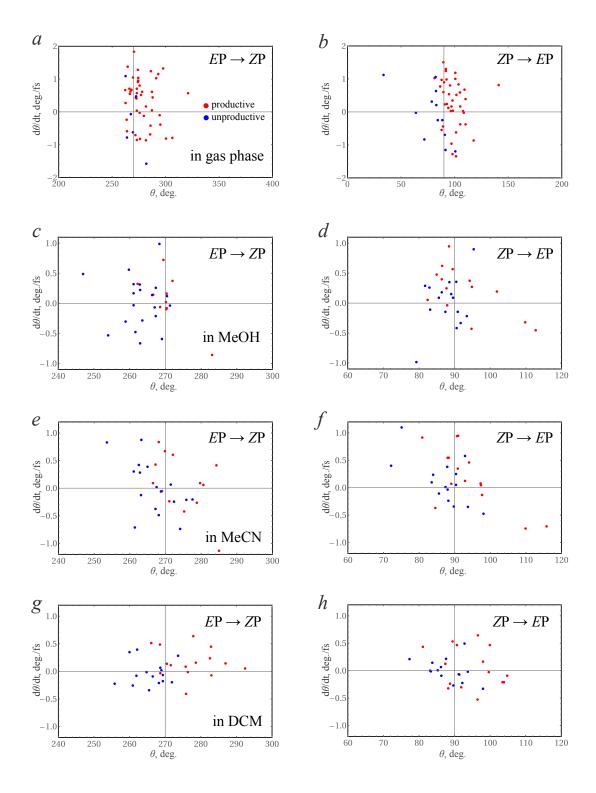
Supplementary Figure 13: Dynamics of the F-MMTDP motor in MeCN. Panels (a) and (b) show the time evolution of the central dihedral angle θ for the EP \to ZP and ZP \to EP photoreactions, respectively. Panels (c) and (d) show the population dynamics of the S₁ state for the two photoreactions. The red curves are fitted to the computed populations (the blue dots) by a monoexponential function $n_{fit}(t) = e^{-(t_0 - t)/\tau}$, where t_0 is the latency time and τ is the exponential decay constant. The excited-state lifetimes $t_{\rm S_1}$ are shown in the insets. The margins of error of the fitted quantities were obtained by bootstrapping with 10^4 replicas. ⁵⁹ Panels (e) and (f) show the distribution of the dihedral angles θ and τ for the EP \to ZP and ZP \to EP photoreactions, respectively. The initial structures are shown by the blue dots, the final structures by the red dots, and the structures at the moment of the S₁ \to S₀ transition (hops) by the grey dots. The quantum yields of photoisomerisation for both photoreactions are shown in the insets.



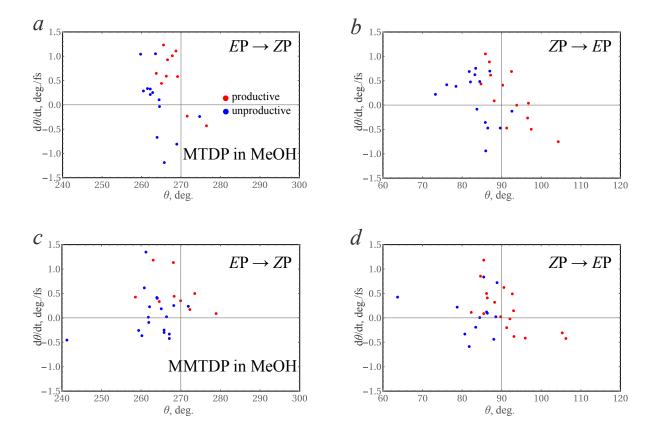
Supplementary Figure 14: Dynamics of the F-MMTDP motor in DCM. Panels (a) and (b) show the time evolution of the central dihedral angle θ for the EP \to ZP and ZP \to EP photoreactions, respectively. Panels (c) and (d) show the population dynamics of the S₁ state for the two photoreactions. The red curves are fitted to the computed populations (the blue dots) by a monoexponential function $n_{fit}(t) = e^{-(t_0 - t)/\tau}$, where t_0 is the latency time and τ is the exponential decay constant. The excited-state lifetimes $t_{\rm S_1}$ are shown in the insets. The margins of error of the fitted quantities were obtained by bootstrapping with 10^4 replicas. Fanels (e) and (f) show the distribution of the dihedral angles θ and τ for the EP \to ZP and ZP \to EP photoreactions, respectively. The initial structures are shown by the blue dots, the final structures by the red dots, and the structures at the moment of the S₁ \to S₀ transition (hops) by the grey dots. The quantum yields of photoisomerisation for both photoreactions are shown in the insets.



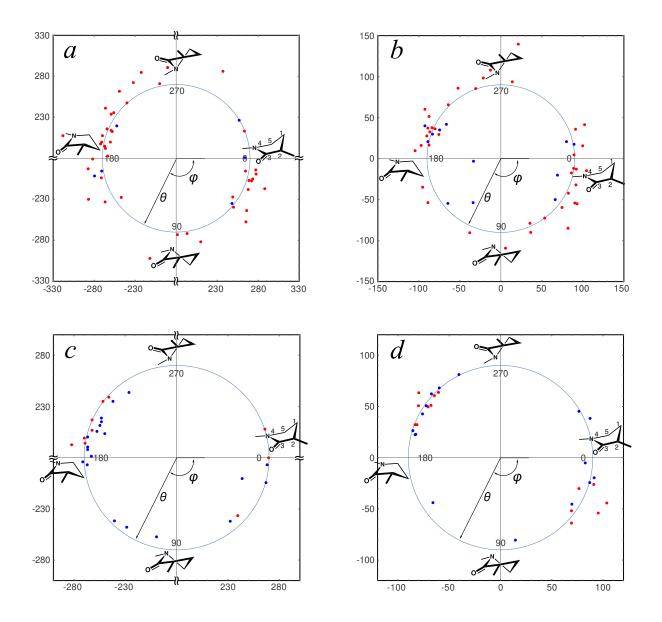
Supplementary Figure 15: Distributions of the time-of-arrival (TOA) at the final structure for the trajectories simulated in the (a) gas phase, (b) MeOH solution, (c) MeCN solution, and (d) DCM solution.

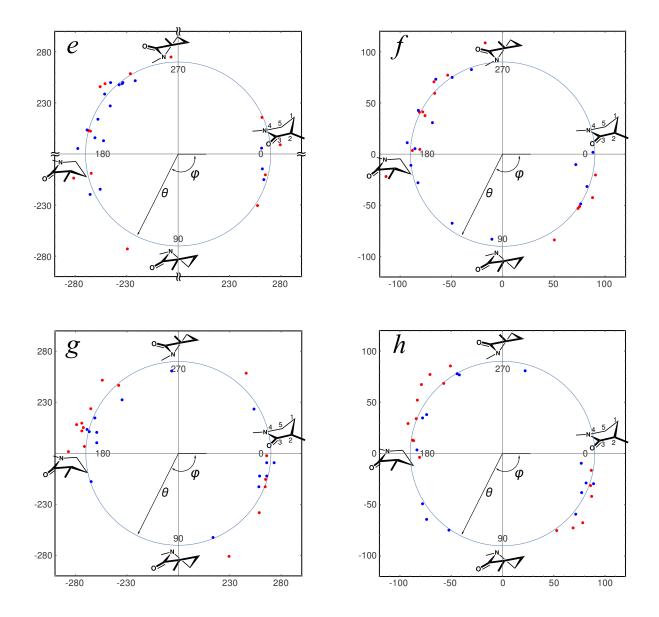


Supplementary Figure 16: Distributions of the central dihedral angle θ and angular velocity $d\theta/dt$ at the time of the $S_1 \to S_0$ transition (surface hop) for the EP \to ZP and ZP \to EP photoreactions. Panels (a) and (b) show the gas phase reactions, (c) and (d) show reactions in MeOH, (e) and (f) in MeCN, and (g) and (h) in DCM. The red dots show the productive trajectories and the blue dots show the unproductive trajectories.

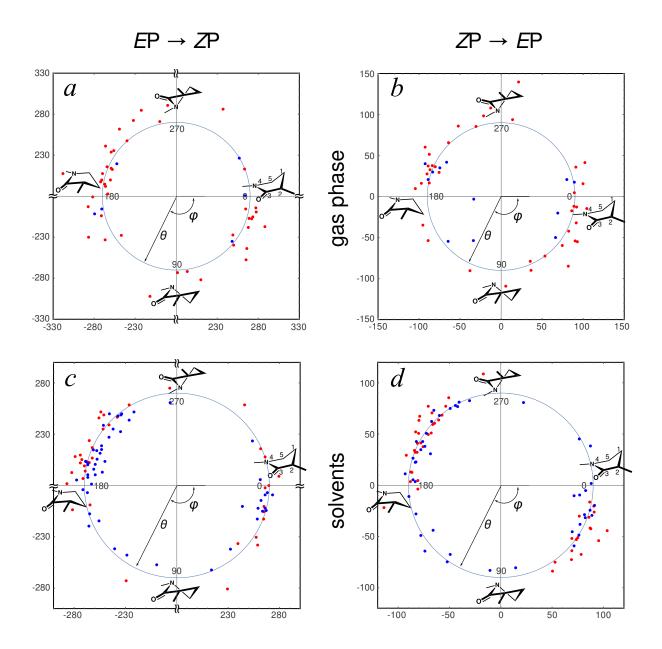


Supplementary Figure 17: Distributions of the central dihedral angle θ and angular velocity $d\theta/dt$ at the time of the $S_1 \to S_0$ transition (surface hop) for the (a) $EP \to ZP$ and (b) $ZP \to EP$ photoreactions occurring in MTDP (1) in MeOH, and (c) $EP \to ZP$ and (d) $EP \to EP$ photoreactions occurring in MMTDP (2) in MeOH. The red dots show the productive trajectories, and the blue dots show the unproductive trajectories. The graphs were plotted using the data from ref. $EP \to EP$ and $EP \to EP$ photoreactions occurring in MMTDP (2) in MeOH. The red dots show the





Supplementary Figure 19: Conformations of the pyrrolidinone ring in the motor 3 at the time of the $S_1 \to S_0$ surface hop. The conformations are characterized by two parameters: the central dihedral angle θ and the Cremer-Pople puckering angle 50 ϕ . The numbering of the atoms in the ring, used when calculating the puckering angle, is given in the insets. The circles of fixed radius show the threshold values of the angle θ used in Figure 16; θ = 270° for EP \to ZP and θ = 90° for ZP \to EP. The red dots show the (θ,ϕ) values for the productive trajectories and the blue dots for the unproductive trajectories. Panels (a) and (b) show the distribution of the (θ,ϕ) values for the EP \to ZP and ZP \to EP reactions in the gas phase, respectively. Panels (c) and (d) show the same distributions for photoreactions in MeOH. Panels (e) and (f) show photoreactions in MeCN, and panels (g) and (h) in DCM.



Supplementary Figure 20: Conformations of the pyrrolidinone ring in the motor 3 at the time of the $S_1 \to S_0$ surface hop. The conformations are characterized by two parameters: the central dihedral angle θ and the Cremer-Pople puckering angle 50 ϕ . The numbering of the atoms in the ring, used when calculating the puckering angle, is given in the insets. The circles of fixed radius show the threshold values of the angle θ used in Figure 16; θ = 270° for EP \to ZP and θ = 90° for ZP \to EP. The red dots show the (θ, ϕ) values for the productive trajectories and the blue dots for the unproductive trajectories. Panels (a) and (b) show the distribution of the (θ, ϕ) values for the EP \to ZP and ZP \to EP reactions in the gas phase, respectively. Panels (c) and (d) show the same distributions for photoreactions in all solvents considered in the QM/MM simulations; i.e., MeOH, MeCN, and DCM.

Supplementary Table 2 shows the analysis of the $S_1 \to S_0$ hopping times, where the average times for F-MMTDP in various media are shown together with the standard deviation (std) as mean \pm std. The hopping times $t_{2\text{hop}}$ are divided into two contributions: one is the time $t_{2\text{NACR}}$ required to reach the region of strong nonadiabatic coupling (NACR) and the other is the time t_{inNACR} spent within the region of strong nonadiabatic coupling. As seen in Supplementary Fig. 16, nearly all surface hops occur in the region within $\pm 20^{\circ}$ of the dividing line between the reactants and products. Therefore, the region of strong nonadiabatic coupling was defined by the angle θ within the range of values starting from 250° for $EP \to ZP$ and 90° for $ZP \to EP$.

In the gas phase, the largest contribution to $t_{2\text{hop}}$ is the time required to reach the region of strong coupling. The time spent within this region is approximately 40–45% of this value. The situation remains mainly the same in the MeON and MeCN solvents, where the longest contribution to $t_{2\text{hop}}$ is from $t_{2\text{NACR}}$. In DCM, this changes and it is t_{inNACR} that makes the greatest contribution to $t_{2\text{hop}}$, while $t_{2\text{NACR}}$ remains basically the same as in other solvents and gas phase.

MTDP in the gas phase takes a somewhat longer time to reach the region of strong coupling during the $EP \to ZP$ reaction. This is explained by the effect of the $O \cdot \cdot \cdot F$ repulsion in F-MMTDP, which accelerates the torsion in the early stage of the photoreaction. Very little difference is observed for the $ZP \to EP$ step. In MeOH, MTDP and F-MMTDP show very similar characteristics of the hopping times. The greatest difference between F-MMTDP and its parent molecules is with MMTDP, where, during the $EP \to ZP$ reaction, torsion towards the strong coupling region twice the time. This is likely caused by the effect of the intramolecular attractive interactions, which retard the movement towards the conical intersection seam.

For some selected QM/MM NAMD trajectories, the geometries of the minimumenergy conical intersections (MECI) and S_1 minima have been optimised starting from the snapshots at the hopping time. The resulting energies are reported in Supplementary Supplementary Table 2: Analysis of times to $S_1 \to S_0$ surface hop in F-MMTDP, MMTDP, and MTDP in various media. The following characteristics are reported: $t_{2\text{hop}}$ - a time spent in the S_1 state until the surface hop; $t_{2\text{NACR}}$ - a time required to reach the region of strong nonadiabatic coupling (when the torsion angle θ reaches 70° from the starting structure; that is, $\theta = 250^\circ$ for $EP \to ZP$ and $\theta = 70^\circ$ for $ZP \to EP$); $t_{in\text{NACR}}$ - a time spent within the region of strong nonadiabatic coupling before hop. The numeric values are given as mean \pm std, where std is the standard deviation.

medium	reaction	$t_{2\text{hop}}$, fs	t_{2NACR} , fs	$t_{in NACR}$, fs	
F-MMTDP					
gas phase		283 ± 82 365 ± 84	205 ± 63 250 ± 70	78 ± 50 115 ± 74	
МеОН		372 ± 158 410 ± 127	261 ± 115 269 ± 88	111 ± 120 141 ± 92	
MeCN		374 ± 127 457 ± 130	275 ± 118 291 ± 92	99 ± 70 166 ± 88	
DCM	$\begin{array}{c} EP \rightarrow ZP \\ ZP \rightarrow EP \end{array}$	568 ± 219 709 ± 254	264 ± 118 295 ± 132	$304 \pm 208 \ 414 \pm 267$	
MMTDP					
МеОН	$\begin{array}{c} EP \rightarrow ZP \\ ZP \rightarrow EP \end{array}$	652 ± 419 361 ± 195	588 ± 385 272 ± 172	64 ± 61 89 ± 55	
MTDP					
gas phase		370 ± 64 281 ± 45	324 ± 52 223 ± 30	46 ± 50 58 ± 38	
МеОН	$EP \to ZP$ $ZP \to EP$	390 ± 189 302 ± 113	292 ± 210 226 ± 96	98 ± 111 76 ± 50	

Table 3, where they are compared with the gas phase geometry optimisations.

As seen in the table, the MECI geometries in the gas phase and MeOH and MeCN

Supplementary Table 3: Energies of the MECI structures and S_1 minima obtained for F-MMTDP in the gas phase and various solvents. For the latter, the geometries of MECI and S_1 minima were optimized for randomly selected trajectories. The energies are given in the atomic units in the following order: S_0 energy / S_1 energy.

medium	geometry $EP \rightarrow ZP$		$Z\mathrm{P} o E\mathrm{P}$				
F-MMTDP							
gas phase	MECI	-1284.815241 / -1284.815123	-1284.807118 / -1284.806990				
	S _{1,min}	-1284.829370 / -1284.816355	-1284.827968 / -1284.809623				
МеОН	MECI	-1285.511046 / -1285.510933	-1285.497033 / -1285.496982				
	S _{1,min}	-1285.519370 / -1285.511736	a)				
MeCN	MECI	-1286.452198 / -1286.452083	-1286.588414 / -1286.588307				
	S _{1,min}	-1286.471122 / -1286.454118	-1286.594934 / -1286.588864				
DCM	MECI	-1255.164351 / -1255.164207	-1236.370859 / -1236.370771				
	S _{1,min}	-1255.183322 / -1255.168121	-1236.384683 / -1236.375000				
MMTDP							
MeOH	MECI	-1186.381437 / -1186.381367	-1186.359001 / -1186.358955				
	S _{1,min}	a)	a)				
MTDP							
gas phase	MECI	-1146.332571 / -1146.332557	-1146.332365 / -1146.332350				
	S _{1,min}	a)	a)				
МеОН	MECI	-1147.032463 / -1147.032392	-1147.051732 / -1147.051668				
	S _{1,min}	a)	a)				

^{a)} Oscillatory convergence; MECI occurs at the bottom of the S₁ minimum.

solvents correspond to peaked topography. In the gas phase, the MECI energies are elevated above the respective S_1 minima by 0.8 - 1.7 kcal/mol. In MeOH, the elevation is

 \lesssim 0.5 kcal/mol. In the $ZP \to EP$ step, MECI occurs at the bottom of the S_1 minimum. In MeCN, the MECI elevation is 0.3 – 1.3 kcal/mol. These values correspond to peaked or nearly so topographies of the respective MECIs. In DCM, the MECI geometries are elevated by more than 3 kcal/mol. This suggests that these MECIs have a sloped topography. In both MTDP and MMTDP, MECIs have peaked topography, both in the gas phase and in MeOH.

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