

Supplementary Information

Fluorine-free binder-based dry thick electrodes with Parafilm[®] M for sustainable and efficient battery manufacturing

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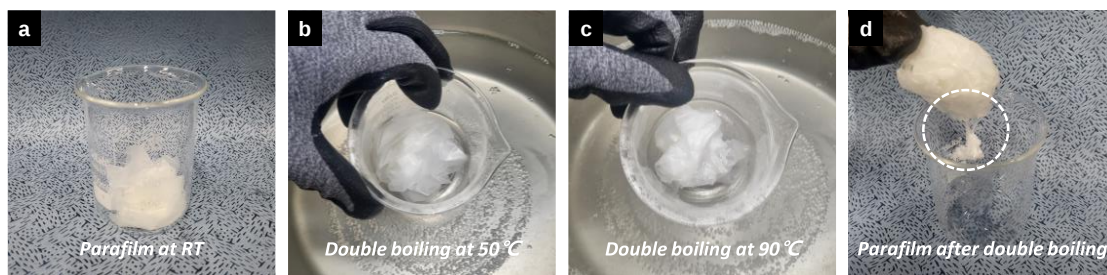
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Keywords: fluorine-free binder, paraffin, dry electrode, thick electrode, lithium-ion batteries

1 **Supplementary Table 1. Assignment of FT-IR characteristic peaks**

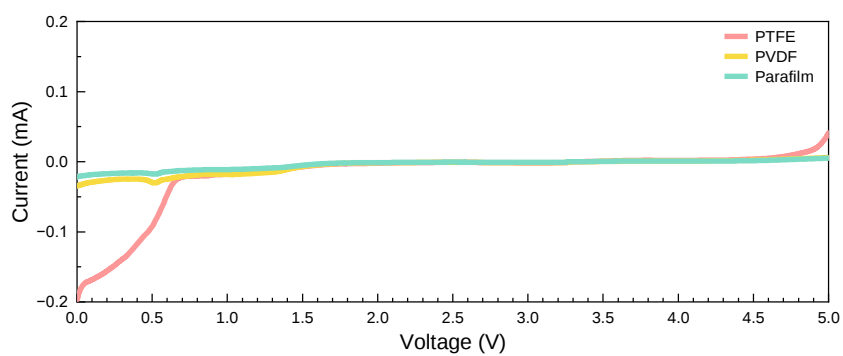
Bonding	Wavenumber (cm ⁻¹)
CF ₂	612
CH ₂ rocking	719
C-F	1175
CF ₃	1213
C-H	1396
-CH=CH- stretching	1462
	1472
-CH ₂ stretching	2847
	2915
-CH ₃ stretching	2952

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2 **Supplementary Fig. 1. Physical changes of Parafilm at the temperature around 90 °C.**



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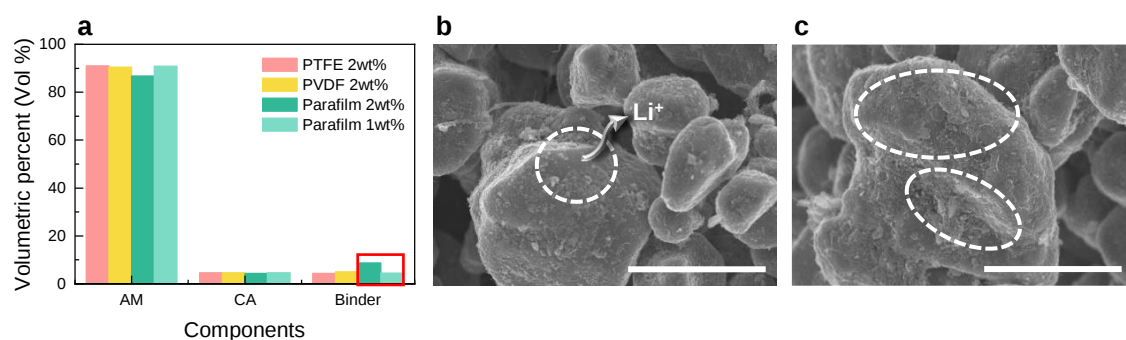
2 **Supplementary Fig. 2. Reduction and oxidation stability of binder in Li|binder film cells**

3 **tested by linear sweep voltammetry(LSV) at a scan rate of 1mV s^{-1} .**

1 **Supplementary Table 2. Volume and mass ratios according to binder.**

	Binder Type	Unit	Material			
			Active material	Conductive agent	Binder	Sum
PTFE	Material true density	g/cc	4.76	1.95	2.10	-
	Gravimetric composition	wt%	96	2	2	100.00
	Volumetric composition	vol%	91.07	4.63	4.30	100.00
PVDF	Material true density	g/cc	4.76	1.95	1.80	-
	Gravimetric composition	wt%	96	2	2	100.00
	Volumetric composition	vol%	90.42	4.60	4.98	100.00
Parafilm	Material true density	g/cc	4.76	1.95	0.98	-
	Gravimetric composition	wt%	96	2	2	100.00
	Volumetric composition	vol%	86.80	4.41	8.78	100.00
	Material true density	g/cc	4.76	1.95	0.98	-
	Gravimetric composition	wt%	97	2	1	100.00
	Volumetric composition	vol%	90.88	4.57	4.55	100.00

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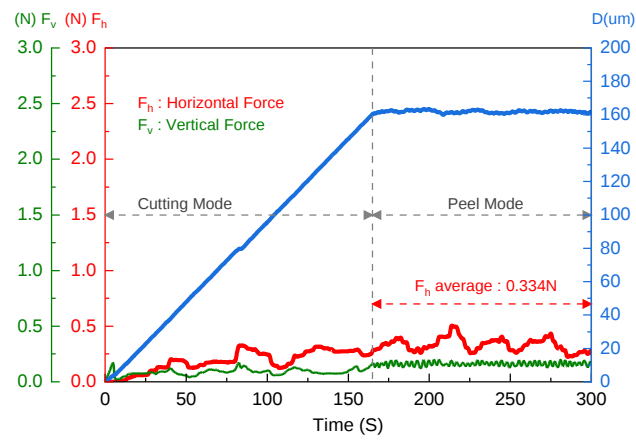


Supplementary Fig. 3. Parafilm mixed powder at various volume ratios. **a** Variation in volume ratio with different mass fractions of various binders. SEM images of active material particle surfaces with **b** 1 wt% and **c** 2 wt% Parafilm binder addition. The scale bar represents 5 μm .

1 **Supplementary Table 3. Comparison of adhesion strength using peel-off test of the**
2 **electrode between this work and other references.**

Electrode			Loading (mg cm ⁻²)	Adhesion strength (N cm ⁻¹)
This study	Dry coating Electrode	Parafilm	50	5.9
		PTFE	50	-
	Wet coating Electrode	PVDF	50	-
Reference	Dry coating Electrode(PTFE) (NCM622/CNT/binder = 92/5/3)		30	4.6
	Wet coating Electrode (PVDF) (NCM811/carbon black additive/binder = 92/3/5)		40	2.3
			60	0

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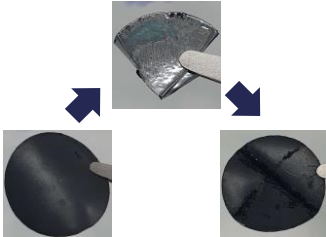
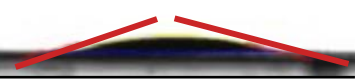
Supplementary Fig. 4. Representative force/depth profiles measured by SAICAS.

Analysis of adhesion properties of Parafilm-based electrode.

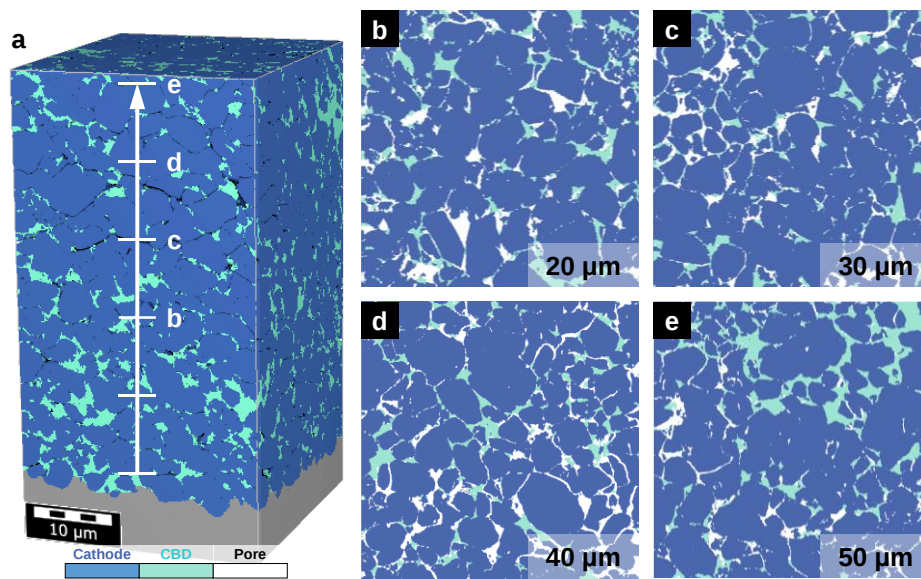
1 **Supplementary Table 4. Comparison of adhesion strength using SAICAS of the electrode**
2 **between this work and other references.**

Electrode		Loading (mAh cm ⁻²)	Adhesion strength (N)
This work	Dry coating Electrode(Parafilm)	8	0.334
Reference	Dry coating Electrode (PTFE) (SC-NCA/LPSCI/CNF/PTFE = 85/13/1/1)	3	0.6
	Wet double coating Electrode (PVDF) (NCM622/carbon conductor/PVDF = 93/3/4)	3.5	0.04~0.05

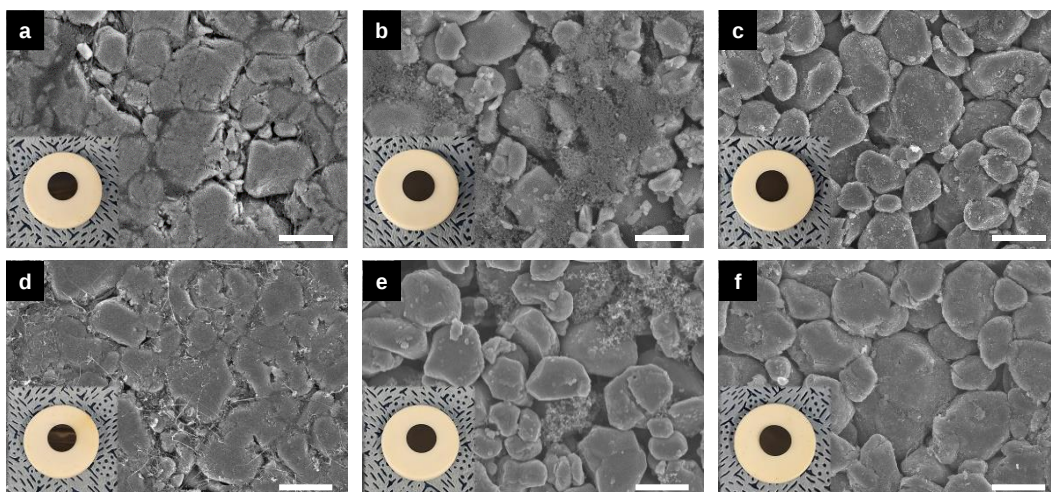
1 **Supplementary Table 5. Comparison of physical properties according to coating method.**

Ø14 Electrode (40 mg cm ⁻² , 3.3~3.4 g/cc)	Wet Coating (PVDF binder)	Dry Coating (Parafilm binder)
Folding Test		
Wettability Test	20.51 ° 	17.76 ° 

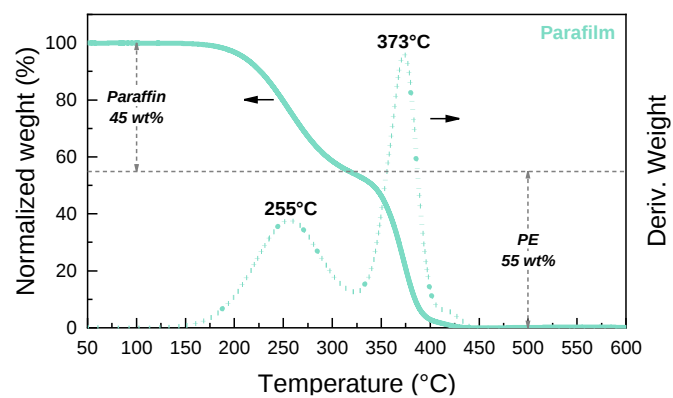
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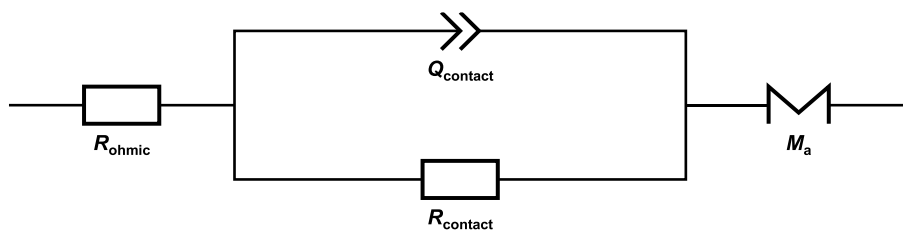
Supplementary Fig. 5. 3D FIB-SEM reconstructions of the Parafilm-based electrode. a Reconstructed 3D volume of the electrode. Sliced 2D images at **b** 20, **c** 30, **d** 40, and **e** 50 μm through thickness direction.



Supplementary Fig. 6. SEM image of the surface morphology of electrode. The inserted picture is a photograph of the electrode surface. On the surface of the electrode use each **a** PTFE, **b** PVDF, and **c** Parafilm binders before scorching test. **d** PTFE, **e** PVDF, and **f** Parafilm binders after scorching test. The scale bars represents 10 μm .



Supplementary Fig. 7. TGA curves of Parafilm binder.



R_{ohmic} : ohmic resistance

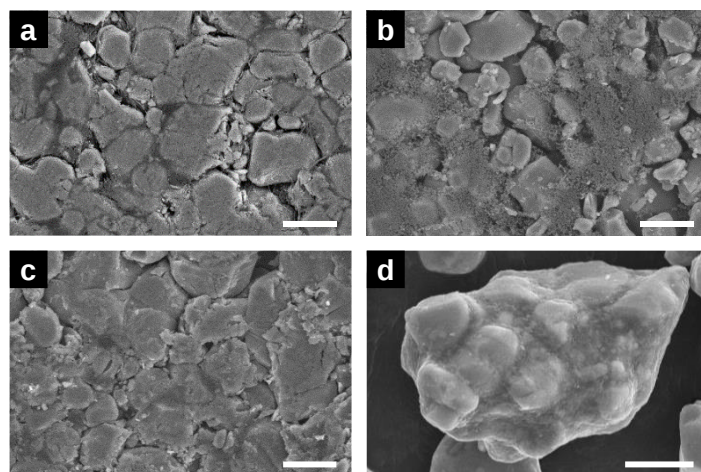
$R_{contact}$: contact resistance

$Q_{contact}$: contact constant phase element

M_a : modified restricted diffusion

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2 **Supplementary Fig. 8. Equivalent circuit model for fitting impedance spectra.**

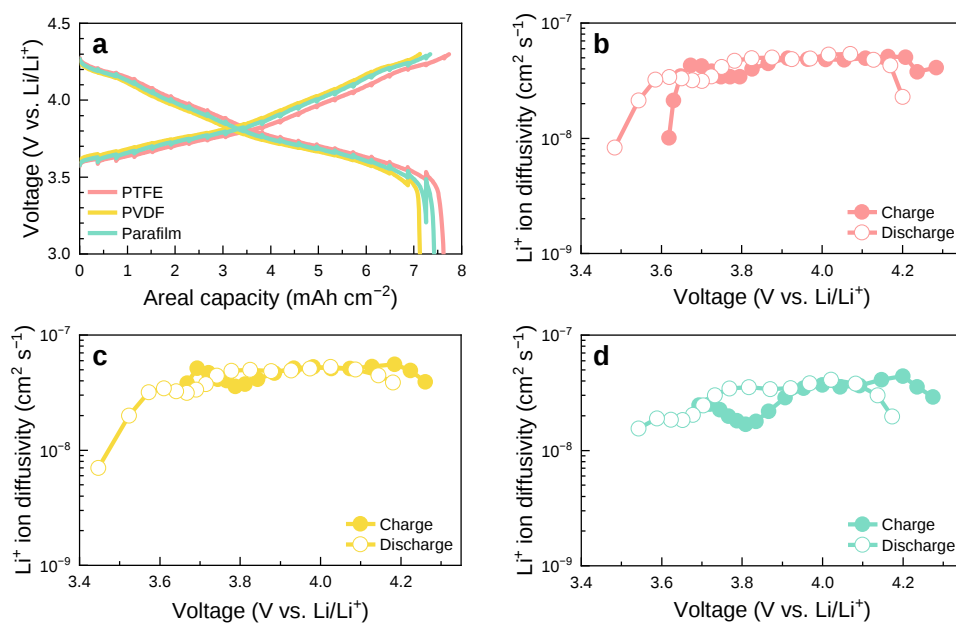


Supplementary Fig. 9. SEM image of the surface morphology of electrodes. On the surface of the electrode use each **a** PTFE, **b** PVDF, and **c** Parafilm binders before the formation cycle. The scale bars are 10 μm . **d** Zoom in on the powder mixed with the parafilm binder (Scale bar: 2 μm).

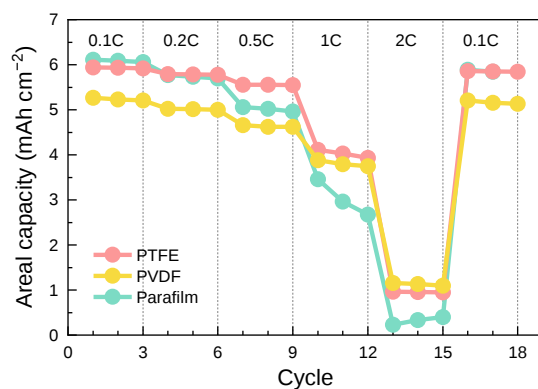
1 **Supplementary Table 6. The parameters for ionic tortuosity calculation.**

K_{bulk} (S/cm)			Surface (cm ²)		
0.0077			1.5394		
Binder	Thickness (cm)	Porosity	R_{ion} (Ω)	R_{ion} ($\Omega \cdot \text{cm}^2$)	Tortuosity
PTFE	0.0172	0.1956	53.62	85.54	7.23
PVDF	0.0243	0.3589	90.88	139.90	15.91
Parafilm	0.0167	0.1874	39.15	60.27	5.21

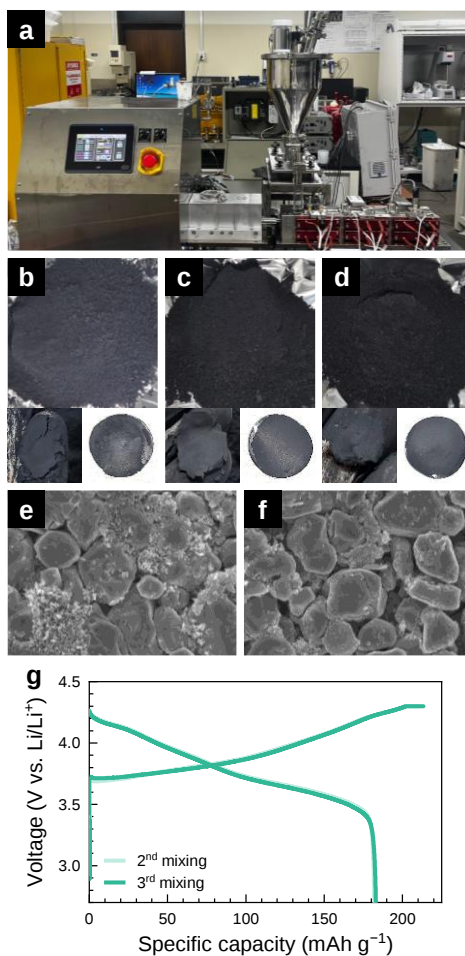
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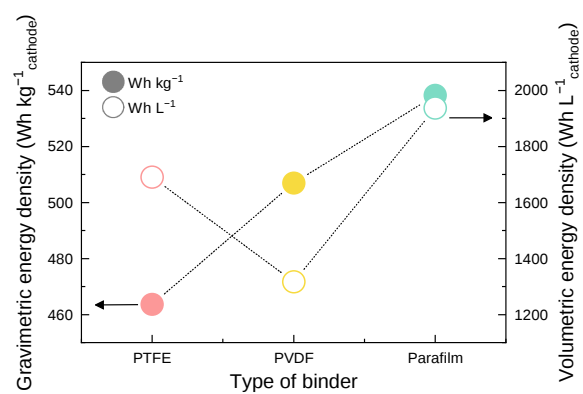
Supplementary Fig. 10. GITT profiles of each binder and corresponding diffusion coefficient. **a** GITT cycle data for voltage as a function of areal capacity. Calculated diffusion coefficient (D_{Li^+}) of the half cell using **b** PTFE, **c** PVDF, and **d** Parafilm from the GITT for charge/discharge.



Supplementary Fig. 11. Rate performance of half cell using different binders in the voltage range of 3.0-4.3V. Charge/discharge at rates of C/10, C/5, C/2, 1C, 2C, and the back to C/10.



Supplementary Fig. 12. Twin-screw-based continuous dry mixing process with Parafilm binder. Photographs of **a** continuous twin-screw mixer, and the dry mixed powder with the iteration of **b** first, **c** second, and **d** third. The left-bottom insets show the hand-clumped dry mixed powder, and the right-bottom ones show the fabricated dry electrodes. The microstructure with dry mixing iterations of **e** second and **f** third. **g** The initial electrochemical profiles of the second and third mixed dry electrode powder.



Supplementary Fig. 13. The gravimetric energy density and volumetric energy density of coin cells with different binders.

1 **Supplementary Table 7. Initial charge/discharge capacities and columbic efficiencies for**
2 **NCM811 with each binder half coin cells (40 mg cm⁻², 0.1C) and NCM811/graphite full**
3 **coin cells (40 mg cm⁻², 0.3C).**

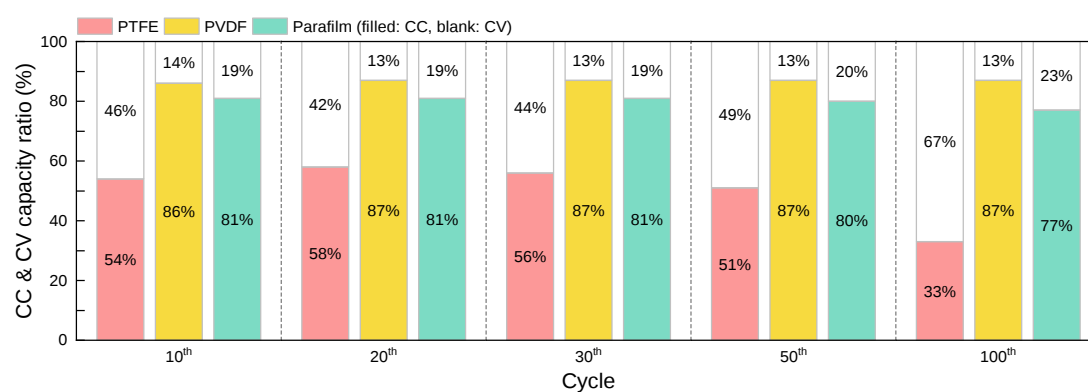
Cell type	Cell number	Charge capacity (mAh cm ⁻²)	Average charge capacity (mAh cm ⁻²)	Discharge capacity (mAh cm ⁻²)	Average discharge capacity (mAh cm ⁻²)	Columbic efficiency (%)	Average coulombic efficiency (%)
NCM811 with PTFE half-coin cells	1	8.634	8.444	7.859	7.706	90.902	91.186
	2	8.664		7.920		91.288	
	3	8.033		7.340		91.369	
NCM811 with PVDF half-coin cells	1	8.487	7.540	6.824	6.649	80.297	88.891
	2	7.619		6.670		87.423	
	3	6.515		6.453		98.952	
NCM811 with Parafilm half-coin cells	1	8.432	8.527	7.502	7.658	88.843	89.726
	2	8.552		7.733		90.421	
	3	8.596		7.739		89.915	
NCM811 with PTFE /graphite full-coin cells	1	7.251	8.058	5.872	6.063	80.979	75.523
	2	8.484		5.870		69.191	
	3	8.439		6.447		76.400	
NCM811 with PVDF /graphite full-coin cells	1	7.208	7.078	6.396	6.529	88.728	91.922
	2	7.572		6.670		88.087	
	3	6.453		6.522		98.952	
NCM811 with Parafilm /graphite full-coin cells	1	7.337	7.207	6.582	6.428	89.715	89.183
	2	6.924		6.121		88.409	
	3	7.536		7.443		89.424	

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- 1 **Supplementary Table 8. The mass loading, density, and capacity loading of NCM811**
- 2 **electrode using Parafilm binder and graphite electrode.**

Electrode Type	Mass ratio (%)	Mass loading (mg cm ⁻²)	Density (g cm ⁻³)	Capacity loading (mAh cm ⁻²)	N/P ratio
NCM811	AM/SP/CNT/Parafilm =97/1.5/0.5/1	37.8	3.8	7.33	1.05
graphite	AM/SP/SBR/CMC =95/2/1.5/1.5	22.4	1.04	7.66	

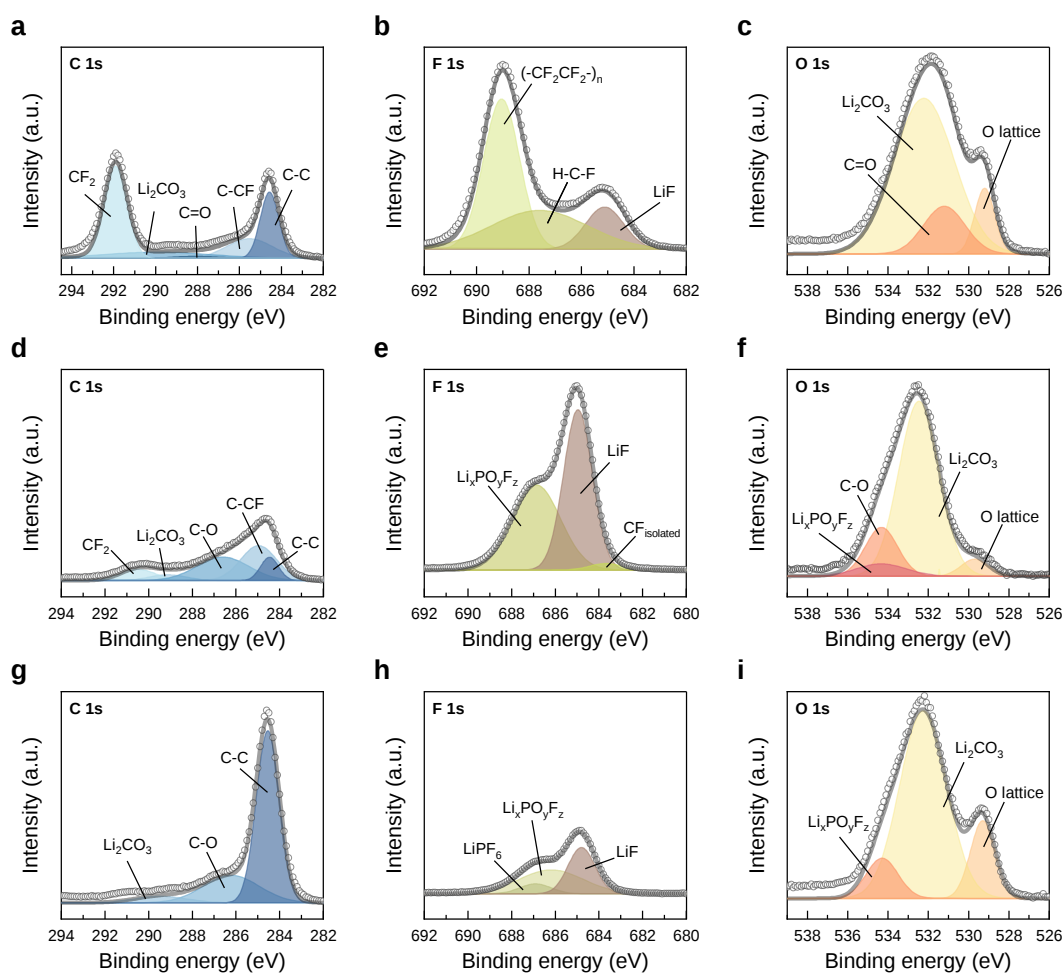
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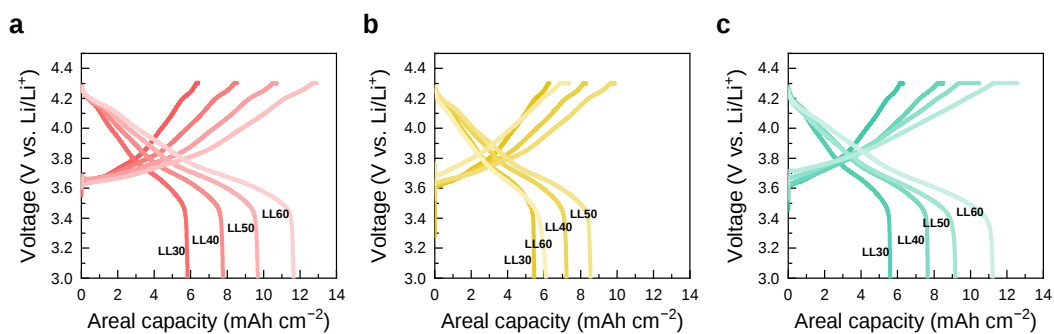
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2 **Supplementary Fig. 14. Comparison of CC and CV mode charge capacities with each**

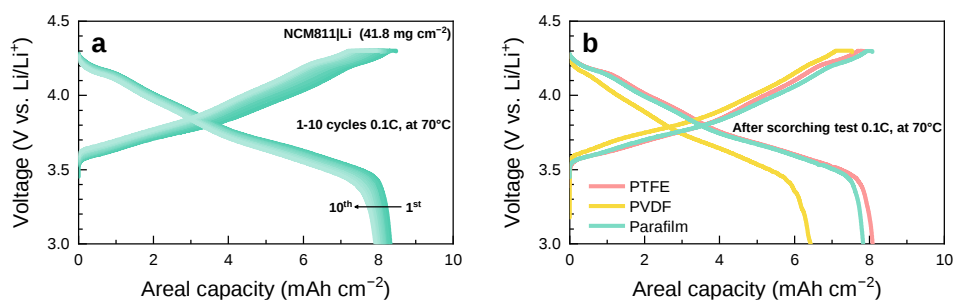
3 **binder after the 10th, 20th, 30th, 50th, and 100th cycles.**



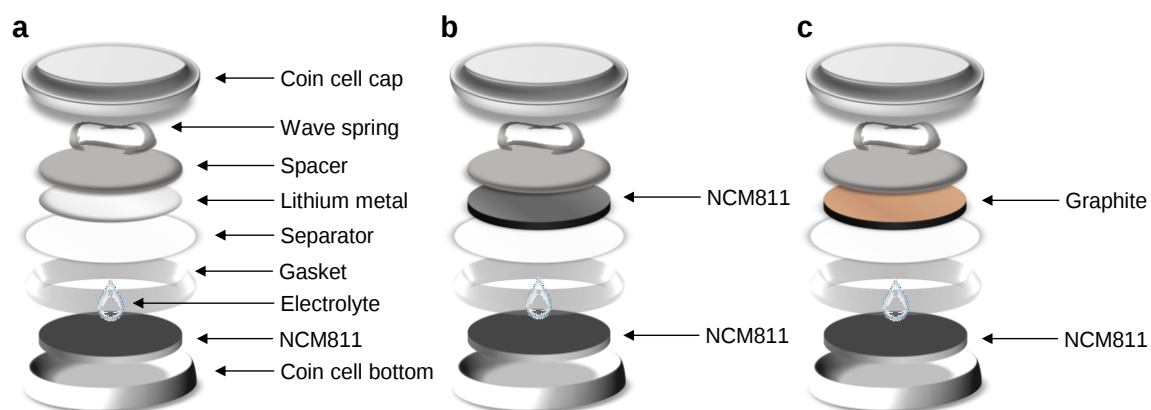
Supplementary Fig. 15. XPS analysis of the CEI layers on the formation cycled NCM811 cathode surface. C 1s, F 1s, and O 1s XPS spectra of the a-c PTFE, d-f PVDF, and g-i Parafilm.



Supplementary Fig. 16. Formation voltage profile. At areal mass loading 30, 40, 50, and 60mg cm⁻² for **a** PTFE, **b** PVDF, and **c** parafilm half cell in the voltage range of 3.0-4.3V at 0.1C.



Supplementary Fig. 17. Cycle data for thermal stability verification. a Voltage profile of Parafilm-based cell at 70°C above T_g for 10 cycles. **b** Cycle voltage profiles of the electrodes after scorching test.



1
2 **Supplementary Fig. 18. Coin cell architecture. a** NCM811|Li half coin cell **b**
3 NCM811|NCM811 Symmetric cell **c** NCM811|Graphite full coin cell.