

Supporting Information

Boron-mediated One-pot Access to Salicylaldehydes via *ortho*-C–H hydroxylation of benzaldehydes

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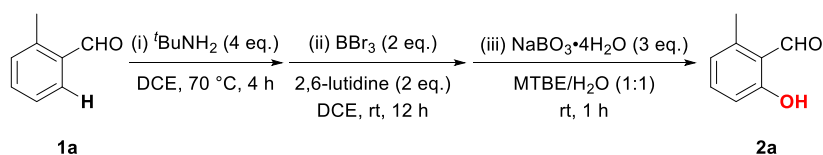
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I. General remarks

NMR spectra were prepared on an Agilent 400-MR DD2 spectrometer (^1H NMR at 400 MHz, ^{13}C NMR at 100 MHz, ^{19}F NMR at 376 MHz). The ^1H NMR (400 MHz) chemical shifts and the ^{13}C NMR (100 MHz) chemical shifts were measured relative to CDCl_3 as the internal reference. High resolution mass spectra (HRMS) were prepared with a Shimadzu LCMS-IT-TOF (ESI) or an Agilent 1260/6530 (ESI). GC-MS spectra were recorded by an Agilent 8860-5977B.

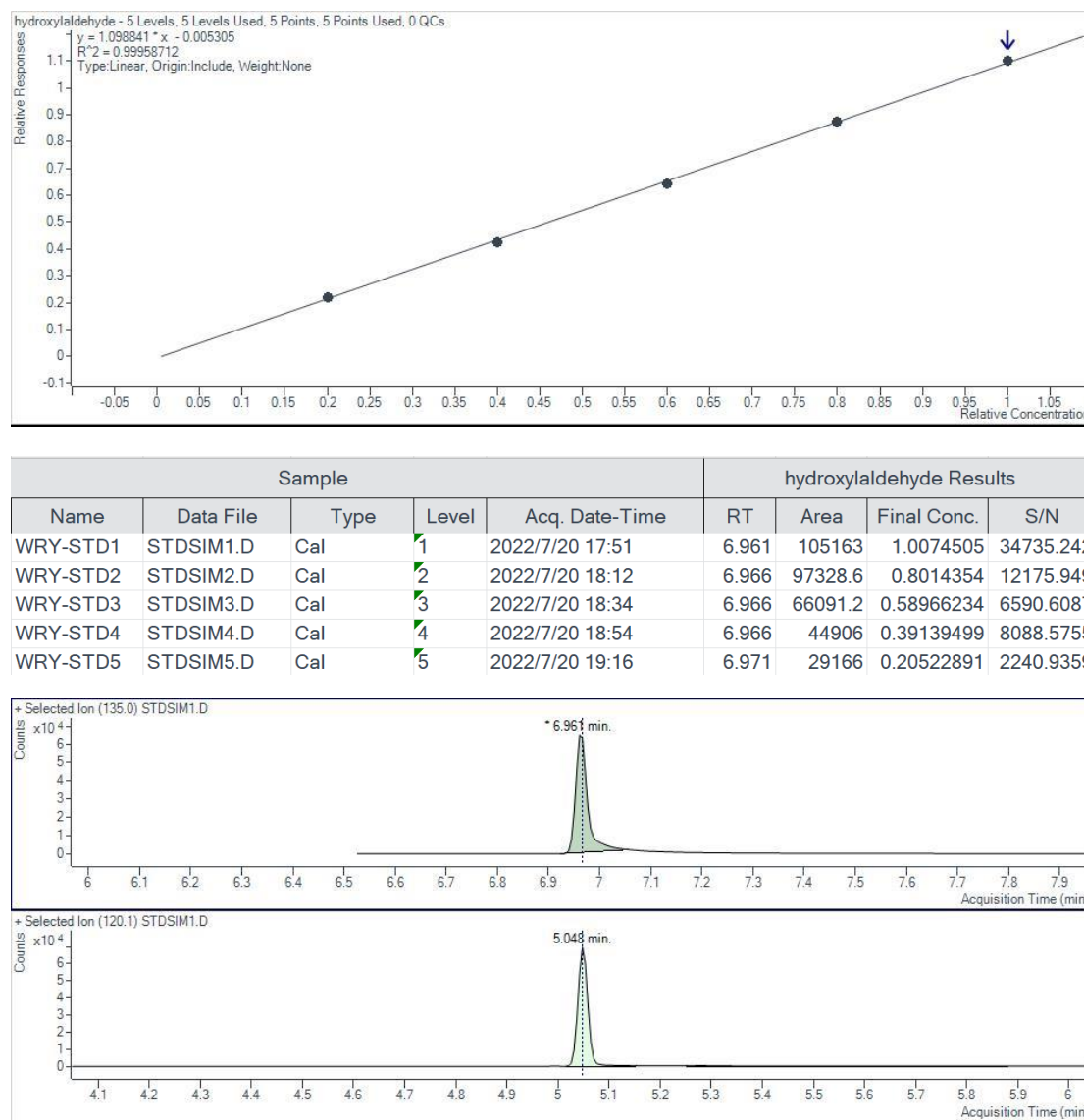
Unless otherwise noted, all reagents were obtained from commercial suppliers and used without further purification. DCE were distilled over CaCl_2 before use. $t\text{BuNH}_2$, benzaldehydes (**1a-1e**, **1g-1r**, **1t**), BBr_3 (2*N* in DCM), BBr_3 (neat) and CuI were purchased from Energy Chemical. $\text{NaBO}_3 \cdot 4\text{H}_2\text{O}$ and $2\text{Na}_2\text{CO}_3 \cdot 3\text{H}_2\text{O}_2$ was purechased from Aladdin Chemical. $\text{Pd}(\text{OAc})_2$ was purchased from Shanxi Kaida Chemical Engineering. 4-Benzylbenzaldehyde (**1f**)(1) and 9-methyl-9*H*-carbazole-2-carbaldehyde (**1r**)(2) were synthesized according to the literature. The yields of compound **2a** were determined by GC-MS analysis using calibration curves based on the data from authentic samples of the corresponding compound. For all GC-MS calibration curves, the ratio of molar concentration is taking as the horizontal axis and the ratio of GC area is taking as the vertical axis. Unless otherwise noted, all reactions were performed with dry solvents under argon in dried glassware with standard vacuum-line techniques.

II. General procedure for the optimization study



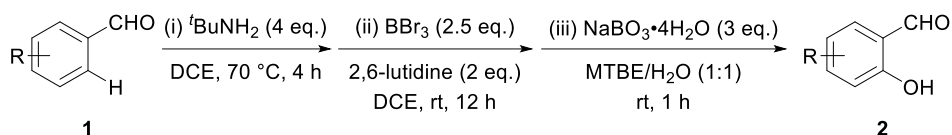
- (i) An oven-dried Schlenk tube equipped with a stirring bar was charged with 2-methylbenzaldehyde (48 mg, 0.4 mmol, 1 equiv), $t\text{BuNH}_2$ (168 μL , 1.6 mmol, 4 equiv) under argon. Then solvent DCE (1 mL) was added at room temperature. The reaction mixture was placed in a preheated oil bath at 70 °C, and stirred for 4 h.
- (ii) Next, $t\text{BuNH}_2$ and solvent was removed under reduced pressure. Base (0.8 mmol, 2.0 equiv), DCE (1 mL) and BBr_3 (0.8 mmol, 2 equiv) were added under argon. The reaction mixture was stirred at rt for 12h.
- (iii) Then, solvent was removed under reduced pressure, followed by the addition of

methyl *tert*-butyl ether (MTBE, 1 mL), sodium perborate tetrahydrate (185 mg, 1.2 mmol, 3.0 equiv), H₂O (1 mL) under argon. The reaction mixture was stirred at rt for another 1h. Afterwards, the reaction mixture was diluted with ethyl acetate (5 mL), mesitylene (55.6 μ L, 0.4 mmol) was subjected as internal standard. The yield of product was determined by GC-MS analysis using standard calibration curves (Scheme S1) based on data from the authentic sample of **2a** and mesitylene.



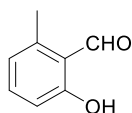
Scheme S1 Standard GC-MS calibration curves of **2a**

III. General procedure for the *ortho*-C–H hydroxylation



- (i) An oven-dried Schlenk tube equipped with a stirring bar was charged with benzaldehyde **1** (0.4 mmol, 1.0 equiv), $^t\text{BuNH}_2$ (168 μL , 4.0 equiv) under argon. Then solvent DCE (1 mL) was added at room temperature. The reaction mixture was placed in a preheated oil bath at 70 $^\circ\text{C}$, and stirred for 4 h.
- (ii) Next, $^t\text{BuNH}_2$ and solvent was removed under reduced pressure. 2,6-Lutidine (85.7 mg, 2.0 equiv), DCE (1 mL) and BBr_3 (0.5 mL, 2*N* in DCM, 2.5 equiv) were added under argon. The reaction mixture was stirred at rt for 12h.
- (iii) Then, solvent was removed under reduced pressure, followed by the addition of MTBE (1 mL), sodium perborate tetrahydrate (185 mg, 1.2 mmol, 3.0 equiv) and water (1 mL) under argon. The reaction mixture was stirred at rt for another 1h. Afterwards, the reaction mixture was extracted with ethyl acetate (10 mL $\times$ 3). The combined organic layer was further purified by column chromatography on silica gel (200-300 mesh) to afford the corresponding product **2**.

IV. Experimental data for the described substances



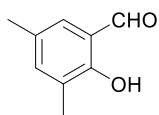
2-Hydroxy-6-methylbenzaldehyde (2a)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 50/1) on silica gel afforded compound **2a** as colorless oil (43 mg, 79% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 2.60 (s, 3H), 6.70 (d, J = 7.6 Hz, 1H), 6.80 (d, J = 8.4 Hz, 1H), 7.37 (t, J = 8.0 Hz, 1H), 10.31 (s, 1H), 11.91 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 17.4, 115.7, 118.6, 121.9, 137.5, 142.2, 163.3, 194.7 ppm.

The NMR spectrum data are consistent with the literature.⁽³⁾



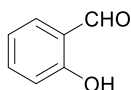
2-Hydroxy-3,5-dimethylbenzaldehyde (2b)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 50/1) on silica gel afforded compound **2b** as yellow oil (47 mg, 78% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 2.24 (s, 3H), 2.30 (s, 3H), 7.17 (s, 1H), 7.21 (s, 1H), 9.83 (s, 1H), 11.08 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 15.1, 20.4, 119.9, 126.7, 128.7, 131.1, 139.2, 158.1, 196.8 ppm.

The NMR spectrum data are consistent with the literature.(4)



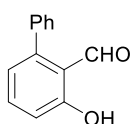
2-Hydroxybenzaldehyde (2c)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 50/1) on silica gel afforded compound **2c** as colorless oil (43 mg, 89% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 7.00 (dd, J = 16.8, 7.6 Hz, 2H), 7.53 (dd, J = 16.0, 7.6 Hz, 2H), 9.89 (s, 1H), 11.03 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 117.6, 119.9, 120.7, 133.8, 137.0, 161.6, 196.7 ppm.

The NMR spectrum data are consistent with the literature.(5)



3-Hydroxy-[1,1'-biphenyl]-2-carbaldehyde (2d)

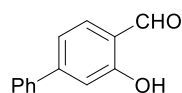
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde,

purification by column chromatography (petroleum ether/ethyl acetate = 50/1) on silica gel afforded compound **2d** as pale yellow oil (50 mg, 63% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 6.90 (dd, J = 7.6, 1.2 Hz, 1H), 7.00 (d, J = 8.4 Hz, 1H), 7.37 (m, 2H), 7.45 (m, 2H), 7.46 – 7.49 (m, 1H), 7.51 – 7.57 (t, J = 8.0 Hz, 1H), 9.84 (s, 1H), 11.93 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 117.1, 118.1, 121.7, 128.4, 128.6, 130.2, 136.8, 137.5, 147.6, 162.9, 197.3 ppm.

The NMR spectrum data are consistent with the literature.(3)



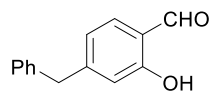
3-Hydroxy-[1,1'-biphenyl]-4-carbaldehyde (2e)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 20/1) on silica gel afforded compound **2e** as a white solid (34 mg, 43% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ 7.22 (s, 1H), 7.24 – 7.28 (m, 1H), 7.42 – 7.45 (m, 1H), 7.46 – 7.50 (m, 2H), 7.60 – 7.63 (m, 2H), 7.64 (t, J = 2.0 Hz, 1H), 9.92 (s, 1H), 11.13 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ 116.0, 119.0, 119.7, 127.5, 129.0, 129.1, 134.2, 139.4, 150.0, 162.0, 196.2 ppm.

The NMR spectrum data are consistent with the literature.(3)



4-Benzyl-2-hydroxybenzaldehyde (2f)

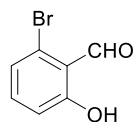
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 80/1) on silica gel afforded compound **2f** as orange oil (52 mg, 61% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 3.99 (s, 2H), 6.86 (m, 2H), 7.18 – 7.22 (m, 2H), 7.25 (t, J = 7.6 Hz, 1H), 7.28 – 7.35 (m, 2H), 7.46 (d, J = 8.0 Hz, 1H), 9.83 (s, 1H),

11.07 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 42.4, 117.8, 119.1, 120.9, 126.7, 128.8, 129.1, 133.9, 139.4, 151.8, 161.9, 196.1 ppm.

The ^1H NMR spectrum data are consistent with the literature.(6)



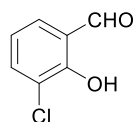
2-Bromo-6-hydroxybenzaldehyde (2g)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 30/1) on silica gel afforded compound **2g** as a yellow solid (56 mg, 70% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ 7.12 – 7.21 (m, 2H), 7.41 (d, J = 8.4 Hz, 1H), 9.86 (s, 1H), 11.12 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ 119.7, 121.2, 123.7, 131.4, 134.7, 162.6, 195.9 ppm.

The NMR spectrum data are consistent with the literature.(3)



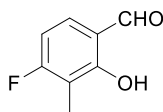
3-Chloro-2-hydroxybenzaldehyde (2h)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 12/1) on silica gel afforded compound **2h** as an orange solid (18 mg, 30% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 7.00 (t, J = 7.6 Hz, 1H), 7.51 (dd, J = 8.0, 1.6 Hz, 1H), 7.62 (dd, J = 8.0, 1.6 Hz, 1H), 9.90 (s, 1H), 11.49 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 120.4, 121.6, 122.4, 132.2, 137.0, 157.3, 196.2 ppm.

The NMR spectrum data are consistent with the literature.(7)



4-Fluoro-2-hydroxy-3-methylbenzaldehyde (2i)

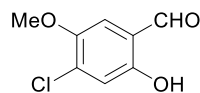
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 20/1) on silica gel afforded compound **2i** as a white solid (32 mg, 52% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 2.17 (d, J = 2.0 Hz, 3H), 6.71 (t, J = 8.8 Hz, 1H), 7.39 (dd, J = 8.8, 6.4 Hz, 1H), 9.81 (s, 1H), 11.61 (d, J = 2.0 Hz, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 7.0, 107.9 (d, $J_{\text{C-F}}$ = 25.3 Hz), 113.8 (d, $J_{\text{C-F}}$ = 24.2 Hz), 117.3 (d, $J_{\text{C-F}}$ = 2 Hz), 132.9 (d, $J_{\text{C-F}}$ = 13.1 Hz), 162.6 (d, $J_{\text{C-F}}$ = 12.1 Hz), 166.4 (d, $J_{\text{C-F}}$ = 256.5 Hz), 195.6 ppm.

^{19}F NMR (376 MHz, Chloroform-*d*) δ = -101.69 – -101.56 (m) ppm.

HRMS (ESI $^+$) calcd for $\text{C}_8\text{H}_8\text{FO}_2$ $[\text{M}+\text{H}]^+$ 155.0503, found 155.0505.



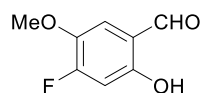
4-Chloro-2-hydroxy-3-methoxybenzaldehyde (2j)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 8/1) on silica gel afforded compound **2j** as a pale yellow solid (29 mg, 39% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 3.91 (s, 3H), 7.02 (s, 1H), 7.07 (s, 1H), 9.85 (s, 1H), 10.77 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 56.9, 114.6, 118.9, 119.9, 133.3, 148.9, 156.2, 195.3 ppm.

The NMR spectrum data are consistent with the literature.⁽⁸⁾



4-Fluoro-2-hydroxy-3-methoxybenzaldehyde (2k)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 5/1) on silica gel afforded compound **2k** as a brown solid (42 mg, 62% yield).

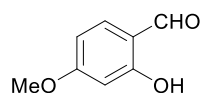
^1H NMR (400 MHz, Chloroform-*d*) δ = 3.89 (s, 3H), 6.73 (d, J = 12.0 Hz, 1H), 7.07 (d, J = 8.0 Hz, 1H), 9.79 (s, 1H), 11.07 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 57.1, 106.1 (d, $J_{\text{C-F}}$ = 21.7 Hz), 116.5 (d, $J_{\text{C-F}}$ = 2.5 Hz), 116.8 (d, J = 5.3 Hz, 1H), 142.0 (d, $J_{\text{C-F}}$ = 12.1 Hz), 158.1 (d, $J_{\text{C-F}}$ = 12.8 Hz), 158.7 (d, $J_{\text{C-F}}$ = 261.6 Hz), 194.9 ppm.

^{19}F NMR (376 MHz, Chloroform-*d*) δ = -117.92 (dd, J = 11.9, 9.2 Hz) ppm.

The NMR spectrum data are consistent with the literature.(3)

HRMS (ESI⁺) calcd for C₈H₇FO₃ [M+Na]⁺ 193.0271, found 193.0281.



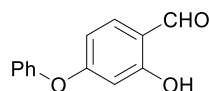
2-Hydroxy-4-methoxybenzaldehyde (2l)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 20/1) on silica gel afforded compound **2l** as a yellow solid (18 mg, 30% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 3.86 (s, 3H), 6.43 (d, J = 2.4 Hz, 1H), 6.54 (dd, J = 8.4, 2.4 Hz, 1H), 7.43 (d, J = 8.8 Hz, 1H), 9.71 (s, 1H), 11.48 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 55.8, 100.1, 108.5, 114.4, 135.4, 164.7, 167.0, 194.5 ppm.

The NMR spectrum data are consistent with the literature.(3)



2-Hydroxy-4-phenoxybenzaldehyde (2m)

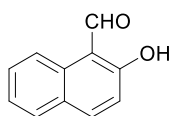
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde,

purification by column chromatography (petroleum ether/ethyl acetate = 30/1) on silica gel afforded compound **2m** as brown oil (40 mg, 47% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 6.41 (d, J = 2.4 Hz, 1H), 6.61 (dd, J = 8.4, 2.4 Hz, 1H), 7.10 (d, J = 7.6 Hz, 2H), 7.25 (t, J = 7.6 Hz, 1H), 7.39 – 7.45 (m, 2H), 7.47 (d, J = 8.4 Hz, 1H), 9.76 (s, 1H), 11.38 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 104.5, 109.8, 109.9, 116.3, 121.0, 125.5, 130.3, 135.7, 154.4, 164.2, 165.7, 194.8 ppm.

The NMR spectrum data are consistent with the literature.(3)



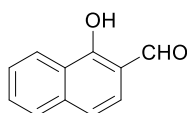
2-Hydroxy-1-naphthaldehyde (2n)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 15/1) on silica gel afforded compound **2n** as a yellow solid (60 mg, 88% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 7.19 (d, J = 7.6 Hz, 1H), 7.42 (d, J = 7.6 Hz, 1H), 7.48 – 7.60 (m, 2H), 8.04 (d, J = 7.2 Hz, 1H), 8.14 (d, J = 8.4 Hz, 1H), 9.86 (s, 1H), 11.67 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 116.1, 120.6, 121.3, 124.4, 129.2, 132.5, 136.4, 139.3, 143.1, 155.3, 198.0 ppm.

The NMR spectrum data are consistent with the literature.(9)



1-Hydroxy-2-naphthaldehyde (2o)

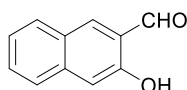
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 30/1) on silica gel afforded compound **2o** as a dark green solid (30 mg, 42% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ 7.38 (d, J = 8.4 Hz, 1H), 7.50 (d, J = 8.8 Hz, 1H), 7.52 – 7.59 (m, 1H), 7.67 (t, J = 7.6 Hz, 1H), 7.79 (d, J = 8.0 Hz, 1H), 8.45 (d, J = 8.4

Hz, 1H), 9.98 (s, 1H), 12.66 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ 114.0, 119.6, 124.5, 124.7, 126.3, 126.6, 127.8, 130.8, 137.7, 162.1, 196.5 ppm.

The NMR spectrum data are consistent with the literature.(9)



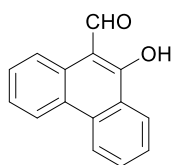
3-Hydroxy-2-naphthaldehyde(2o')

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 30/1) on silica gel afforded compound **2o'** as a yellow solid (21 mg, 30% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 7.28 (s, 1H), 7.38 (t, J = 7.2 Hz, 1H), 7.57 (t, J = 6.8 Hz, 1H), 7.72 (d, J = 8.4 Hz, 1H), 7.87 (d, J = 8.0 Hz, 1H), 8.15 (s, 1H), 10.09 (s, 1H), 10.32 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 112.1, 122.5, 124.6, 126.8, 127.6, 129.5, 130.4, 138.0, 138.4, 156.0, 196.8 ppm.

The NMR spectrum data are consistent with the literature.(3)



10-Hydroxyphenanthrene-9-carbaldehyde (2p)

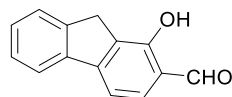
According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 10/1) on silica gel afforded compound **2p** as a yellow solid (50 mg, 56% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 7.23 (d, J = 8.0 Hz, 1H), 7.64 (m, 2H), 7.82 (t, J = 7.6 Hz, 1H), 7.97 (d, J = 8.0 Hz, 1H), 8.21 (d, J = 8.4 Hz, 1H), 8.32 (s, 1H), 8.63 (d, J = 8.4 Hz, 1H), 9.88 (s, 1H), 11.65 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 114.6, 116.8, 117.4, 123.9, 127.8, 129.4, 129.7,

130.7, 131.8, 131.9, 133.3, 134.6, 147.8, 156.3, 197.8 ppm.

The NMR spectrum data are consistent with the literature.(10)



3-Hydroxy-9H-fluorene-2-carbaldehyde (2q)

According to the general one-pot procedure for *ortho*-hydroxylation of benzaldehyde, purification by column chromatography (petroleum ether/ethyl acetate = 10/1) on silica gel afforded compound **2q** as a white solid (55 mg, 66% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 3.90 (s, 2H), 7.39 – 7.48 (m, 3H), 7.57 (d, J = 8.0 Hz, 1H), 7.59 – 7.64 (m, 1H), 7.79 – 7.86 (m, 1H), 9.91 (s, 1H), 11.35 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 33.8, 112.1, 119.4, 121.4, 125.6, 127.1, 128.9, 130.4, 134.8, 140.4, 145.1, 150.9, 158.3, 196.3 ppm.

HRMS (ESI $^+$) calcd for $\text{C}_{14}\text{H}_{11}\text{O}_2$ $[\text{M}+\text{H}]^+$ 211.0754, found 211.0753.

The structure of **2q** was further confirmed by HMBC (for fragment see Figure S1, for the full spectrum see Page S37).

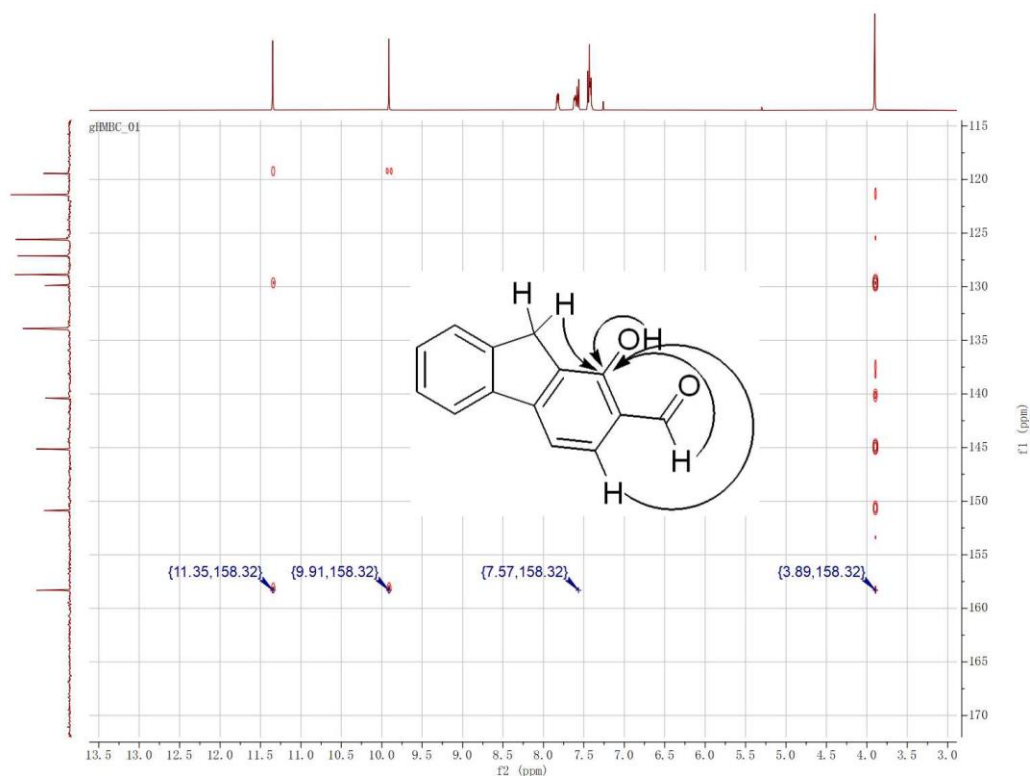
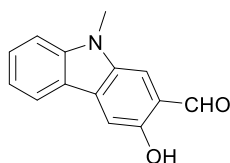


Figure S1 Fragment of HMBC spectrum of **2q**



2-Hydroxy-9-methyl-9H-carbazole-2-carbaldehyde (2r)

According to the general procedure, purification by column chromatography (petroleum ether/ethyl acetate = 10/1) on silica gel afforded compound **2r** as a yellow solid (31 mg, 34% yield).

^1H NMR (400 MHz, Chloroform-*d*) δ = 3.81 (s, 3H), 7.21 (t, J = 8.0 Hz, 1H), 7.34 (d, J = 8.4 Hz, 1H), 7.43 (s, 1H), 7.55 (ddt, J = 8.4, 7.2, 1.2 Hz, 1H), 7.58 (s, 1H), 8.05 (d, J = 8 Hz, 1H), 9.99 (s, 1H), 10.76 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 29.4, 107.0, 109.0, 113.6, 118.9, 119.2, 121.5, 122.1, 128.7, 131.2, 135.7, 144.5, 154.9, 196.1 ppm.

HRMS (ESI $^+$) calcd for $\text{C}_{14}\text{H}_{11}\text{NO}_2$ $[\text{M}+\text{H}]^+$ 226.0863, found 193.0858.

The structure of **2r** was further confirmed by HMBC (for fragment see Figure S2, for the full spectrum see Page S38).

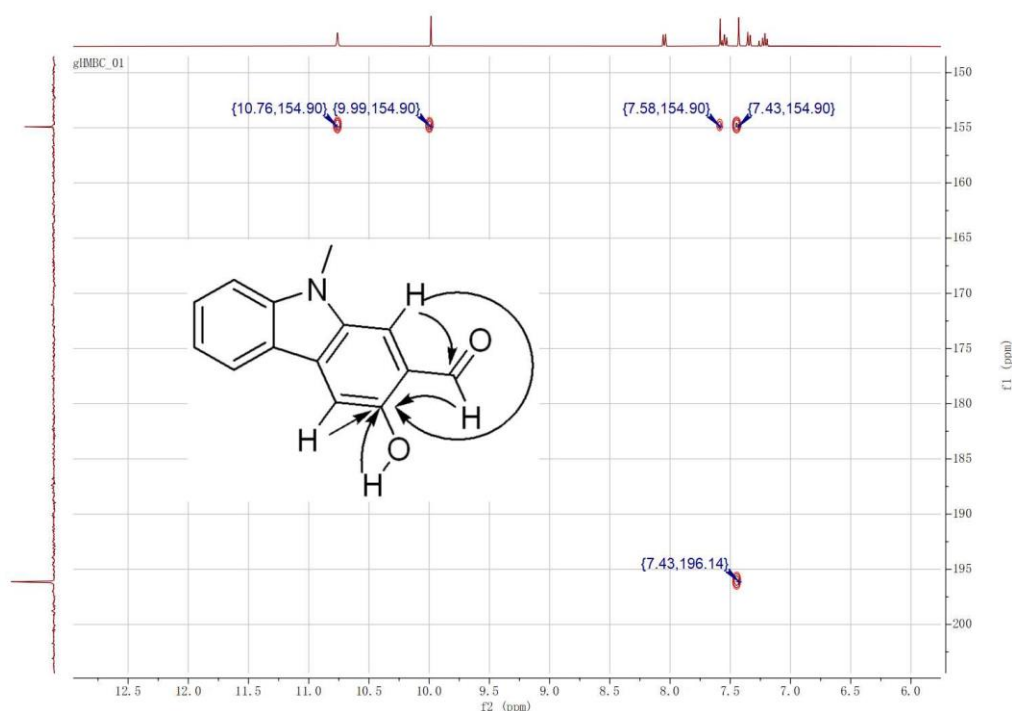
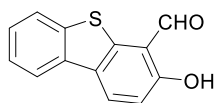


Figure S2 Fragment of HMBC spectrum of **2r**



3-Hydroxydibenzo[*b,d*]thiophene-4-carbaldehyde (**2s**)

According to the general procedure, purification by column chromatography (petroleum ether/ethyl acetate = 20/1) on silica gel afforded compound **2s** as a brown solid (34 mg, 37% yield).

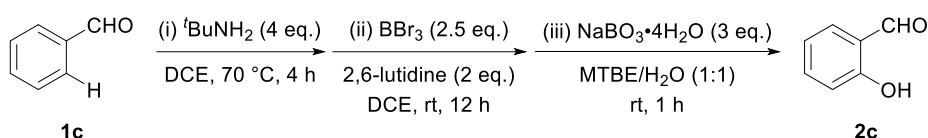
^1H NMR (400 MHz, Chloroform-*d*) δ = 7.06 (d, J = 8.8 Hz, 1H), 7.42 – 7.51 (m, 2H), 7.85 (d, J = 6.8 Hz, 1H), 8.01 (d, J = 8.0 Hz, 1H), 8.18 (d, J = 8.8 Hz, 1H), 10.34 (s, 1H), 11.57 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 114.5, 115.9, 120.9, 122.9, 125.3, 126.4, 129.3, 129.8, 134.7, 137.6, 143.7, 163.1, 192.9 ppm.

HRMS (ESI⁺) calcd for C₁₃H₈O₂S [M+H]⁺ 229.0318, found 229.0321.

V. Scale-up reaction and downstream synthesis

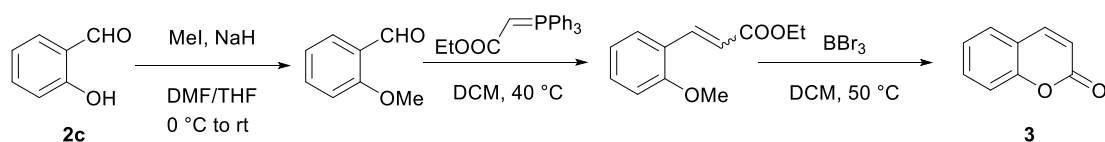
(1) Scale-up synthesis of **2c**



- (iv) An oven-dried 100 mL Schlenk flask equipped with a stirring bar was charged with benzaldehyde **1** (8 mmol, 1.0 equiv), *t*BuNH₂ (3.4 mL, 4.0 equiv) under argon. Then solvent DCE (20 mL) was added at room temperature. The reaction mixture was placed in a preheated oil bath at 70 °C, and stirred for 4 h.
- (v) Next, *t*BuNH₂ and solvent was removed under reduced pressure. 2,6-Lutidine (1.72 g, 2.0 equiv), then a mixture of DCE (20 mL) and BBr₃ (10 mL, 2*N* in DCM, 2.5 equiv) were added dropwisely under Ar. The reaction mixture was stirred at rt for 12h.
- (vi) Then, solvent was removed under reduced pressure, followed by the addition of MTBE (20 mL), sodium perborate tetrahydrate (3.7 g, 16 mmol, 2.0 equiv) and water (1 mL) under argon. The reaction mixture was stirred at rt for another 1

h. Afterwards, the reaction mixture was extracted with ethyl acetate (100 mL×3). The combined organic layer was further purified by column chromatography on silica gel (petroleum ether/ethyl acetate = 50/1) to afford the corresponding product **2c**.

(2) Synthesis of 2*H*-chromen-2-one (**3**)



A modified literature procedure was used for the synthesis of **3**:⁽³⁾

To a 100 mL flask were added NaH (0.12 g, 5.0 mmol), DMF (8.0 mL), and THF (2.0 mL), then cooled to 0 °C, followed by dropwise addition of a mixture of **2c** (0.61 g, 5.0 mmol) and MeI (2.13 g, 15.0 mmol) in THF (5.0 mL). The mixture was stirred for 2 h at 0 °C, and then the reaction was allow to warm to rt and stir for another 3 h. The reaction was quenched with H₂O and extracted with EA (3 × 20 mL). The combined organic phase was washed with brine, dried over Na₂SO₄ and purified by flash chromatography on silica gel (petroleum ether/ethyl acetate = 20/1) to yield the methylated product *o*-anisaldehyde as colorless oil.

The crude *o*-anisaldehyde was dissolved in CH₂Cl₂ (10 mL), Ph₃P=CHCOOEt (2.1 g, 6 mmol) was added. The mixture was stirred at 40 °C for 4 h. The reaction was quenched with H₂O and extracted with CH₂Cl₂ (3 × 15 mL). Purification by flash chromatography on silica gel (petroleum ether/ethyl acetate = 12/1) gave crude ethyl 3-(2-methoxyphenyl)acrylate as colorless oil.

The crude ethyl 3-(2-methoxyphenyl) acrylate was directly transferred to a 100 mL Schlenk tube under Ar, followed by the addition of CH₂Cl₂ (15.0 mL) and BBr₃ (4 mmol, 1*N* in CH₂Cl₂). The mixture was stirred at 50 °C for 4 h. After cooling to rt, the reaction was quenched with H₂O (20 mL), extracted with EA (3×20 mL), washed with brine, and dried over Na₂SO₄. Purification by column chromatography on silica gel (petroleum ether/ethyl acetate = 5/1) gave **3** (446 mg, 61% overall yield) as a white solid.

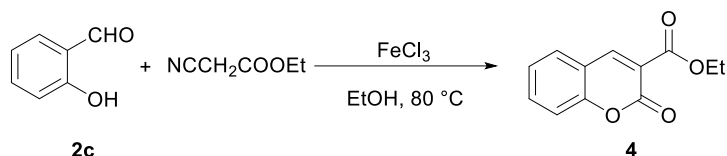
¹H NMR (400 MHz, Chloroform-*d*) δ = 6.42 (d, *J* = 9.6 Hz, 1H), 7.25 – 7.33 (m, 2H), 7.44 – 7.56 (m, 2H), 7.71 (d, *J* = 9.6 Hz, 1H) ppm.

¹³C NMR (101 MHz, Chloroform-*d*) δ = 116.8, 117.0, 118.9, 124.5, 128.0, 131.9, 143.6,

154.1, 160.9 ppm.

The NMR spectrum data are consistent with the literature.(3)

(3) Synthesis of ethyl 2-oxo-2H-chromene-3-carboxylate (4)



A modified literature procedure was used for the synthesis of **4**:(11)

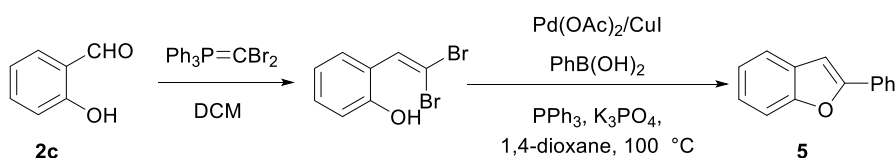
To a stirred solution of salicylaldehyde **2c** (49 mg, 0.4 mmol) and ethyl cyanoacetate (68 mg, 0.6 mmol) in ethanol (2 mL) was added anhydrous FeCl_3 (5 mg, 0.03 mmol). The mixture was heated in an oil bath at 80°C for 24 h and cooled down to room temperature. The reaction was quenched with H_2O (5 mL), extracted with ethyl acetate (3×5 mL), washed with brine, and dried over Na_2SO_4 . Purification by column chromatography on silica gel (petroleum ether/ethyl acetate = 8/1) gave **4** (47.9 mg, 55%) as a white solid.

^1H NMR (400 MHz, Chloroform-*d*) δ = 1.41 (t, J = 6.8 Hz, 3H), 4.41 (q, J = 7.2 Hz, 2H), 7.30 – 7.38 (m, 2H), 7.59 – 7.68 (m, 2H), 8.53 (s, 1H) ppm.

^{13}C NMR (101 MHz, Chloroform-*d*) δ = 14.4, 61.7, 116.6, 118.0, 118.4, 125.7, 128.6, 134.5, 148.2, 154.9, 156.7, 163.2 ppm.

The NMR spectrum data are consistent with the literature.(11)

(4) Synthesis of ethyl 2-oxo-2H-chromene-3-carboxylate (5)



A modified literature procedure was used for the synthesis of **5**:(3)

To a 25 mL flask were added PPh_3 (1.57 g, 6 mmol) and CH_2Cl_2 (6 mL). The mixture was cooled to 0°C , after which CBr_4 (1 g, 3 mmol) dissolved in CH_2Cl_2 (3 mL) was added and stirred for 10 min. NEt_3 (0.61 g, 6 mmol) was added dropwise and stirred for

an additional 5 min, after which **2c** (122 mg, 1 mmol, 1 equiv) in CH₂Cl₂ (1 mL) was added dropwise. The mixture was stirred for another 30 min at 0 °C, then warmed to rt and stirred for 2 h. The reaction was quenched by saturated aqueous NH₄Cl solution (10 mL), extracted with CH₂Cl₂ (3×10 mL). The combined organic layers were concentrated and purified by flash chromatography on silica gel (petroleum ether/ethyl acetate = 8/1) to give the crude 2-(2,2-dibromovinyl)phenol as a white solid.

The crude 2-(2,2-dibromovinyl)phenol was transferred to a Schlenk tube, followed by the addition of PhB(OH)₂ (244 mg, 2 mmol), Pd(OAc)₂ (11.3 mg, 5.0 mol %), CuI (9.5 mg, 5.0 mol %), PPh₃ (39.3 mg, 15 mol %), K₃PO₄ (254 mg, 1.2 mmol), and 1,4-dioxane (10 mL) under Ar. The mixture was stirred at 100 °C for 24 h, extracted with EtOAc (10 mL), washed with brine, and dried over Na₂SO₄. Purification by column chromatography on silica gel (petroleum ether) gave **5** (126 mg, 65% overall yield) as a white solid.

¹H NMR (400 MHz, Chloroform-*d*) δ = 7.05 (s, 1H), 7.25 (t, *J* = 7.6 Hz, 1H), 7.28 – 7.33 (m, 1H), 7.37 (t, *J* = 7.2 Hz, 1H), 7.47 (t, *J* = 7.6 Hz, 2H), 7.55 (d, *J* = 7.6 Hz, 1H), 7.60 (d, *J* = 7.2 Hz, 1H), 7.89 (d, *J* = 6.8 Hz, 2H) ppm.

¹³C NMR (101 MHz, Chloroform-*d*) δ = 101.0, 111.3, 121.0, 123.1, 124.4, 125.1, 127.6, 128.9, 129.3, 130.6, 155.0, 156.0 ppm.

The NMR spectrum data are consistent with the literature.(3)

VI. References

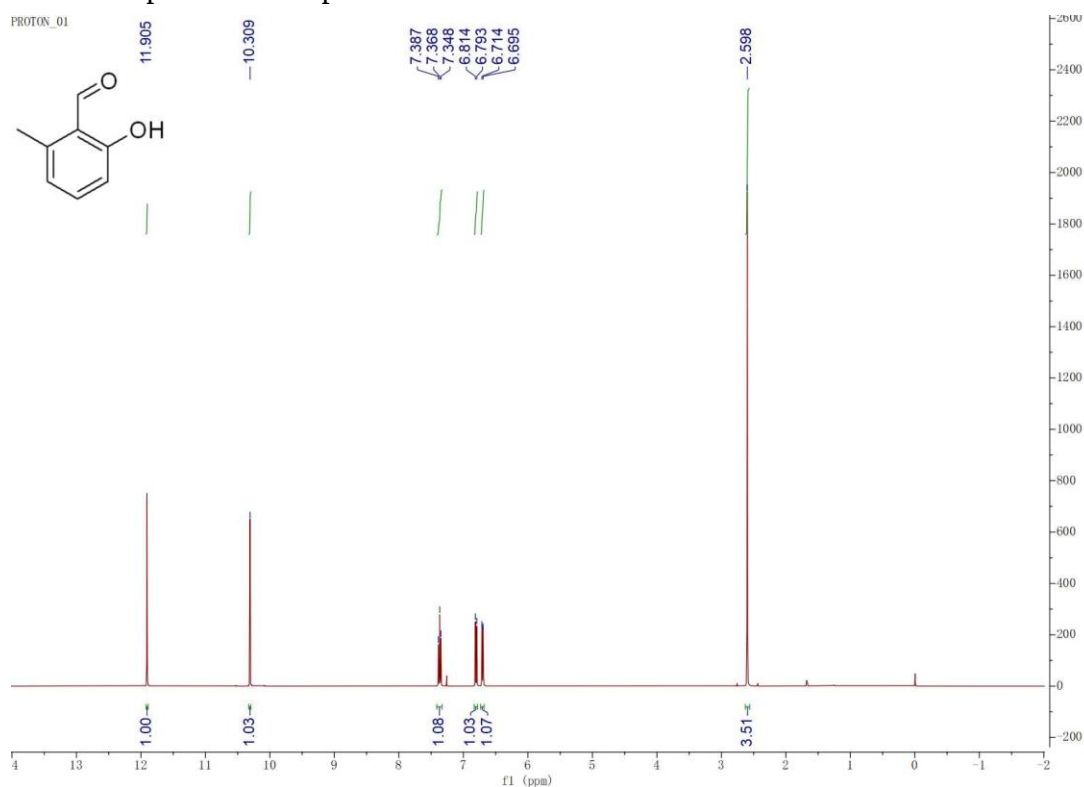
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Using a Transient Directing Group. *Org Lett.* 2017;19(23):6280-3.

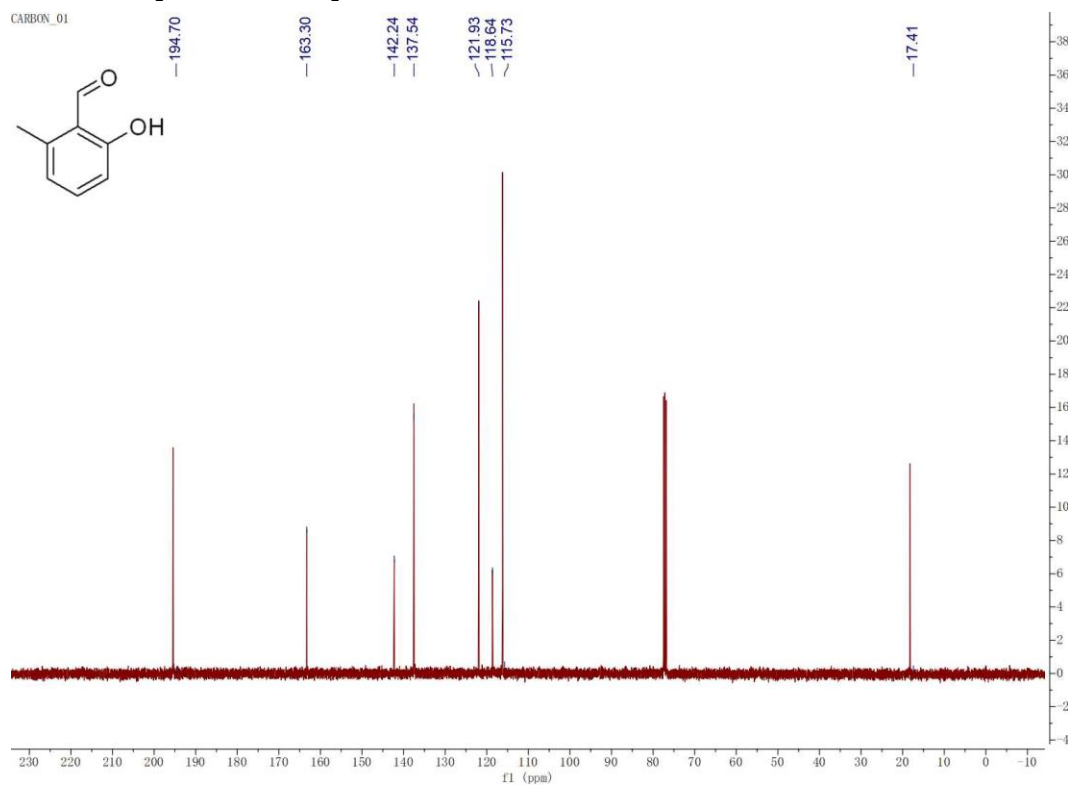
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VII. Copies of NMR spectra

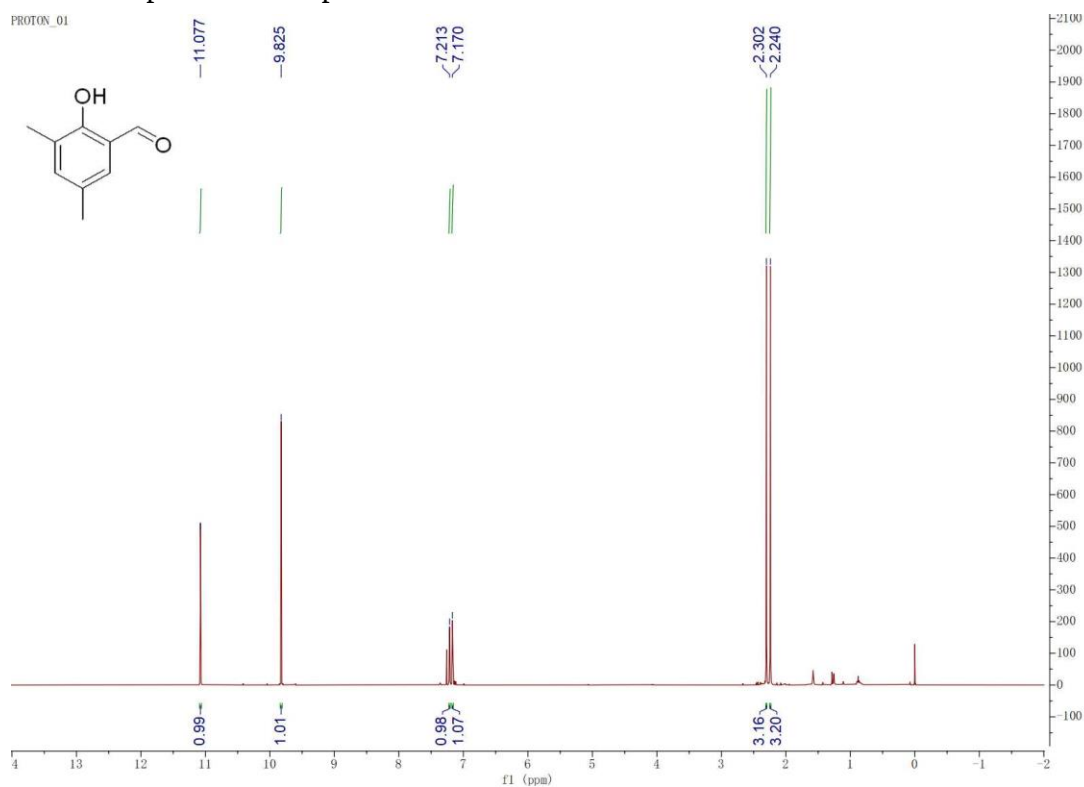
¹H NMR spectra of compound **2a**:



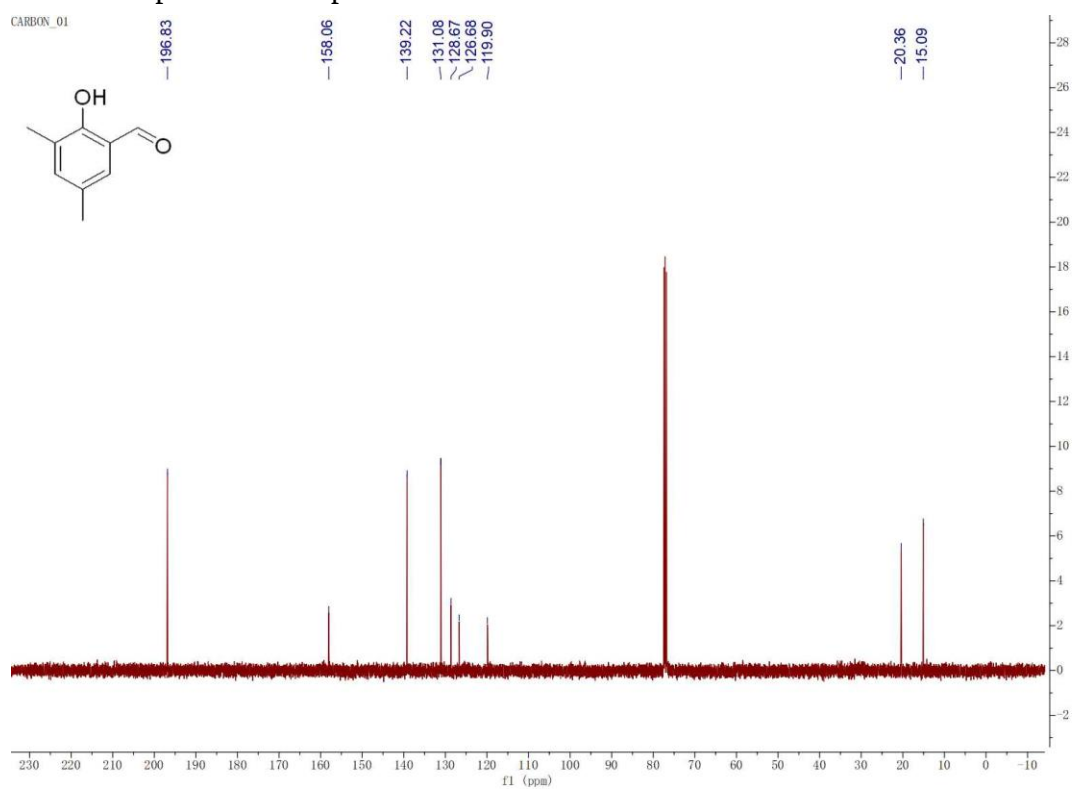
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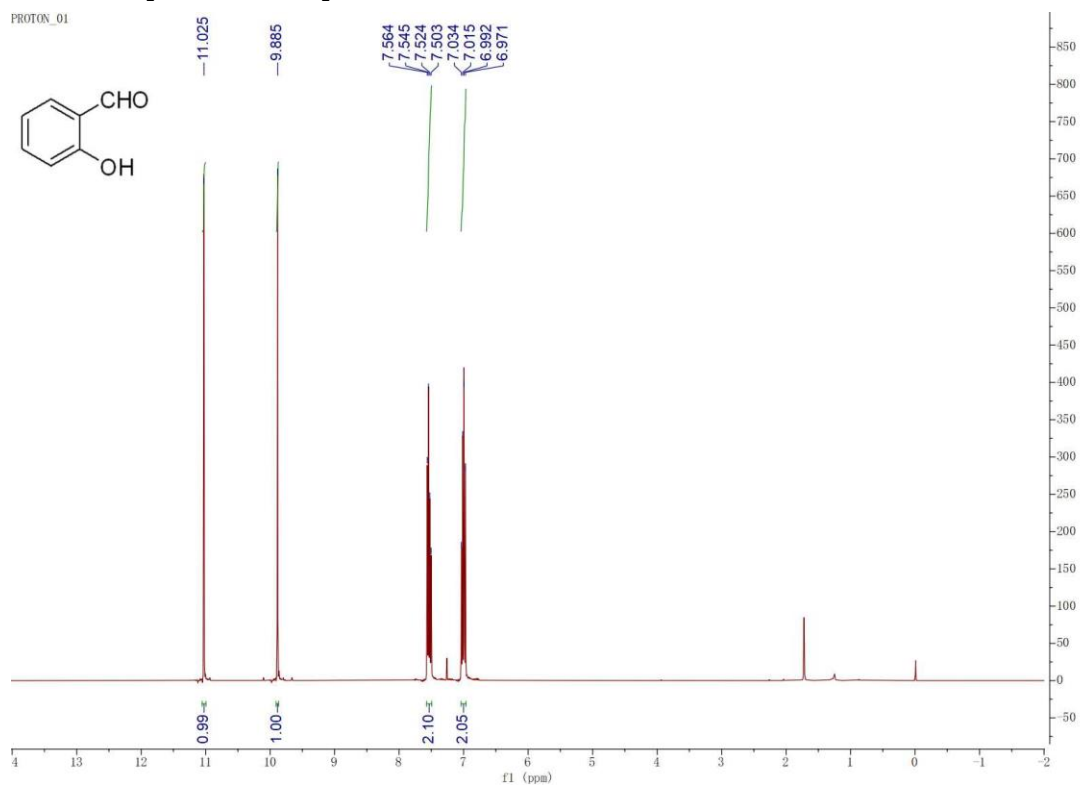
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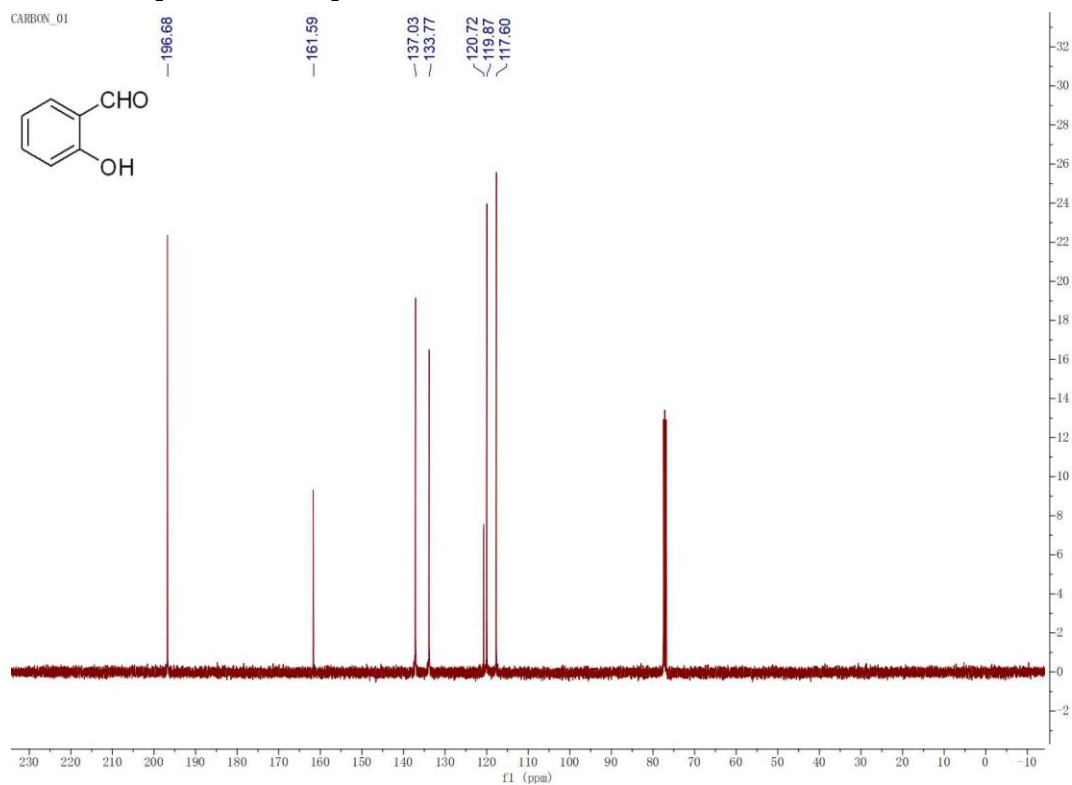
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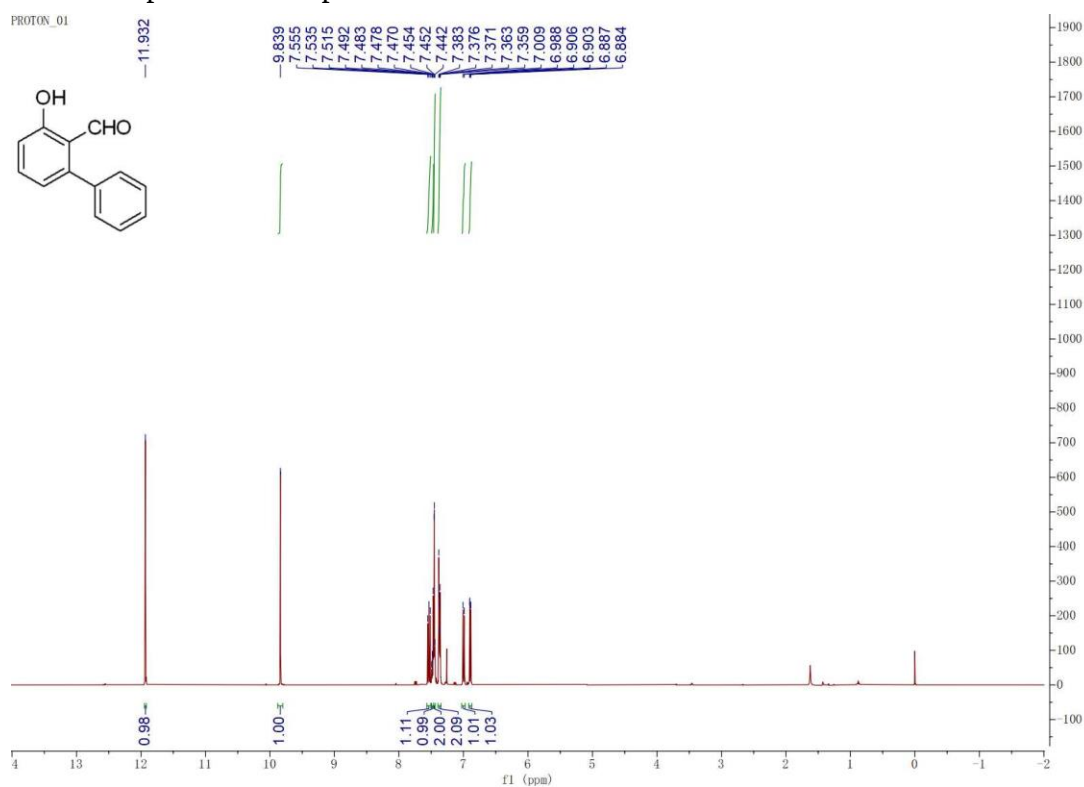
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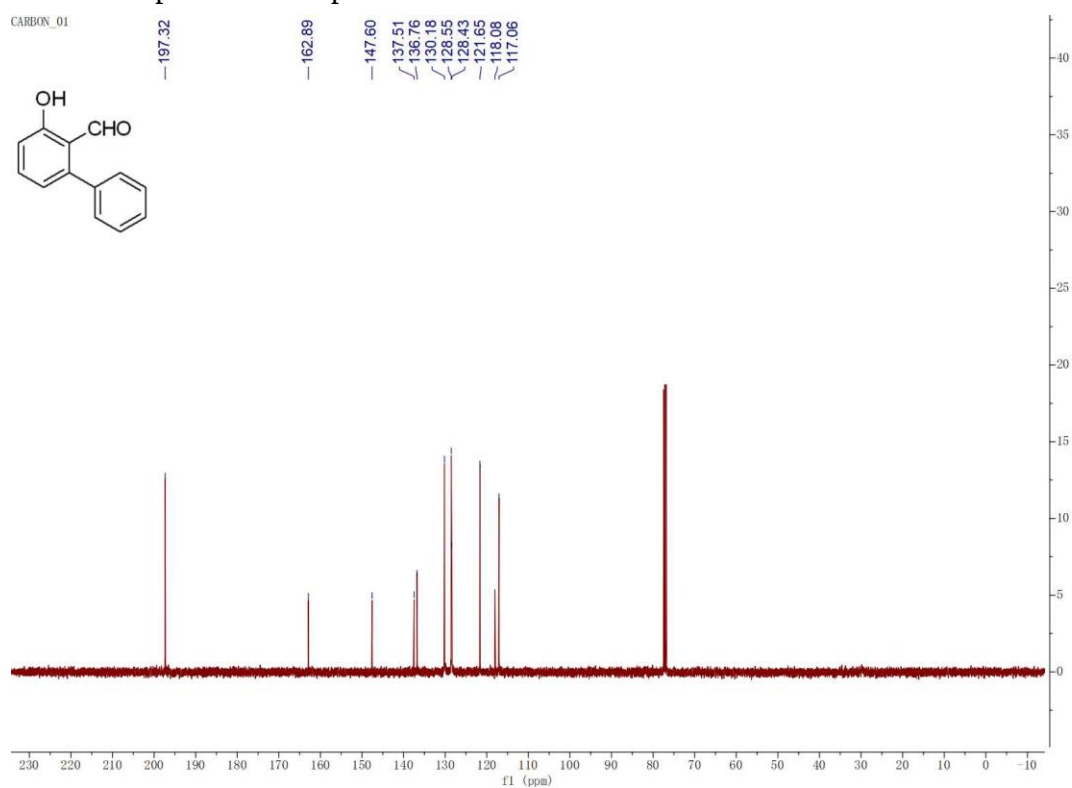
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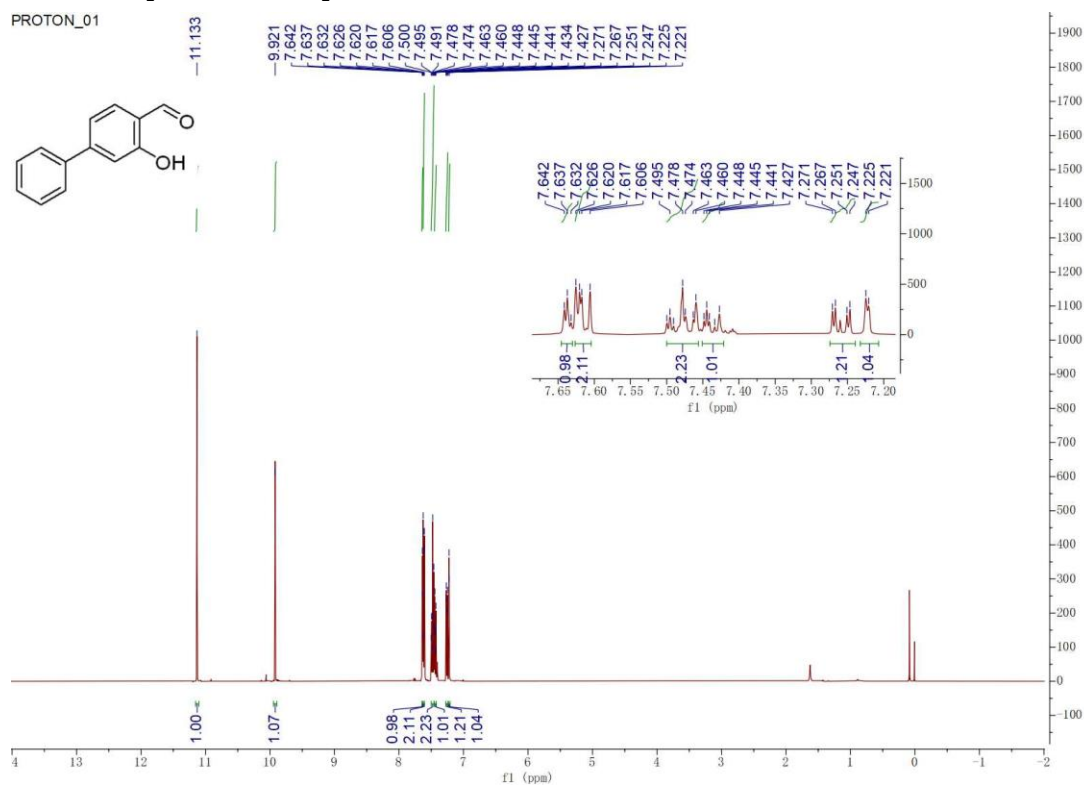
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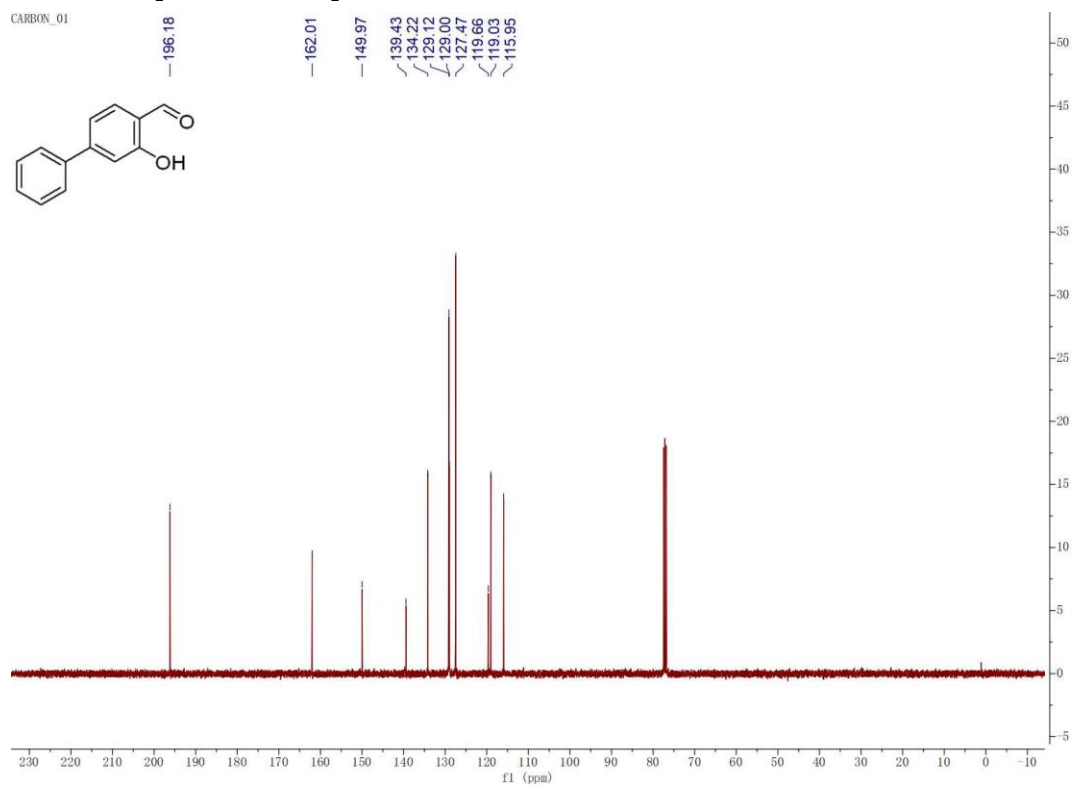
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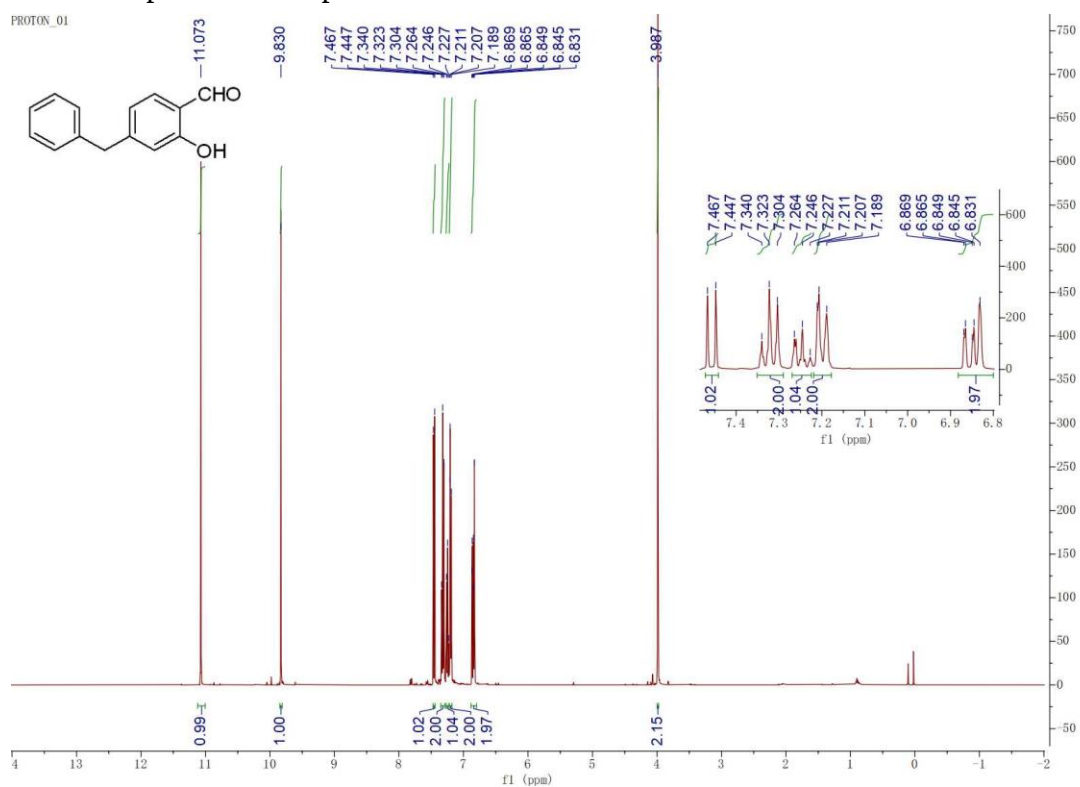
PROTON 01



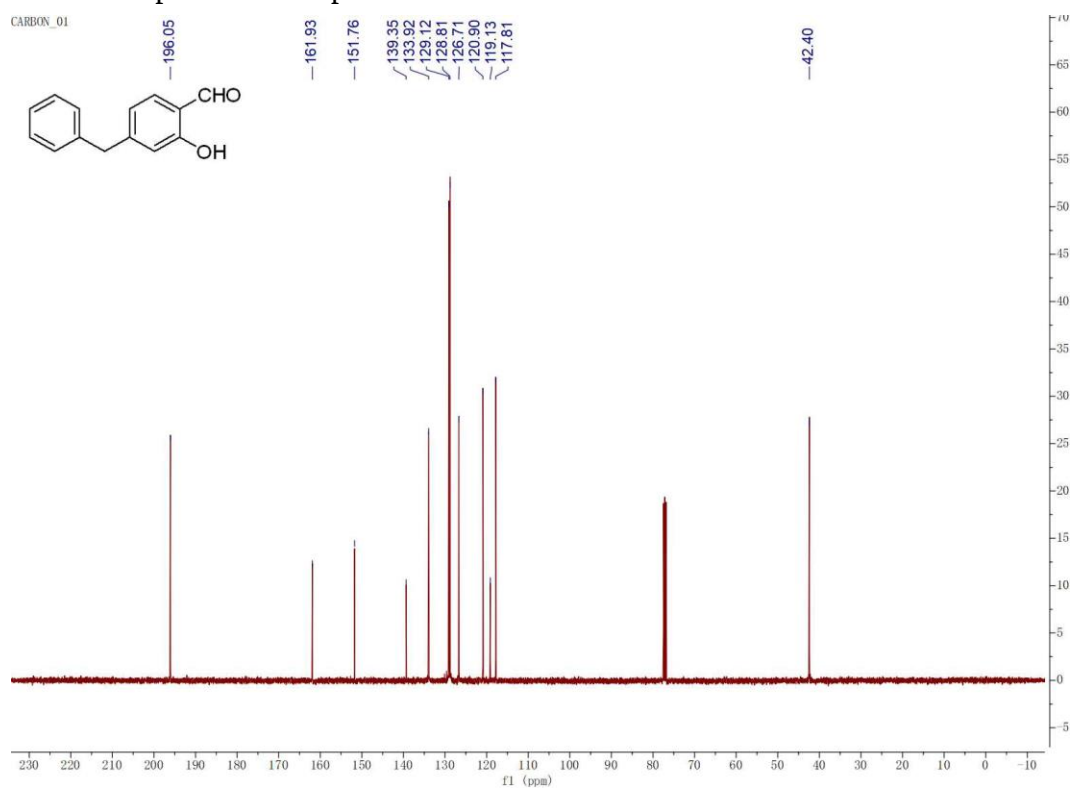
CARBON_01



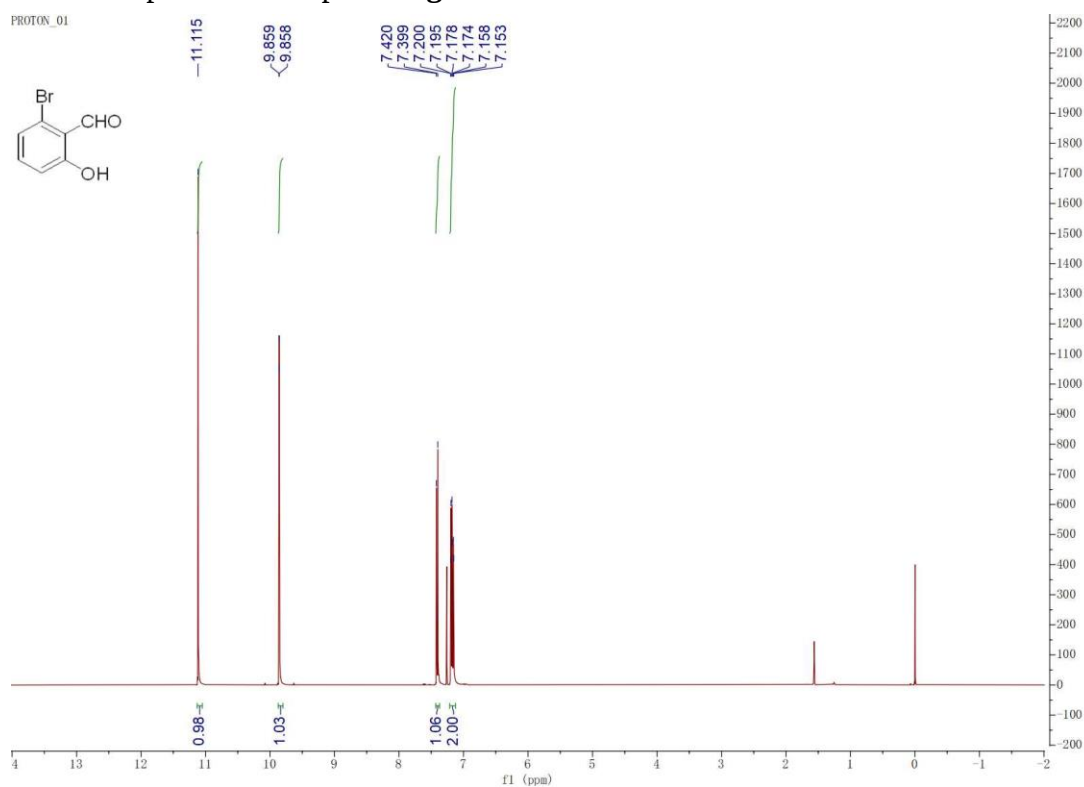
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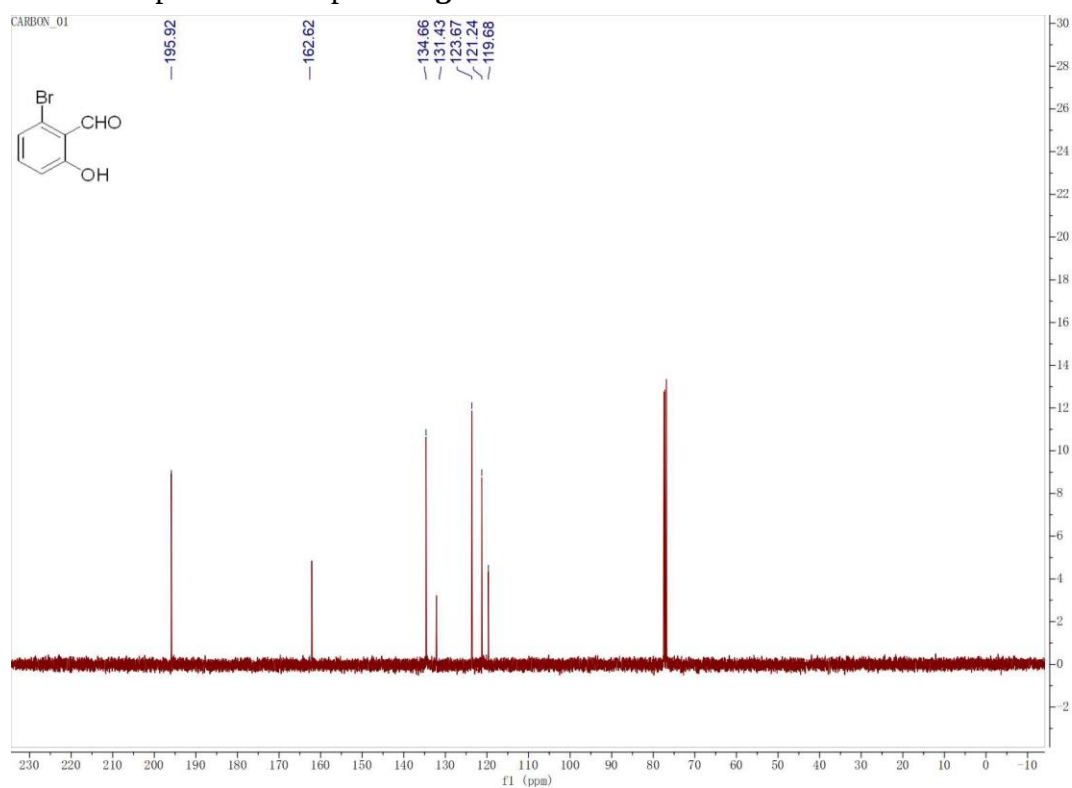
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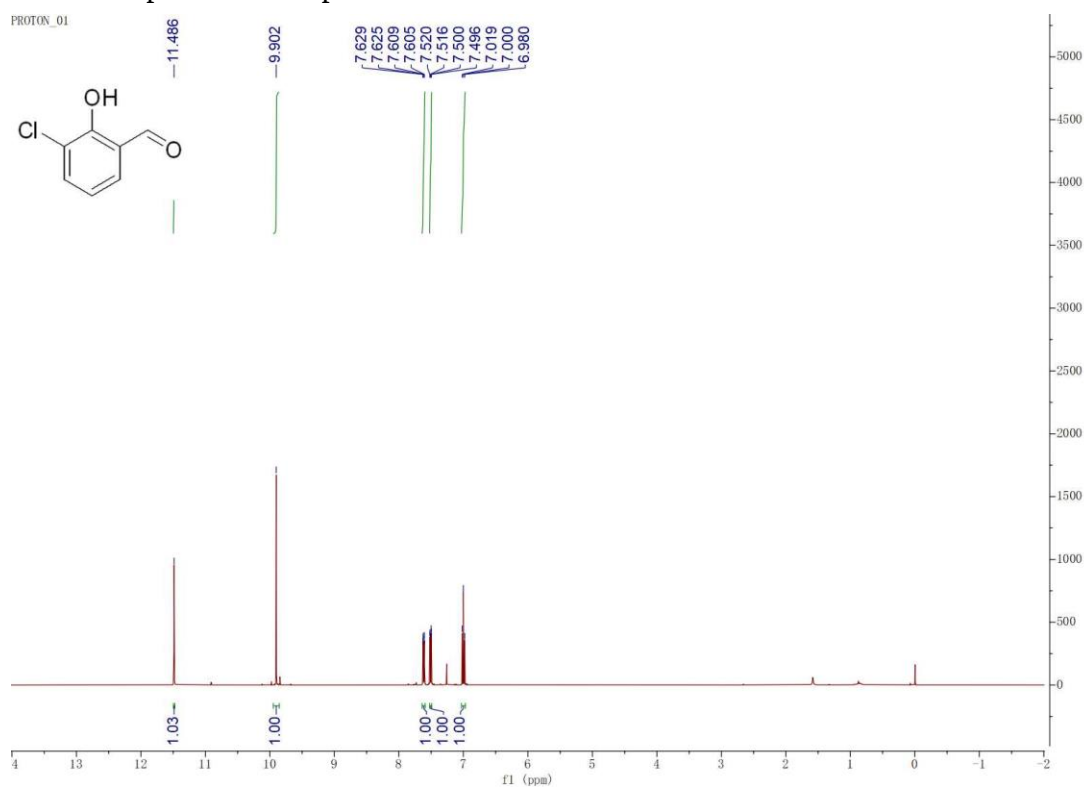
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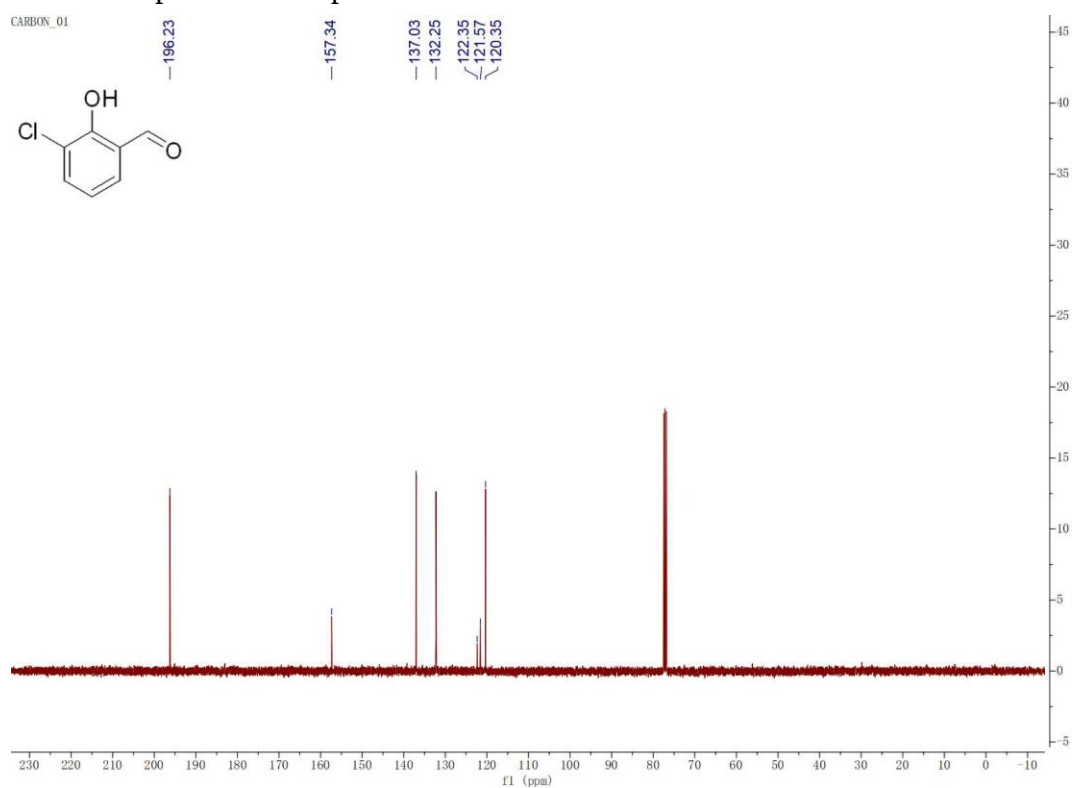
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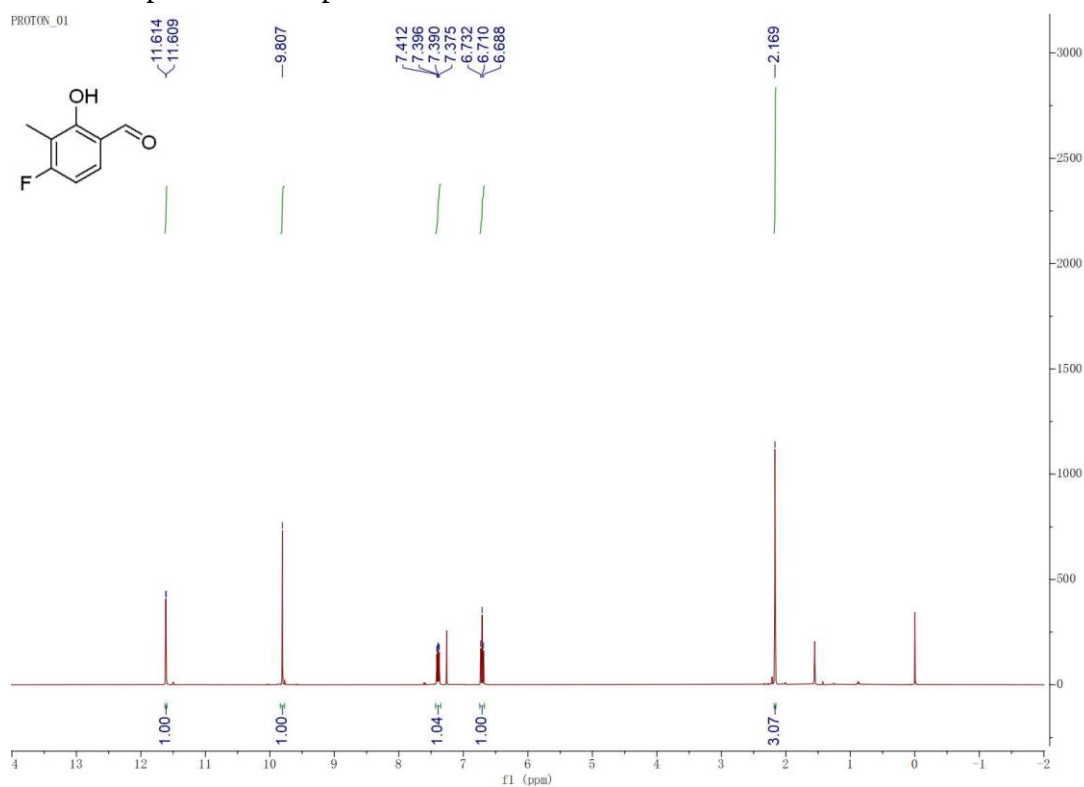
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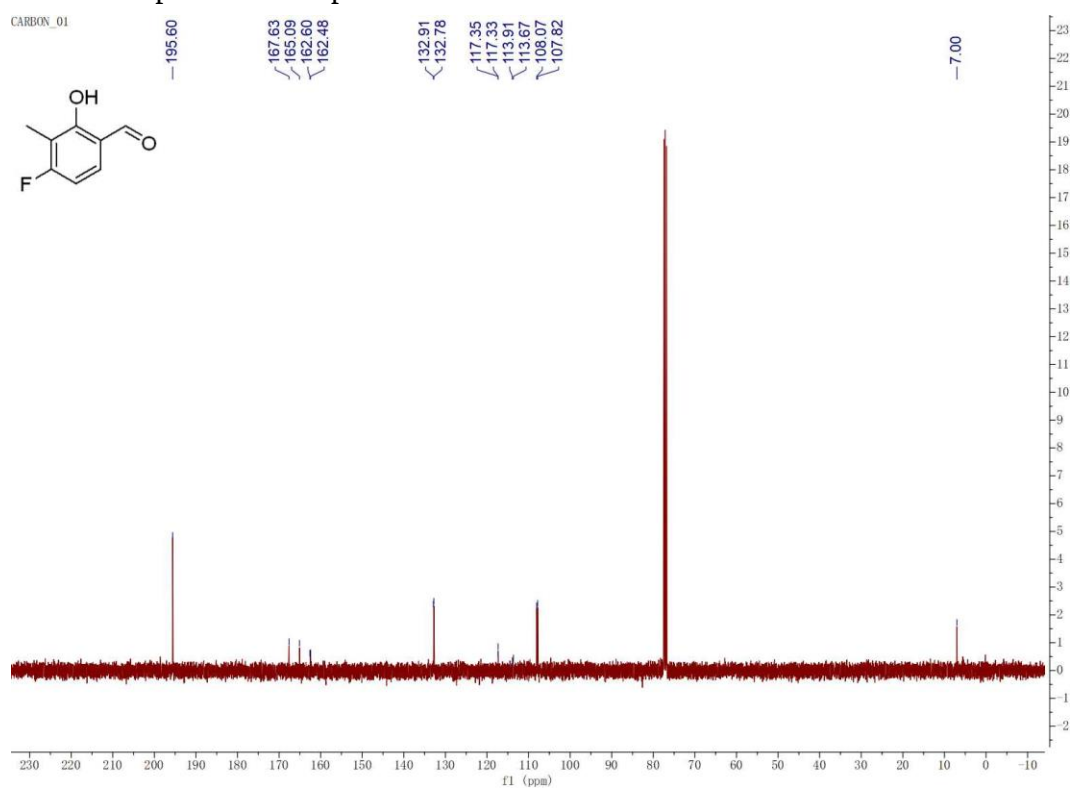
¹³C NMR spectra of compound **2h**:



¹H NMR spectra of compound 2i:

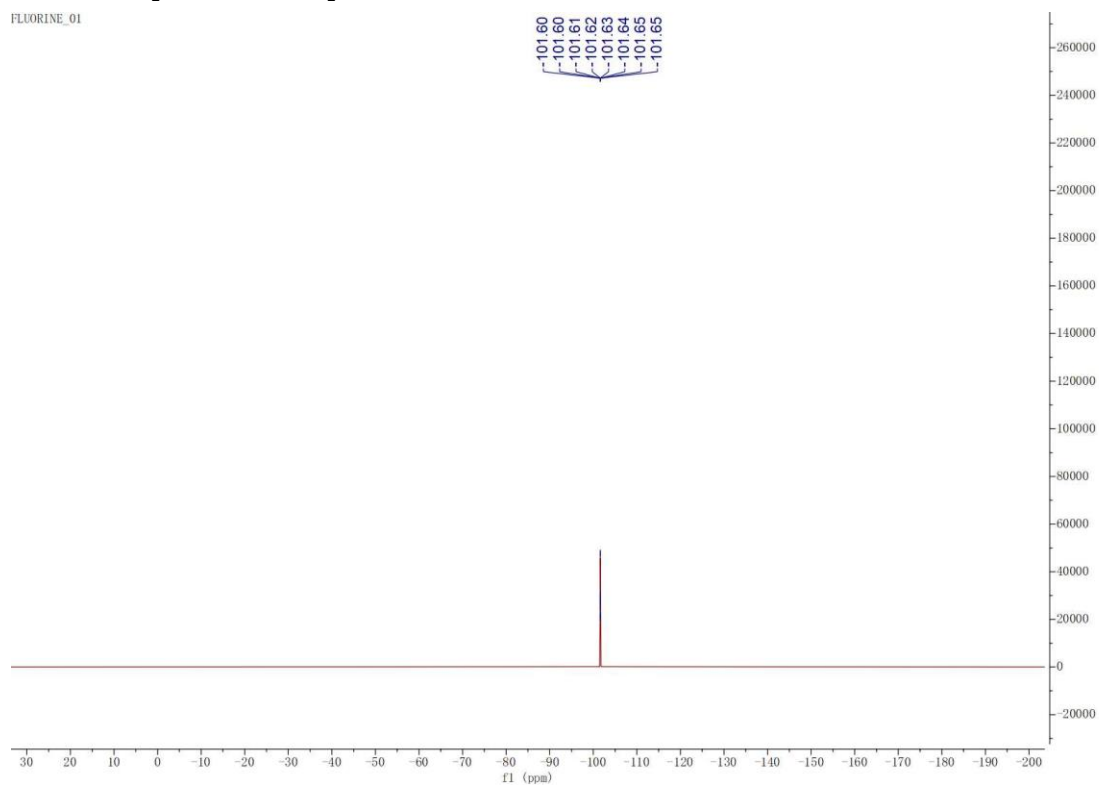


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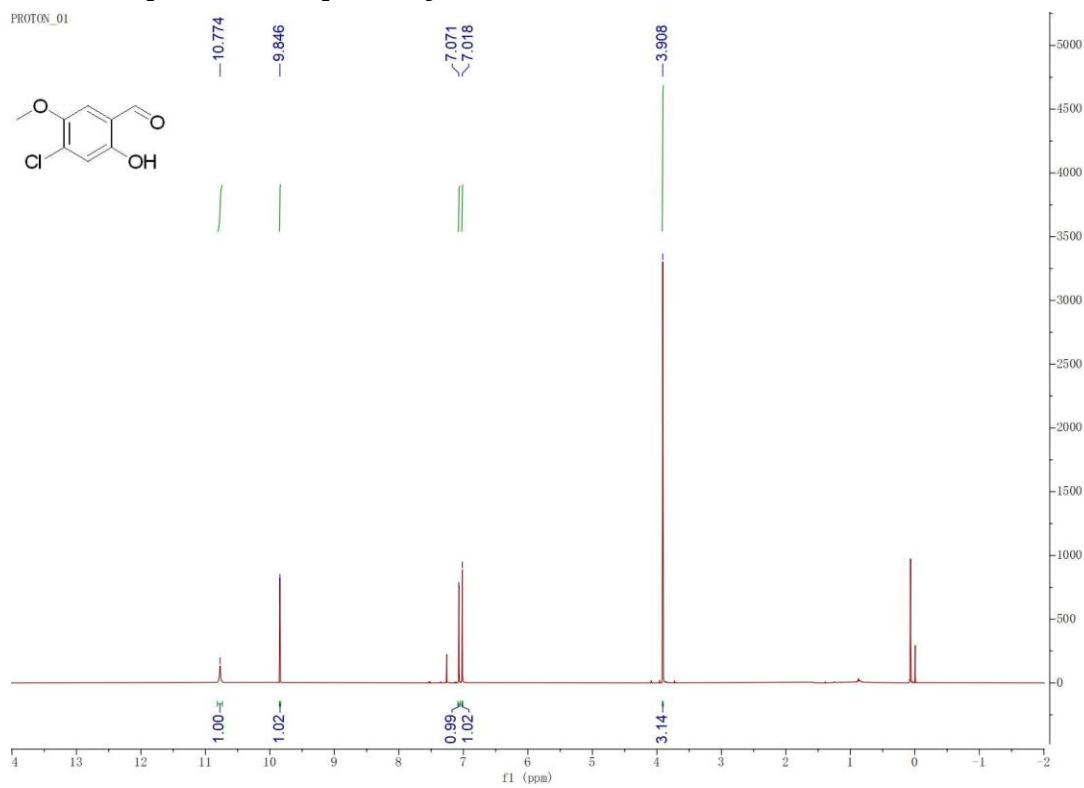
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FLUORINE_01

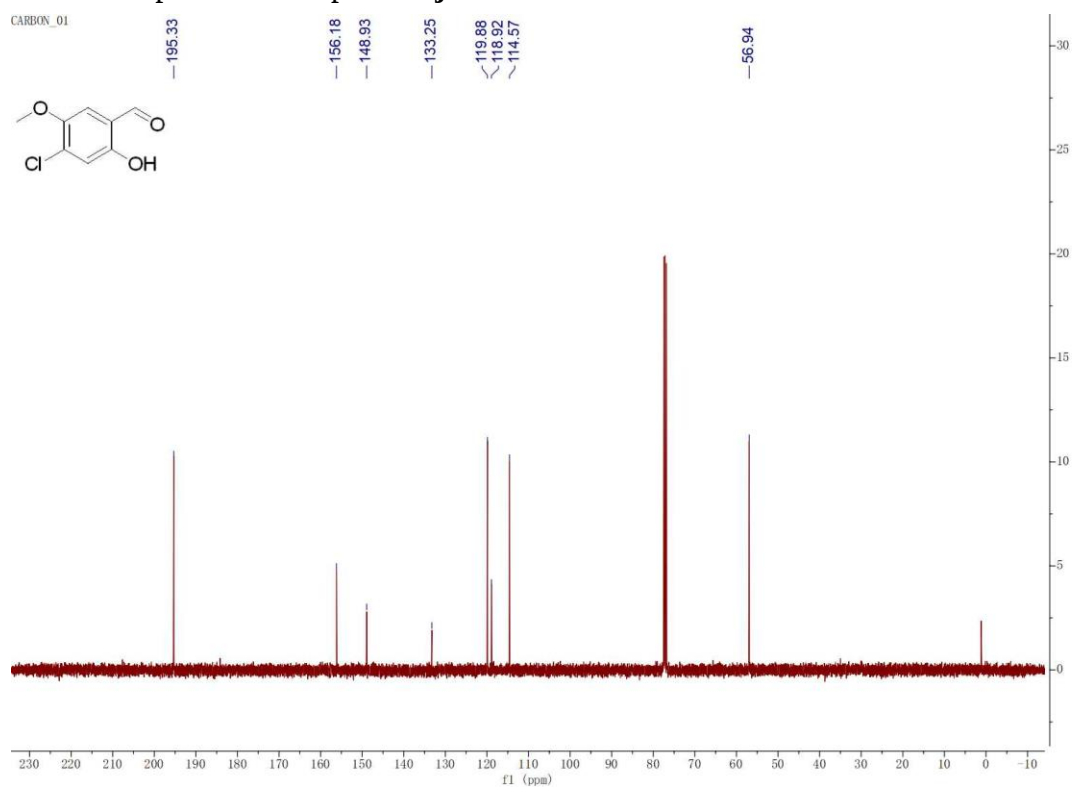


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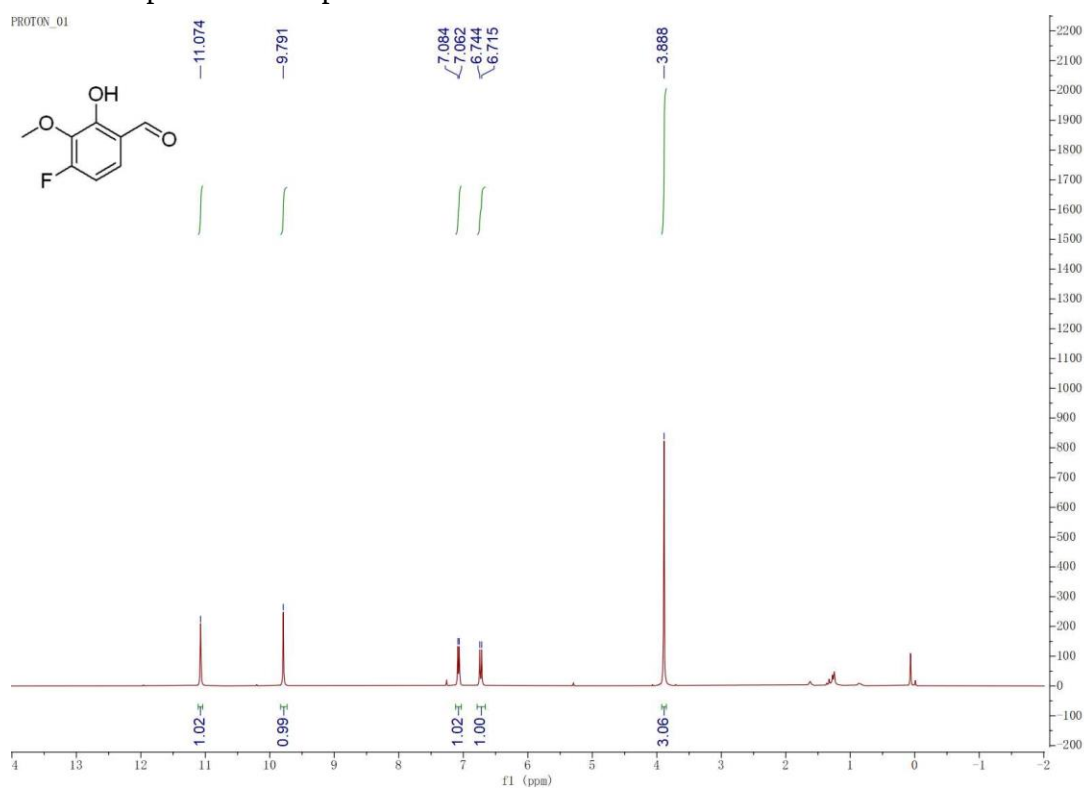
PROTON_01



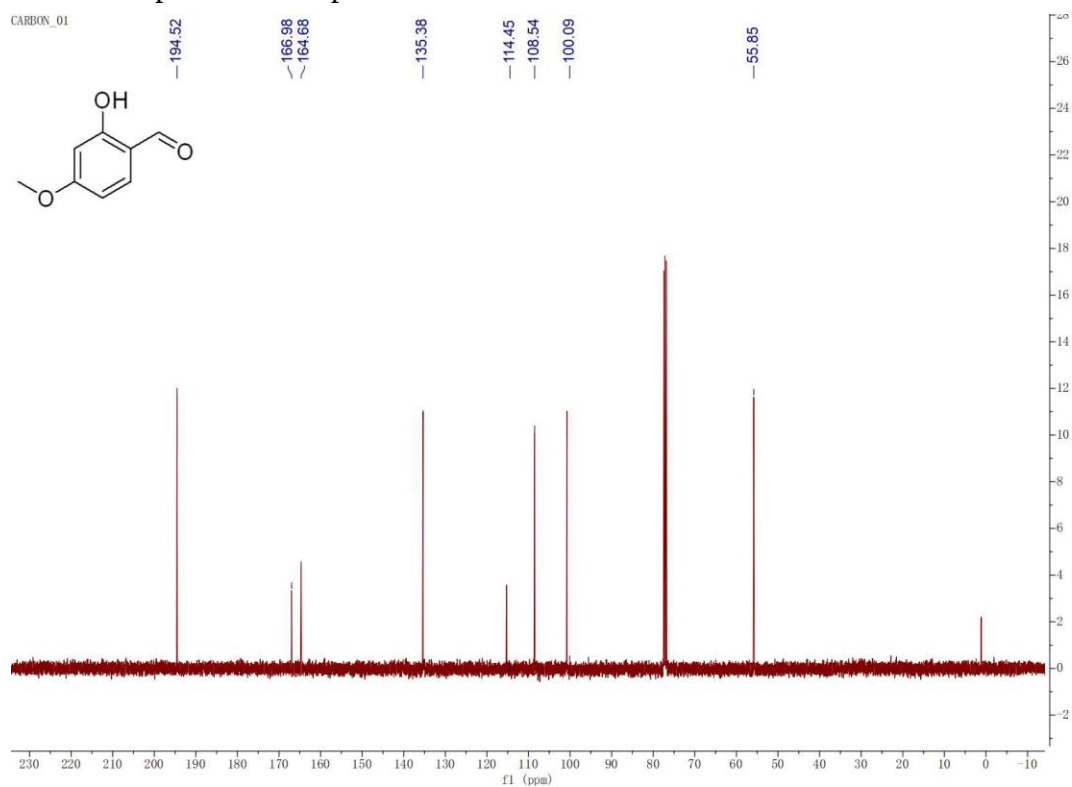
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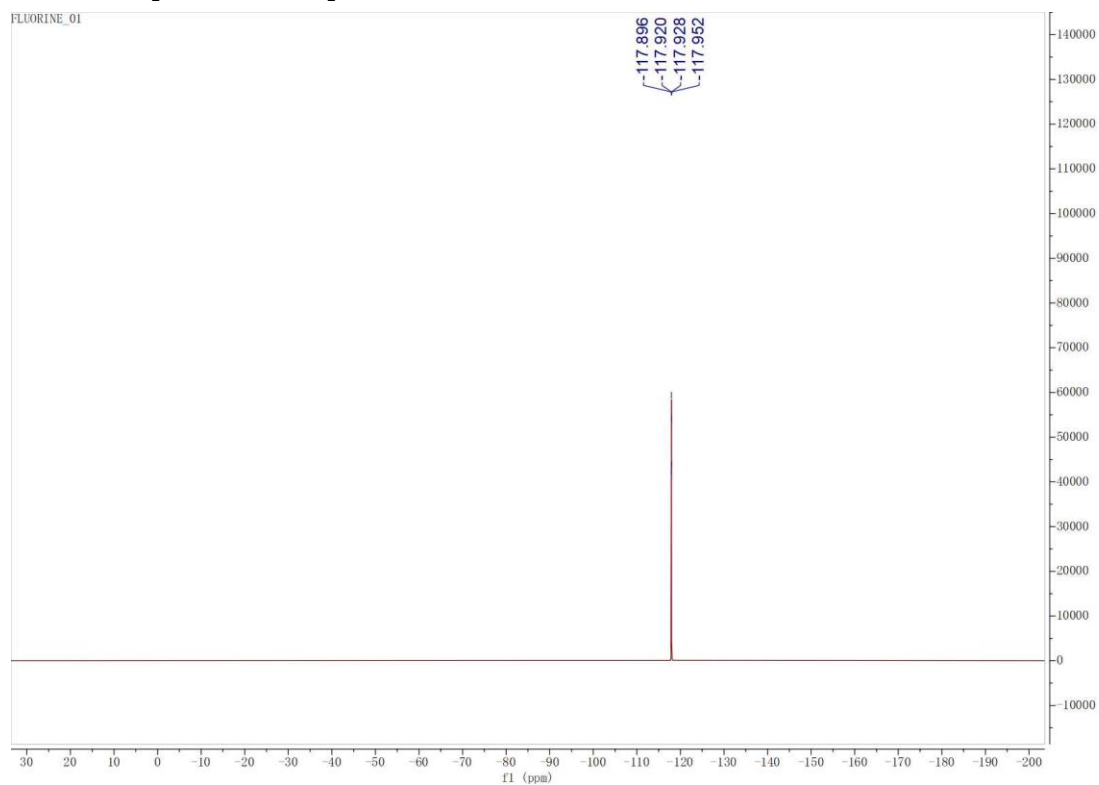
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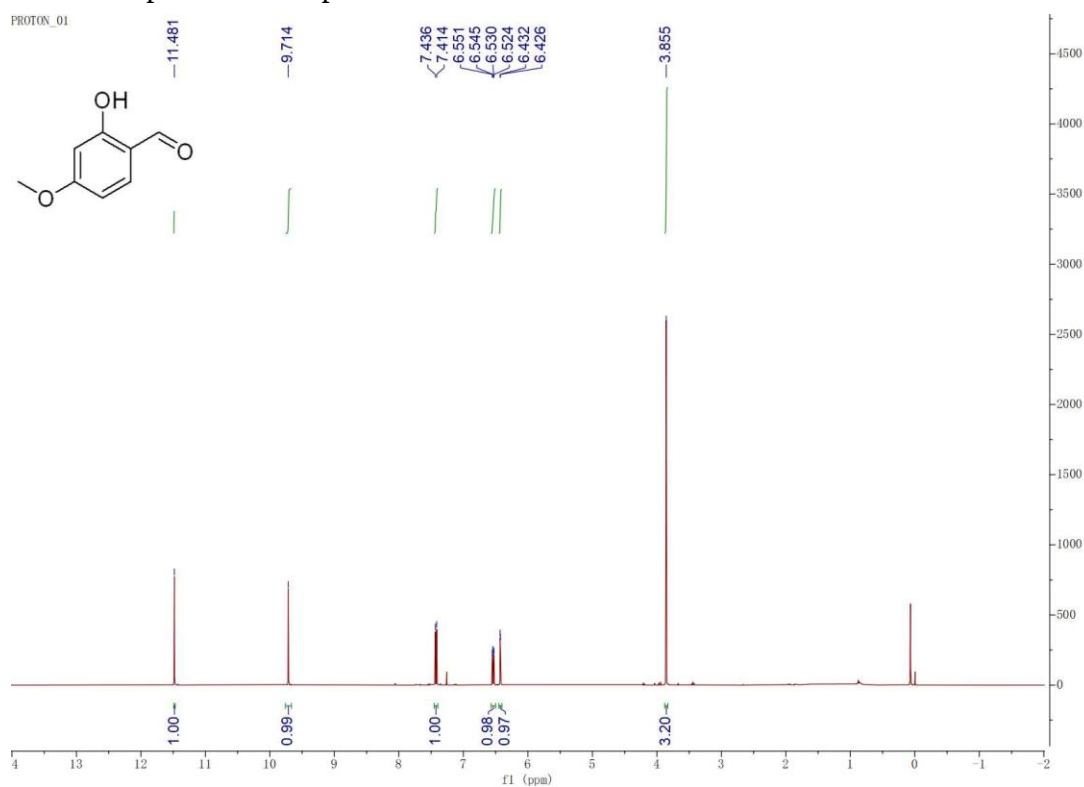
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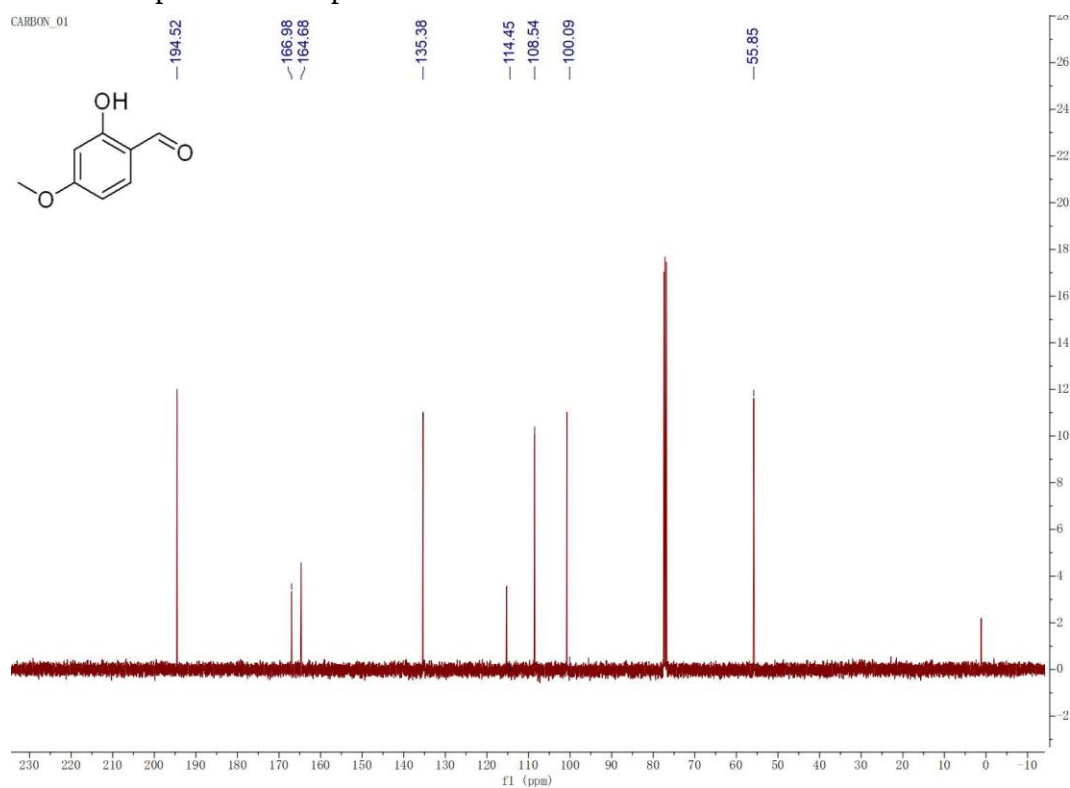
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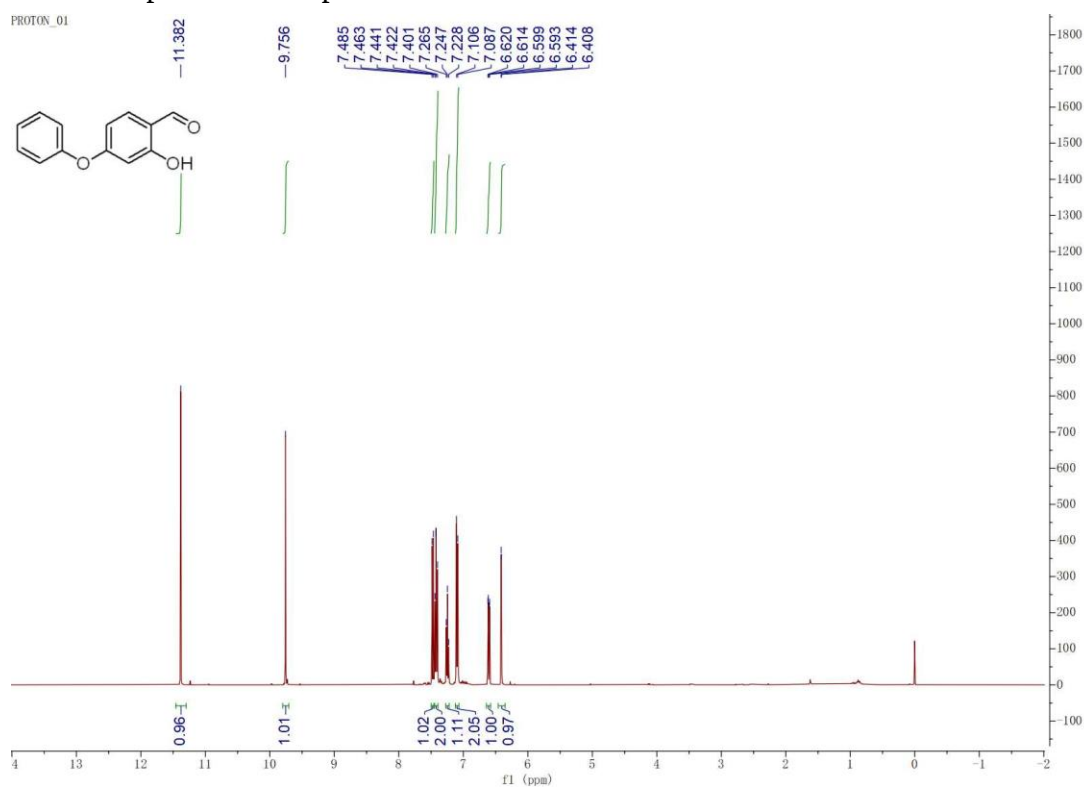
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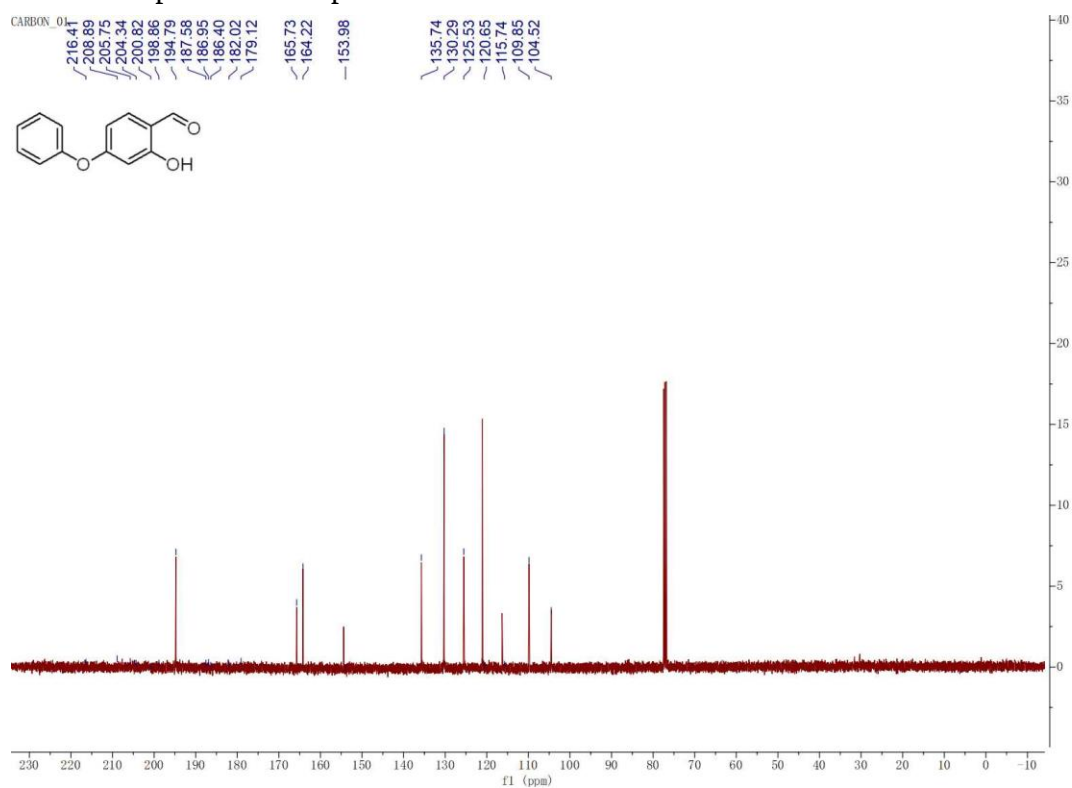
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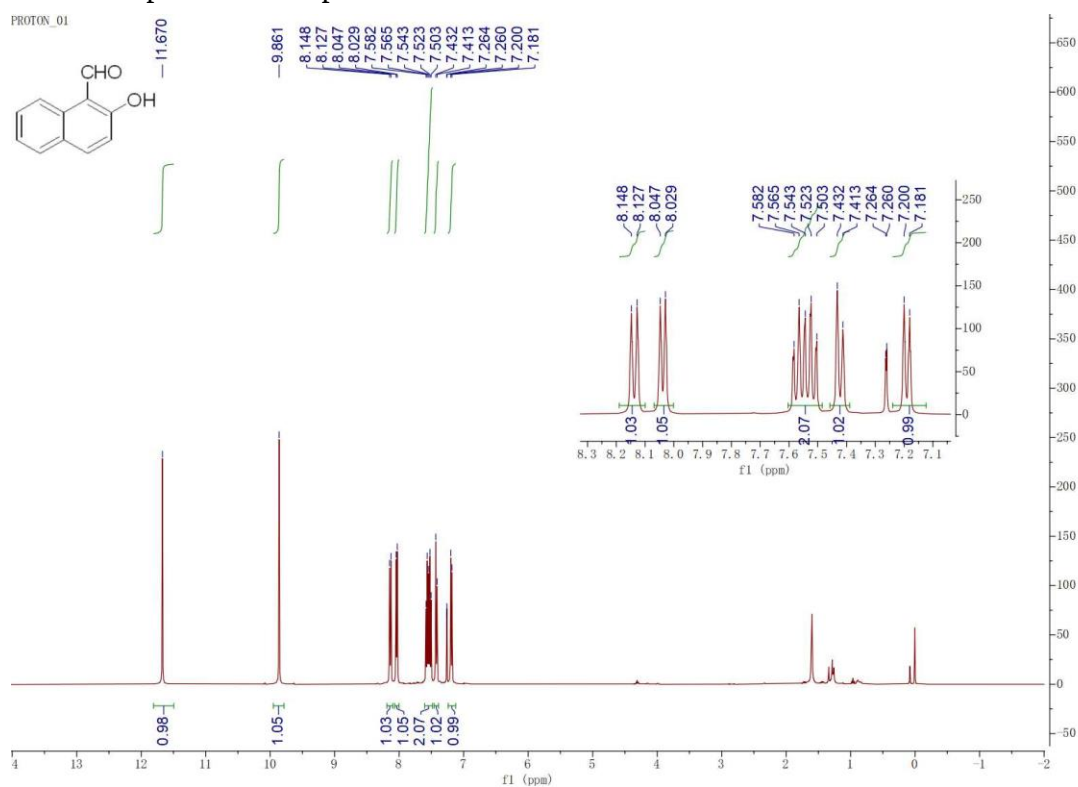
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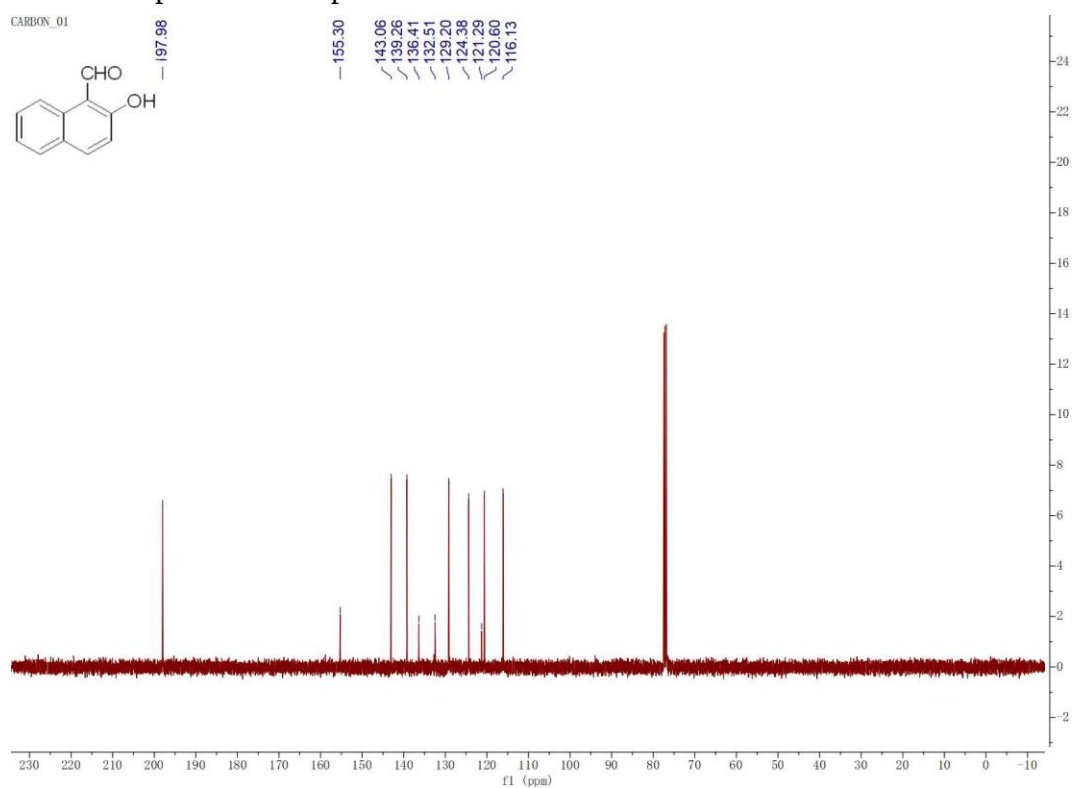
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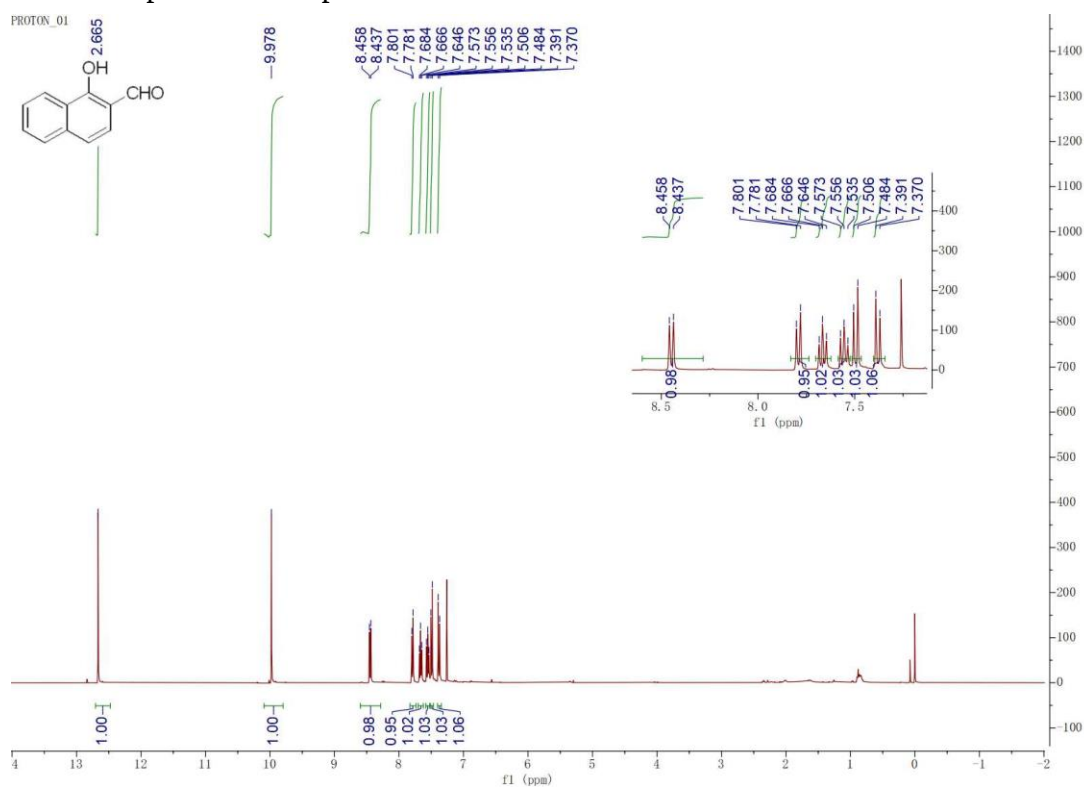
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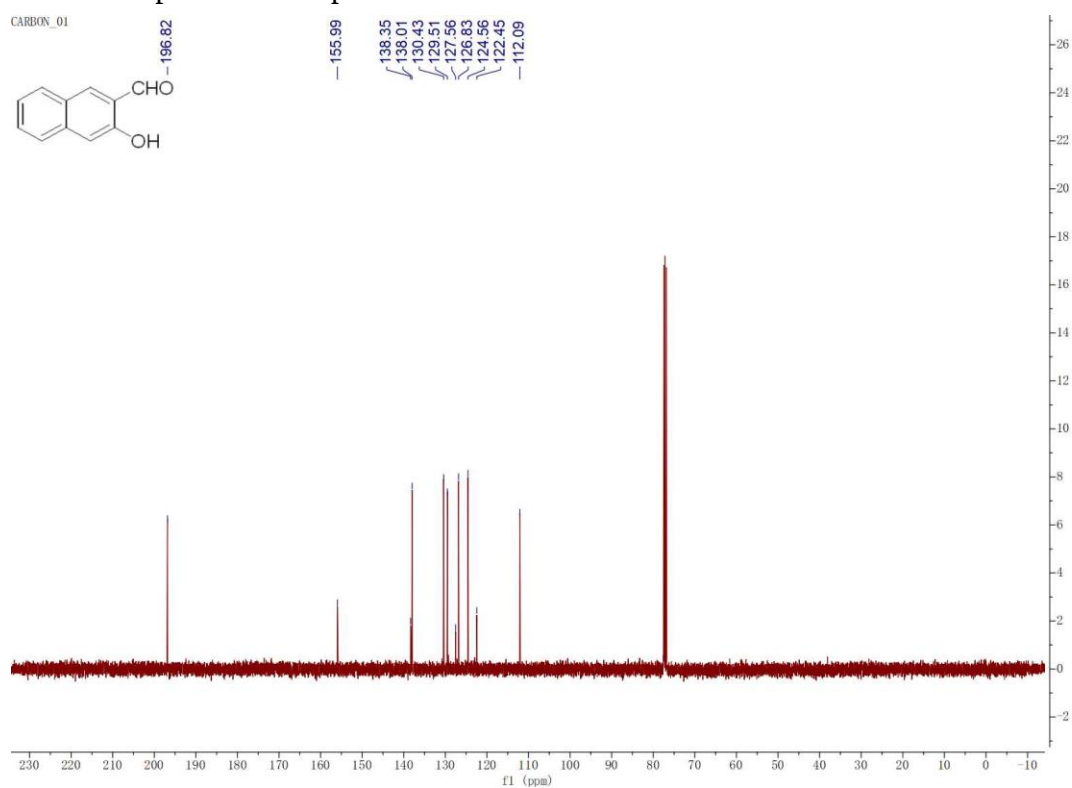
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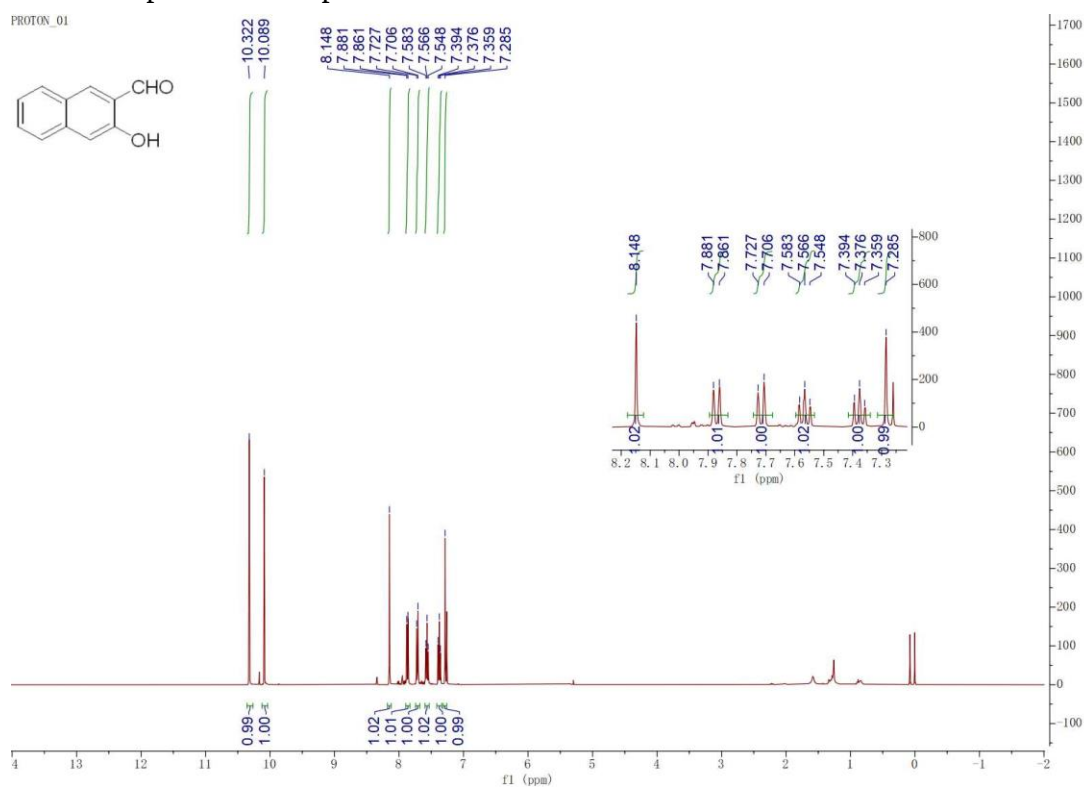
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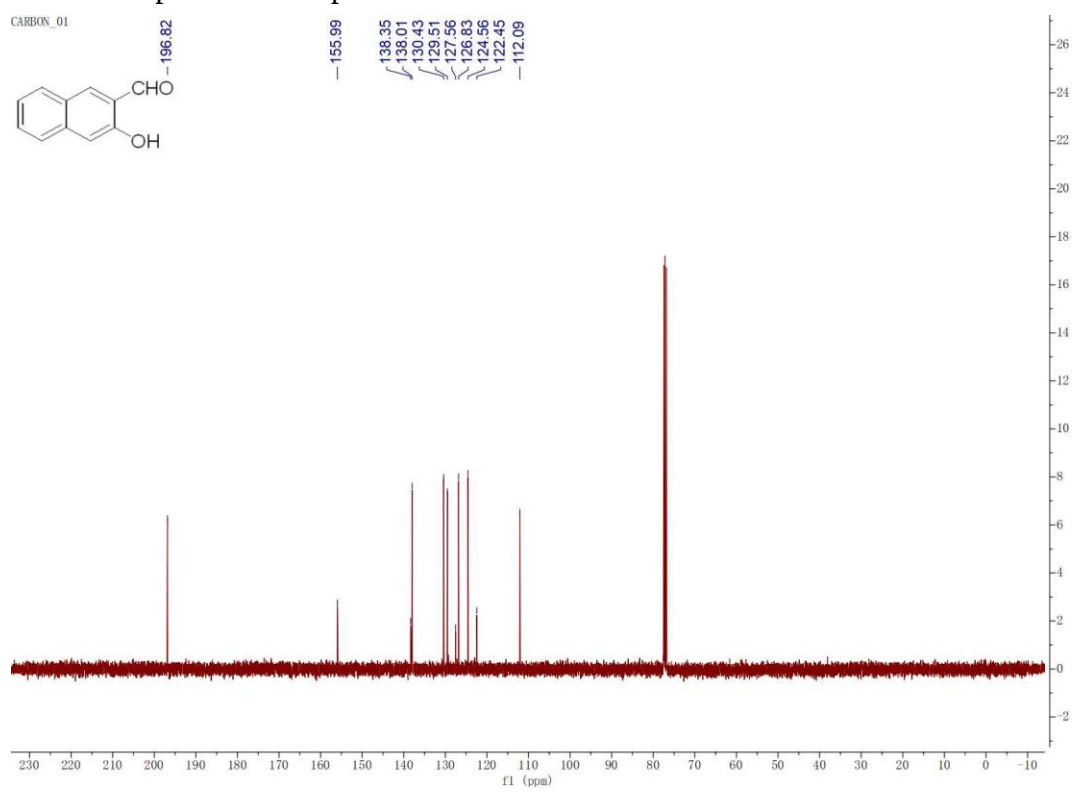
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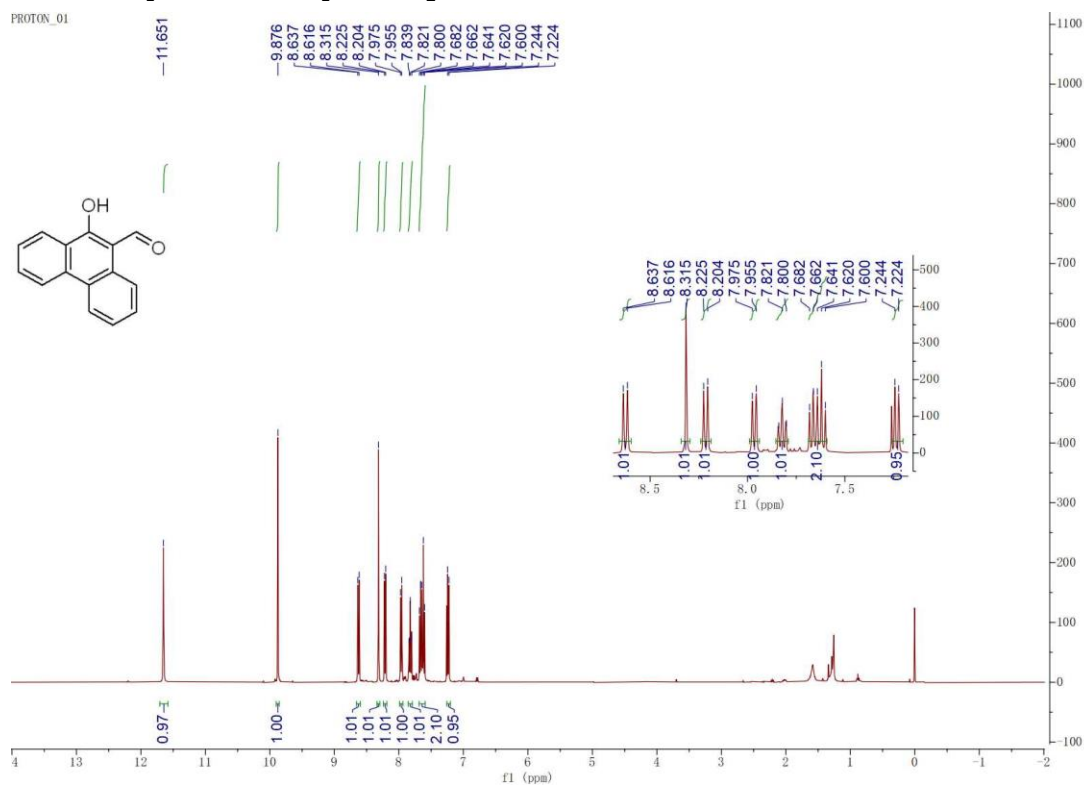
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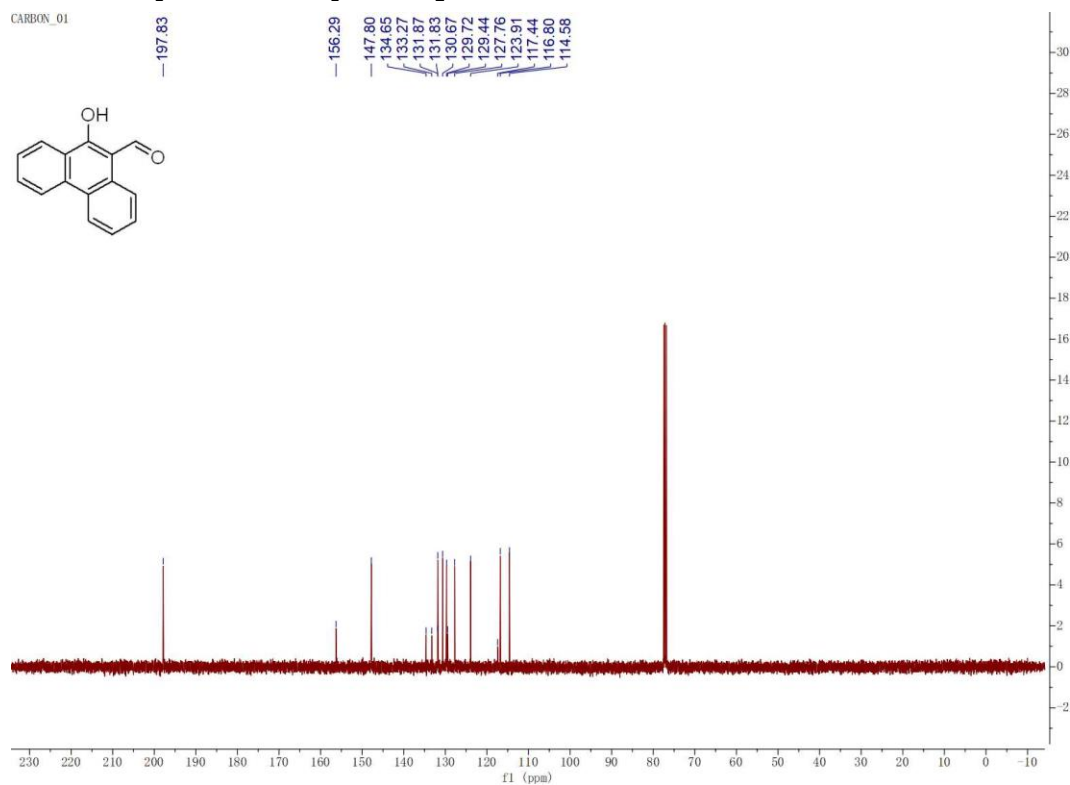
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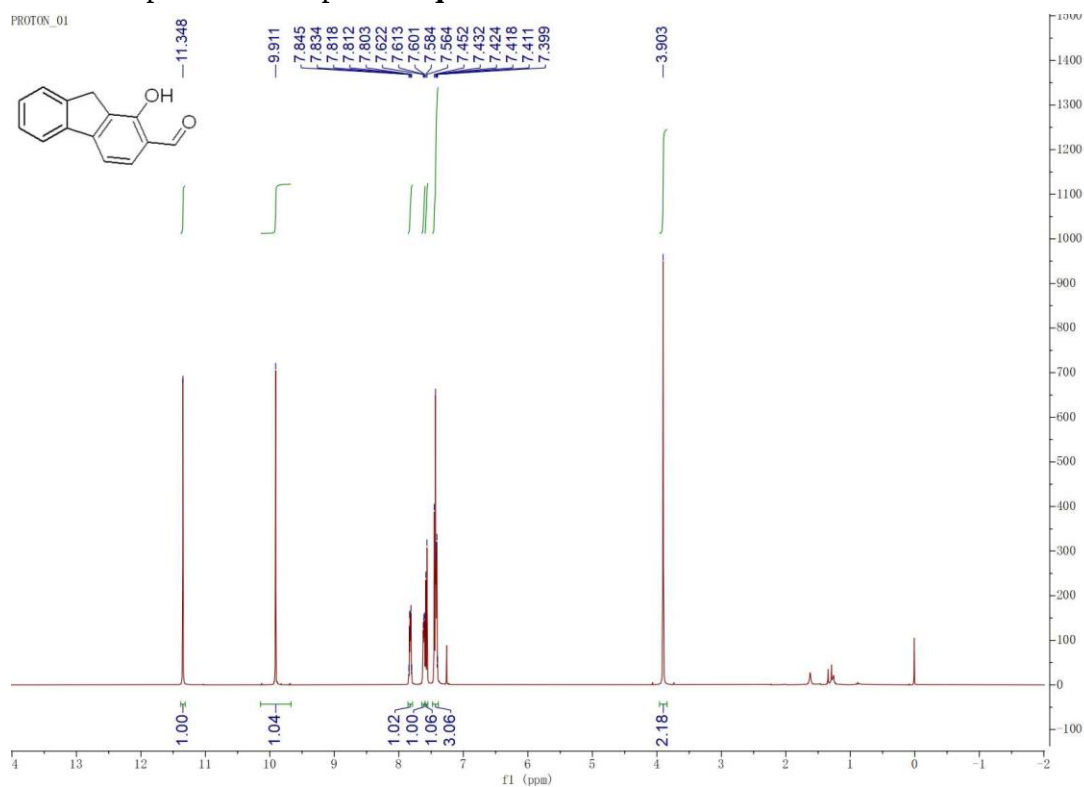
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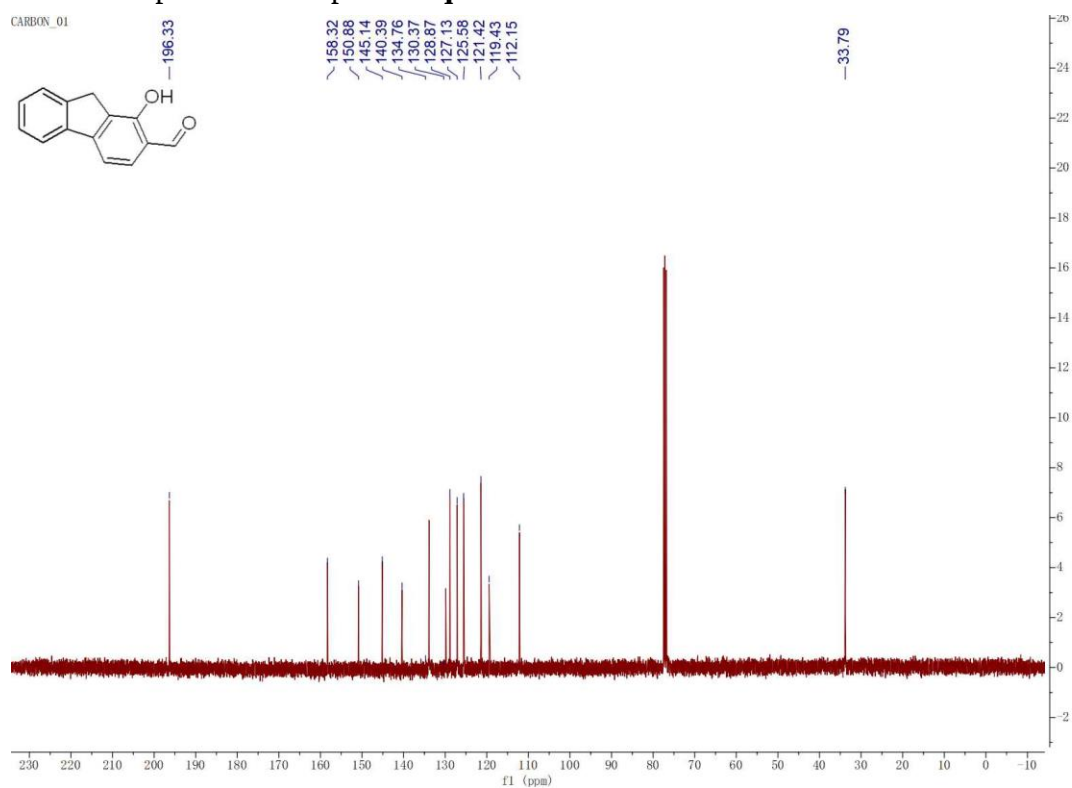
¹³C NMR spectra of compound **2p**:



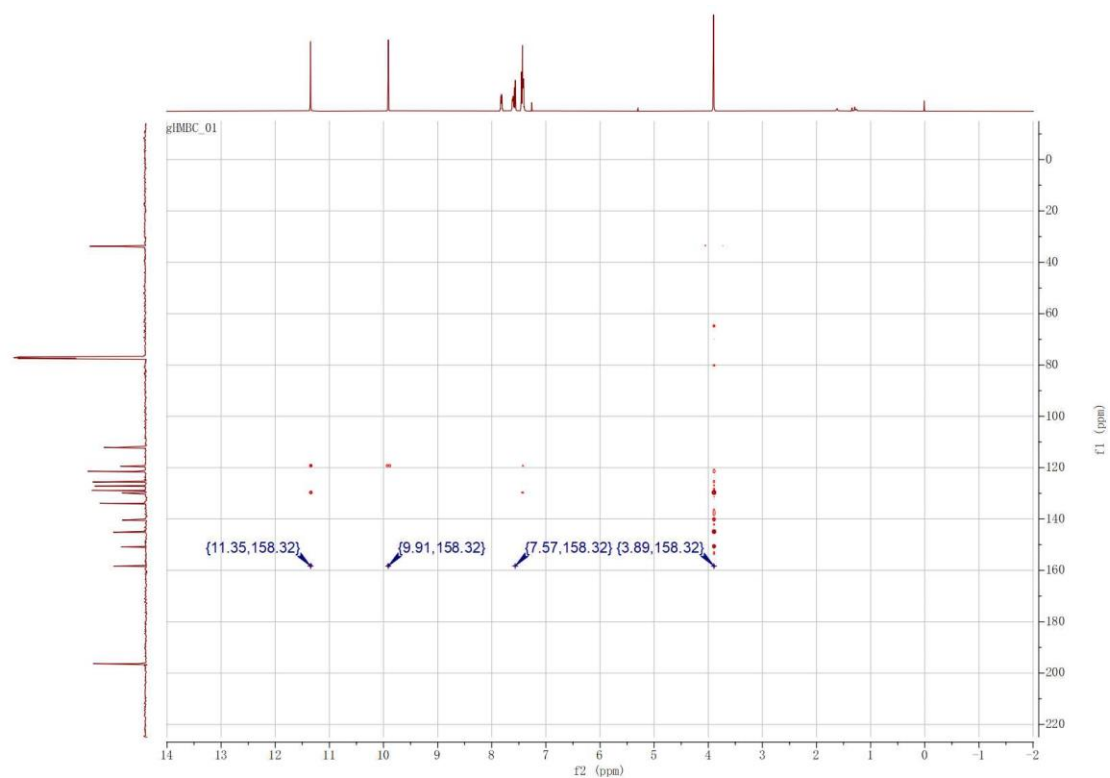
¹H NMR spectra of compound **2q**:



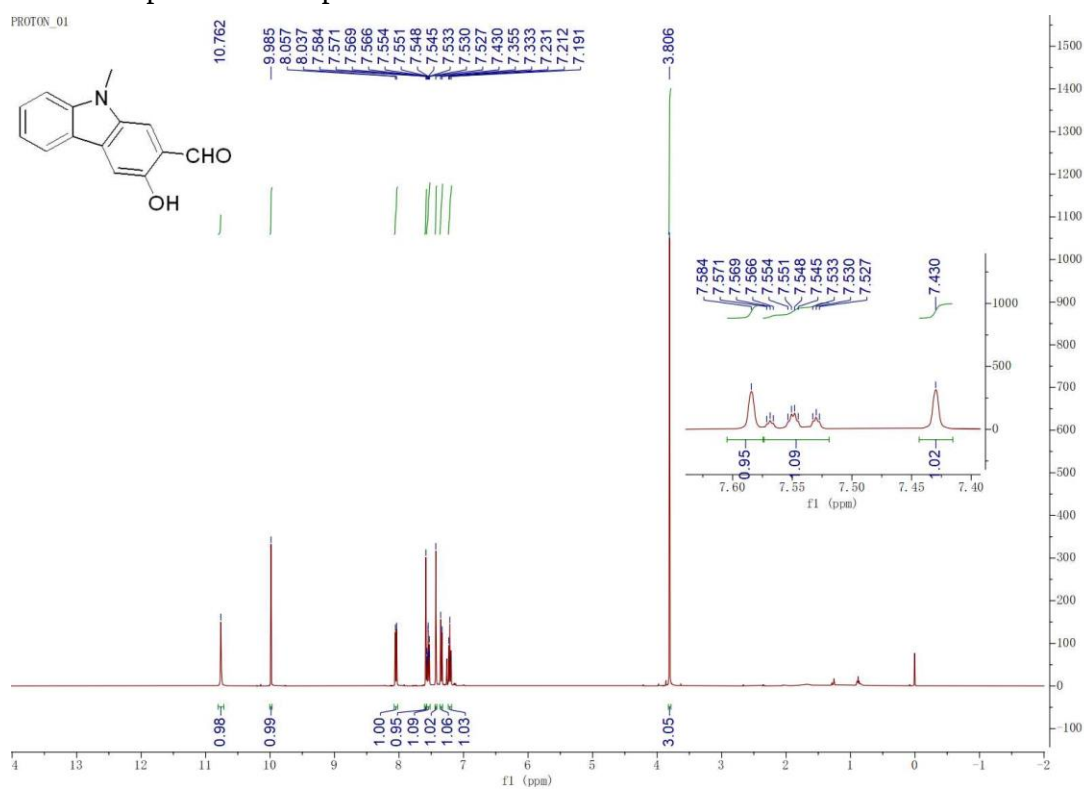
¹³C NMR spectra of compound **2q**:



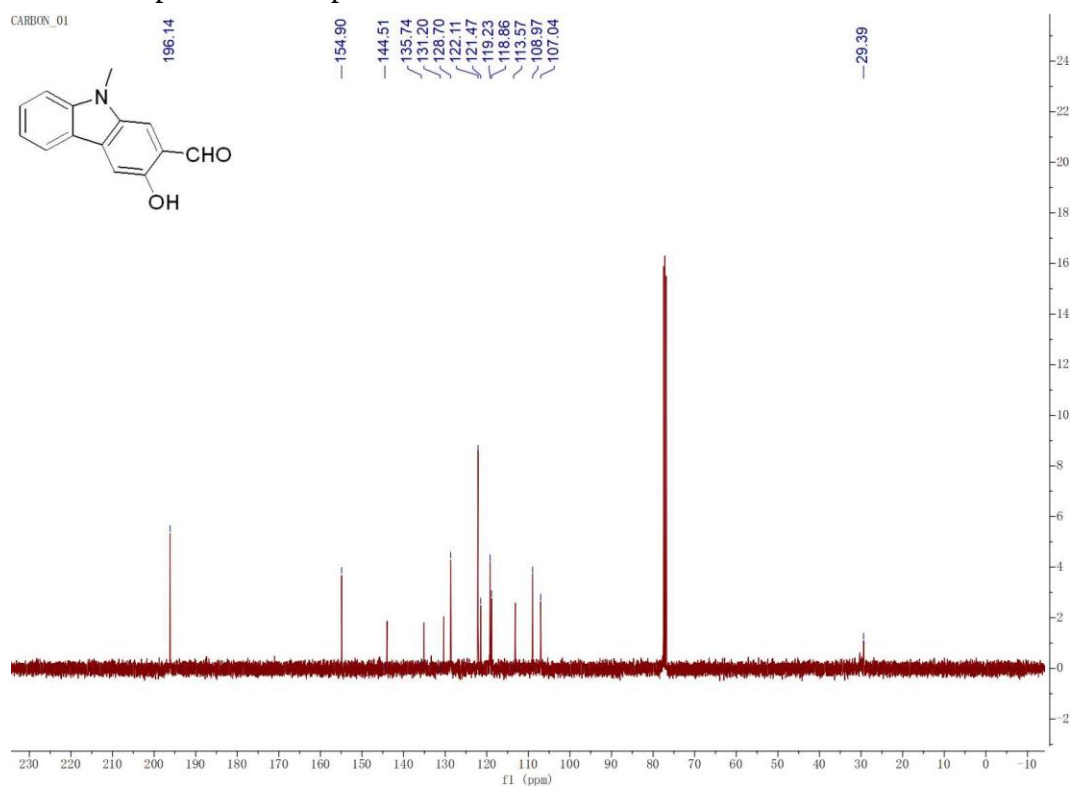
Full HMBC NMR spectra of compound **2q**:



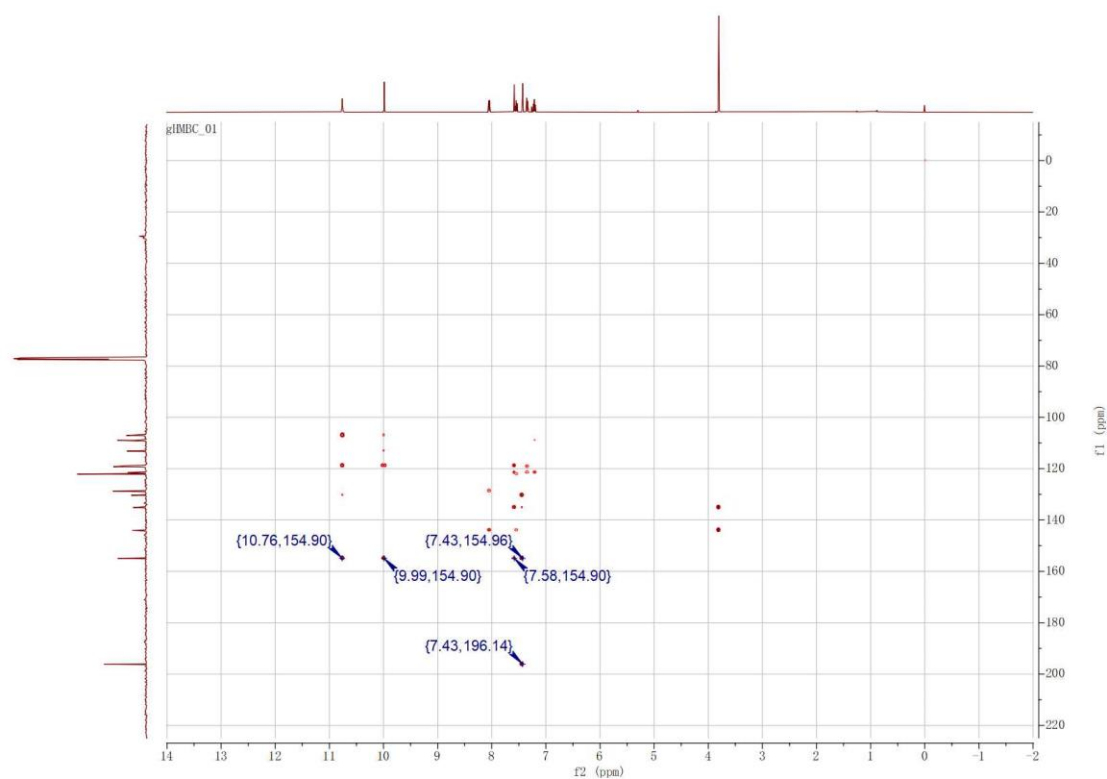
¹H NMR spectra of compound **2r**:



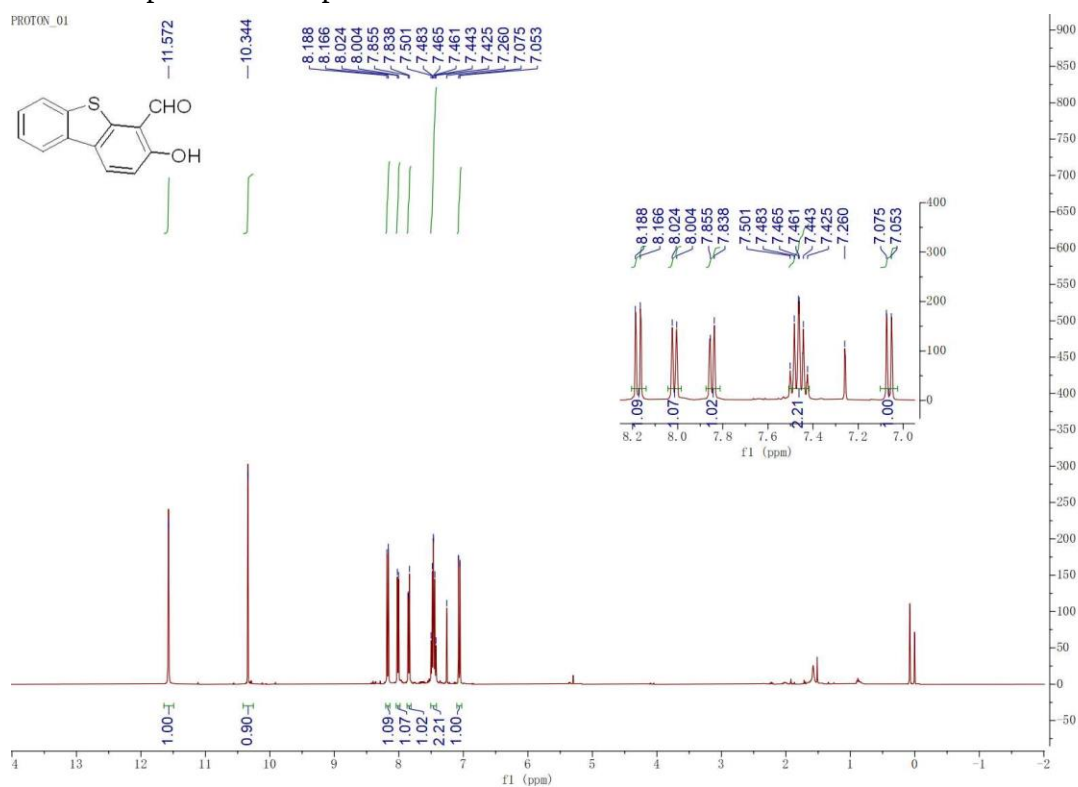
¹³C NMR spectra of compound **2r**:



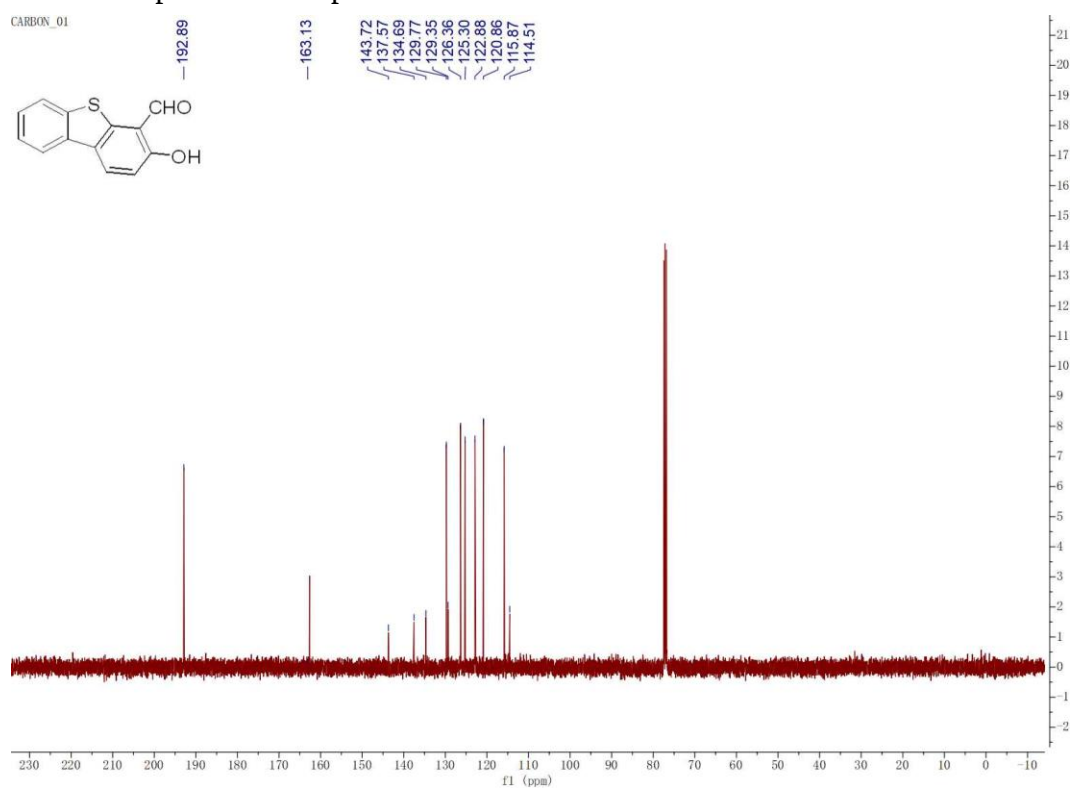
Full HMBC NMR spectra of compound **2r**:



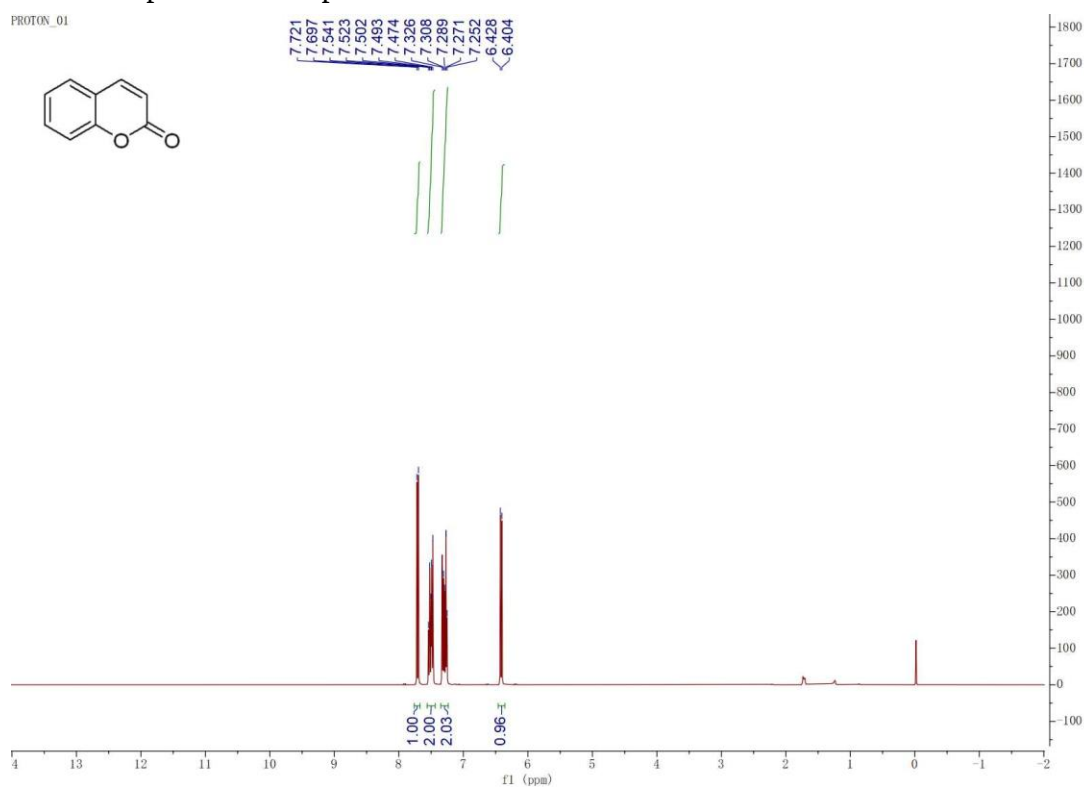
¹H NMR spectra of compound 2s:



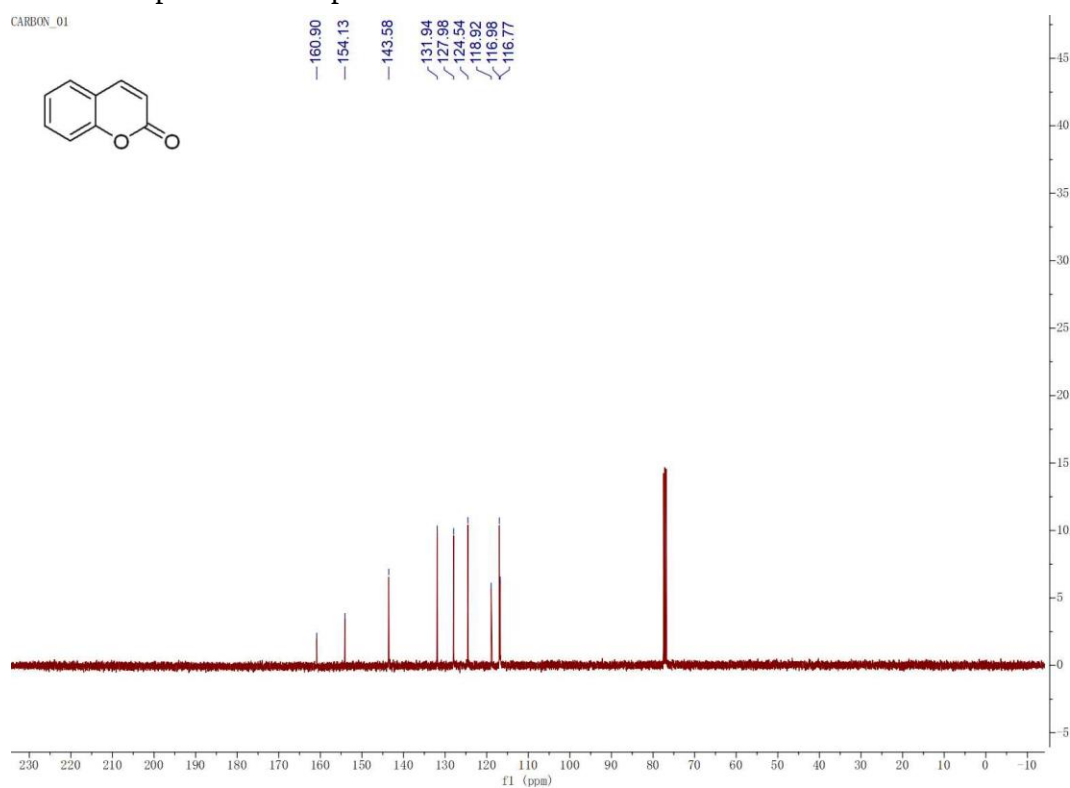
¹³C NMR spectra of compound 2s:



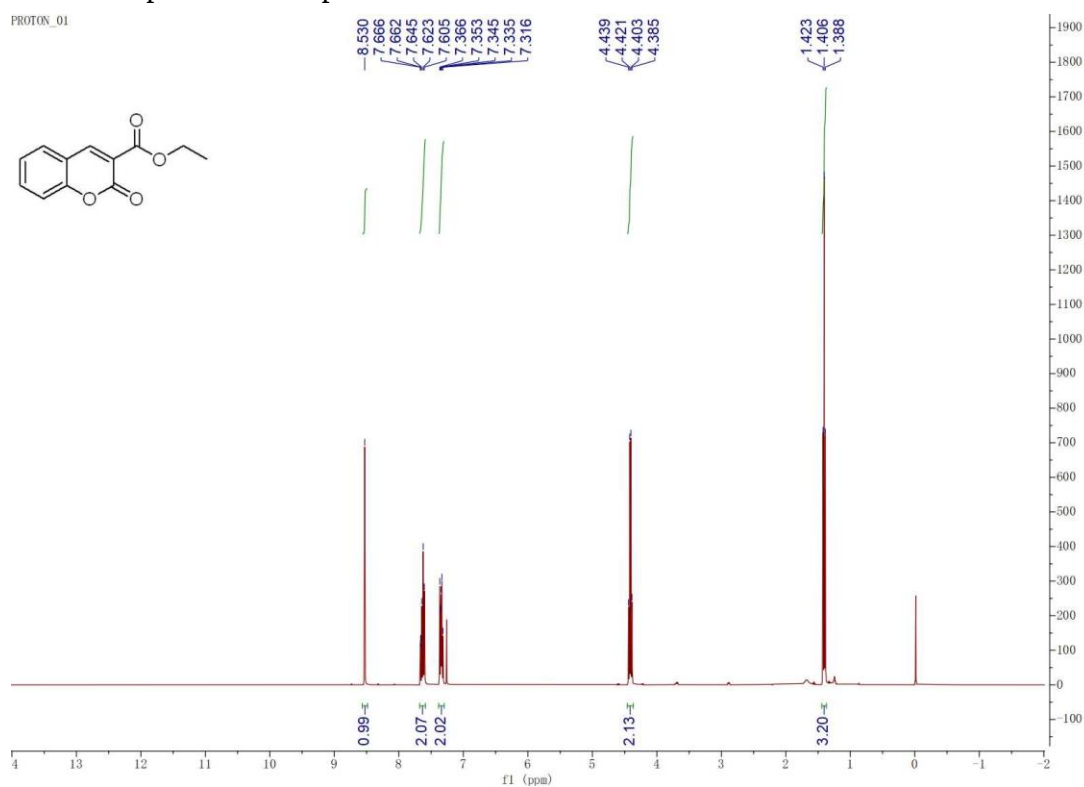
¹H NMR spectra of compound 3:



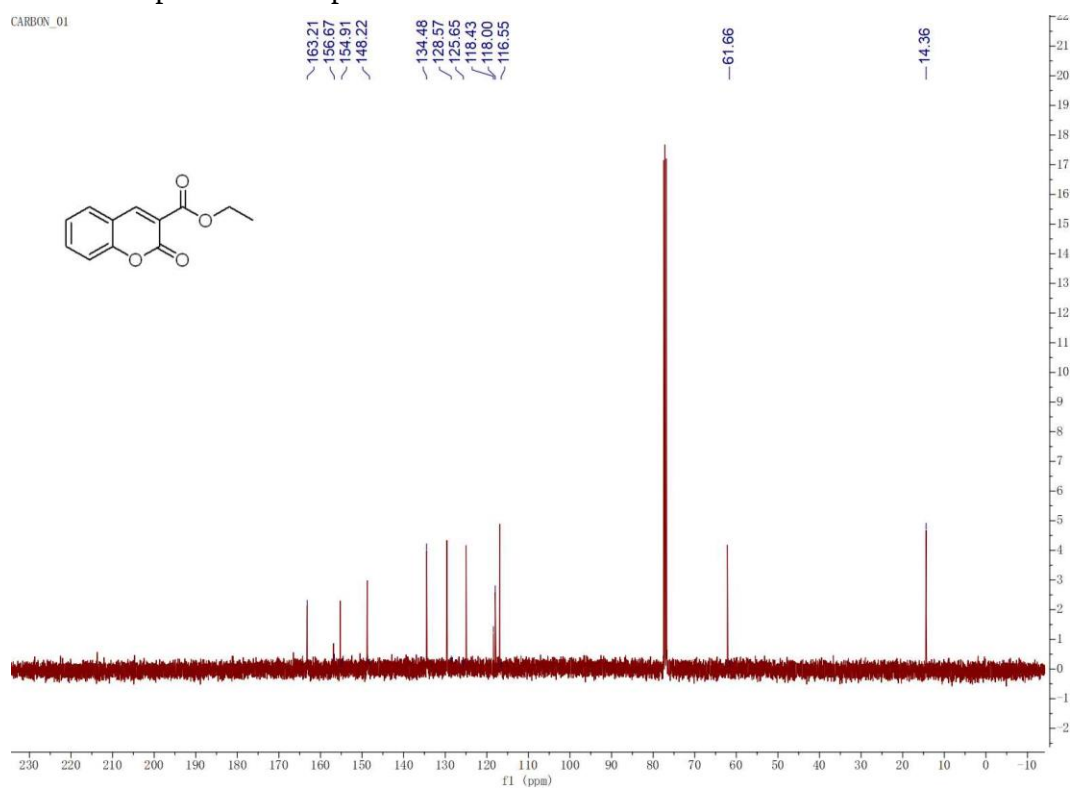
¹³C NMR spectra of compound 3:



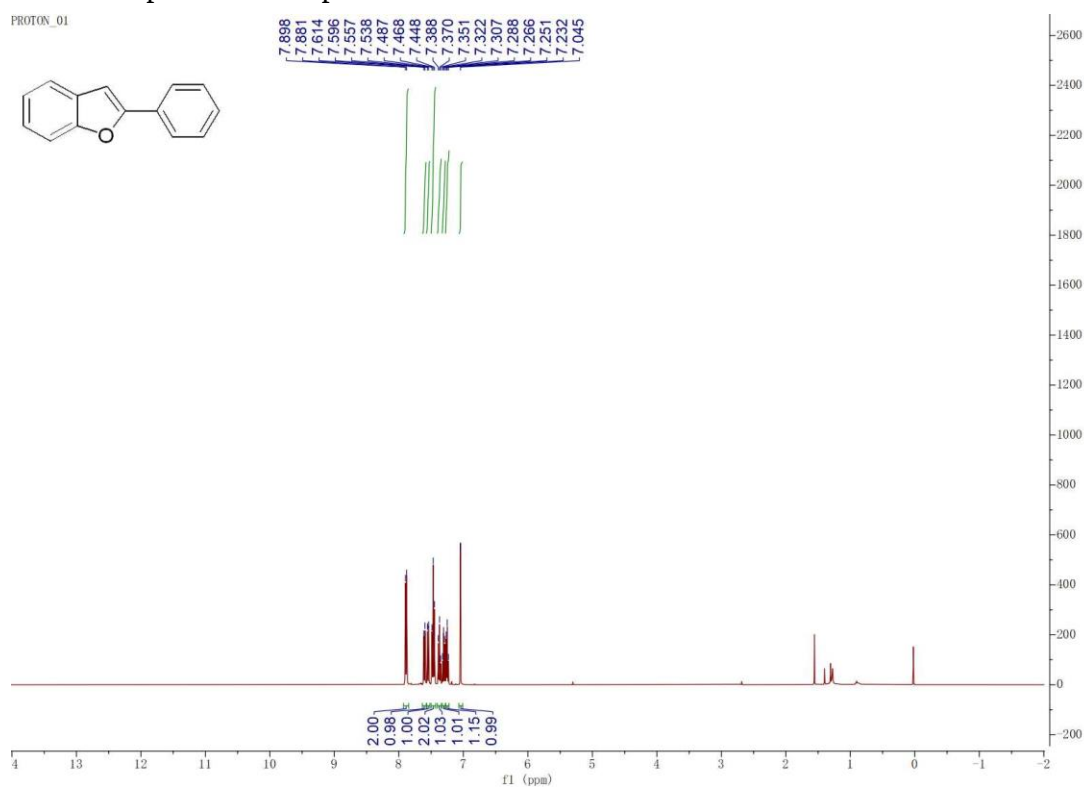
¹H NMR spectra of compound 4:



¹³C NMR spectra of compound 4:



¹H NMR spectra of compound 5:



¹³C NMR spectra of compound 5:

