

Crosslinked ion diffusion inhibitors stabilize 3D/2D perovskite heterostructures for efficient and stable photovoltaics

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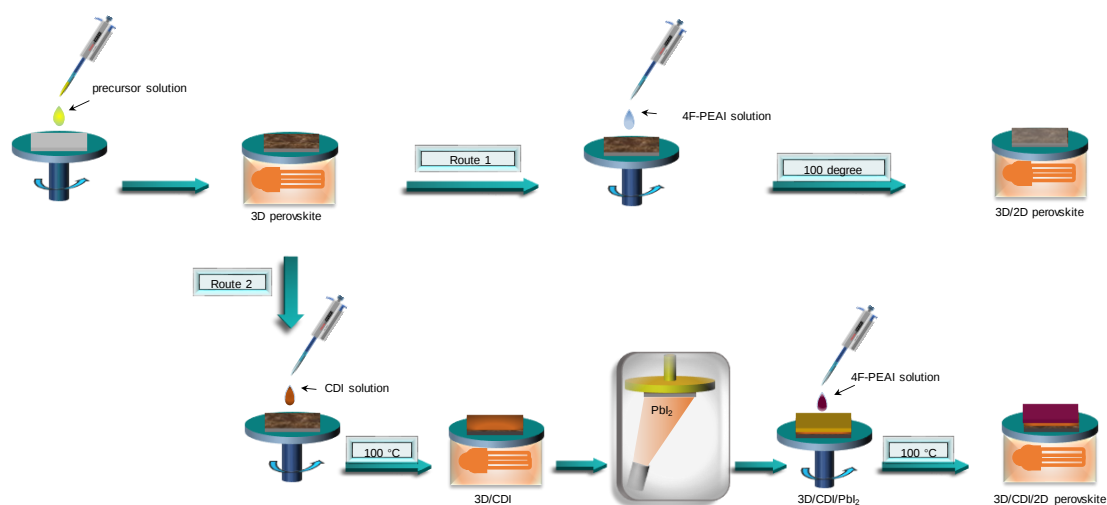
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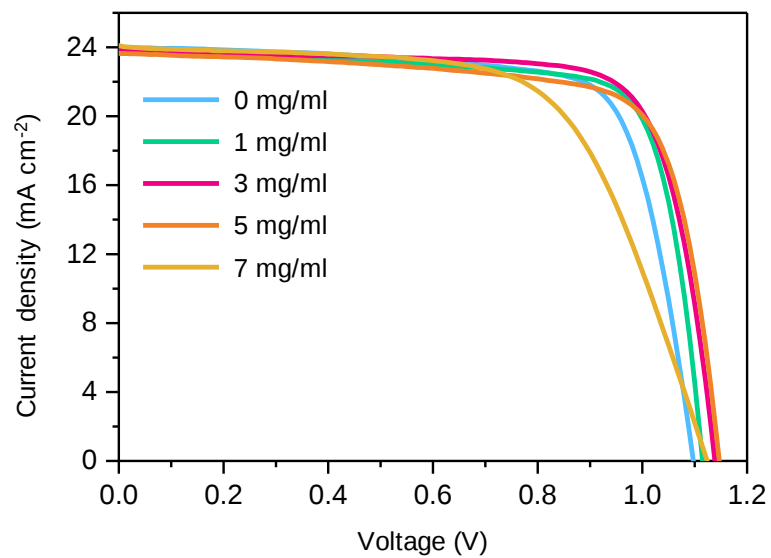
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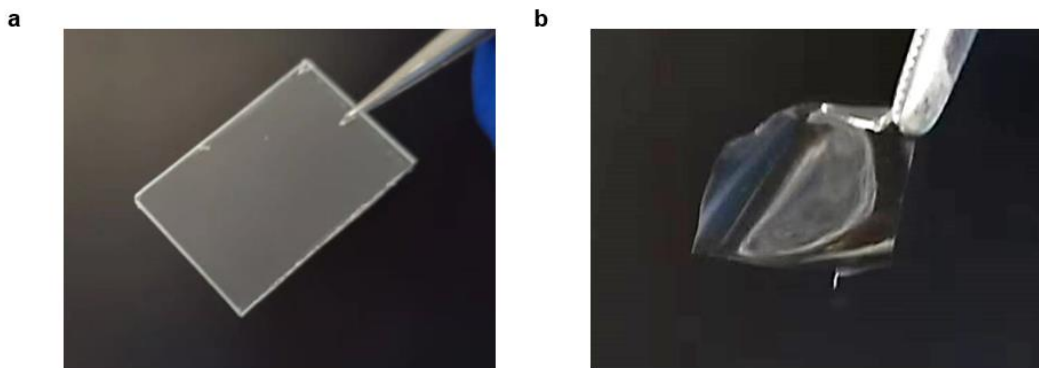
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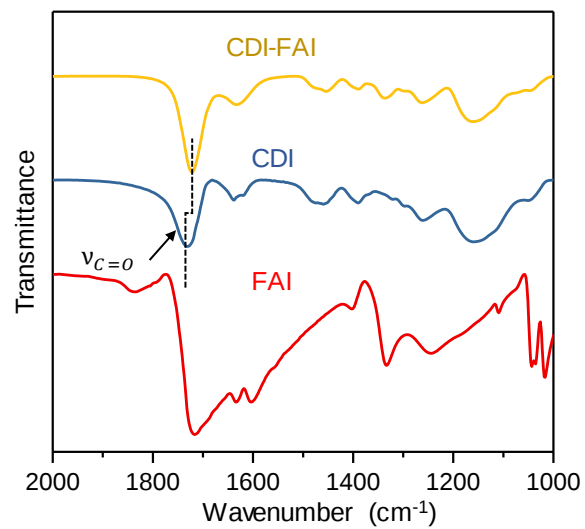
Supplementary Figure 1 | Fabrication routes of perovskite heterostructures. Route 1 is the traditional preparation process of a 3D/2D perovskite heterostructure by directly spin-coating 4F-PEAI/IPA solution on the 3D perovskite surface. Route 2 is the preparation process of a 3D/CDI/2D perovskite heterostructure where the 2D perovskite is prepared by a vapor-assisted two-step process.



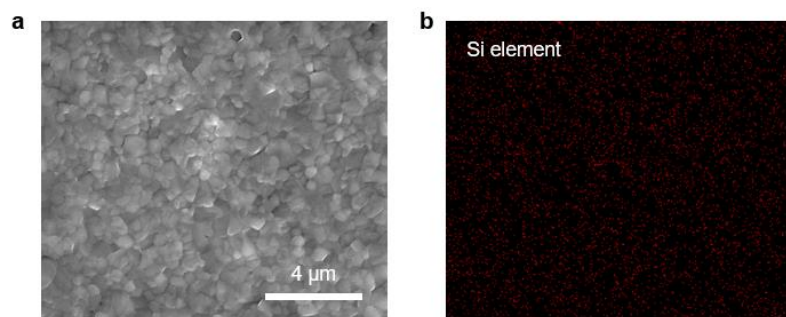
Supplementary Figure 2 | J-V curves of 3D/CDI/2D devices with different concentrations of CDI precursor solutions. The optimized concentration of CDI precursor solutions is 3 mg/ml.



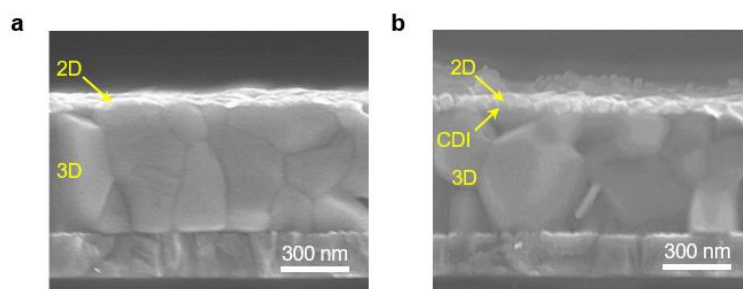
Supplementary Figure 3 | Photo of a CDI film on a glass substrate. **a**, The CDI thin film was deposited on the glass substrate by spin coating the CDI precursor solution (30 mg mL^{-1} in tetrahydrofuran) at 200 rpm for 30 seconds. **b**, The CDI thin film was separated from glass substrate.



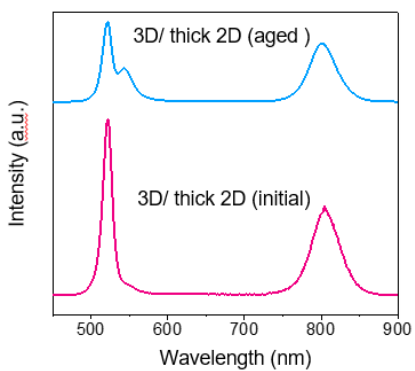
Supplementary Figure 4 | FTIR spectra of FAI, CDI and CDI-FAI.



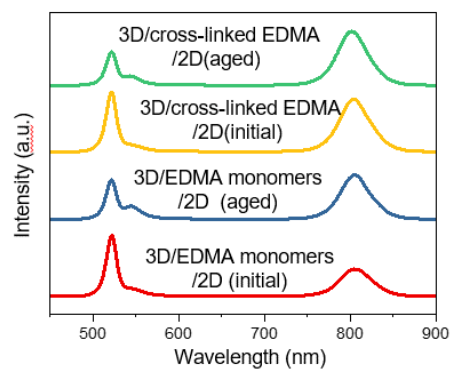
Supplementary Figure 5 | Morphology and elemental distribution of a 3D perovskite/CDI film. Surface SEM image (a) and the corresponding EDS (b) for a 3D perovskite/CDI film.



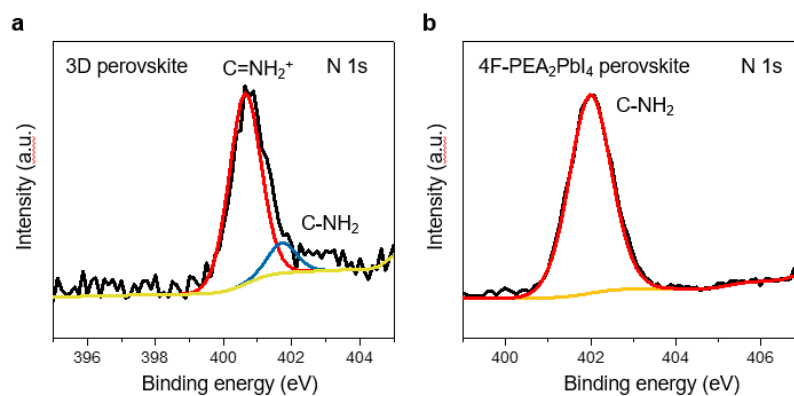
Supplementary Figure 6 | The increased 2D thickness of 3D/CDI/2D perovskite heterostructures. Cross-sectional SEM images of the 3D/2D (a) and 3D/CDI/2D (b) perovskite heterostructures.



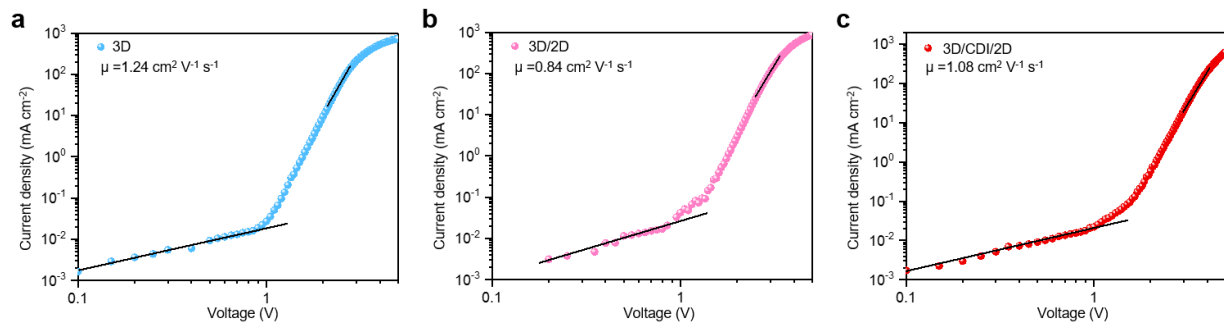
Supplementary Figure 7 | Steady-state PL spectra measurement. Steady-state PL spectra of initial and thermally-aged 3D/2D perovskite heterostructures with a thick 2D perovskite prepared by the same vapor-assisted two-step process as in the 3D/CDI/2D perovskite heterostructure.



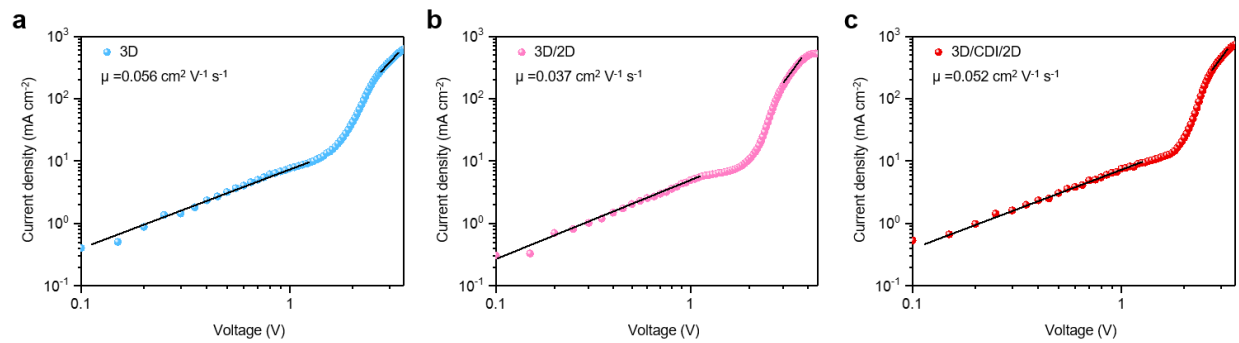
Supplementary Figure 8 | Stability of 3D/2D perovskite heterostructures with different barrier layer. Steady-state PL spectra of initial and thermally-aged 3D/2D perovskite heterostructures with EDMA monomer and crosslinked EDMA polymer as the barrier layer.



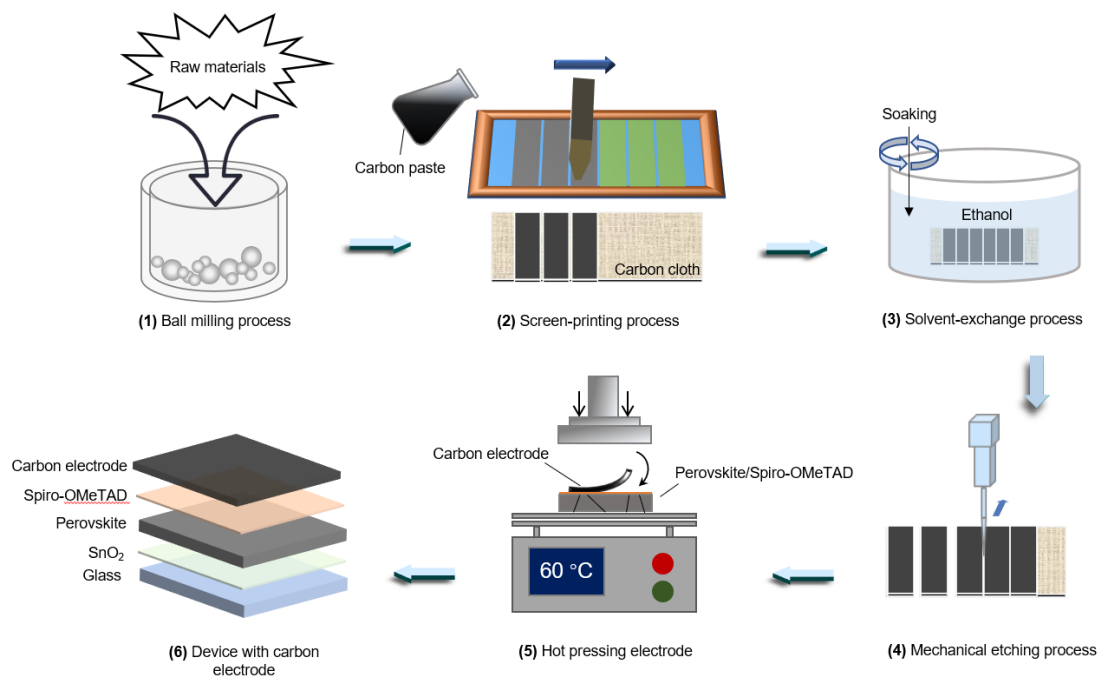
Supplementary Figure 9 | XPS spectra measurement. XPS spectra of 3D (FAPbI₃)_{0.95}(MAPbBr₃)_{0.05} perovskite (a) and 2D 4F-PEA₂PbI₄ perovskite (b).



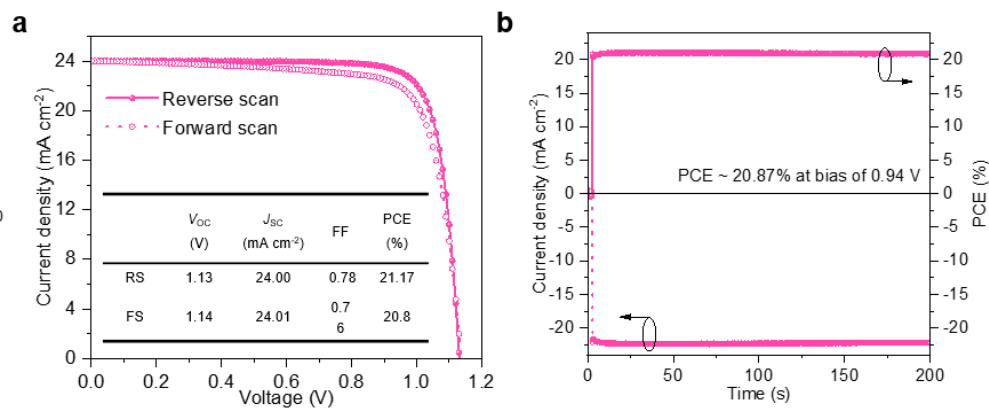
Supplementary Figure 10 | Dark J - V characteristics of electron-only devices (ITO/SnO₂/perovskite heterostructure/PCBM/Au).



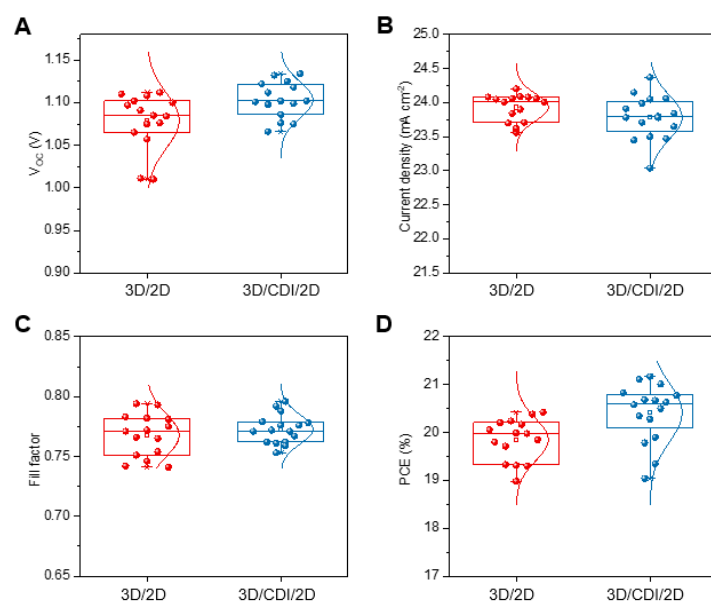
Supplementary Figure 11 | Dark J - V characteristics of hole-only devices (ITO/PTAA/perovskite heterostructure /Spiro/Au).



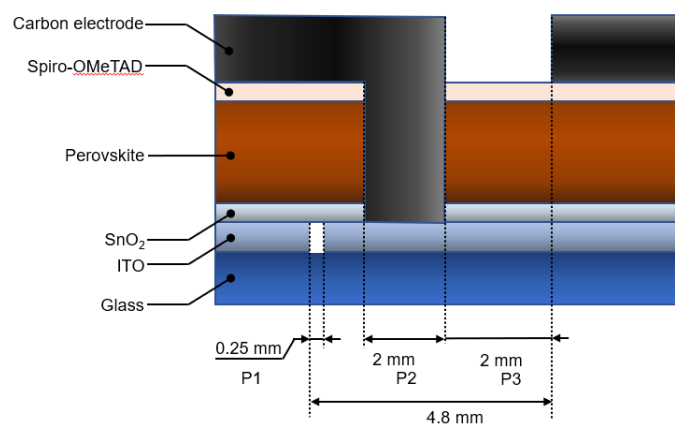
Supplementary Figure 12 | Preparation process of carbon electrode based on the screen-printing technique.



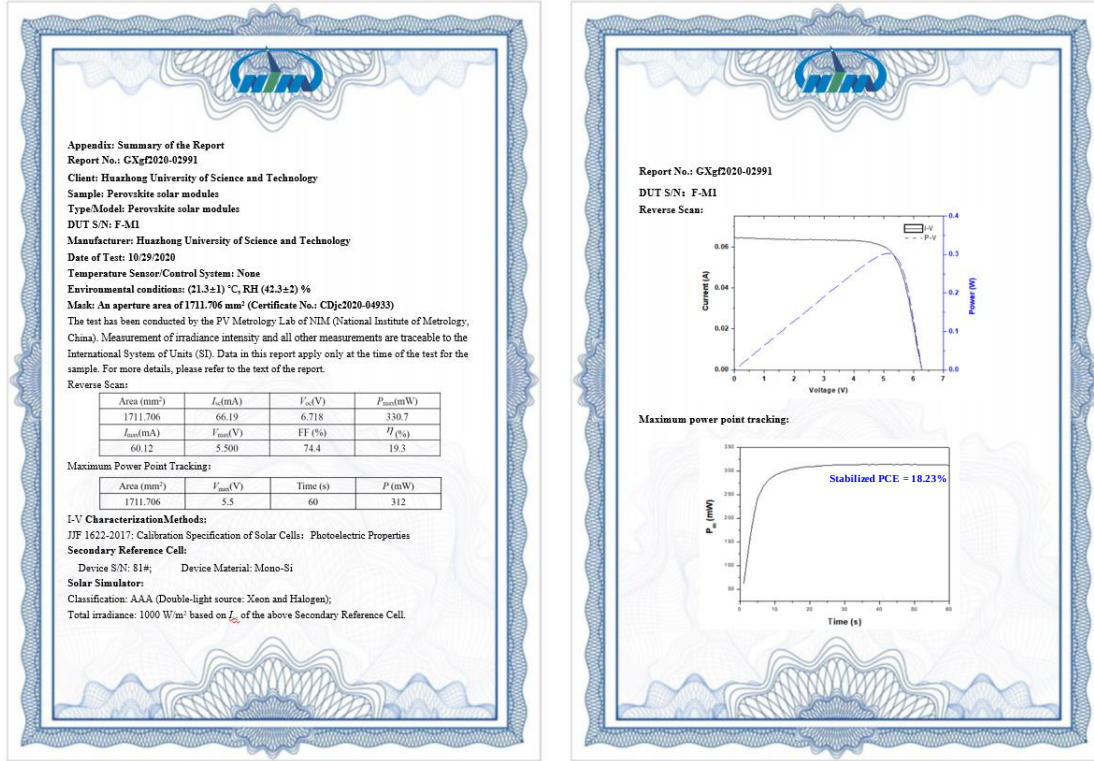
Supplementary Figure 13 | Device performance of 3D/CDI/2D PSCs. J-V curves (a) and the stabilized output and calculated PCE of the device at a fixed bias (b).



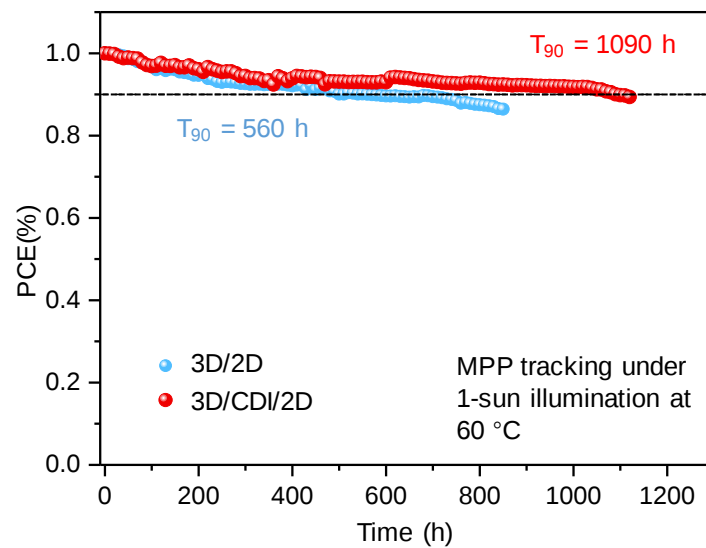
Supplementary Figure 14 | The statistics of photoelectric performance parameters. Comparison of the device performance obtained by 3D/2D and 3D/CDI/2D perovskite heterostructures.



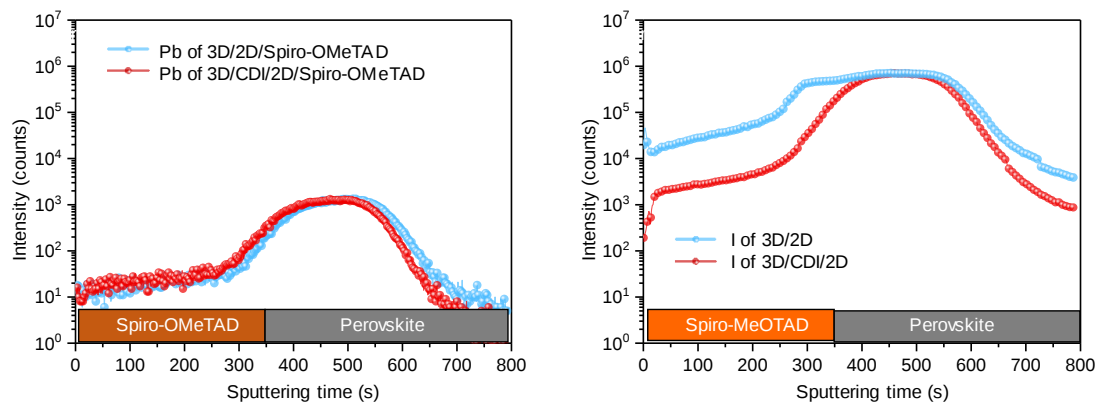
Supplementary Figure 15 | Side view of the PSM.



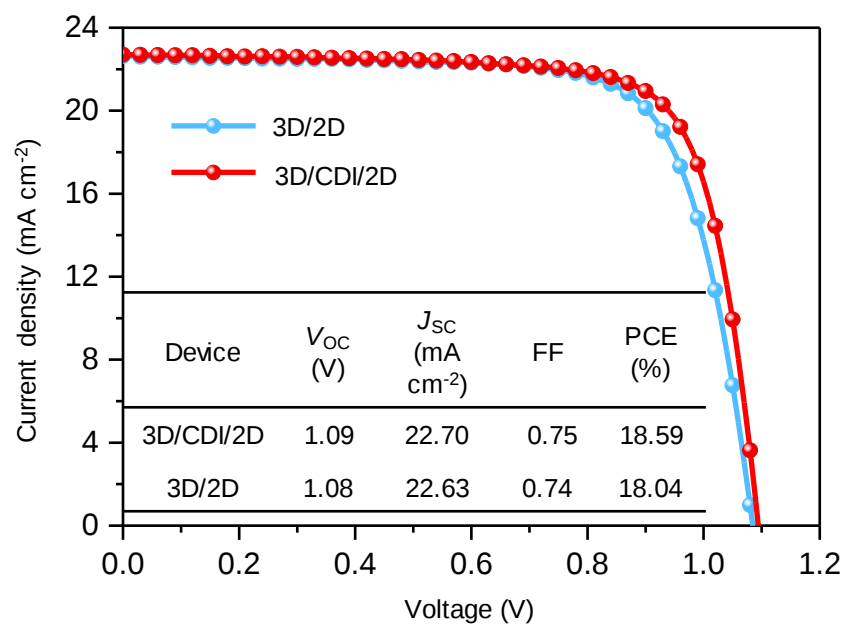
Supplementary Figure 16 | Certification of a PSM based on the 3D/CDI/2D perovskite heterostructure and carbon electrode. The aperture area is 17.12 cm², and the stabilized PCE is 18.23%. Certification was carried out by the National Institute of Metrology in China.



Supplementary Figure 17 | The operational stability of PSCs (SnO₂/perovskite heterostructure/Spiro-OMeTAD/Carbon) under continuous illumination in N₂ at ~60 °C. For the 3D/2D device, the initial efficiency is 20.15%. For the 3D/CDI/2D device, the initial efficiency is 21.02%.



Supplementary Figure 18 | TOF-SIMS of Pb and I ions in Spiro-OMeTAD and the perovskite layer.



Supplementary Figure 19 | Device performance obtained with P3HT-Spiro-OMeTAD. The *J*-*V* curves of PSCs with the structure of SnO_2 / (3D/2D perovskite)/P3HT-Spiro-OMeTAD/Carbon, and SnO_2 / (3D/CDI/2D perovskite)/P3HT-Spiro-OMeTAD/Carbon.

Supplementary Table 1 |. Summary of reported operating stability of PSCs. The perovskite layer is treated with bulky organics, and the back contact is either metal (Au, Ag and Cu) or carbon. (Ref ¹⁻¹⁹).

Perovskite passivation	Electrode	PCE/% (Aperture area/cm ²)	Lifetime (MPP)	T (°C)	Journal/Year
CDI/4F-PEAI	Carbon	21.17 (0.16)	T ₉₀ = 4392 hrs	60 °C	This work
CDI/4F-PEAI	Carbon	18.2 (17.1)	-	-	This work-
MeO-PEAI (ST)	Au	25.8 (0.096)	T ₉₀ = ~ 500 hrs	-	<i>Nature</i> (2021, Ref. 1)
FAHCOO (BT) /OAI (ST)	Au	25.6 (0.08)	T ₈₅ = 450 hrs	35	<i>Nature</i> (2021, Ref. 2)
DMePDAI ₂ (ST)	Au	24.7 (0.12)	T ₉₀ = 1000 hrs	~ 40	<i>Science</i> (2021, Ref. 3)
OLAI (ST)	Ag	24.3 (0.07)	T ₉₅ = 500 hrs	~ 40	<i>Science</i> (2022, Ref. 4)
3F-PEAI/MAI (ST)	Ag	23.3 (0.05)	No degradation = 1000 hrs	RT	<i>Nat. Photon.</i> (2022, Ref. 5)
3F-PEAI/MAI (ST)	Ag	23.3 (0.05)	T ₉₂ = 500 hrs	65	<i>Nat. Photon.</i> (2022, Ref. 5)
choline chloride (ST)	Au	23.1 (0.16)	T ₉₀ = 500 hrs	RT	<i>Science</i> (2020, Ref. 6)
i-BABr (ST)	Au	20.42 (17.1)	T ₈₂ = 500 hrs	RT	<i>Science</i> (2021, Ref. 7)
BA ₂ PbI ₄ (ST)	Au	24.35 (0.094)	T ₉₈ = 1620 hrs	-	<i>Nat. Energy</i> (2021, Ref. 8)
EAI/MAI (ST)	Au	16.6 (22.4)	T ₉₀ = 1570 hrs	40	<i>Nat. Energy</i> (2020, Ref. 9)
OATsO (ST)	Au	24 (0.13)	No degradation = 800 hrs	40	<i>Nature</i> (2022, Ref. 10)
iPAmHCl (BT)	Au	23.9 (0.16)	No degradation, 2000 hrs	RT	<i>Nat. Energy</i> (2021, Ref. 11)
-	Au	24.04 (0.08)	T ₉₅ = ~ 2000 hrs	RT	<i>Science</i> (2021, Ref. 12)
-	rGO/ Au	20.4 (0.16)	T ₉₅ = 1000 hrs	60	<i>Science</i> (2019, Ref. 13)
-	Ag	22.02 (0.13)	T ₉₂ = 1000 hrs	85	<i>Nat. Nanotechnol.</i> (2020, Ref. 14)
PbPyA ₂ /TMS (ST)	Cu	24.3 (0.09)	T ₉₀ = 1000 hrs	55	<i>Science</i> (2022, Ref. 15)
lead sulfate (ST)	Cu	21.1 (-)	T ₉₇ = 1200 hrs	65	<i>Science</i> (2019, Ref. 16)
-	Cu	20.1 (17.9)	No degradation = 550 hrs	60	<i>Science</i> (2021, Ref. 17)
-	Ti ₁ - rGO/FTO	20.6 (0.09)	T ₉₅ = 1300 hrs	60	<i>Nat. energy</i> (2021, Ref. 18)
-	Au/MgF ₂	20.9 (0.113)	No degradation, 1400 hrs	60~65	<i>Nat. Energy</i> (2021, Ref. 19)

Note:

4-methoxy-phenethylammonium iodide: MeO-PEAI; octylammonium iodide: OAI; Octylammonium tosylate: OATsO; N, N-dimethyl-1,3-propane diammonium diiodide: DMePDAI₂; Oleylammonium iodide: OLAI; isobutylamine bromide: i-BABr. 3-iodopropyl trimethoxysilane [Si (OCH₃)₃(CH₂)₃I]: I-SAM; Polyethylenimine: PEIE; poly[2,2'""-bis[[[(2-butyloctyl)oxy]carbonyl] [2,2':5',2'':5'',2'''-quaterthiophene]-5,5'''-diyl]]: PDCBT; BCF-doped poly(triarylamine): PTAA-BCF; 4-fluoro-phenethylammonium iodide: 4F-PEAI; 3-fluoro-phenethylammonium: 3F-PEAI; pyrene-based ethylammonium iodide: PREAI; isopropylammonium chloride: iPAmHCl; n-hexyl trimethyl ammonium bromide: HTABr; ethylenediaminetetraacetic acid dipotassium salt: EDTAK;

ethylammonium iodide: EAI; aminovaleric acid iodide: AVAI; 5-ammoniumvaleric acid iodide: 5-AVAI; 1-butyl-3-methylimidazolium tetrafluoroborate: BMIMBF₄; Pyridine-2-carboxylic acid: PbPyA₂; Hexamethyldisilathiane: TMS; Maximum power point tracking: MPPT; Surface treatment: ST; Bulk treatment: BT; Room temperature: RT.

Supplementary Table 2 | The fitted parameters from TRPL of various perovskite samples.

Samples	A ₁	τ ₁	A ₂	τ ₂	τ _{avg} (ns)
Pristine 3D	0.569	91.3	0.431	455.42	248.24
3D/2D	0.527	83.72	0.473	573.65	315.46
3D/CDI/2D	0.491	89.17	0.509	824.72	423.56
Pristine 3D/ Spiro-OMeTAD	0.645	5.62	0.355	55.02	23.16
3D/2D/Spiro- OMeTAD	0.639	7.71	0.361	68.15	29.53
3D/CDI/2D /Spiro-OMeTAD	0.641	6.73	0.359	62.1	26.61

Note: The measured PL decay kinetics are fitted by bi-exponential decay function as follow

$$F(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2)$$

$$\text{The average lifetime: } \tau_{avg} = (A_1 * \tau_1 + A_2 * \tau_2)/(A_1 + A_2)$$

Supplementary References

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