

Supporting information

Preparation of Gold Nanoparticles Decorated UiO-66-NH₂ Incorporated Epichlorohydrin and Cyclodextrin as Novel Efficient Catalyst in Cross Coupling and Carbonylative Reactions

Leila Mohammadi ^{a. *} and Mohammad Reza Vaezi ^{a.*}

^aDepartment of Nano Technology and Advanced Materials, Materials and Energy Research
Center, Karaj, Iran. E-mail: l.mohammadi3790@gmail.com , m_r_vaezi@merc.ac.ir

Supporting Information

These materials are available in PDF format.

Author Information

- **Leila Mohammadi**

Department of Nano Technology and Advanced Materials, Materials and Energy
Research Center, Karaj, Iran

<https://orcid.org/0000-0001-5741-1503>

Email: l.mohammadi3790@gmail.com, l.mohammadi80@yahoo.com

- **Mohammad Reza Vaezi**

Department of Nano Technology and Advanced Materials, Materials and Energy
Research Center, Karaj, Iran

<https://orcid.org/0000-0002-6238-9678>

Email: m_r_vaezi@merc.ac.ir

Author Contributions

Leila Mohammadi was responsible for synthesizing the UiO-66-NH₂ @ Epichlorohydrin/Cyclodextrin/Au-NPs catalyst, conducting the C-C coupling and Carbonylative Sonogashira reactions, and preparing and assembling the manuscript. Mohammad Reza Vaezi provided valuable contributions to the manuscript's editing.

Conflicts of Interest

The researchers report no conflicts of interest.

Data Availability

All pertinent data for this study are presented within the main article and its supplementary materials.

Acknowledgements

The authors would like to thank the Materials and Energy Research Center (Grant No.: 9911940) and especially grateful of Iran's National Elites Foundation (INEF) for the financial support of this project.

Contents:

1. Table S1. The results of the optimization studies on the Heck reaction.
2. Table S2. The results of the optimization studies on the Sonogashira reaction.
3. Table S3. The results of the optimization studies on the Carbonylative Sonogashira reaction.

4. Table S4. Comparison of the catalytic performance of the proposed catalyst for Heck reaction with some related reports in the *literature*

5. Table S5. Comparison of the Catalytic Performance of the Synthesized Catalyst for the Sonogashira Reaction with Existing Literature Reports

8. Figure S1. Reusability of catalyst for a) Heck and b) Sonogashira reaction.

9. Spectrums

Suzuki derivatives S1-S43

Heck derivatives S44-S67

Sonogashira derivatives S68-S84

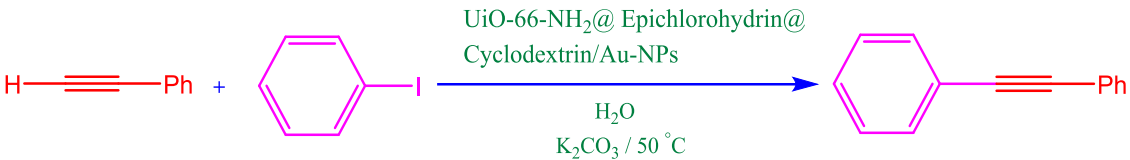
10. REFERENCES

Table S1. The results of the optimization studies on the Heck reaction.

Entry	Catalyst (mg)	Solvent	T (°C)	Base	Time (min)	Yield (%)
1	5	H ₂ O	25	K ₂ CO ₃	240	-
2	10	H ₂ O	25	K ₂ CO ₃	240	10
3	20	H ₂ O	30	K ₂ CO ₃	240	25
4	25	H ₂ O	45	K ₂ CO ₃	240	35
5	30	H ₂ O	55	K ₂ CO ₃	200	60
6	30	H ₂ O	80	K ₂ CO ₃	90	98
7	30	H ₂ O	90	K ₂ CO ₃	90	94
8	30	H ₂ O	100	K ₂ CO ₃	90	90
9	30	H ₂ O	80	K ₂ CO ₃	240	-
10	30	H ₂ O	90	KOH	240	76
11	30	DMSO	80	K ₂ CO ₃	100	90
12	30	DMAc	80	K ₂ CO ₃	100	90
13	30	NMP	80	K ₂ CO ₃	100	90
14	30	DMF	80	K ₂ CO ₃	100	87
15	30	H ₂ O	80	Et ₃ N	100	87
16	30	H ₂ O	80	Na ₂ CO ₃	100	88
17	30	H ₂ O	80	NaOH	100	88
18	30	H ₂ O	80	NaOAc	100	80
19	30	CH ₃ CN	80	K ₂ CO ₃	100	60
20	30	C ₆ H ₅ CH ₃	80	K ₂ CO ₃	100	65

Reaction condition: iodobenzene (1 mmol), styrene (1.1 mmol), base (2 mmol), solvent (3 ml)

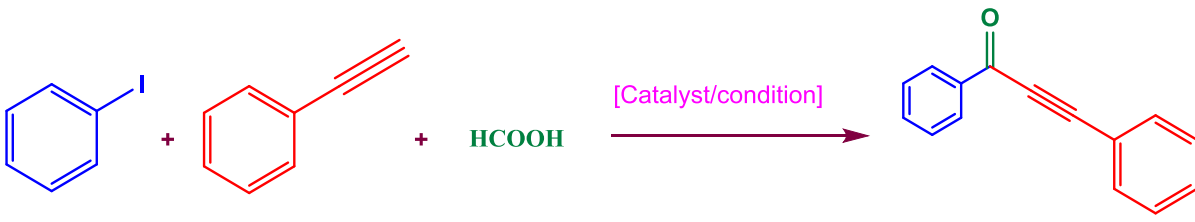
Table S2. The results of the optimization studies on the Sonogashira reaction.



Entry	Catalyst (mg)	Solvent	T (°C)	Base	Time (min)	Yield (%)
1	10	H ₂ O	50	K ₂ CO ₃	20	70
2	20	H ₂ O	50	K ₂ CO ₃	20	95
3	25	H ₂ O	50	K ₂ CO ₃	20	98
4	25	H ₂ O	25	K ₂ CO ₃	20	35
5	25	H ₂ O	100	K ₂ CO ₃	20	80
6	20	EtOH	50	K ₂ CO ₃	20	83
7	25	H ₂ O	50	NaCO ₃	20	15
8	25	H ₂ O	50	CsCO ₃	20	85
9	25	H ₂ O	50	NaOH	20	70
10	25	H ₂ O	50	KOH	20	25

Reaction condition: Aryl halide (1 mmol), terminal alkynes (1.2 mmol), base (2 mmol), solvent (3 mL)

Table S3. The results of the optimization studies on the Carbonylative Sonogashira reaction.



Entry	Base	Ligand	Solvent	Temperature (°C)	Yield (%)
1	Et ₃ N	DPPF	DMF	60	40
2	Et ₃ N	DPPF	toluene	60	42

3	Et ₃ N	DPPF	THF	60	37
4	K ₂ CO ₃	DPPE	toluene	60	41
5	Et ₃ N	DPPE	DMF	60	41
6	Et ₃ N	DPPP	toluene	60	42
7	Et ₃ N	BINAP	DMF	60	40
8	Et ₃ N	DPPM	DMF	60	40
9	Et ₃ N	-	toluene	60	20
10	Et ₃ N	Xantphos	toluene	60	43
11	Et ₃ N	Pcy ₃	DMF	60	38
12	Et ₃ N	Pcy ₃	toluene	60	40
13	Et ₃ N	PPh ₃	toluene	100	96
14	Et ₃ N	PPh ₃	toluene	90	96
15	Et ₃ N	PPh ₃	toluene	80	96
16	Et ₃ N	PPh ₃	toluene	70	96
17	Et ₃ N	PPh ₃	toluene	60	97
18	Et ₃ N	PPh ₃	toluene	55	96
19	Et ₃ N	PPh ₃	toluene	50	96
20	Et ₃ N	PPh ₃	toluene	45	92
21	DBU	PPh ₃	toluene	50	90
22	<i>i</i> Pr ₂ NEt	PPh ₃	toluene	50	89
23	K ₂ CO ₃	PPh ₃	toluene	50	92
24	Et ₃ N	PPh ₃	-	50	75
25	Et ₃ N	PPh ₃	CH ₃ CN	50	50
26	Et ₃ N	PPh ₃	THF	50	45
27	Et ₃ N	PPh ₃	DMF	50	40
28	Et ₃ N	-	toluene	50	20
29	-	PPh ₃	toluene	50	40

Reaction Condition: iodobenzene (0.2 mmol), phenyl acetylene (0.4 mmol), proposed catalyst UiO-66-NH₂@Epichlorohydrin@ Cyclodextrin @Au-NPs (30 mg), Ligand (10 mol%), base (3 mmol), HCOOH (2.0 mmol), Acetic anhydride (Ac₂O) (2 mmol), solvent (2.0 ml), 50°C, 8h.

DPPF = 1,1'-Bis (diphenylphosphino) ferrocene; DPPE = 1,2-Bis (diphenylphosphino) ethane; DPEphose = Bis (diphenylphosphino) butane; DPPE = 1,3-Bis (diphenylphosphino) propane; DPPPE = 1,5-Bis (diphenylphosphino) pentane; DPPM = Bis (diphenylphosphino) methane.

Table S4: Comparison of the catalytic performance of the proposed catalyst for Heck reaction with some related reports in the *literature*

Entry	Catalyst	Reaction condition	Time (h)	Yield %	Reference
1	GA-FSNP@Pd (0.47)	DMF: H ₂ O (2: 1), Cs ₂ CO ₃ , 110°C	0.5	95	[1]

2	Pd-biomagnetic (5)	DMF, TEA, 120°C	3	100	[2]
3	Pd/N-MCNPs	DMAN, TEA, 120°C	3	97.6	[3]
4	SBA-TMG ^a -Pd (0.05)	Solvent-free, 140°C	4	90	[4]
5	Fe ₃ O ₄ @PUNP ^b -Pd (0.1)	H ₂ O, K ₂ CO ₃ , reflux	12	trace	[5]
6	Pd/graphene oxide (0.54)	toluene, NEt ₃ , reflux	5	62	[6]
7	Pd/MCPPY	DMA, TBA, 120°C	3	97	[7]
8	Pd/HCN (0.255)	DMF, K ₂ CO ₃ , 120°C	1	100	[8]
9	Pd/Fe ₃ O ₄ @C (0.308)	DMF, K ₂ CO ₃ , 120°C	2	66	[9]
10	PdNs-PAMAM-g-MWCNTs (0.3)	NMP, K ₂ CO ₃ , 100°C	2.5	95	[10]
11	Pd-MNPSS (0.36)	K ₂ CO ₃ , H ₂ O, 100°C	4	90	[11]
12	PdCl ₂ (1.5), TDTAT (3 W%)	K ₂ CO ₃ , H ₂ O, 80°C	6	96	[12]
13	hydrogel@Pd NPs	K ₂ CO ₃ , H ₂ O, 85°C	0/5	98	[13]
14	<i>h</i> -BN@Sal@Pd(OAc) ₂ (0.05 mol)	H ₂ O, 90°C	3.5	92	[14]
15	PdL _n @β-CD (3)	K ₂ CO ₃ , H ₂ O, reflux	4	94	[15]
16	Pd-MPTA-1 (10 mg)	K ₂ CO ₃ , H ₂ O, 100°C	6	92	[16]
17	Pd@CSP (0.5), TBAB	NEt ₃ , H ₂ O, reflux	7	90	[17]
18	UiO-66-NH ₂ @Epichlorohydrin/CD/Au-NPs	K ₂ CO ₃ , H ₂ O, 80°C, 2h	1.5	98	This research

Table S5. Comparison of the Catalytic Performance of the Synthesized Catalyst for the Sonogashira Reaction with Existing Literature Reports



Entry	Catalyst	Reaction condition	Time (min)	Yield (%)	Ref.
1	Pd-B ₃ -Azo ₄ (0.08 mol%)	PL-XQ 350 W, Xe lamp with 420 nm UV cut-off, hv	3	94	[18]
2	(Pd ^{II} -NHCs) _n @nSiO ₂ (0.54 mol%)	Et ₃ N, H ₂ O, 90°C	0.15	92	[19]
3	Pd(II)-acylthiourea complex	KO ^t Bu, DMAc, 80°C	3	99	[20]
4	Pd@PTC-POP (0.3 mol%)	Et ₃ N, H ₂ O, 100°C	2	99	[21]

5	Pd-catalyst CuI/trimethylamine (5 mol%)	Microwave, H ₂ O, 100°C	2	60	[22]
6	Pd@CS/PEO (7.0 mol%)	DMF, glycol, K ₂ CO ₃ , 110°C	5	93	[23]
7	Pd@ONO-IL (0.15 mol%)	Piperidine, DMF, 70°C	2	94	[24]
8	Pd-NPs/CENFs (0.06 mol%)	Na ₃ PO ₄ ·12H ₂ O, <i>i</i> -PrOH, reflux	3	88	[25]
9	Pd-LHMS-3 (0.03 g)	DMF, Hexamine, N ₂ atm, 120°C	14	80	[26]
10	Pd(0)/MCoS-1 (0.02 g)	H ₂ O, Et ₃ N, 80°C	6	94	[27]
11	Ru/Al ₂ O ₃ (5 mol%)	CH ₃ CN, Et ₃ N, 90°C	24	85	[28]
12	UiO-66-NH ₂ @Epichlorohydrin/ Cyclodextrin/Au-NPs	K ₂ CO ₃ , H ₂ O, 50°C	20	99	This research

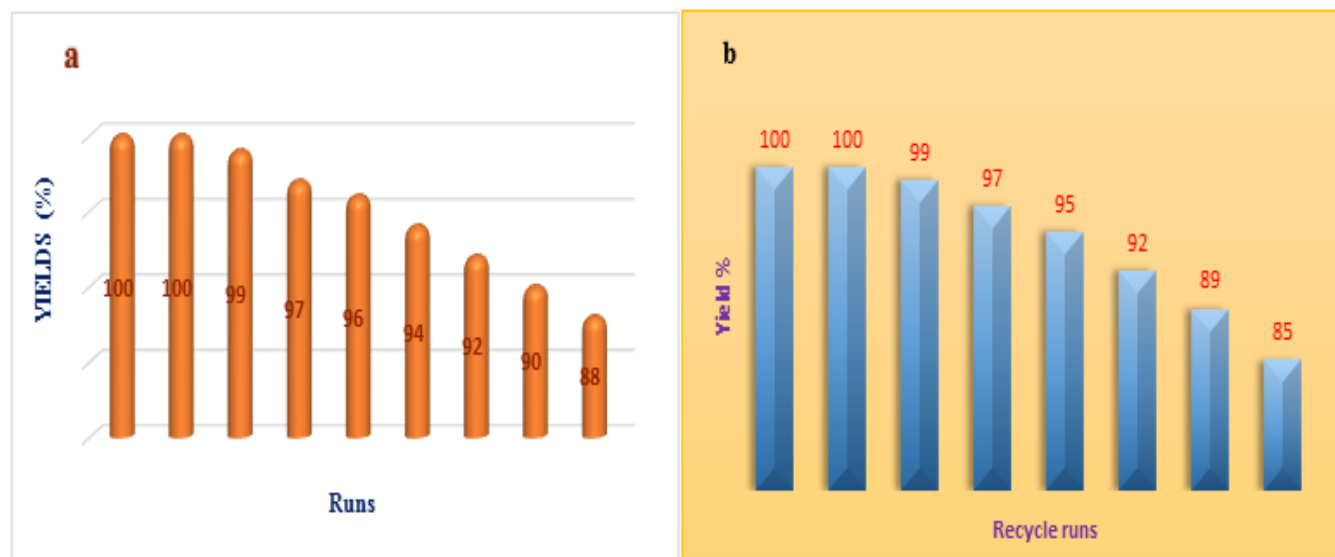
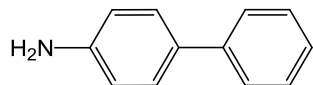


Figure S1. Reusability of catalyst for a) Heck and b) Sonogashira reaction.

Suzuki derivatives

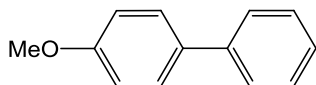
4-amine-[1,1'-biphenyl]



m.p. = 67-68 °C.

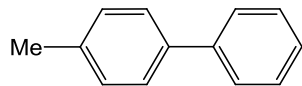
^1H NMR (400 MHz, DMSO): δ = 5.23 (2H, NH₂), 6.67 (2H, d, J = 8.3), 7.20 (1H, t, J = 7.2), 7.34-7.38 (4H, m), 7.53 (2H, d, J = 7.6). ^{13}C NMR (100 MHz, DMSO): δ = 114.3, 125.4, 125.7, 127.2 (4 $\times$ CH Ar), 127.5 (Ar), 128.8 (CH, Ar), 140.7, 148.4 (Ar). m/z: 170.1

4-Methoxy-1,1'-biphenyl



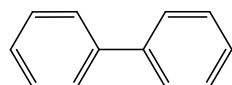
White waxy solid, yield 90%, ^1H NMR (500 MHz, CDCl₃): 7.57-7.52 (m, 4H), 7.41 (t, J = 8.0 Hz, 2H), 7.30 (t, J = 8.0 Hz, 1H), 6.98 (d, J = 8.5 Hz, 2H), and 3.86 (s, 3H).

4-methyl-1,1'-biphenyl



white yellow solid in 92% yield, MP: 46-48 °C; ^1H NMR (500 MHz, CDCl₃): δ 7.61 (s, 1H), 7.51 (t, J = 7.8 Hz, 1H), 7.39 (m, 3H), 7.34 (m, 1H), 7.24 (t, J = 7.2 Hz, 1H), 7.16 (t, J = 8.1 Hz, 2H), 2.31 (s, 3H); ^{13}C NMR (126 MHz, CDCl₃): δ 141.19, 139.80, 138.34, 137.05, 136.72, 129.48, 129.01, 128.74, 128.21, 127.38, 127.01, 126.83, 125.73, 21.11; FTIR (cm⁻¹): 2918, 2318, 1743, 1521, 1325, 1114, 802; GC-MS m/z: 168.

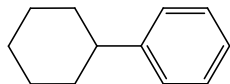
Biphenyl



white solid in 94% yield, MP: 69-71 °C

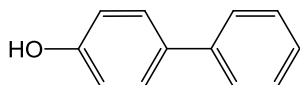
^1H NMR (500 MHz, CDCl_3): δ 7.59 (d, J = 7.7 Hz, 4H), 7.44 (t, J = 7.7 Hz, 4H), 7.34 (t, J = 7.4 Hz, 2H); ^{13}C NMR (126 MHz, CDCl_3): δ 141.29, 128.82, 127.27; FTIR (cm^{-1}): 2358, 1510, 1029, 580; GC-MS (EI, 70eV) m/z : found: 154 ($\text{C}_{12}\text{H}_{10}$).

Cyclohexylbenzene



Colorless oil in 73% yield ^1H NMR (400 MHz, chloroform- d) δ = 7.31 – 7.25 (m, 2H), 7.23 – 7.13 (m, 3H), 2.49 (m, 1H), 1.92 – 1.82 (m, 4H), 1.74 (m, 1H), 1.51 – 1.34 (m, 4H), 1.25 (m, 1H); ^{13}C NMR (100 MHz, chloroform- d) δ = 148.2, 128.4, 127.0, 125.9, 44.8, 34.6, 27.1, 26.3.

4-hydroxyl-[1,1'-biphenyl]

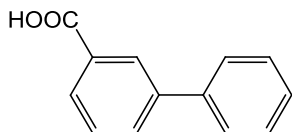


m.p. = 161-162 $^{\circ}\text{C}$.

^1H NMR (400 MHz, DMSO): δ = 6.85 (2H, d, J = 8.5), 7.27 (1H, t, J = 7.4), 7.40 (2H, t, J = 7.4), 7.48

(2H, d, J = 8.5), 7.57 (2H, d, J = 8.5), 9.58 (1H, br s, OH). ^{13}C NMR (100 MHz, DMSO): δ = 115.8, 126.0, 126.4, 127.8, 128.8 ($5 \times \text{CH}$, Ar), 131.0, 140.3, 157.2 (Ar). m/z : 169.1.

3-carboxylic acid-[1,1'-biphenyl]

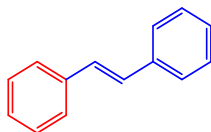


m.p. = 160-161 $^{\circ}\text{C}$. ^1H NMR (400 MHz, DMSO): δ = 7.42 (1H, tt, J = 7.3, 1.1), 7.50 (2H, t, J = 7.3), 7.58 (1H, t, J = 7.7), 7.66 (2H, m), 7.87 (1H, dq, J = 7.7, 1.4), 8.14 (1H, dt, J = 7.7, 1.4), 8.40 (1H, t, J = 1.4).

^{13}C NMR (100 MHz, DMSO): δ = 127.2, 127.8, 128.85, 128.93, 128.96, 128.99 (CH, Ar), 129.8 (Ar), 132.4 (CH Ar), 139.9, 141.6 ($2 \times \text{CH}$ Ar), 172.3 (C=O). m/z : 197.0

Heck derivatives

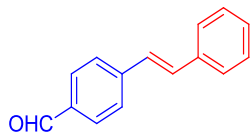
Trans-1, 2-diphenylethene (*E*-Stilbene)



White solid, Mp. 123-124°C (lit. 123.9-124.6°C)

GC-MS (m/z): 180; ¹HNMR (400 MHz, CDCl₃) δ 7.07-7.08 (s, 2H), 7.21-7.23 (m, 2H), 7.32- 7.34 (m, 4H,) 7.46-7.48 (m, 2H) ppm. ¹³CNMR (100 MHz, CDCl₃) δ 122.1, 123.2, 124.3, 132.9.

(*E*)-4-styrylbenzaldehyde

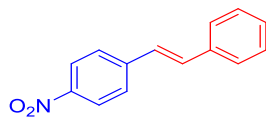


White solid, Mp. 116°C (lit. 116.0-116.8°C)

GC-MS (m/z): 208; ¹HNMR (400 MHz, CDCl₃): δ = 9.97 (s, 1H), 7.84 (d, J=8.0 Hz, 2H), 7.62 (d, J = 8.0 Hz, 2H), 7.52 (d, J= 7.6 Hz, 2H), 7.36 (t, J = 7.6 Hz, 2H), 7.28 (t, J= 7.6 Hz, 1H), 7.22 (d, J = 5.2 Hz, 1H), 7.11 (d, J=16.4 Hz, 1H).

¹³CNMR (100 MHz, CDCl₃): δ = 191.5, 143.4, 136.5, 135.3, 132.2, 130.2, 128.8, 128.4, 127.3, 126.9. MS (EI) m/z: 208, 179, 165, 152, 102, 89, 76, 63, 51.

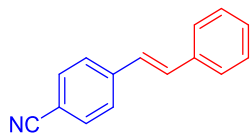
(*E*)-4-Nitrostilbene



Yellow solid, Mp. 154-155°C (lit. 154.2-155.0°C)

GC-MS (m/z): 225; ¹HNMR (400 MHz, CDCl₃) δ: 8.22 (d, J = 8.6 Hz, 2H, Ar-H), 7.63 (d, J = 8.7 Hz, 2H, Ar-H), 7.55 (d, J = 7.7 Hz, 2H, Ar-H), 7.40 (t, J = 7.6 Hz, 2H, Ar-H), 7.36-7.31 (m, 1H, Ar-H), 7.27 (d, J = 16.3 Hz, 1H, CH=), 7.14 (d, J = 16.3, 1H, Hz, CH=). ¹³CNMR (100 MHz, CDCl₃), δ = 125.2, 127.4, 128.0, 128.2, 130.0, 134.4, 137.3, 145.0, 147.8 ppm.

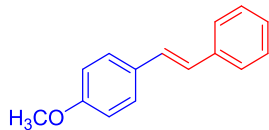
(E)-4-Styrylbenzonitrile



White solid, Mp. 114-115°C (lit. 114.9-115.2°C)

GC-MS (m/z): 205; ¹H-NMR (400 MHz, CDCl₃): δ 7.09 (d, 1H, *J* = 16.0 Hz), 7.22 (d, 1H, *J* = 16.0 Hz), 7.31-7.34 (t, 1H, *J* = 4.0 Hz), 7.38-7.41 (t, 2H, *J* = 4.0 Hz), 7.54 (d, 2H, *J* = 7.2 Hz), 7.58 (d, 2H, *J* = 8.4 Hz), 7.64 (d, 2H, *J* = 8.4 Hz) ppm; ¹³C-NMR (100 MHz, CDCl₃): δ 111.7, 120.2, 127.8, 128.0, 128.1, 129.8, 130.0, 133.5, 133.6, 137.4, 142.9 ppm.

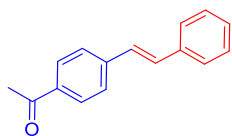
(E)-1-methoxy-4-styrylbenzene



White solid, Mp. 135-136°C (lit. 135.3-135.9°C)

GC-MS (m/z): 210; ¹H-NMR (400 MHz, CDCl₃) δ (ppm): 3.70 (3H, s), 6.78 (1H, d, *J* = 8.75 Hz), 6.90 (1H, d, *J* = 10.75 Hz), 7.00-7.38 (9H, m). ¹³C-NMR (100 MHz, CDCl₃) δ (ppm): 55.3, 114.2, 126.3, 126.6, 127.25, 127.8, 128.2, 128.7, 130.2, 137.7, 159.3.

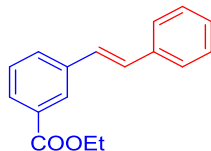
(E)-1-(4-styrylphenyl) ethanone



White solid, Mp. 141-142°C (lit. 141.3-142°C)

GC-MS (m/z): 222; ¹H-NMR (400 MHz, CDCl₃): δ 7.97-7.95 (d, 2H), 7.58-7.57 (d, 2H), 7.53-7.52 (d, 2H), 7.40-7.36 (t, 2H), 7.29-7.25 (t, 1H), 7.22-7.20 (d, 1H), 7.12 (d, 1H), 2.64 (s, 3H), ¹³C-NMR (100 MHz, CDCl₃): δ 197.6, 142.1, 136.7, 136.1, 131.5, 128.9, 128.8, 128.4, 127.6, 126.8, 126.5, 26.8.

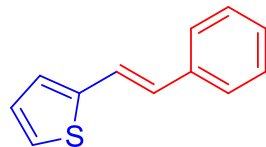
(E)-Ethyl-3-styrylbenzoate



Oily liquid

GC-MS (m/z): 252; ¹H NMR (CDCl₃, 400 MHz): δ (ppm) 8.22 (s, 1H), 7.95 (d, *J* = 7.7 Hz, 1H), 7.70 (d, *J* = 7.7 Hz, 1H), 7.55 (d, *J* = 7.4 Hz, 2H), 7.46–7.27 (m, 4H), 7.21 (d, *J* = 16.0 Hz, 1H), 7.15 (d, *J* = 16.0 Hz, 1H), 4.43 (q, *J* = 7.1 Hz, 3H), 1.44 (t, *J* = 7.1 Hz); ¹³C NMR (CDCl₃, 100 MHz): δ (ppm) 166.6, 137.7, 137.0, 131.0, 130.7, 129.9, 128.8, 128.7, 128.5, 128.0, 127.6, 127.5, 126.7.

(E)-2-styrylthiophene



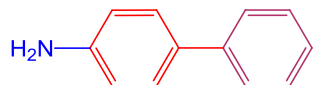
white solid, MP. 111-112°C (lit. 112°C)

GC-MS (m/z): 186; ¹H NMR (CDCl₃, 400 MHz): δ (ppm) 7.49 (d, *J* = 7.5 Hz, 2H), 7.37 (t, *J* = 7.5 Hz, 2H), 7.28 (d, *J* = 5.1 Hz, 1H), 7.24 (d, *J* = 16 Hz, 1H), 7.21 (d, *J* = 4.9 Hz, 1H), 7.09 (d, *J* = 3.4 Hz, 1H), 7.03 (t, *J* = 4 Hz, 1H), 6.95 (d, *J* = 16.2 Hz, 1H); ¹³C NMR (CDCl₃, 100 MHz): δ (ppm) 142.9, 137.0, 128.7, 128.4, 127.6, 126.3, 126.1, 124.3, 121.8.

Spectrums

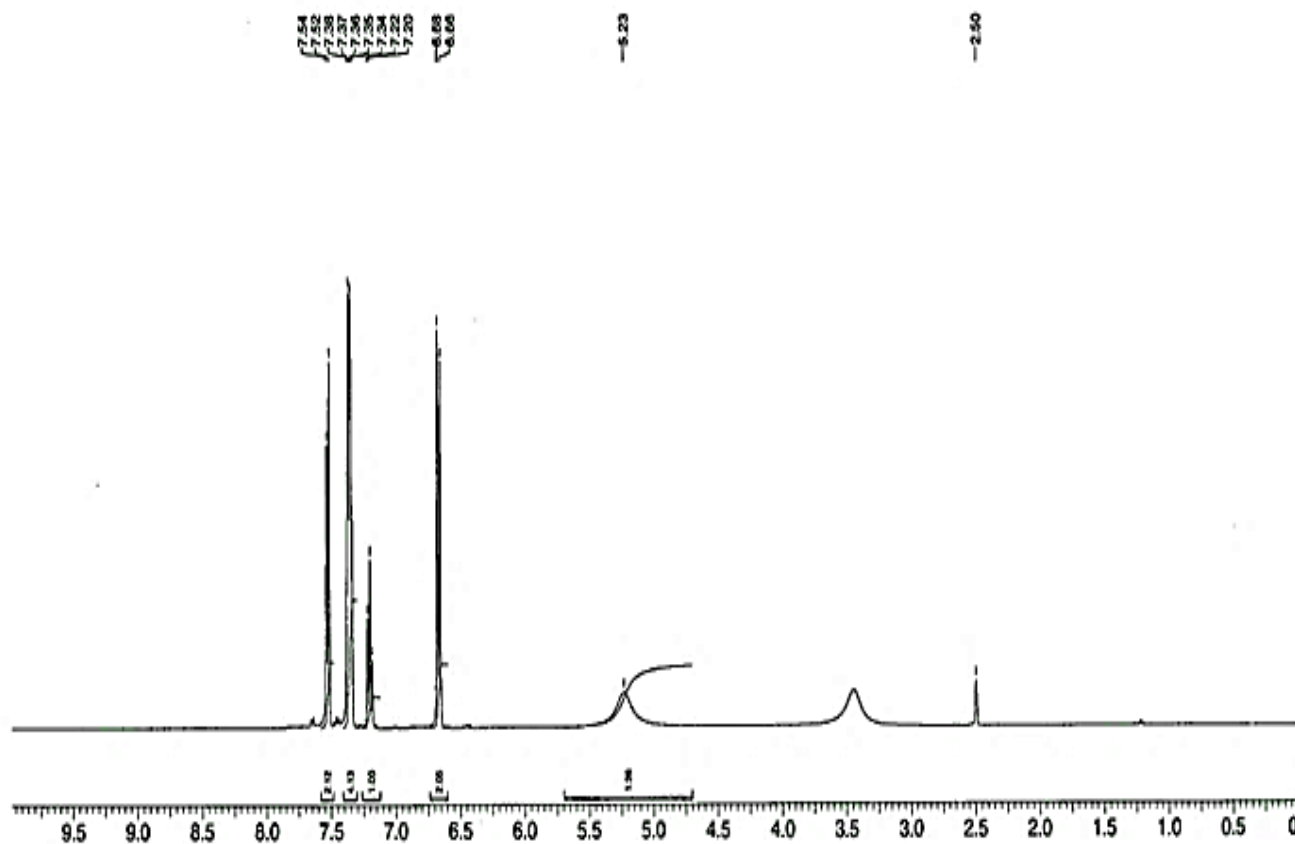
Suzuki derivatives

4-amine-[1,1'-biphenyl]

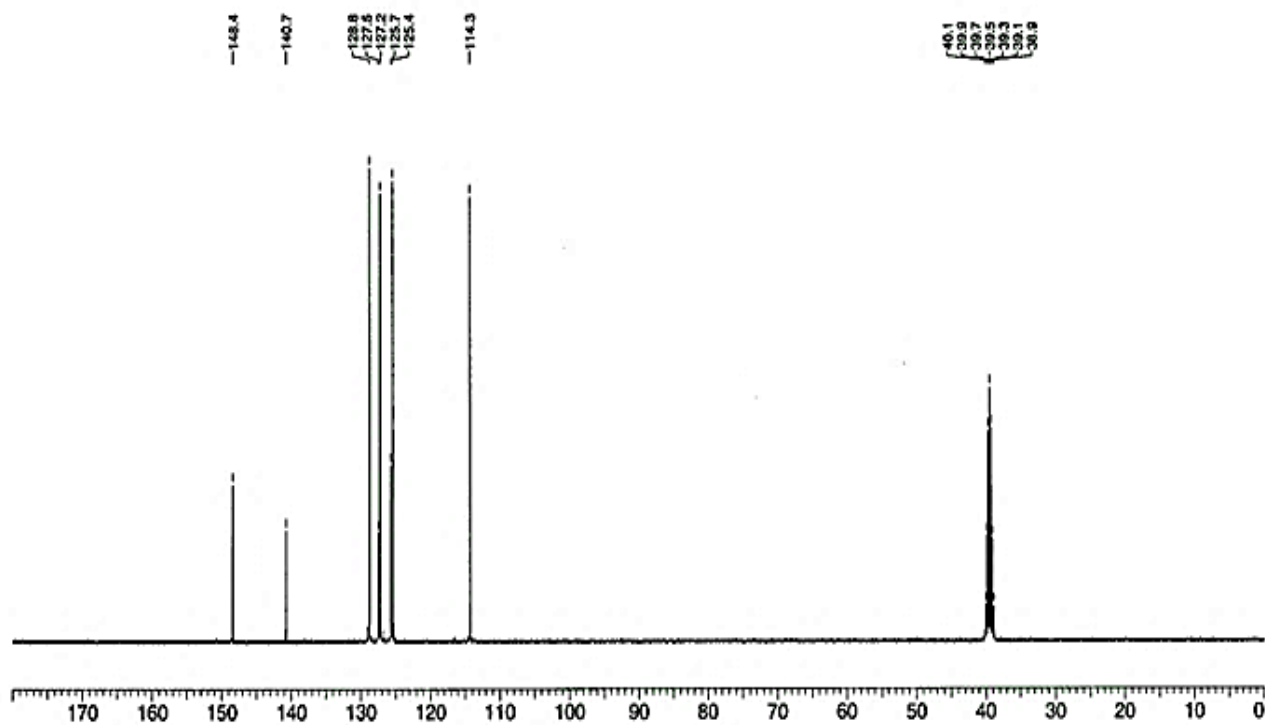


m.p. = 67-68 °C.

¹HNMR (400 MHz, DMSO): δ = 5.23 (2H, NH₂), 6.67 (2H, d, J = 8.3), 7.20 (1H, t, J = 7.2), 7.34-7.38 (4H, m), 7.53 (2H, d, J = 7.6). ¹³CNMR (100 MHz, DMSO): δ = 114.3, 125.4, 125.7, 127.2 (4 $\times$ CH Ar), 127.5 (Ar), 128.8 (CH, Ar), 140.7, 148.4 (Ar). m/z: 170.1

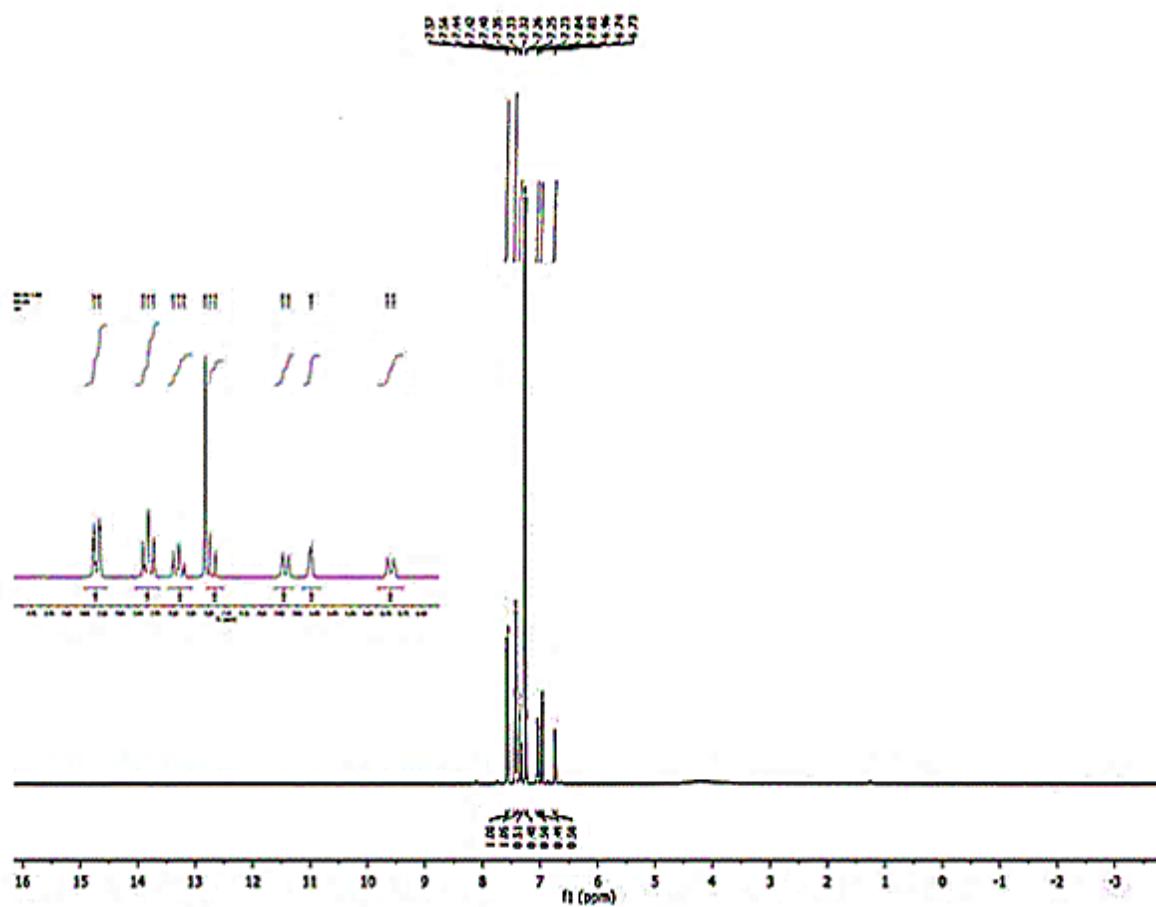
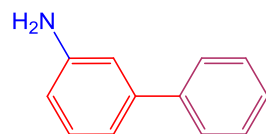


S1. The ¹HNMR of 4-amine-[1,1'-biphenyl]

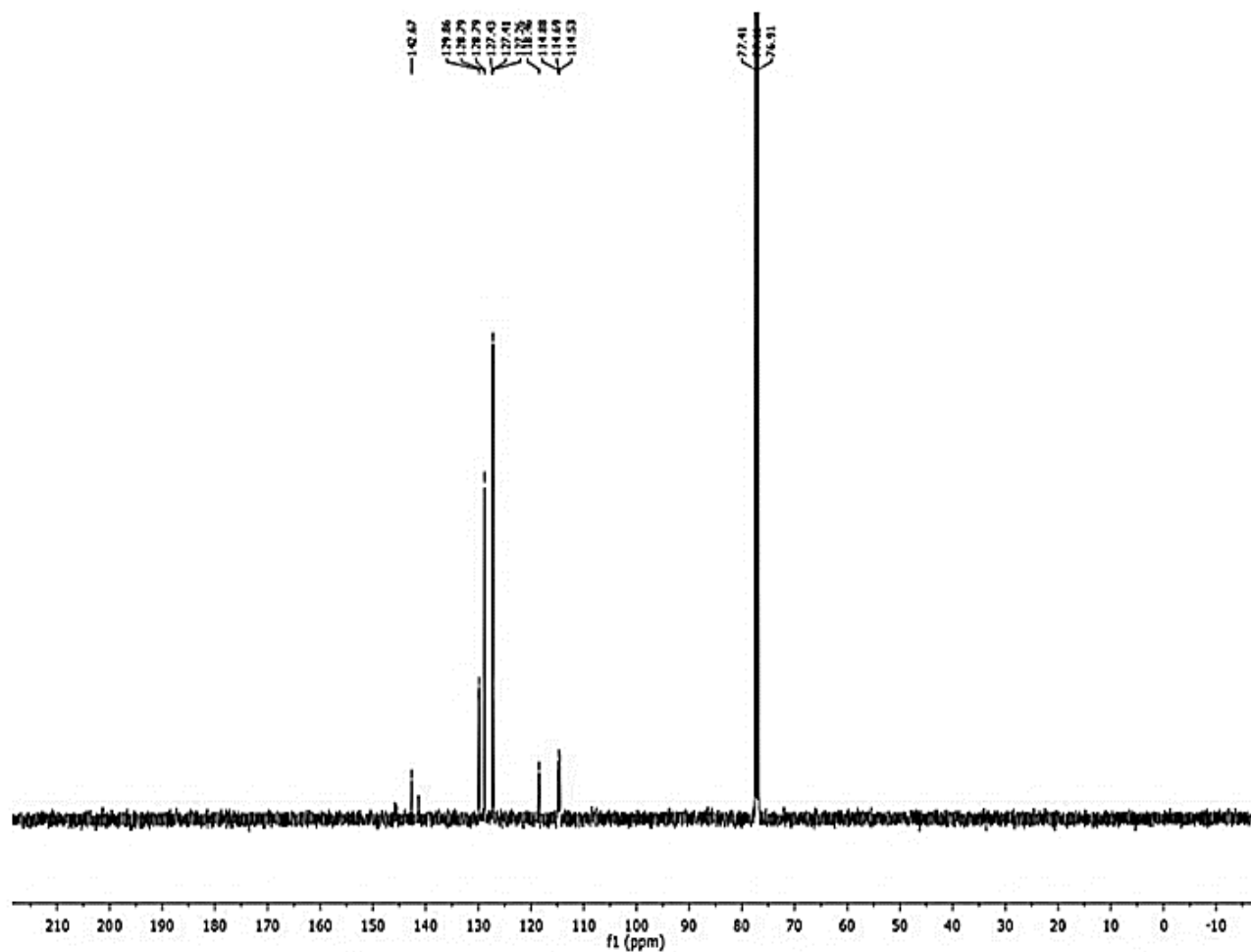


S2. The ^{13}C NMR of 4-amine-[1,1'-biphenyl]

3-amine-[1,1'-biphenyl]

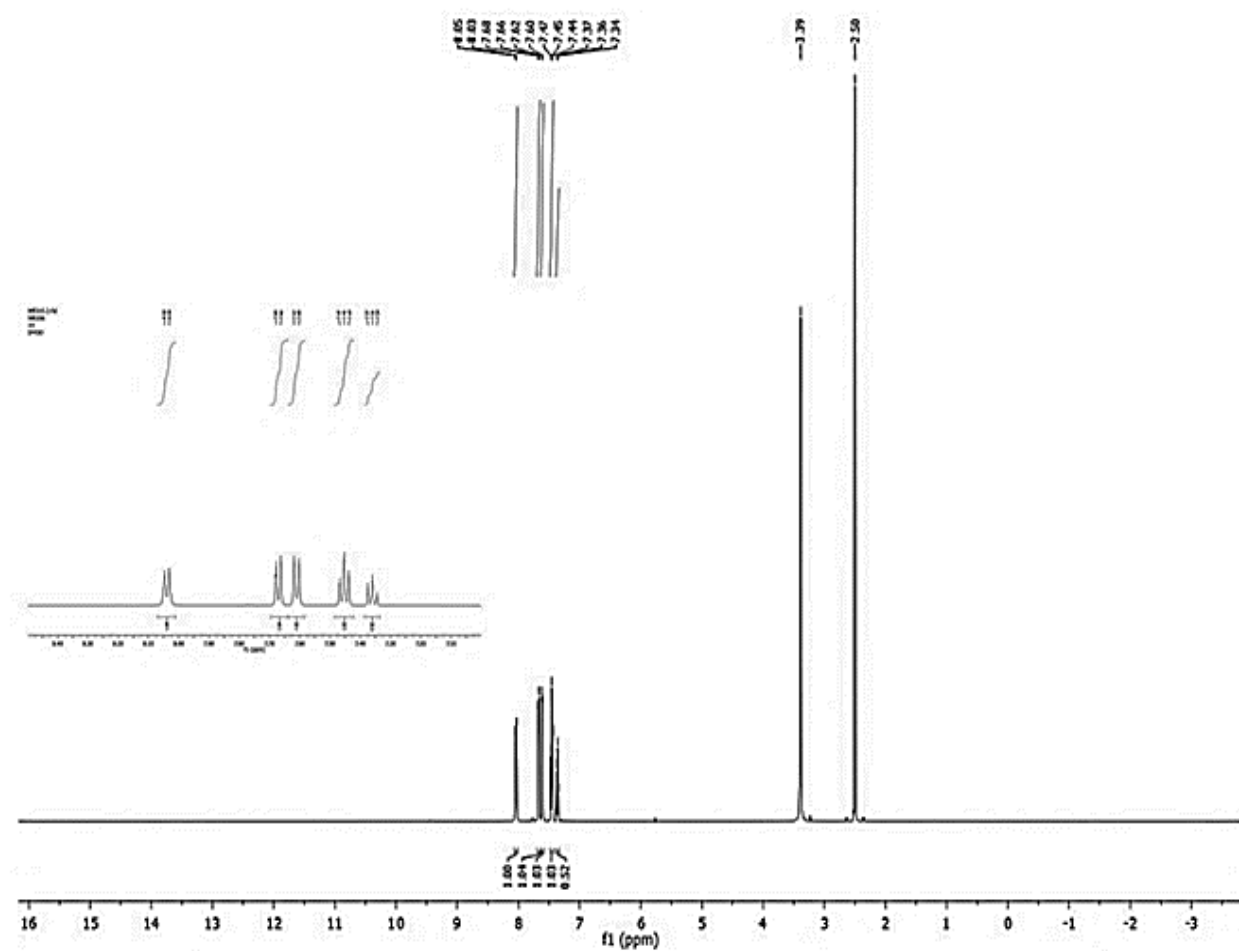
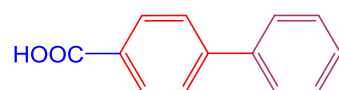


S3. The ^1H NMR of 3-amine-[1,1'-biphenyl]

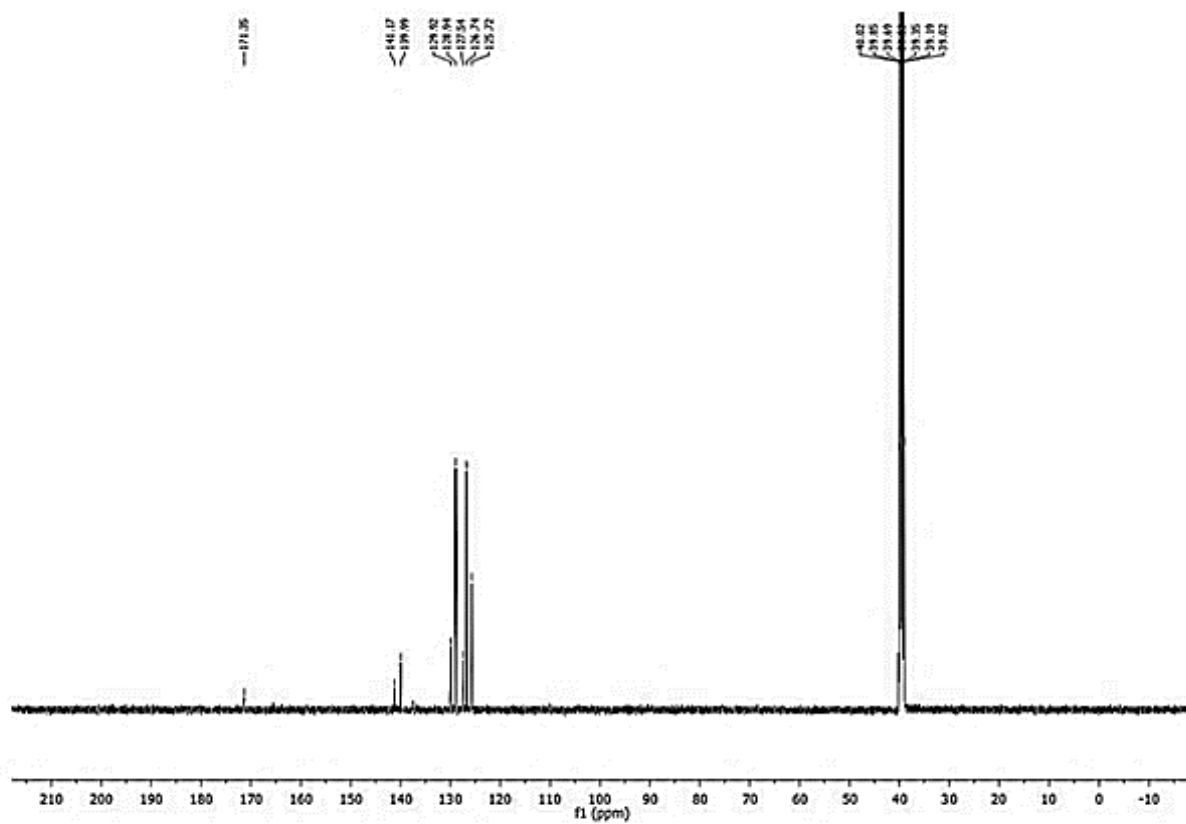


S4. The ^{13}C NMR of 3-amine-[1,1'-biphenyl]

4-carboxylic acid-[1,1'-biphenyl]

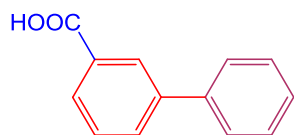


S5. The ¹H NMR of 4-carboxylic acid-[1,1'-biphenyl]



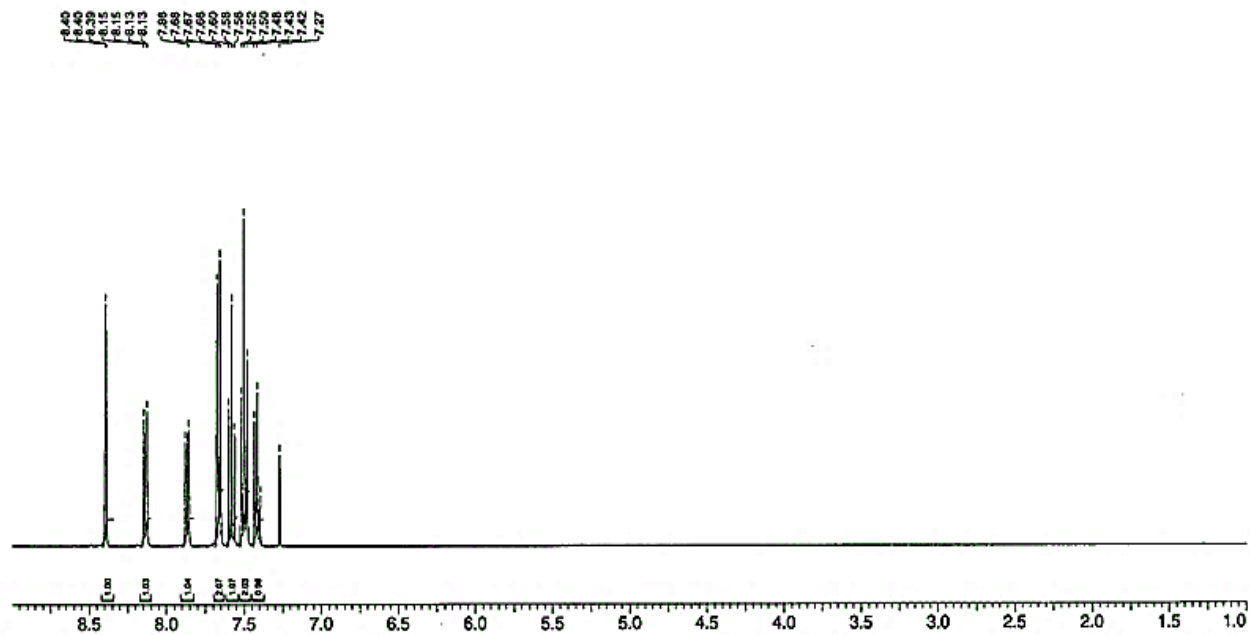
S6. The ^{13}C NMR of 4-carboxylic acid-[1,1'-biphenyl]

3-carboxylic acid-[1,1'-biphenyl]

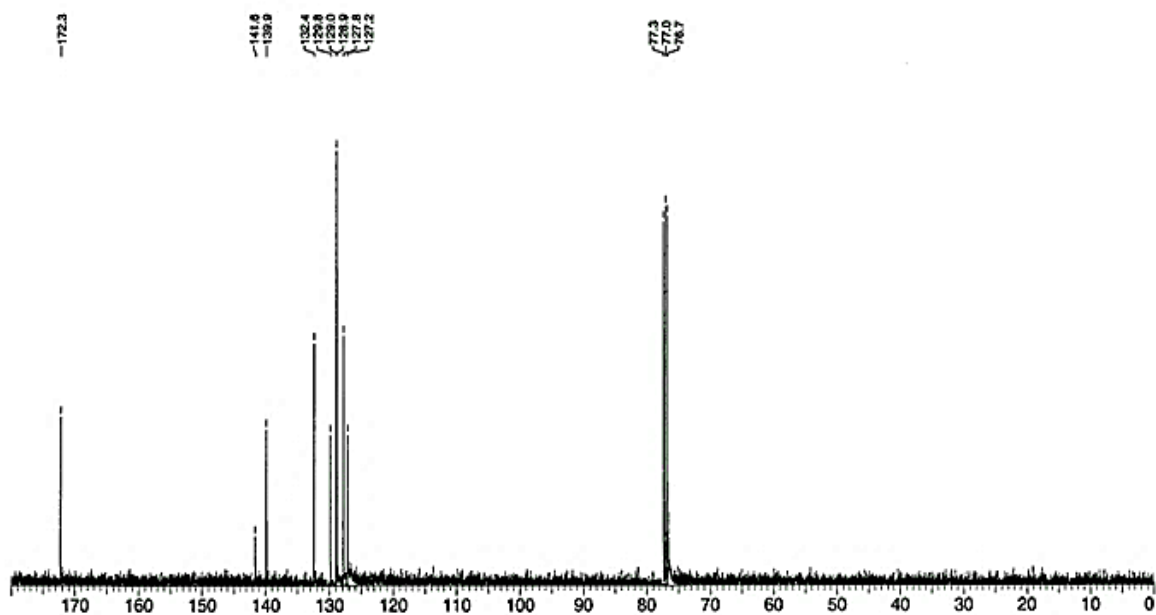


m.p. = 160-161 °C. ^1H NMR (400 MHz, DMSO): δ = 7.42 (1H, tt, J = 7.3, 1.1), 7.50 (2H, t, J = 7.3), 7.58 (1H, t, J = 7.7), 7.66 (2H, m), 7.87 (1H, dq, J = 7.7, 1.4), 8.14 (1H, dt, J = 7.7, 1.4), 8.40 (1H, t, J = 1.4).

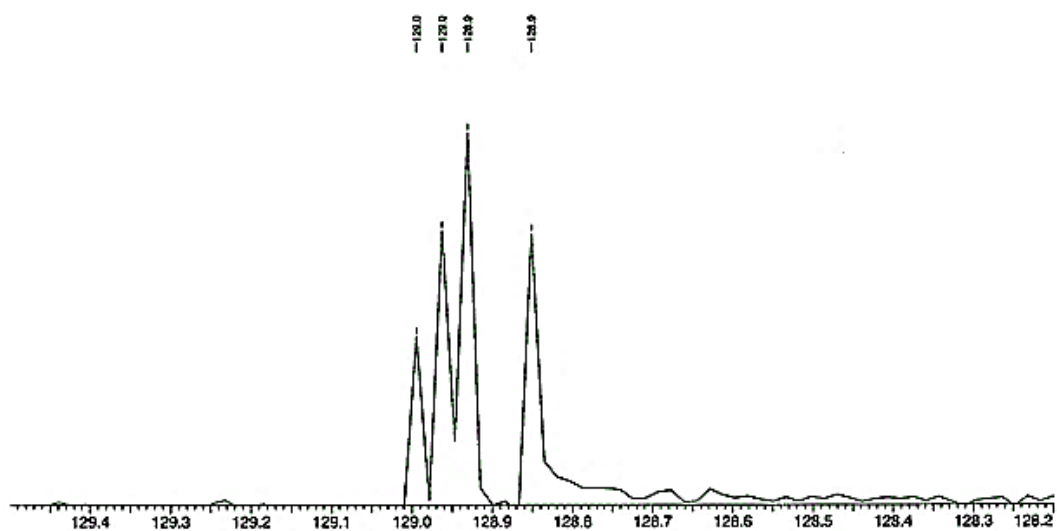
^{13}C NMR (100 MHz, DMSO): δ = 127.2, 127.8, 128.85, 128.93, 128.96, 128.99 (CH, Ar), 129.8 (Ar), 132.4 (CH Ar), 139.9, 141.6 ($2 \times$ CH Ar), 172.3 (C=O). m/z : 197.0



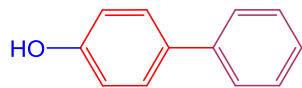
S7. The ^1H NMR of 3-carboxylic acid-[1,1'-biphenyl]



S8. The ^{13}C NMR of 3-carboxylic acid-[1,1'-biphenyl]



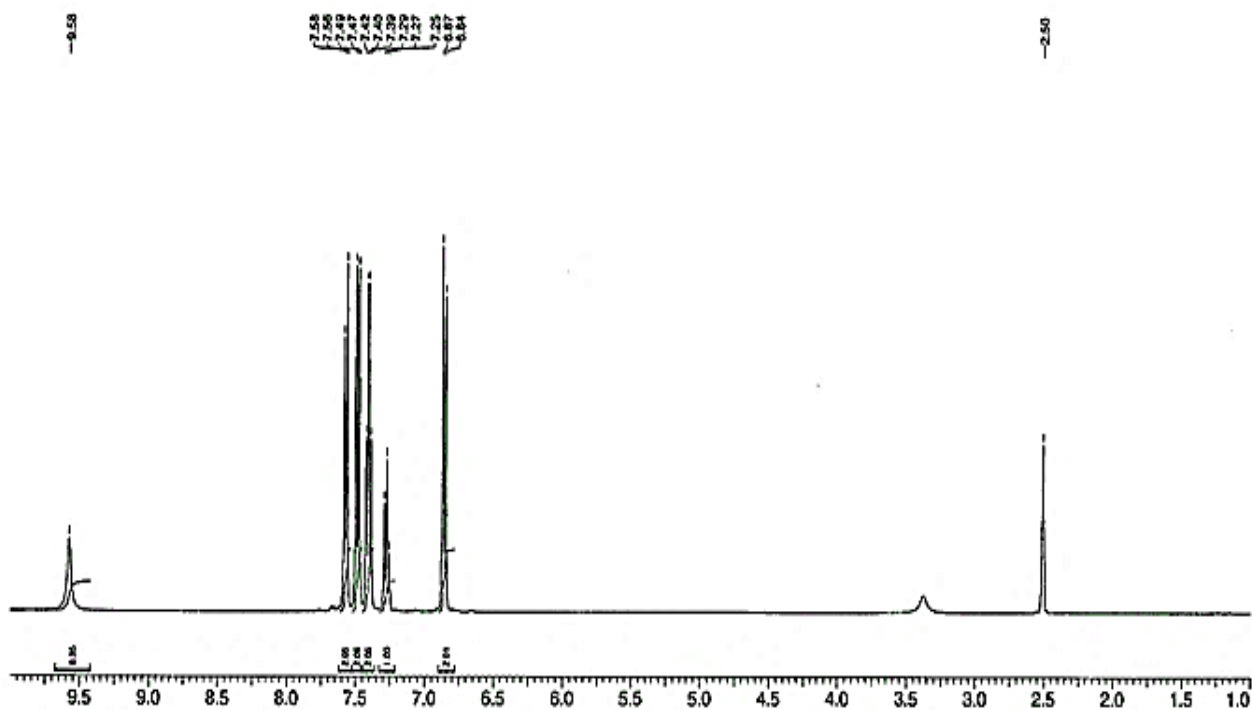
4-hydroxyl-[1,1'-biphenyl]



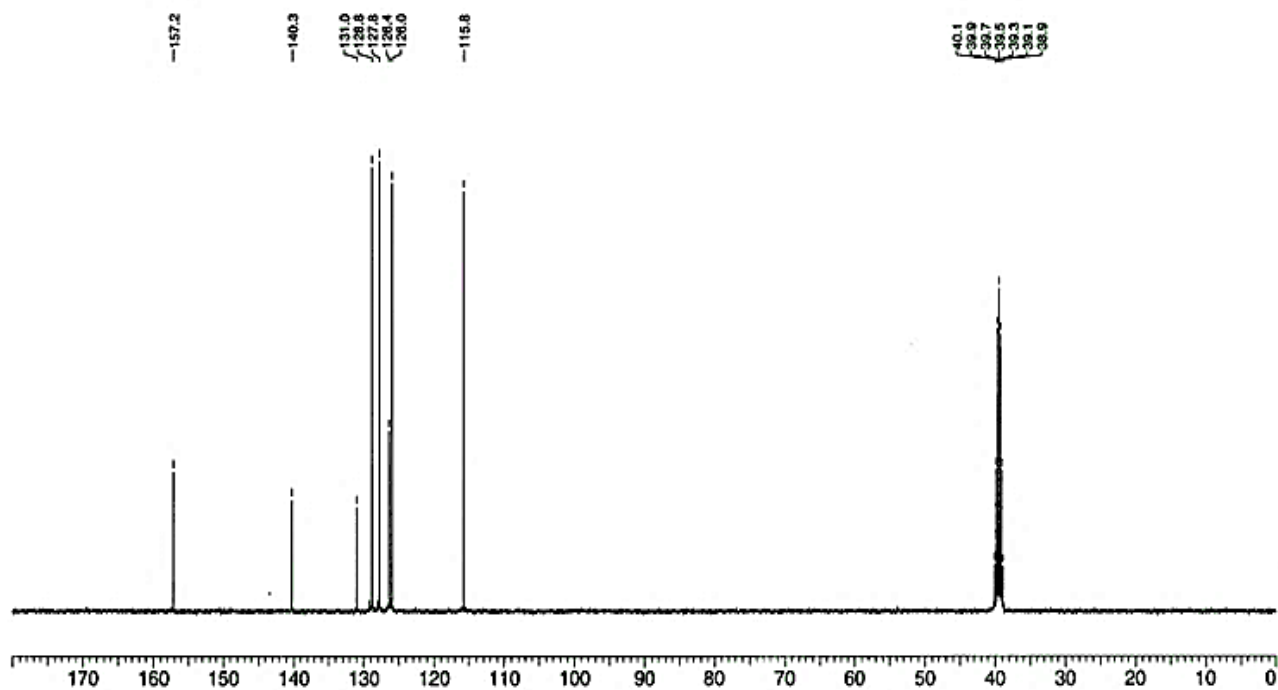
m.p. = 161-162 °C.

^1H NMR (400 MHz, DMSO): δ = 6.85 (2H, d, J = 8.5), 7.27 (1H, t, J = 7.4), 7.40 (2H, t, J = 7.4), 7.48 (2H, d, J = 8.5), 7.57 (2H, d, J = 8.5), 9.58 (1H, br s, OH). ^{13}C NMR (100 MHz, DMSO):

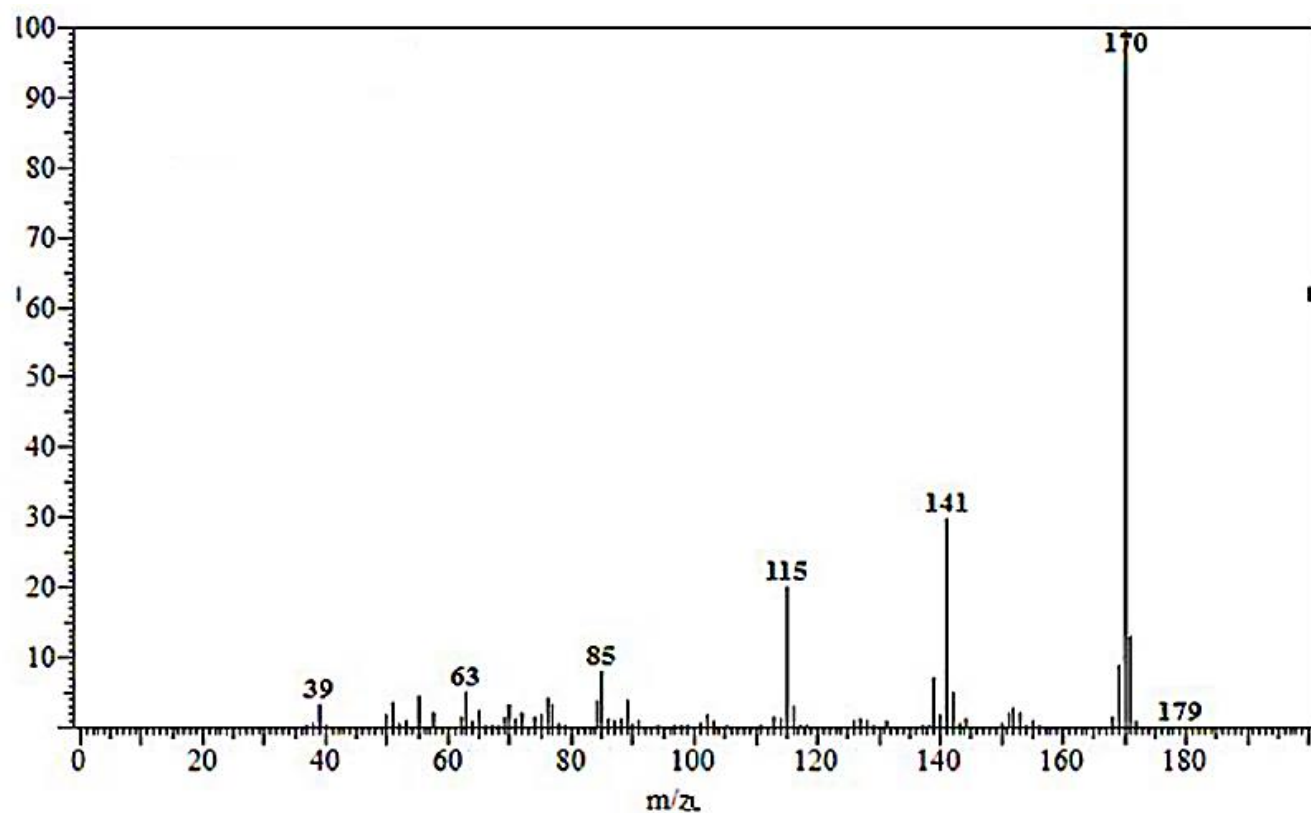
δ = 115.8, 126.0, 126.4, 127.8, 128.8 ($5 \times \text{CH}$, Ar), 131.0, 140.3, 157.2 (Ar). m/z : 169.1.



S9. The ^1H NMR of 4-hydroxyl-[1,1'-biphenyl]



S10. The ¹³CNMR of 4-hydroxyl-[1,1'-biphenyl]

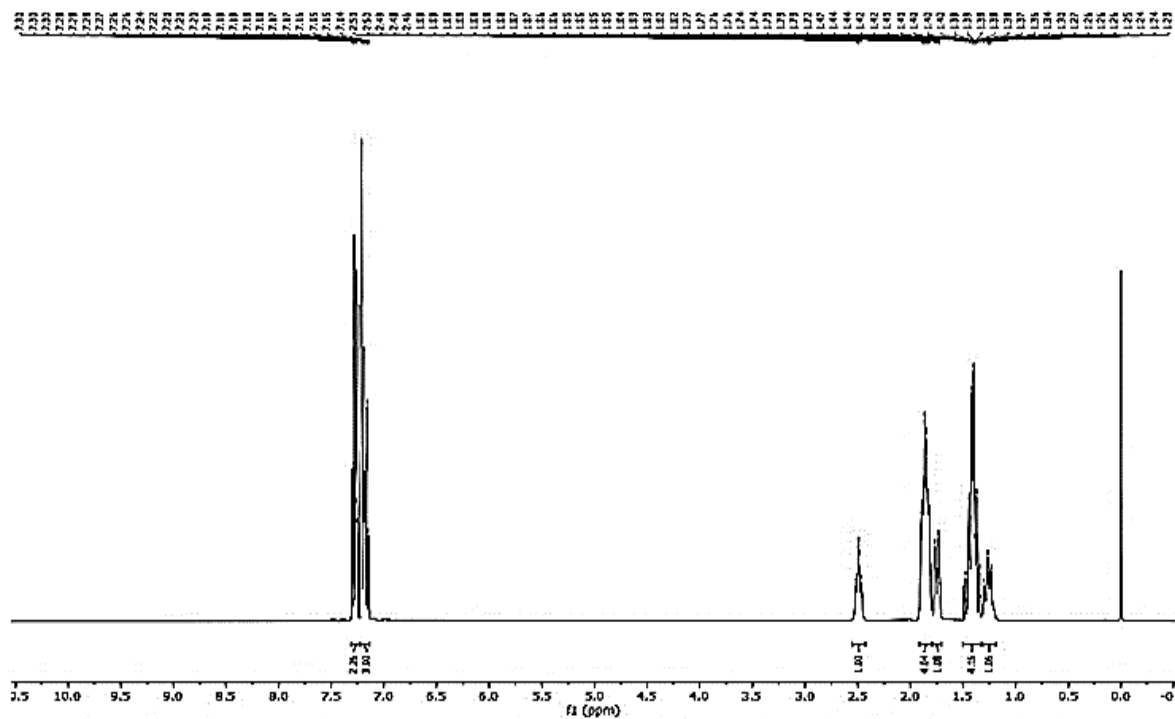


S11. The GC-Mass of 4-hydroxyl-[1,1'-biphenyl]

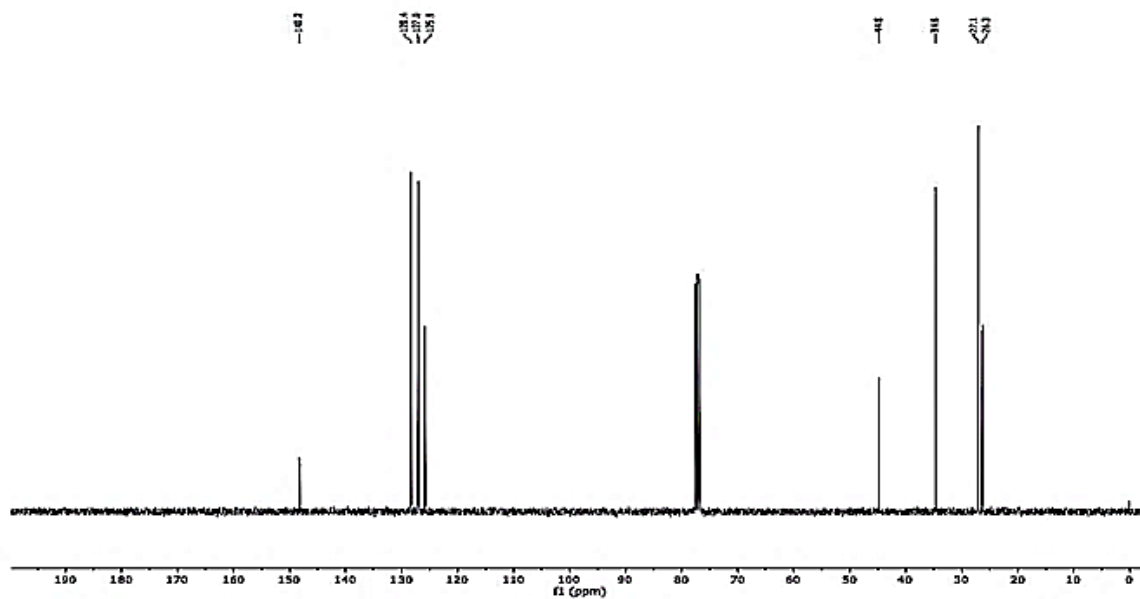
Cyclohexylbenzene



Colorless oil in 73% yield ^1H NMR (400 MHz, chloroform- d) δ = 7.31 – 7.25 (m, 2H), 7.23 – 7.13 (m, 3H), 2.49 (m, 1H), 1.92 – 1.82 (m, 4H), 1.74 (m, 1H), 1.51 – 1.34 (m, 4H), 1.25 (m, 1H); ^{13}C NMR (100 MHz, chloroform- d) δ = 148.2, 128.4, 127.0, 125.9, 44.8, 34.6, 27.1, 26.3.

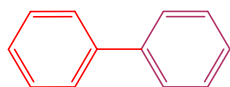


S12. The ^1H NMR of Cyclohexylbenzene



S13. The ^{13}C NMR of Cyclohexylbenzene

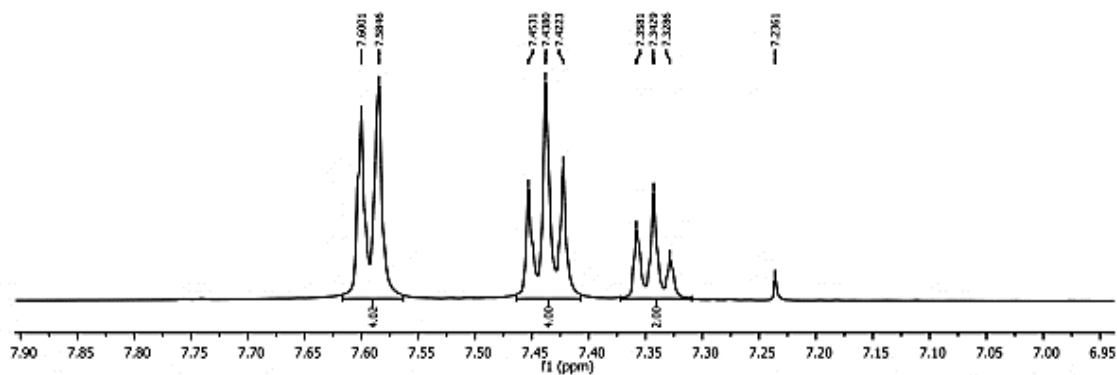
Biphenyl



white solid in 94% yield, MP: 69-71 °C

^1H NMR (500 MHz, CDCl_3): δ 7.59 (d, J = 7.7 Hz, 4H), 7.44 (t, J = 7.7 Hz, 4H), 7.34 (t, J = 7.4 Hz, 2H);

^{13}C NMR (126 MHz, CDCl_3): δ 141.29, 128.82, 127.27; FTIR (cm^{-1}): 2358, 1510, 1029, 580; GC-MS (EI, 70eV) m/z : found: 154 ($\text{C}_{12}\text{H}_{10}$).

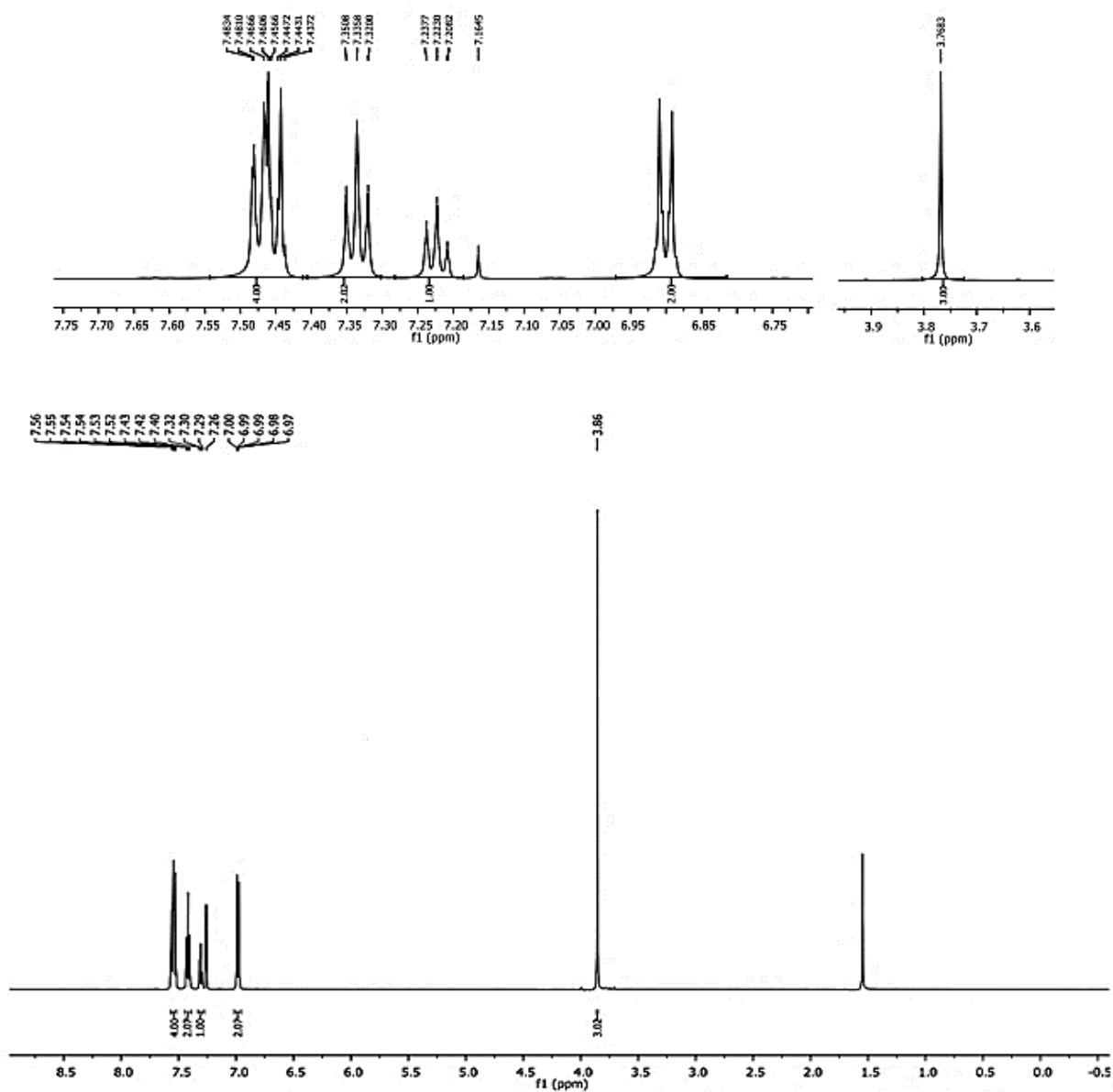




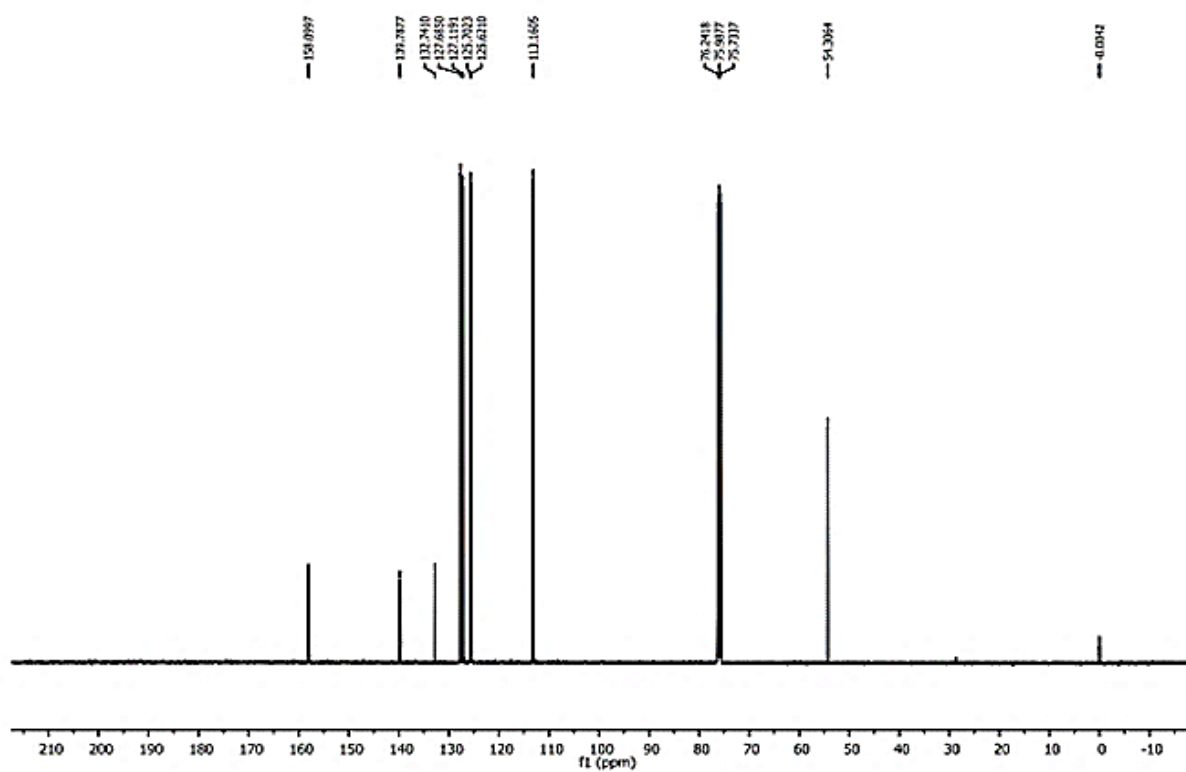
4-Methoxy-1,1'-biphenyl



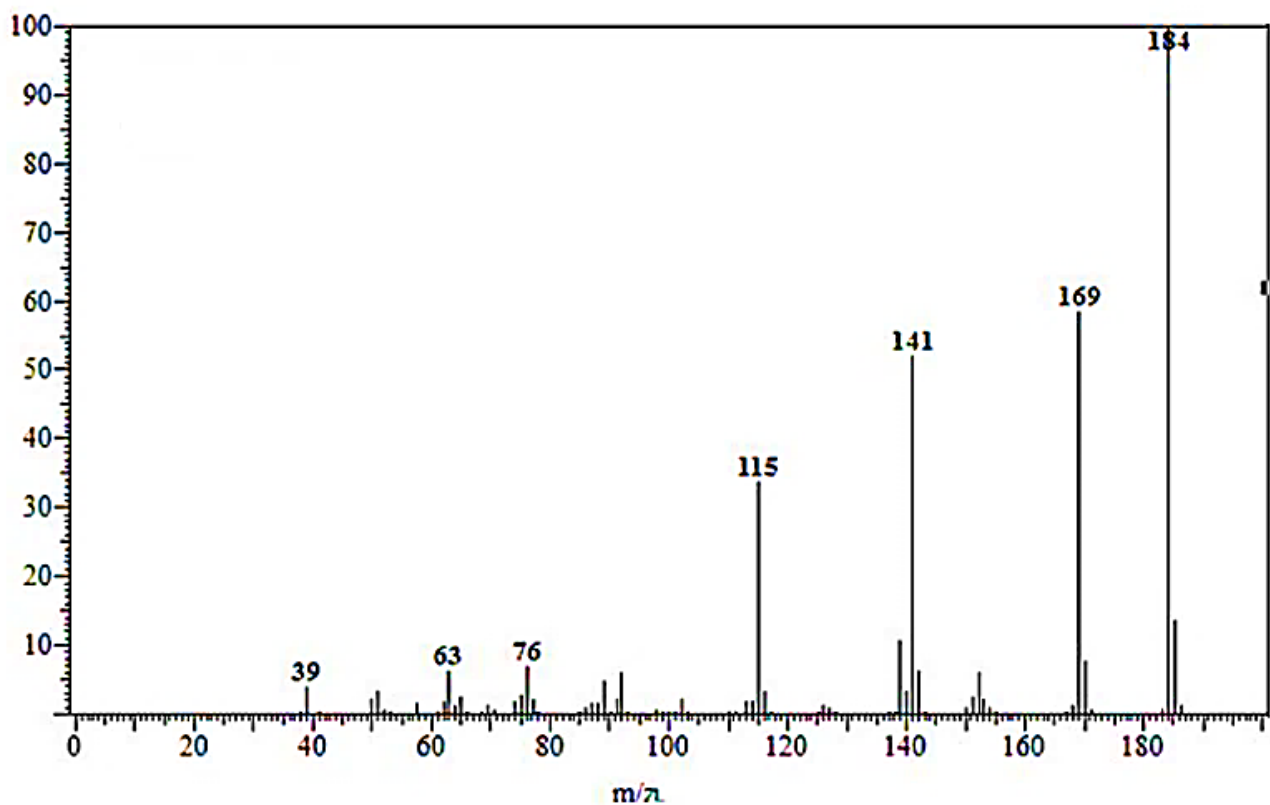
White waxy solid, yield 90%, ^1H NMR (500 MHz, CDCl_3): 7.57-7.52 (m, 4H), 7.41 (t, $J = 8.0$ Hz, 2H), 7.30 (t, $J = 8.0$ Hz, 1H), 6.98 (d, $J = 8.5$ Hz, 2H), and 3.86 (s, 3H).



S16. The ^1H NMR of 4-Methoxy-1,1'-biphenyl



S17. The ^{13}C NMR of 4-Methoxy-1,1'-biphenyl

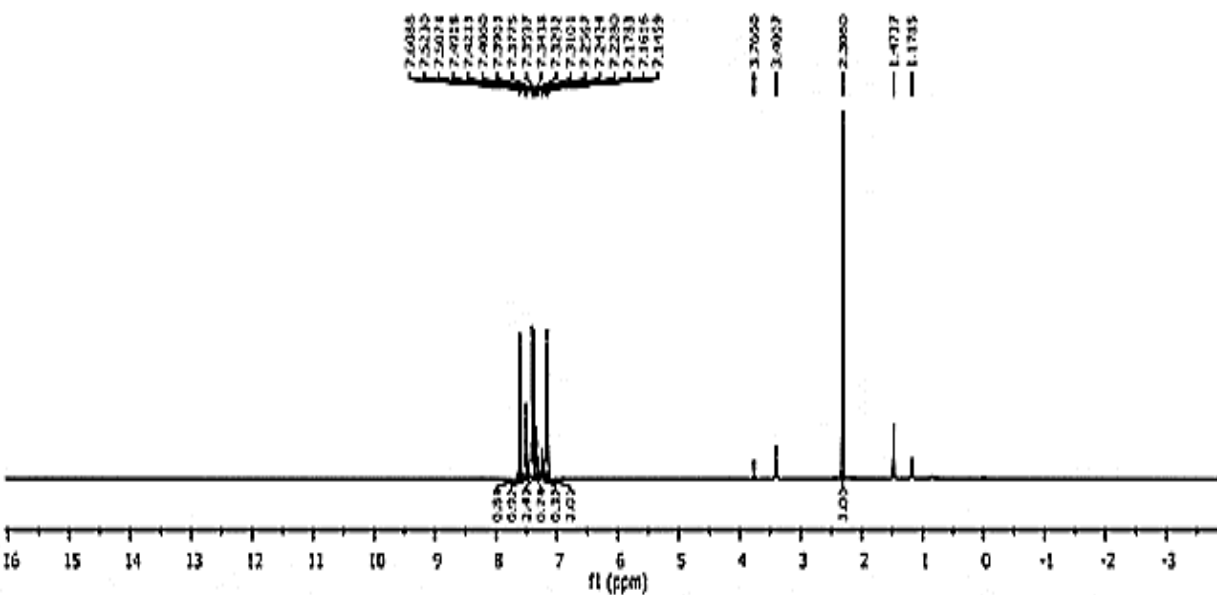


S18. The GC-Mass of 4-Methoxy-1,1'-biphenyl

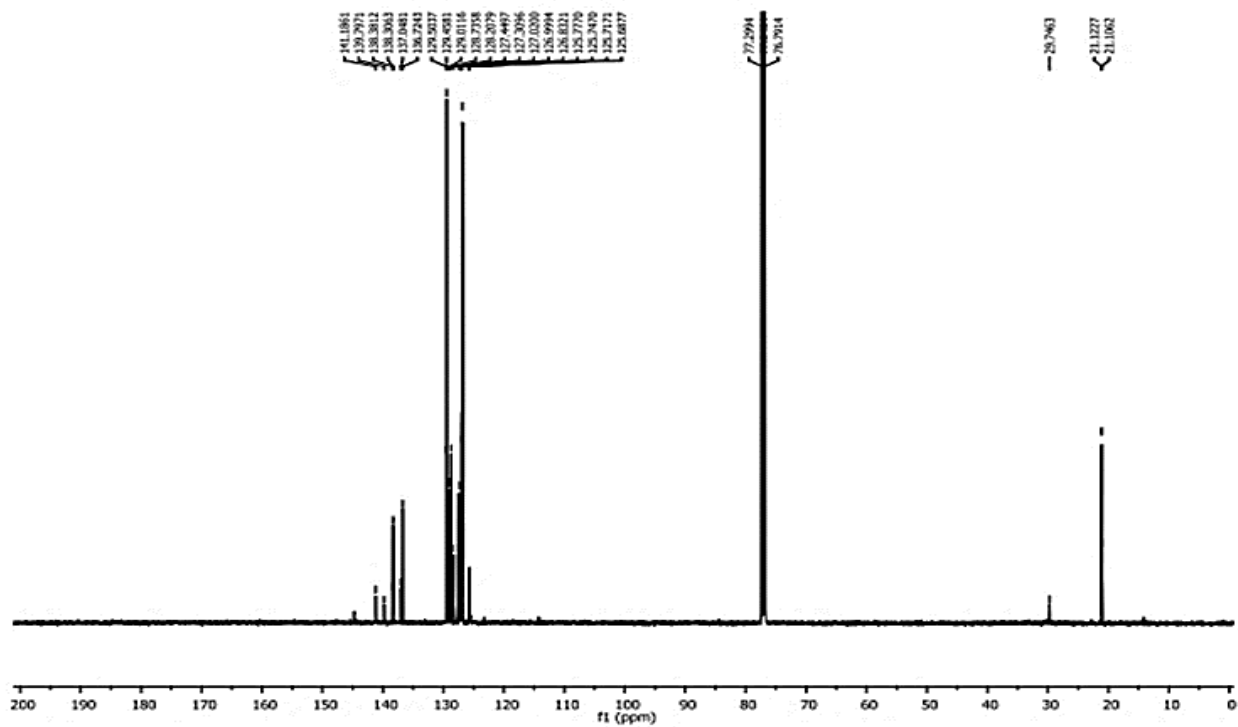
4-Methyl-1,1'-biphenyl



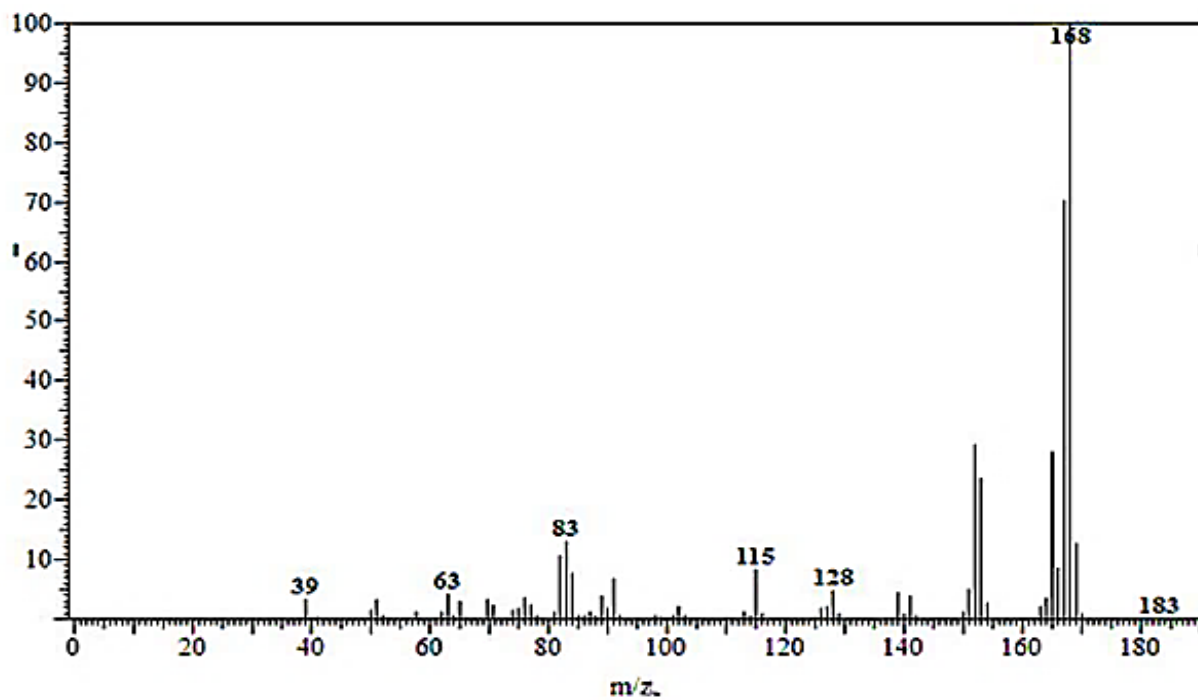
white yellow solid in 92% yield, MP: 46-48 °C; ^1H NMR (500 MHz, CDCl_3): δ 7.61 (s, 1H), 7.51 (t, J = 7.8 Hz, 1H), 7.39 (m, 3H), 7.34 (m, 1H), 7.24 (t, J = 7.2 Hz, 1H), 7.16 (t, J = 8.1 Hz, 2H), 2.31 (s, 3H); ^{13}C NMR (126 MHz, CDCl_3): δ 141.19, 139.80, 138.34, 137.05, 136.72, 129.48, 129.01, 128.74, 128.21, 127.38, 127.01, 126.83, 125.73, 21.11; FTIR (cm^{-1}): 2918, 2318, 1743, 1521, 1325, 1114, 802; GC-MS m/z : 168.



S30

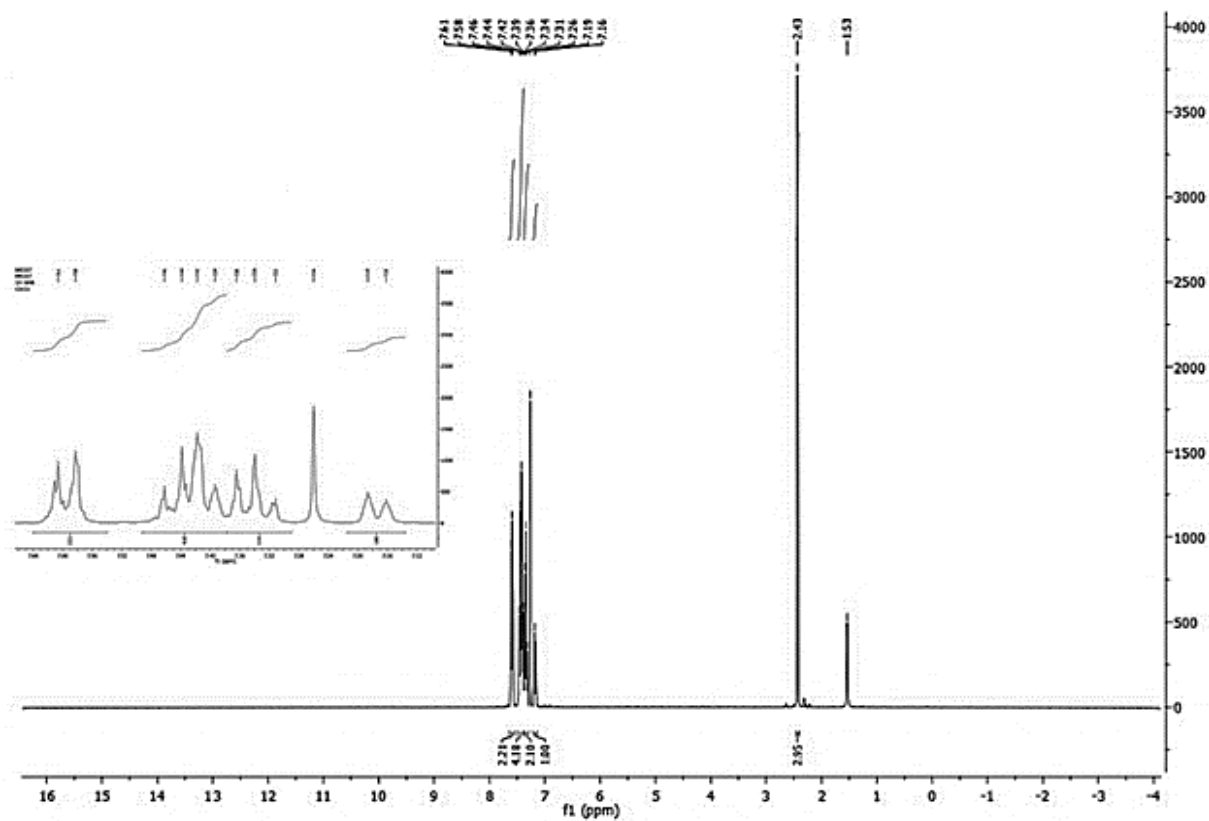
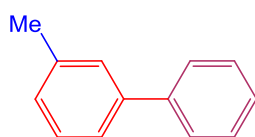


S20. The ^{13}C NMR of 4-Methyl-1,1'-biphenyl

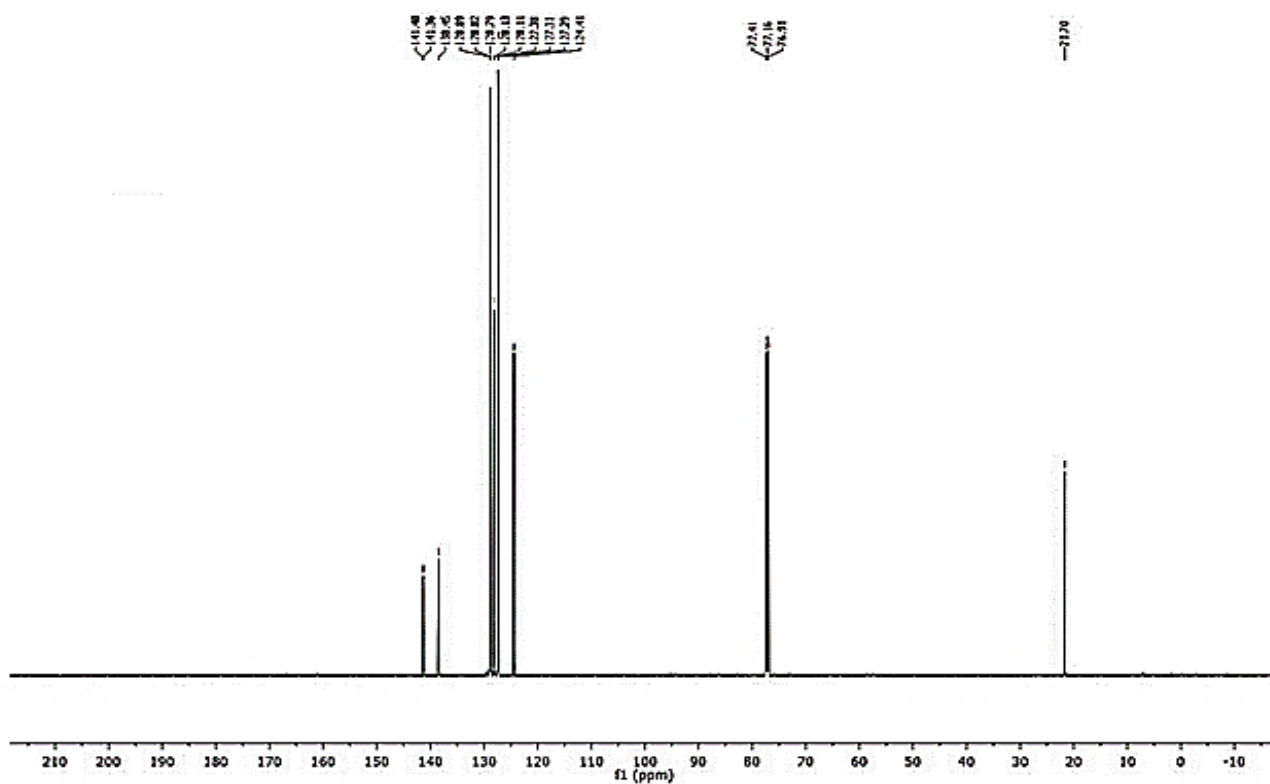


S21. The GC-Mass of 4-Methyl-1,1'-biphenyl

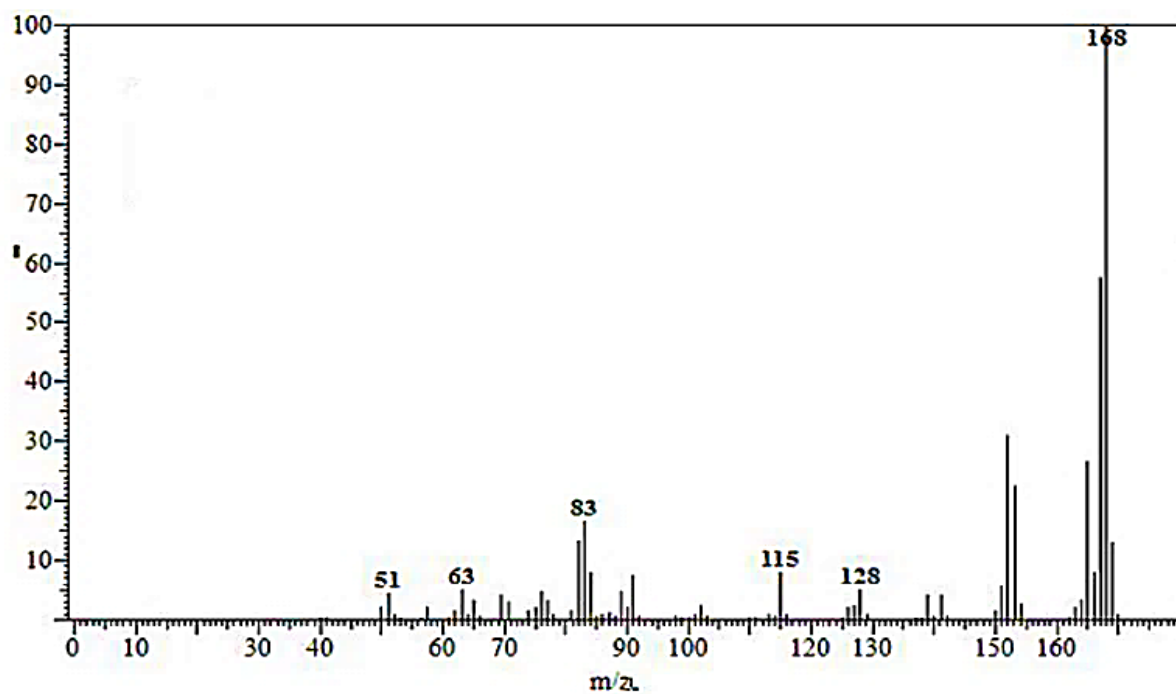
3-methyl-1,1'-bipheny



S22. The ^1H NMR of 3-methyl-1,1'-bipheny

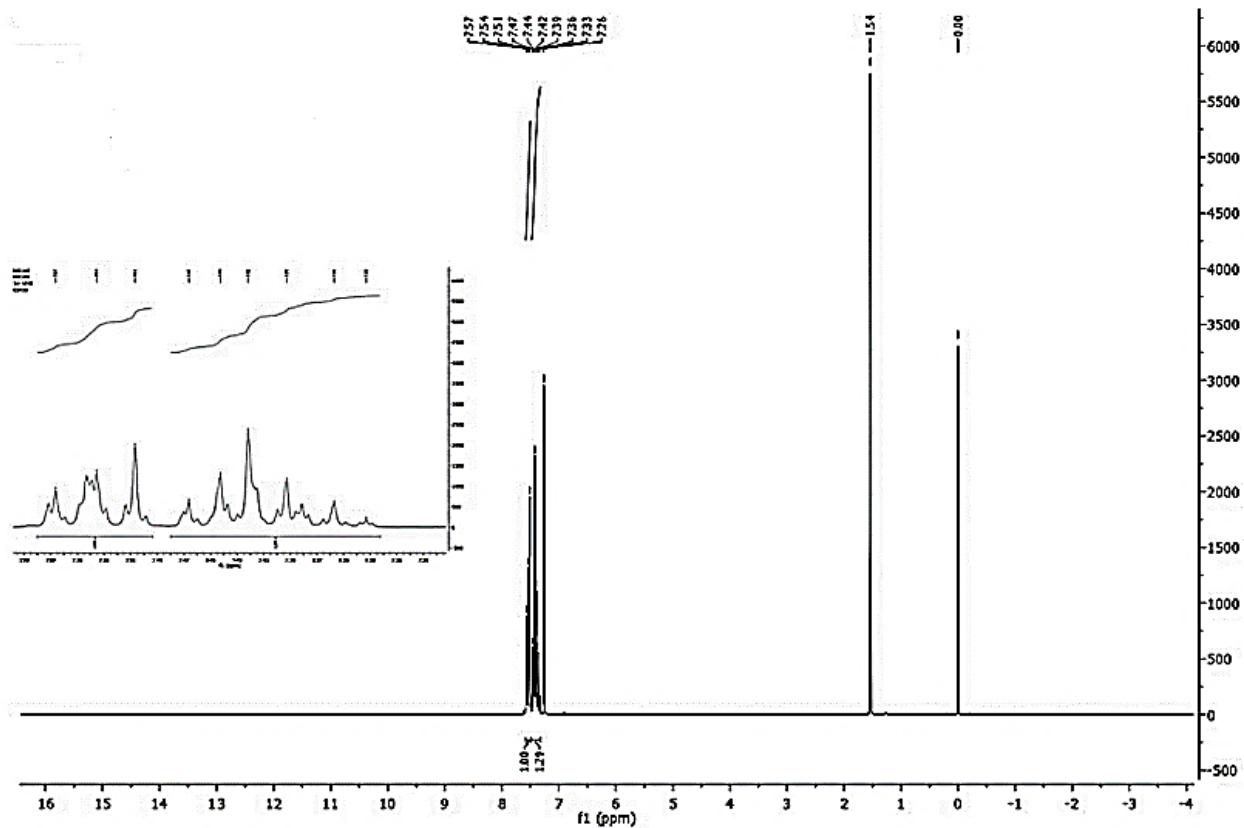
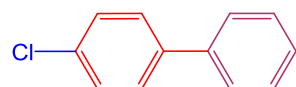


S23. The ^{13}C NMR of 3-methyl-1,1'-biphenyl

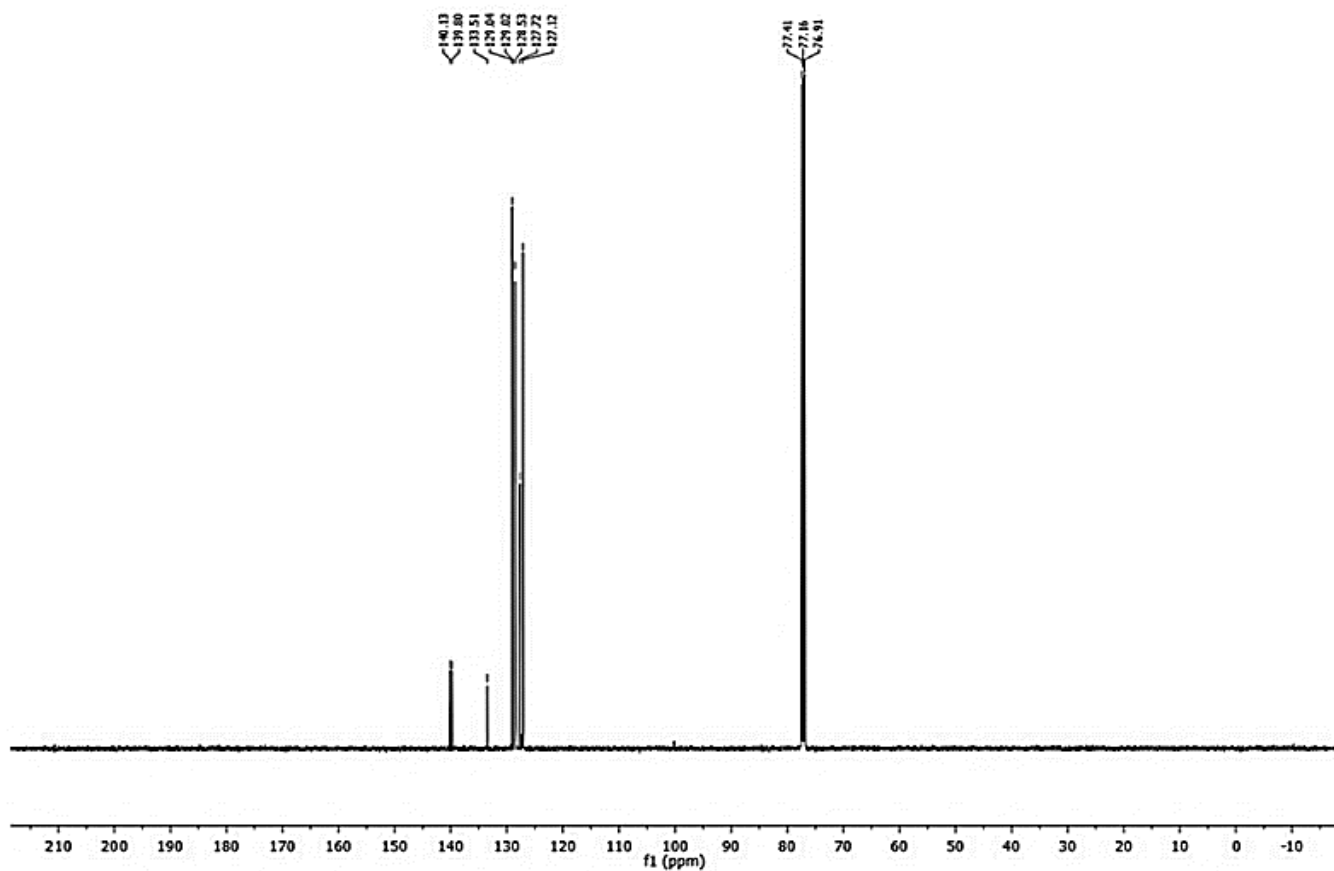


S24. The GC-Mass of 3-methyl-1,1'-biphenyl

4-chloro-1,1'-bipheny

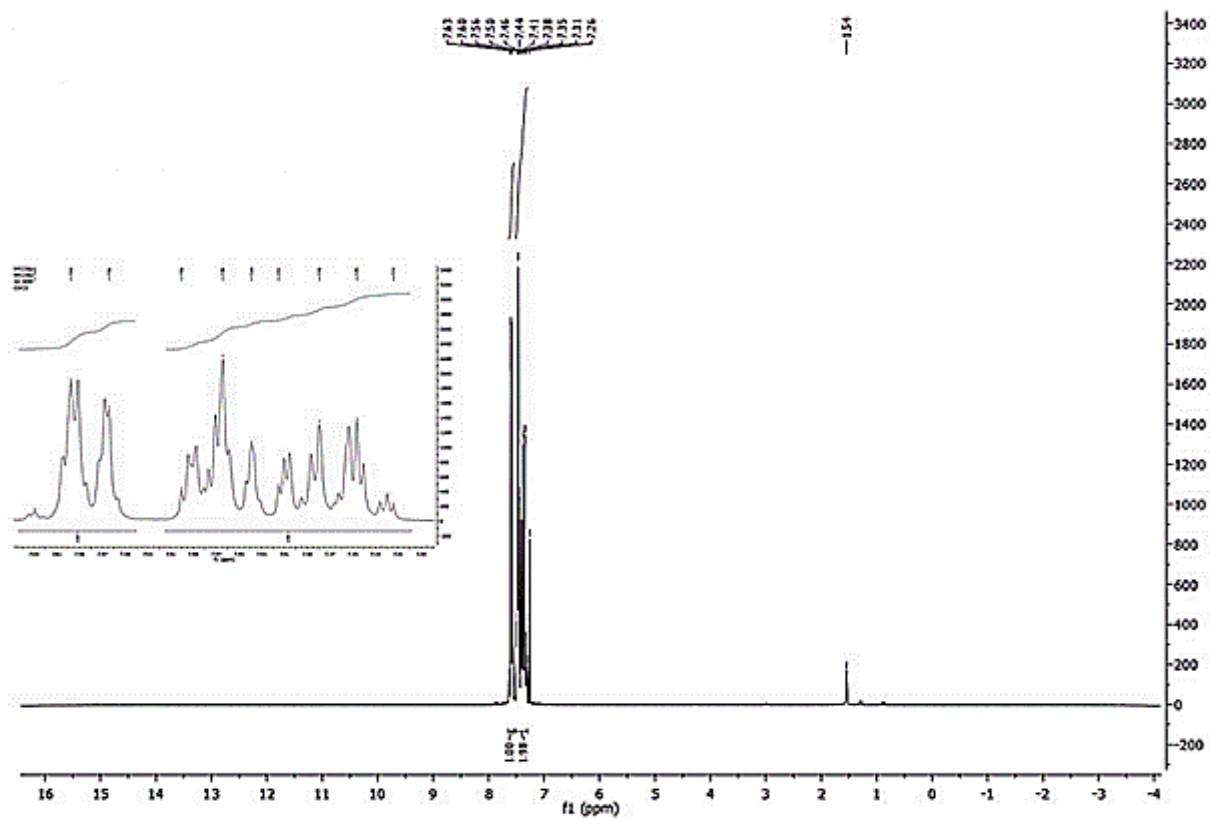


S25. The ^1H NMR of 4-chloro-1,1'-bipheny

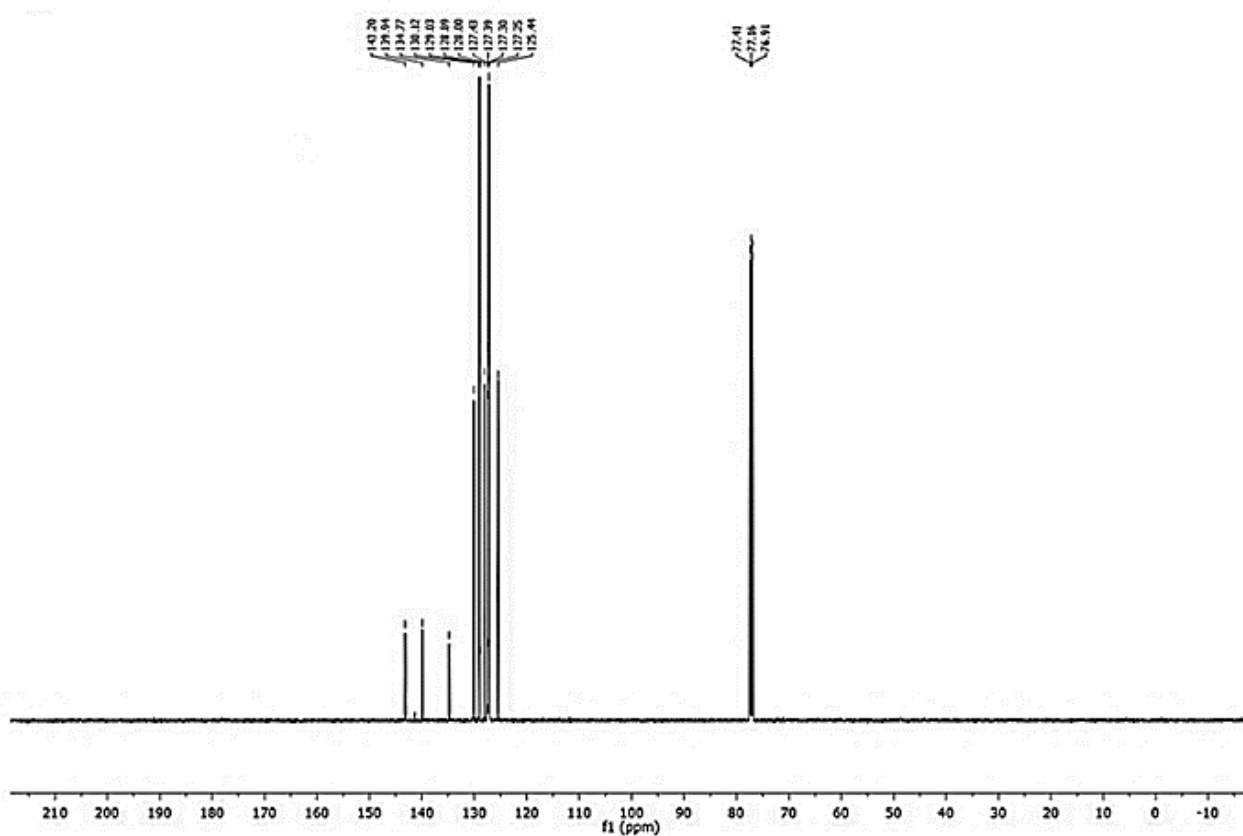


S26. The ^{13}C NMR of 4-chloro-1,1'-biphenyl

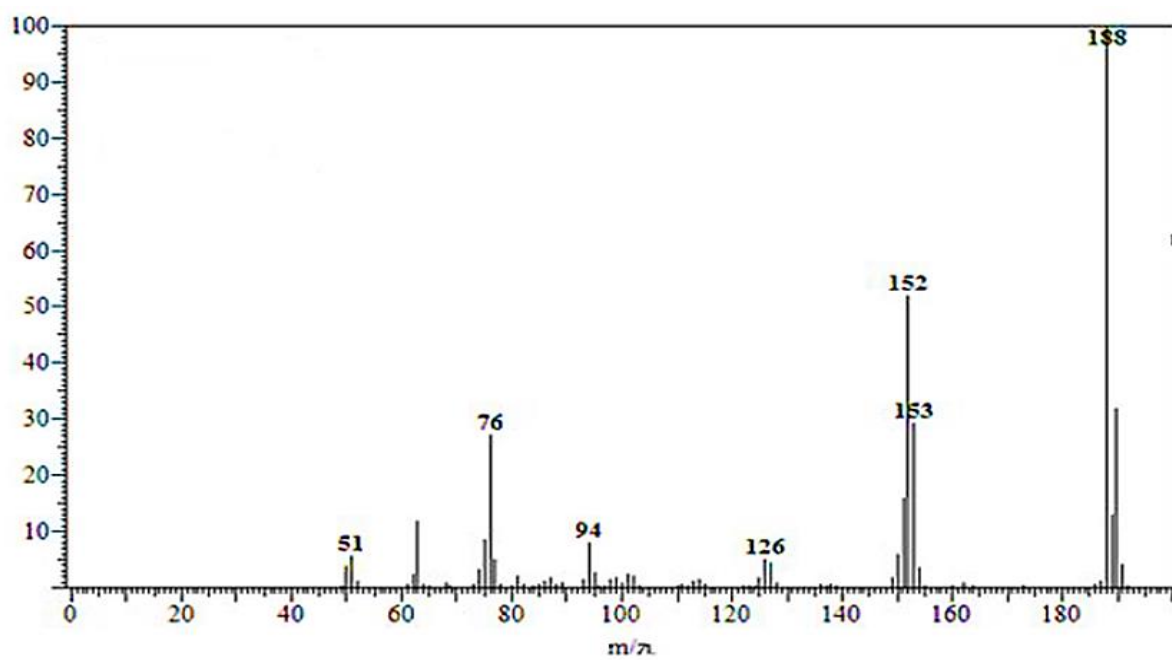
3-chloro-1,1'-bipheny



S27. The ^1H NMR of 3-chloro-1,1'-bipheny

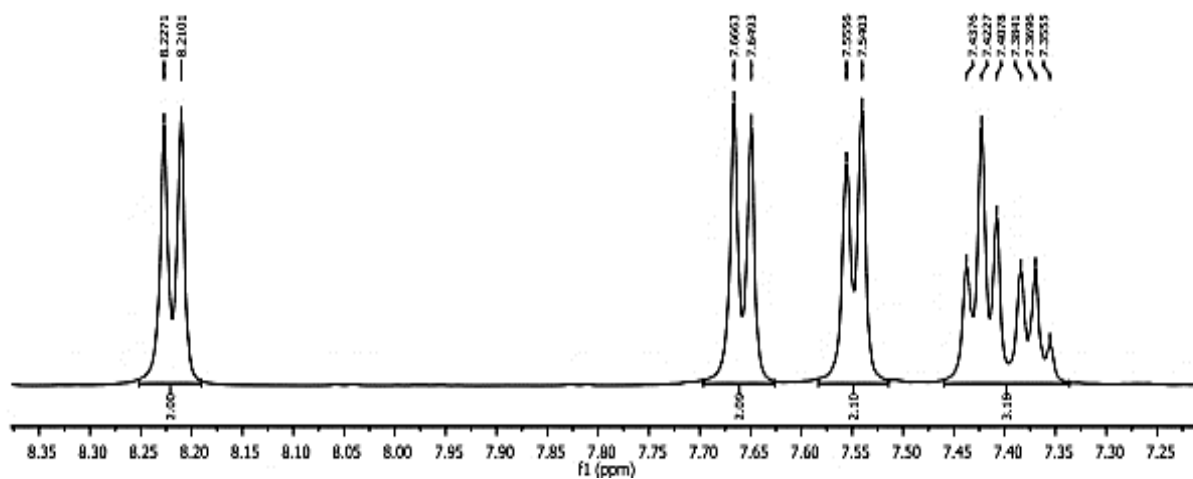
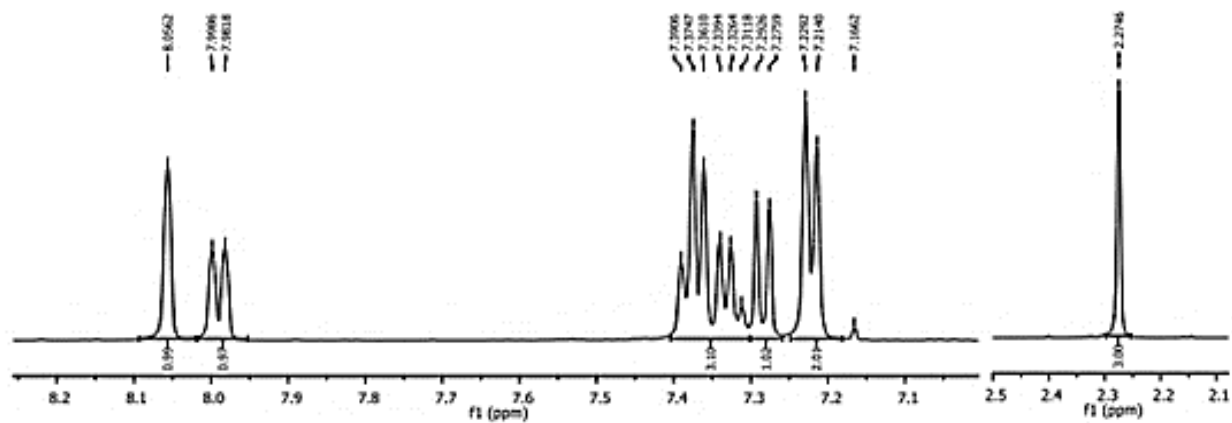
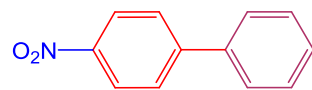


S28. The ¹³CNMR of 3-chloro-1,1'-biphenyl

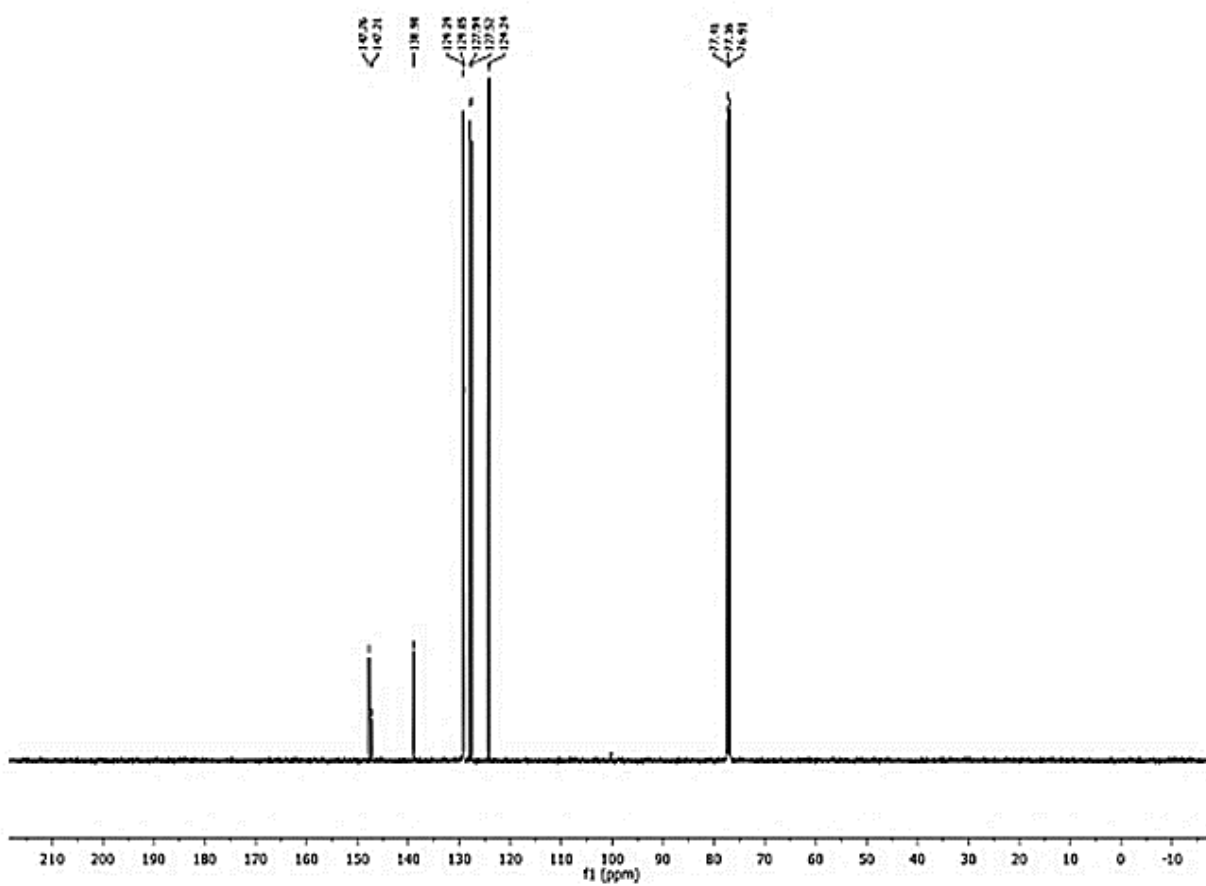


S29. The GC-Mass of 3-chloro-1,1'-biphenyl

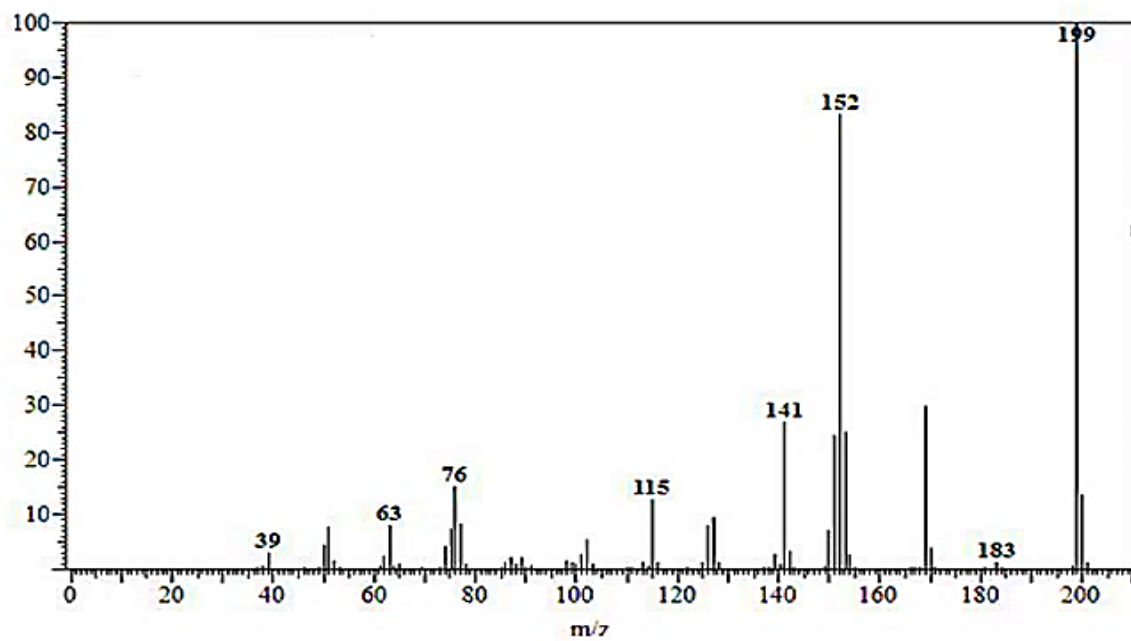
4-nitro-[1,1'-biphenyl]



S30. The ¹H NMR of 4-nitro-[1,1'-biphenyl]

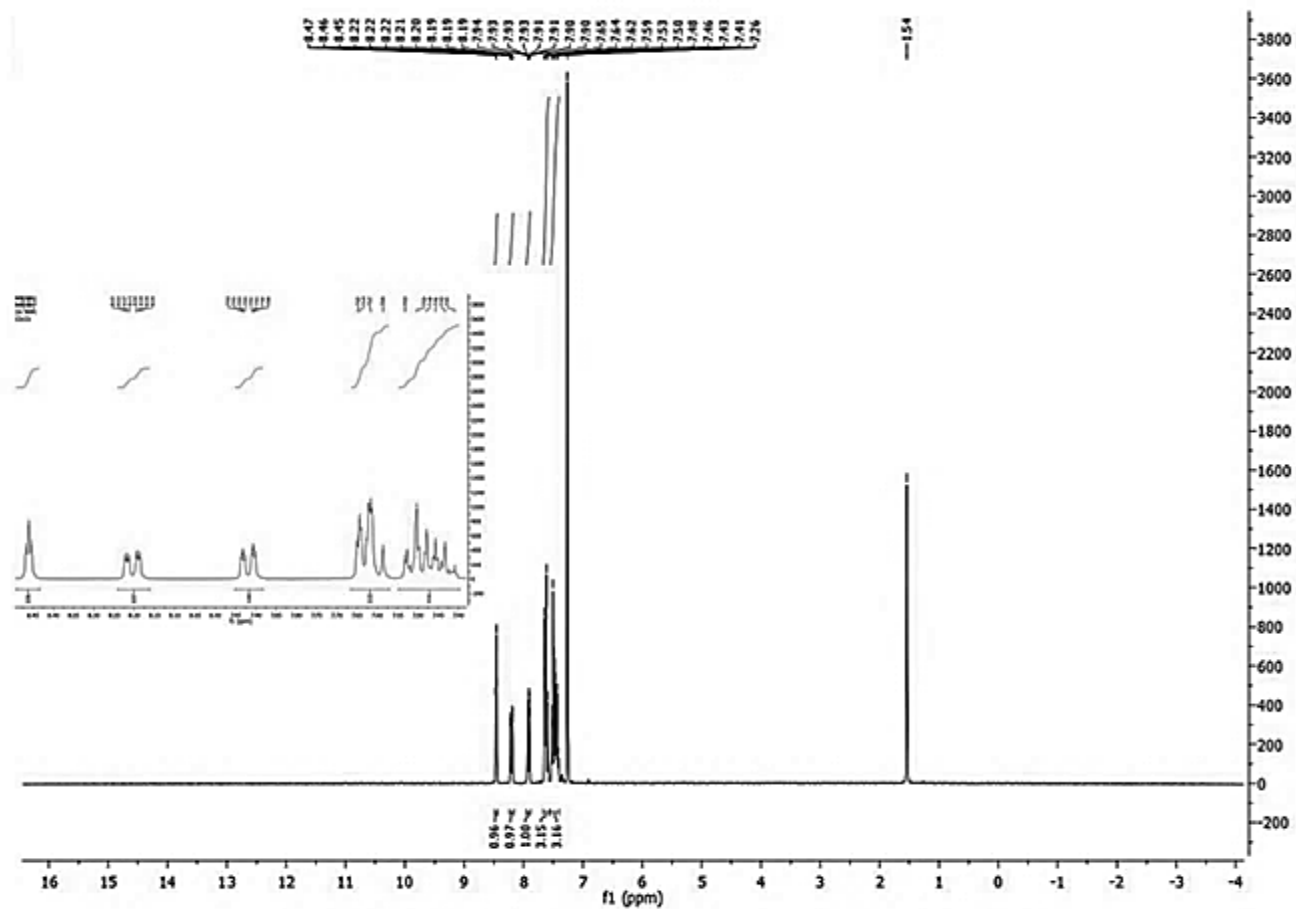
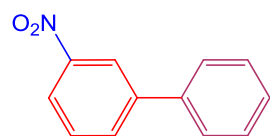


S31. The ^{13}C NMR of 4-nitro-[1,1'-biphenyl]

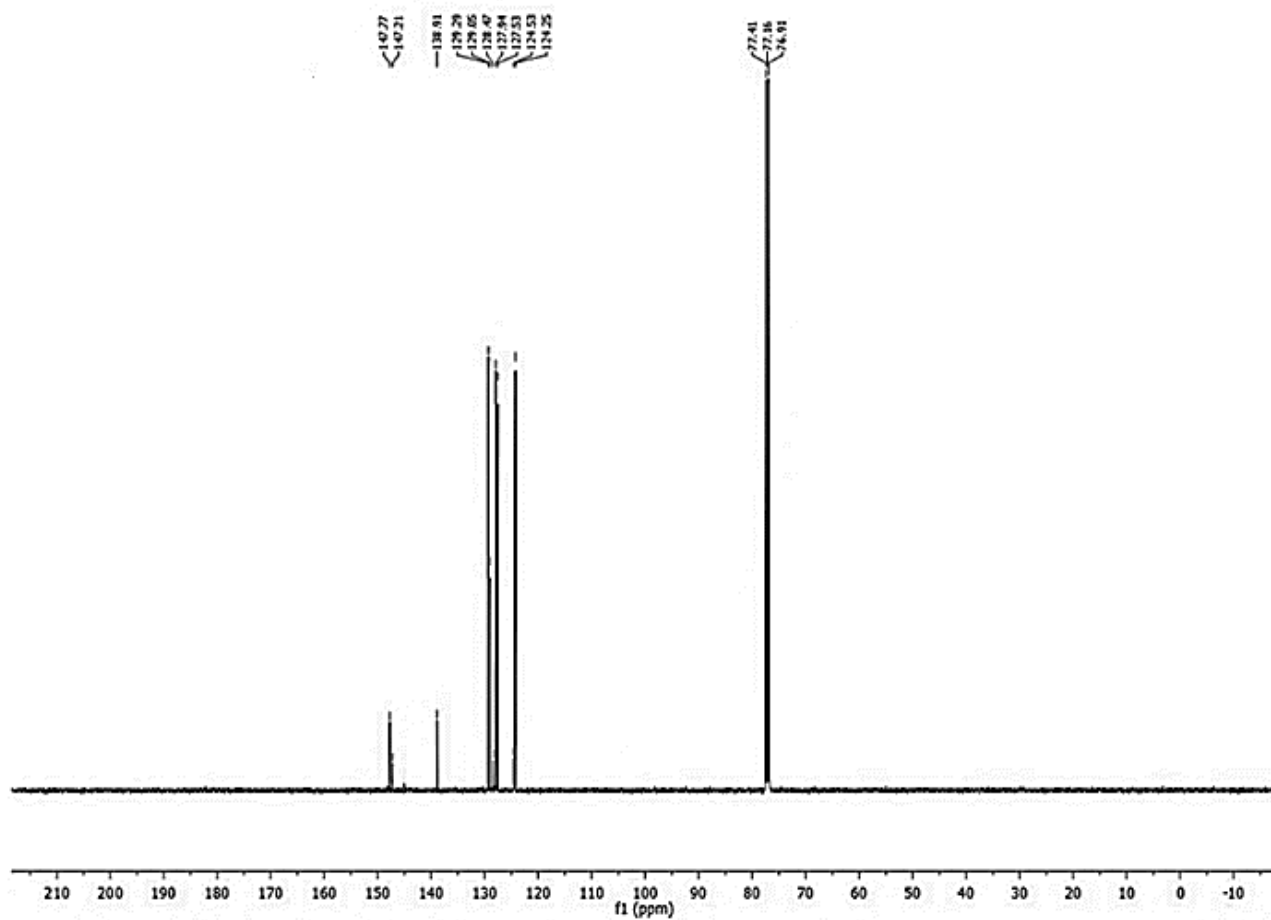


S32. The GC-Mass of 4-nitro-[1,1'-biphenyl]

3-nitro-1,1'-bipheny

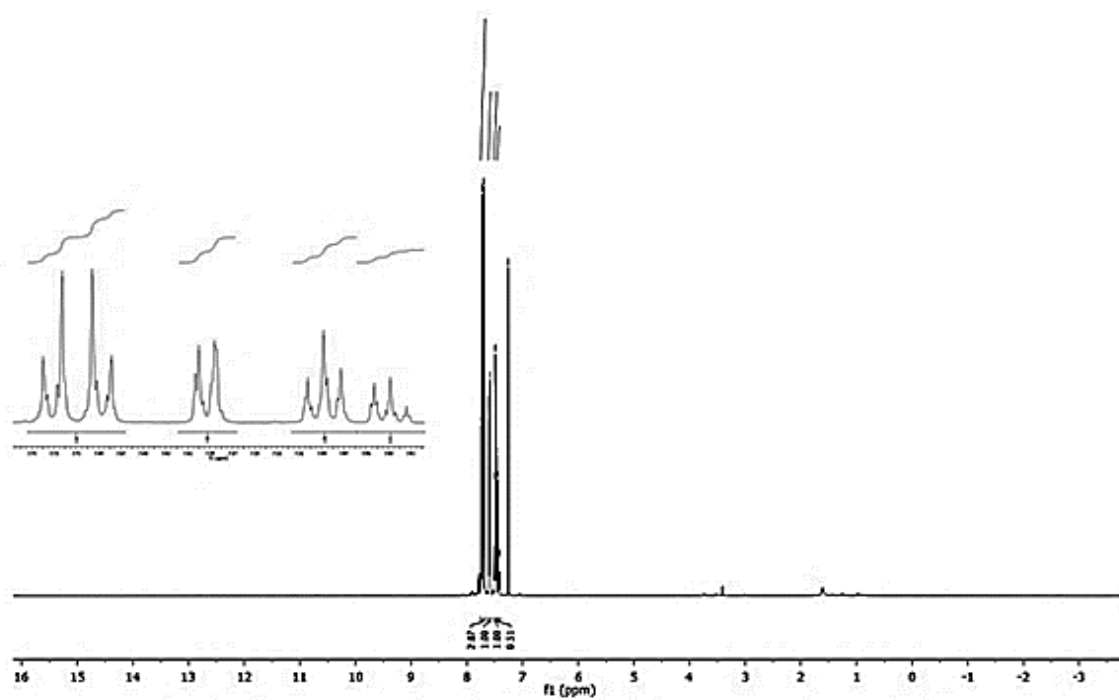
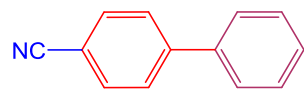


S33. The ¹H NMR of 3-nitro-[1,1'-biphenyl]

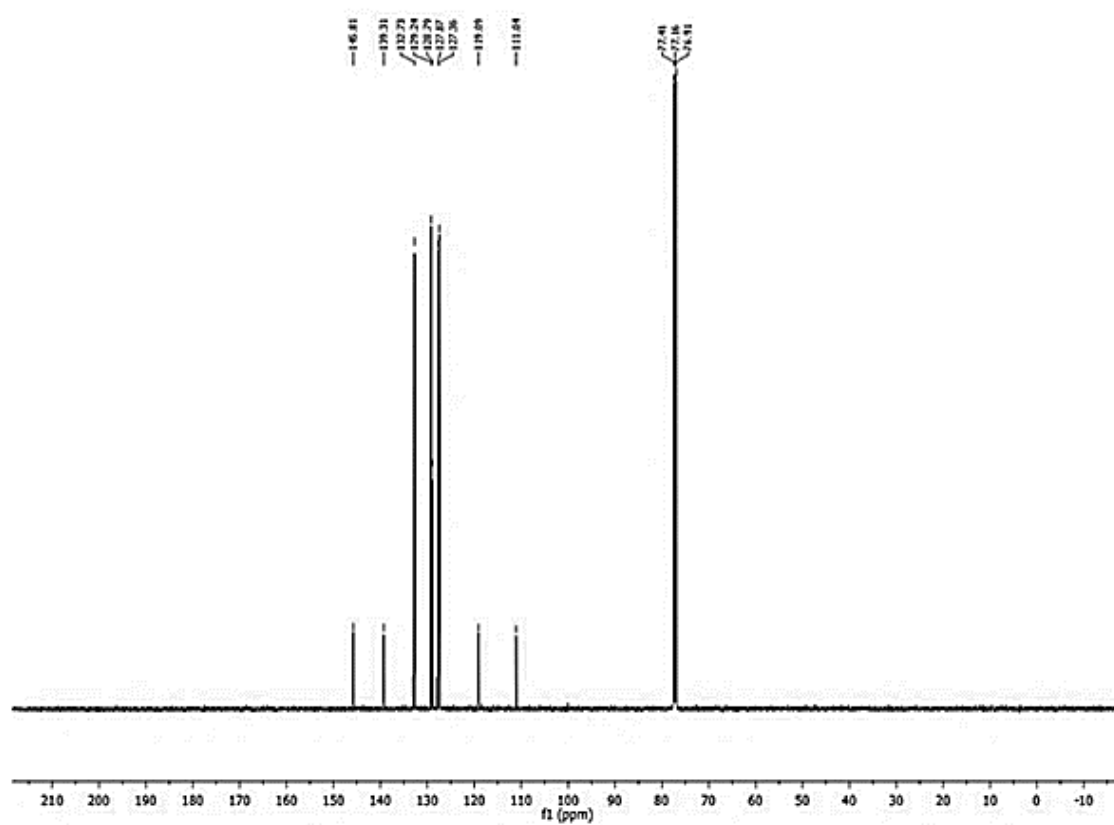


S34. The ^{13}C NMR of 3-nitro-[1,1'-biphenyl]

4-carbonitril-[1,1'-biphenyl]

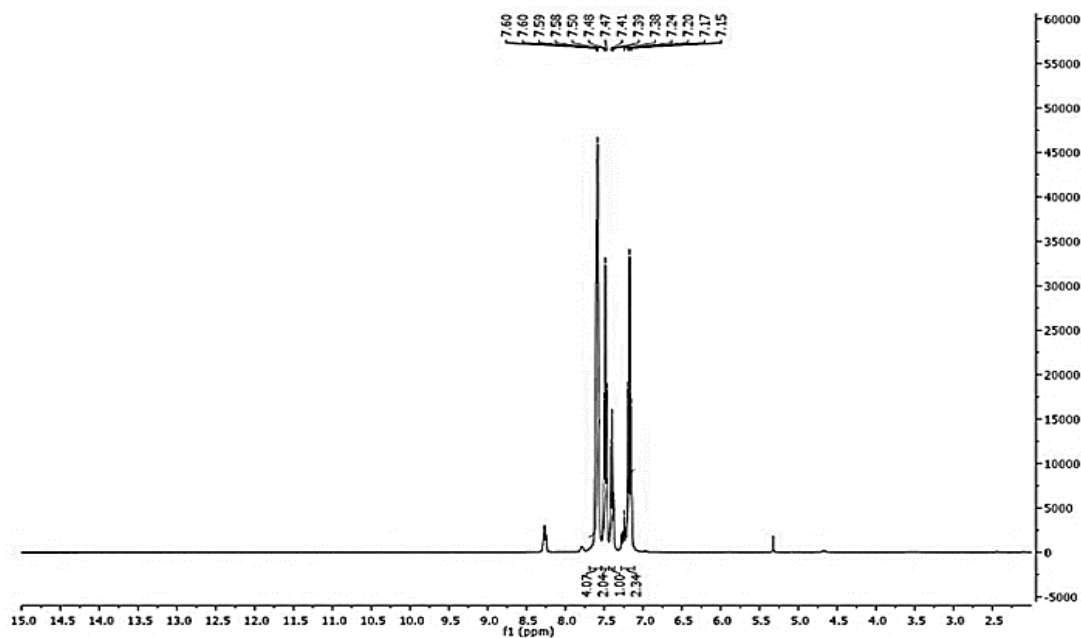
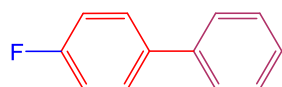


S35. The ^1H NMR of 4-carbonitril-[1,1'-biphenyl]

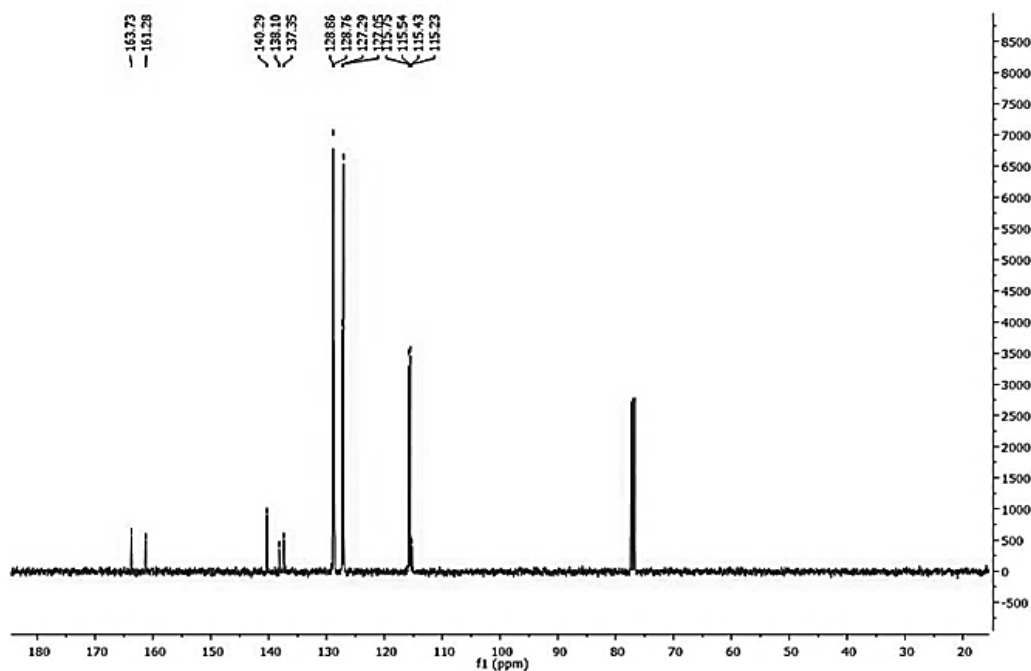


S36. The ^{13}C NMR of 4-carbonitril-[1,1'-biphenyl]

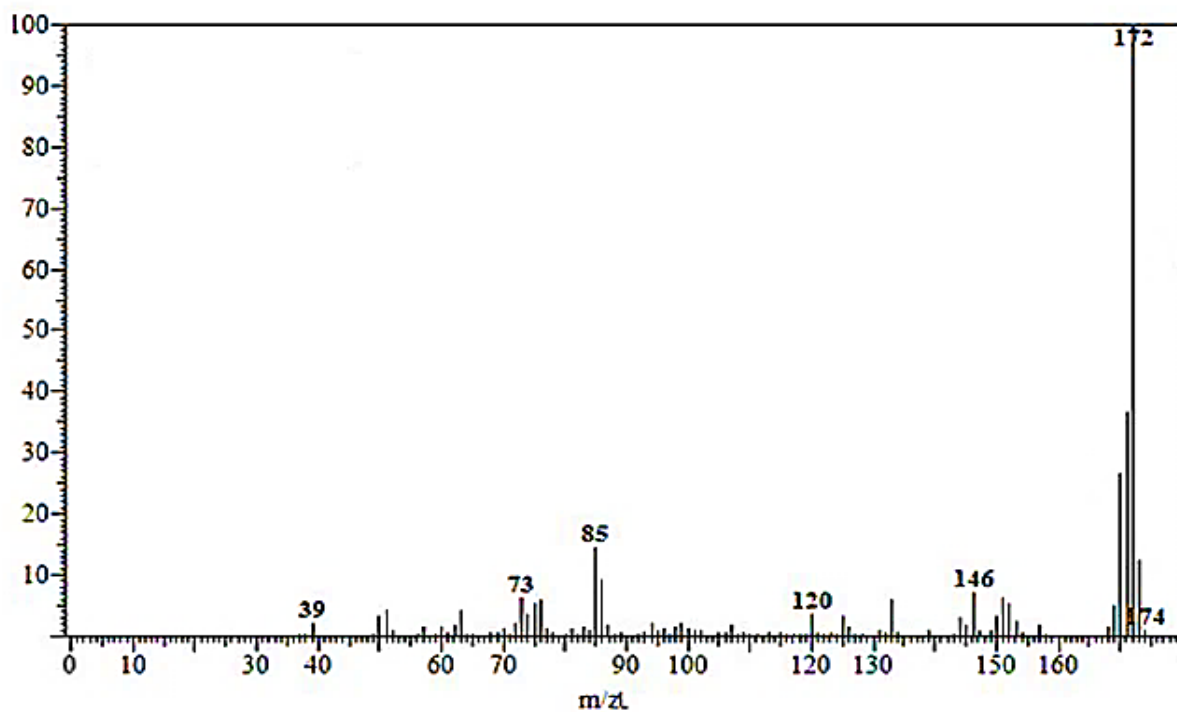
4-fluoro-[1,1'-biphenyl]



S37. The ¹H NMR of 4-fluoro-[1,1'-biphenyl]

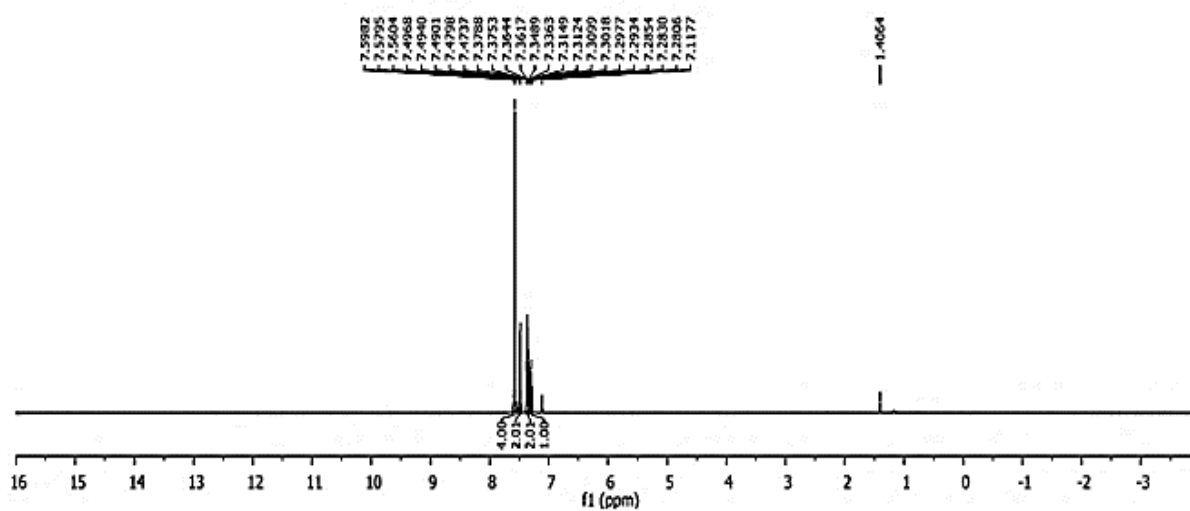
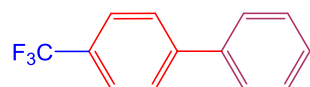


S38. The ¹³C NMR of 4-fluoro-[1,1'-biphenyl]

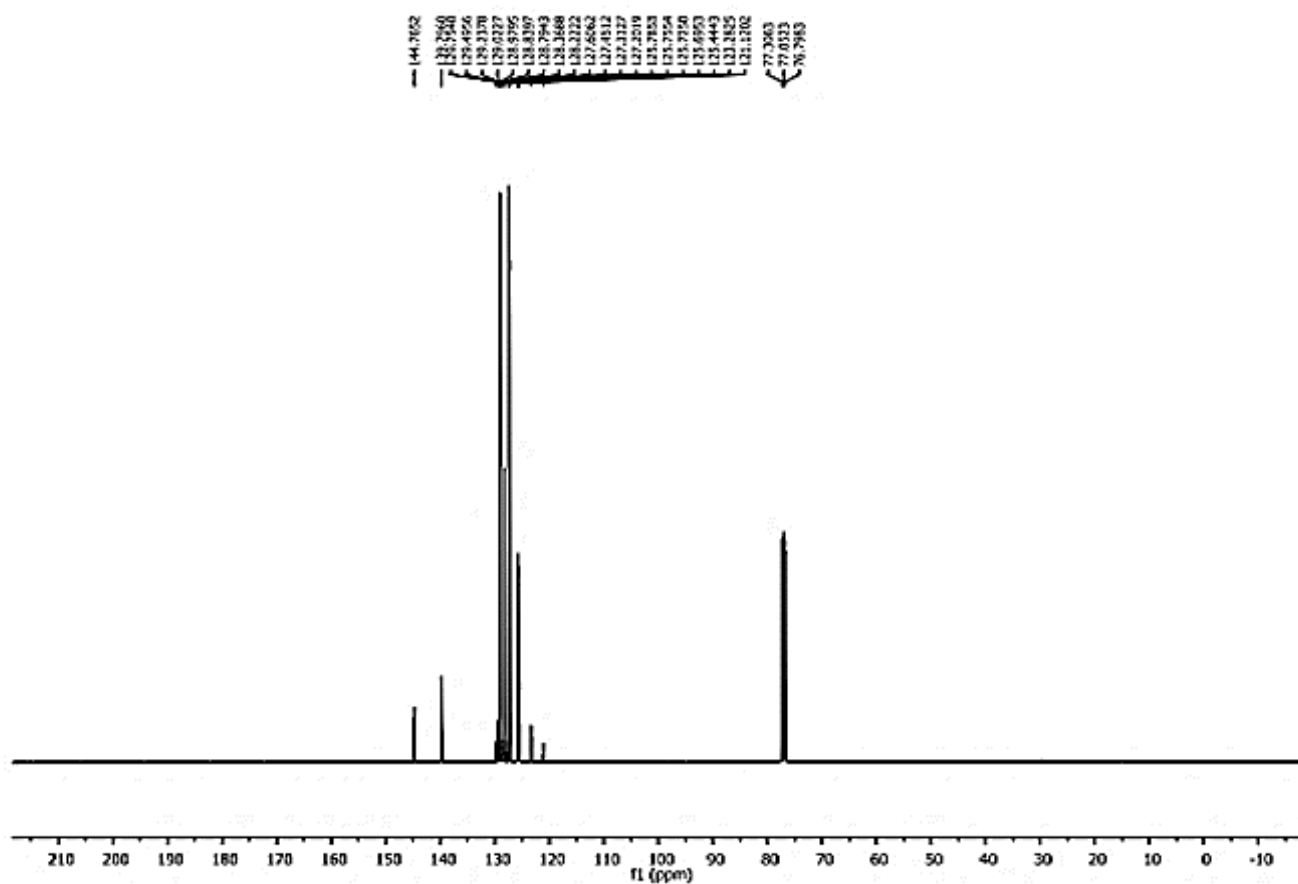
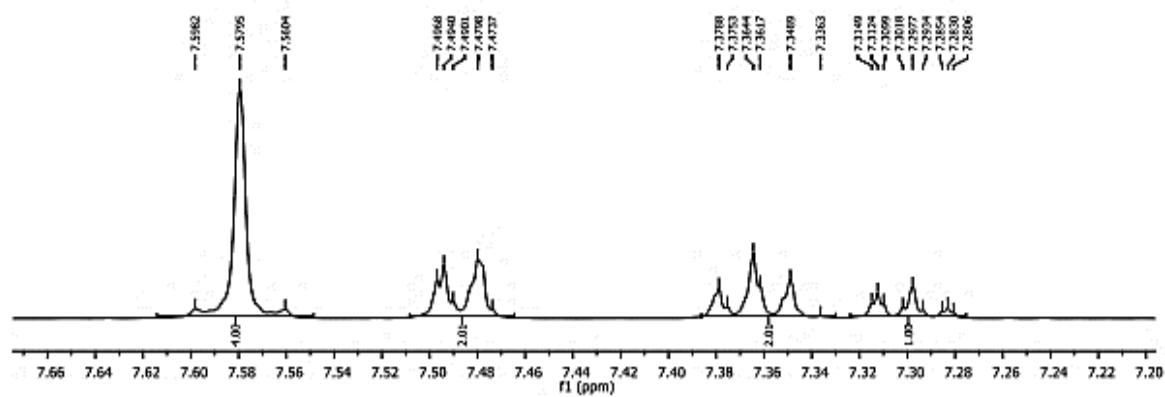


S39. The GC-Mass of 4-fluoro-[1,1'-biphenyl]

4-(trifluoromethyl)-1,1'-biphenyl

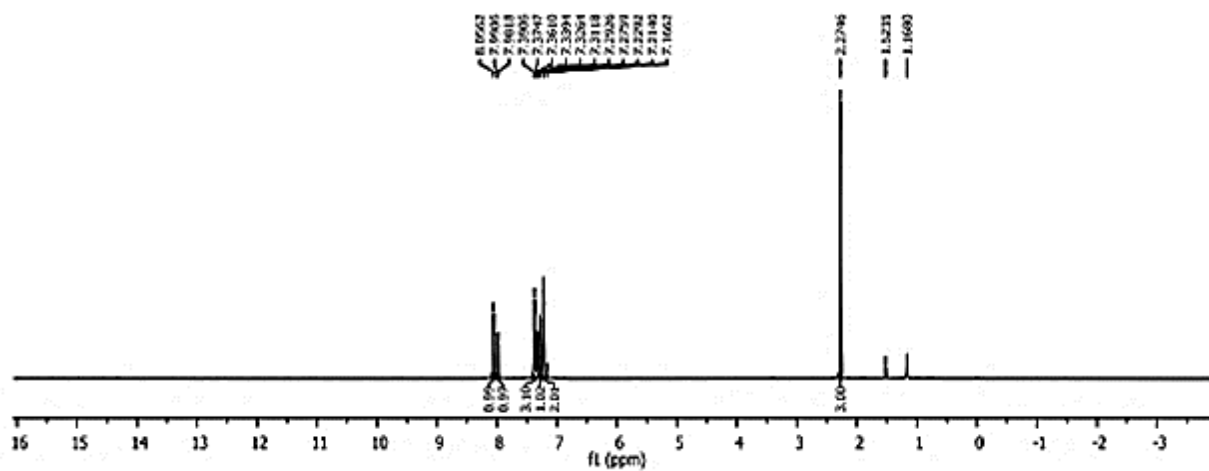
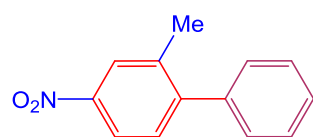


S40. The ^1H NMR of 4-(trifluoromethyl)-1,1'-biphenyl

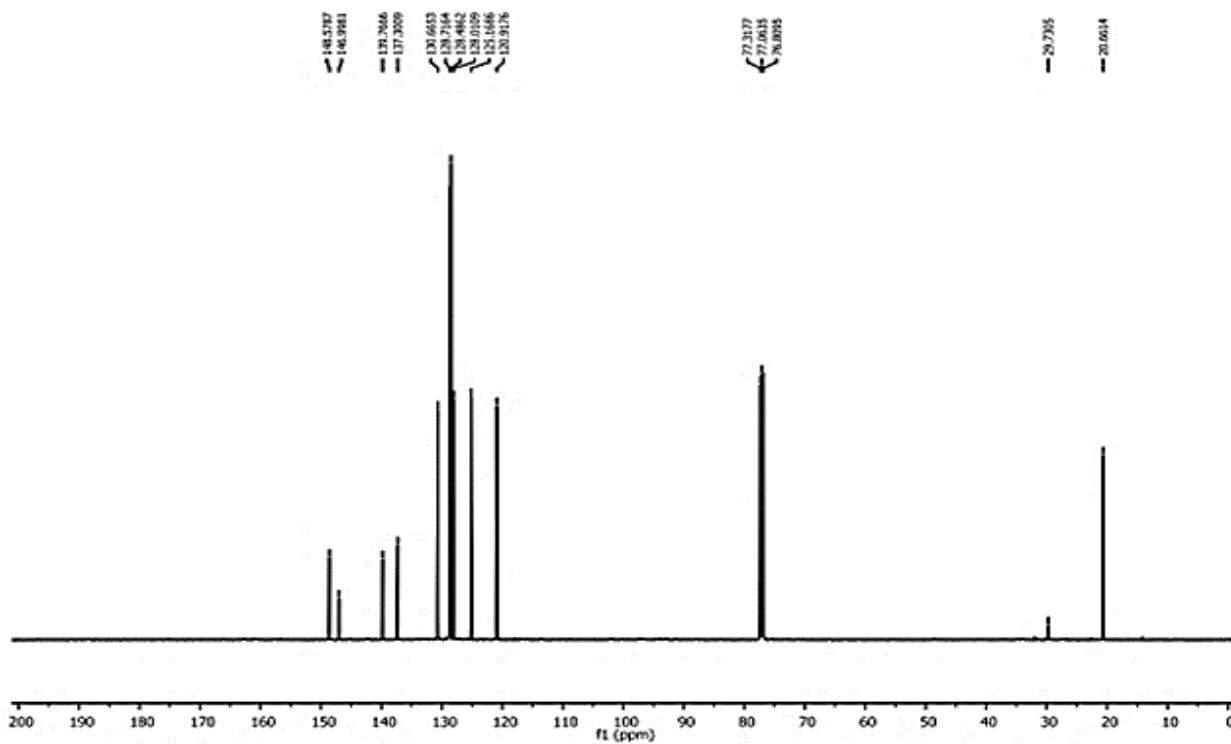


S41. The ^{13}C NMR of 4-(trifluoromethyl)-1,1'-biphenyl

2-methyl-4-nitro-1,1'-biphenyl



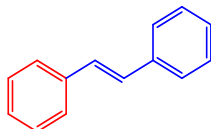
S42. The ^1H NMR of 2-methyl-4-nitro-1,1'-biphenyl



S43. The ^{13}C NMR of 2-methyl-4-nitro-1,1'-biphenyl

Heck derivatives

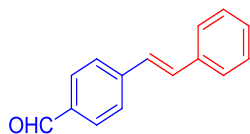
Trans-1, 2-diphenylethene (*E*-Stilbene)



White solid, Mp. 123-124°C (lit. 123.9-124.6°C)

GC-MS (m/z): 180; ^1H NMR (400 MHz, CDCl_3) δ 7.07-7.08 (s, 2H), 7.21-7.23 (m, 2H), 7.32- 7.34 (m, 4H), 7.46-7.48 (m, 2H) ppm. ^{13}C NMR (100 MHz, CDCl_3) δ 122.1, 123.2, 124.3, 132.9.

(*E*)-4-styrylbenzaldehyde

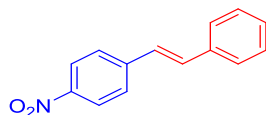


White solid, Mp. 116°C (lit. 116.0-116.8°C)

GC-MS (m/z): 208; ^1H NMR (400 MHz, CDCl_3): δ = 9.97 (s, 1H), 7.84 (d, J =8.0 Hz, 2H), 7.62 (d, J = 8.0 Hz, 2H), 7.52 (d, J = 7.6 Hz, 2H), 7.36 (t, J = 7.6 Hz, 2H), 7.28 (t, J = 7.6 Hz, 1H), 7.22 (d, J = 5.2 Hz, 1H), 7.11 (d, J =16.4 Hz, 1H).

^{13}C NMR (100 MHz, CDCl_3): δ = 191.5, 143.4, 136.5, 135.3, 132.2, 130.2, 128.8, 128.4, 127.3, 126.9. MS (EI) m/z: 208, 179, 165, 152, 102, 89, 76, 63, 51.

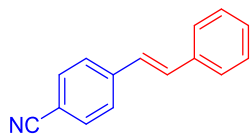
(*E*)-4-Nitrostilbene



Yellow solid, Mp. 154-155°C (lit. 154.2-155.0°C)

GC-MS (m/z): 225; ^1H NMR (400 MHz, CDCl_3) δ : 8.22 (d, J = 8.6 Hz, 2H, Ar-H), 7.63 (d, J = 8.7 Hz, 2H, Ar-H), 7.55 (d, J = 7.7 Hz, 2H, Ar-H), 7.40 (t, J = 7.6 Hz, 2H, Ar-H), 7.36-7.31 (m, 1H, Ar-H), 7.27 (d, J = 16.3 Hz, 1H, CH=), 7.14 (d, J = 16.3, 1H, Hz, CH=). ^{13}C NMR (100 MHz, CDCl_3), δ = 125.2, 127.4, 128.0, 128.2, 130.0, 134.4, 137.3, 145.0, 147.8 ppm.

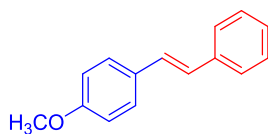
(E)-4-Styrylbenzonitrile



White solid, Mp. 114-115°C (lit. 114.9-115.2°C)

GC-MS (m/z): 205; ¹H-NMR (400 MHz, CDCl₃): δ 7.09 (d, 1H, *J* = 16.0 Hz), 7.22 (d, 1H, *J* = 16.0 Hz), 7.31-7.34 (t, 1H, *J* = 4.0 Hz), 7.38-7.41 (t, 2H, *J* = 4.0 Hz), 7.54 (d, 2H, *J* = 7.2 Hz), 7.58 (d, 2H, *J* = 8.4 Hz), 7.64 (d, 2H, *J* = 8.4 Hz) ppm; ¹³C-NMR (100 MHz, CDCl₃): δ 111.7, 120.2, 127.8, 128.0, 128.1, 129.8, 130.0, 133.5, 133.6, 137.4, 142.9 ppm.

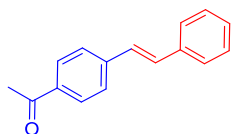
(E)-1-methoxy-4-styrylbenzene



White solid, Mp. 135-136°C (lit. 135.3-135.9°C)

GC-MS (m/z): 210; ¹H-NMR (400 MHz, CDCl₃) δ (ppm): 3.70 (3H, s), 6.78 (1H, d, *J* = 8.75 Hz), 6.90 (1H, d, *J* = 10.75 Hz), 7.00-7.38 (9H, m). ¹³C-NMR (100 MHz, CDCl₃) δ (ppm): 55.3, 114.2, 126.3, 126.6, 127.25, 127.8, 128.2, 128.7, 130.2, 137.7, 159.3.

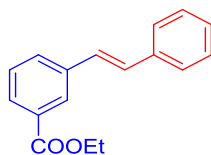
(E)-1-(4-styrylphenyl) ethanone



White solid, Mp. 141-142°C (lit. 141.3-142°C)

GC-MS (m/z): 222; ¹H-NMR (400 MHz, CDCl₃): δ 7.97-7.95 (d, 2H), 7.58-7.57 (d, 2H), 7.53-7.52 (d, 2H), 7.40-7.36 (t, 2H), 7.29-7.25 (t, 1H), 7.22-7.20 (d, 1H), 7.12 (d, 1H), 2.64 (s, 3H), ¹³C-NMR (100 MHz, CDCl₃): δ 197.6, 142.1, 136.7, 136.1, 131.5, 128.9, 128.8, 128.4, 127.6, 126.8, 126.5, 26.8.

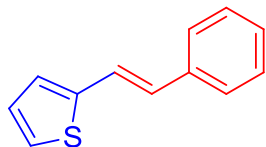
(E)-Ethyl-3-styrylbenzoate



Oily liquid

GC–MS (m/z): 252; ^1H NMR (CDCl_3 , 400 MHz): δ (ppm) 8.22 (s, 1H), 7.95 (d, $J = 7.7$ Hz, 1H), 7.70 (d, $J = 7.7$ Hz, 1H), 7.55 (d, $J = 7.4$ Hz, 2H), 7.46–7.27 (m, 4H), 7.21 (d, $J = 16.0$ Hz, 1H), 7.15 (d, $J = 16.0$ Hz, 1H), 4.43 (q, $J = 7.1$ Hz, 3H), 1.44 (t, $J = 7.1$ Hz); ^{13}C NMR (CDCl_3 , 100 MHz): δ (ppm) 166.6, 137.7, 137.0, 131.0, 130.7, 129.9, 128.8, 128.7, 128.5, 128.0, 127.6, 127.5, 126.7.

(*E*)-2-styrylthiophene

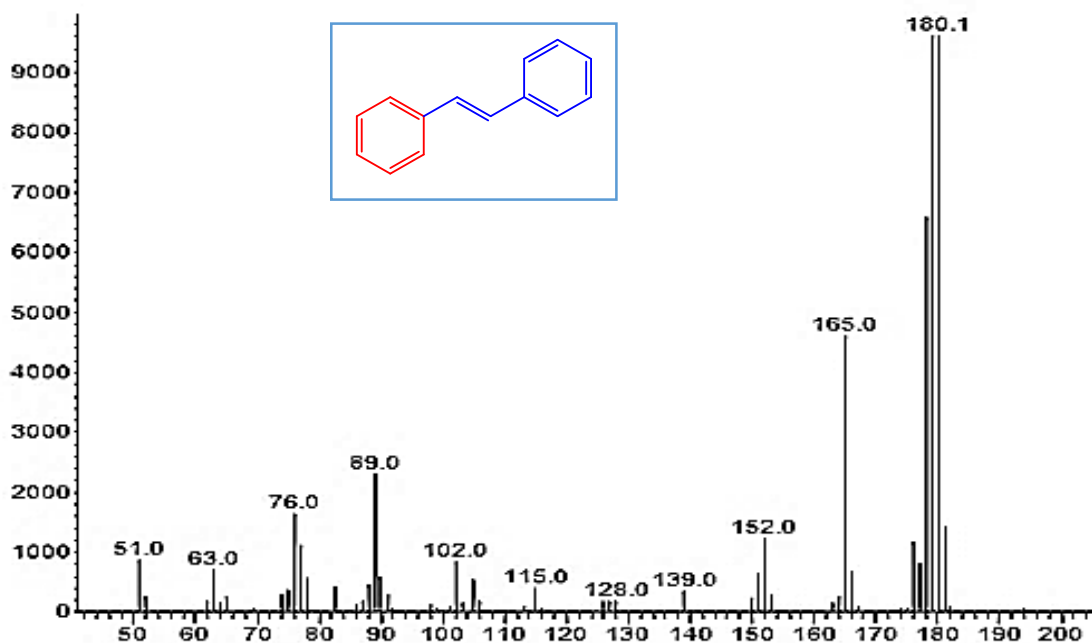


white solid, MP. 111–112°C (lit. 112°C)

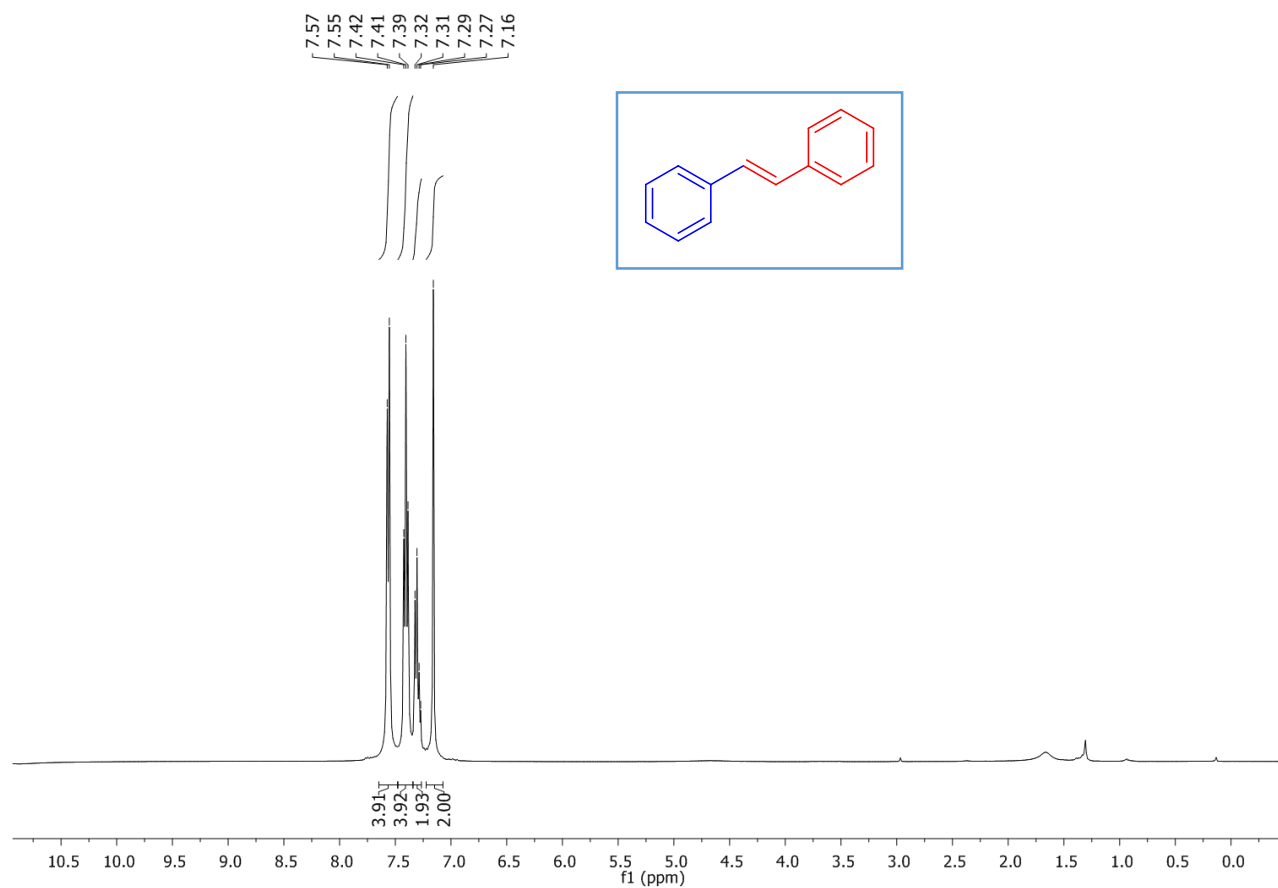
GC–MS (m/z): 186; ^1H NMR (CDCl_3 , 400 MHz): δ (ppm) 7.49 (d, $J = 7.5$ Hz, 2H), 7.37 (t, $J = 7.5$ Hz, 2H), 7.28 (d, $J = 5.1$ Hz, 1H), 7.24 (d, $J = 16$ Hz, 1H), 7.21 (d, $J = 4.9$ Hz, 1H), 7.09 (d, $J = 3.4$ Hz, 1H), 7.03 (t, $J = 4$ Hz, 1H), 6.95 (d, $J = 16.2$ Hz, 1H); ^{13}C NMR (CDCl_3 , 100 MHz): δ (ppm) 142.9, 137.0, 128.7, 128.4, 127.6, 126.3, 126.1, 124.3, 121.8.

GC-MS and NMR Spectra for Heck Coupling Reaction

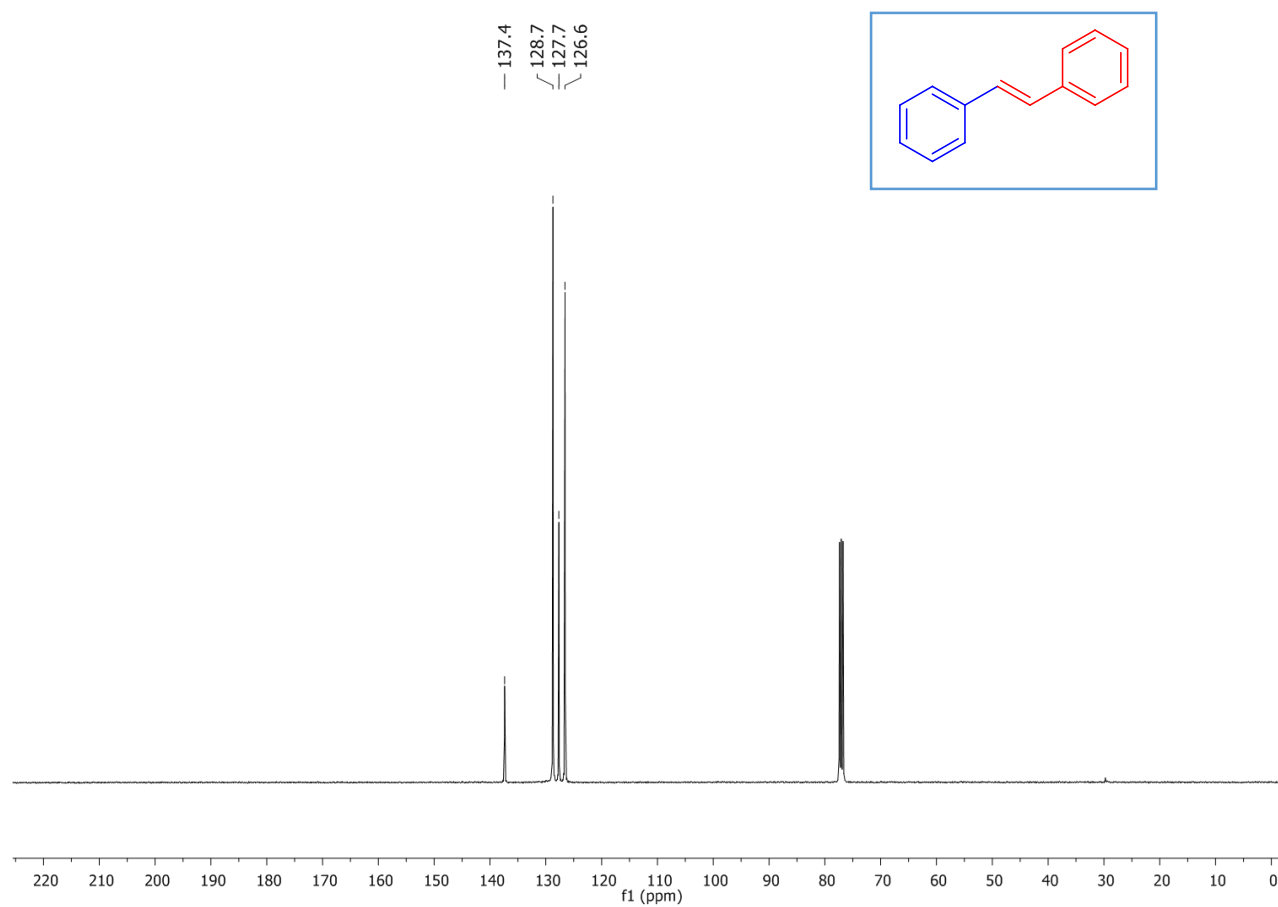
1, 2-diphenylethene



S44. GC-Mass of 1, 2-diphenylethene

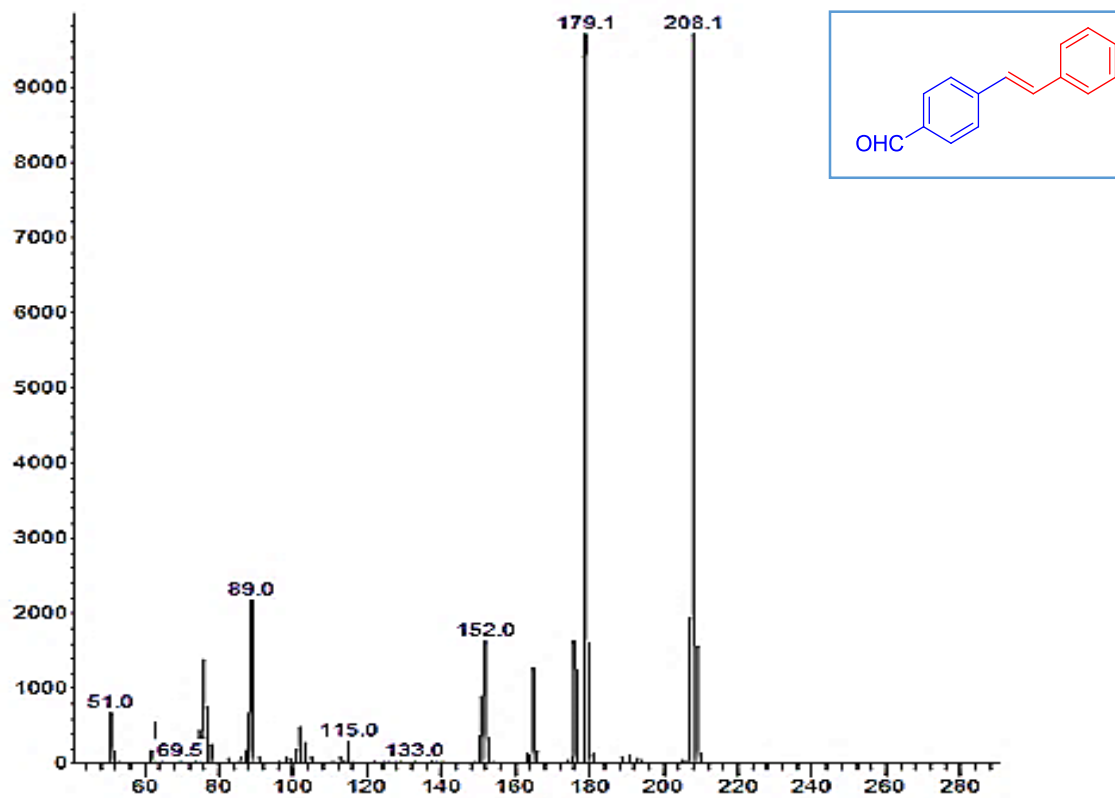


S45. ^1H NMR of 1, 2-diphenylethene

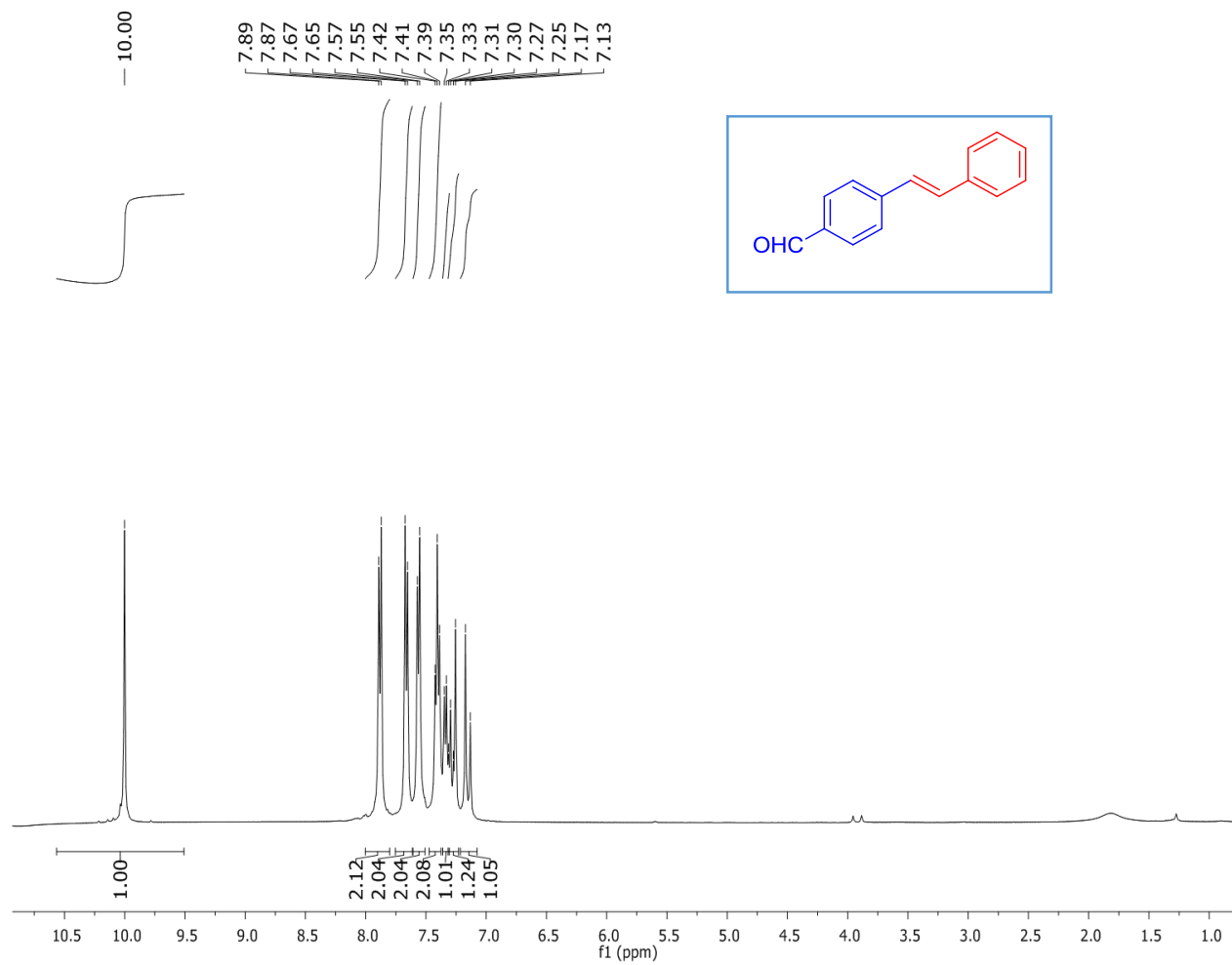


S.46. ^{13}C NMR of 1, 2-diphenylethene

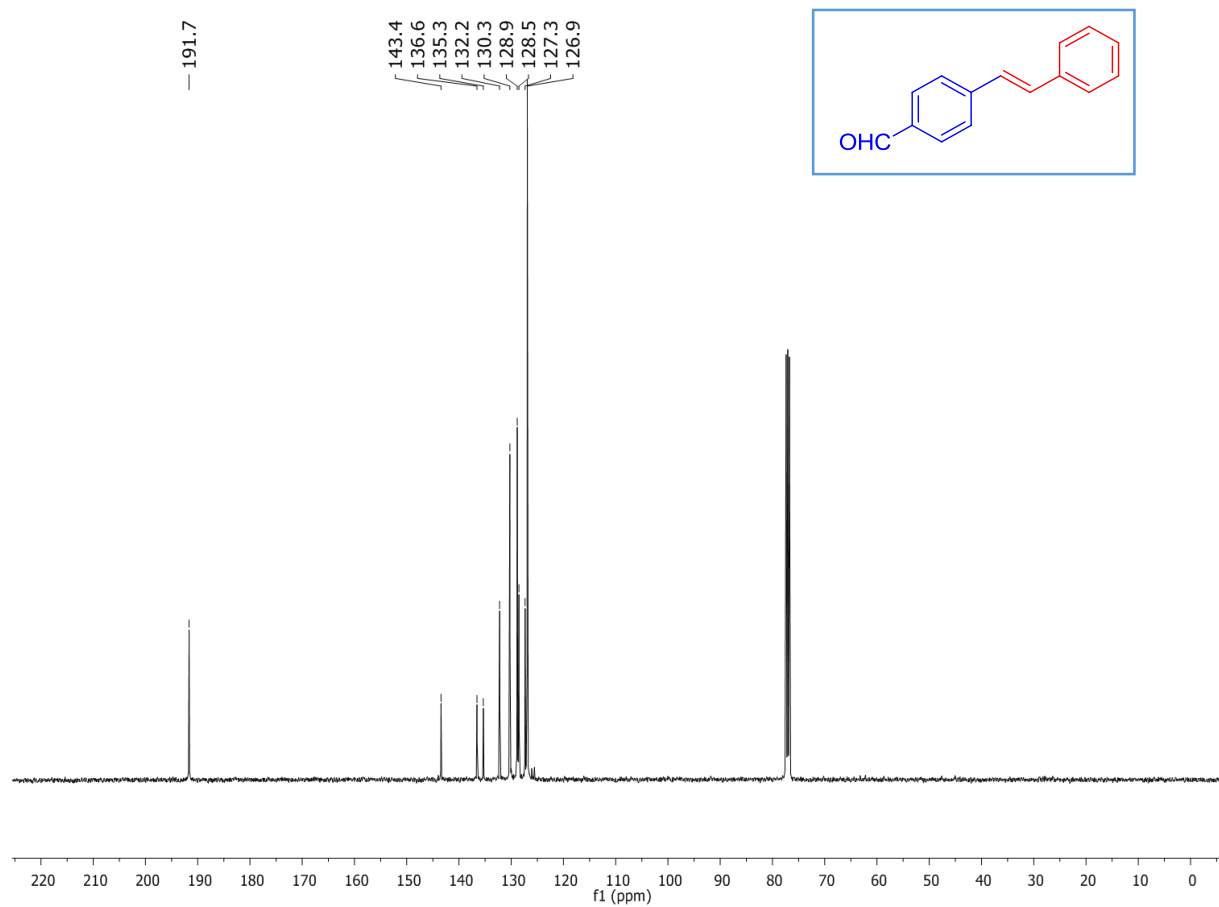
(E)-4-styrylbenzaldehyde



S47. GC-Mass of (E)-4-styrylbenzaldehyde

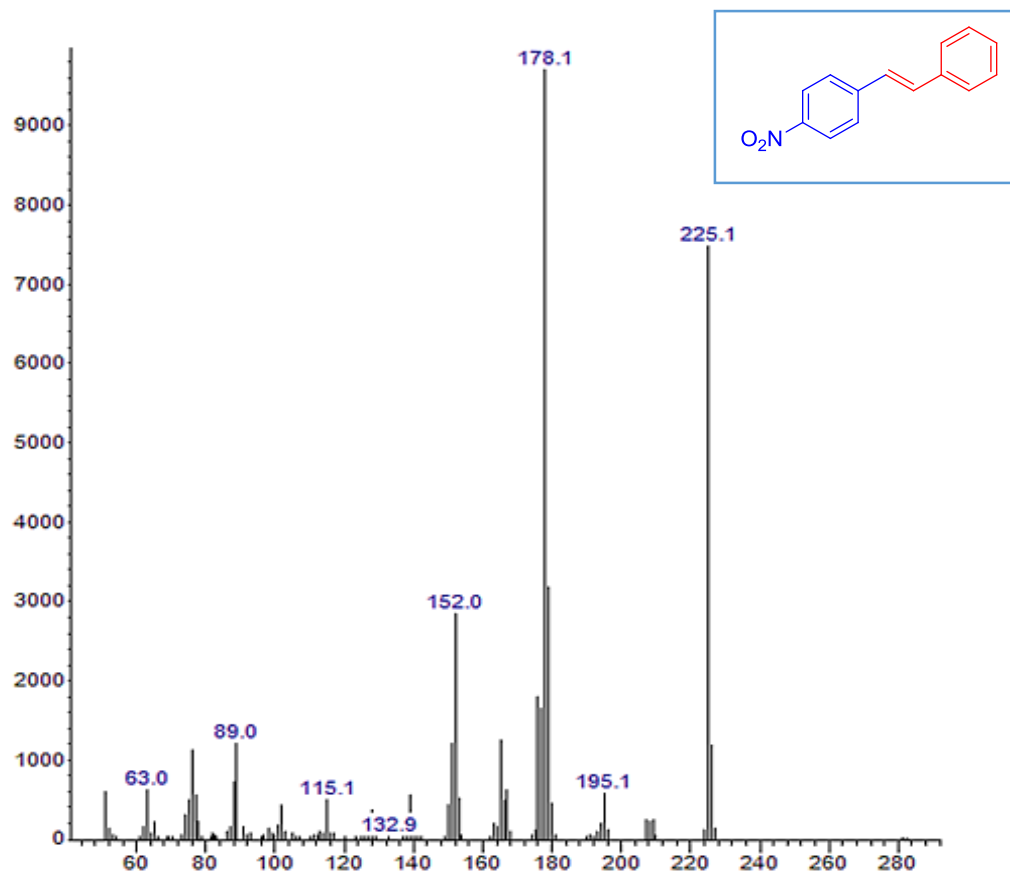


S48. ¹H NMR of (*E*)-4-styrylbenzaldehyde

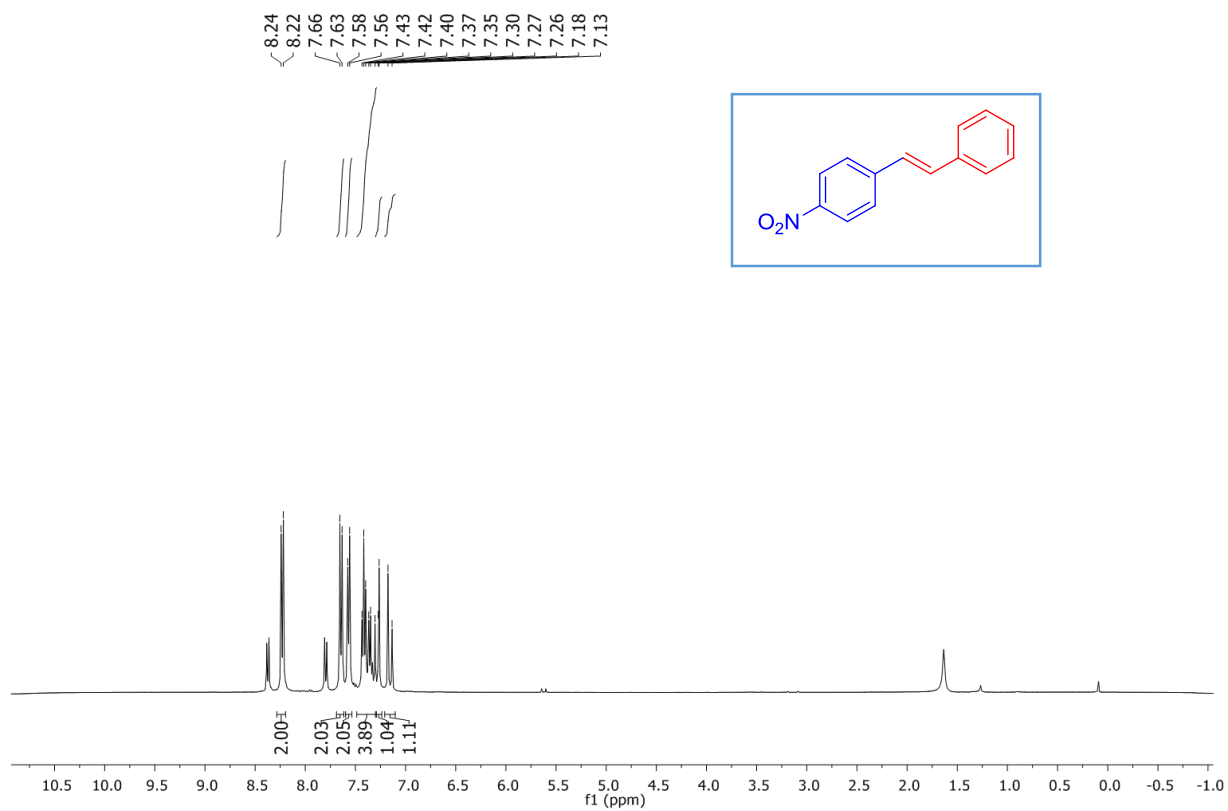


S49. ^{13}C NMR of (*E*)-4-styrylbenzaldehyde

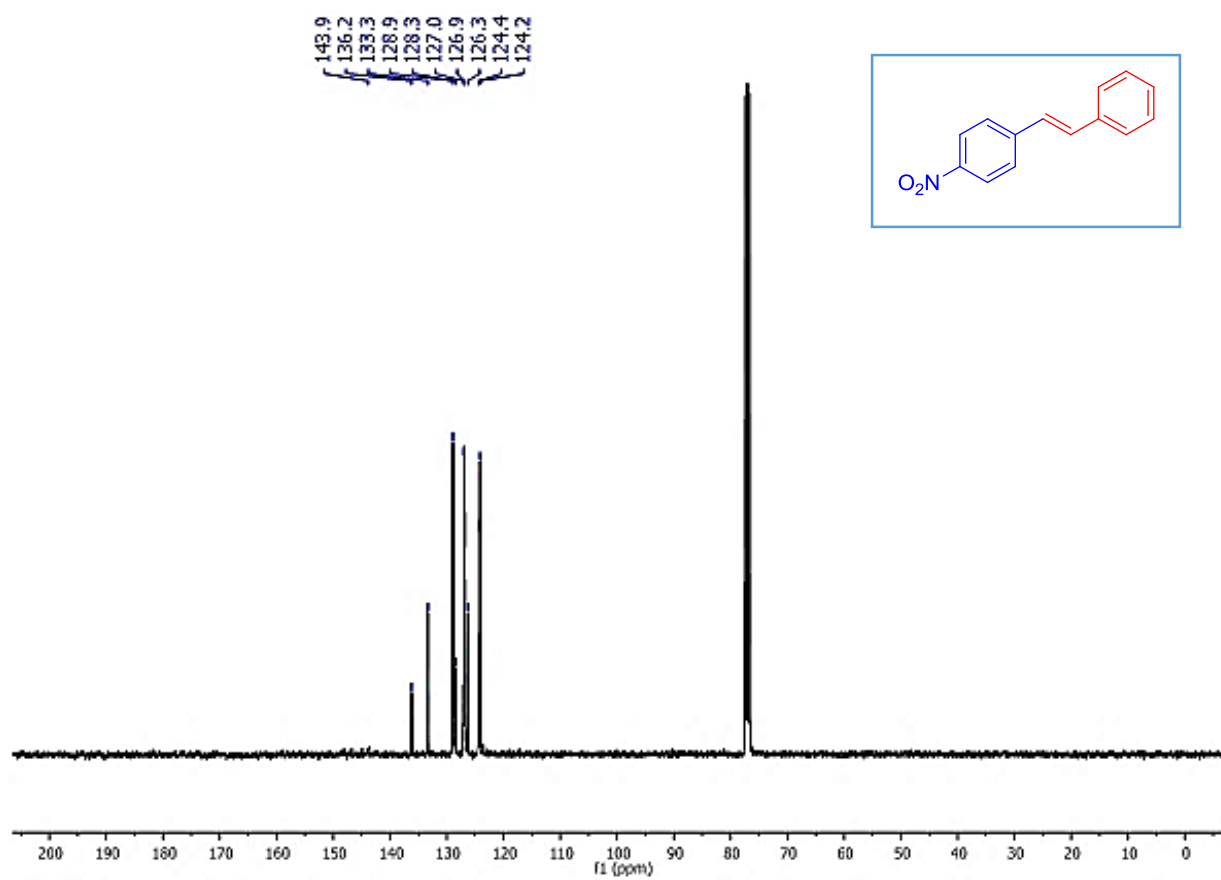
(E)-1-nitro-4-styrylbenzene



S50. GC-Mass of (E)-1-nitro-4-styrylbenzene

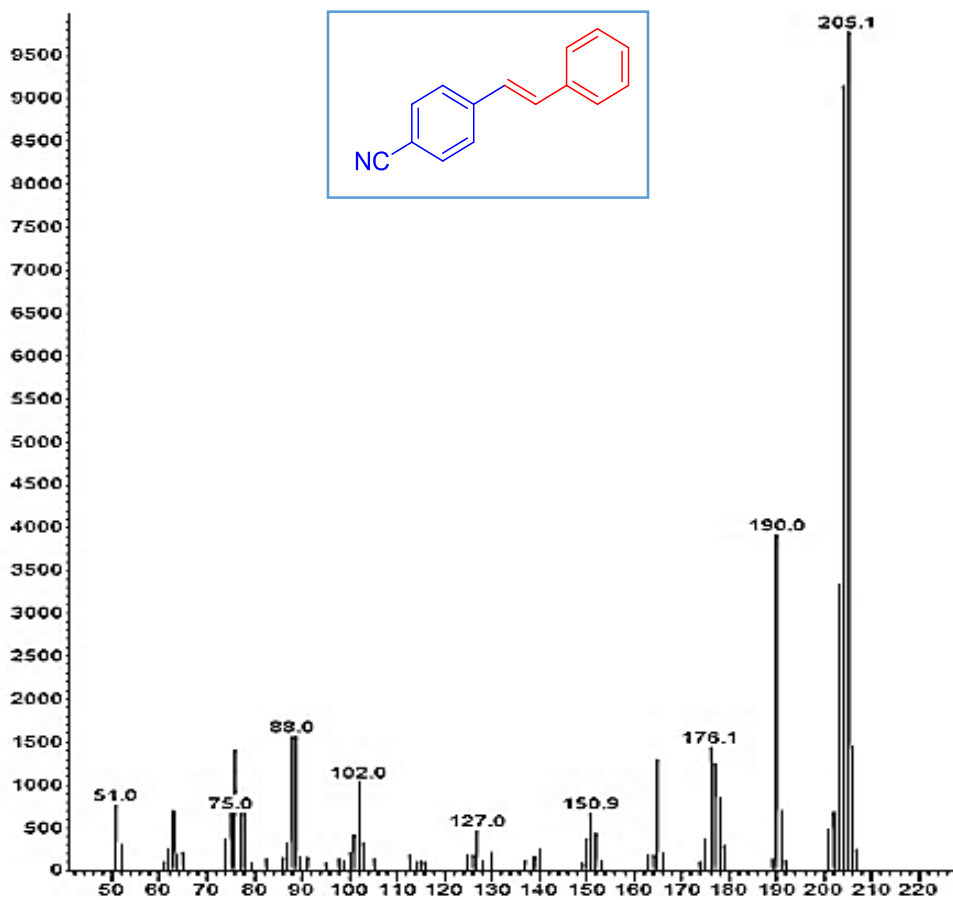


S51. ¹HNMR for (E)-1-nitro-4-styrylbenzene

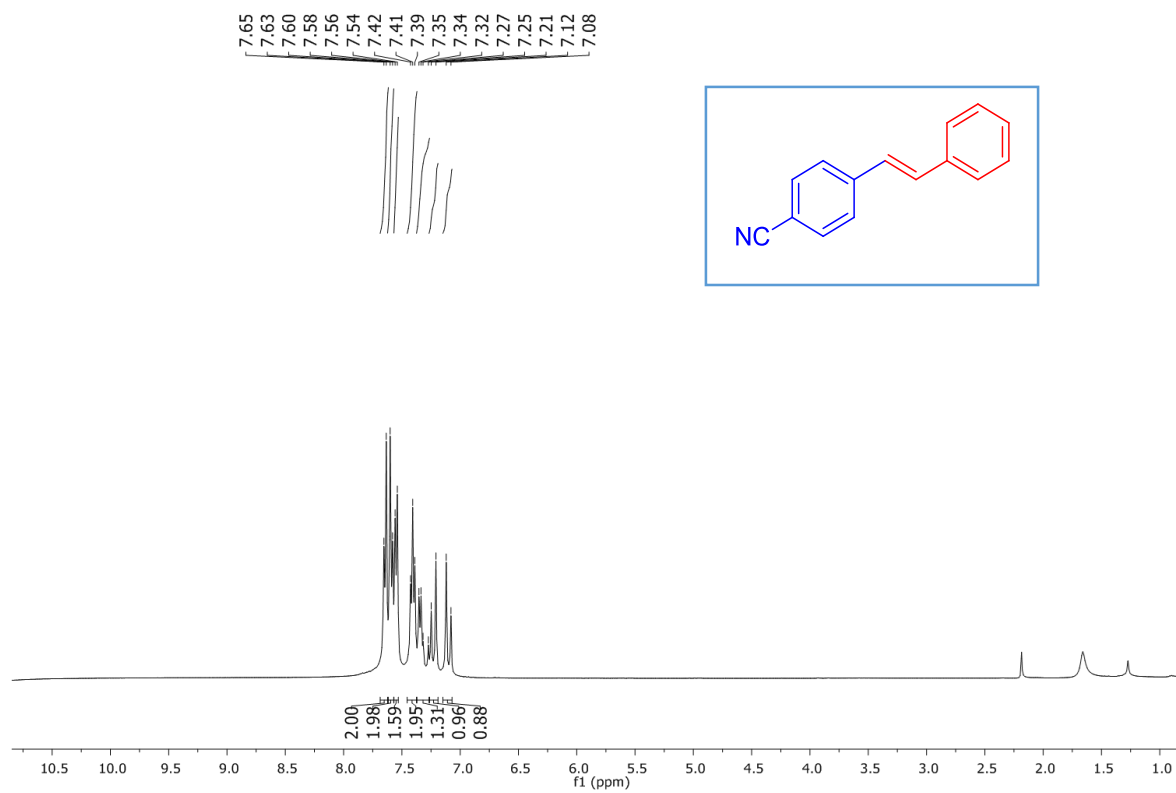


S52. ¹³CNMR for (*E*)-1-nitro-4-styrylbenzene

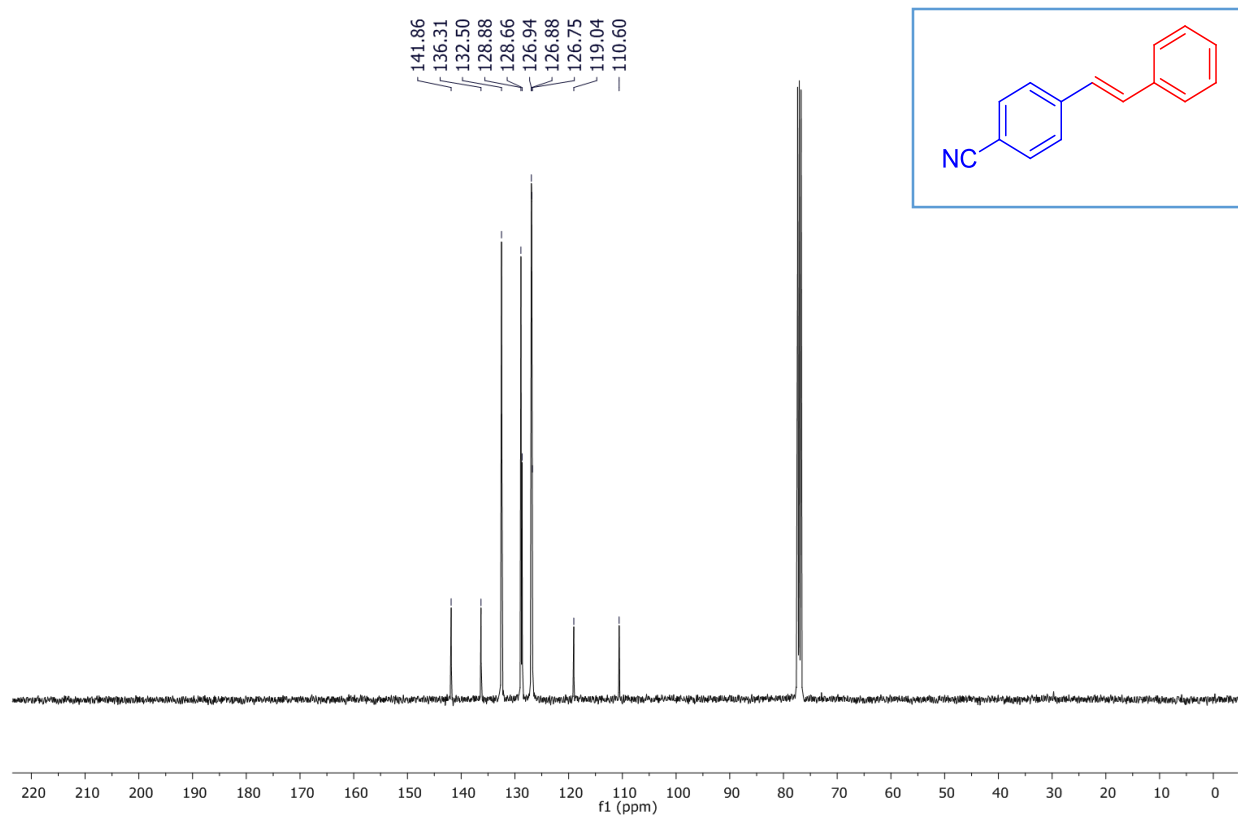
(E)-4-styrylbenzonitrile



S53. GC-Mass for *(E)*-4-styrylbenzonitrile

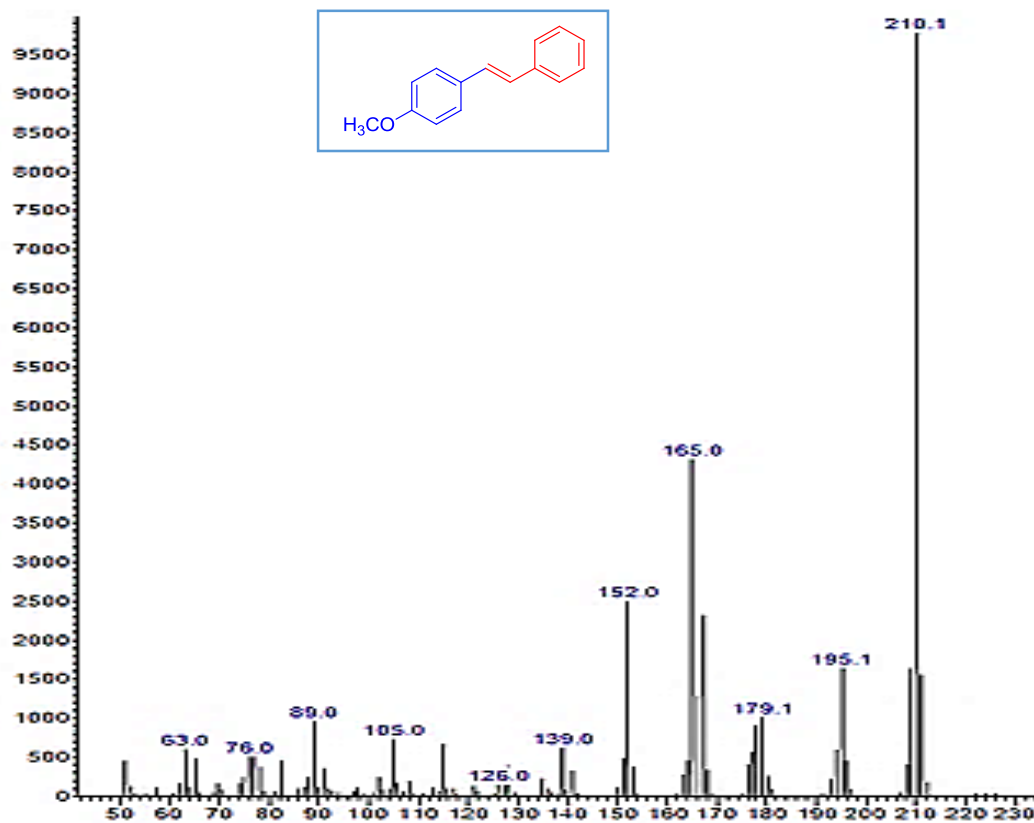


S54. ¹HNMR for (*E*)-4-styrylbenzonitrile

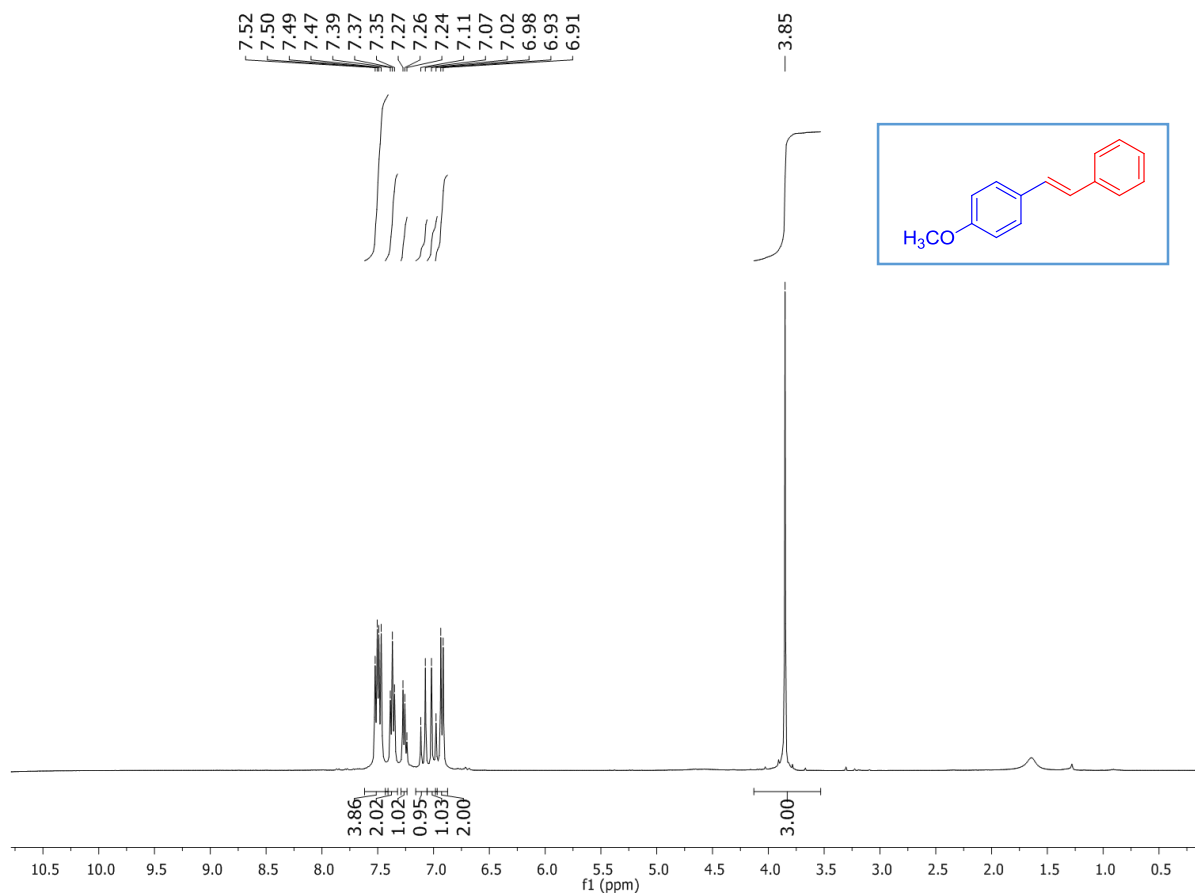


S55. ^{13}C NMR for (*E*)-4-styrylbenzonitrile

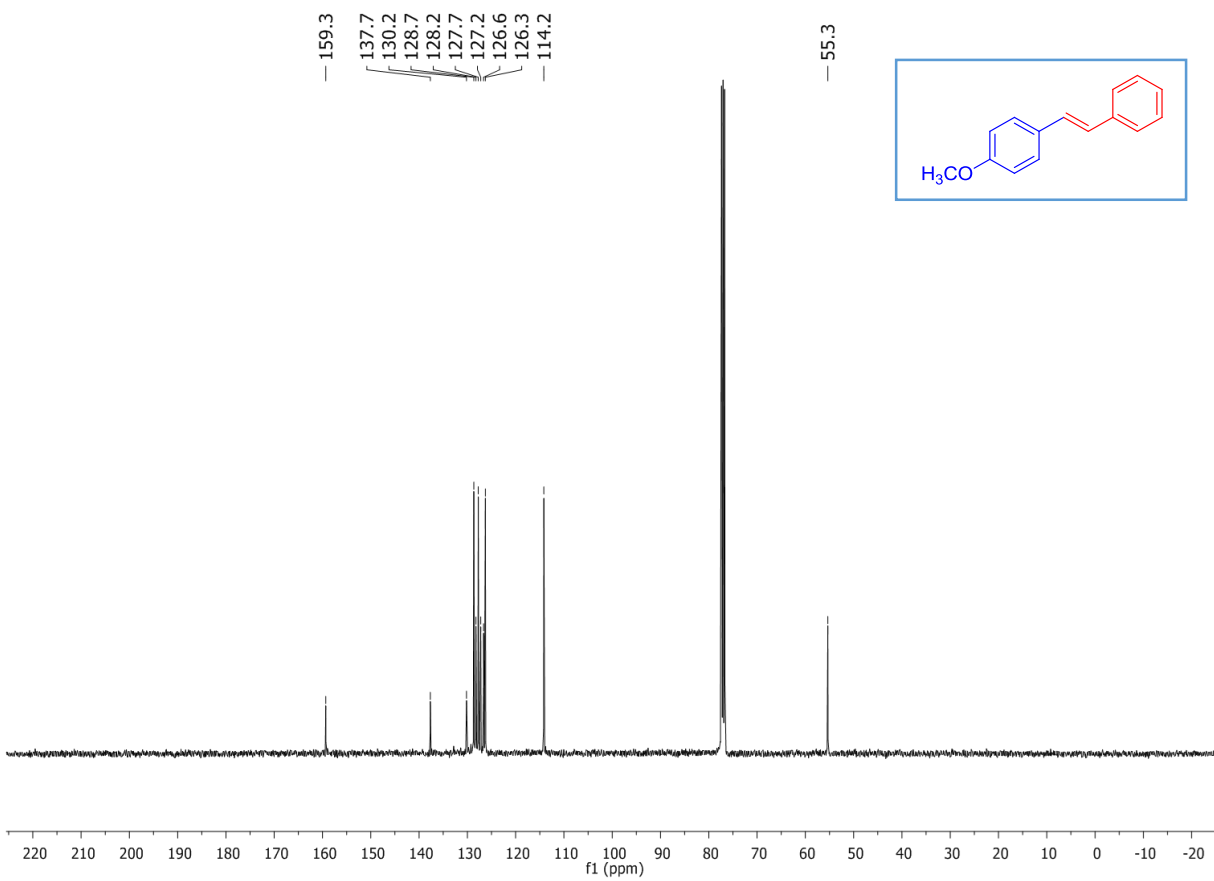
(E)-1-methoxy-4-styrylbenzene



S56. GC-Mass for (E)-1-methoxy-4-styrylbenzene

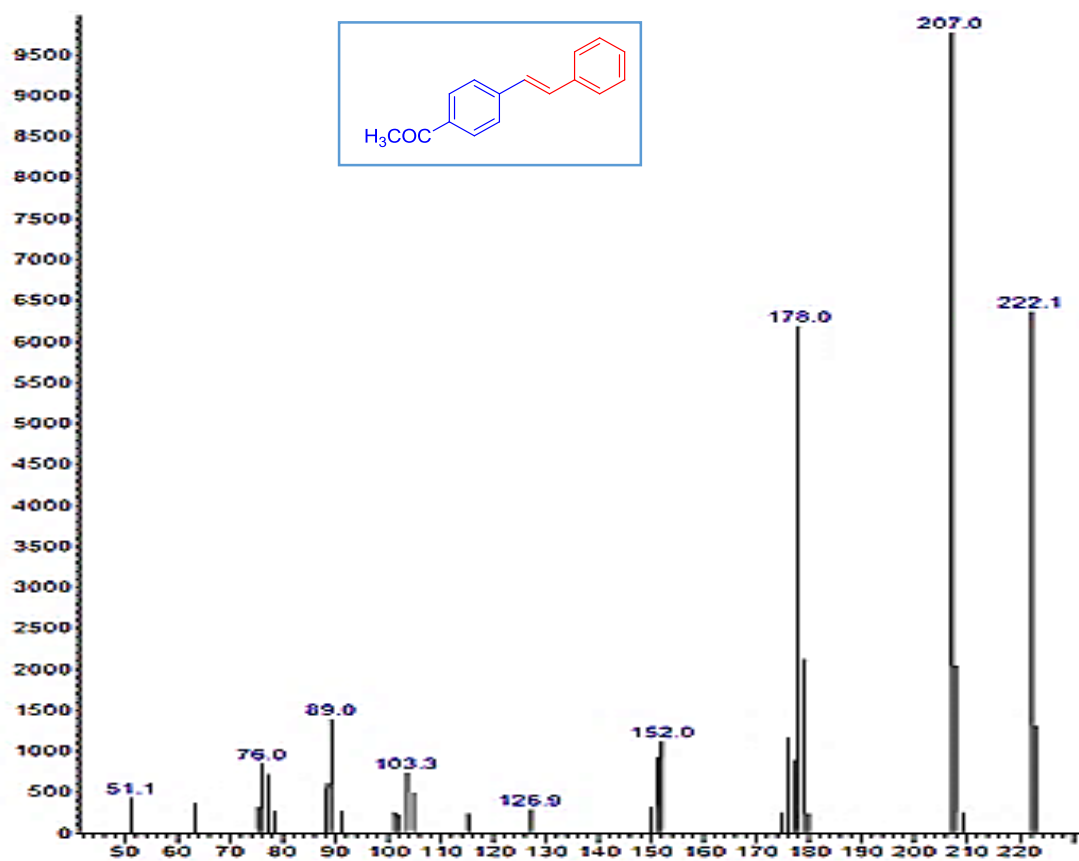


S57. ¹HNMR for (*E*)-1-methoxy-4-styrylbenzene

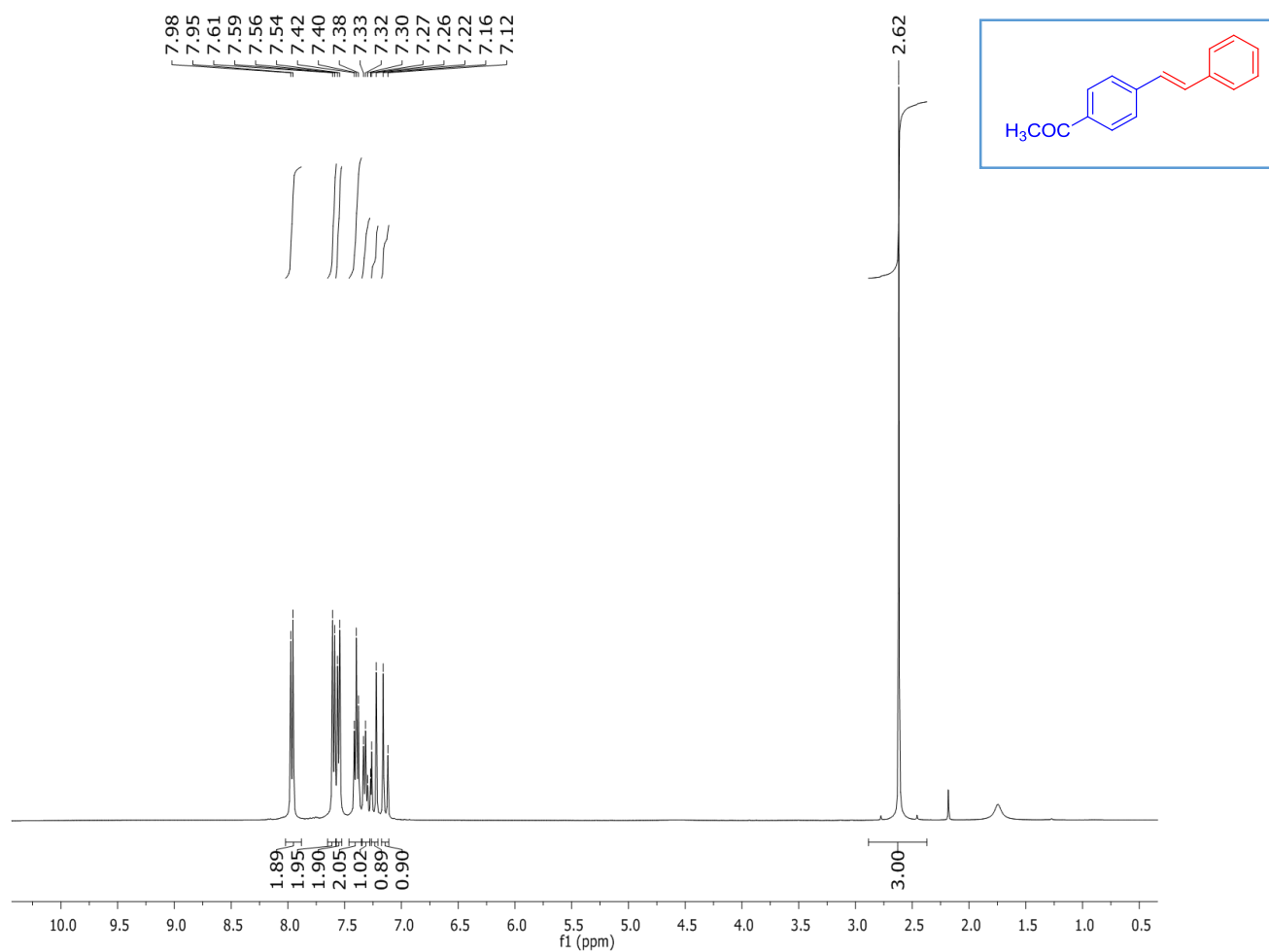


S58. ^{13}C NMR for (E)-1-methoxy-4-styrylbenzene

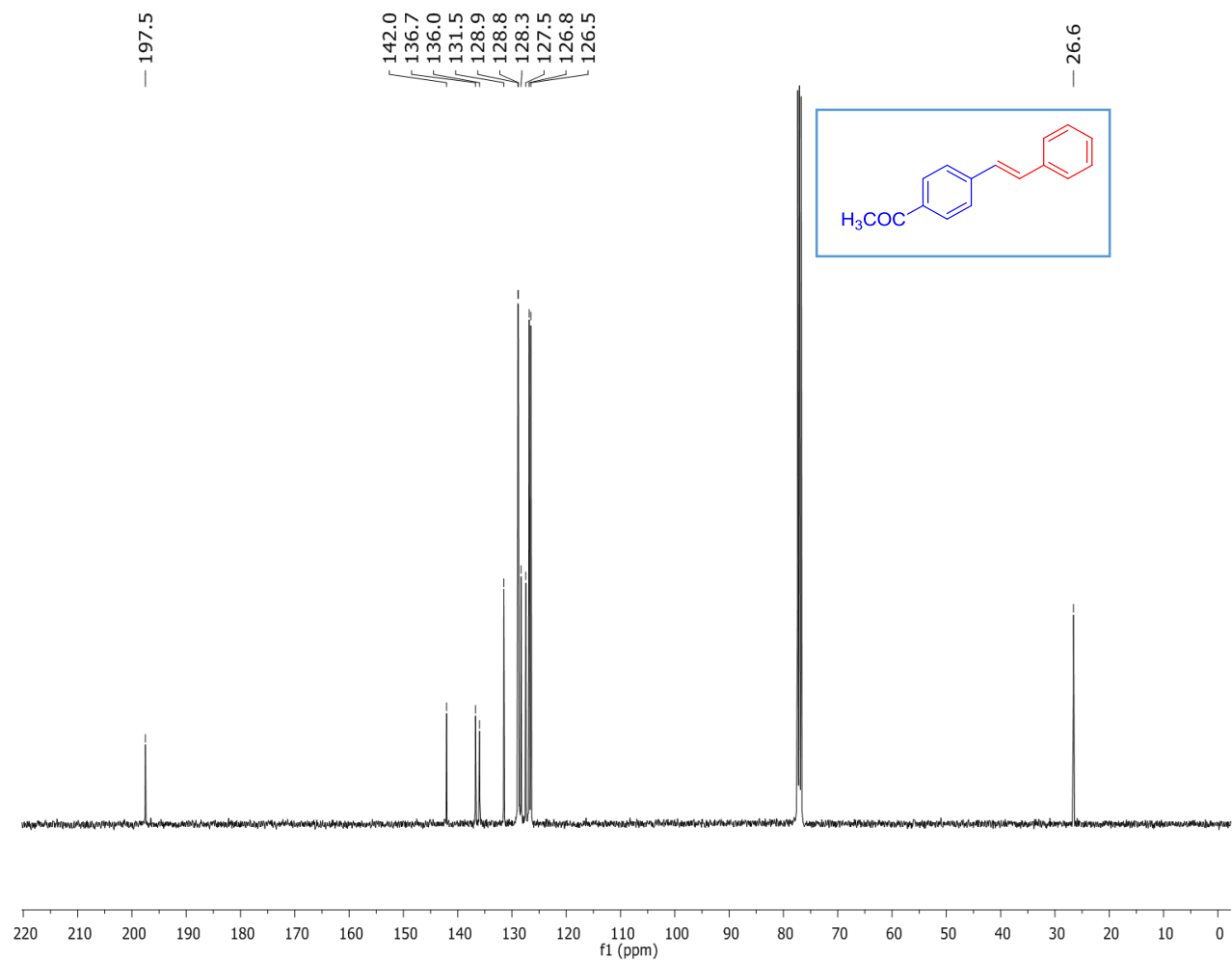
(E)-1-(4-styrylphenyl)ethan-1-one



S59. GC-Mass for *(E)*-1-methoxy-4-styrylbenzene

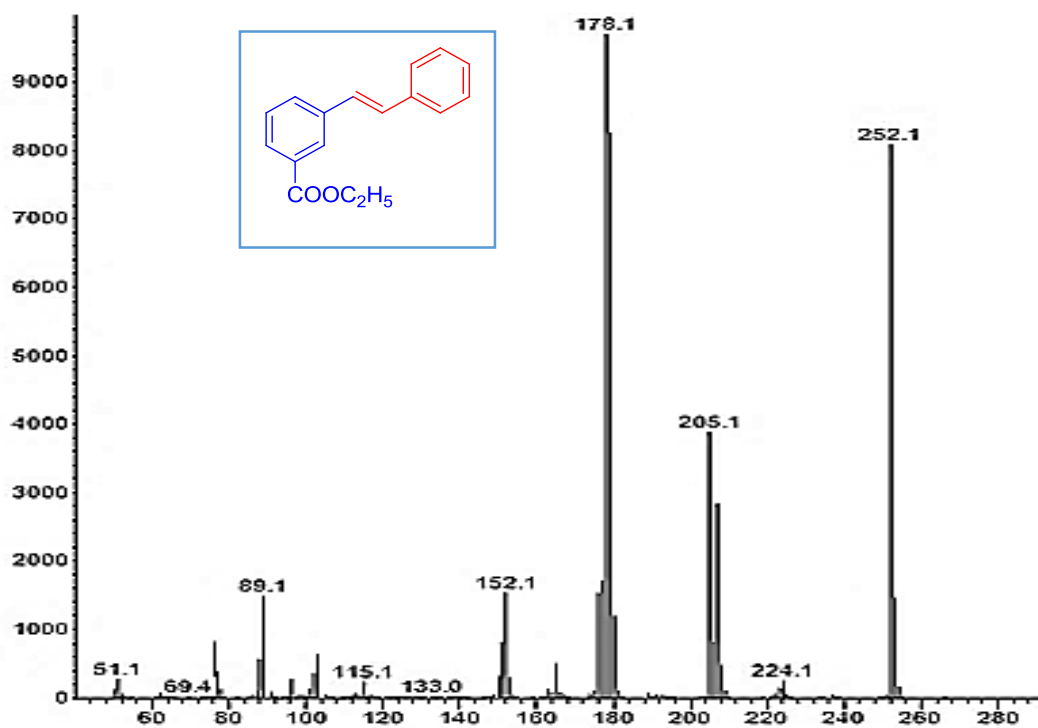


S60. ¹H NMR for (*E*)-1-(4-styrylphenyl)ethan-1-one

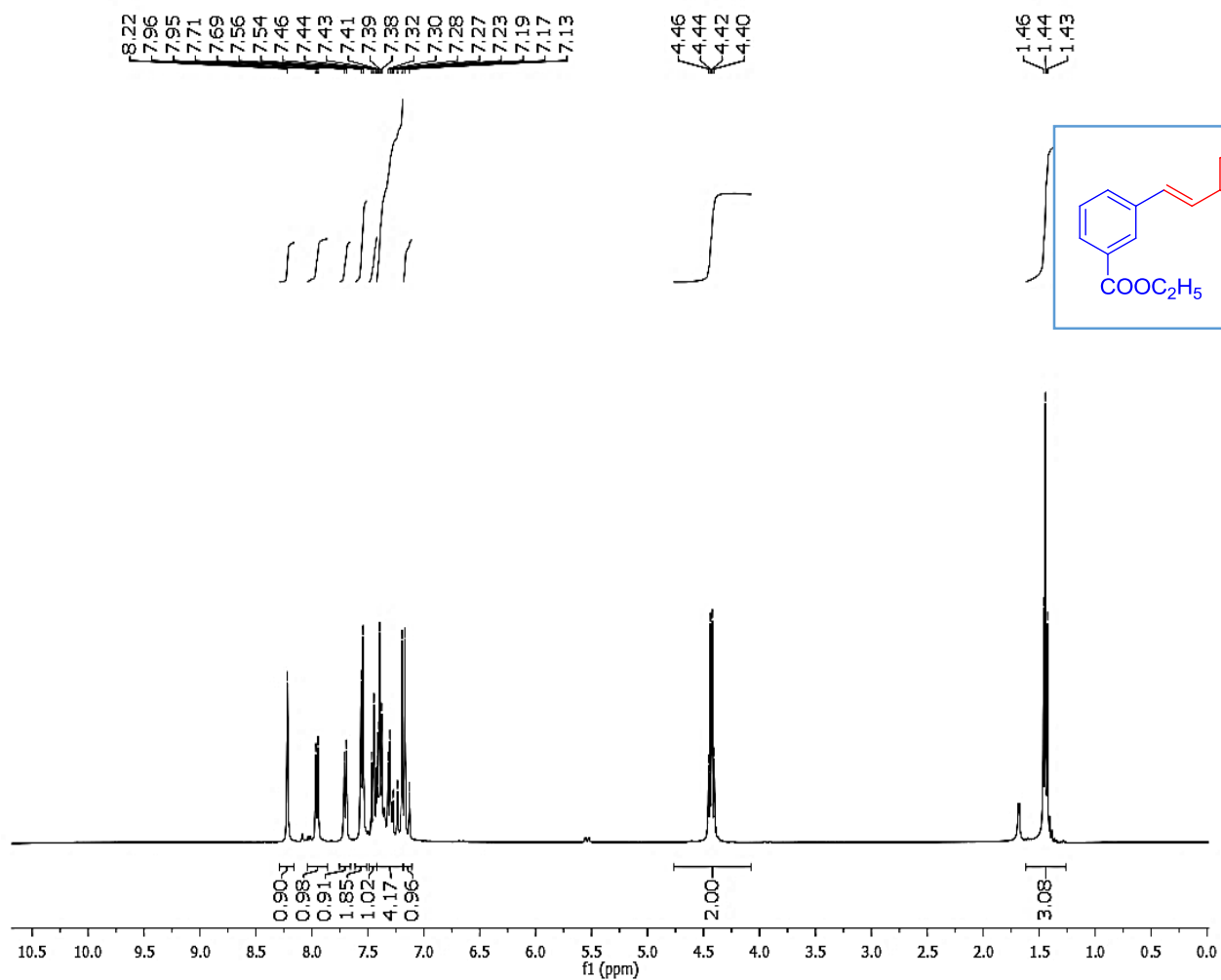


S61. ¹³CNMR for (E)-1-(4-styrylphenyl)ethan-1-one

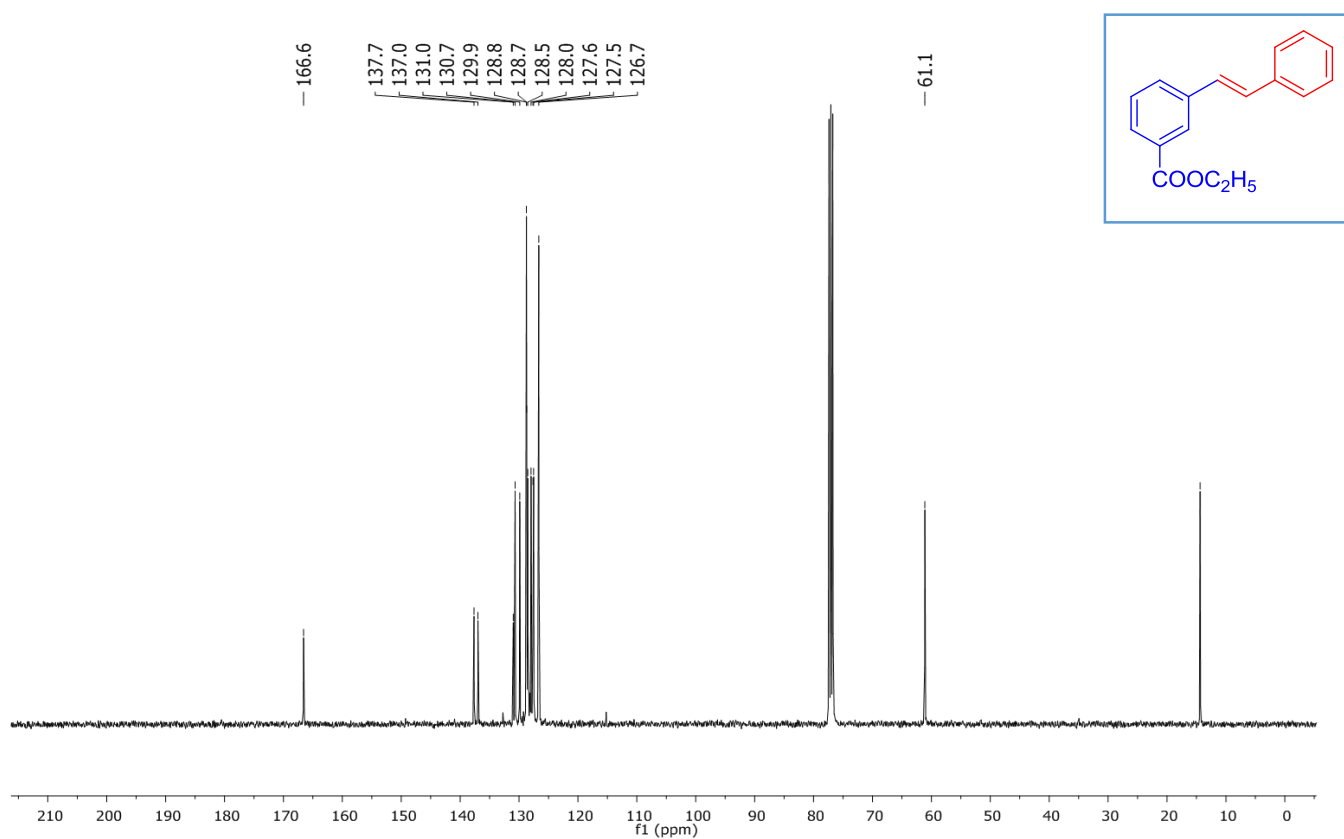
Ethyl (*E*)-3-styrylbenzoate



S62. GC-Mass for Ethyl (*E*)-3-styrylbenzoate

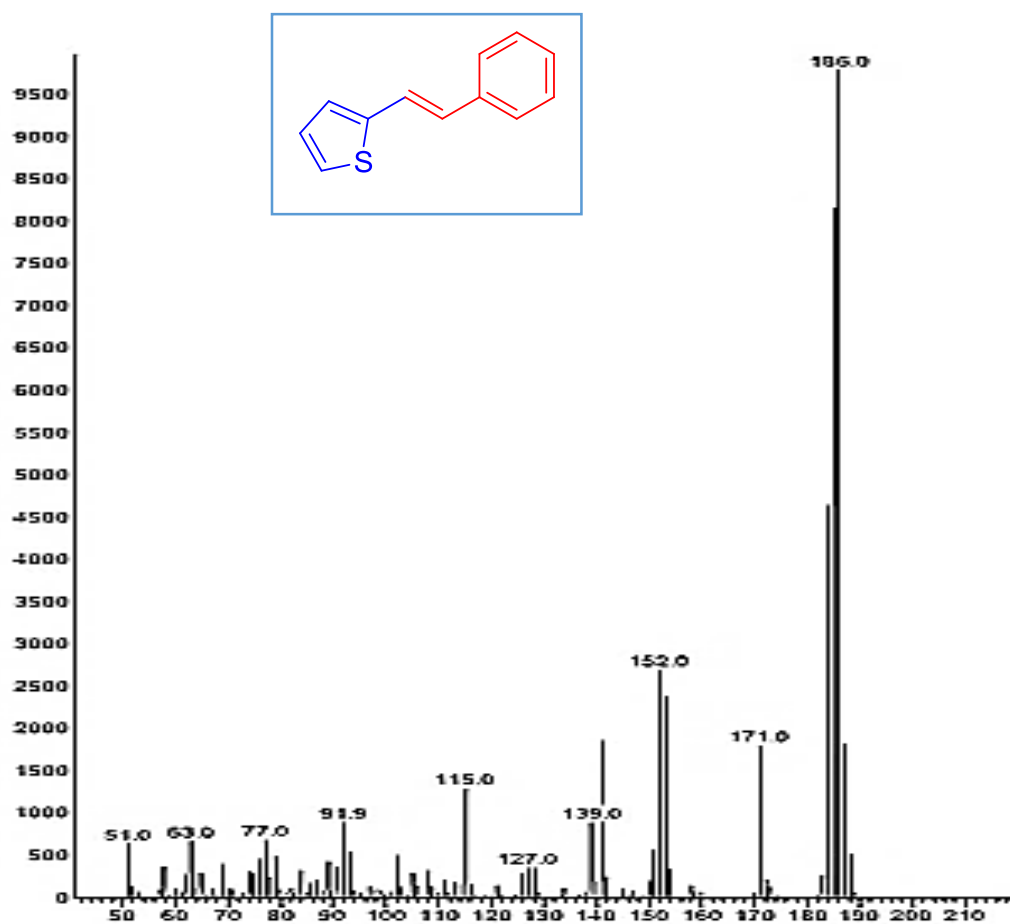


S63. ^1H NMR for Ethyl (*E*)-3-styrylbenzoate

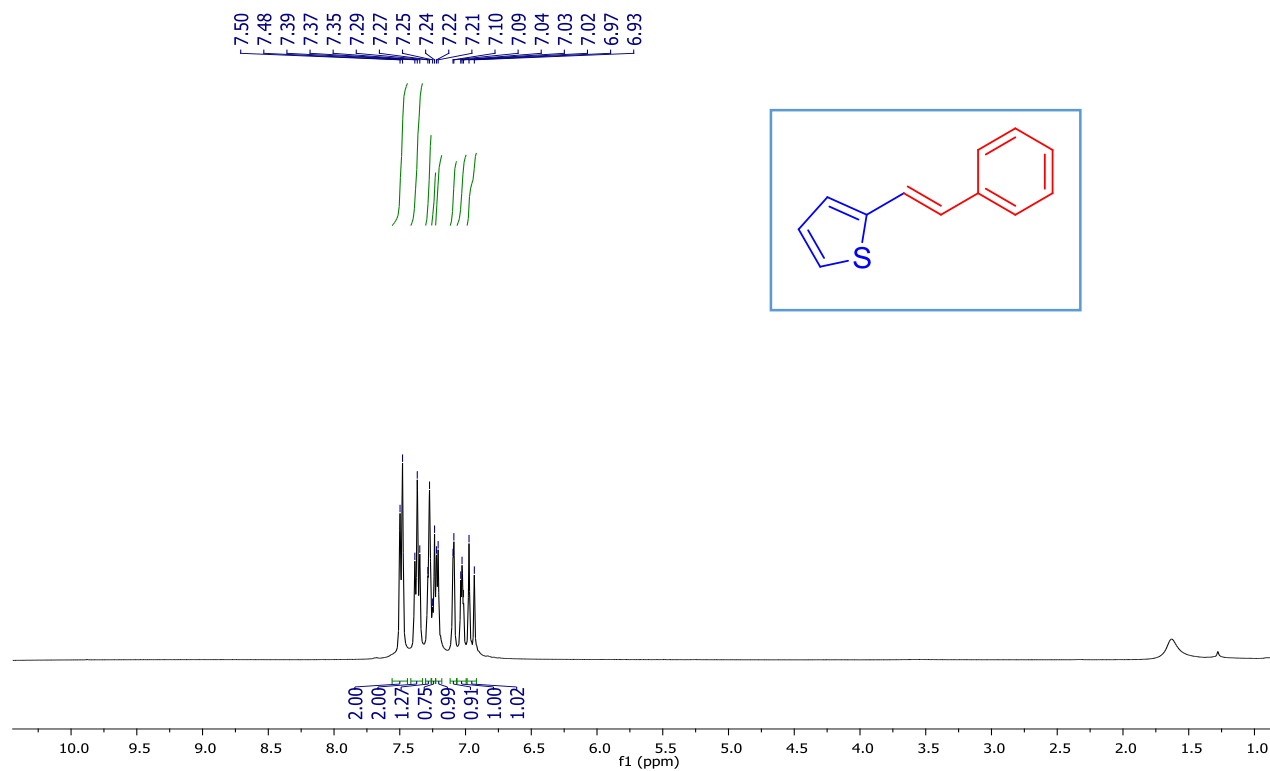


S64. ¹³CNMR for Ethyl (*E*)-3-styrylbenzoate

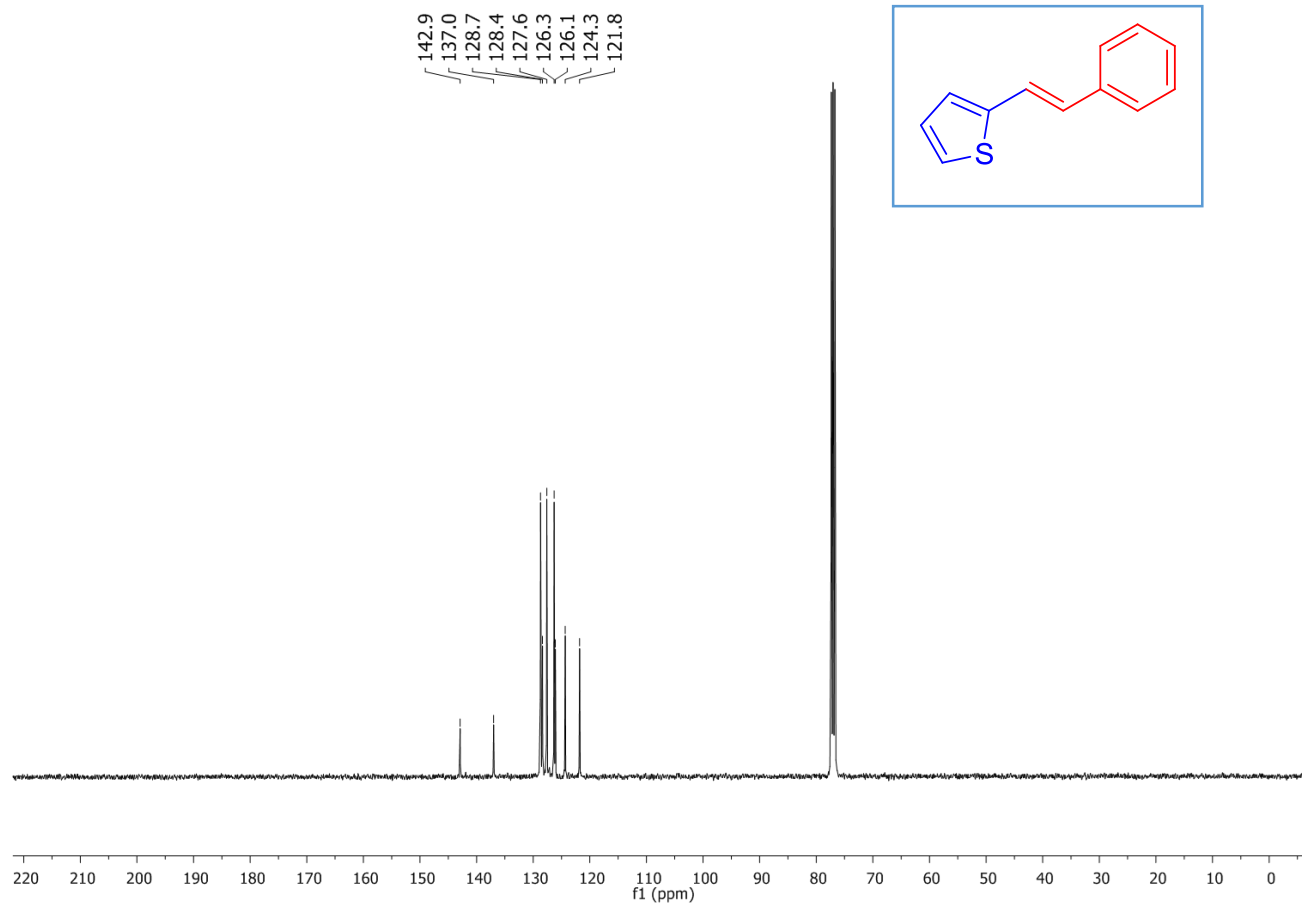
(*E*)-2-styrylthiophene



S65. GC-Mass of (*E*)-2-styrylthiophene



S66. ¹HNMR Compound of (E)-2-styrylthiophene

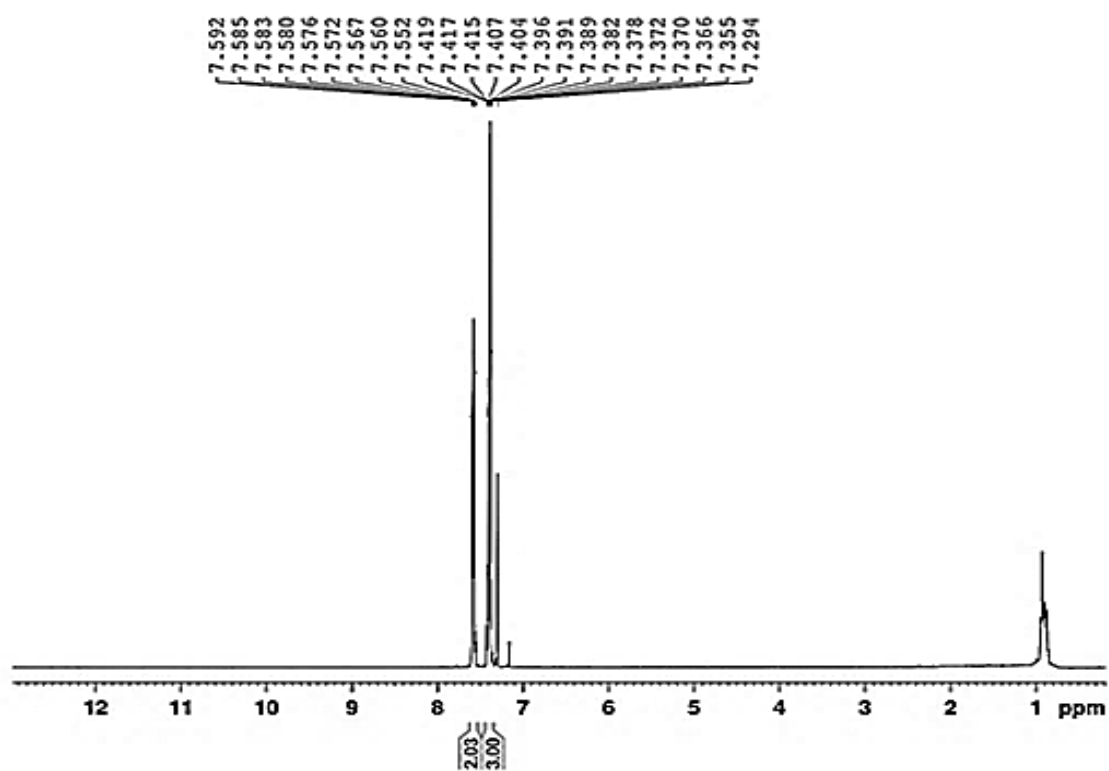
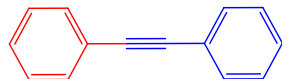


S67. ¹³CNMR Compound of (E)-2-styrylthiophene

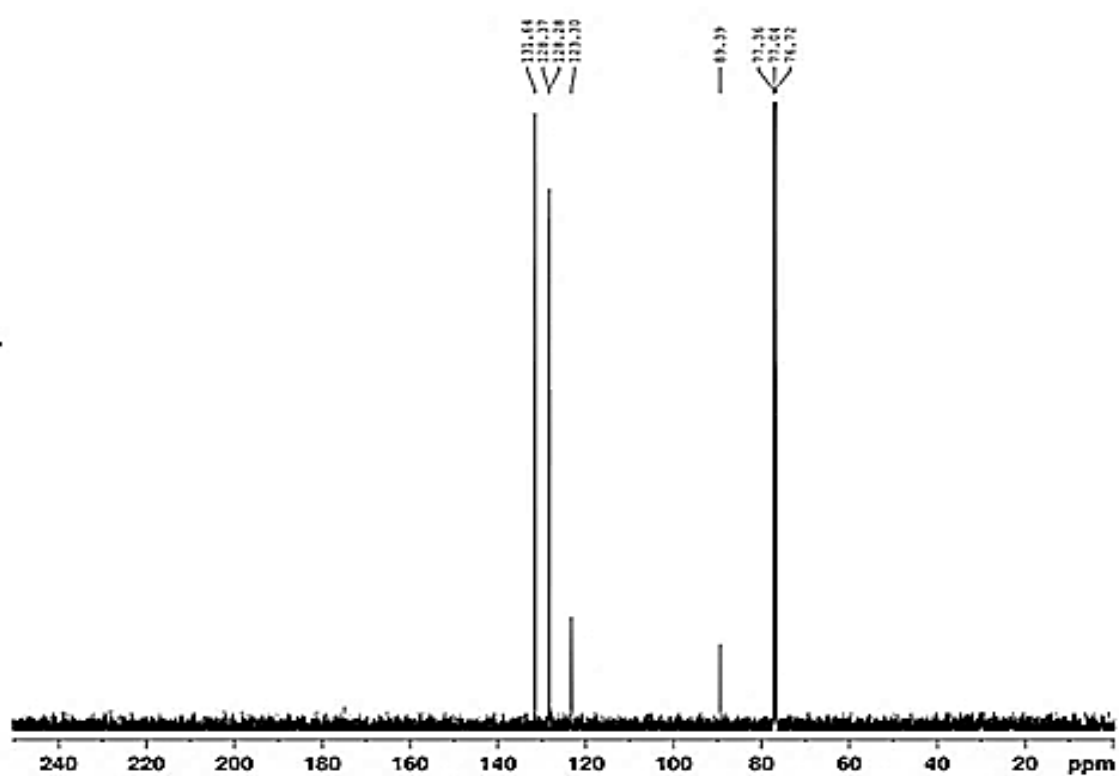
Sonogashira derivatives

1. 1,2-diphenylethyne

Melting Point (M.P) = 60-62°C



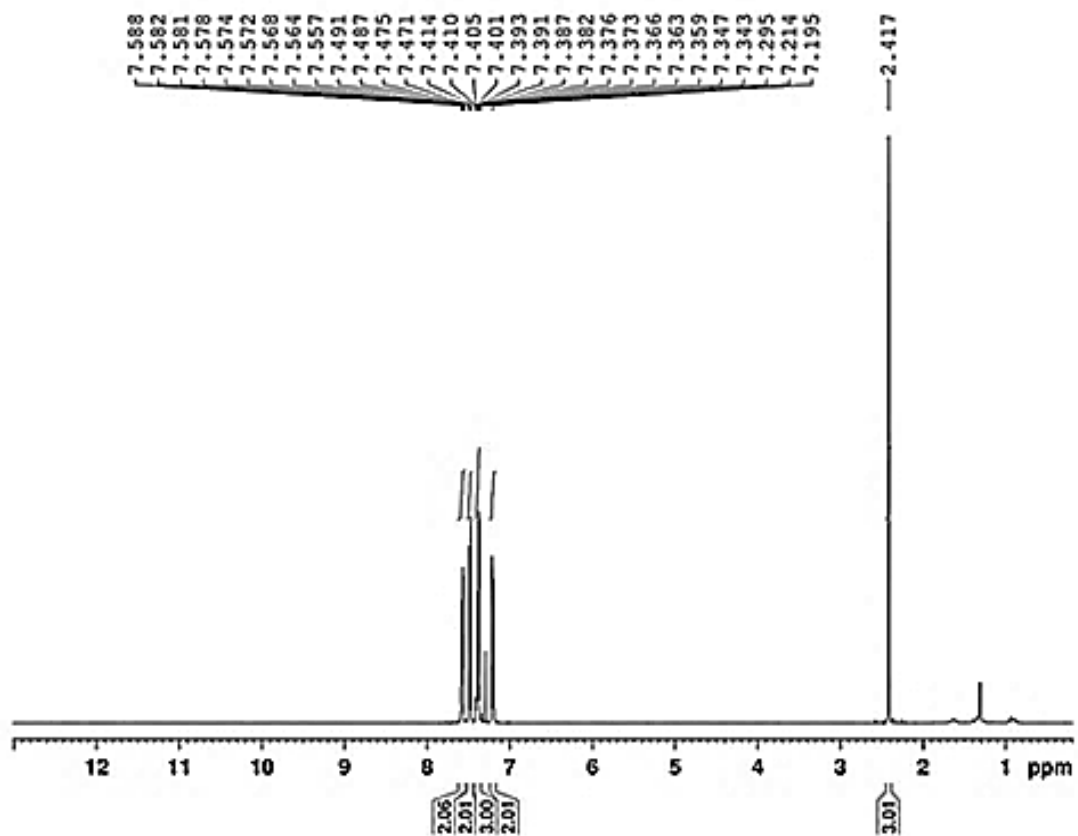
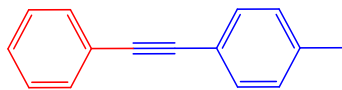
S68. ¹H NMR Compound of 1,2-Diphenylethyne



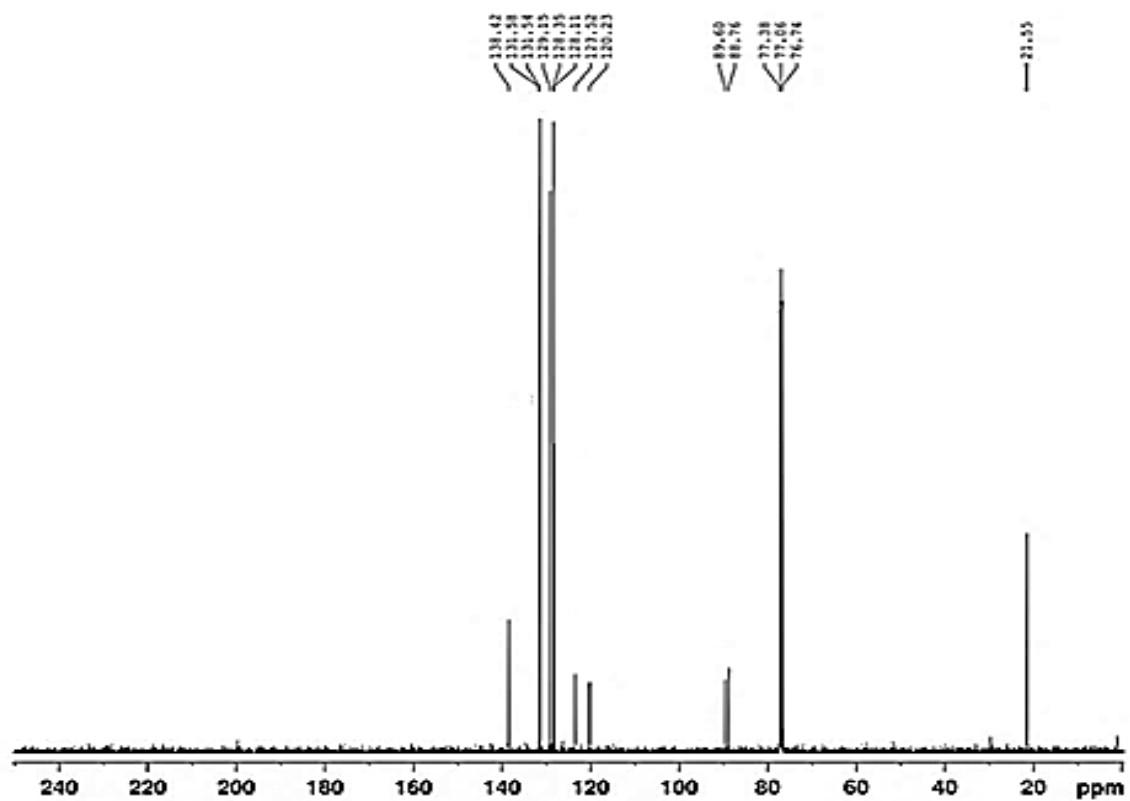
S69. ^{13}C NMR Compound of 1,2-Diphenylethyne

2. 1-methyl-4-(phenylethynyl)benzene

Melting Point (M.P) = 70-71°C



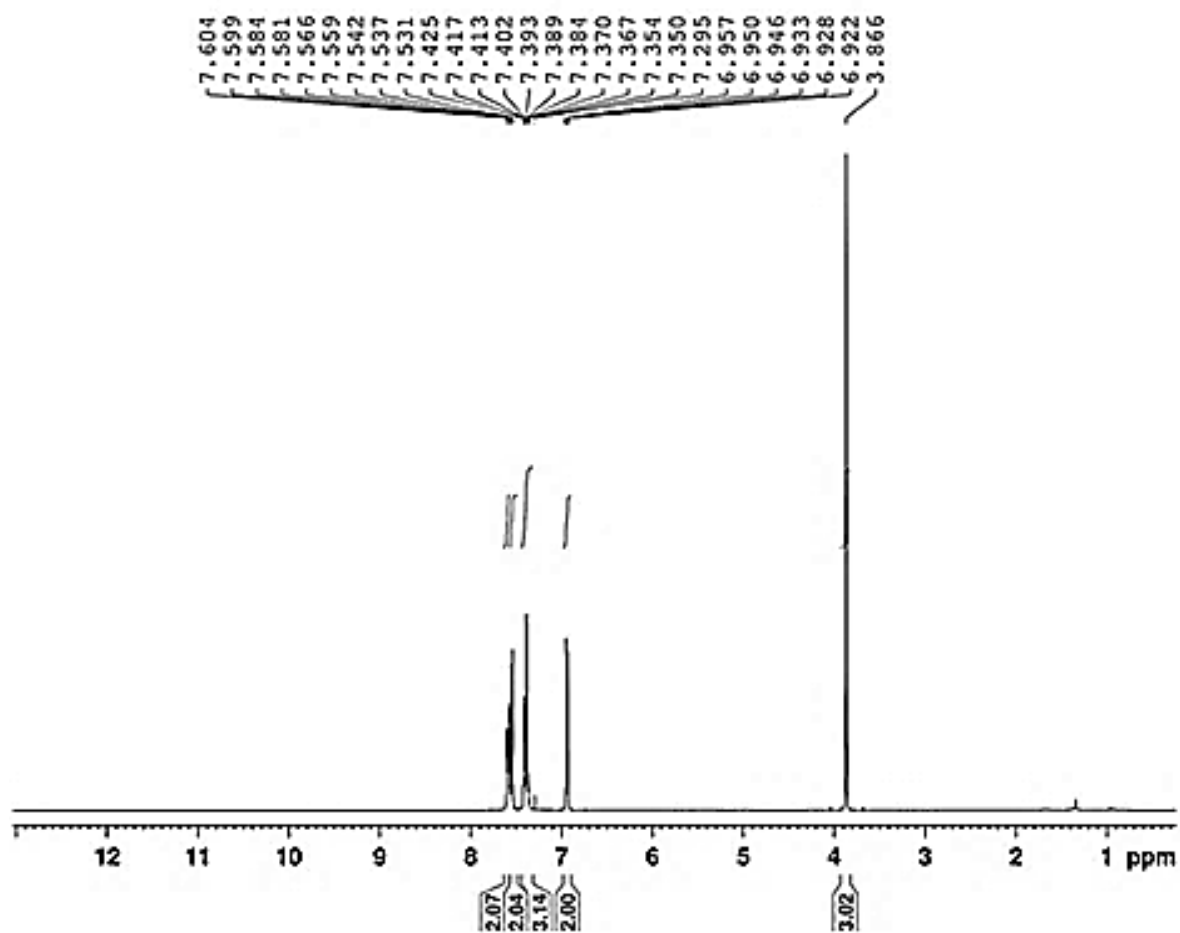
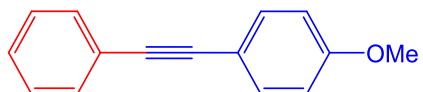
S70. ¹H NMR Compound of 1-methyl-4-(phenylethynyl) benzene



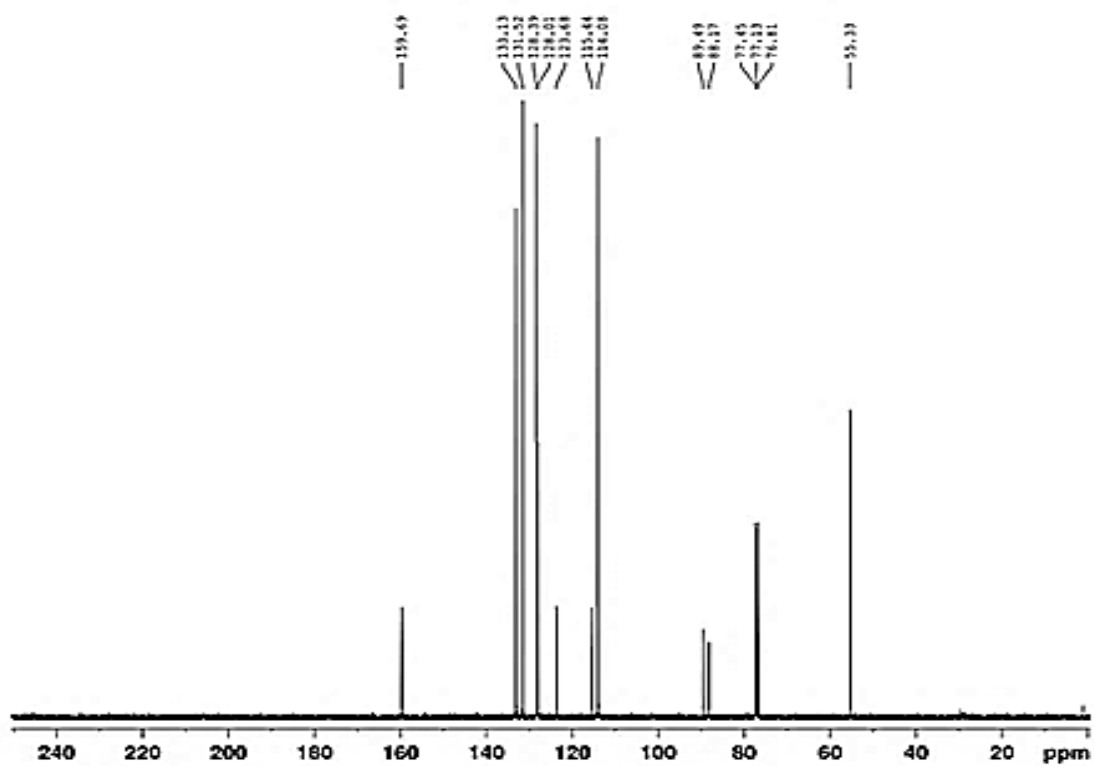
S71. ^{13}C NMR Compound of 1-methyl-4-(phenylethynyl) benzene

3. 1-methoxyl-4-(phenylethynyl)benzene

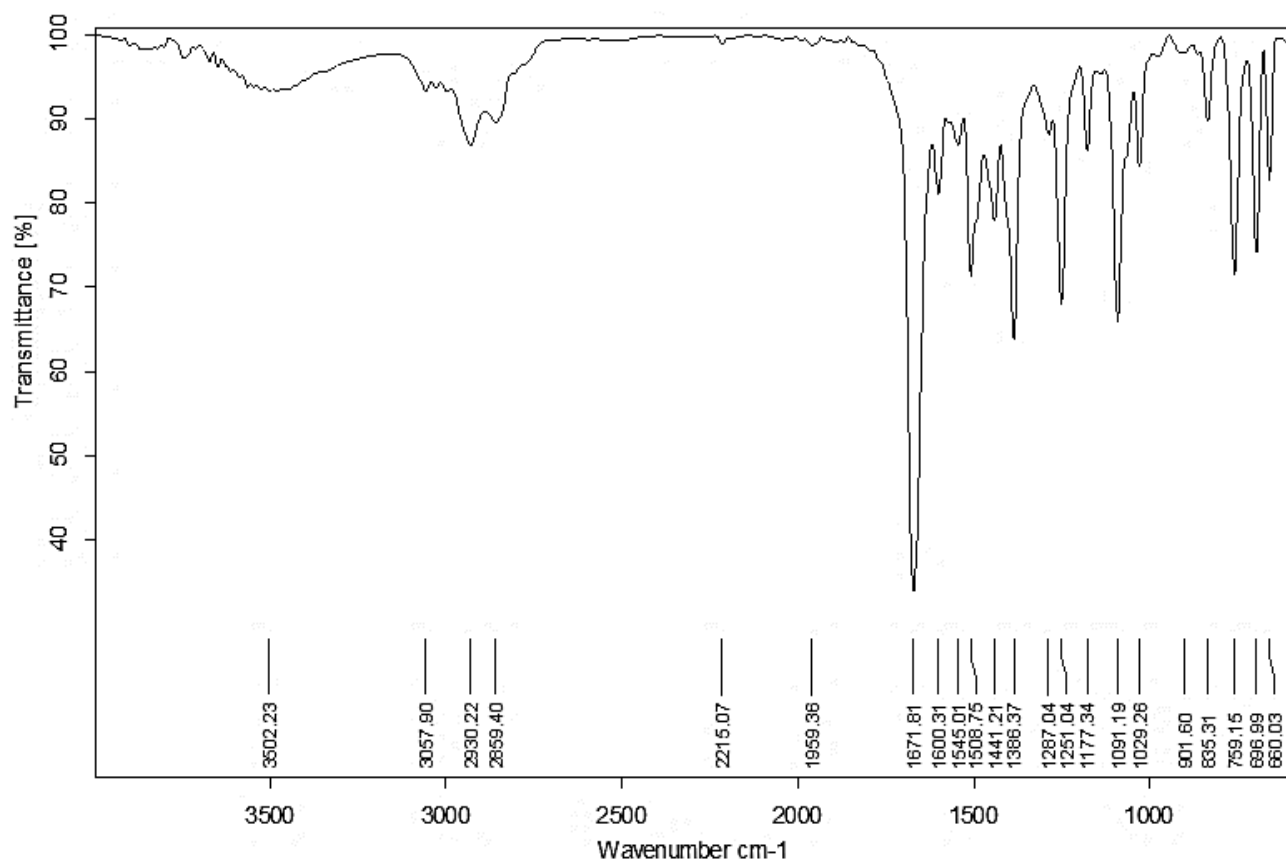
Melting Point (M.P) = 61-63°C



S72. ¹H NMR Compound of 1-methoxy-4-(phenylethynyl) benzene



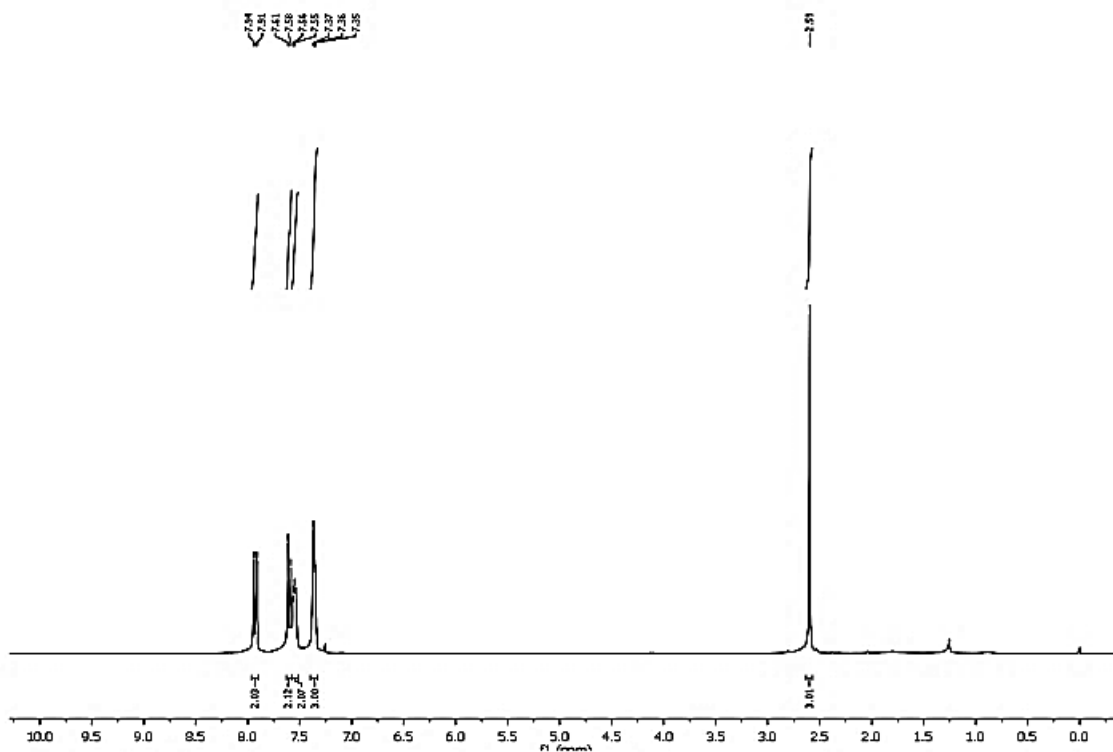
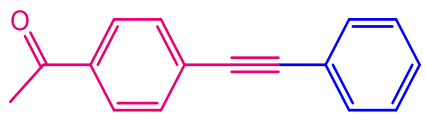
S73. ^{13}C NMR Compound of 1-methoxy-4-(phenylethynyl) benzene



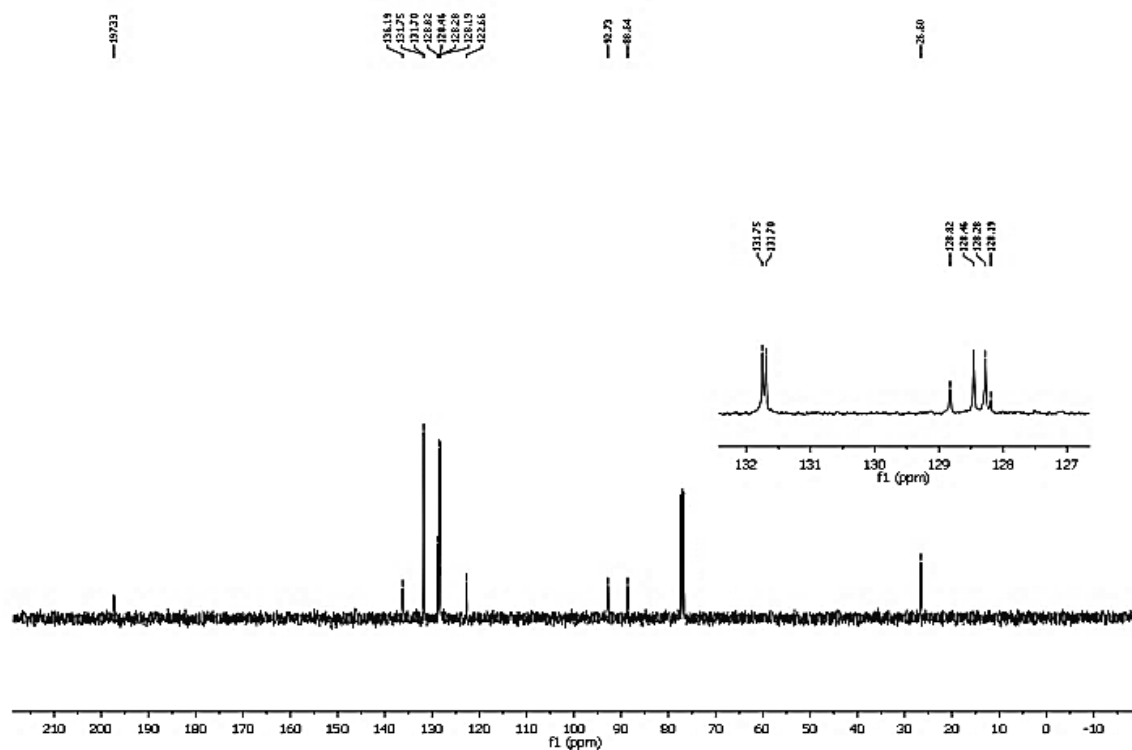
S74. FT-IR Compound of 1-methoxy-4-(phenylethynyl) benzene

4. 1-(4-(phenylethynyl) phenyl) ethan-1-one

Melting Point (M.P) = 99-101°C



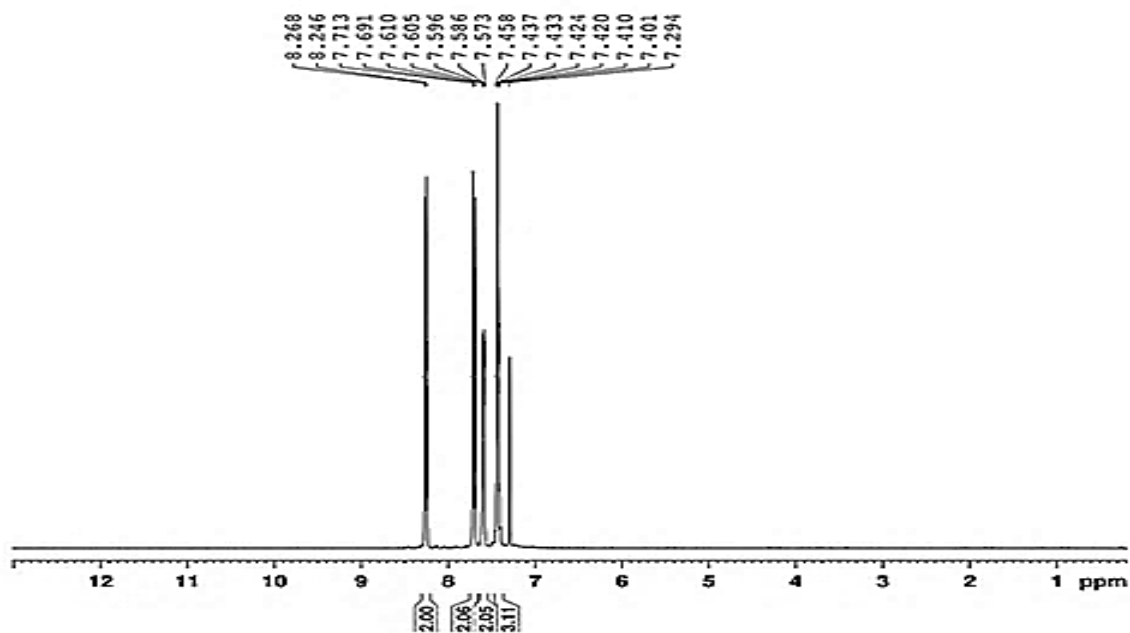
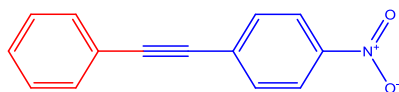
S75. ¹H NMR Compound of 1-(4-(phenylethynyl) phenyl) ethan-1-one



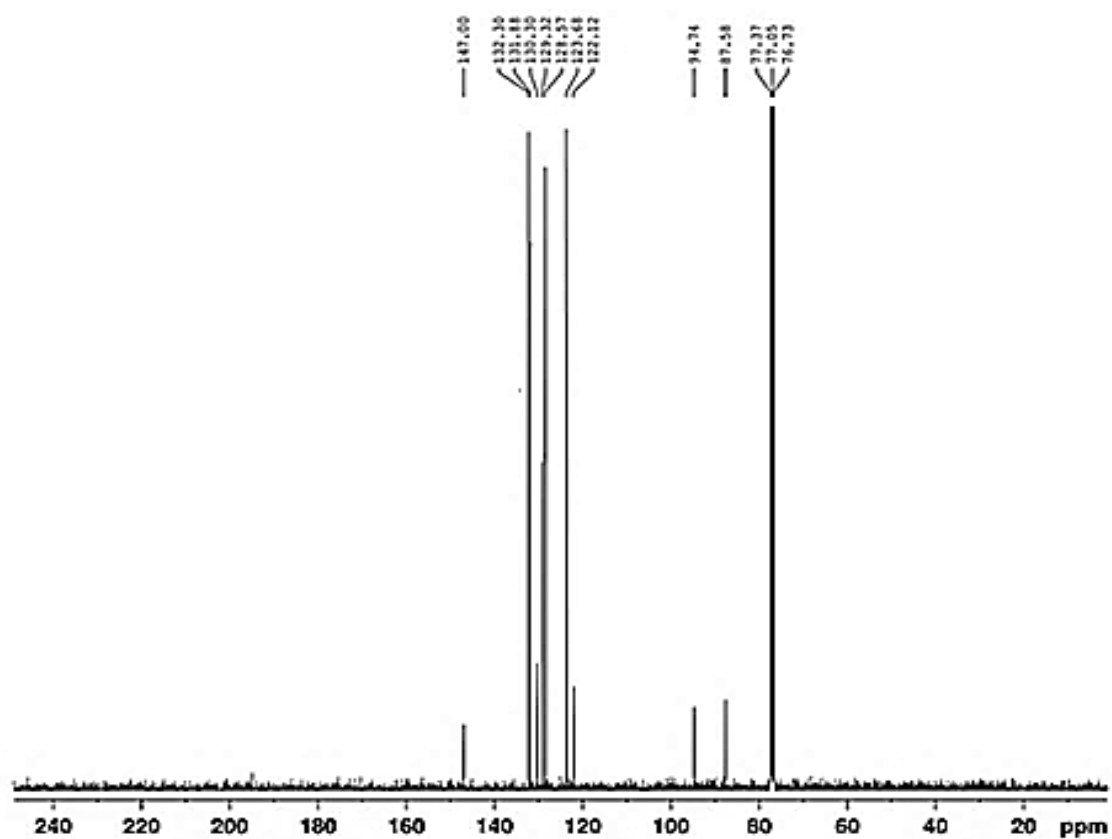
S76. ¹³CNMR Compound of 1-(4-(phenylethynyl) phenyl) ethan-1-one

5. 1-nitro-4-(phenylethynyl) benzene

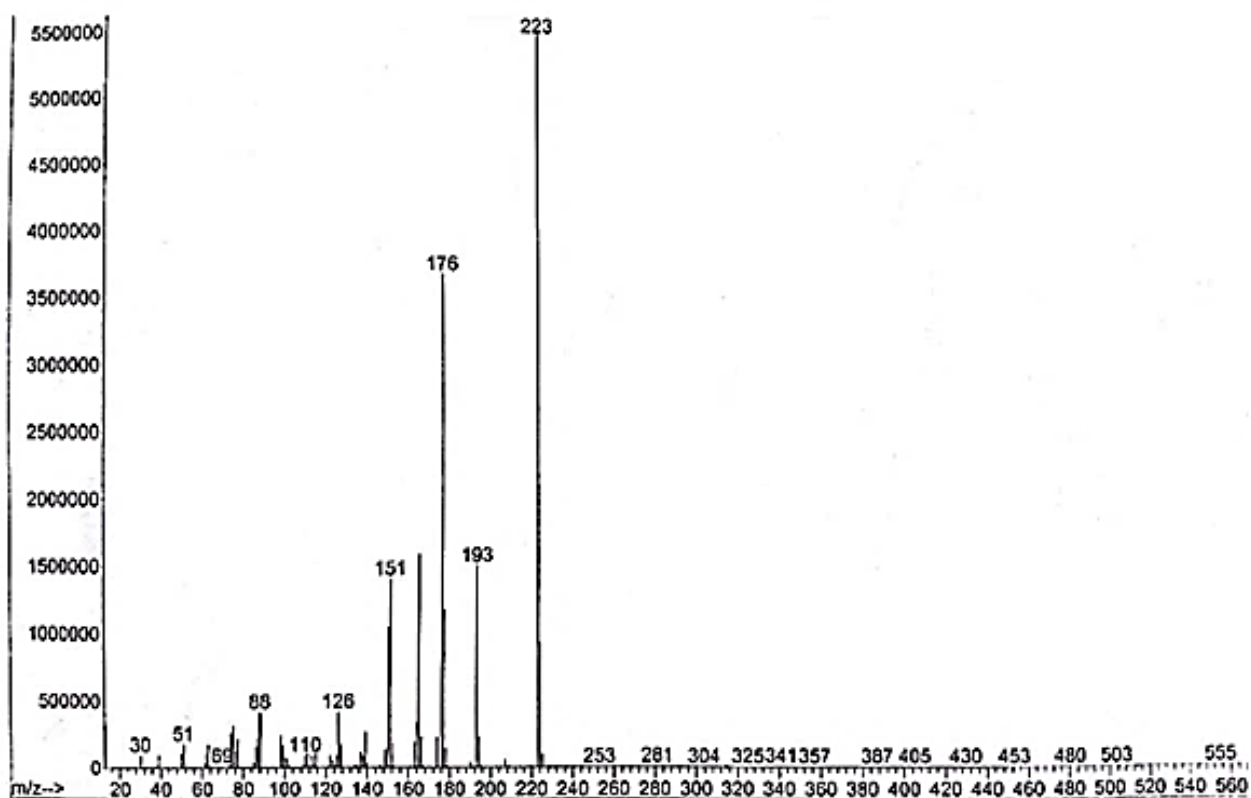
Melting Point (M.P) = 112-114°C



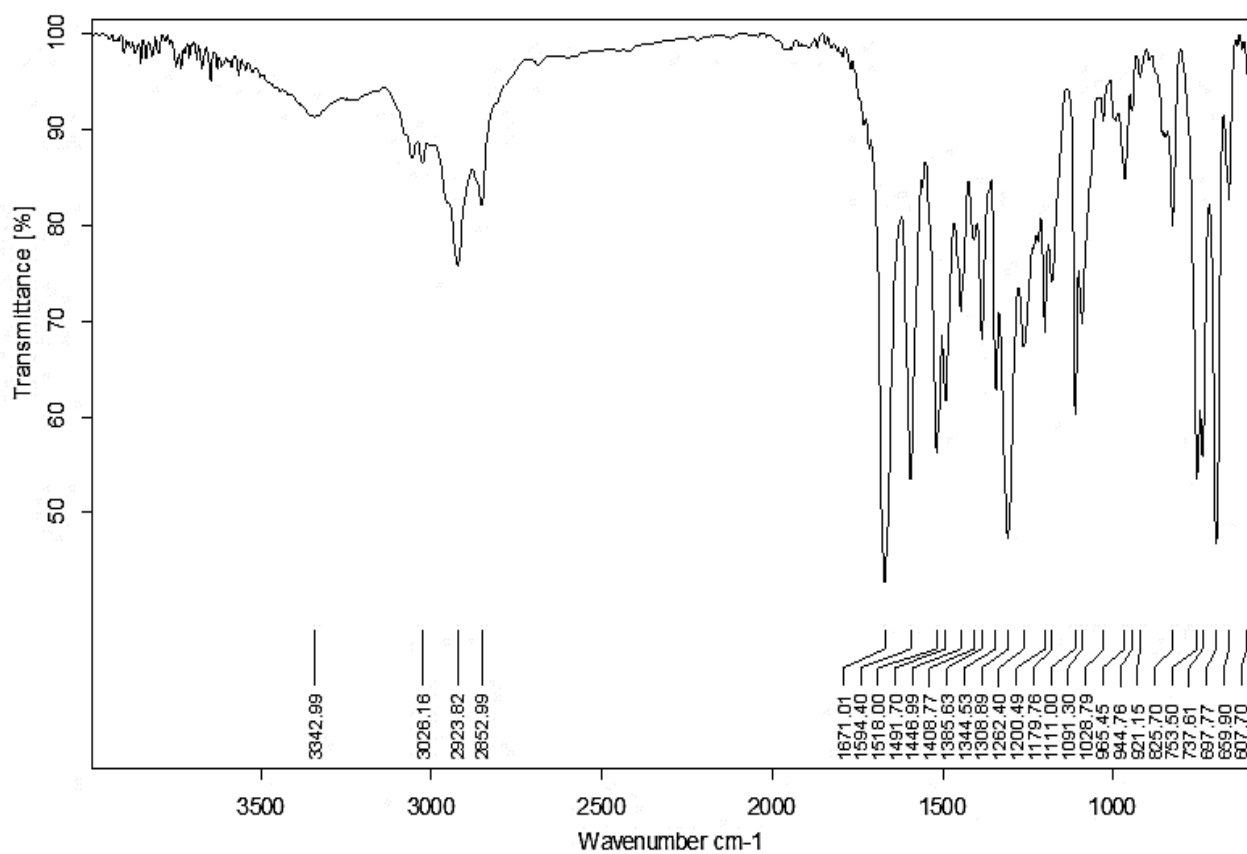
S77. ¹H NMR Compound of 1-nitro-4-(phenylethynyl) benzene



S78. ¹³CNMR Compound of 1-nitro-4-(phenylethynyl) benzene

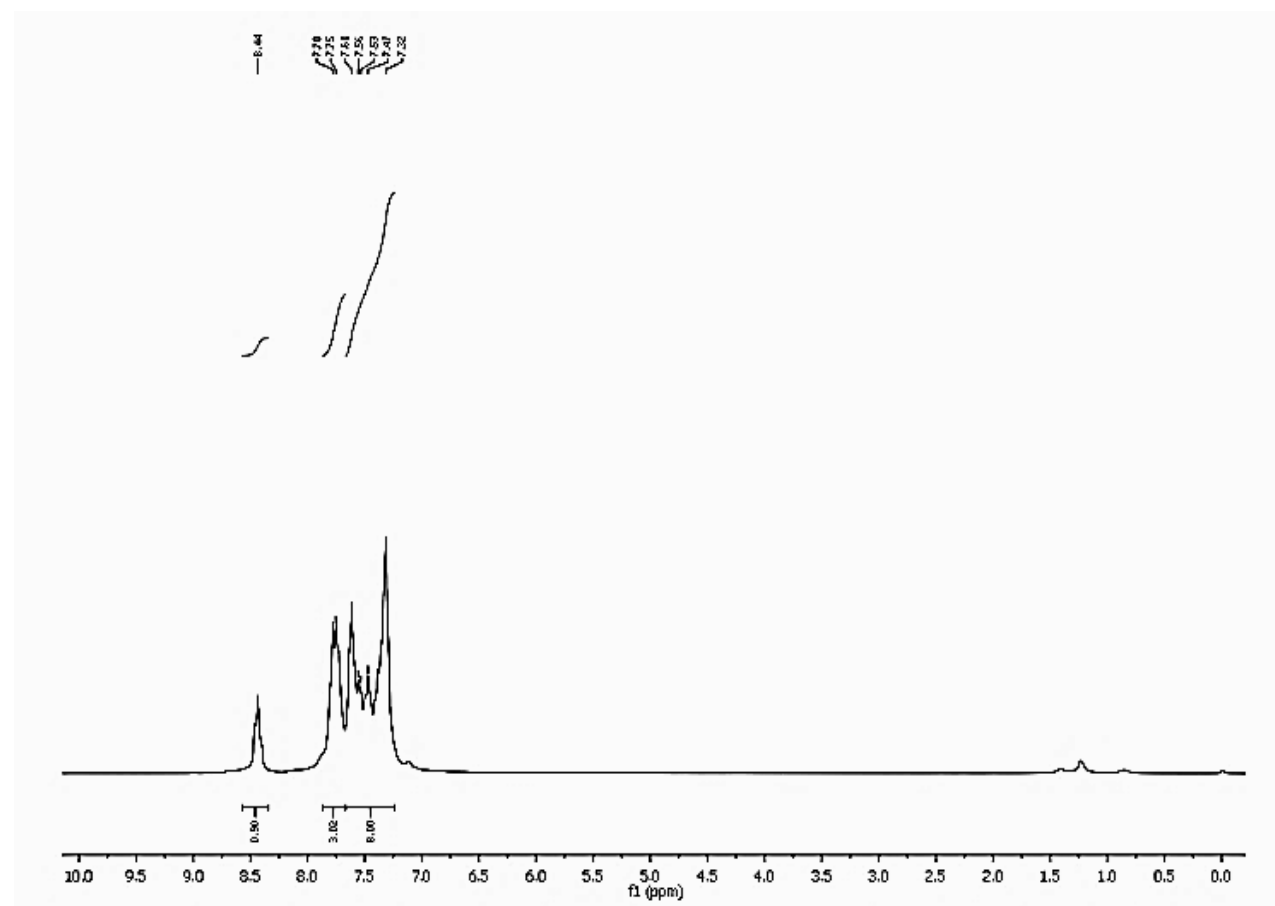
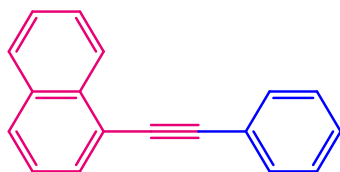


S79. GC-Mass Compound of 1-nitro-4-(phenylethynyl)benzene

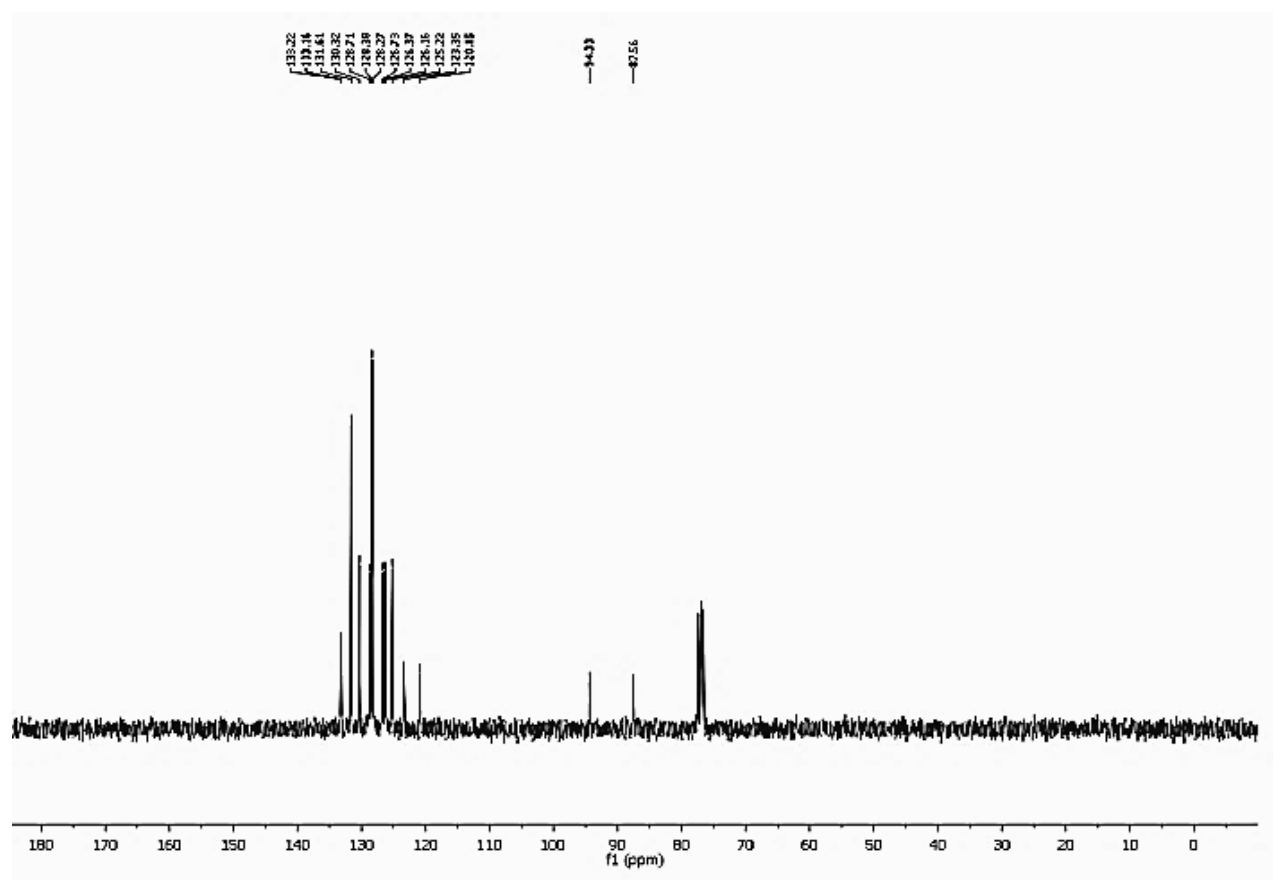


S80. FT-IR Compound of 1-nitro-4-(phenylethynyl)benzene

6. 1-(phenylethynyl)naphthalene



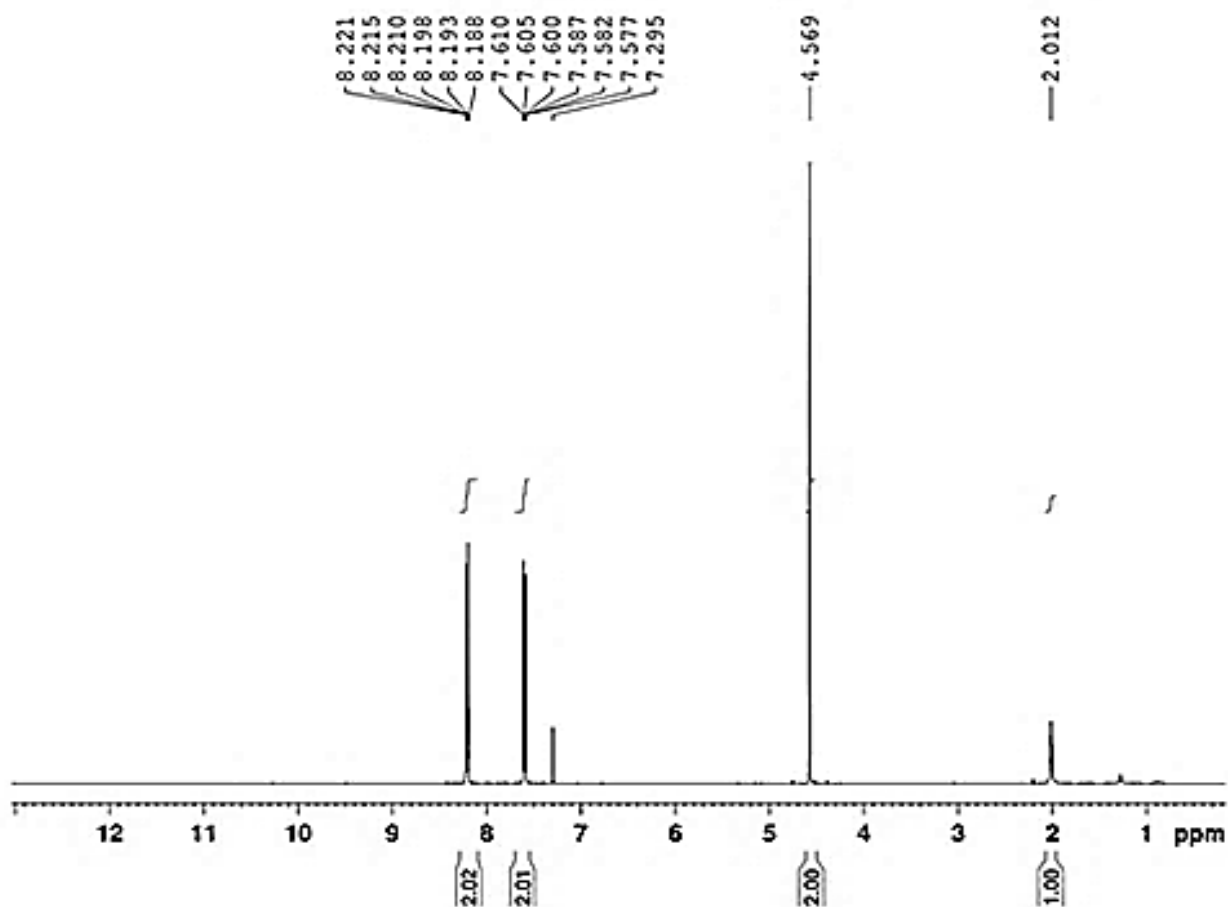
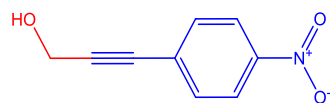
S81. ¹H NMR Compound of 1-(phenylethynyl)naphthalene



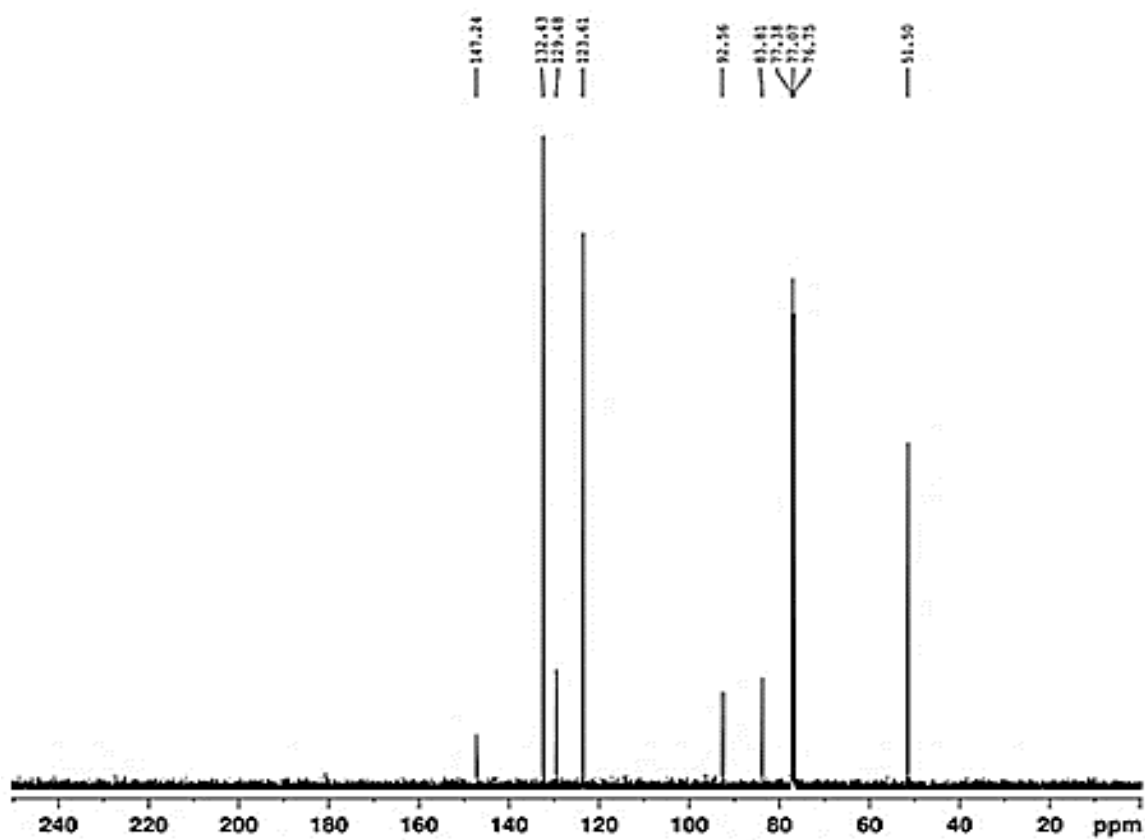
S82. ¹³CNMR Compound of 1-(phenylethynyl) naphthalene

7. 3-(4-nitrophenyl) prop-2-yn-1-ol

Melting Point (M.P) = 94-96°C



S83. ¹H NMR Compound of 3-(4-nitrophenyl) prop-2-yn-1-ol



S84. ¹³CNMR Compound of 3-(4-nitrophenyl) prop-2-yn-1-ol

References

- [1] M. Tanhaei, A. Mahjoub & R. Nejat, *Catal. Lett.* 2018, **148**, 1549-1561.
- [2] V. S. Coker, J. A. Bennett, N. D. Telling, T. Henkel & J. M. Charnock, *Acs Nano*. 2010, **4**, 2577-2584.
- [3] H. Yoon, S. Ko & J. Jang, *Chem. Commun.* 2007, **14**, 1468-1470.
- [4] X. Ma, Y. Zhou, J. Zhang, A. Zhu, T. Jiang & B. Han, *Green Chem.* 2008, **10**, 59-66.
- [5] J. Yang, D. Wang, W. Liu, X. Zhang, F. Bian & W. Yu, *Green Chem.* 2013, **15**, 3429-3437.
- [6] Y. V. Ioni, S. E. Lyubimov, V. A. Davankov & S. P. Gubin, *Russ. J. Inorg. Chem.* 2013, **58**, 392-394.
- [7] S. Ko & J. Jang, *Angew. Chem. Int. Ed.* 2006, **45**, 7564-7567.
- [8] M. Zhu, Y. Wang, C. Wang, W. Li & G. Diao, *Catal. Sci. Technol.* 2013, **3**, 952-961.
- [9] M. Zhu & G. Diao, *J. Phys. Chem. C*. 2011, **115**, 24743-24749.
- [10] M. R. Nabid, Y. Bide & S. J. T. Rezaei, *Appl. Catal. A: Gen.* 2011, **406**, 124-132.
- [11] M. Tukhani, F. Panahi & A. Khalafi-Nezhad, *ACS Sustain. Chem. Eng.* 2018, **6**, 1456-1467.
- [12] N. Iranpoor, S. Rahimi & F. Panahi, *RSC Adv.* 2015, **5**, 49559-49567.
- [13] S. Ashiri & E. Mehdipour, *RSC Adv.* 2018, **8**, 32877-32885.
- [14] X. Cheng, W. Li, R. Nie, X. Ma, R. Sang, L. Guo & Y. Wu, *Adv. Synth. Catal.* 2017, **359**, 454-466.
- [15] S. D. Dindulkar, D. Jeong, H. Kim & S. Jung, *Carbohydr. Res.* 2016, **430**, 85-94.
- [16] J. Mondal, A. Modak & A. Bhaumik, *J. Mol. Catal. A: Chem.* 2011, **350**, 40-48.
- [17] A. Kamal, V. Srinivasulu, B. N. Seshadri & N. Shankaraiah, *Green Chem.* 2012, **14**, 2513-2522.
- [18] B. L. *Chem. Commun.* 1998, **13**, 1361-1362.
- [19] Y. Liao, L. He, J. Huang, J. Zhang, L. Zhuang, H. Shen, C.-Y., Su, *ACS Appl. Mater. Interfaces*. 2010, **2** (8), 2333-2338.
- [20] S. R. Borhade, S. B. Waghmode, *Beilstein J. org. chem.*, 2011, **7** (1), 310-319.
- [21] H. A. Elazab, A. R. Siamaki, S. Moussa, B. F. Gupton, M. S. El-Shall, *Appl Catal A: Gen.* 2015, **491**, 58-69.
- [22] M. Zeltner, A. Schätz, M. L. Hefti, W. J. Stark, *J. Mater. Chem.* 2011, **21** (9), 2991-2996.
- [23] S. Faisal, ROMP-Derived Alkylating Reagents and Scavengers: Application in Library Development and Sequestration. University of Kansas, 2016.
- [24] R. Li, P. Zhang, Y. Huang, P. Zhang, H. Zhong, Q. Chen, *J. Mater. Chem.* 2012, **22** (42), 22750-22755.
- [25] A. J. Amali, R. K. Rana, *Green Chem.* 2009, **11** (11), 1781-1786.

- [26] J. Yang, D. Wang, W. Liu, X. Zhang, F. Bian, W. Yu, *Green chem.* 2013, **15** (12), 3429-3437.
- [27] M. Tanhaei, A. Mahjoub, R. Nejat, *Catal. Lett.* 2018, **148** (6), 1549-1561.
- [28] T. H. Kwon, K. Y. Cho, K.-Y. Baek, H. G. Yoon, B. M. Kim, *RSC adv.* 2017, **7** (19), 11684-11690.