

Supplementary

Selective Isotropic Etching of SiO₂ over Si₃N₄ using NF₃/H₂ Remote Plasma and Methanol Vapor

**Hong Seong Gil^{1,=}, Doo San Kim^{1,=}, Yun Jong Jang¹, Dea Whan Kim³, Hea In Kwon¹,
Gyoung Chan Kim¹, Dong Woo Kim^{1,*}, and Geun Young Yeom^{1,2,*}**

¹ School of Advanced Materials Science and Engineering, Sungkyunkwan University, Suwon 16419, Republic of Korea

² SKKU Advanced Institute of Nano Technology (SAINT), Sungkyunkwan University, Suwon 16419, Republic of Korea

³ Department of Semiconductor Display Engineering, Sungkyunkwan University, Suwon 16419, Republic of Korea

=These authors contributed equally to this work.

*E-mail: dwkim111@gmail.com, gyyeom@skku.edu

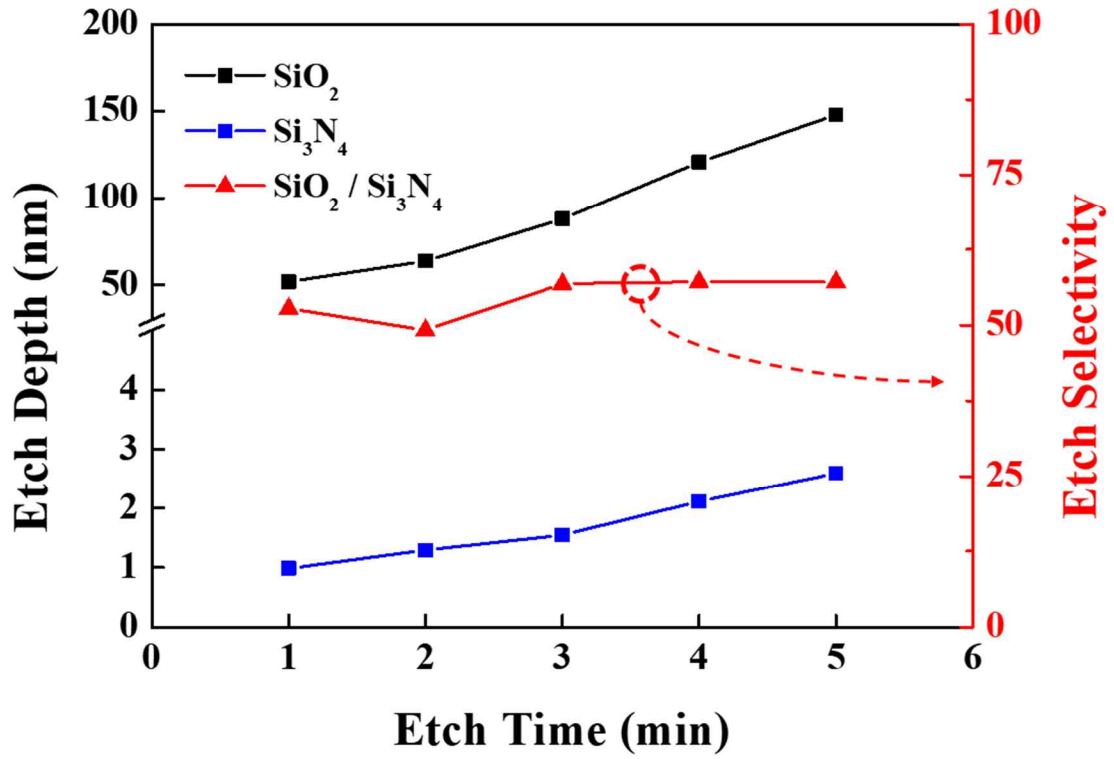


Figure S1. Wet etch rates of SiO₂ and Si₃N₄ samples used in the experiment and their etch selectivities by etching in a HF solution (HF:deionized water = 1:100, RT).

Figure S1 shows the etch rates of Si₃N₄ and SiO₂ used in this experiment with 1 % HF solution and, as shown in Figure S1, the etch rate of SiO₂ was ~ 30 nm/min and the etch selectivity of SiO₂ over Si₃N₄ was ~ 50.

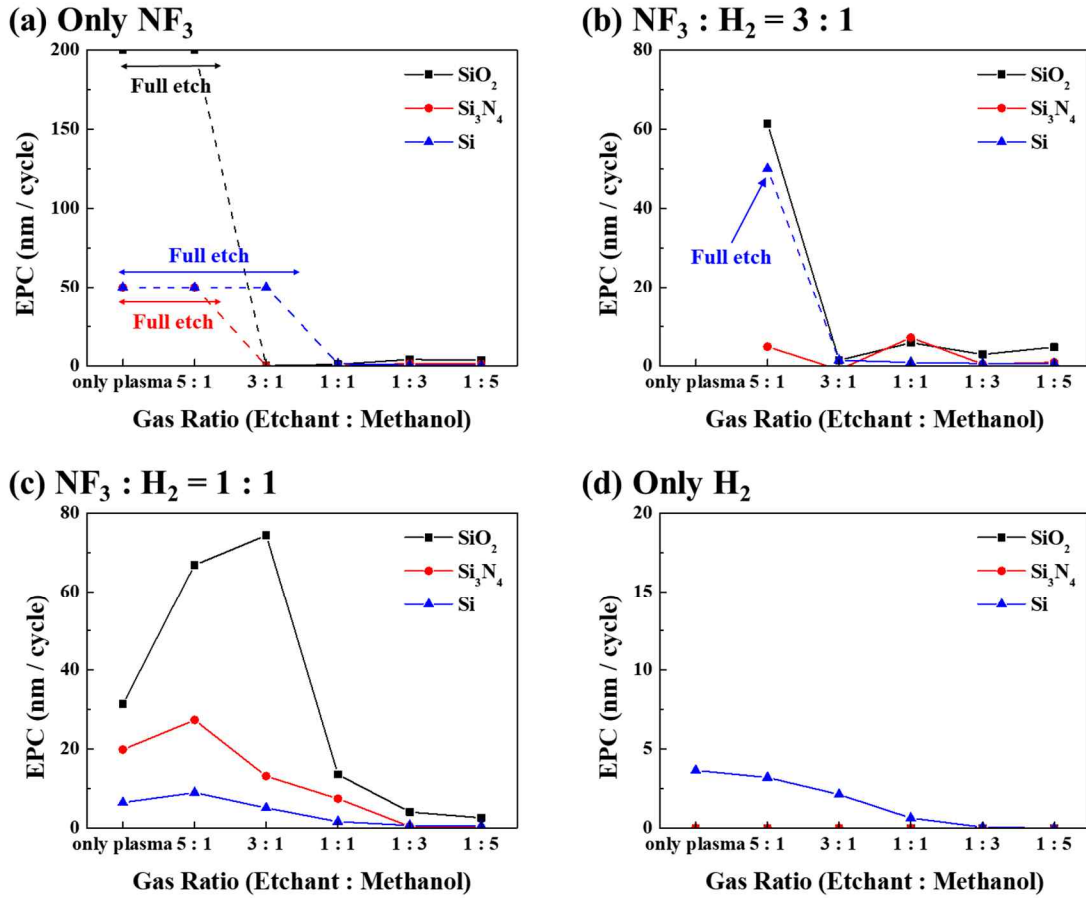


Figure S2. EPCs as a function of the partial pressure ratio between NF_3/H_2 and methanol for different ratios of $\text{NF}_3 : \text{H}_2$; (a) only NF_3 plasma, (b) $\text{NF}_3 : \text{H}_2$ ratio of 3 : 1, (c) $\text{NF}_3 : \text{H}_2$ ratio of 1 : 1, and (d) only H_2 plasma. Other process conditions are the same as those in Figure 2(b).

Figure S2 shows EPCs as a function of the partial pressure ratio between NF_3/H_2 and methanol for different ratios of $\text{NF}_3:\text{H}_2$. As shown in Figure S2(a), for NF_3 only, 200 nm thick SiO_2 and 50 nm thick Si_3N_4 and Si were etched away within one cycle when the ratio of NF_3 :methanol was lower than 5:1, however, when the ratio of NF_3 :methanol was higher, no etching was observed. Similar results were obtained for the NF_3 -rich conditions such as $\text{NF}_3:\text{H}_2 = 3:1$ as shown in Figure S2(b). However, for the conditions of $\text{NF}_3:\text{H}_2 = 1:1$, the EPC was the highest when both NF_3/H_2 and methanol were abundant as shown in Figure S2(c). For the H_2 rich conditions, only silicon was etched as shown in Figure S2(d).

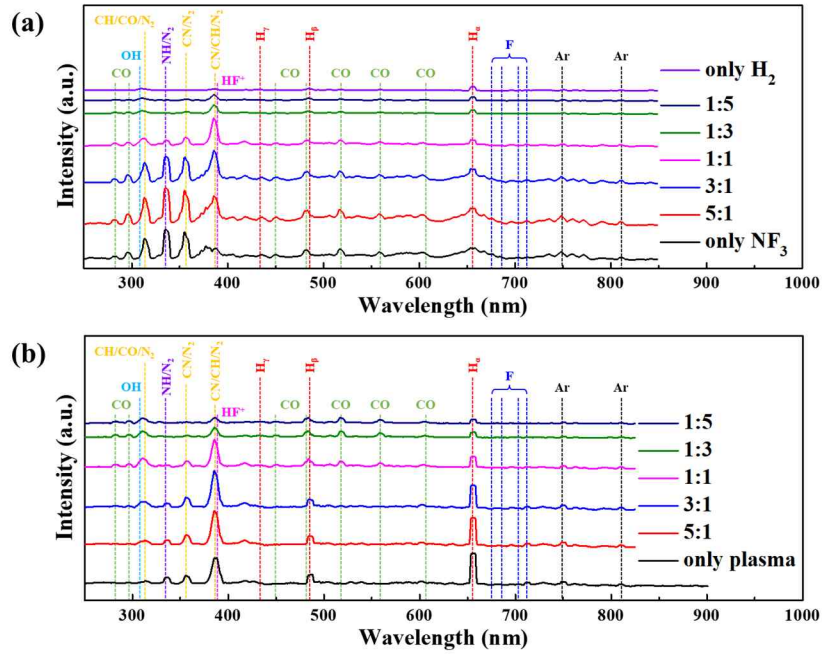


Figure S3. OES spectra in plasma region measured as a function of (a) NF₃ : H₂ gas ratio and (b) NF₃/H₂ : methanol ratio for the conditions in Figure 2(a) and (b), respectively.

Figure S3 shows the raw OES data in plasma region after the removal of backgrounds measured as a function of (a) NF₃:H₂ gas ratio and (b) NF₃/H₂:methanol ratio for Figure 4(a) and (b), respectively. For Ar actinometry, small Ar was added in the plasma and the Ar peaks could be measured at 750.4 nm and 811.5 nm. [1-6] Methanol related peaks were observed; OH peak at 309 nm [2,7], CO peaks at 297.7, 451, 483.5, 519.8, 559, 607 nm, etc. [4,8-10], NH peak at 336 nm, and CN peaks at 357, 388 nm. [4,6,8,9,11] N₂ peak was observed in the region of 300-400 nm, [4-8,11] H peaks at 434, 486, 656 nm, and F peaks at 677, 685.6, 703.7, 712.8 nm. [2-5,7] For Ar actinometry, Ar peak at 750 nm was used with H peak at 656 nm, F peak at 703.7 nm, NH at 336 nm, CN/CH at 388 nm, and CO at 519 nm.

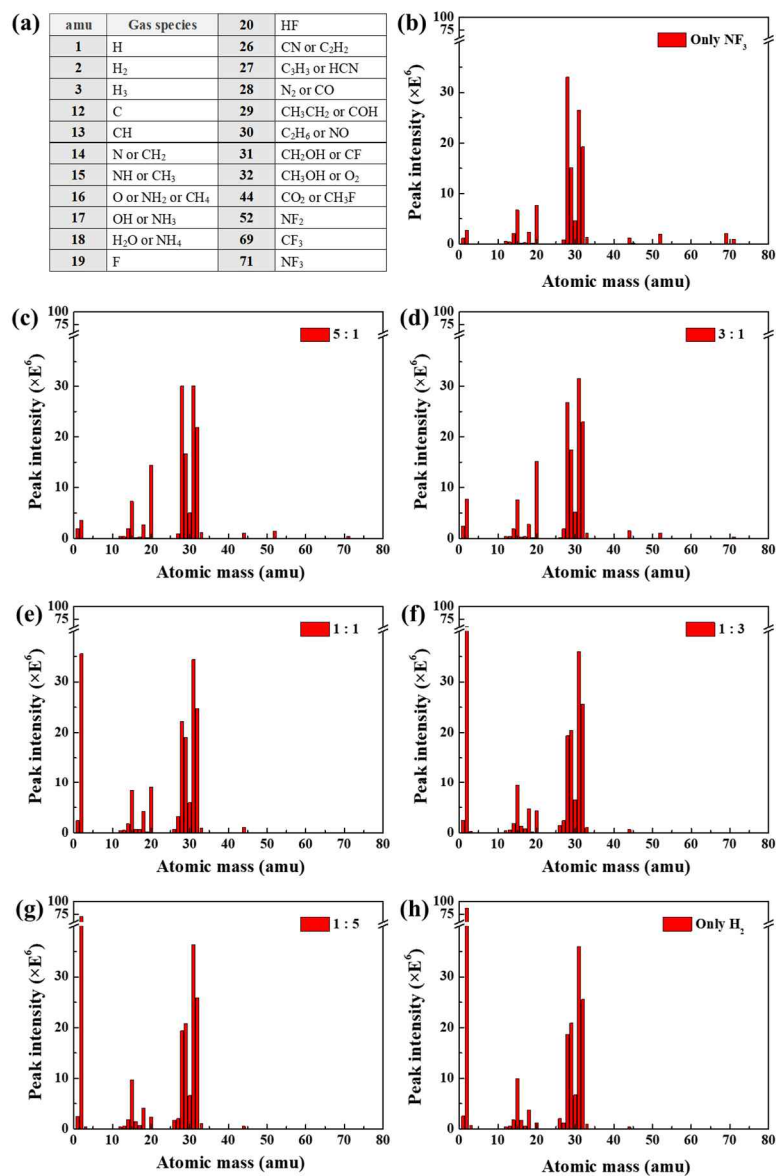


Figure S4. RGA spectra in main chamber region for different NF₃ : H₂ gas ratios; (a) species according to amu, (b) only NF₃ condition, (c) 5 : 1 condition, (d) 3 : 1 condition, (e) 1 : 1 condition, (f) 1 : 3 condition, (g) 1 : 5 condition, and (h) only H₂ condition.

Figure S4 shows RGA spectra in main chamber region for different NF₃ : H₂ gas ratios. Atomic species such as H and F which could be observed by OES in the plasma region were not observed in the process chamber region possibly due to recombination. Also, possibly due to the ionization in the RGA, dissociated species such as CH₂OH, NF₂ representing CH₃OH, NF₃ were observed.

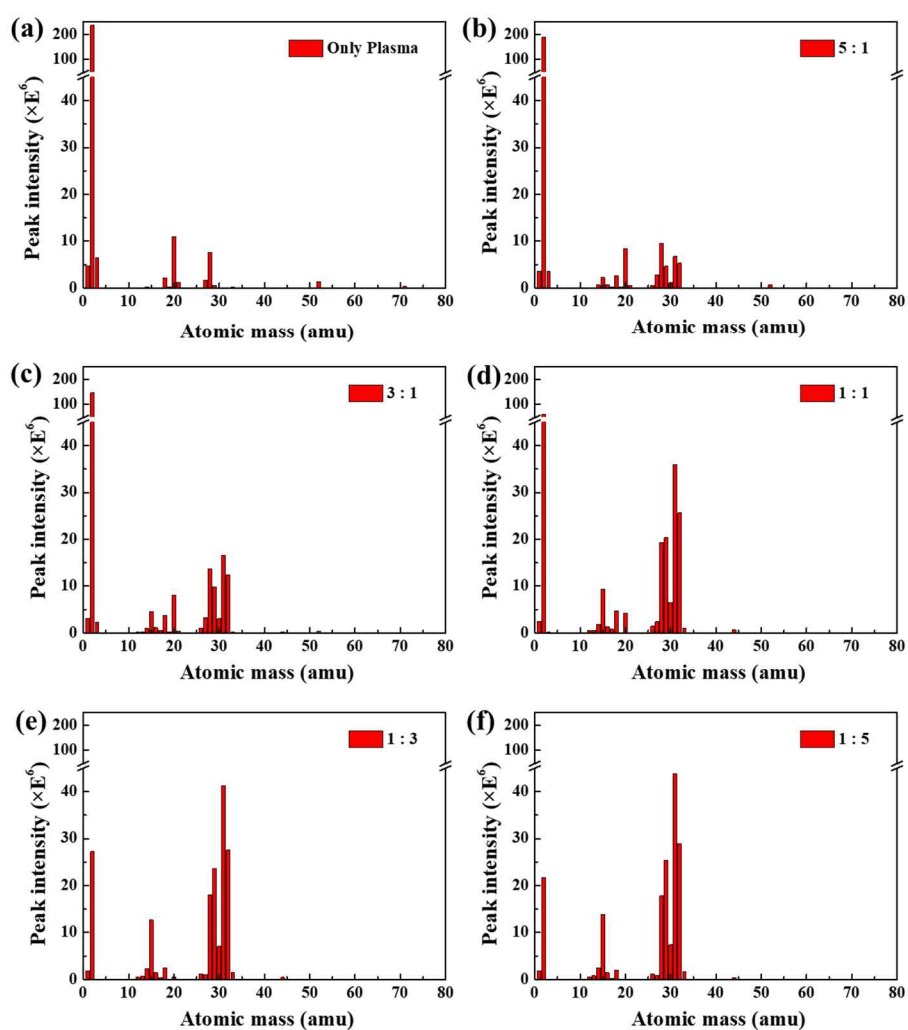


Figure S5. RGA spectra in main chamber region for different NF_3/H_2 : methanol ratios; (a) only plasma condition, (b) 5 : 1 condition, (c) 3 : 1 condition, (d) 1 : 1 condition, (e) 1 : 3 condition, and (f) 1 : 5 condition.

Figure S5 shows RGA spectra in main chamber region for different NF_3/H_2 : methanol ratios.

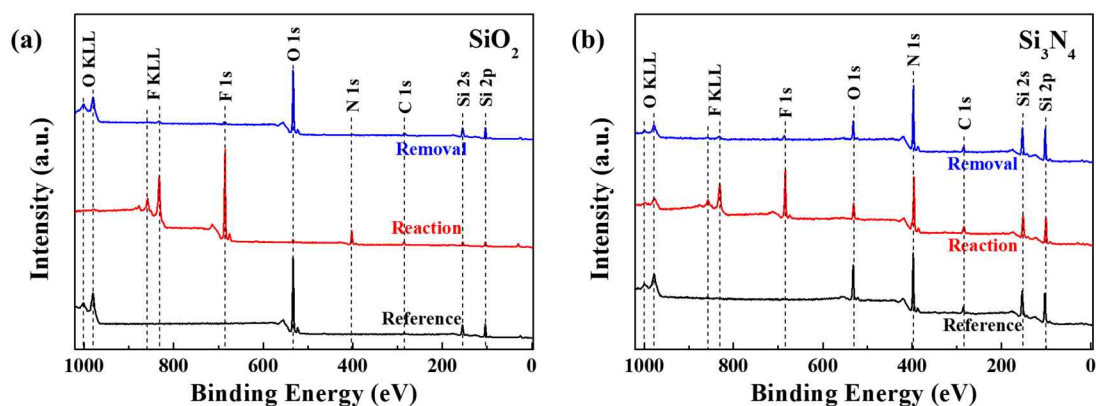


Figure S6. Wide scan XPS data on the surfaces of (a) SiO_2 and (b) Si_3N_4 for each process step of cyclic etching under optimized conditions in Figure 3.

Figure S6 shows the wide scan XPS data for results in Figure 8. As shown in Figure S6(a), on the SiO_2 surface, after the reaction step, F and N peaks were emerged and, after the removal step, those peaks were removed in addition to the increase of O peak. In the case of Si_3N_4 , as shown in Figure S6(b), the change of F after each process step similar to that on SiO_2 surface was observed on the Si_3N_4 surface. [12-15]

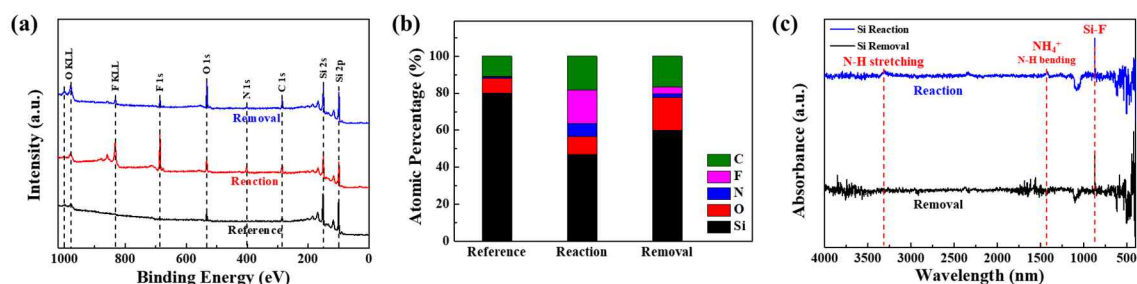


Figure S7. surface characteristics of poly Si measured by XPS and FTIR for each process step of cyclic etching under optimized conditions in Figure 3; (a) XPS wide scan data, (b) surface atomic percentages from (a), and (c) FTIR data.

Figure S7 shows surface characteristics of poly Si measured by XPS and FTIR for each process step of cyclic etching under optimized conditions in Figure 3. As shown in XPS results in Figure S7(a) and (b), the increase of N, O, and F peaks was observed after the reaction step, and, after the removal step, the decrease of N and F peaks was observed but with the increase of O peak possibly due to the oxidation of poly-silicon surface. The FTIR results in Figure S7(c) also showed the slight increase of N-H bonding after the reaction step.

Supporting References

- [1] D. H. Kim, J. E. Choi, S. J. Hong, Analysis of optical emission spectroscopy data during silicon etching in SF₆/O₂/Ar plasma, *Plasma Sci. Technol.* 23 (2021), 125501. <https://doi.org/10.1088/2058-6272/ac24f4>
- [2] Q. Lou, S. Kaler, V. M. Donnelly, D. J. Economou, Optical emission spectroscopic studies and comparisons of CH₃F/CO₂ and CH₃F/O₂ inductively coupled plasmas, *J. Vac. Sci. Technol. A* 33 (2015), 021305. <http://dx.doi.org/10.1116/1.4904213>
- [3] A. A. Osipov, A. A. Osipov, V. I Berezenko, S. E. Alexandrov, OES diagnostic of SF₆/Ar gas mixture of ICP discharges for LiNbO₃ etching, *IOP Conf. Ser.: Mater. Sci. Eng.* 919 (2020), 022018. <https://doi.org/10.1088/1757-899X/919/2/022018>
- [4] W. S. Song, J. E. Kang, S. J. Hong, Spectroscopic Analysis of CF₄/O₂ Plasma Mixed with N₂ for Si₃N₄ Dry Etching, *Coatings*, 12 (2022), 1064. <https://doi.org/10.3390/coatings12081064>
- [5] S. An, S. J. Hong, Spectroscopic Analysis of NF₃ Plasmas with Oxygen Additive for PECVD Chamber Cleaning, *Coatings*, 13 (2023), 91. <https://doi.org/10.3390/coatings13010091>
- [6] S. J. Kang, V. M. Donnelly, Optical absorption and emission spectroscopy studies of ammonia-containing plasmas, *Plasma Sources Sci. Technol.* 16 (2007), 265-272. <https://doi.org/10.1088/0963-0252/16/2/008>
- [7] J. Lee, K. Kim, Y. S. Kim, A Study on the NF₃ Plasma Etching Reaction with Cobalt Oxide Grown on Inconel Base Metal Surface, *Plasma. Chem. Plasma. Process.* 39 (2019), 1145–1159. <https://doi.org/10.1007/s11090-019-09979-4>
- [8] H. W. Liu, T. H. Chen, C. H. Chang, S. K. Lu, Y. C. Lin, D. S. Liu, Impact on the Gas Barrier Property of Silicon Oxide Films Prepared by Tetramethylsilane-Based PECVD Incorporating with Ammonia, *Appl. Sci.* 7 (2017), 56. <https://doi.org/10.3390/app7010056>
- [9] S. Zhang, X. Zeng, H. Bai, C. Zhang, T. Shao, Optical emission spectroscopy measurement of plasma parameters in a nanosecond pulsed spark discharge for

CO₂/CH₄ dry reforming, *Spectrochim. Acta A Mol. Biomol. Spectrosc.* 267 (2022), 120590. <https://doi.org/10.1016/j.saa.2021.120590>

- [10] S. Yoo, D. Seok, Y. Jung, K. Lee, Hydrophilic Surface Treatment of Carbon Powder Using CO₂ Plasma Activated Gas. *Coatings*. 11 (2021), 925. <https://doi.org/10.3390/coatings11080925>
- [11] H. C. Hsueh, H. C. Li, D. Chiang, S. Lee, Effects of Ammonia/Methane Mixtures on Characteristics of Plasma Enhanced Chemical Vapor Deposition n-Type Carbon Films, *J. Electrochem. Soc.* 159 (2011), D77. <https://doi.org/10.1149/2.051202jes>
- [12] B. Liu, W. Xu, J. Tao, P. Yan, J. Zheng, M. H. Engelhard, D. Lu, C. Wang, J. G. Zhang, Enhanced Cyclability of Lithium–Oxygen Batteries with Electrodes Protected by Surface Films Induced via In Situ Electrochemical Process, *Adv. Energy Mater.* 8 (2018), 1702340. <https://doi.org/10.1002/aenm.201702340>
- [13] R. Dalmau, R. Collazo, S. Mita, Z. Sitar, X-Ray Photoelectron Spectroscopy Characterization of Aluminum Nitride Surface Oxides: Thermal and Hydrothermal Evolution. *J. Electron. Mater.* 36 (2007), 414–419. <https://doi.org/10.1007/s11664-006-0044-x>
- [14] M. Fantauzzi, B. Elsener, D. Atzei, A. Rigoldiab, A. Rossi, Exploiting XPS for the identification of sulfides and polysulfides, *RSC Adv.* 5 (2015), 75953. <https://doi.org/10.1039/c5ra14915k>
- [15] K. T. Wang, W. Y. Wang, T. C. Wei, Photomask-Free, Direct Selective Electroless Deposition on Glass by Controlling Surface Hydrophilicity, *ACS Omega*, 4 (2019), 7706–7710. <http://pubs.acs.org/journal/acsodf>