

## **Supplementary information**

### **Experimental and computational investigation on the charge storage performance of a novel Al<sub>2</sub>O<sub>3</sub>-graphene hybrid electrode**

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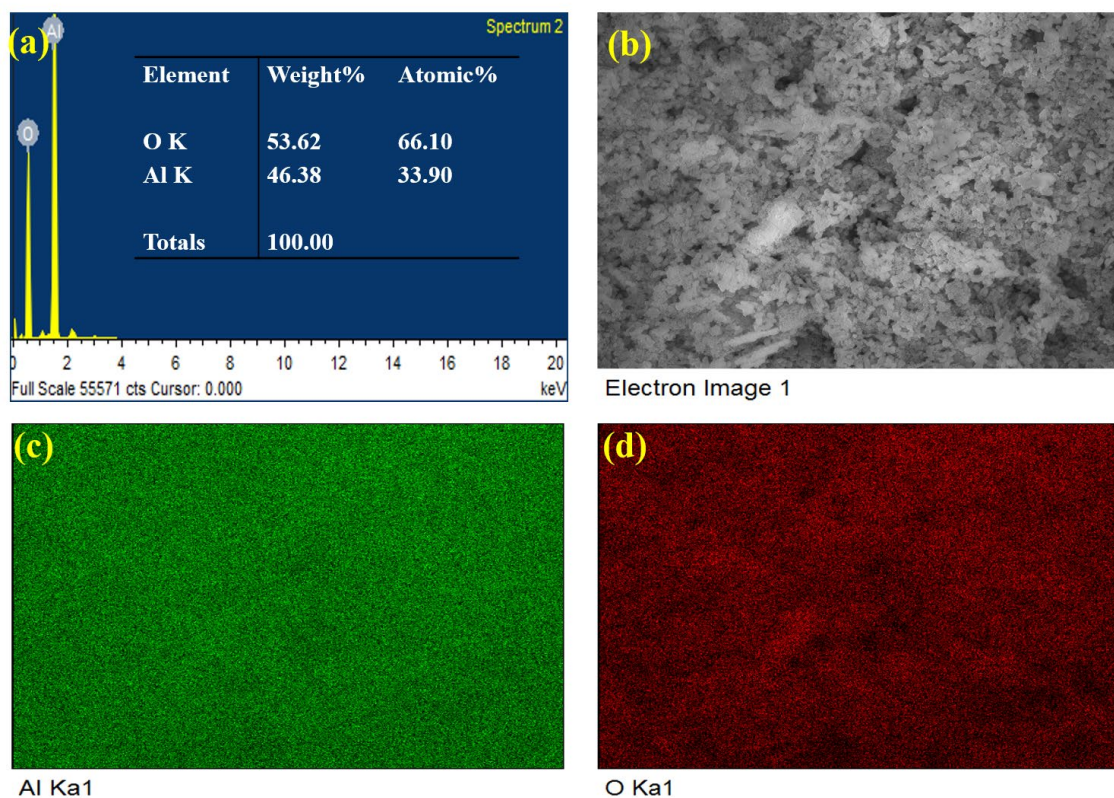
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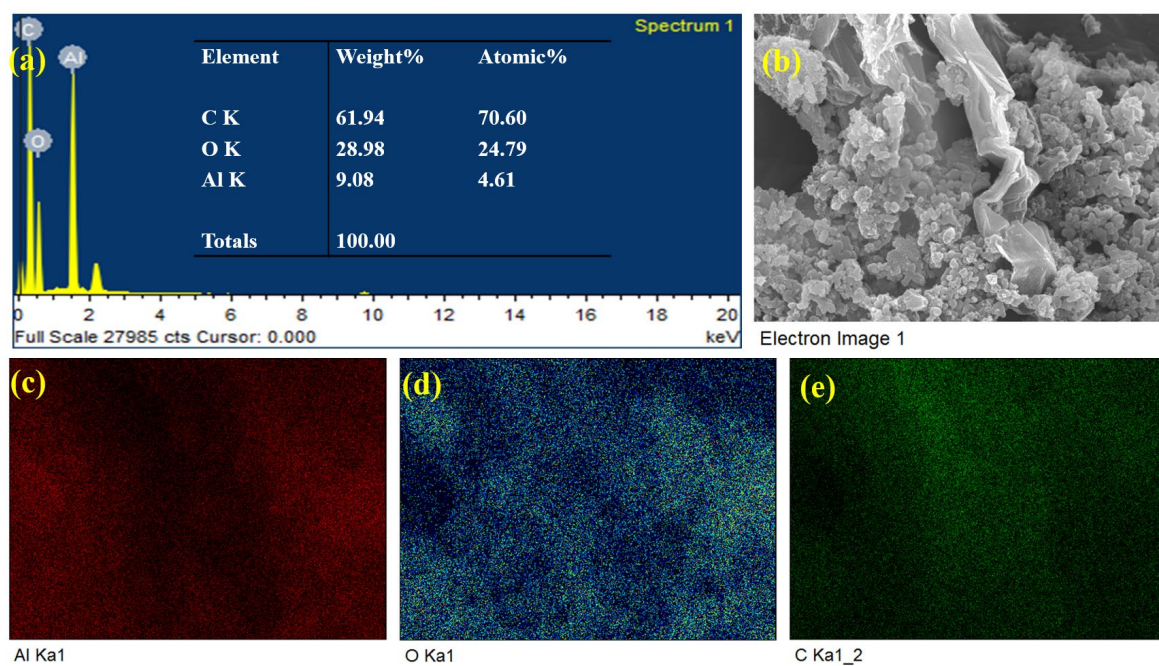
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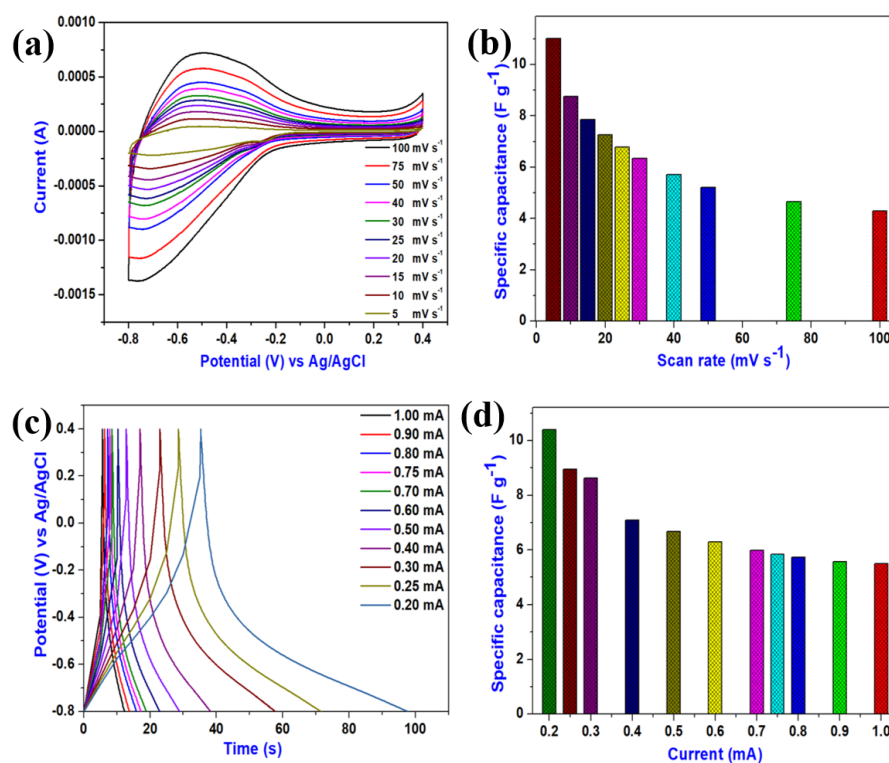
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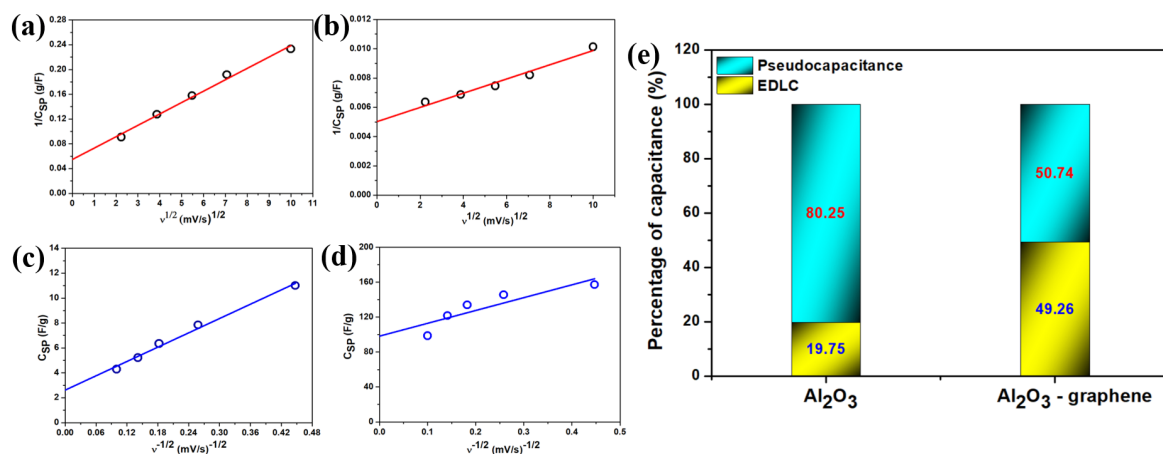
**Figure S1.** (a) The EDX spectrum  $\text{Al}_2\text{O}_3$ , (b) FE-SEM micrograph for elemental mapping of  $\text{Al}_2\text{O}_3$ , (c) aluminium and (d) oxygen.



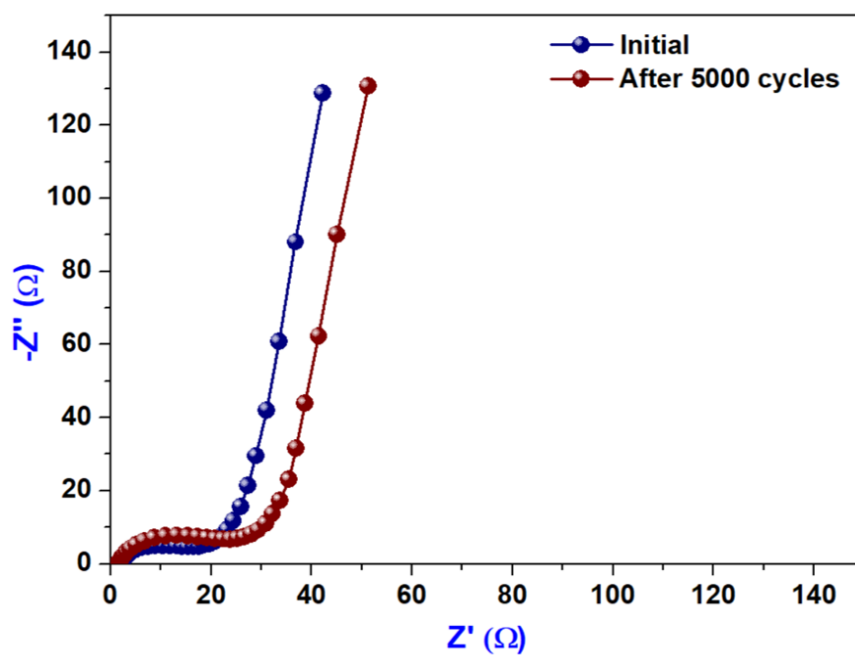
**Figure S2.** (a) The EDX spectrum Al<sub>2</sub>O<sub>3</sub>-graphene hybrid, (b) FE-SEM micrograph for elemental mapping of Al<sub>2</sub>O<sub>3</sub>-graphene hybrid, (c) aluminium, (d) oxygen and (e) carbon.



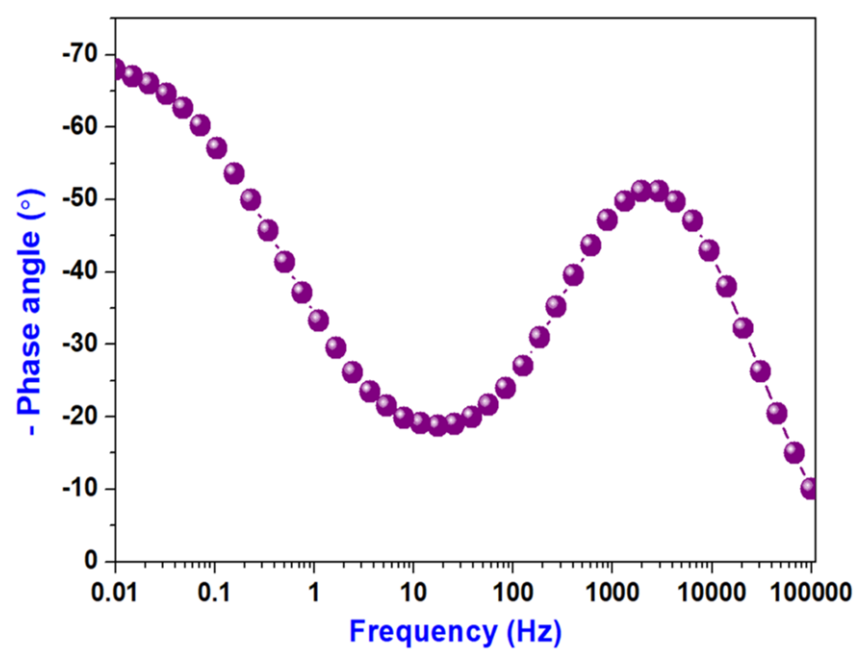
**Figure S3.** Electrochemical characterization of  $\text{Al}_2\text{O}_3$  electrode. (a) Cyclic voltammetric profiles of  $\text{Al}_2\text{O}_3$  electrode at various scan rates 5-100  $\text{mV s}^{-1}$ , (b) effect of scan rates on specific capacitances of  $\text{Al}_2\text{O}_3$  electrode, (c) charge-discharge profile of  $\text{Al}_2\text{O}_3$  electrode at various applied currents and (d) effect of applied currents on specific capacitances of  $\text{Al}_2\text{O}_3$  electrode.



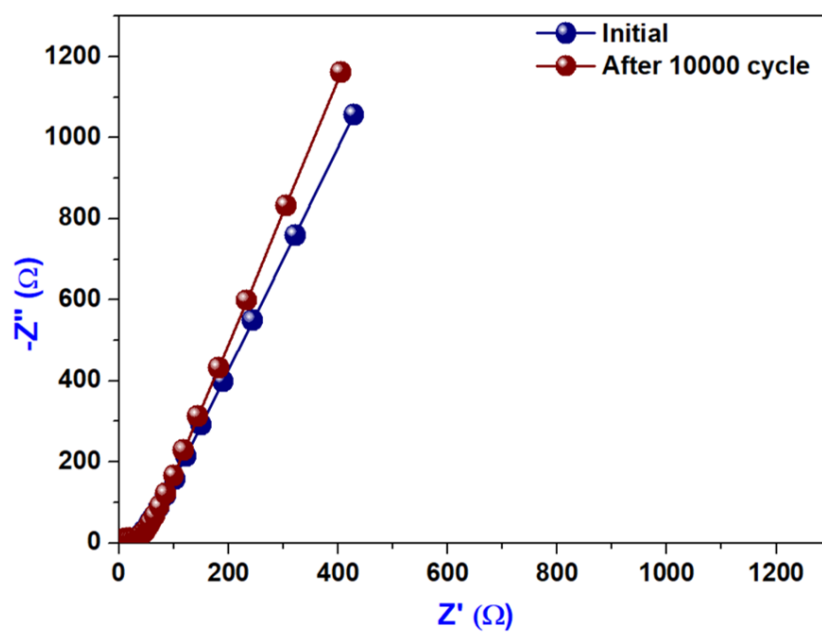
**Figure S4.** Trasatti plot for  $\text{Al}_2\text{O}_3$  and  $\text{Al}_2\text{O}_3$ -graphene hybrid. Dependence of  $1/C_{sp}$  on  $v^{1/2}$  for (a)  $\text{Al}_2\text{O}_3$  electrode and (b)  $\text{Al}_2\text{O}_3$ -graphene hybrid electrode. Dependence of  $C_{sp}$  on  $v^{-1/2}$  for (c)  $\text{Al}_2\text{O}_3$  electrode and (d)  $\text{Al}_2\text{O}_3$ -graphene hybrid electrode in  $\text{Na}_2\text{SO}_4$  electrolyte. (e) The contribution of EDLC and pseudocapacitance of  $\text{Al}_2\text{O}_3$  electrode and  $\text{Al}_2\text{O}_3$ -graphene hybrid electrode.



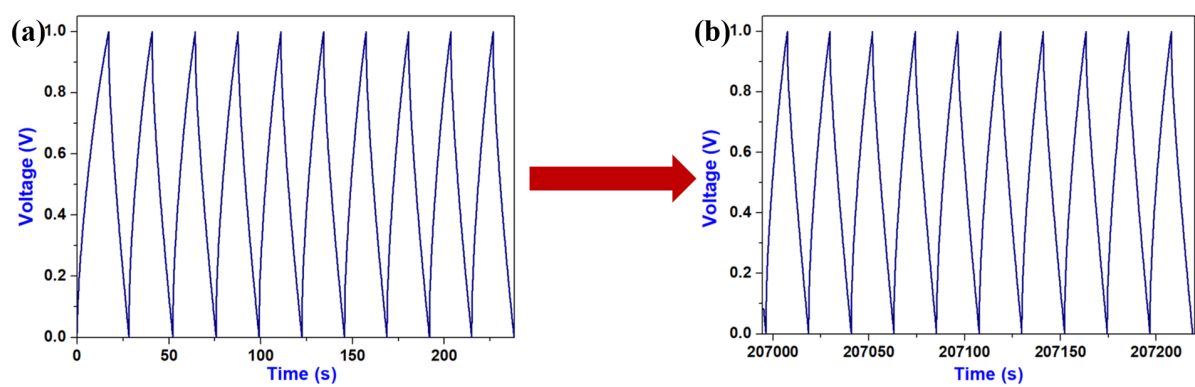
**Figure S5.** The comparative Nyquist plot for  $\text{Al}_2\text{O}_3$ -graphene hybrid electrode at initial and after cyclic stability test.



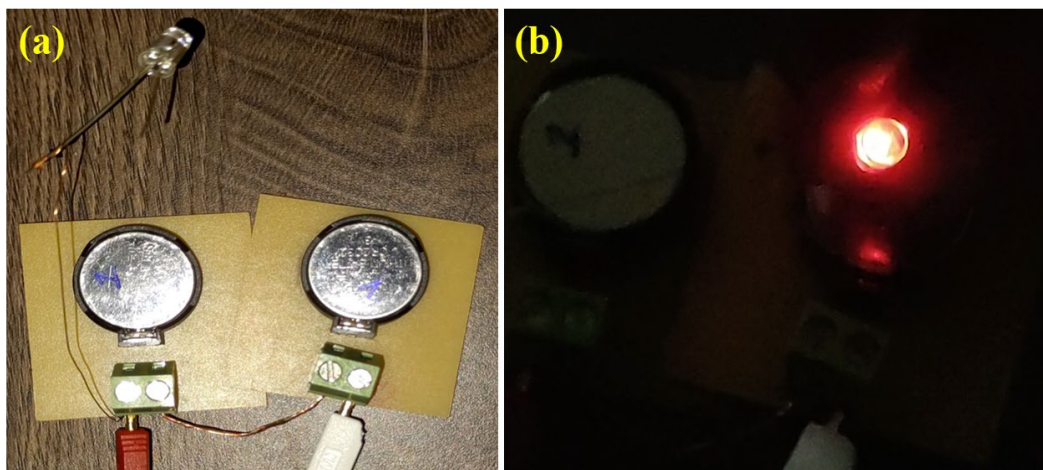
**Figure S6.** Bode phase angle plot of Al<sub>2</sub>O<sub>3</sub>-graphene hybrid//Al<sub>2</sub>O<sub>3</sub>-graphene hybrid SSD.



**Figure S7.** The comparative Nyquist plot for  $\text{Al}_2\text{O}_3$ -graphene hybrid// $\text{Al}_2\text{O}_3$ -graphene hybrid SSD at initial and after cyclic stability test.



**Figure S8.** The initial and final ten cycles of  $\text{Al}_2\text{O}_3$ -graphene hybrid// $\text{Al}_2\text{O}_3$ -graphene hybrid SSD in cyclic stability test.



**Figure S9.** The practical application of Al<sub>2</sub>O<sub>3</sub>-graphene hybrid//Al<sub>2</sub>O<sub>3</sub>-graphene hybrid SSD in series connection of two-coin cells (a) without glowing an LED, (b) glowing a red LED by fully charged Al<sub>2</sub>O<sub>3</sub>-graphene hybrid//Al<sub>2</sub>O<sub>3</sub>-graphene hybrid SSD in series connection of two-coin cells.

**Table S1.** Summary of supercapacitive performances of Al<sub>2</sub>O<sub>3</sub>-graphene hybrid electrode and reported supercapacitor electrode materials using three-electrode system.

| Material                                         | Preparation method         | Specific capacitance (F g <sup>-1</sup> ) | Reference        |
|--------------------------------------------------|----------------------------|-------------------------------------------|------------------|
| CuSbS <sub>2</sub>                               | Colloidal method           | 22                                        | 1                |
| CuSbSe <sub>2</sub>                              | Colloidal                  | 48                                        | 1                |
| Cu <sub>3</sub> SbS <sub>4</sub>                 | Hydrothermal               | 60                                        | 2                |
| FeMoO <sub>4</sub>                               | Hydrothermal               | 135                                       | 3                |
| /Graphene                                        |                            |                                           |                  |
| Cu <sub>2</sub> MoS <sub>4</sub>                 | Hydrothermal               | 135.78                                    | 4                |
| ZnS                                              | Solvothermal               | 32.8                                      | 5                |
| WS <sub>2</sub>                                  | Chemical exfoliation       | 40                                        | 6                |
| CuO                                              | Wet chemical method        | 88.5                                      | 7                |
| RuO <sub>2</sub>                                 | Chemical synthesis         | 50                                        | 8                |
| MoO <sub>3</sub>                                 | Solution combustion method | 108.88                                    | 9                |
| MoS <sub>2</sub>                                 | Hydrothermal               | 92.85                                     | 10               |
| <b>Al<sub>2</sub>O<sub>3</sub>-<br/>graphene</b> | <b>Hydrothermal</b>        | <b>157.29</b>                             | <b>This work</b> |

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