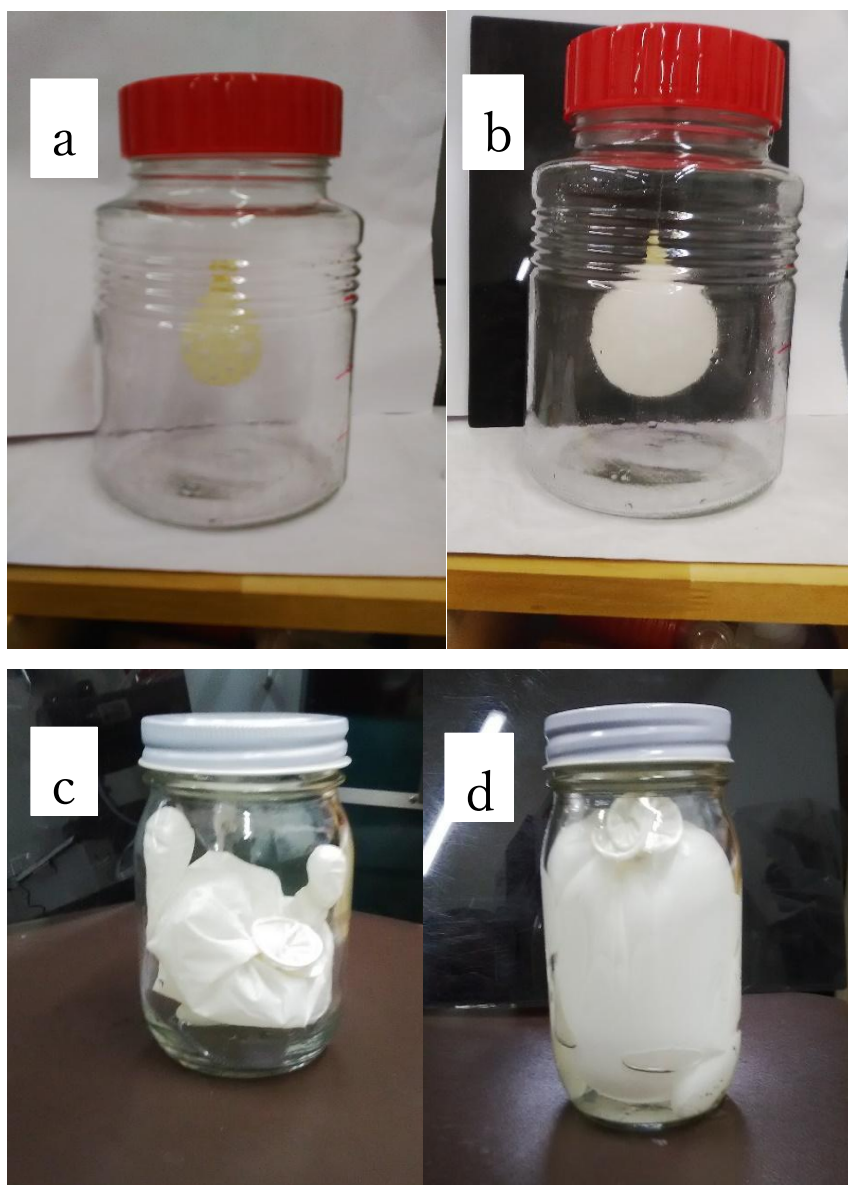


Supplementary Fig. 1, Table 1 and Fig. 2

### Supplementary Figure 1.

“Pseudo-Osmosis” in the gas phase.



(a) The initial state of latex balloon, (b) the ballon treated with 80% CO<sub>2</sub> in 4 L glass bottle for 4h, (c) the initial state of a latex glove, and (d) the glove treated with 100% CO<sub>2</sub> in 450 mL glass bottle for 4 h.

## Supplementary Table I.

Solubility of gas into the water.

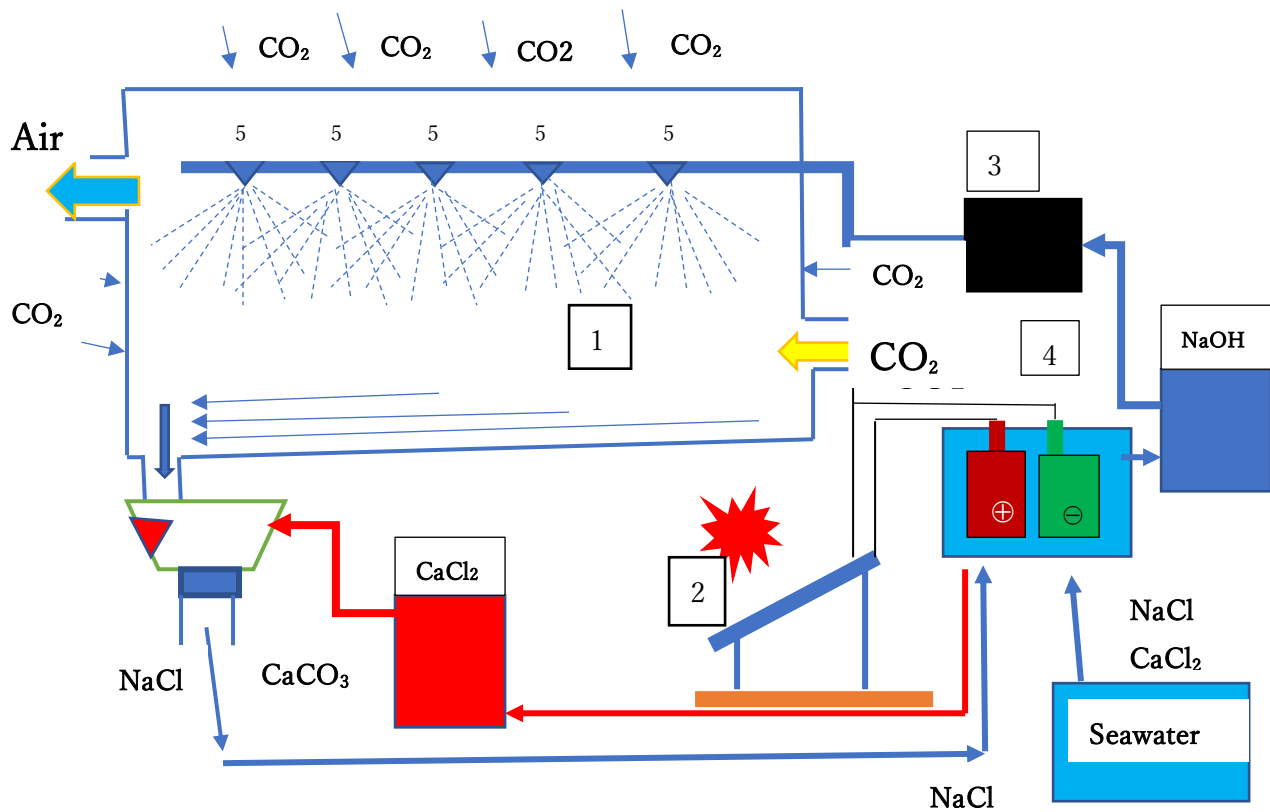
Gas	0°C	20°C	40°C	60°C	80°C	100°C
CO <sub>2</sub>	1.71	0.88	0.53	0.36		
	1.71	0.869				
O <sub>2</sub>	0.049	0.031	0.023	0.019	0.018	
	0.049	0.0312				
N <sub>2</sub>	0.024	0.016	0.012	0.01	0.0096	0.0095
	0.0238	0.0159				
Air	0.029	0.019	0.014	0.012	0.011	0.011
H <sub>2</sub>	0.0219	0.0181				
CH <sub>4</sub>		0.0347				

The value means the volume of the gas which absorbs into 1 mL water at 1 atm.

The black value is a citation from the Chronological Scientific Table, Maruzen Publication Company. The red value is a citation from the Handbook of Chemistry, 3.5 edition.

Supplementary Figure 2

An “Artificial Forest” Model.



1):  $\text{CO}_2$  fixing space, 2): solar panel, 3) pump, 4): battery.

Using a polytunnel made of polymer sheets, which allows  $\text{CO}_2$  penetration, instead of hard chamber, steel, or plastic could be cost effective.

