

Supplementary Information

Precision synthesis and atomistic analysis of deep blue cubic quantum dots made via self-organization

Authors: Olivier J. G. L. Chevalier¹, Takayuki Nakamuro^{1*}, Wataru Sato¹, Satoru Miyashita¹, Takayuki Chiba², Junji Kido², Rui Shang^{1*}, Eiichi Nakamura^{1*}.

Affiliations:

¹ Department of Chemistry, The University of Tokyo, 7-3-1 Bunkyo-ku, Tokyo, 113-0033, Japan.

² Graduate School of Organic Materials Science, Yamagata University, Yonezawa, Yamagata, 992-8510, Japan.

*Corresponding authors. Email: muro@chem.s.u-tokyo.ac.jp, ruishang@chem.s.u-tokyo.ac.jp, nakamura@chem.s.u-tokyo.ac.jp.

Table of contents

1. Supplementary Notes.....	3
2.1. Determination of the ideal QD ₄ structure.....	3
2.2. Estimation of the purity via size distribution.....	3
2.3. Estimation of the purity via photoluminescence spectra.....	4
2. Mass spectrum of the precursor suspension	5
3. Representative SEM images of MAPbBr ₃ microcrystals	6
4. Representative SEM images of MAPbBr ₃ microcrystals with malic acid	7
5. Representative SEM images of MAPbBr ₃ microcrystals with lactic acid.....	8
6. Representative SEM images of MAPbBr ₃ microcrystals with succinic acid	9
7. Representative SEM images of MAPbBr ₃ microcrystals with oleic acid	10
8. Heatmap of the reaction kinetics	11
9. Emitted wavelength and linewidth through reaction time.....	12
10. Circular dichroism spectra.....	13
11. Representative TEM images of MLA•MA•QD ₄ and MLA•Cs•QD ₄ on aC	14
12. Fluorescence lifetimes	15
13. Fluorescence spectra through storage at low temperature.....	16
14. Infrared spectra	17
15. EDX table (in equivalents to Pb ²⁺)	18
16. Representative SMART-EM images of MLA•Cs•QD ₄	19
17. Representative SMART-EM images of MLA•Cs•QD ₄	20
18. Representative SMART-EM images of MLA•MA•QD ₄	21
19. Overall SMART-EM pictures of MLA•Cs•QD ₄ and MLA•MA•QD ₄	22
20. FFT patterns of MLA•Cs•QD ₄	23
21. Captions for Video S1 to S3	24
22. Supplementary References	24

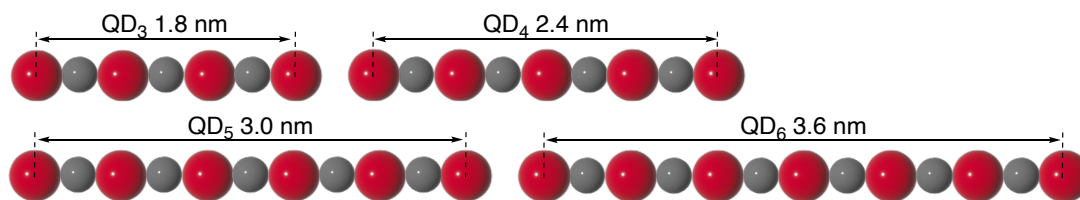
Supplementary Notes

Supplementary Note 1: **Cs-QD₄**: neutral Cs₁₁₂Pb₆₄Br₂₄₀ vs. geometrically perfect [Cs₁₂₅Pb₆₄Br₂₄₀]¹³⁺.

Cs-QD₄, Cs₁₁₂Pb₆₄Br₂₄₀, discussed in Fig. 1c, 5a, and 5b is an archetypal "neutral" lead perovskite QD₄, where all Pb atoms bear eight Br atoms and hence coordinatively saturated, and 112 negative charges are neutralized by 112 Cs cations^{48,49}. Of these 112 Cs, 27 are located in the core of the QD, and 85 on the surface. A cautionary note is that this structure leaves some potential Cs cation sites vacant as illustrated in its side view in Fig. 5b, where we find only 10 Cs cations on the edge instead of 12. The formation of a structure of reduced symmetry arises from mismatch between the charge neutrality of **Cs-QD₄** and the geometry of the cube; that is, **Cs-QD₄** requires only 112 Cs cation, but the cubic geometry creates 125 potential Cs sites on the surface. On other word, if we fill all 125 sites by Cs cations, **Cs-QD₄** become [Cs₁₂₅Pb₆₄Br₂₄₀]¹³⁺, which we do not necessarily consider here for species stably forming in a non-polar solvent such as toluene. Note that we find none of this problem in a bulk crystal where the number of Cs cation on edges is negligibly small.

Supplementary Note 2: Estimation of the size distribution of QDs on amorphous carbon (aC) film

The size distribution of the QDs on aC summarized in Fig. 4 was determined by using the standard deviation of the size determined by repeated measurement of single QDs of different sizes on aC (cf. Fig. 3b, Supplementary Fig. 10). This estimation should be valid because the QD are cubic as supported by the SMART-EM images (Fig. 5), and the QDs are defect-free and their surface fully coordinated by Br atoms as supported by the PLQY, EDX and SMART-EM data (Fig. 5). Note that the contrast of the MA, OAM, and OLA groups is too low to be detectable under the present TEM conditions, and hence Pb, Br, and Cs atoms can be seen (cf. Fig. 3a). As such, a QD₃ has 3 Pb atoms and 4 Br atoms edge wise, measuring 1.8 nm. Similarly, a QD₄ measures 2.4 nm, a QD₅ 3.0 nm and a QD₆ 3.6 nm. By measuring a single QD on aC, we can estimate a standard deviation on the QD measurement of 0.17 nm. We then use Gaussian distributions centered on each size. Line widths were adjusted to 0.17 nm and intensities were adjusted until the sum of gaussian functions best matched the total size distribution. Population fractions were estimated as a percentage of the area below the curve compared to the total area of the reconstructed size distribution.

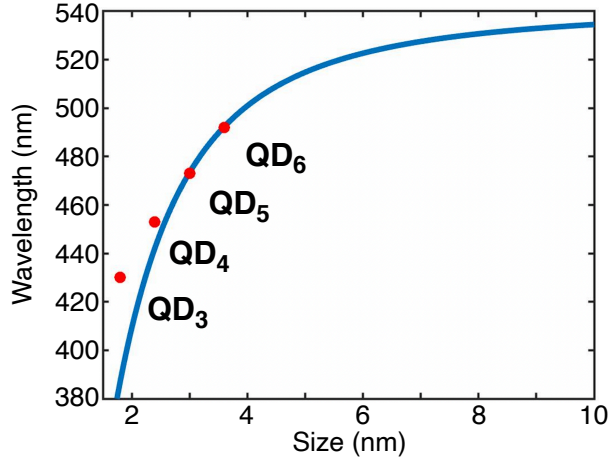


Supplementary Note 3: Estimation of the purity via photoluminescence spectra

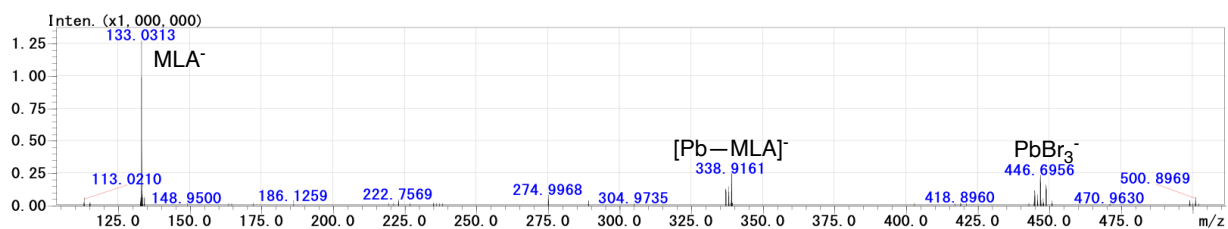
First, we estimated the emitted wavelength for each one of the four species we identified in the size distribution analysis via the Brus equation. Namely, we fitted the following equation:

$$E(r) = E_{bulk} + \frac{h^2}{8r^2} * \frac{1}{\mu}$$

where h is Planck's constant, $E_{\text{bulk}} = 2292 \text{ meV}$ and $\mu = 0.127 m_e$ ⁵⁰. We factor in a deviation to ideality for the theoretical QD₃ estimated from the deviation seen with the experimental QD₄. We then used theoretical values for both QD₅ and QD₆ and Gaussian distributions centered on each wavelength. Linewidths and intensities were adjusted until the sum of gaussian functions best matched the photoluminescence spectra. Population fractions were estimated as a percentage of the area below the curve compared to the total area of the reconstructed PL spectra.

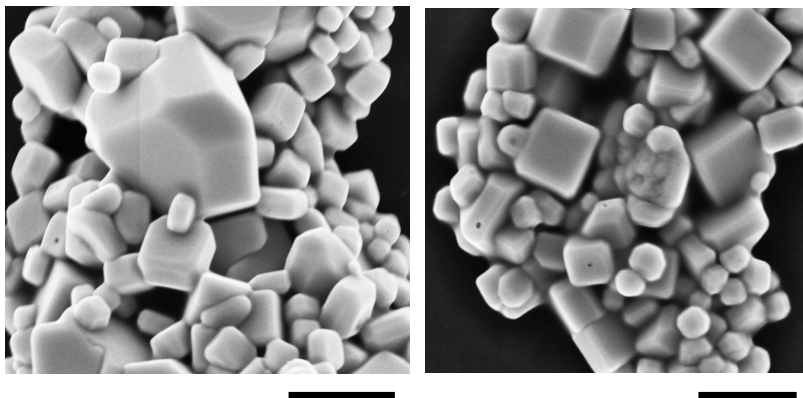


Supplementary Figures

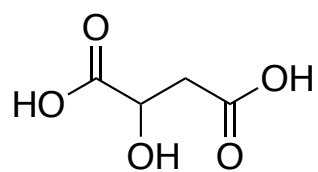


Supplementary Figure 1 | Negative ion mass spectra of the white precursor suspension, where we find three negative ions that serve as building blocks of the QD₄ as illustrated in Fig. 1d.

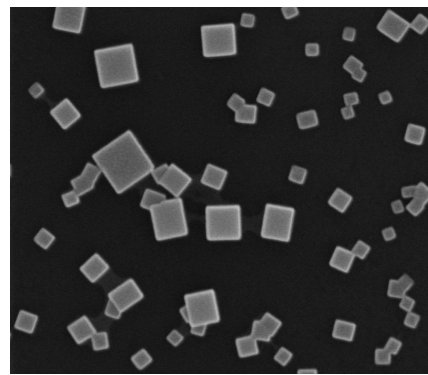
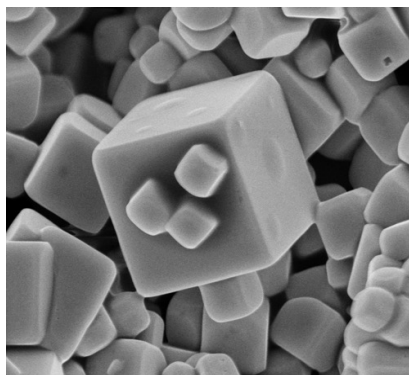
1:1 PbBr_2 MABr in DMF
diluted with toluene
without acid and amine



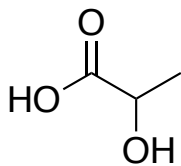
Supplementary Figure 2 | Representative SEM images of MAPbBr_3 microcrystals prepared without any ligands. Scale bars 500 nm.



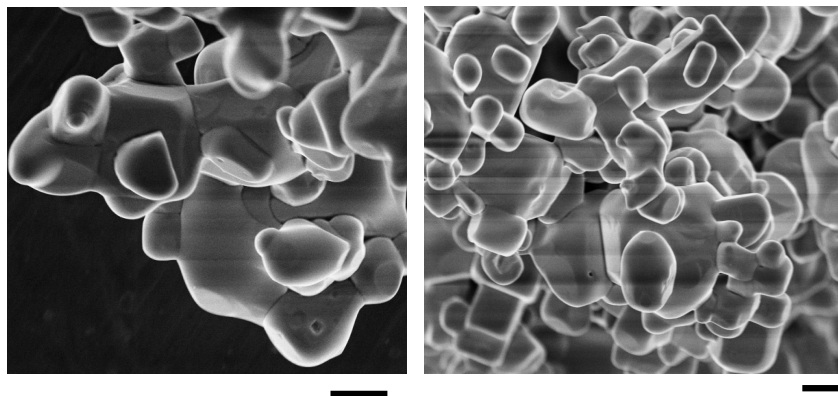
4.0 equiv malic acid
(MLA)



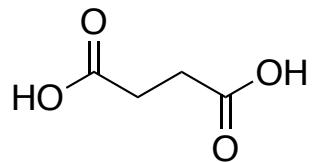
Supplementary Figure 3 | Representative SEM images of MAPbBr₃ microcrystals prepared with 4.0 equiv MLA to Pb. Scale bars 1 μ m.



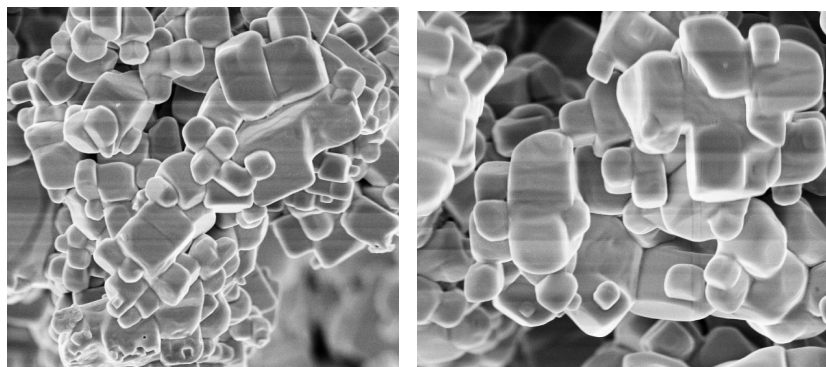
0.5 equiv lactic acid



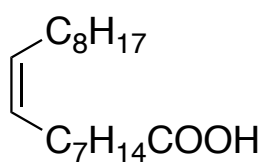
Supplementary Figure 4 | Representative SEM images of MAPbBr₃ microcrystals prepared with 0.5 equiv lactic acid to Pb. Scale bars 1 μ m.



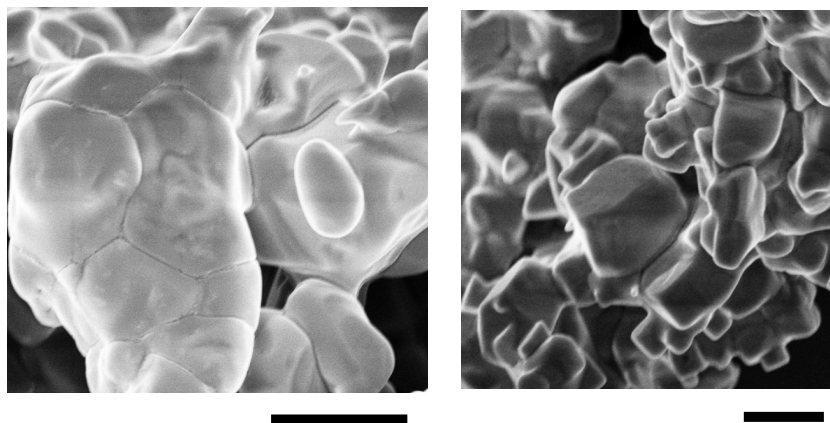
0.5 equiv succinic acid



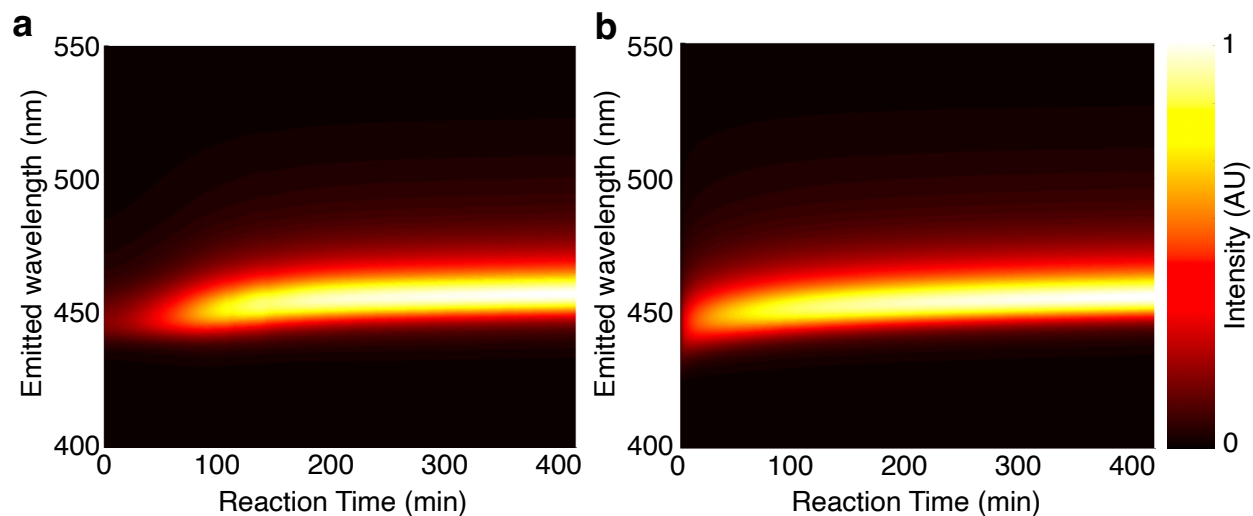
Supplementary Figure 5 | Representative SEM images of MAPbBr₃ microcrystals prepared with 0.5 equiv succinic acid to Pb. Scale bars 1 μm.



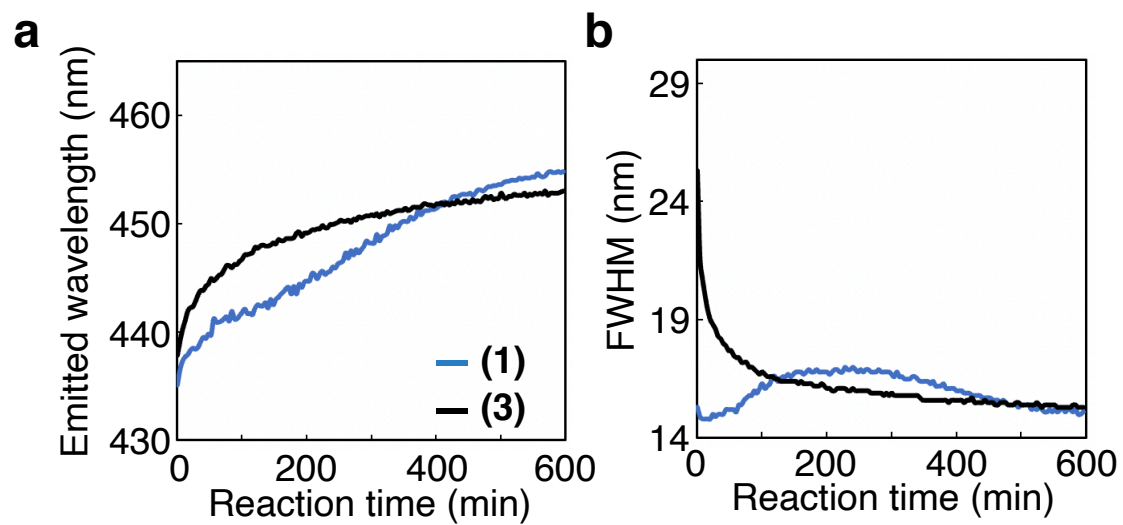
0.5 equiv oleic acid
(OA)



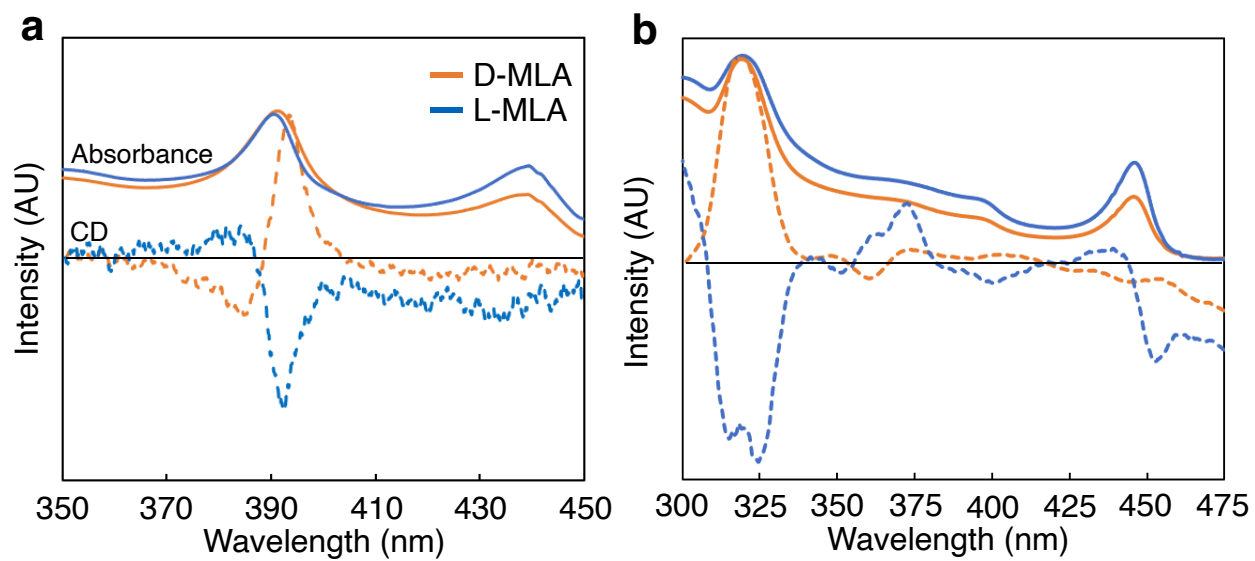
Supplementary Figure 6 | Representative SEM images of MAPbBr₃ microcrystals prepared with 0.5 equiv OA to Pb. Scale bars 1 μm.



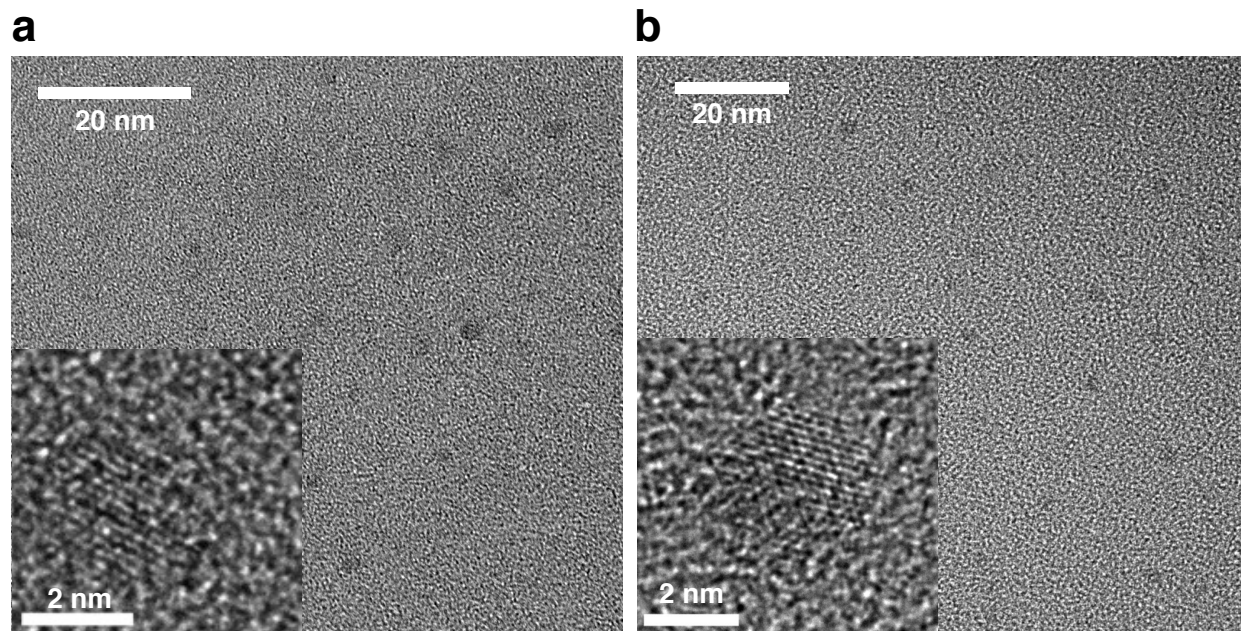
Supplementary Figure 7 | Heatmap of the reaction kinetics for the first 400 minutes. PL spectra were taken every 3 minutes over the course of the reaction for **a**, MLA•MA•QD₄ (**1**), and **b**, OA•MA•QD₄ (**3**).



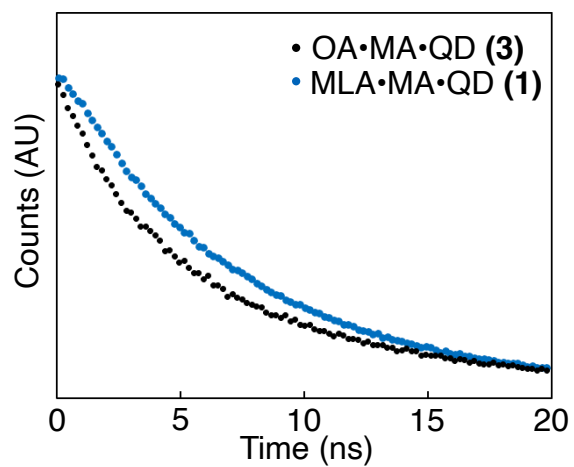
Supplementary Figure 8 | Evolution of the emitted wavelength **a**, and FWHM **b**, for **1** and **3** through the reaction, extracted from the heatmaps in the Supplementary Figure 7.



Supplementary Figure 9 | Circular dichroism and absorbance spectra with chiral MLA. **a**, MLA•MA•QD₄. **b**, MLA•Cs•QD₄.

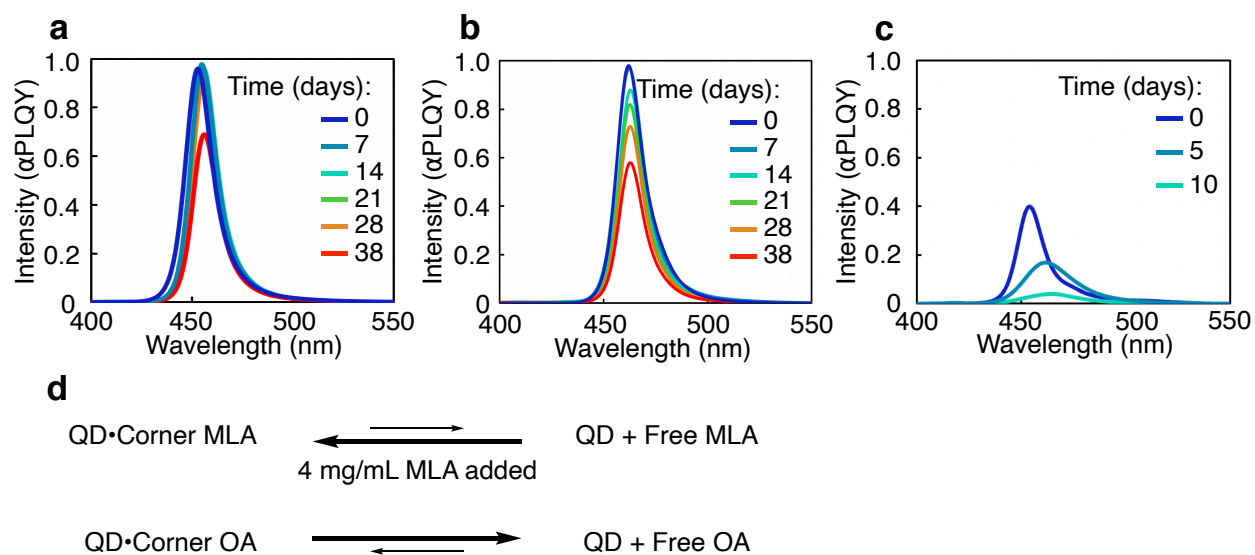


Supplementary Figure 10 | Representative TEM images of QDs on the surface of aC of 6 nm thickness. Low and high magnification of images are shown. **a**, MLA•MA•QD₄ (1). **b**, MLA•Cs•QD₄ (2).

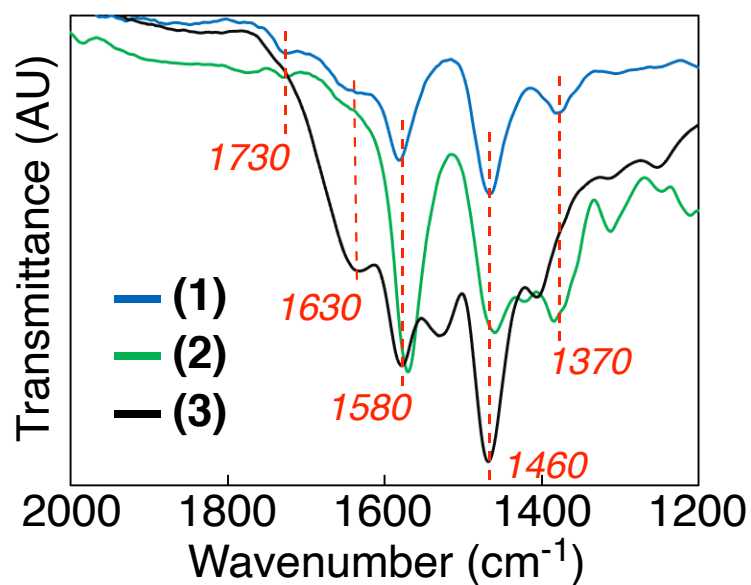


Ligand	A ₁ (%)	τ ₁ (ns)	A ₂ (%)	τ ₂ (ns)	<τ> (ns)	k _r (x10 ⁷ s ⁻¹)	k _{nr} (x10 ⁷ s ⁻¹)	τ _r (ns)	τ _{nr} (ns)
OA•MA•QD	89.5	6.3	10.5	26.1	12.7	1.89	5.98	53	16.7
MLA•MA•QD	87.5	7	12.5	17.3	9.0	10.6	0.5	9.4	200

Supplementary Figure 11 | Fluorescence lifetime and results for a biexponential modeling.



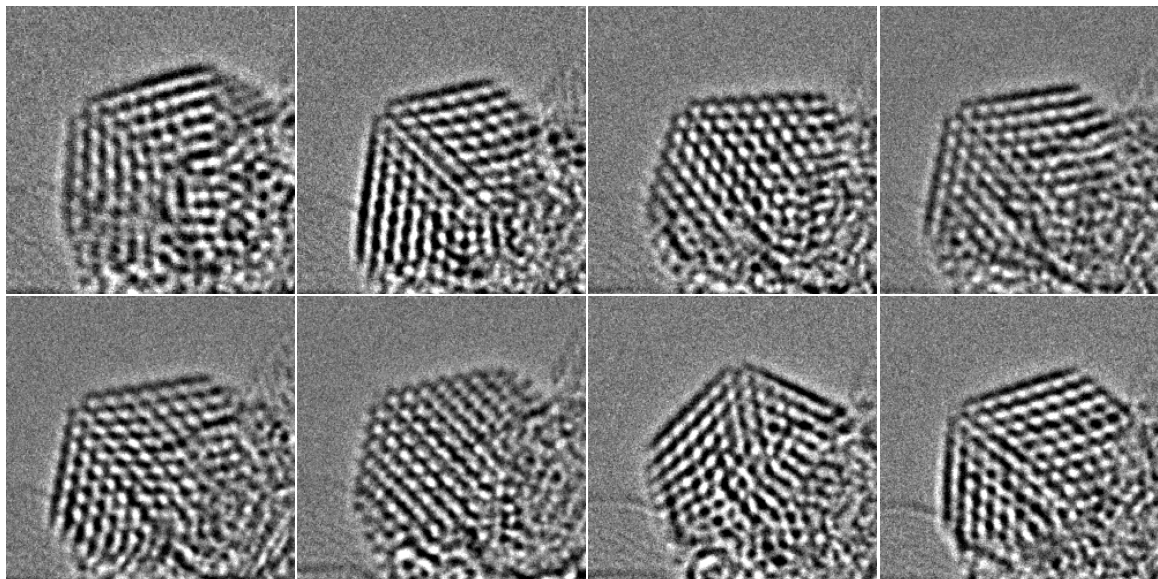
Supplementary Figure 12 | Fluorescence spectra through storage time at 5 °C for **a**, MLA•MA•QD₄ (**1**) in the presence of added MLA. **b**, MLA•Cs•QD₄ (**2**) in the presence of added MLA. **c**, OA•MA•QD₄ (**3**). Intensity for each spectrum was set equal to the photoluminescence quantum yield. **d**, Addition of MLA displaces the equilibrium towards bound MLA, protecting the QD.



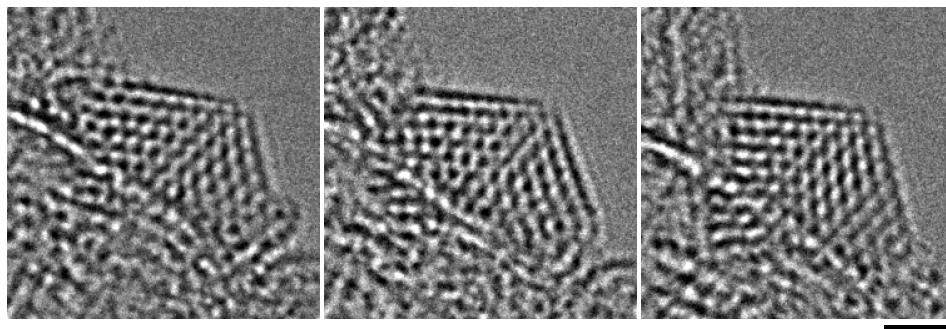
Supplementary Figure 13 | Film FTIR spectra for MLA•MA•QD₄ **(1)**, MLA•Cs•QD₄ **(2)** and OA•MA•QD₄ **(3)** Peaks are attributed as follow: C=O stretching (1730 cm^{-1}), N-H bending (1630 cm^{-1}), antisymmetric COO⁻ stretching (1580 cm^{-1}), symmetric COO⁻ stretching (1460 cm^{-1}) and alcohol O-H bending (1370 cm^{-1})⁵¹.

Compound	O (eq to Pb)	Br (eq to Pb)	Cs (eq to Pb)	Ligand (eq to Pb)
MLA•MA•QD ₄ (1)	0.7 ± 0.1	3.3 ± 0.1	-	0.14 ± 0.02
MLA•Cs•QD ₄ (2)	0.75 ± 0.2	3.7 ± 0.1	0.9 ± 0.03	0.15 ± 0.03
OA•MA•QD ₄ (3)	0.27 ± 0.02	3.0 ± 0.1	-	0.14 ± 0.01

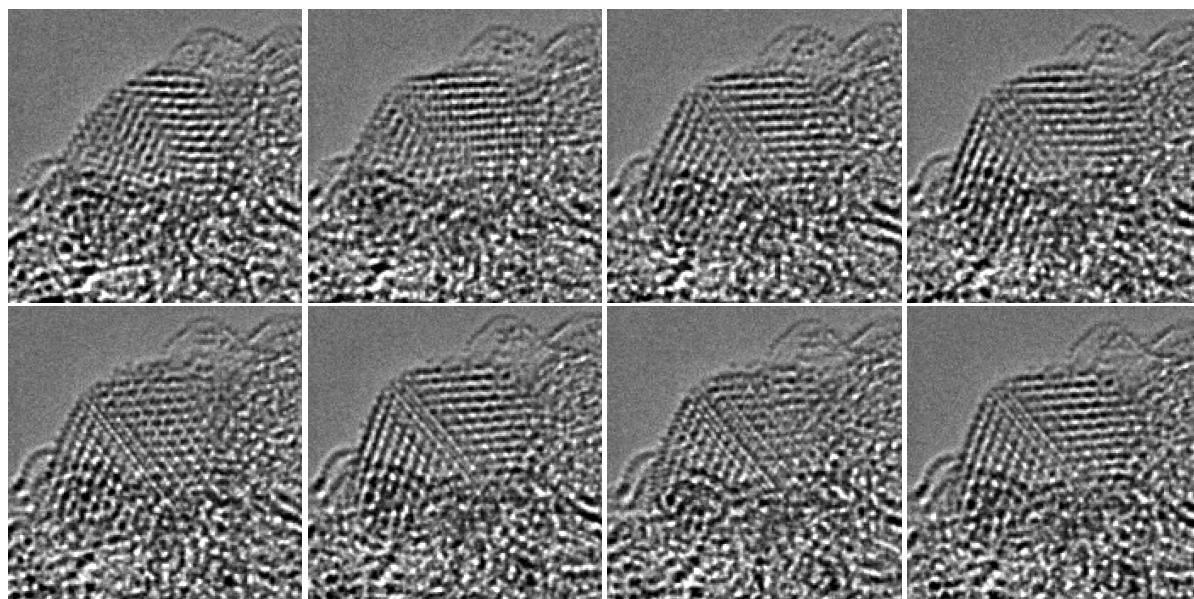
Supplementary Figure 14 | EDX results using the amount of Pb as an internal standard. Results averaged over 20 measurements.



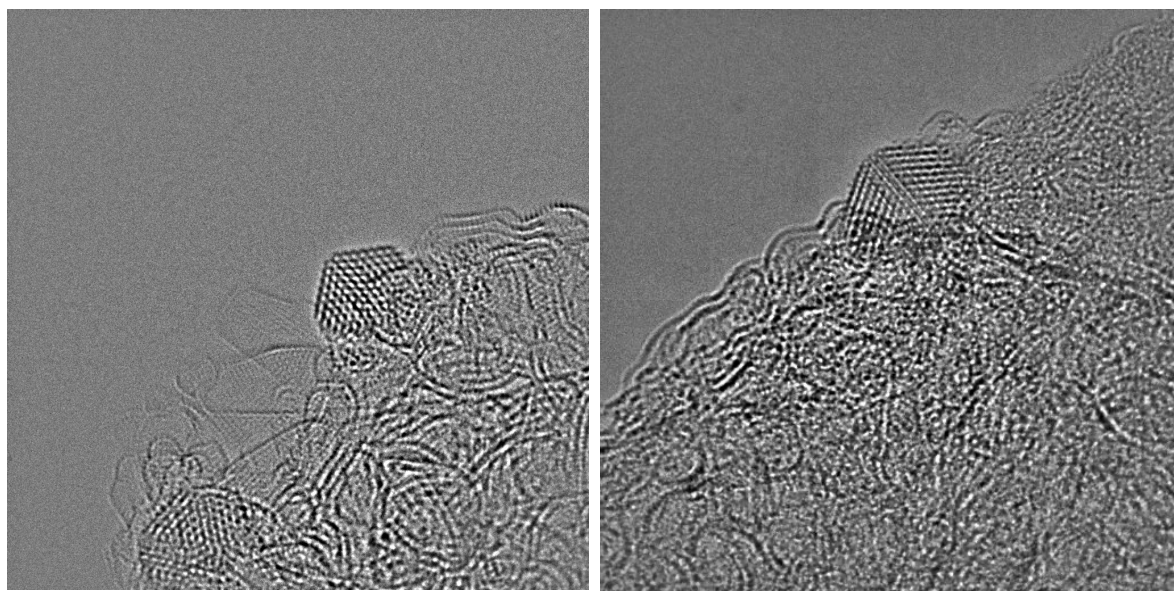
Supplementary Figure 15 | Representative SMART-EM images of **2** tumbling on CNT (Movie S1). EDR = $2.8 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$. Frame rate of 20.0 ms frame⁻¹ or 50 fps, and scale bar 1 nm.



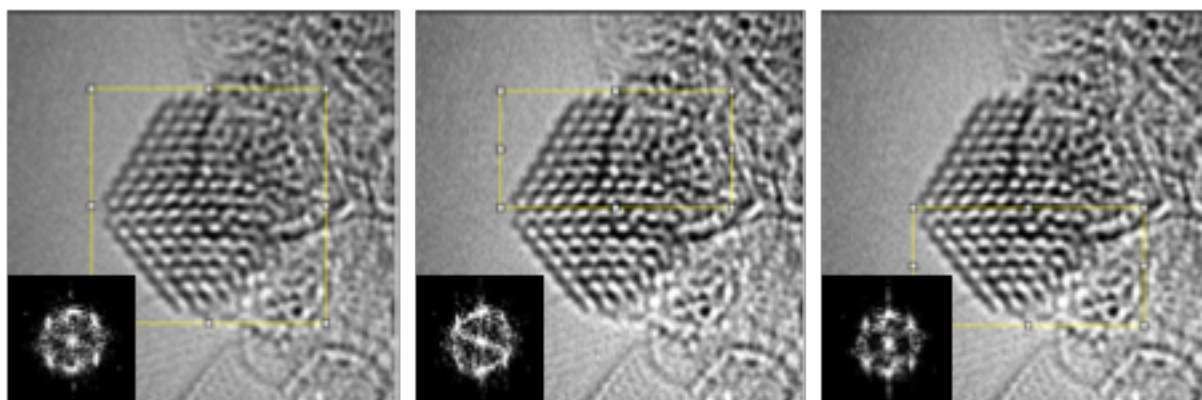
Supplementary Figure 16 | Representative SMART-EM images of **2** tumbling on CNT. EDR = $2.4 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$. Frame rate of $20.0 \text{ ms frame}^{-1}$ or 50 fps, and scale bar 1 nm.



Supplementary Figure 17 | Representative SMART-EM images of **1** vibrating on CNT (Movie S3). EDR = $2.1 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$. Frame rate of $13.3 \text{ ms frame}^{-1}$ or 75 fps , and scale bar 1 nm .



Supplementary Figure 18 | Overall views of QDs seen as a cube on the surface of aggregate of CNTs and graphitic structures observed by SMART-EM. Graphitic materials are more conductive by thousand times and hence more electron-donative than amorphous carbon film. Left: **2**. Right: **1**. The TEM scale was internally calibrated using the graphitic sites of CNTs. Scale bar 1 nm.



Supplementary Figure 19 | FFT patterns of **2**. We find the same pattern over different sections of the QD indicating that it is single crystal.

Captions for Video S1 to S3

Video S1. Structural fluctuation of **2** on CNT (**Fig. 5f-j**).

This is the movie of **2** on CNT without drift and rotation correction. The experimental conditions: Acceleration voltage of 80 kV, electron dose rate of $2.8 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$, and exposure time of 20.0 milliseconds for each frame. The playback speed of the video is 25 fps, half of the original video recording. Scale bar is 1 nm.

Video S2. Structural fluctuation of **1** on CNT (**Fig. 5l**).

This is the movie of **1** on CNT without drift and rotation correction. The experimental conditions: Acceleration voltage of 80 kV, electron dose rate of $2.6 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$, and exposure time of 5.0 milliseconds for each frame or 200 fps. The playback speed of the video is 50 fps. Scale bar is 1 nm.

Video S3. Structural fluctuation of **1** on CNT (**Fig. 5m-n**).

This is the movie of **1** on CNT without drift and rotation correction. The experimental conditions: Acceleration voltage of 80 kV, electron dose rate of $2.1 \times 10^7 \text{ e}^- \text{ nm}^{-2} \text{ s}^{-1}$, and exposure time of 13.3 milliseconds for each frame. The playback speed of the video is 37.5 fps, half of the original video recording. Scale bar is 1 nm.

Supplementary References

-
- ⁴⁸ Ten Brinck, S. & Infante, I. Surface Termination, Morphology, and Bright Photoluminescence of Cesium Lead Halide Perovskite Nanocrystals. *ACS Energy Lett.* **1**, 1266-1272 (2016).
- ⁴⁹ Ravi, V. K. *et al.* Origin of the Substitution Mechanism for the Binding of Organic Ligands on the Surface of CsPbBr₃ Perovskite Nanocubes. *J. Phys. Chem. Lett.* **8**, 4988-4994 (2017).
- ⁵⁰ Galkowski, K. *et al.* Determination of the exciton binding energy and effective masses for methylammonium and formamidinium lead tri-halide perovskite semiconductors. *Energy Environ. Sci.* **9**, 962-970 (2016).
- ⁵¹ Oomens, J. & Steill, J. D. Free Carboxylates Stretching Modes. *J. Phys. Chem. A* **112**, 3281-3283 (2008).