

Elucidating the Reaction Pathways of Veratrylglycero- β - guaiacyl Ether Degradation Over Metal-Free Solid Acid Catalysts with Hydrogen

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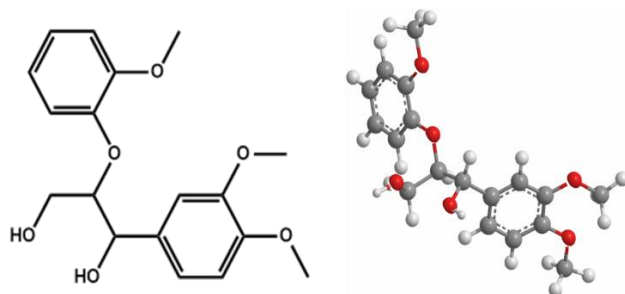


Figure S1. 2D and 3D optimized structure of the β -O-4 model compound (veratrylglycero- β -guaiacyl ether) .

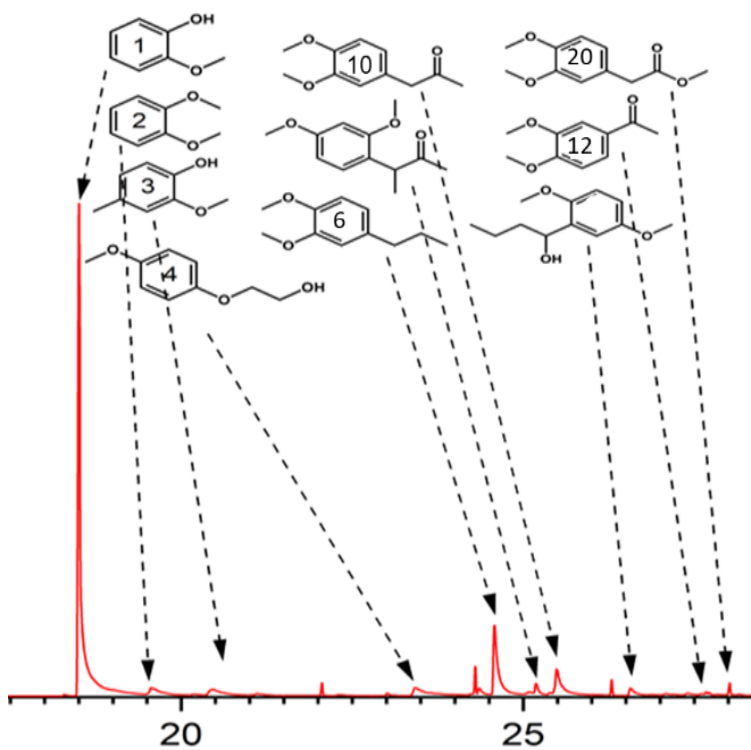


Figure S2. GC-MS chromatogram of the products generated during the zeolite cracking of β -O-4 model compound over CBV 400.

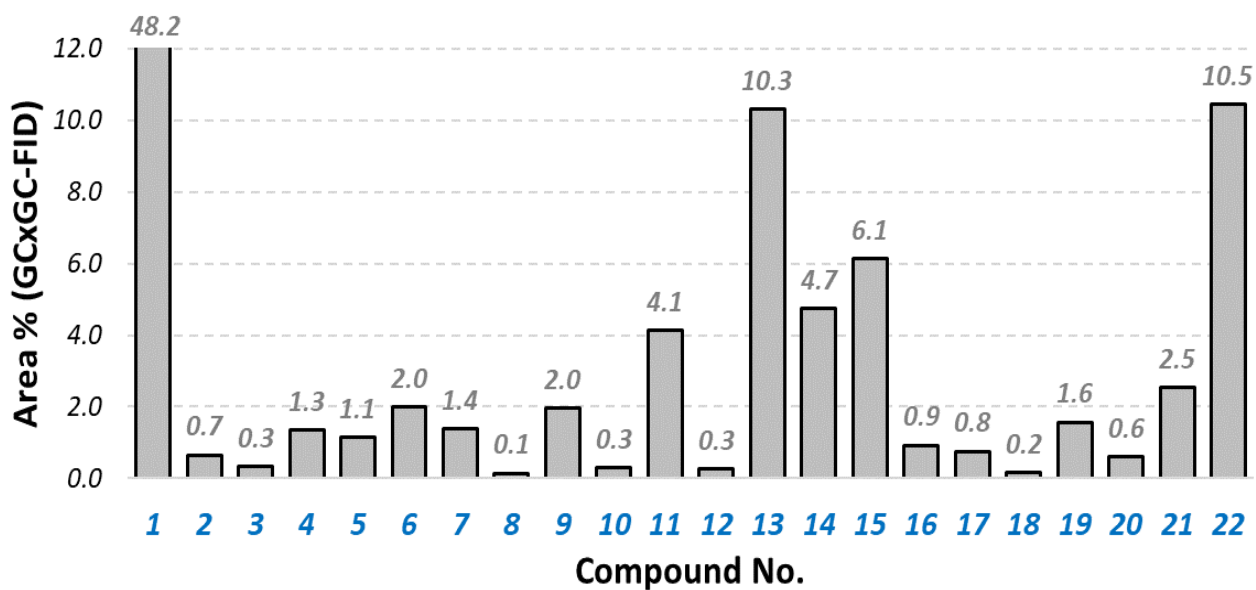
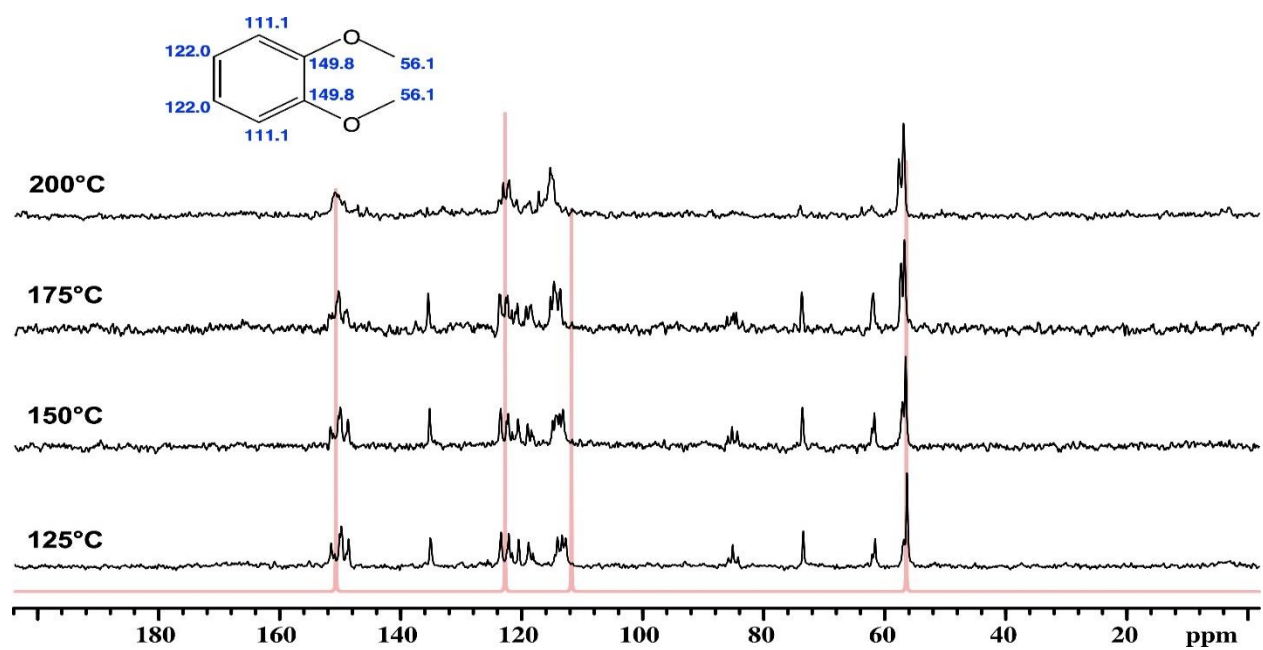
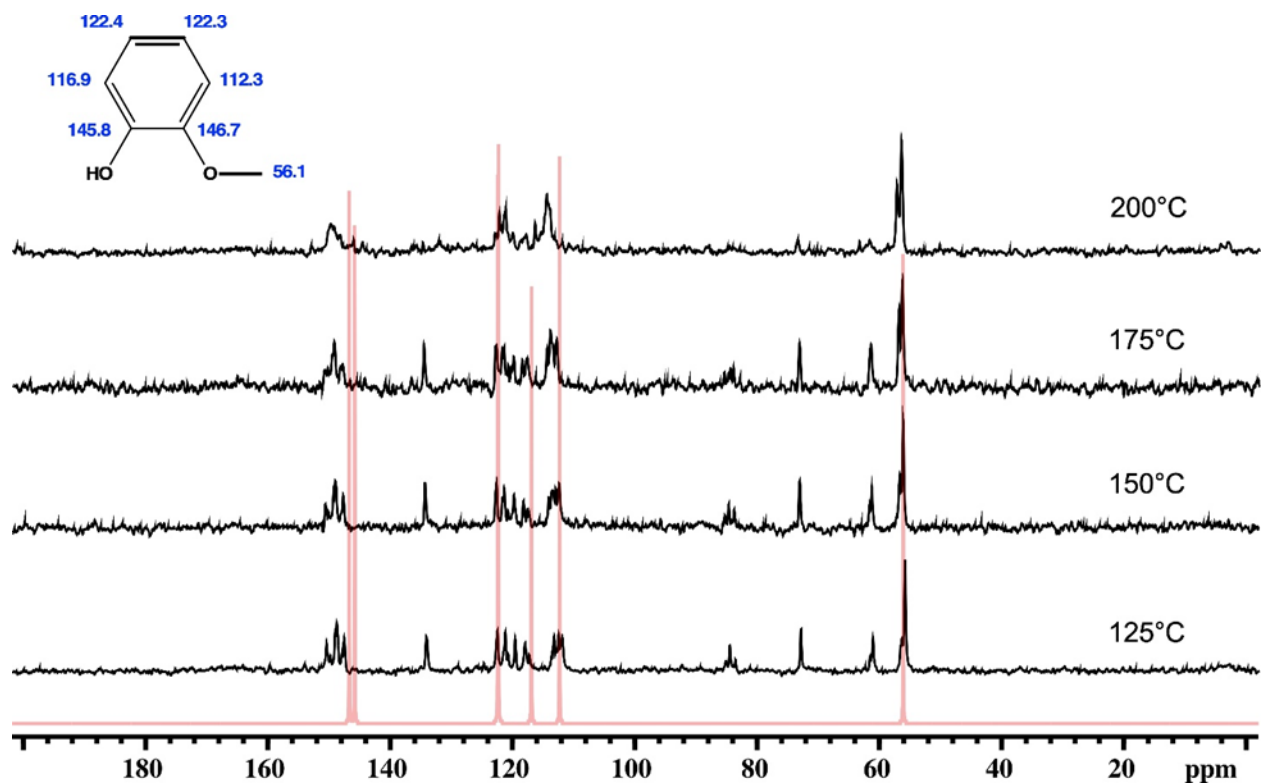
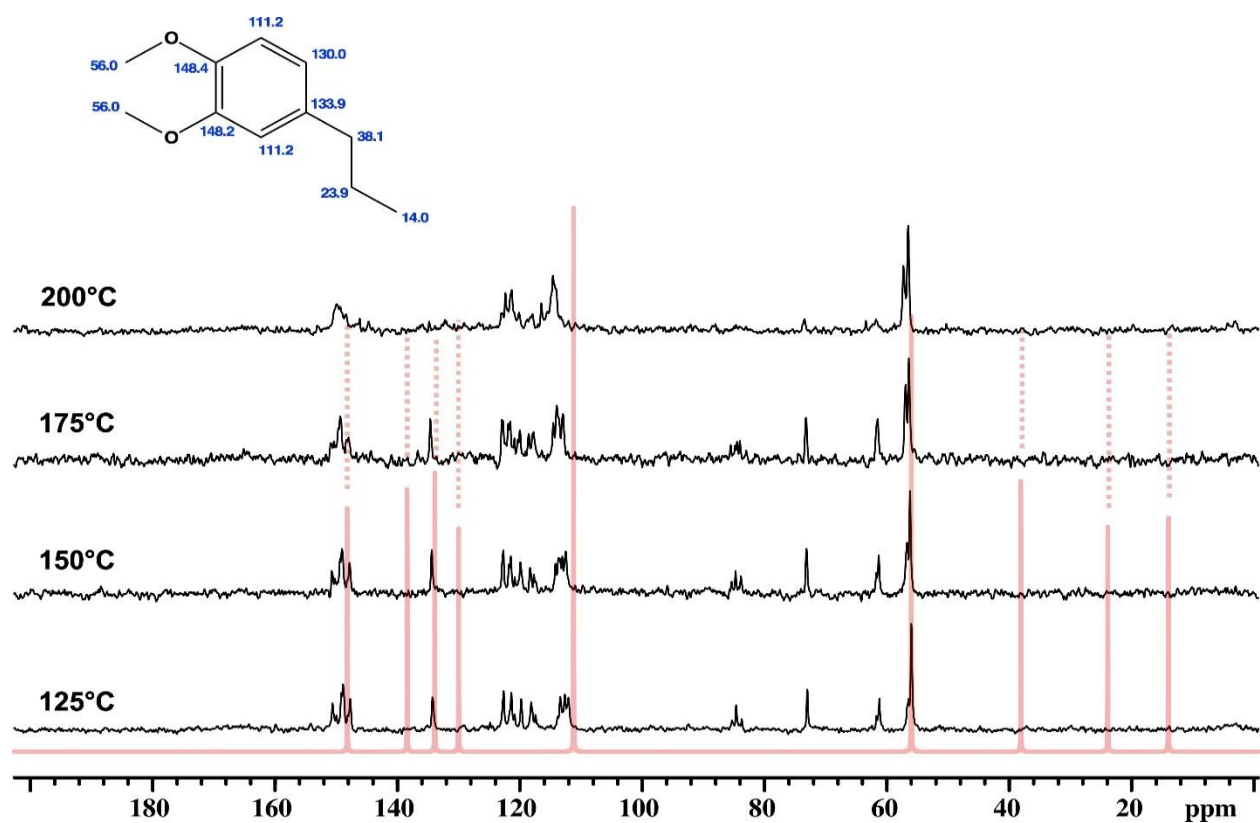
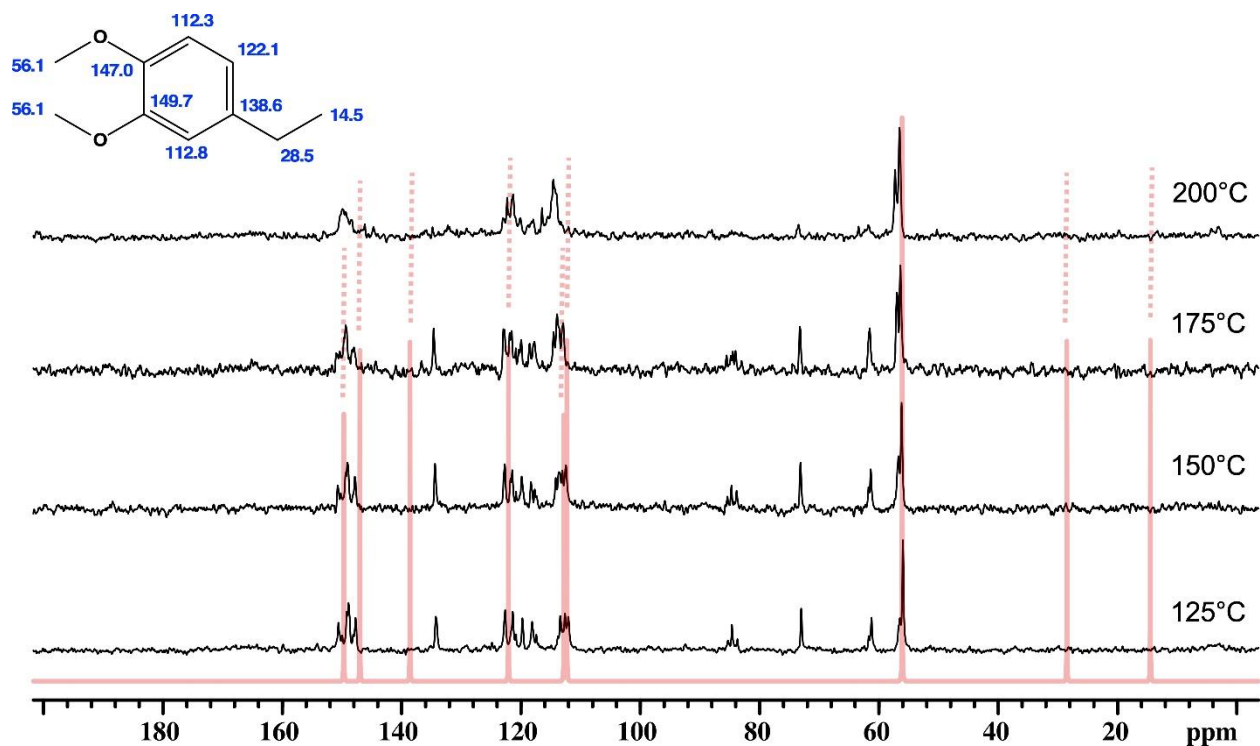
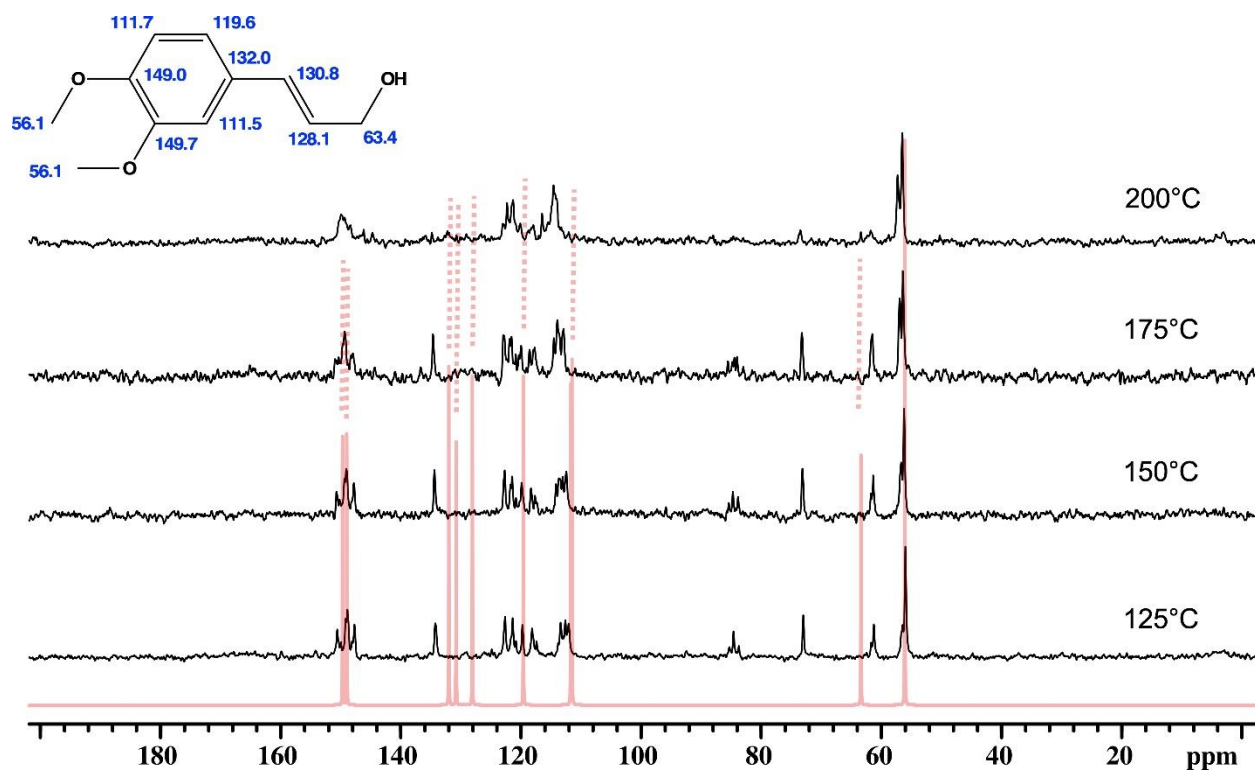
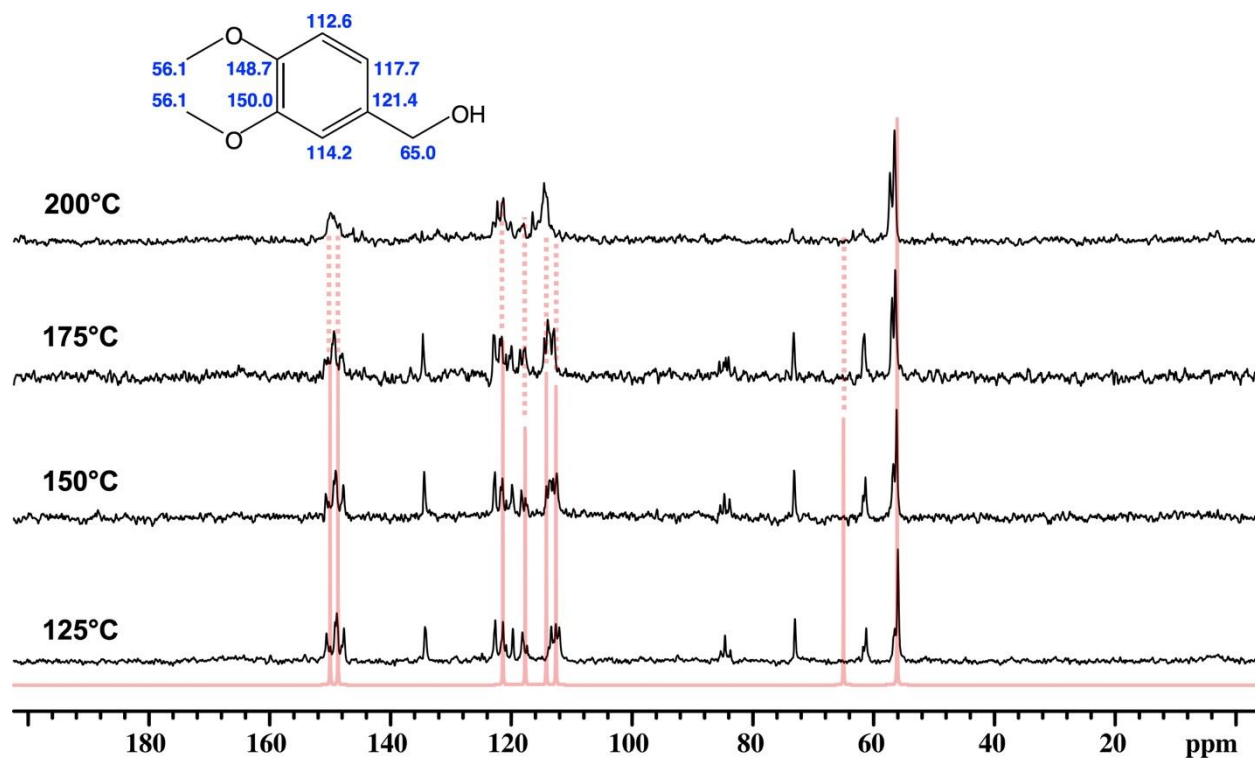
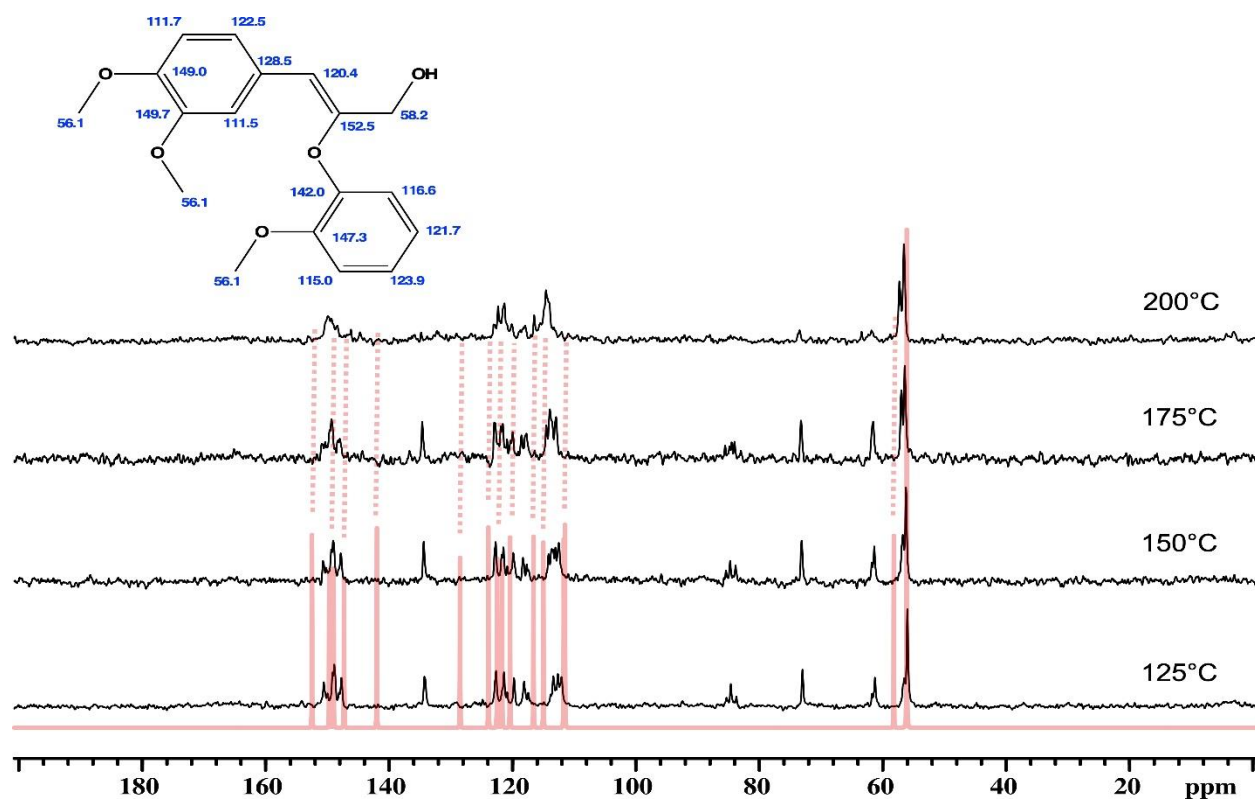
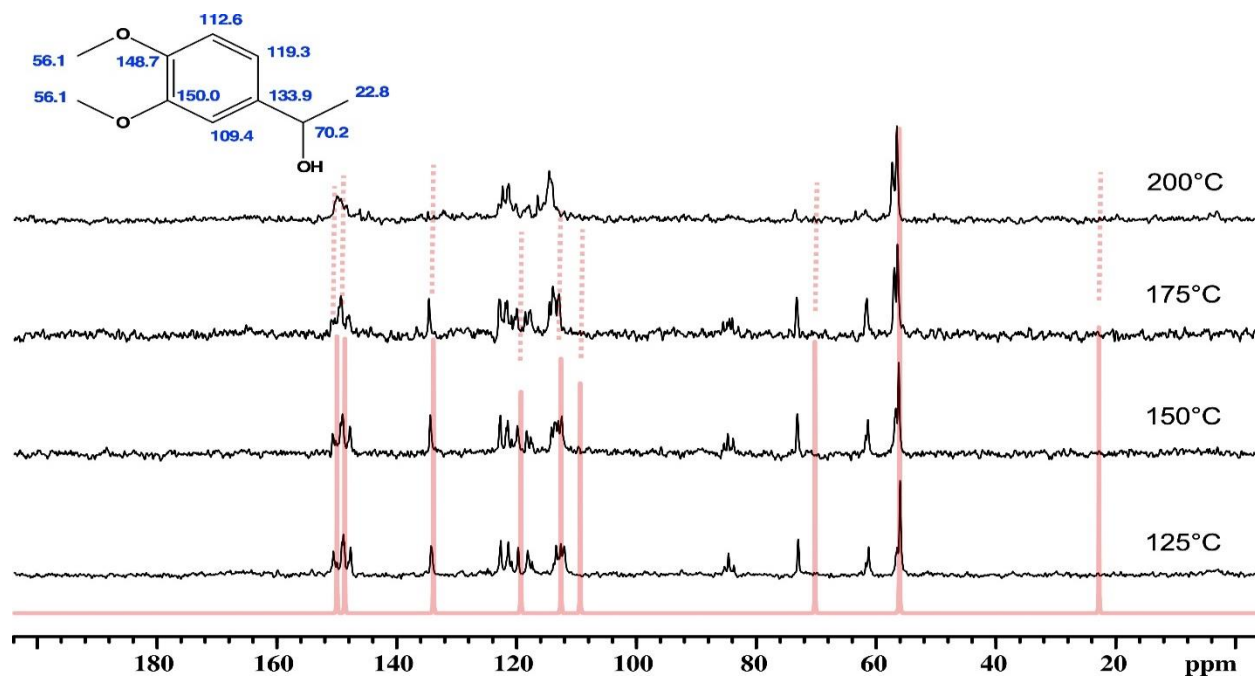


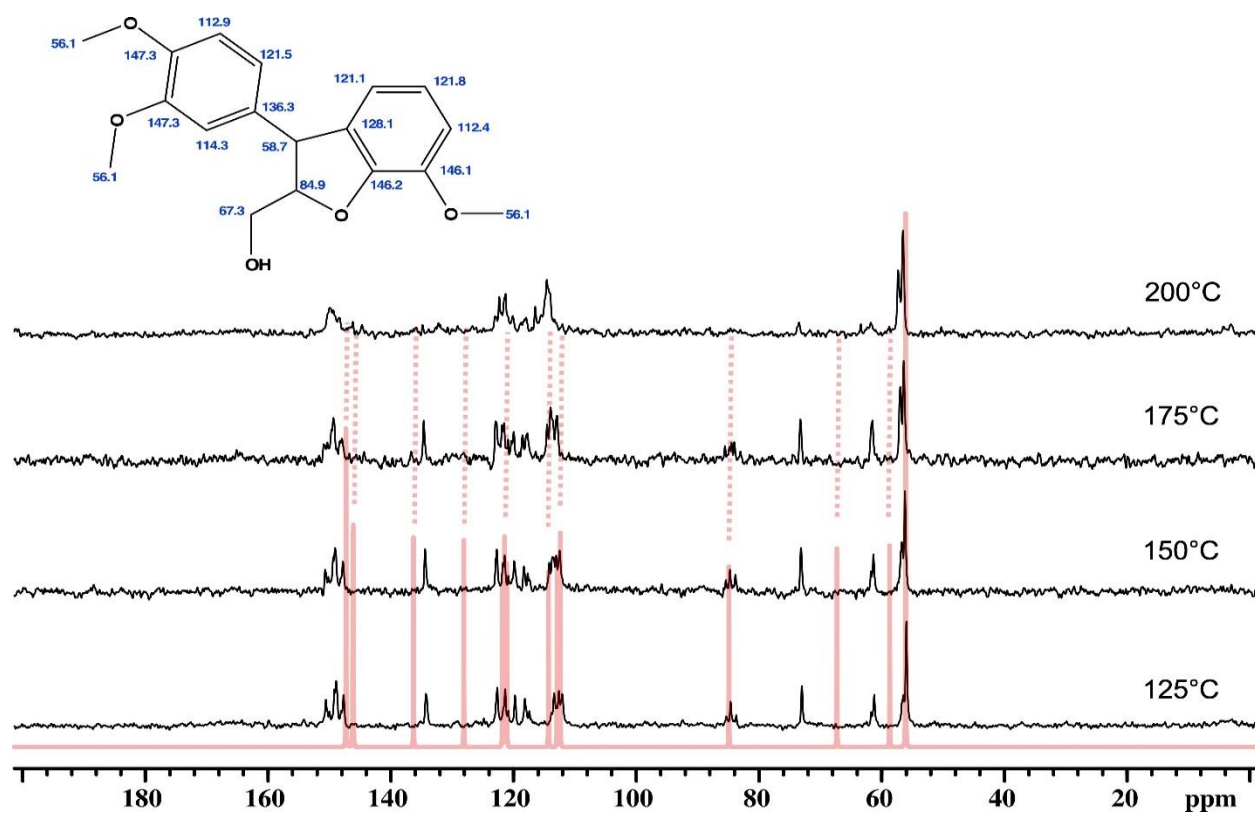
Figure S3: Semi-qualitative results of identified products as reported by GC $\times$ GC-FID (in area%) from zeolite decomposition of β -O-4 model compound.











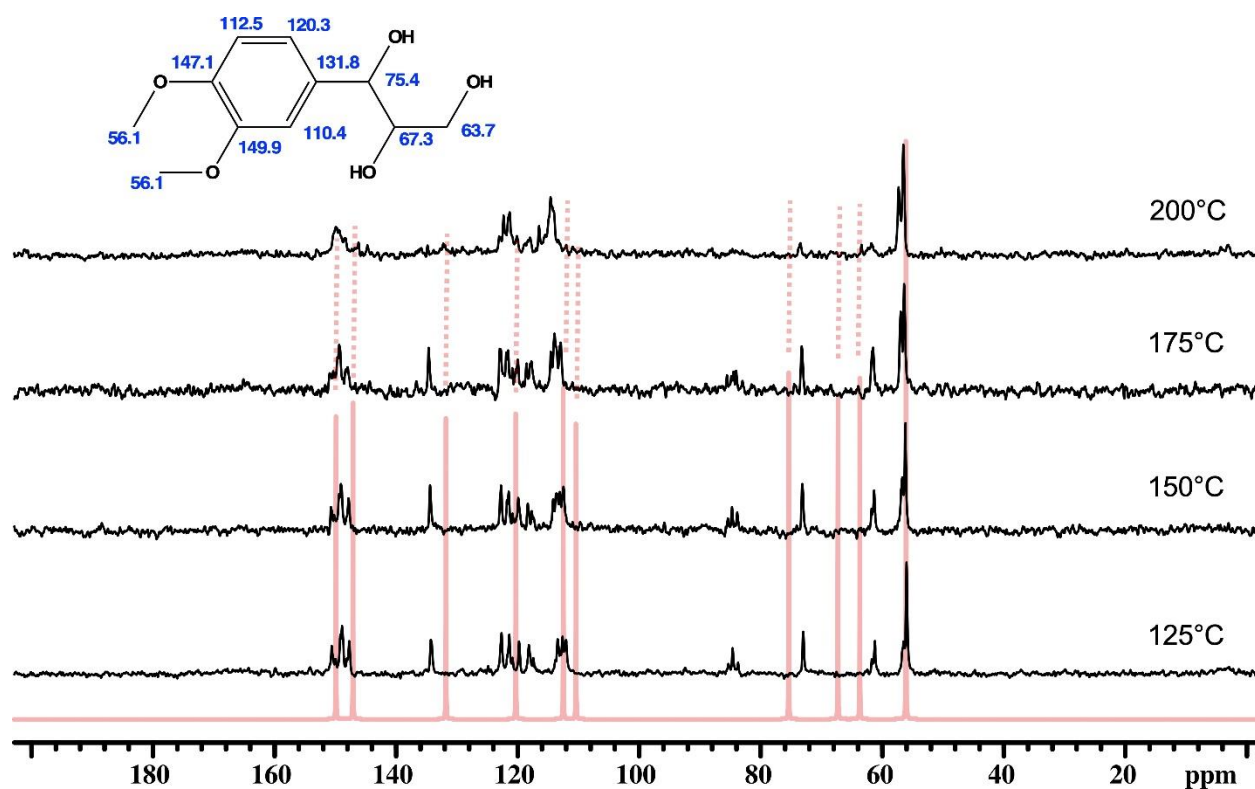
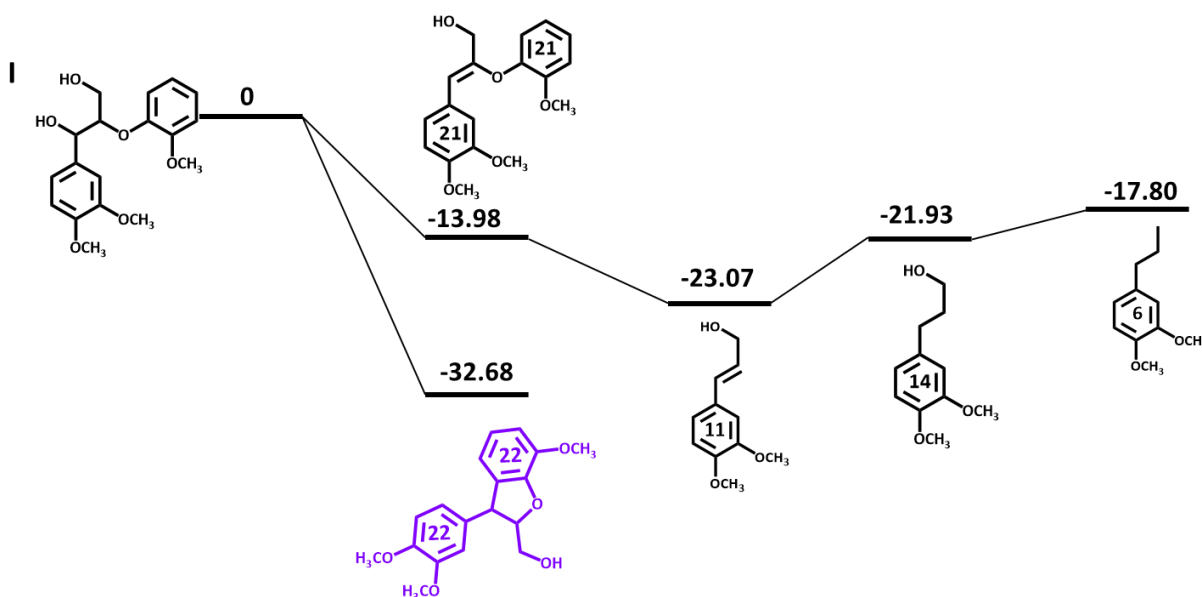


Figure S4: Identification of product and intermediates based on ^{13}C NMR profile with *in-situ* recorded spectra. Red vertical lines show the ChemDraw 19.0 (© PerkinElmer Infomatics) predicted ^{13}C NMR of compounds (1, 2, 5, 6, 7, 11, 13, 21, 22, 26).



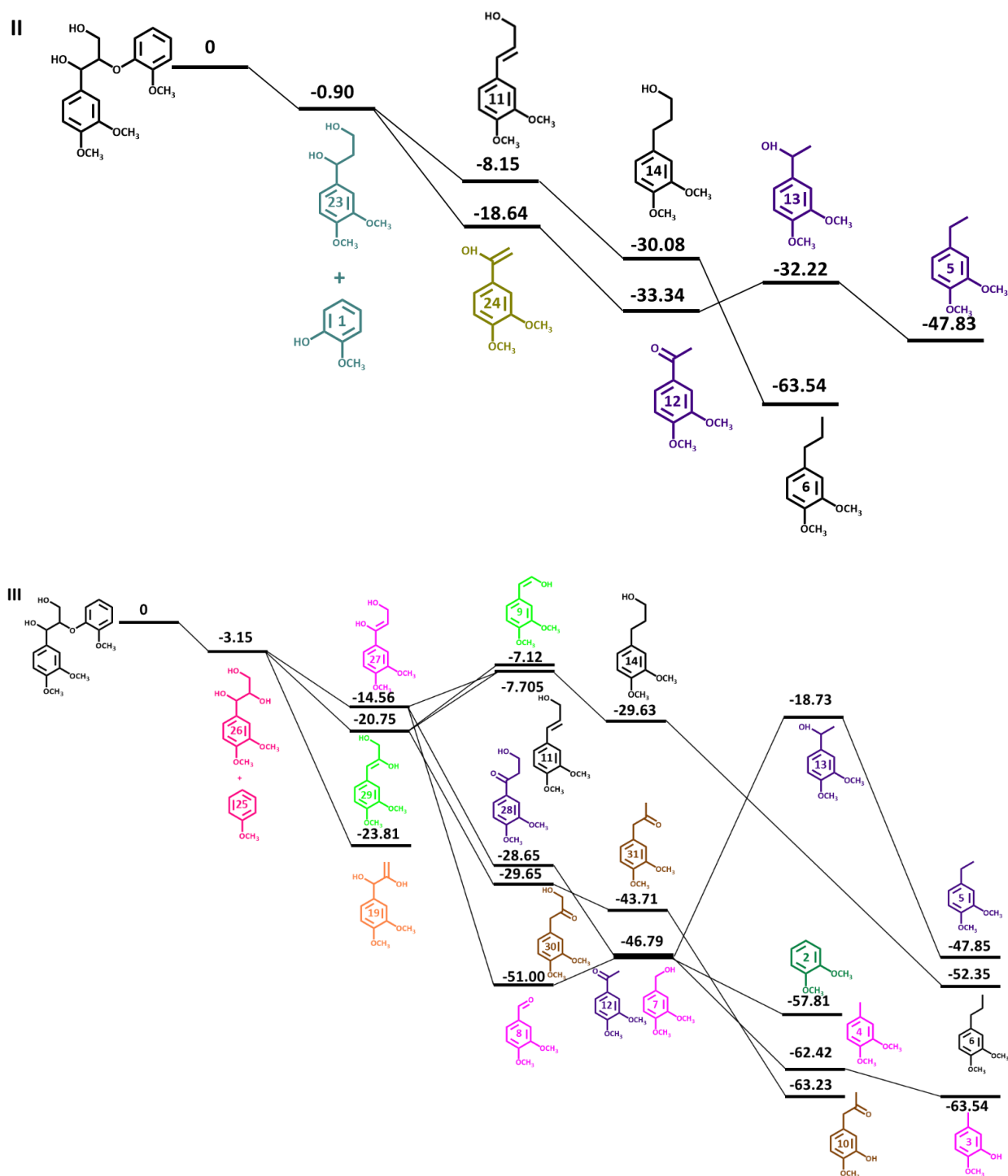
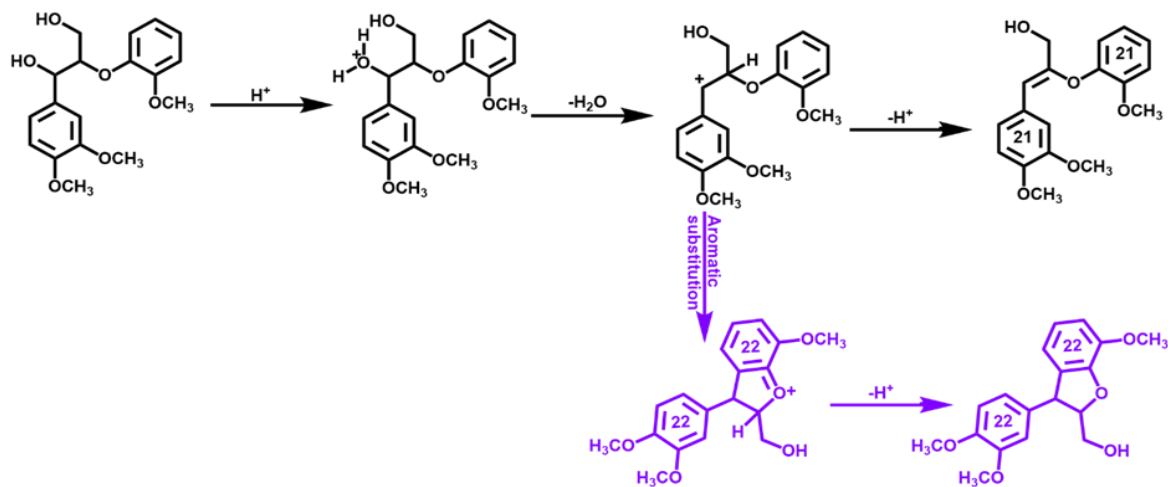


Figure S5. The calculated energy profiles of three main pathways shown in Figure 4. Unit of the energy is kcal/mol.



Scheme S1: Primary ion mechanism of β -O-4 acidolysis over zeolite.