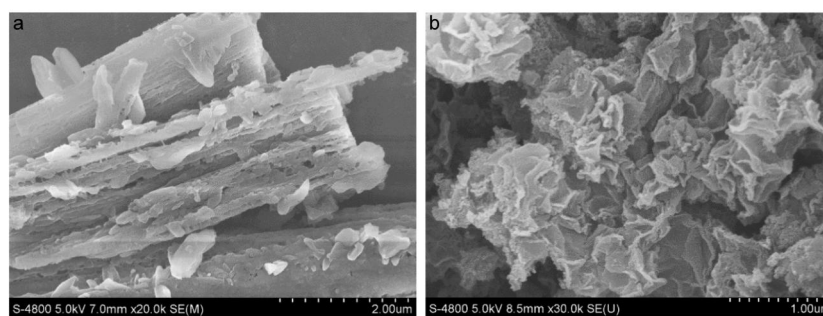


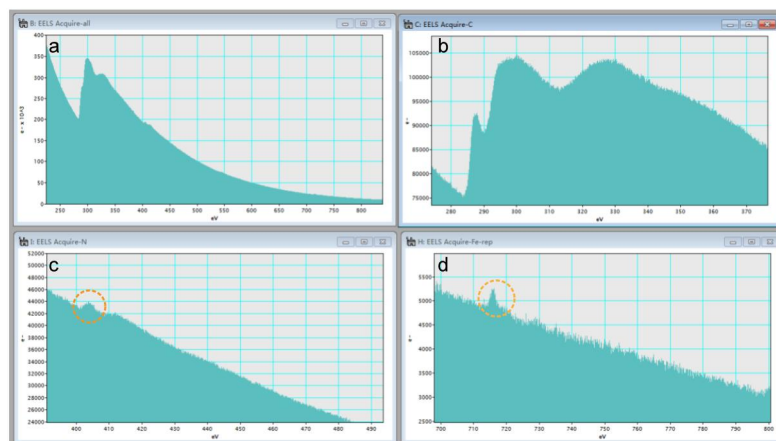
SUPPLEMENTARY INFORMATION

Supplementary figures



Supplementary Fig. 1 | **a**, SEM images of NC. **b**, SEM images of FeSAC-NC.

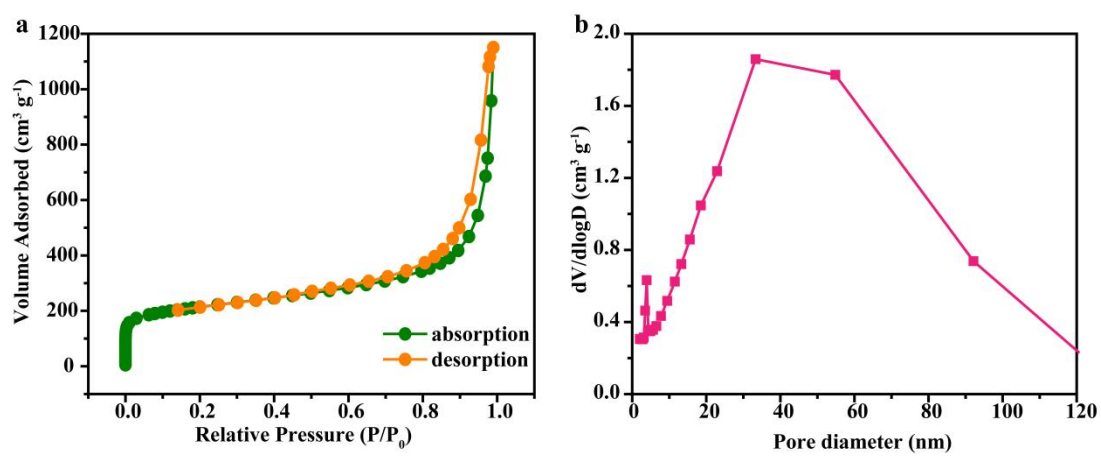
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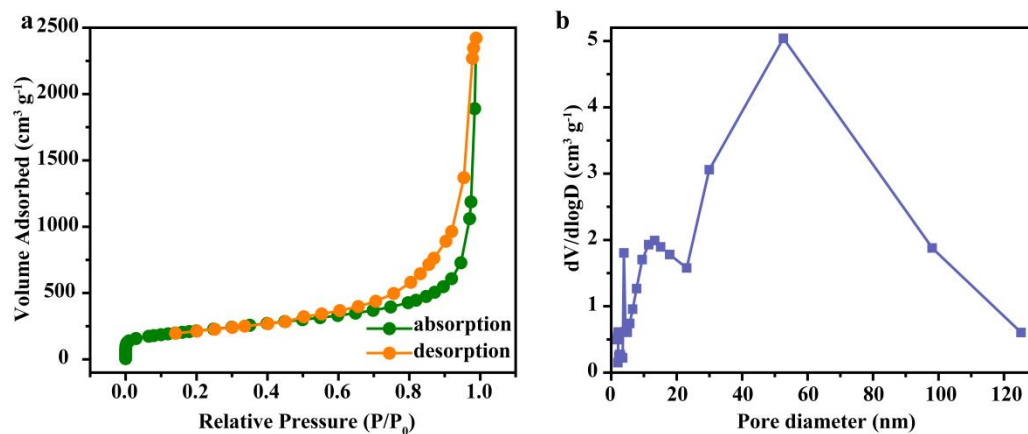
3 **Supplementary Fig. 2** | EELS point spectrum of FeSAC-NC. **a**, full spectrum. **b**, C spectrum. **c**, N
4 spectrum. **d**, Fe spectrum.

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Supplementary Fig. 3 | a, N₂ adsorption-desorption curves of FeSAC-NC. **b,** Pore distribution curve of FeSAC-NC.

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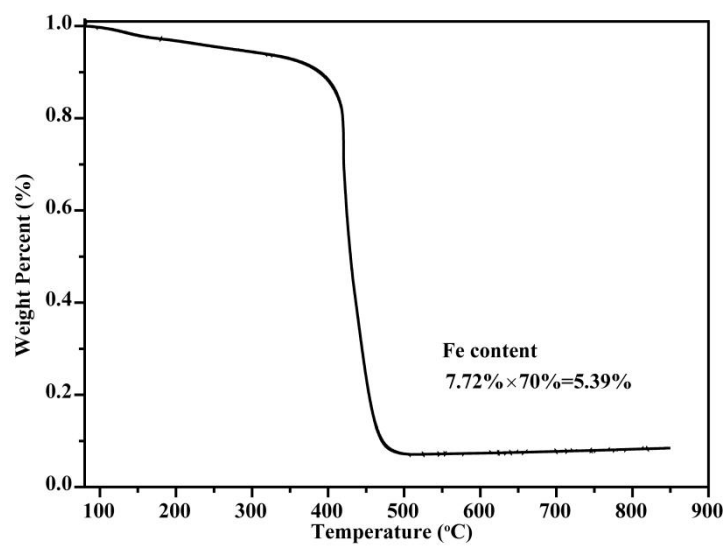


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3 **Supplementary Fig. 4 | a,** N₂ adsorption-desorption curves of FeN₄. **b,** Pore distribution curve of FeN₄.

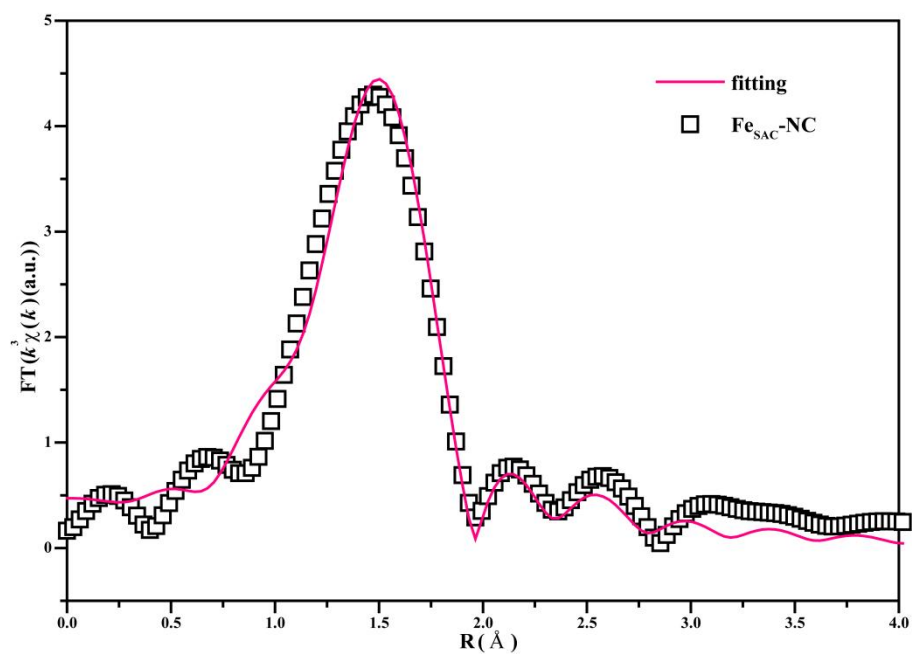
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Supplementary Fig. 5 | TGA curve of Fe_{SAC}-NC tested in air ambient.

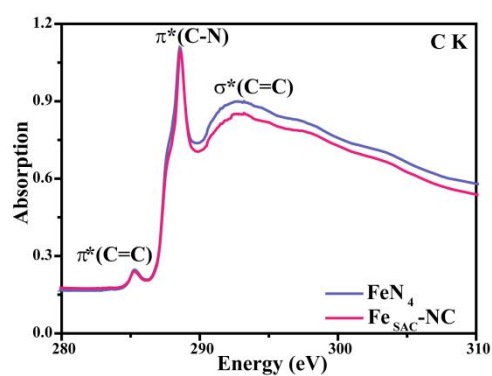
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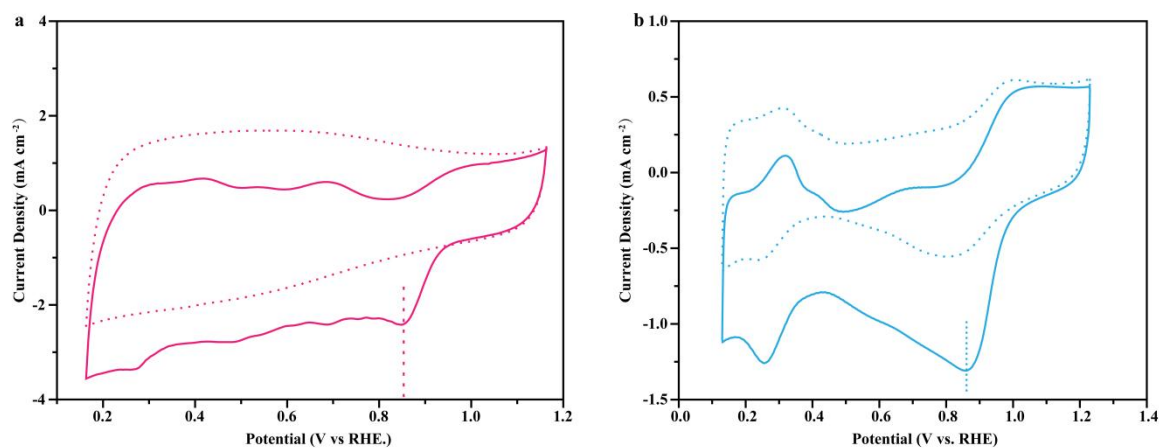
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3 **Supplementary Fig. 6** | k^3 -weighted Fourier transform spectra of $\text{Fe}_{\text{SAC}}\text{-NC}$ and corresponding fitting
4 curve.

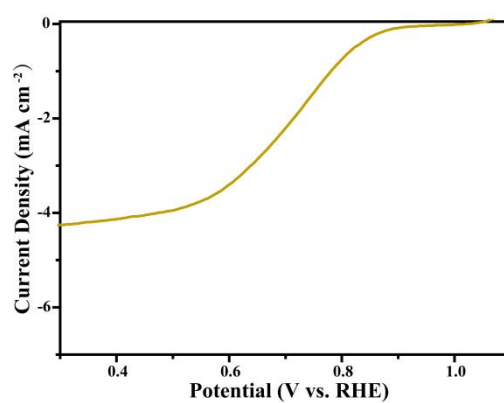
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Supplementary Fig. 7 | The normalized XANES spectra at the C K-edge of FeN₄ and FeSAC-NC.

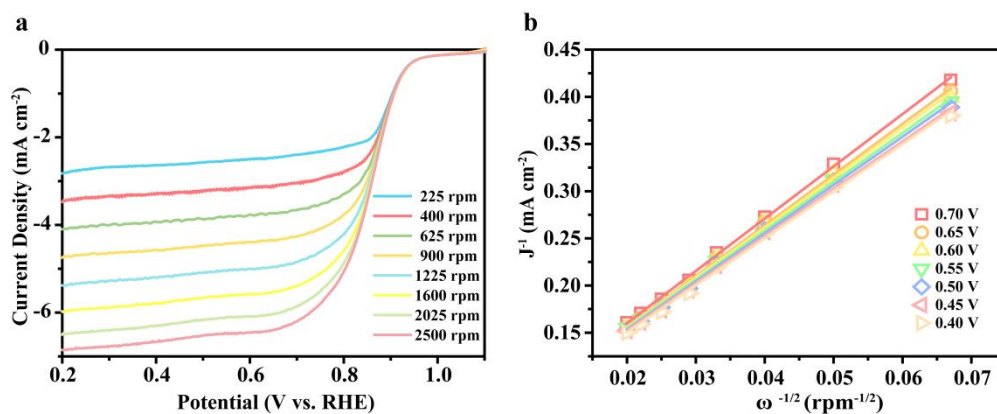


Supplementary Fig. 8 | a, The CV curves of FeSAC-NC in O₂ (solid) or Ar (dotted)-saturated 0.1M KOH. **b**, The CV curves of 20% Pt/C in O₂ (solid) or Ar(dotted)-saturated 0.1M KOH.



Supplementary Fig. 9 | LSV curve of NC.

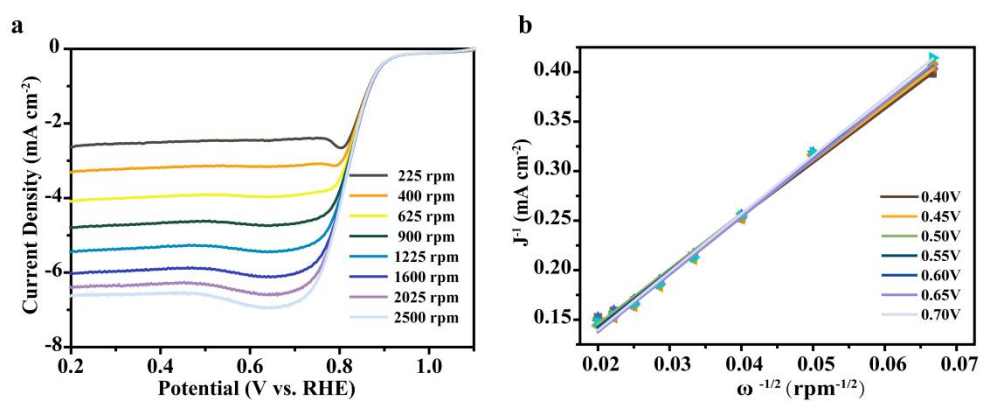
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Supplementary Fig. 10 | a, LSVs curves of Fe_{SAC}-NC measured at different RDE rotating speed in O₂-saturated electrolytes. **b**, The corresponding K-L plot of Fe_{SAC}-NC.

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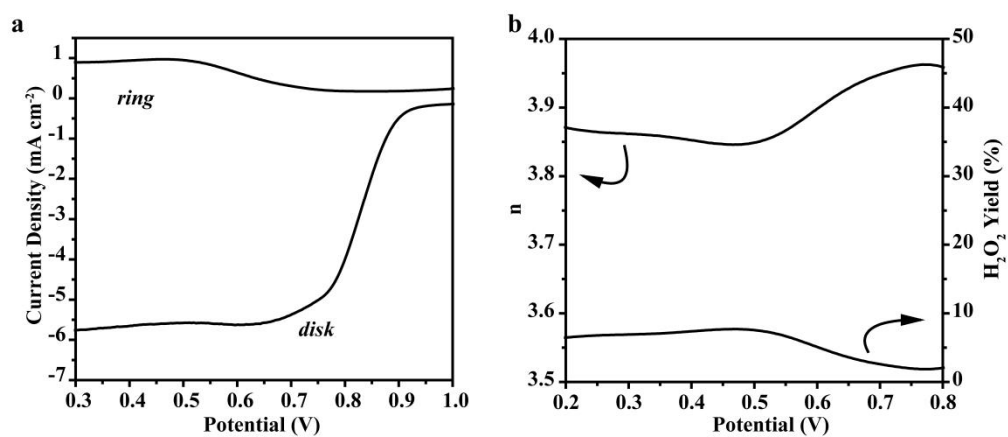


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3 **Supplementary Fig. 11 | a**, LSVs curves of FeN₄ measured at different RDE rotating speed in O₂-
4 saturated electrolytes. **b**, The corresponding K-L plot of FeN₄.

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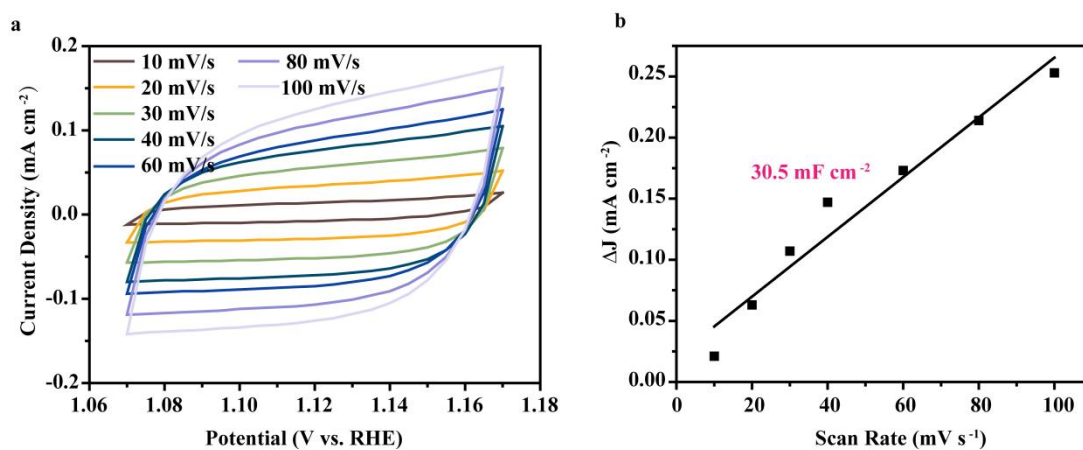
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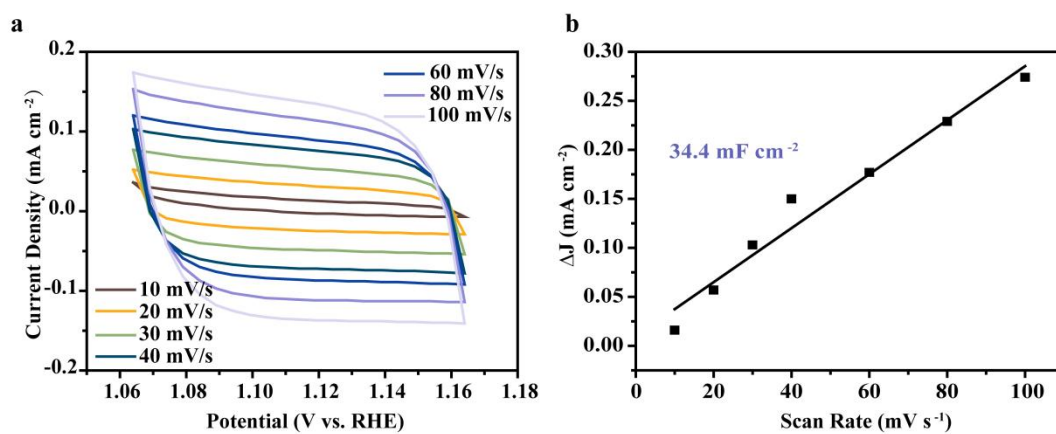
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3 **Supplementary Fig. 12 | a**, RRDE test of FeN₄. **b**, Electron transfer number (*n*) and H₂O₂ yield
4 calculated from RRDE measurement of FeN₄.

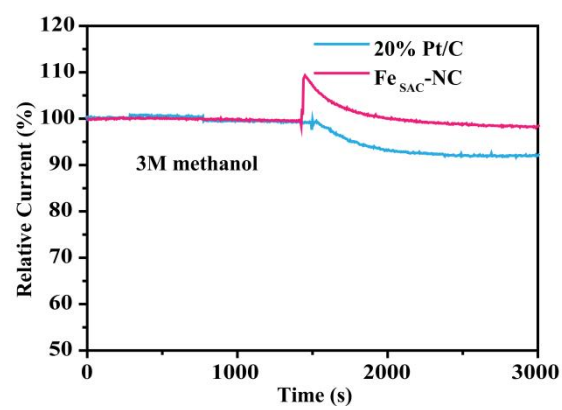
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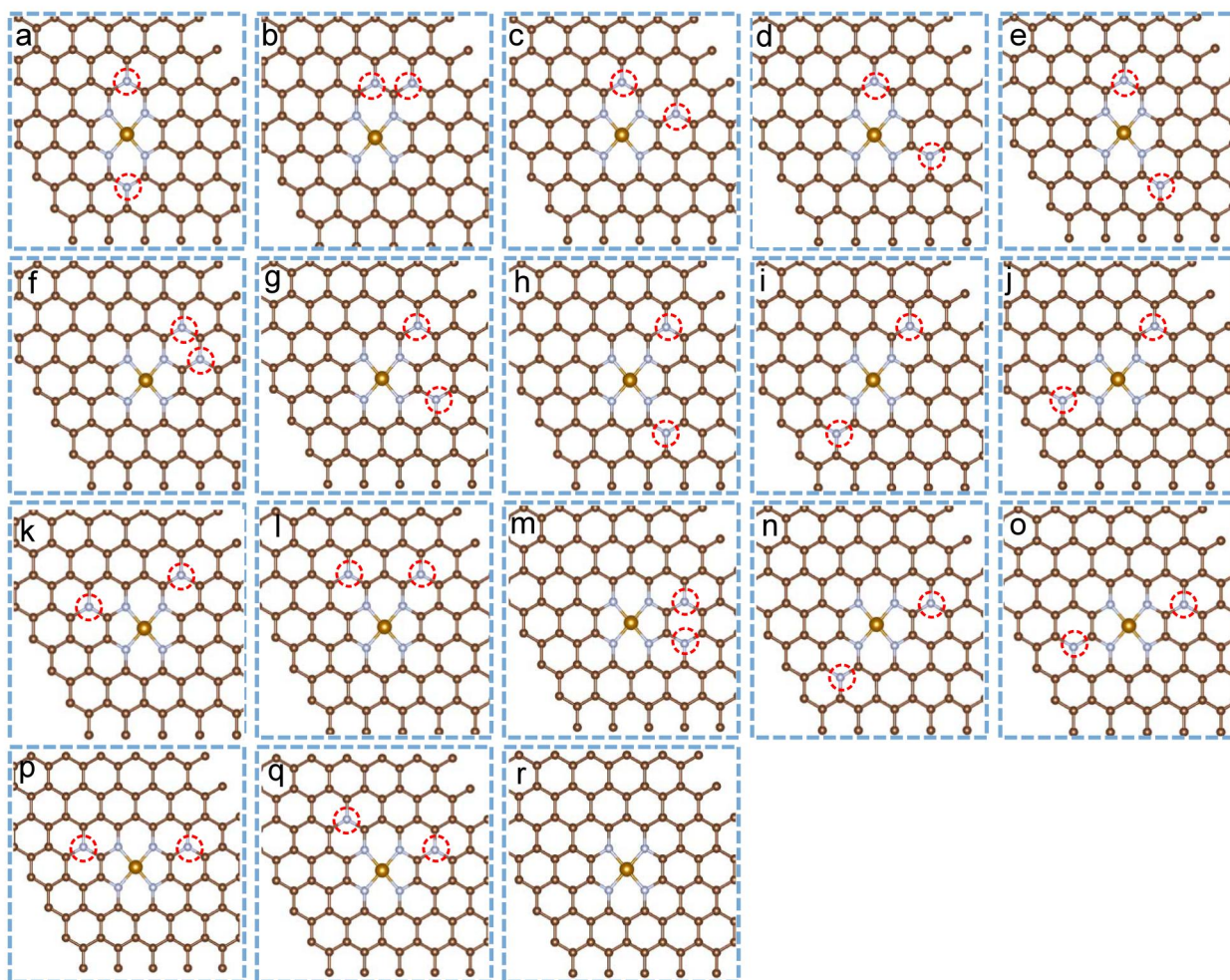
Supplementary Fig. 13 | a, CV curves for FeSAC-NC in the region of 1.06-1.18 V vs. RHE with various scan rates. **b**, Scan-rate dependency of the current density for FeSAC-NC.



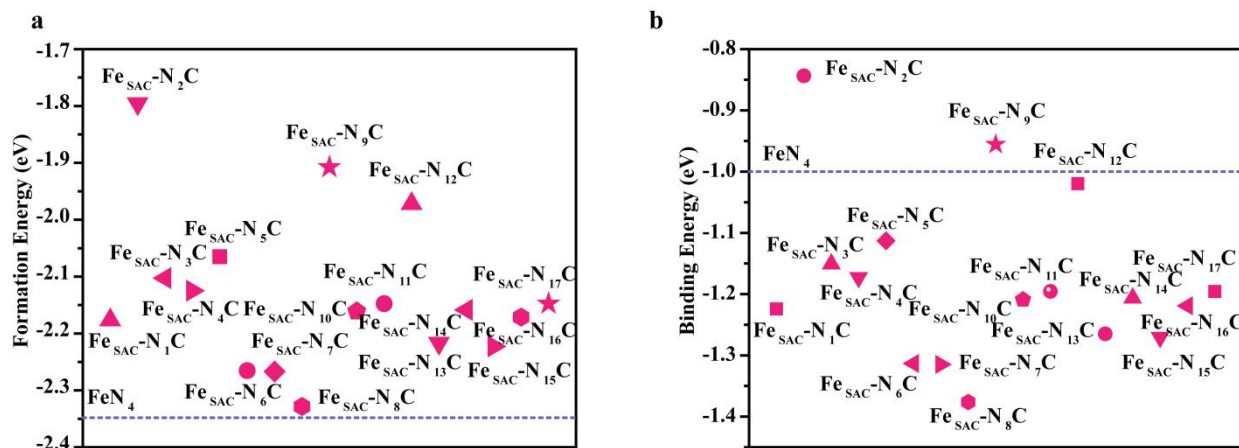
Supplementary Fig. 14 | **a**, CV curves for FeN₄ in the region of 1.06-1.18 V vs. RHE with various scan rates. **b**, Scan-rate dependency of the current density for FeN₄.



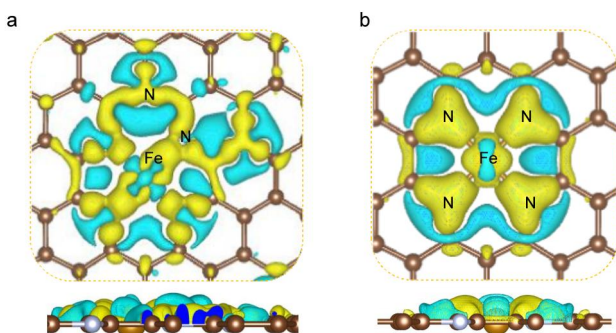
Supplementary Fig. 15 | Methanol crossover effect test of Fe_{SAC}-NC and 20% Pt/C.

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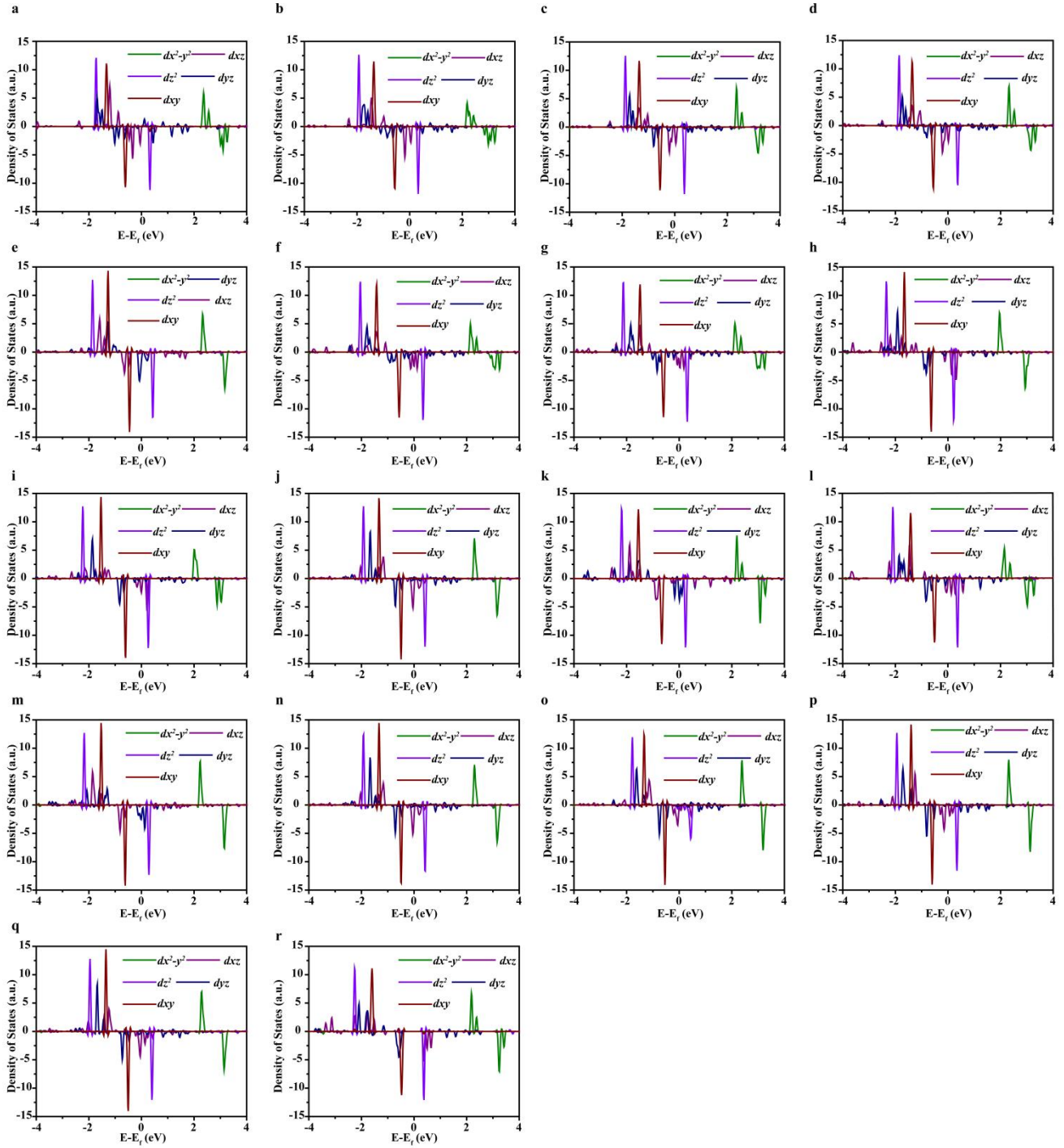
Supplementary Fig. 16 | Optimized structure of **(a)** FeSAC-N₁C. **(b)** FeSAC-N₂C. **(c)** FeSAC-N₃C. **(d)** FeSAC-N₄C. **(e)** FeSAC-N₅C. **(f)** FeSAC-N₆C. **(g)** FeSAC-N₇C. **(h)** FeSAC-N₈C. **(i)** FeSAC-N₉C. **(j)** FeSAC-N₁₀C. **(k)** FeSAC-N₁₁C. **(l)** FeSAC-N₁₂C. **(m)** FeSAC-N₁₃C. **(n)** FeSAC-N₁₄C. **(o)** FeSAC-N₁₅C. **(p)** FeSAC-N₁₆C. **(q)** FeSAC-N₁₇C. **(r)** FeN₄. (The names are derived from the N atoms respective locations.)



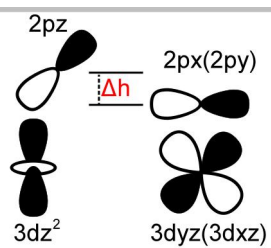
Supplementary Fig. 17 | a, The formation energy of $\text{Fe}_{\text{SAC}}\text{-N}_X\text{C}$ ($X=1-17$). **b**, The binding energy of Fe site on $\text{Fe}_{\text{SAC}}\text{-N}_X\text{C}$ ($X=1-17$).



Supplementary Fig. 18 | **a**, Differential charge density of FeSAC-N₃C. **b**, Differential charge density of FeN₄. Yellow regions indicate charge accumulation, while cyan regions indicate charge depletion.

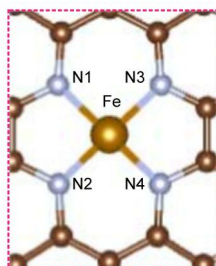


Supplementary Fig. 19 | Detailed DOS of Fe d orbitals: (a) FeSAC-N₁C, (b) FeSAC-N₂C, (c) FeSAC-N₃C, (d) FeSAC-N₄C, (e) FeSAC-N₅C, (f) FeSAC-N₆C, (g) FeSAC-N₇C, (h) FeSAC-N₈C, (i) FeSAC-N₉C, (j) FeSAC-N₁₀C, (k) FeSAC-N₁₁C, (l) FeSAC-N₁₂C, (m) FeSAC-N₁₃C, (n) FeSAC-N₁₄C, (o) FeSAC-N₁₅C, (p) FeSAC-N₁₆C, (q) FeSAC-N₁₇C and (r) FeN₄.



Supplementary Fig. 20 | Schematic illustration of different hybridization modes of $3dyz$ ($3dxz$), $3dz^2$ and OH $2\pi^*$ orbital. (Δh means the spatial height difference.)

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Supplementary Fig. 21 | The schematic of each Fe-N bond length of $\text{FeSAC-N}_x\text{C}$ ($x=1-17$).

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Supporting tables

Supplementary Table 1 | Summary of NC, Fe_{SAC}-NC, and FeN₄ BET data.

Sample	BET surface area (m ² ·g ⁻¹)	Total volume (cm ³ ·g ⁻¹)	Mesopore size (nm)
NC	255	0.29	5.4
Fe _{SAC} -NC	725	1.66	15.4
FeN ₄	752	3.77	21.2

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2 **Supplementary Table 2** | Elemental composition of Fe_{SAC}-NC.
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elemental analyzer				ICP	TGA
C (wt.%)	N (wt.%)	O (wt.%)	H (wt.%)	Fe(wt.%)	
81.38	8.97	3.21	0.96	5.48	5.39

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1 **Supplementary Table 3** | EXAFS fitting parameters of Fe foil and Fe_{SAC}-NC.

Samples	Shell	N	R (Å)	$\Delta\sigma^2 \times 10^3$ (Å ²)	ΔE_0 (eV)	R-factor
Fe foil	Fe-Fe	8	2.42±0.01	6.8 ± 1.2	10.6 ± 0.8	0.037
	Fe-Fe	6	2.84±0.01	9.0 ± 2.1	10.1 ± 0.4	
FeN ₄	Fe-N	4	1.89±0.01	3.1±1.6	1.2±0.1	0.019
Fe _{SAC} -NC	Fe-N	3.7	1.90±0.02	8.17± 1.1	-7.8±0.5	0.042

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Supplementary Table 4 | Summarize the E_0 values of standard reference materials and Fe_{SAC}-NC.

Sample	Fe foil	Fe ₂ O ₃	FeN ₄	Fe _{SAC} -NC
E_0 (eV)	7112.216	7126.305	7122.311	7123.292

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2**Supplementary Table 5** | Summarize the most near-term reported single atoms-based electrocatalysts.

Sample	Onset potential (V vs. RHE)	Half-wave potential (V vs. RHE)	Ref
Fe _{SAC} -NC	1.06	0.89	our work
Ru-SAS/SNC	0.998	0.861	1
Cu/Zn-NC	0.98	0.83	2
Co-N/C+NG	1.04	0.915	3
Fe-N-CDC/CNT	0.99	0.86	4
Fe-NiNC-50	1.0	0.84	5
FeSA/HNPC	0.92	0.83	6
Cu-S ₁ N ₃	1.05	0.918	7
Ni-N ₄ /GHSs/Fe-N ₄	0.93	0.83	8
Co ₁ -N ₃ PS/HC	1.00	0.91	9
PSTA-Co-1000	0.96	0.878	10
Fe-N/P-C-700	0.941	0.867	11
Fe/OES	1.00	0.85	12

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2**Supplementary Table 6** | Adsorption free energies of *O, *OH, and *OOH on Fe_{SAC}-N_XC (X=1-17).

Catalysts	ΔG^*_{O} (eV)	ΔG^*_{OH} (eV)	ΔG^*_{OOH} (eV)
Fe _{SAC} -N ₁ C	1.93	0.84	3.56
Fe _{SAC} -N ₂ C	1.74	0.86	3.54
Fe _{SAC} -N ₃ C	1.70	0.84	3.60
Fe _{SAC} -N ₄ C	1.81	0.86	3.51
Fe _{SAC} -N ₅ C	1.70	0.80	3.52
Fe _{SAC} -N ₆ C	1.66	0.86	3.48
Fe _{SAC} -N ₇ C	1.65	0.83	3.48
Fe _{SAC} -N ₈ C	1.65	0.85	3.49
Fe _{SAC} -N ₉ C	1.66	0.84	3.53
Fe _{SAC} -N ₁₀ C	1.66	0.82	3.53
Fe _{SAC} -N ₁₁ C	1.66	0.86	3.61
Fe _{SAC} -N ₁₂ C	1.75	0.82	3.49
Fe _{SAC} -N ₁₃ C	1.65	0.75	3.40
Fe _{SAC} -N ₁₄ C	1.65	0.83	3.49
Fe _{SAC} -N ₁₅ C	1.77	0.82	3.51
Fe _{SAC} -N ₁₆ C	1.76	0.85	3.53
Fe _{SAC} -N ₁₇ C	1.66	0.84	3.50
FeN ₄	1.39	0.58	3.56

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3**Supplementary Table 7** | Reaction free energy of elementary step for ORR at U = 0 V vs. RHE on FeSAC-N_XC(X=1-17).

Catalysts	ΔG_1 (eV)	ΔG_2 (eV)	ΔG_3 (eV)	ΔG_4 (eV)	ΔG_5 (eV)
FeSAC-N ₁ C	-0.51	-0.85	-1.63	-1.09	-0.84
FeSAC-N ₂ C	-0.52	-0.86	-1.80	-0.87	-0.86
FeSAC-N ₃ C	-0.55	-0.76	-1.90	-0.86	-0.84
FeSAC-N ₄ C	-0.55	-0.86	-1.82	-0.90	-0.80
FeSAC-N ₅ C	-0.52	-0.89	-1.70	-0.95	-0.86
FeSAC-N ₆ C	-0.70	-0.74	-1.82	-0.80	-0.86
FeSAC-N ₇ C	-0.50	-0.94	-1.83	-0.82	-0.83
FeSAC-N ₈ C	-0.42	-1.01	-1.84	-0.80	-0.85
FeSAC-N ₉ C	-0.46	-0.93	-1.87	-0.82	-0.84
FeSAC-N ₁₀ C	-0.51	-0.87	-1.88	-0.83	-0.82
FeSAC-N ₁₁ C	-0.71	-0.72	-1.85	-0.89	-0.86
FeSAC-N ₁₂ C	-0.67	-0.84	-1.82	-0.84	-0.82
FeSAC-N ₁₃ C	-0.54	-0.98	-1.75	-0.90	-0.75
FeSAC-N ₁₄ C	-0.53	-0.90	-1.84	-0.82	-0.83
FeSAC-N ₁₅ C	-0.53	-0.88	-1.74	-0.95	-0.82
FeSAC-N ₁₆ C	-0.53	-0.87	-1.77	-0.91	-0.85
FeSAC-N ₁₇ C	-0.53	-0.89	-1.84	-0.82	-0.84
FeN ₄	-0.70	-0.67	-2.17	-0.81	-0.58

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1 **Supplementary Table 8** | Fe-N1, Fe-N2, Fe-N3, Fe-N4 and the average Fe-N bond length over Fe-N-C
2 catalysts.

Catalysts	Fe-N1 (Å)	Fe-N2 (Å)	Fe-N3 (Å)	Fe-N4 (Å)	Average Fe-N (Å)
Fe _{SAC} -N ₁ C	1.90	1.90	1.90	1.90	1.90
Fe _{SAC} -N ₃ C	1.91	1.90	1.91	1.89	1.90
Fe _{SAC} -N ₄ C	1.90	1.90	1.89	1.91	1.90
Fe _{SAC} -N ₁₆ C	1.92	1.89	1.92	1.89	1.91

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Supplementary References

1. Qin, J. *et al.* Altering ligand fields in single-atom sites through second-shell anion modulation boosts the oxygen reduction reaction. *J. Am. Chem. Soc.* **144**, 2197-2207, (2022).
2. Tong, M. *et al.* Operando cooperated catalytic mechanism of atomically dispersed Cu-N₄ and Zn-N₄ for promoting oxygen reduction reaction. *Angew. Chem. Int. Ed.* **60**, 14005-14012, (2021).
3. Zhang, X. *et al.* Boosting electrocatalytic activity of single atom catalysts supported on nitrogen-doped carbon through N coordination environment engineering. *Small* **18**, e2105329, (2022).
4. Lilloja, J. *et al.* Transition-metal- and nitrogen-doped carbide-derived carbon/carbon nanotube composites as cathode catalysts for anion-exchange membrane fuel cells. *ACS Catal.* **11**, 1920-1931, (2021).
5. Zhu, X. *et al.* Harnessing the interplay of Fe-Ni atom pairs embedded in nitrogen-doped carbon for bifunctional oxygen electrocatalysis. *Nano Energy* **71**, 104597-104607, (2020).
6. Jin, X. *et al.* A highly efficient Fe-N-C electrocatalyst with atomically dispersed FeN₄ sites for the oxygen reduction reaction. *ChemCatChem* **13**, 2683-2690, (2021).
7. Shang, H. *et al.* Engineering unsymmetrically coordinated Cu-S₁N₃ single atom sites with enhanced oxygen reduction activity. *Nat. Commun.* **11**, 3049, (2020).
8. Chen, J. *et al.* Dual single-atomic Ni-N₄ and Fe-N₄ sites constructing janus hollow graphene for selective oxygen electrocatalysis. *Adv. Mater.* **32**, e2003134, (2020).
9. Chen, Y. *et al.* Atomic-level modulation of electronic density at cobalt single-atom sites derived from metal-organic frameworks: enhanced oxygen reduction performance. *Angew. Chem. Int. Ed.* **60**, 3212-3221, (2021).
10. Wei, X. *et al.* Cross-linked polyphosphazene hollow nanosphere-derived N/P-doped porous carbon with single nonprecious metal atoms for the oxygen reduction reaction. *Angew. Chem. Int. Ed.* **59**, 14639-14646, (2020).
11. Yuan, K. *et al.* Boosting oxygen reduction of single iron active sites via geometric and electronic engineering: nitrogen and phosphorus dual coordination. *J. Am. Chem. Soc.* **142**, 2404-2412, (2020).
12. Hou, C.-C. *et al.* Single-atom iron catalysts on overhang-eave carbon cages for high-performance oxygen reduction reaction. *Angew. Chem. Int. Ed.* **59**, 7384-7389, (2020).