

# Viscous fingering in volatile fluid towards spontaneous evolution of well-ordered 3D patterns

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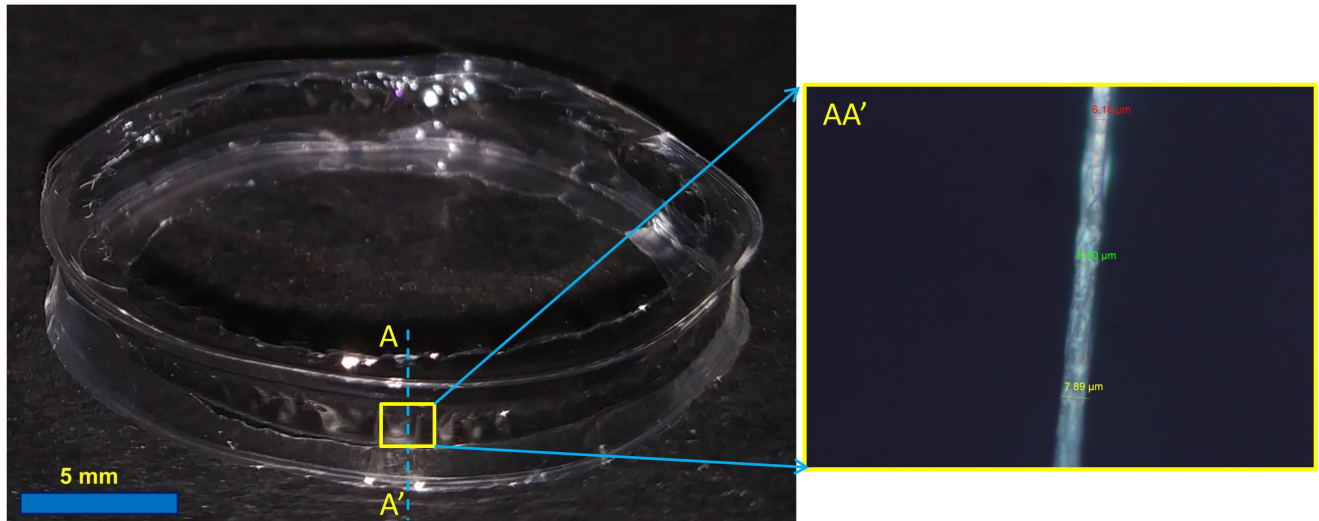
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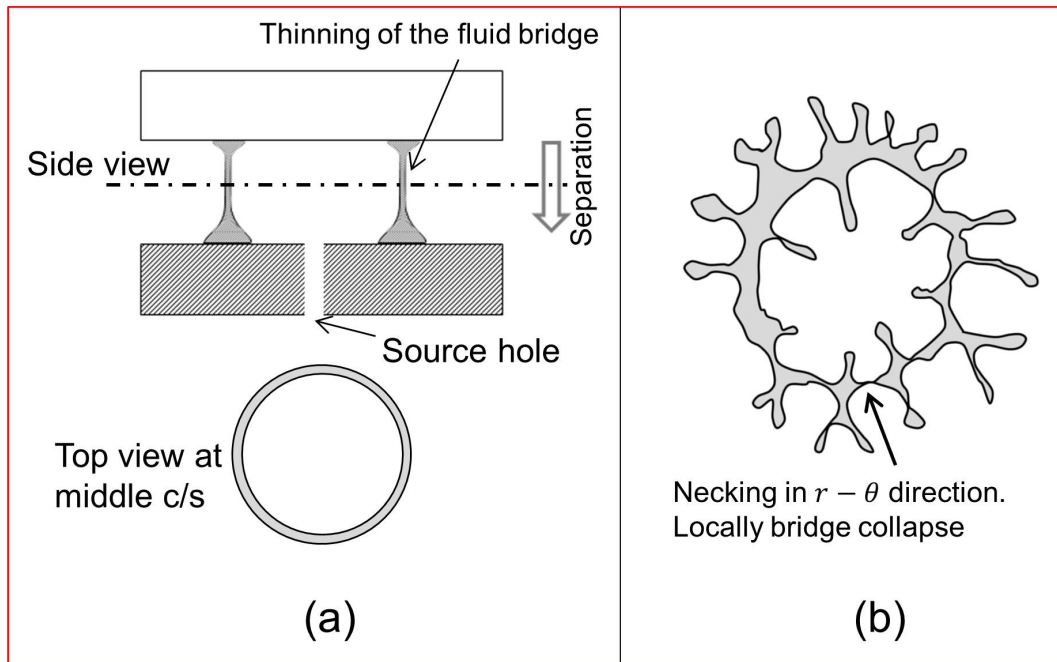
**Supplementary Movie. 1.** Experiment shown in Figure 5.O<sub>1</sub> :The supplementary movie shows stable evolution followed by fluid ring retraction hence structure is not retained(No retention).

**Supplementary Movie. 2.** Experiment shown in Figure 6.B<sub>1'</sub>:The supplementary movie shows a sudden bridge breaking.

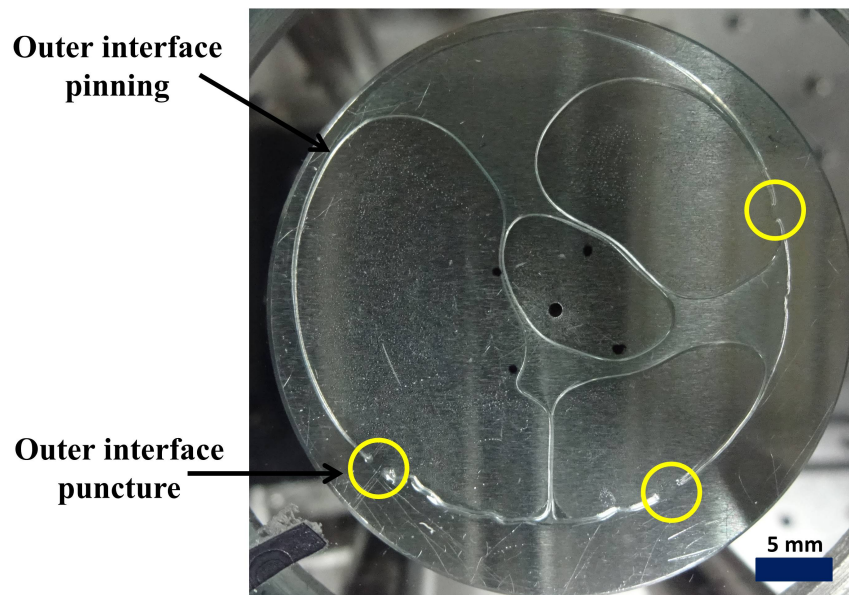
**Supplementary Movie. 3.** Experiment shown in Figure 6.B<sub>2'</sub> :The supplementary movie shows localized bridge breaking during the unstable evolution stage.



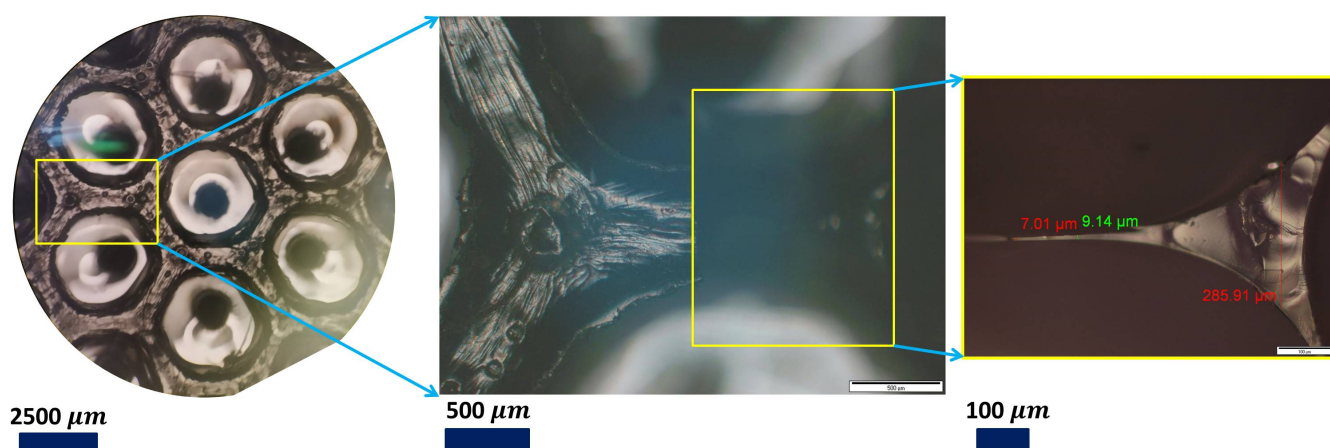
**Supplementary Fig. 1.** Wall thickness at the mid-plane of the hollow cylindrical well fabricated on ULHSC. Left side image showing microscopic view of the cross-section(AA') of the wall at the mid-plane(highlighted section)



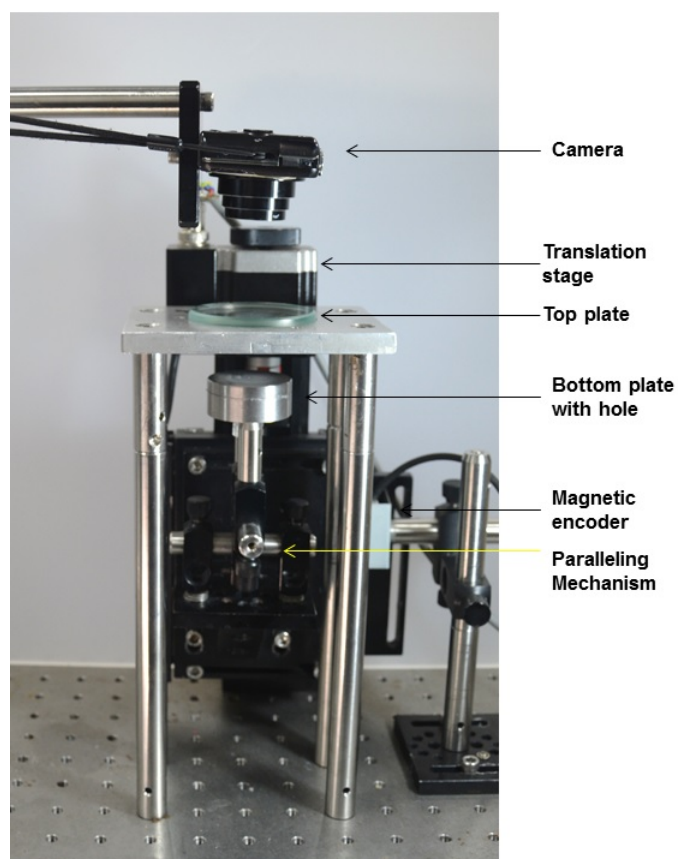
**Supplementary Fig. 2.** Bridge breaking schematic a) Thining of the fluid bridge during plate separation b) Necking in azimuthal direction during viscous fingering



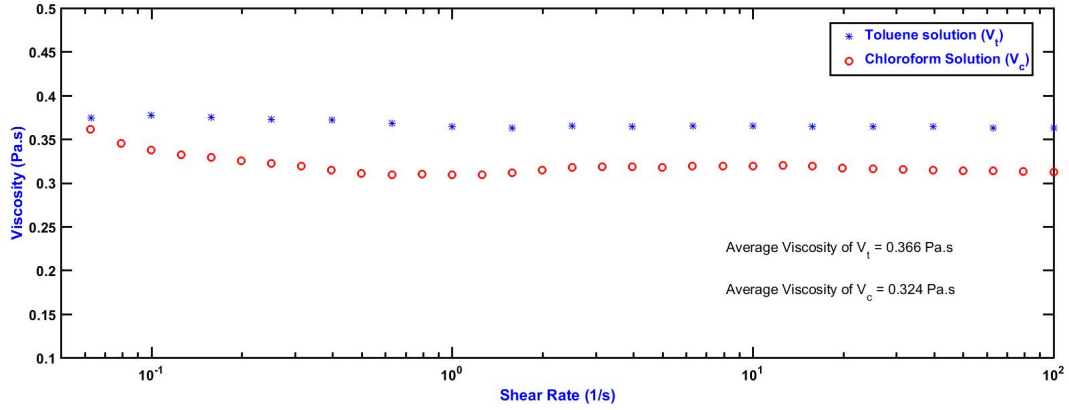
**Supplementary Fig. 3.** Pinning and Puncture of the outer interface



**Supplementary Fig. 4.** Wall thickness of the mid-plane of the HAR hexagonal mesh fabricated on MLHSC. In left side image focus is on top plane and in right side image focus is on mid-plane of highlighted section.



**Supplementary Fig. 5.** Setup

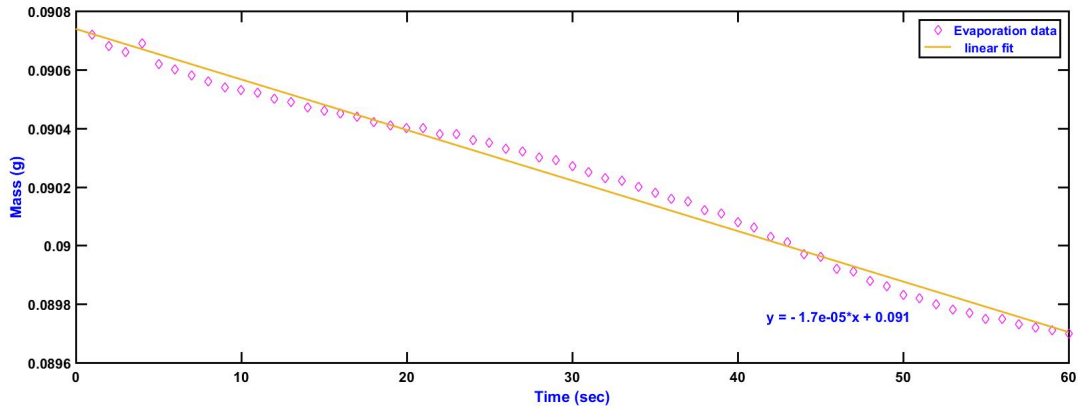


**Supplementary Fig. 6.** Viscosity of the chloroform solution( $V_c$ ) and toluene solution( $V_t$ ) .

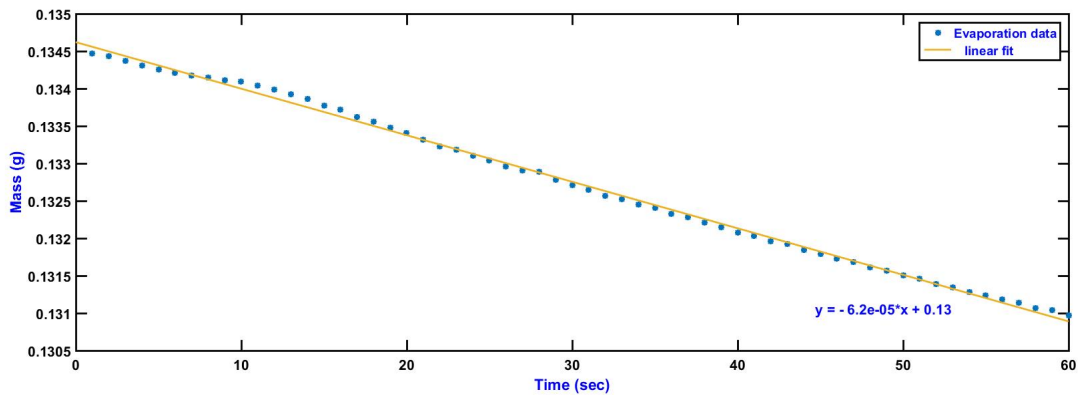
### Evaporation rate calculation

Mass flux in  $Kg/m^2s$  = Evaporation rate in  $Kg/s$   $\div$  exposing area to atmosphere in  $m^2$

Evaporation rate in m/s = mass flux in  $Kg/m^2s$   $\div$  density of the fluid in  $Kg/m^3$



**Supplementary Fig. 7.** Evaporation rate(g/sec) of the toluene solution( $V_t$ ).



**Supplementary Fig. 8.** Evaporation rate(g/sec) of the chloroform solution( $V_c$ ).