

Supplementary Materials for

“Ultra-Clean Angstromporous Monolayer Amorphous Carbon Yields High Precision Proton Beam”

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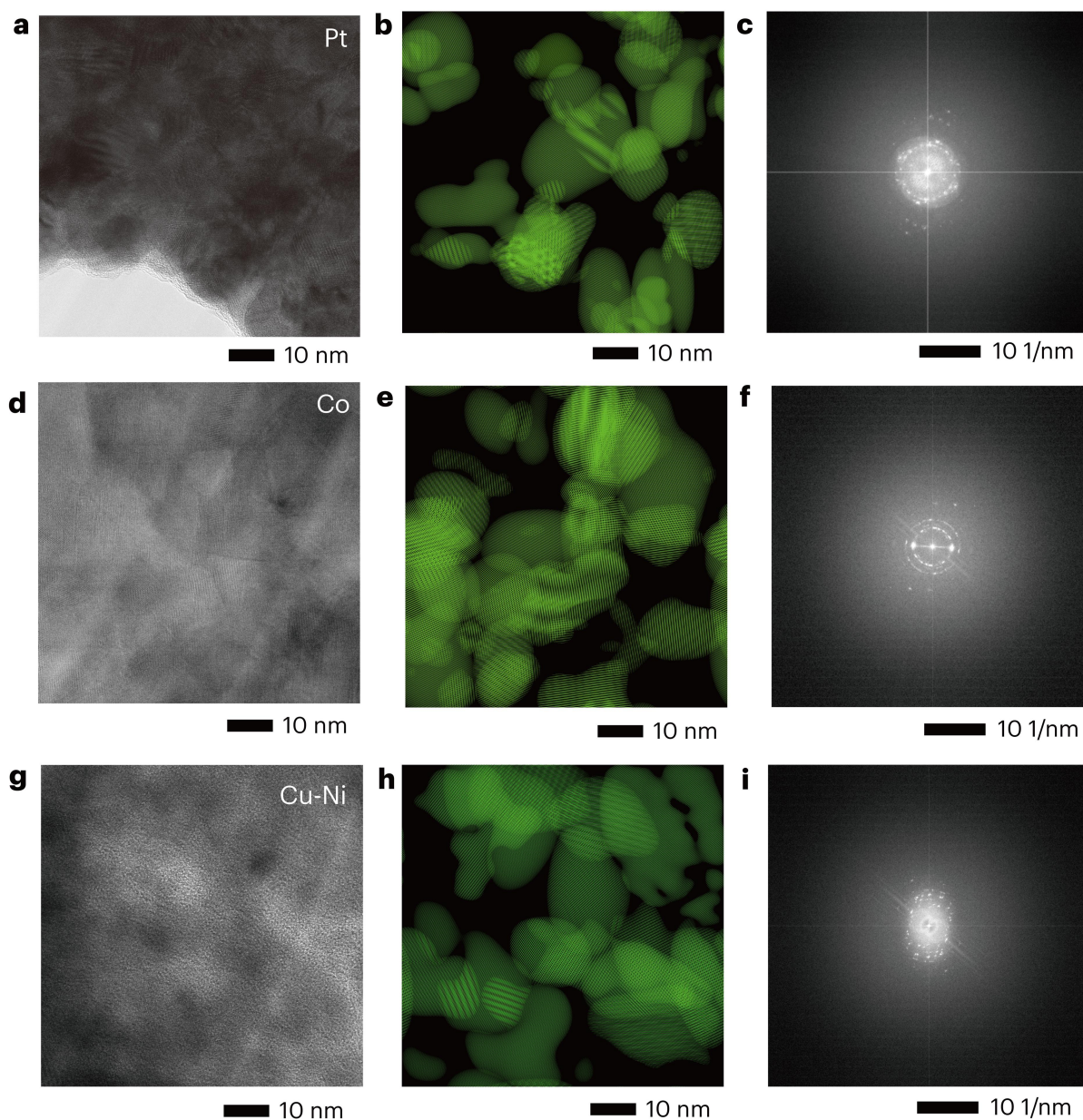
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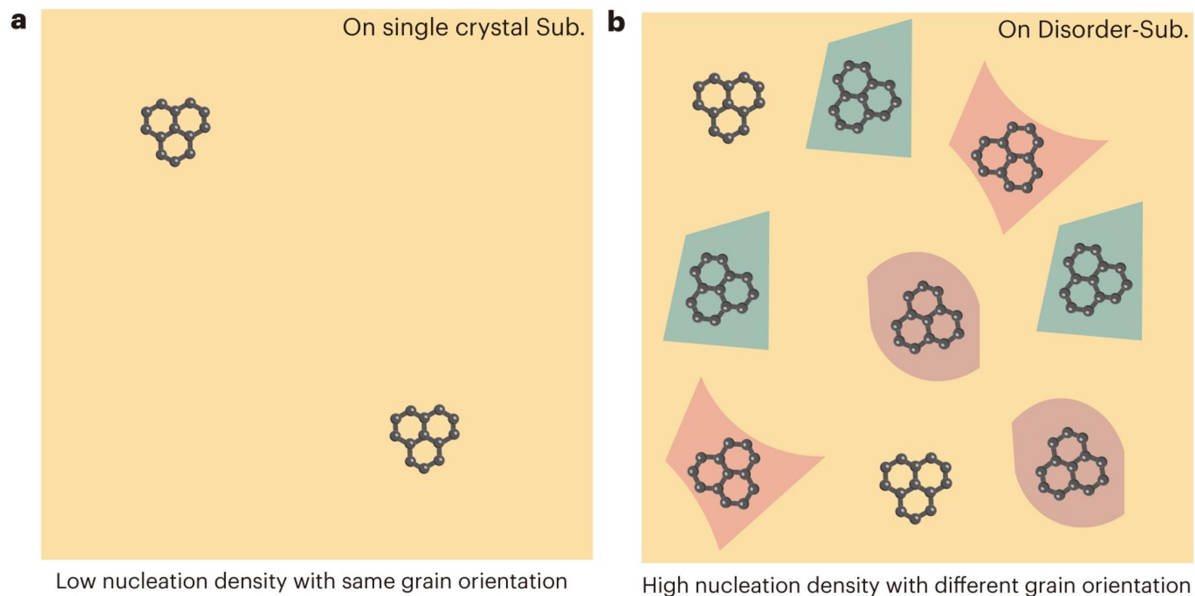
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Table 2 Representative preparation methods of monolayer graphene/amorphous carbon.

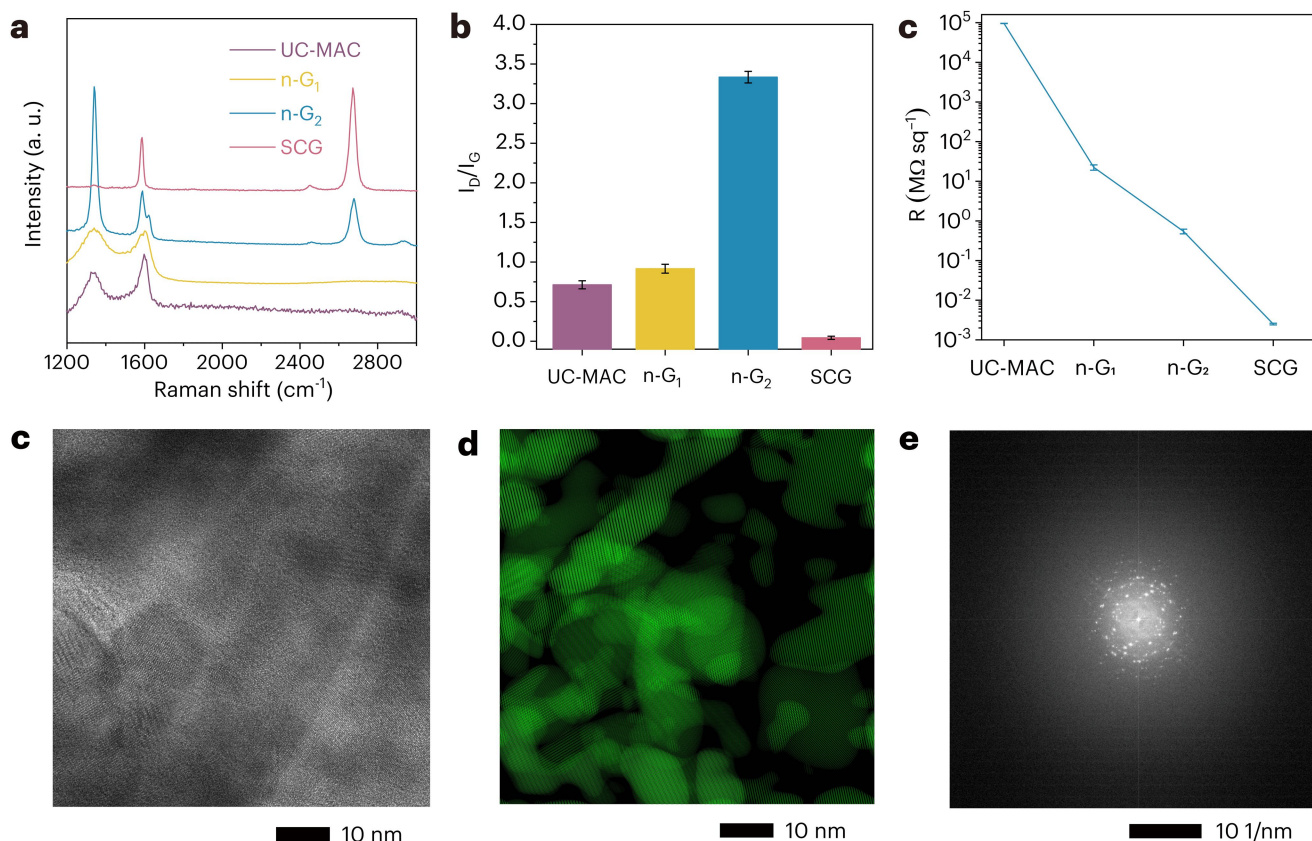
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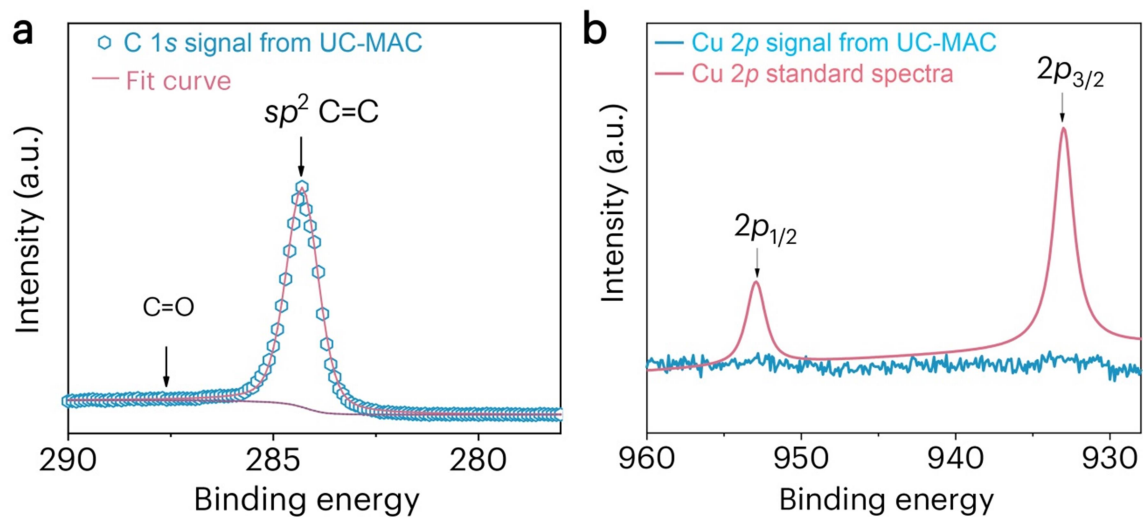
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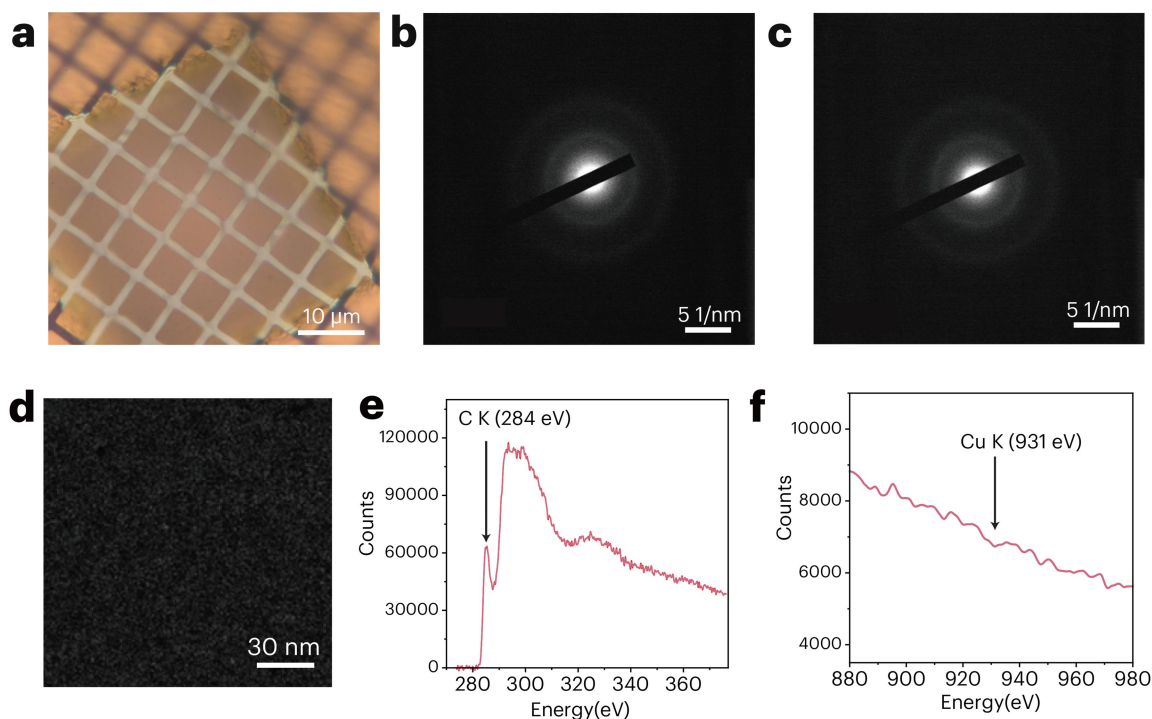
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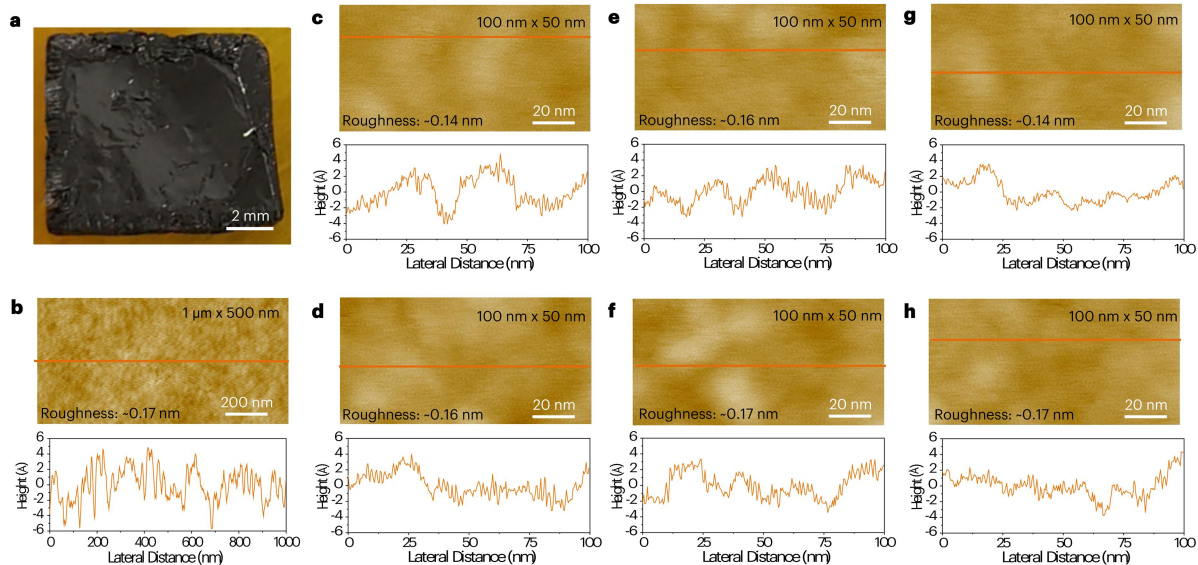
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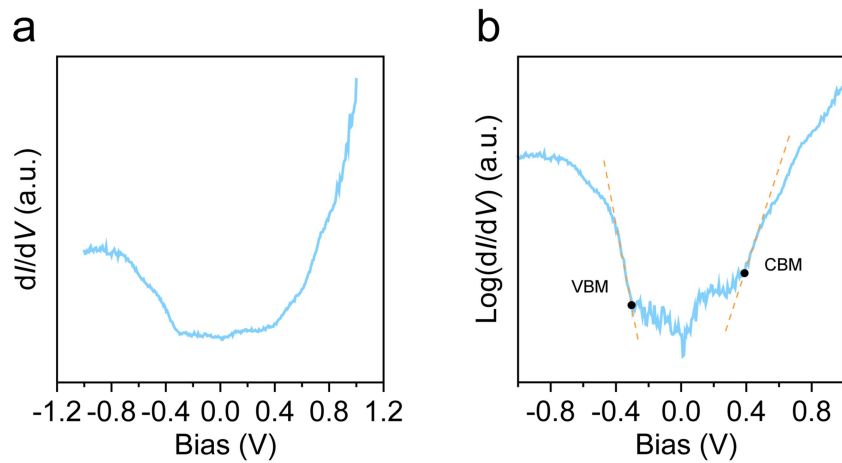
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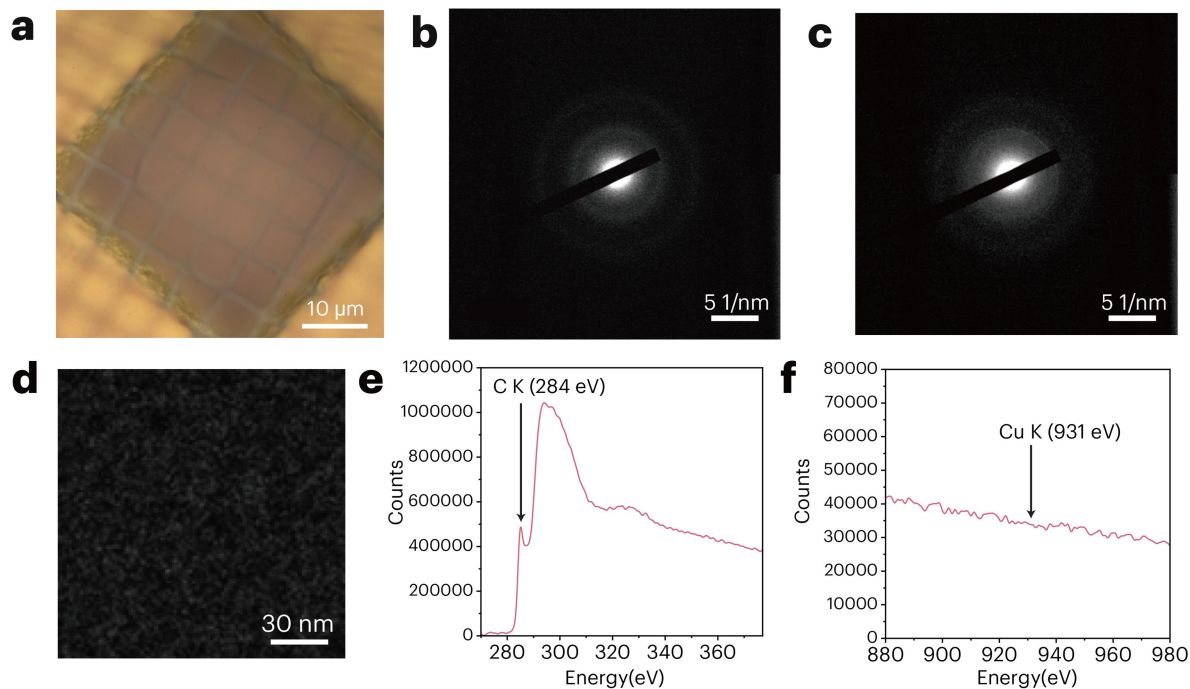
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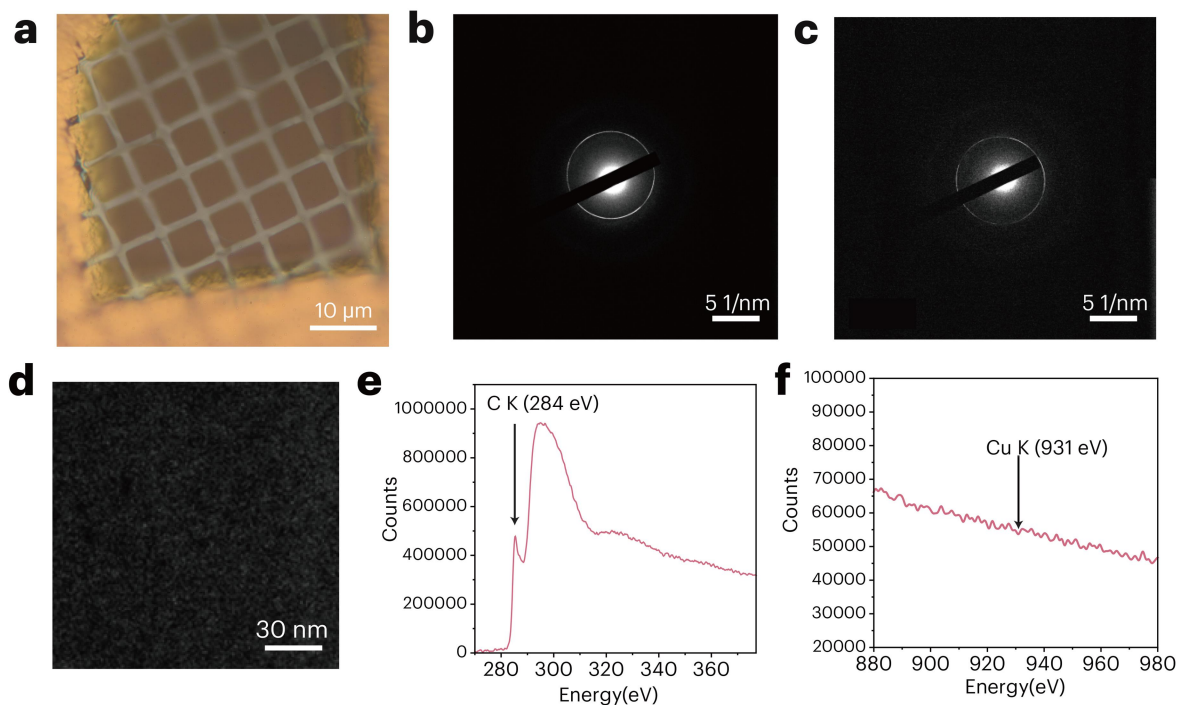
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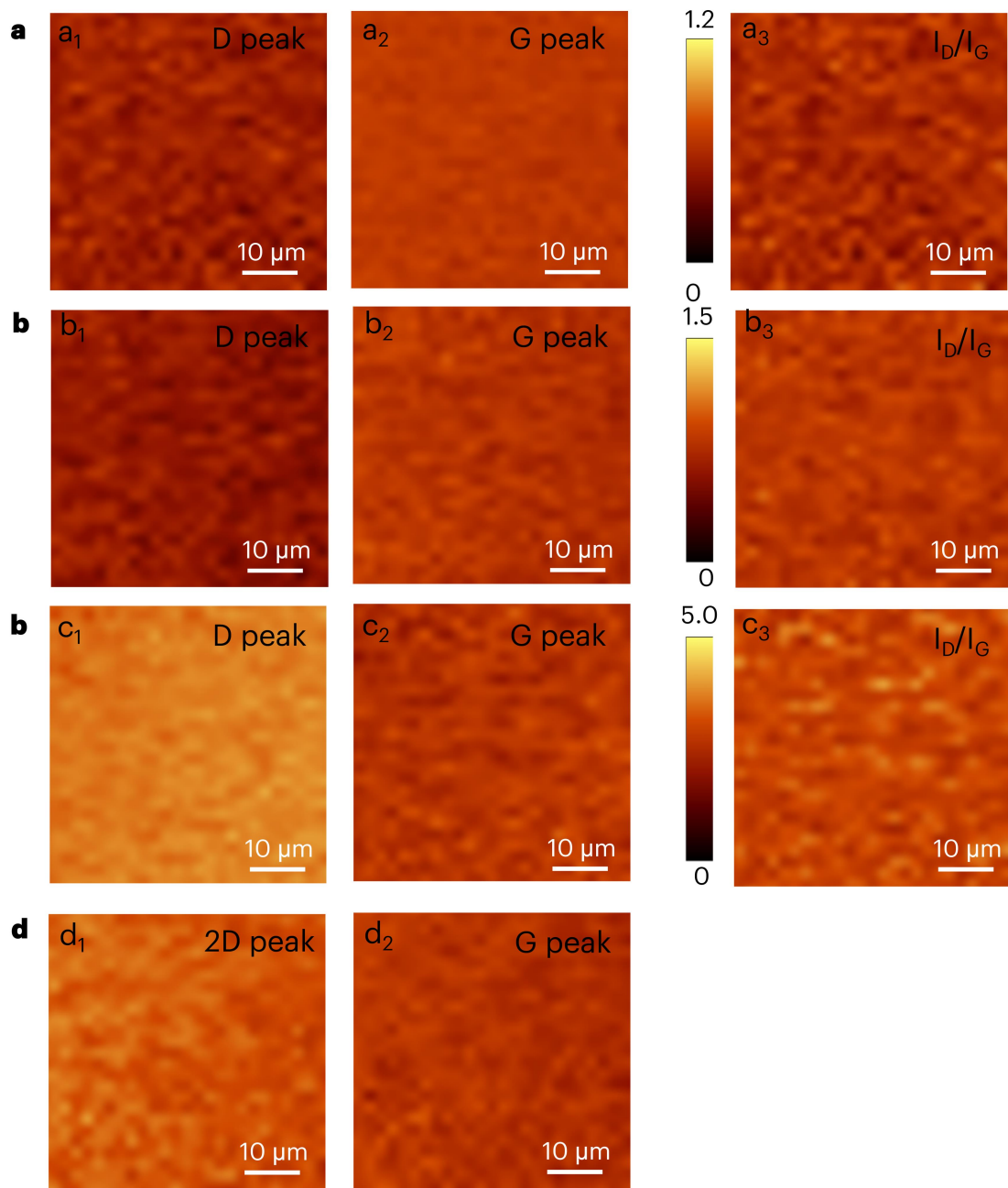
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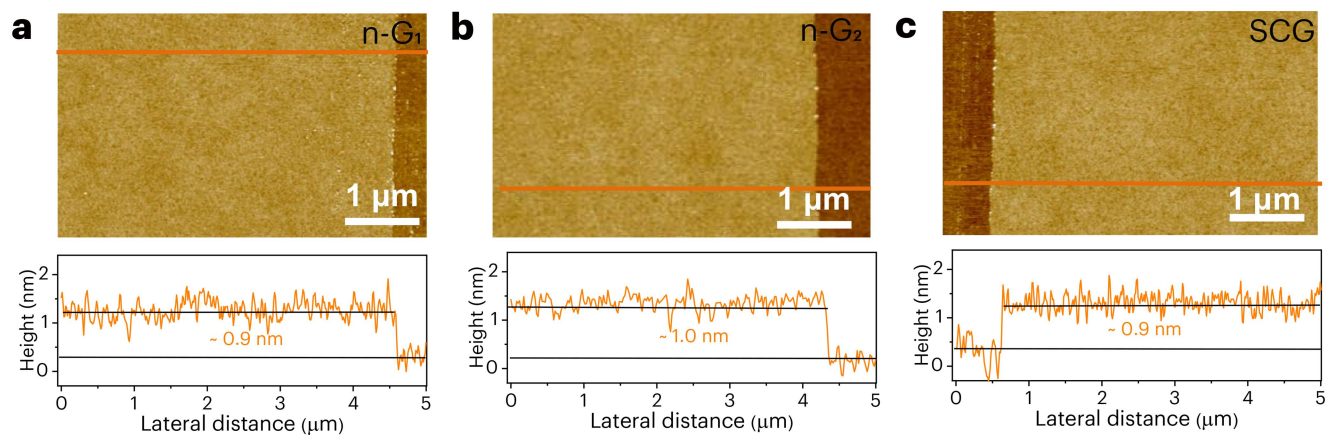
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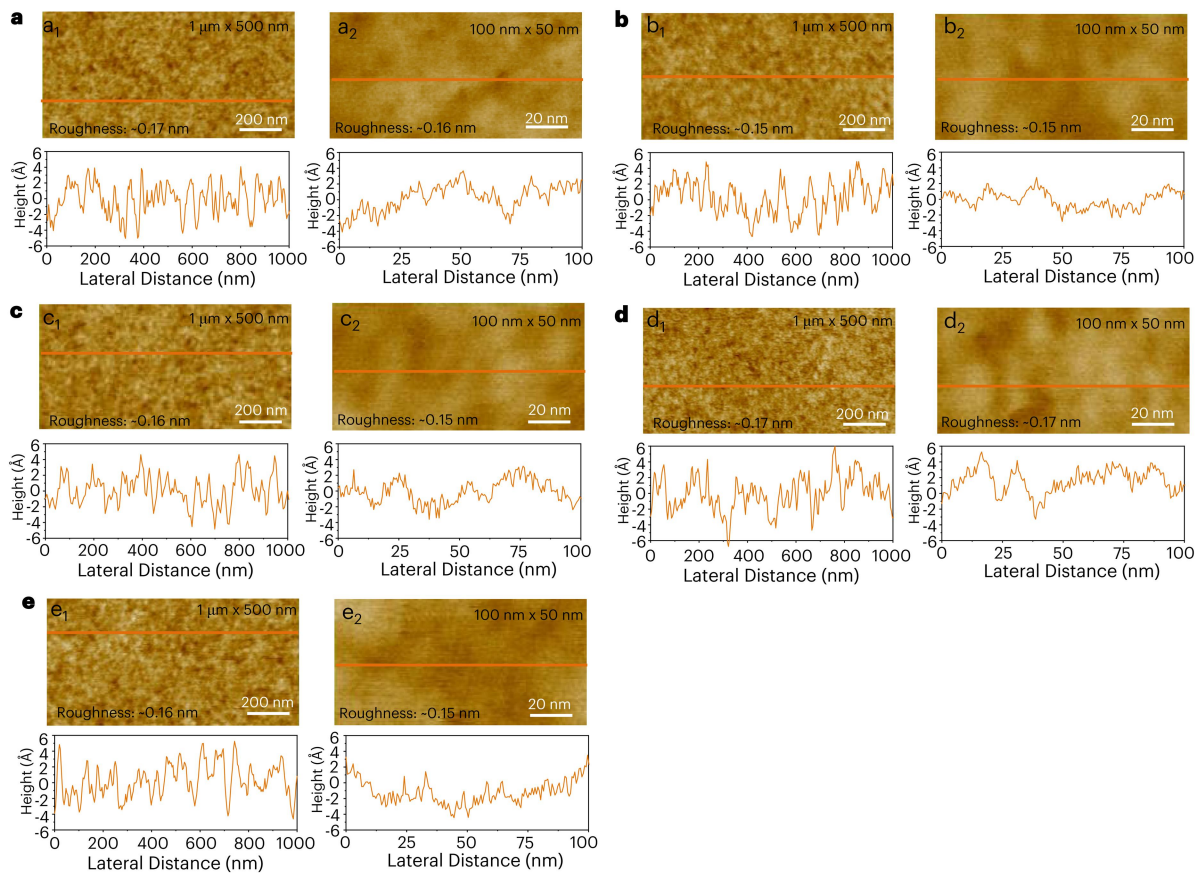
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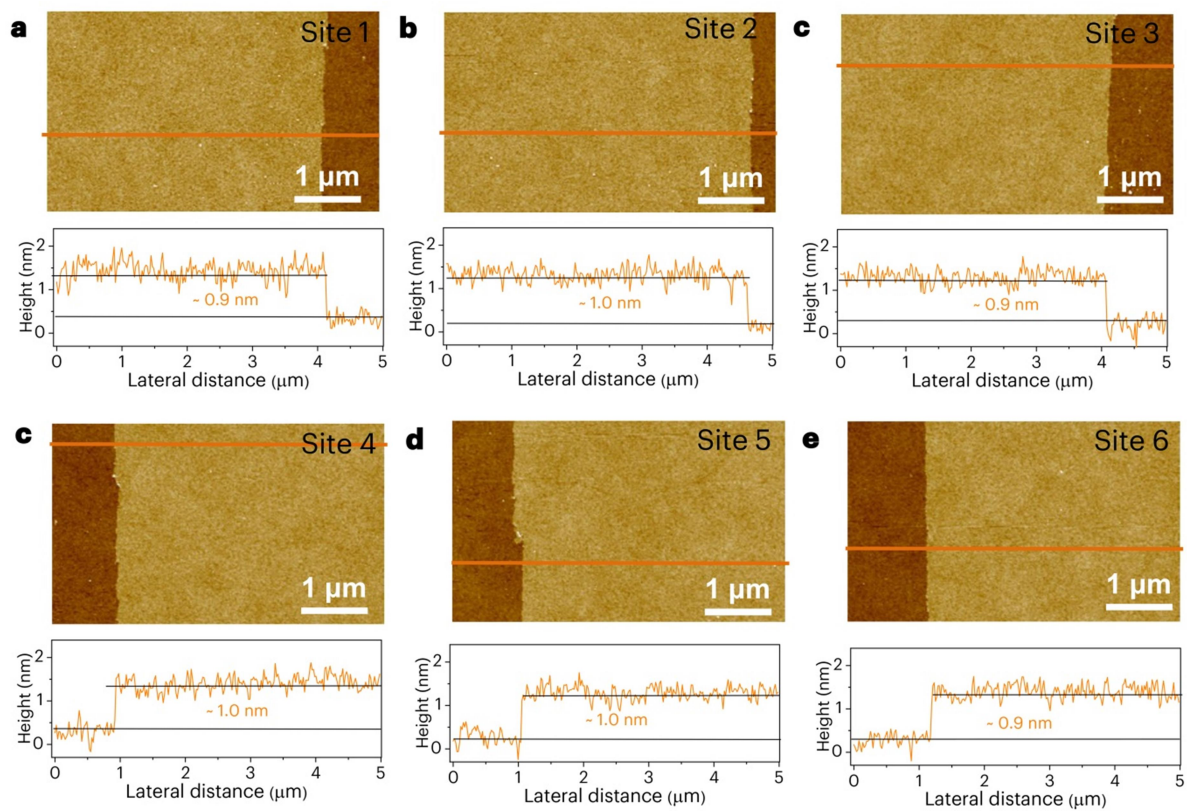
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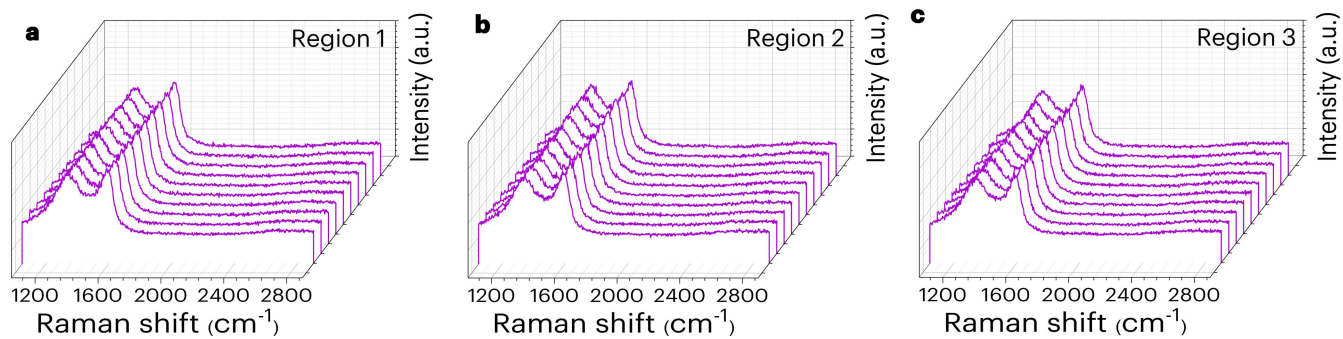
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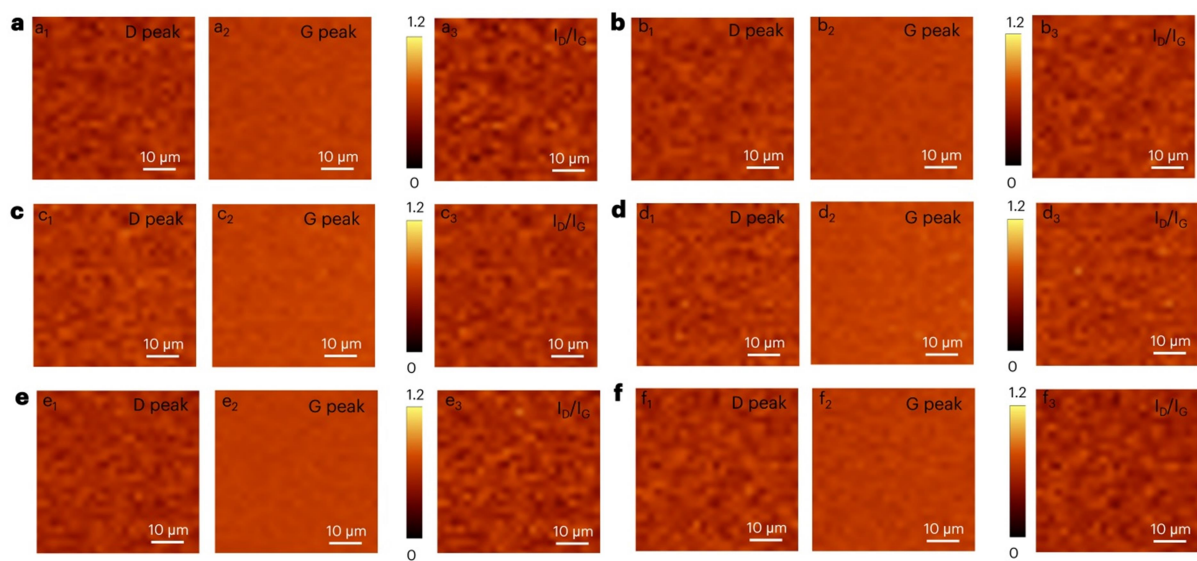
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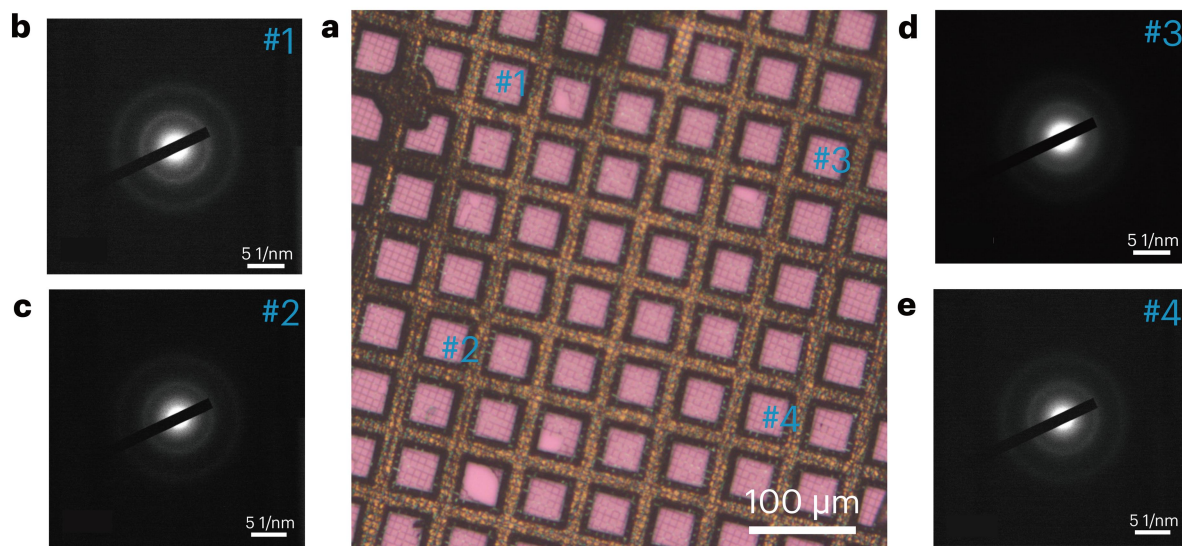
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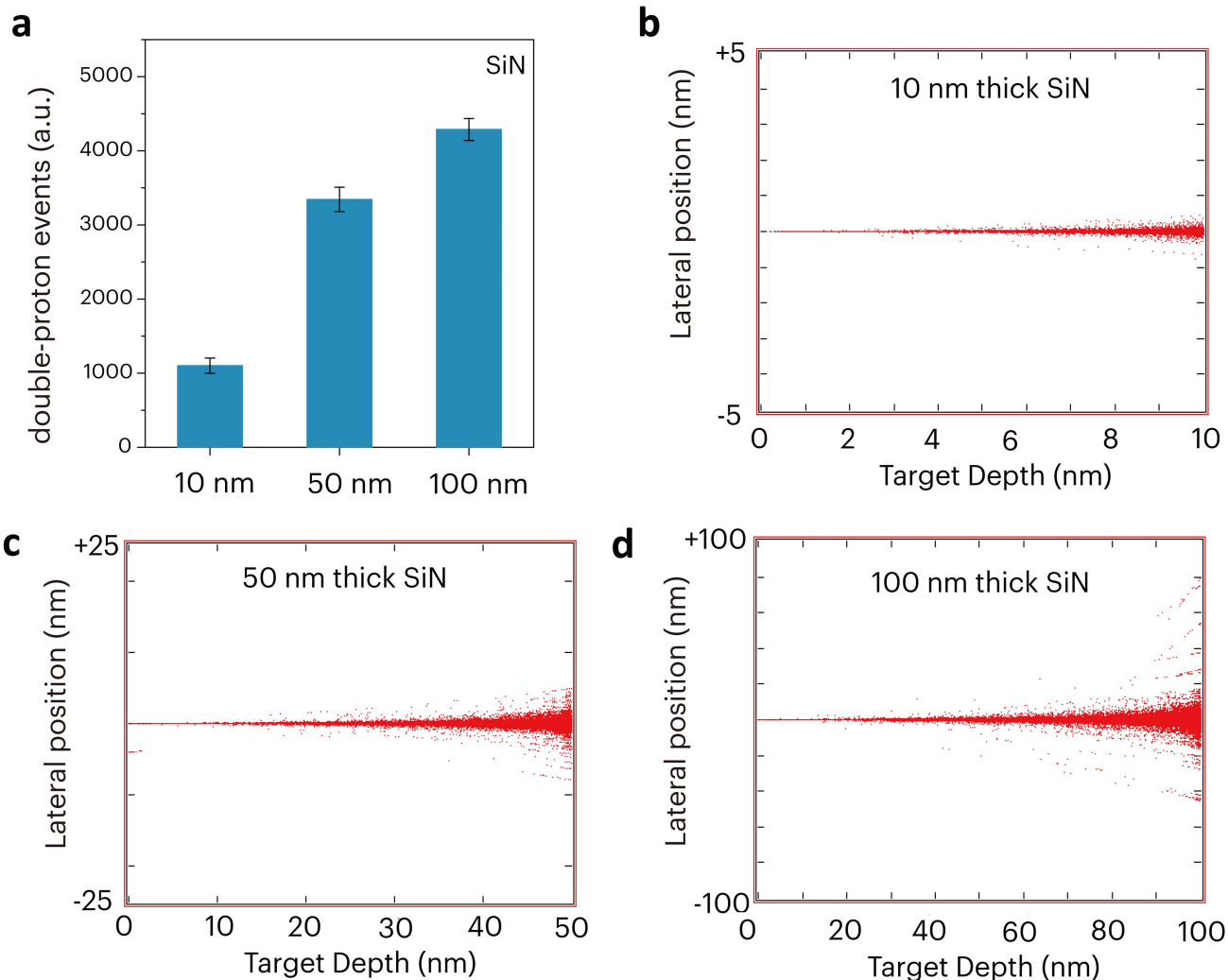
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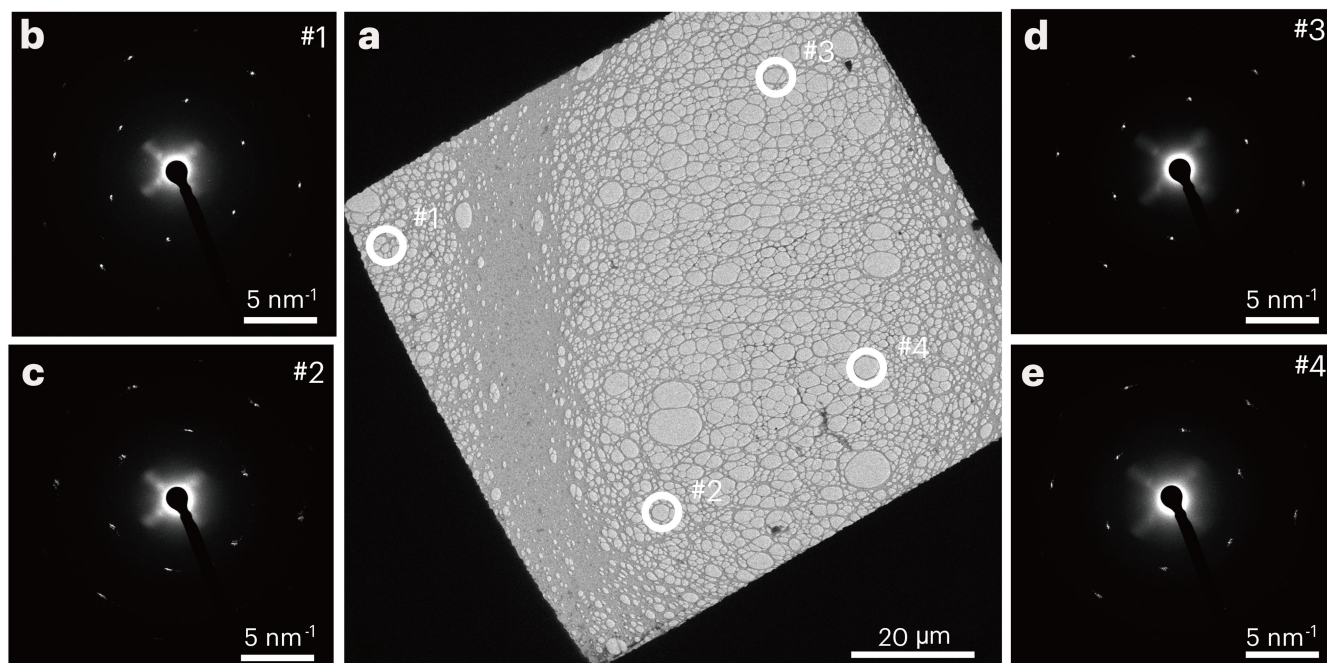
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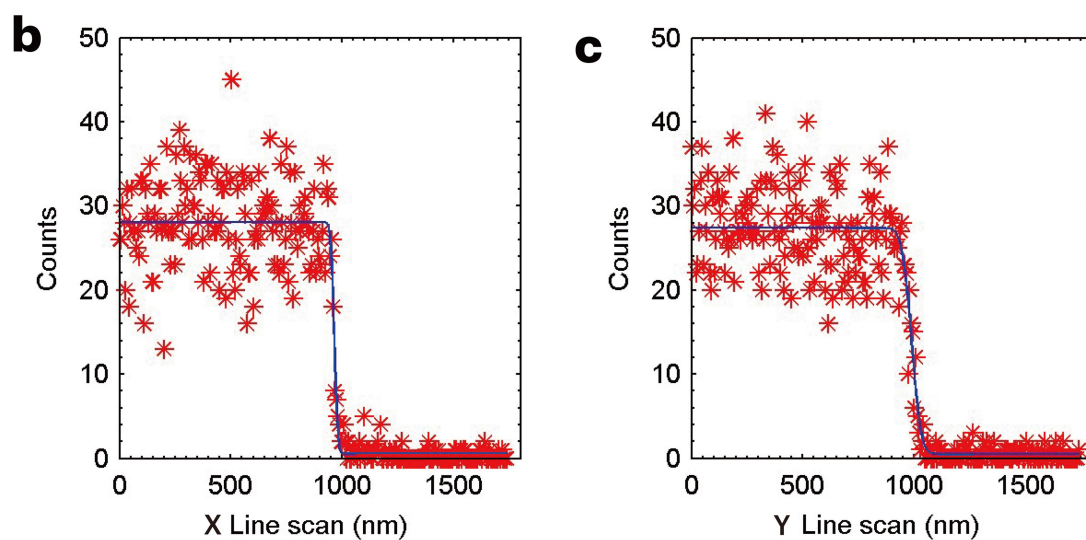
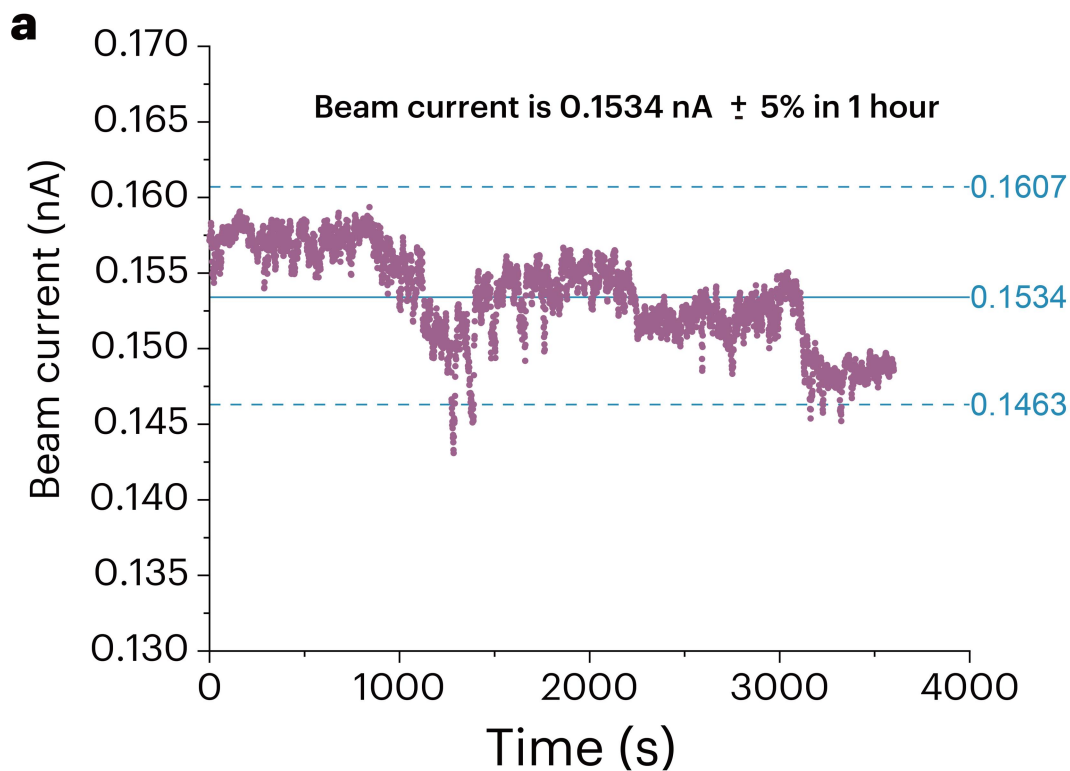
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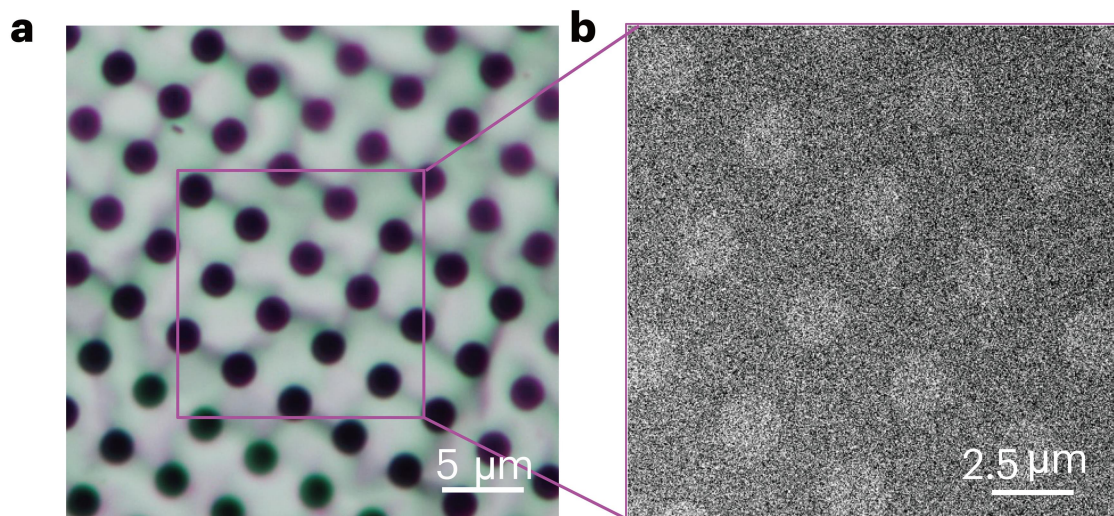
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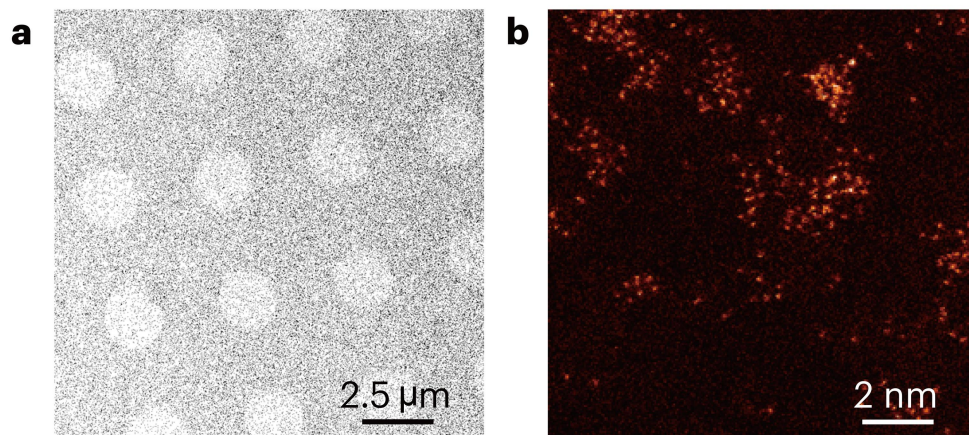
Supplementary Figure S18. TEM image of SCG grown from as-sputtered single crystal Cu(111) substrate. a-e, Low magnification TEM image of SCG (a) and the SAED (b-e) of the marked sites in (a) proved that the SCG has the same crystal orientation.



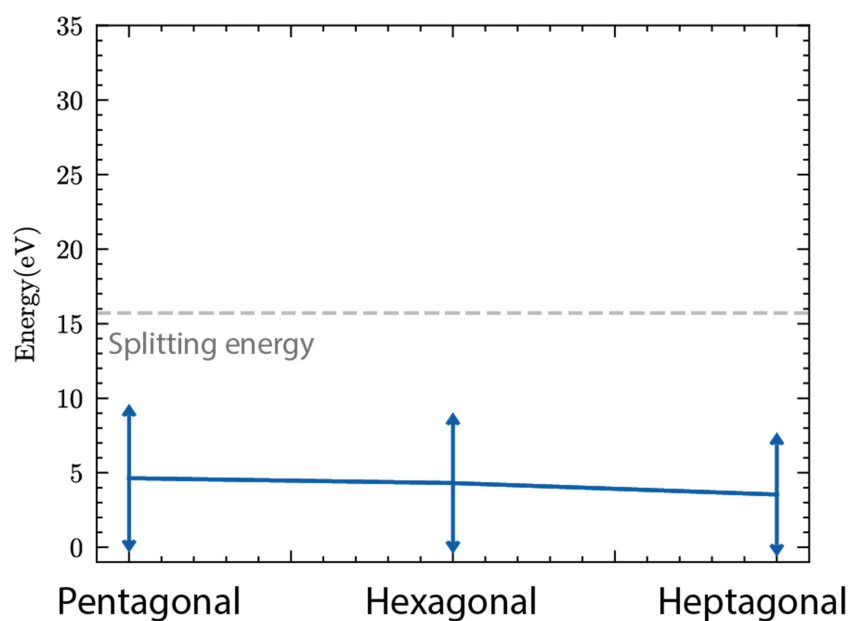
Supplementary Figure S19. Performance of the high energy ion nanobeam for splitting of hydrogen molecular ions. **a**, Typical long-term beam current stabilities data. **b**, **c**, Focused beam spot size (FWHM) of 28 nm \times 80 nm.



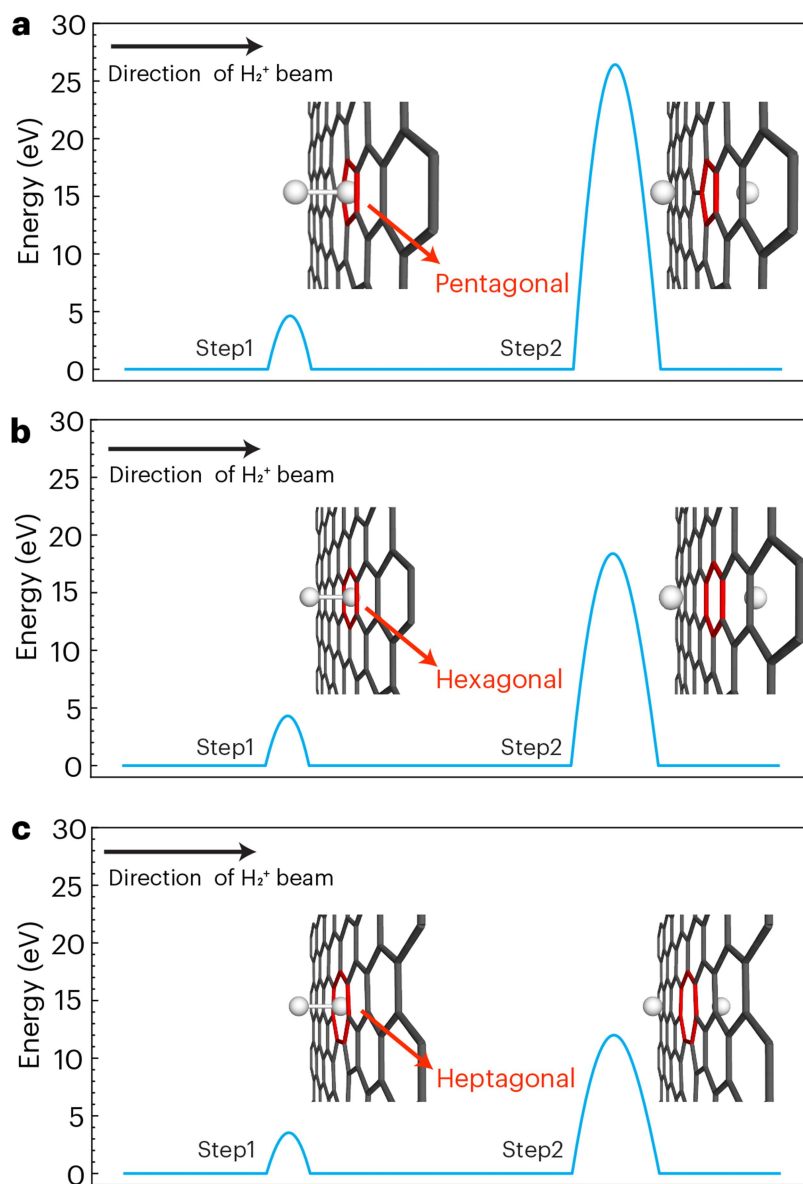
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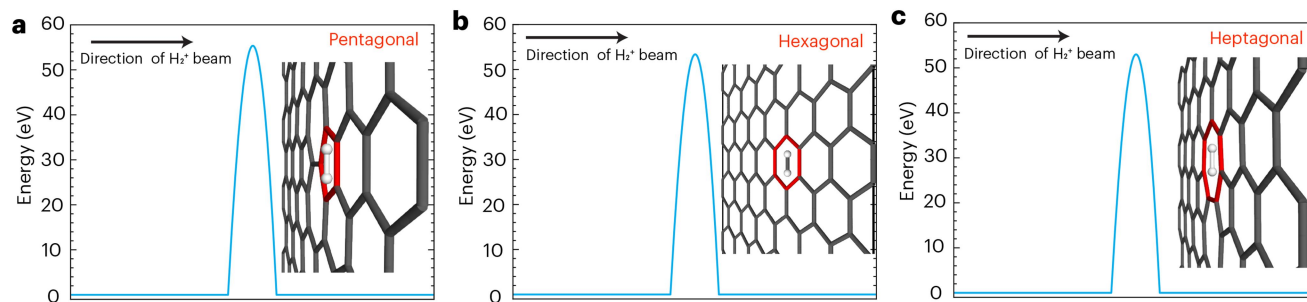
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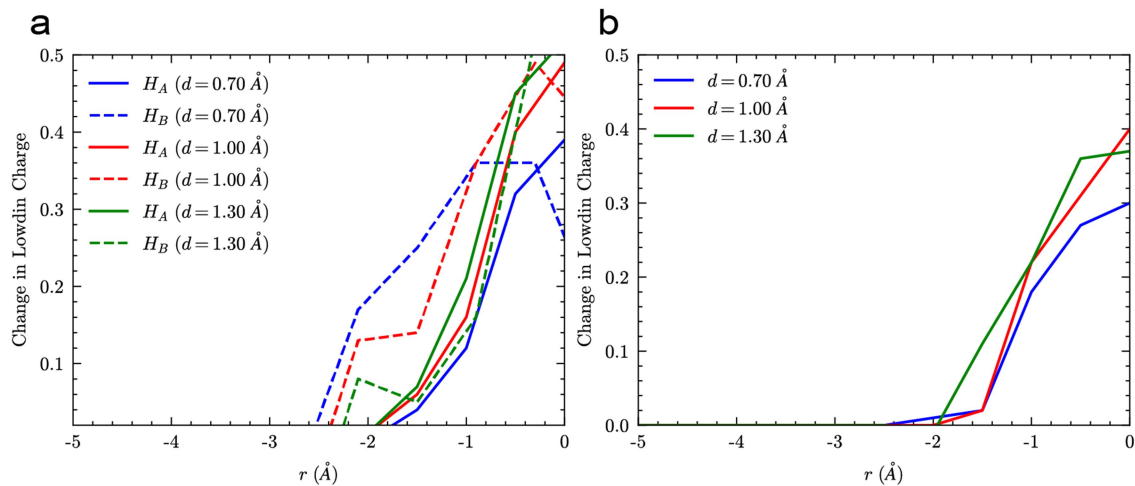
Supplementary Figure S22. Computed energy barrier for the first-step splitting of H_2^+ when the frontend H_A atom passes through either the hexagonal or non-hexagonal (pentagonal or heptagonal) ring. The initial H_2^+ ion being in the vertical configuration toward SCG/UC-MAC monolayer.



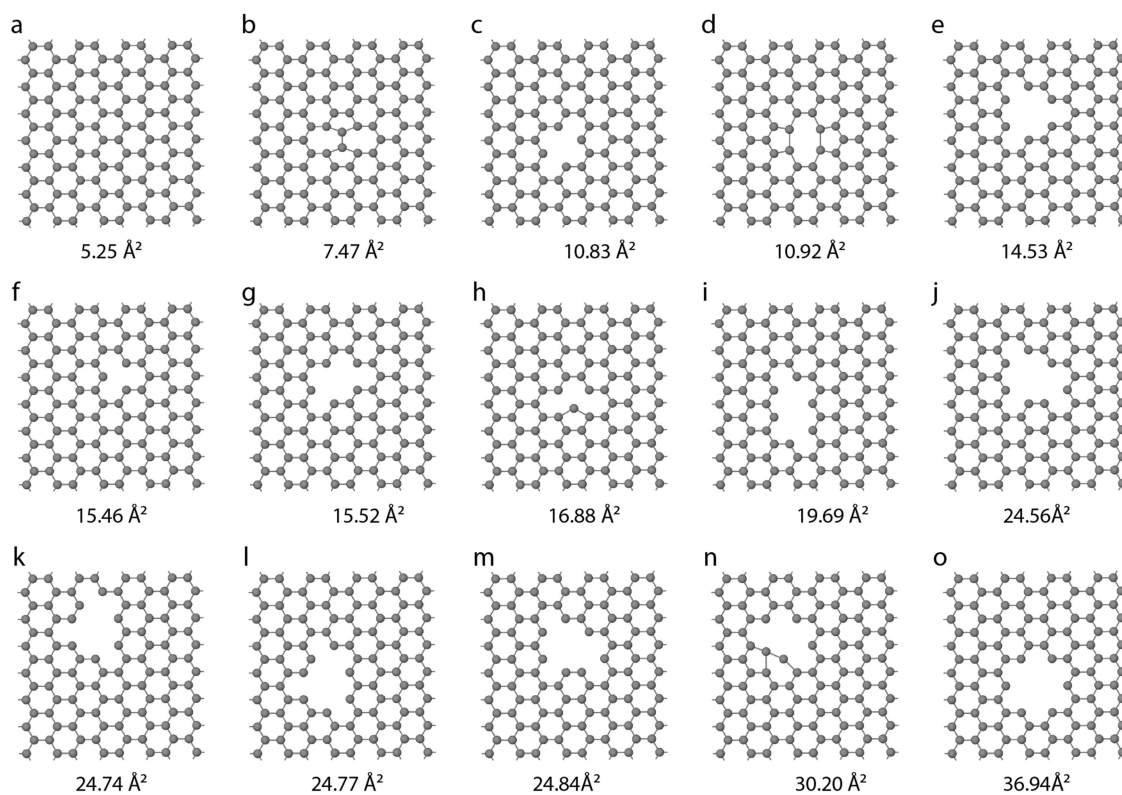
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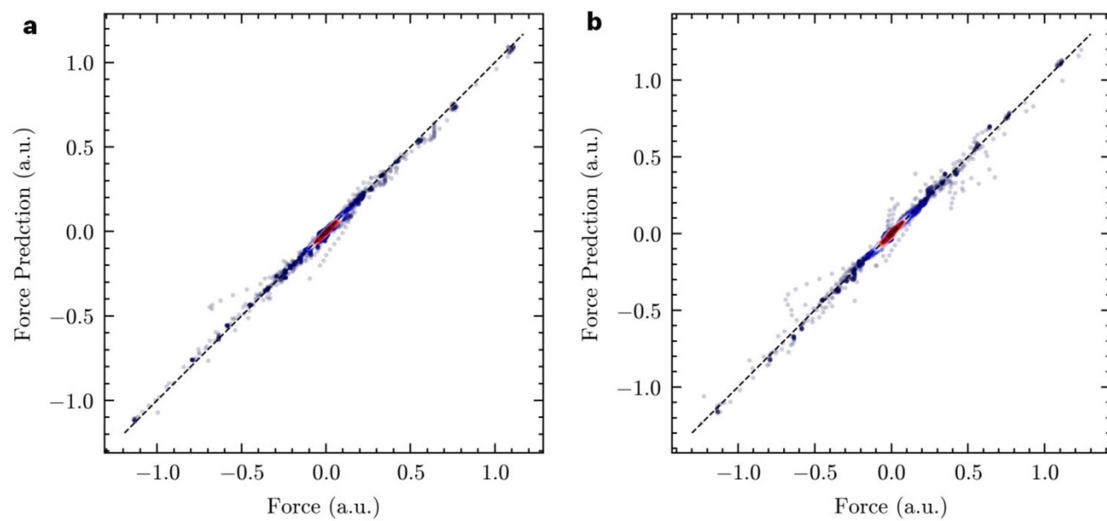
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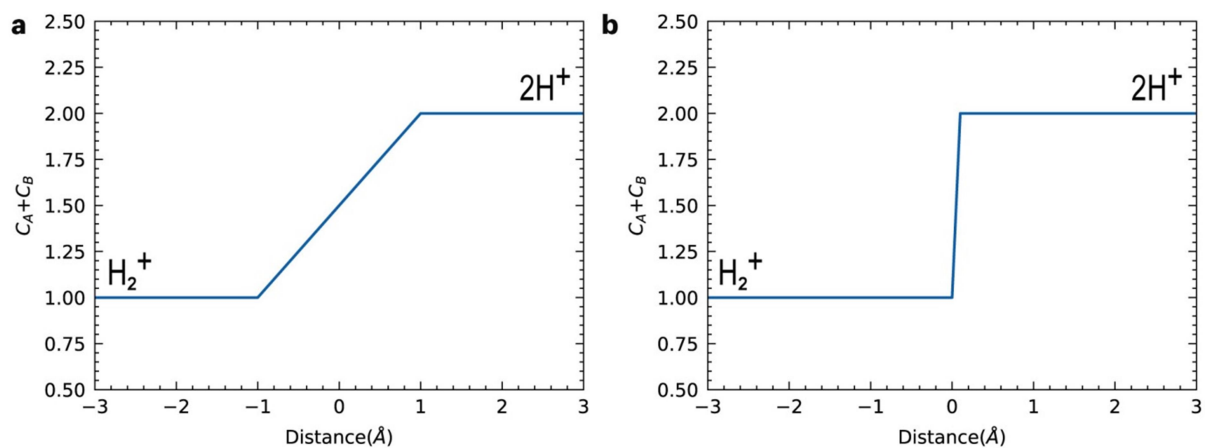
Supplementary Figure S25. Charge analysis of individual H in the H_2^+ ion when approaching to the graphene monolayer at different H-H bond distance (d), and in different initial H_2^+ configuration: (a) Vertical and (b) parallel. Here, r refers to the distance between the hydrogen atom and the graphene layer in (a), or between the center of mass of H_2^+ ion and the graphene layer in (b). The position of the graphene layer is at 0.



Supplementary Figure S26. Computed geometric area of various defect or pores (b-o) carved in atomic structure model of graphene.



Supplementary Figure S27. The training results of MLP based on the (a) linear and (b) binary step model of electron transfer.



Supplementary Figure S28. The charge transfer scheme based on the (a) linear and (b) binary step model.

Table 1. Sputtering parameters for different sputtered substrates.

Sputtered substrate	Thickness (nm)	Metal deposition	Substrate	Nanograins density
		rate (Å/s)	temperature (°C)	
Cu (111)	1000	0.5	800	0
Cu Disorder-Sub	50	0.1	200	$\sim 2260 \pm 20 \text{ /um}^2$
Cu substrate #1	50	0.1	200	$\sim 2260 \pm 20 \text{ /um}^2$
Cu substrate #2	150	0.5	500	$\sim 72 \pm 12 \text{ /um}^2$

Note: Cu substrate #1 and Cu Disorder-Sub were sputtered under the same conditions.

Table 2. Representative CVD preparation methods of monolayer graphene/amorphous carbon.

Monolayer carbon film	Preparation method	Carbon source	Growth temperature/ °C	Growth time/s	Refs.
Graphene	CVD	CH ₄	1000	1800	<i>Science</i> , 2009, 324, 1312-1314. ⁵
Graphene	CVD	C ₂ H ₄	~727	3600	<i>Nature</i> , 2021, 596, 519-524. ⁶
Graphene	ICP-CVD	CH ₄	750	300	<i>Nature</i> , 2014, 505, 190-194. ⁷
n-G	ICP-CVD	CH ₄	750 ~ 800	150	<i>2D Materials</i> , 2016, 3, 021001. ⁸
MAC	Laser-assisted depositions	CH ₄	200 ~ 500	60 ~ 600	<i>Nature</i> , 2020, 577, 199-203. ³
MAC	CVD	BN-doped C ₁₀ H ₆ Br ₂	300 ~ 500	300 ~ 1800	<i>Nature</i> , 2023, 615, 56-61. ⁴
UC-MAC	Disorder-to-disorder synthesis	CH ₄	820	3 ~ 20	This work

Table 3. Specific DTD growth parameters of the as-grown monolayer carbon films in our work.

Sample	Sputtered substrate	Power/W	Growth temperature/ °C	Growth time/s
SCG	Cu (111)	50	820	300
n-G ₁	Cu Disorder-Sub	20	820	60
n-G ₂	Cu Disorder-Sub	20	820	30
UC-MAC	Cu Disorder-Sub	100	820	3~20
n-G ₃	Cu substrate #1	50	820	30
n-G ₄	Cu substrate #2	50	820	30

Adjusting the reaction power and growth duration allows fine-tuning of the density of non-hexagonal rings in the film. With these optimized parameters, various monolayer carbon films can be produced, ranging from SCG to UC-MAC.

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